

EFFECTS OF MANGANESE OXIDE NANOPARTICLES EMBEDMENT ON  
OPTIC-MAGNETIC FEATURES OF MAGNESIUM BOROTELLURITE  
DOPED EUROPIUM GLASS

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*...dedicated to my lovely parent:*

**AZIZ IBRAHIM ♥ CHE NONG SU**

*for their unconditional love, motivation and support..*

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## ABSTRACT

This thesis reports the modified optical and magnetic properties of europium doped and manganese oxide nanoparticles ( $\text{Mn}_3\text{O}_4$ NPs) embedded magnesium borotellurite glass. Glass samples with composition of  $(59-x)\text{TeO}_2-30\text{B}_2\text{O}_3-10\text{MgO}-1\text{Eu}_2\text{O}_3-x\text{Mn}_3\text{O}_4$  with  $0.0 \leq x \leq 2.0$  mol % and  $(59-y)\text{TeO}_2-30\text{B}_2\text{O}_3-10\text{MgO}-y\text{Eu}_2\text{O}_3-1\text{Mn}_3\text{O}_4$  with  $0.0 \leq y \leq 2.0$  mol % were prepared by melt-quenching technique. The amorphous nature of the glass was determined by X-Ray Diffractometer (XRD) and the presence of  $\text{Mn}_3\text{O}_4$  NPs was verified by using Transmission Electron Microscope (TEM) and High Resolution Transmission Electron Microscope (HRTEM). The thermal parameters were determined by Differential Thermal Analyzer (DTA) and spectroscopic properties were measured by Raman, Ultraviolet-Visible (UV-Vis) and Photoluminescence (PL) spectrometer. Magnetic properties were determined by Vibrating Sample Magnetometer (VSM) and Electron Spin Resonance (ESR) spectrometer. The XRD patterns confirmed the amorphous nature of all glasses and TEM images manifested the growth of  $\text{Mn}_3\text{O}_4$  NPs with average diameter 15 nm. HRTEM result revealed that the lattice spacing of  $\text{Mn}_3\text{O}_4$  NPs was 0.308 nm at (112) plane. The thermal analysis showed that the glass transition temperature,  $T_g$  increases with the increase of  $\text{Mn}_3\text{O}_4$  NPs and it was attributed to the arrangement of  $\text{Mn}_3\text{O}_4$  NPs in the glass matrix. The glass with 1.0 mol % of  $\text{Mn}_3\text{O}_4$  NPs and  $\text{Eu}_2\text{O}_3$  showed the highest thermal stability, 126 °C and the glass forming tendency, 0.76. The Raman spectra displayed  $\text{Mn}_3\text{O}_4$  NPs assisted alteration in the Te-O-Te,  $\text{BO}_3$ ,  $\text{BO}_4$ ,  $\text{TeO}_3$  trigonal pyramidal and  $\text{TeO}_4$  trigonal bipyramidal bonding vibrations. The UV-Vis spectra consist of three bands attributed to absorption from ground state ( $^7\text{F}_0$ ) to  $^5\text{D}_0$ ,  $^5\text{D}_1$  and  $^5\text{D}_2$  excited states. Two surface plasmon resonance (SPR) peaks of  $\text{Mn}_3\text{O}_4$  NPs were detected at 388 nm and 516 nm. The emission spectra of  $\text{Eu}^{3+}$  ion under 390 nm excitations revealed four prominent peaks centered at 591 nm, 614 nm, 651 nm and 700 nm assigned to the transitions from  $^5\text{D}_0 \rightarrow ^7\text{F}_J$  ( $J=1, 2, 3, 4$ ) states, respectively. Quenching effect in the luminescence intensity due to the incorporation of  $\text{Mn}_3\text{O}_4$  NPs was ascribed to the energy transfer from the  $\text{Eu}^{3+}$  ion to  $\text{Mn}_3\text{O}_4$  NPs. The calculated Judd-Ofelt intensity parameters ( $\Omega_\lambda$   $\lambda=2, 4, 6$ ), radiative parameter and stimulated emission cross section of  $\text{Eu}^{3+}$  ions were found to be strongly influenced by  $\text{Mn}_3\text{O}_4$  NPs. Prepared glass systems exhibit paramagnetic behavior with glass magnetization and susceptibility at room temperature in the range of  $(4.95-13.31 \times 10^{-2}) \text{ emug}^{-1}$  and  $(4.12-11.09 \times 10^{-6}) \text{ emuOe}^{-1}\text{g}^{-1}$  respectively. The ESR spectra of all glass samples exhibit two resonance signals with g values at 1.9 and 4.3 with higher signal observed at 1.9. In addition, correlations between the size of  $\text{Mn}_3\text{O}_4$  NPs with saturation magnetization ( $M_s$ ) and coercivity ( $H_c$ ) were established. In conclusion, incorporation of  $\text{Mn}_3\text{O}_4$  NPs in the glass system has improved the optical and magnetic properties of the glass.

## ABSTRAK

Tesis ini melaporkan sifat optik dan magnet terubah suai yang berdopkan europium dan zarah nano mangan oksida ( $\text{Mn}_3\text{O}_4$  NPs) berbenam kaca magnesium borotellurite. Sampel kaca dengan komposisi  $(59-x)\text{TeO}_2-30\text{B}_2\text{O}_3-10\text{MgO}-1\text{Eu}_2\text{O}_3-x\text{Mn}_3\text{O}_4$  dengan  $0.0 \leq x \leq 2.0$  mol % dan  $(59-y)\text{TeO}_2-30\text{B}_2\text{O}_3-10\text{MgO}-y\text{Eu}_2\text{O}_3-1\text{Mn}_3\text{O}_4$  dengan  $0.0 \leq y \leq 2.0$  mol % disediakan menggunakan teknik pelindapan leburan. Sifat amorfus kaca ditentukan oleh difraktometer Sinar-X (XRD) dan kehadiran  $\text{Mn}_3\text{O}_4$  NPs disahkan dengan menggunakan mikroskop elektron penghantaran (TEM) dan mikroskop elektron penghantaran beresolusi tinggi (HRTEM). Parameter terma ditentukan dengan penganalisa terma pembeza (DTA) dan sifat spektroskopi diukur dengan spektrometer Raman, penyerapan ultra lembayung boleh nampak (UV-Vis) dan fotoluminesens (PL). Sifat magnet ditentukan menggunakan magnetometer getaran sampel (VSM) dan spektrometer resonans putaran elektron (ESR). Corak XRD mengesahkan sifat amorfus bagi semua kaca dan imej TEM menunjukkan kewujudan  $\text{Mn}_3\text{O}_4$  NPs dengan diameter purata 15 nm. Keputusan HRTEM mendedahkan jarak kekisi bagi  $\text{Mn}_3\text{O}_4$  NPs ialah 0.308 nm pada satah (112). Analisis terma menunjukkan bahawa suhu peralihan kaca,  $T_g$  meningkat dengan pertambahan  $\text{Mn}_3\text{O}_4$  NPs yang disebabkan oleh susunan  $\text{Mn}_3\text{O}_4$  NPs dalam matriks kaca. Kaca dengan 1.0 mol %  $\text{Mn}_3\text{O}_4$  NPs dan  $\text{Eu}_2\text{O}_3$  menunjukkan kestabilan terma tertinggi, 126 °C dan kecenderungan pembentukan kaca tertinggi, 0.76. Spektrum Raman menunjukkan  $\text{Mn}_3\text{O}_4$  NPs mempengaruhi perubahan dalam getaran ikatan Te-O-Te,  $\text{BO}_3$ ,  $\text{BO}_4$ , piramid trigonal  $\text{TeO}_3$  dan bipyramidal trigonal  $\text{TeO}_4$ . Spektrum UV-Vis terdiri daripada tiga jalur yang berpadanan dengan penyerapan dari keadaan dasar ( $^7\text{F}_0$ ) kepada keadaan teruja  $^5\text{D}_0$ ,  $^5\text{D}_1$  dan  $^5\text{D}_2$ . Dua puncak resonans plasmon permukaan (SPR) bagi  $\text{Mn}_3\text{O}_4$  NPs dicerap pada 388 nm dan 516 nm. Spektrum pancaran ion  $\text{Eu}^{3+}$  apabila diuja pada 390 nm memaparkan empat puncak yang berpusat pada 591 nm, 614 nm, 651 nm dan 700 nm masing-masing merujuk kepada peralihan  $^5\text{D}_0 \rightarrow ^7\text{F}_J$  ( $J = 1, 2, 3, 4$ ). Kesan penurunan keamatan luminesens yang disebabkan oleh kehadiran  $\text{Mn}_3\text{O}_4$  NPs adalah disebabkan berlakunya pemindahan tenaga daripada  $\text{Eu}^{3+}$  ion kepada  $\text{Mn}_3\text{O}_4$  NPs. Pengiraan parameter keamatan Judd-Ofelt ( $\Omega_\lambda = 2, 4, 6$ ), parameter sinaran dan keratan rentas pancaran terangsang ion  $\text{Eu}^{3+}$  sangat dipengaruhi oleh  $\text{Mn}_3\text{O}_4$  NPs. Semua kaca yang disediakan bersifat paramagnet dengan kemagnetan kaca dan kerentanan pada suhu bilik masing-masing dalam julat  $(4.95-13.31 \times 10^{-2})$  emug<sup>-1</sup> dan  $(4.12-11.09 \times 10^{-6})$  emuOe<sup>-1</sup>g<sup>-1</sup>. Spektrum ESR menunjukkan semua kaca mempamerkan dua isyarat resonans dengan nilai g pada 1.9 dan 4.3 dengan isyarat resonans lebih tinggi pada 1.9. Sebagai tambahan, korelasi antara saiz  $\text{Mn}_3\text{O}_4$  NPs dengan pemagnetan tepu ( $M_s$ ) dan coerciviti ( $H_c$ ) telah ditentukan. Kesimpulannya, kemasukan  $\text{Mn}_3\text{O}_4$  NPs dalam sistem kaca telah menambahbaikkan sifat optik dan magnet kaca.

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## LIST OF SYMBOLS

$2\theta$	-	Angle of diffraction
$\Delta E$	-	Urbach energy
$\Delta H_{pp}$	-	Peak-to-peak line width
$\Delta S$	-	Thermal stability
$B$	-	Magnetic field
$d$	-	Crystal lattice planar spacing
$E$	-	Energy
$\vec{E}$	-	Electric field
$E_{opt}$	-	Optical band gap
$g$	-	Gyromagnetic
$H_c$	-	Coercivity
$H_r$	-	Resonance magnetic field
$h\nu$	-	Photon energy
$J$	-	Angular momentum
$M$	-	Magnetization
$M_s$	-	Remanence magnetization
$M_r$	-	Saturation magnetization
$n$	-	Refractive index
$O$	-	Orbital
$S$	-	Spin
$\alpha$	-	Absorption coefficient
$\mu_B$	-	Bohr magneton
$\nu$	-	Wavenumber
$\chi$	-	Susceptibility
$T_m$	-	Melting temperature

- $T_c$  - Crystallization temperature
- $T_g$  - Transition temperature

**LIST OF ABBREVIATIONS**

BO	-	Bridging oxygens
DTA	-	Differential thermal analysis
ESR	-	Electron spin resonance
ET	-	Energy transfer
GSA	-	Ground state absorption
HRTEM	-	High resolution transmission electron microscope
JCPDS	-	Joint Committee on Powder Diffraction Standard
J-O	-	Judd Ofelt
NBO	-	Non-bridging oxygen
NPs	-	Nanoparticles
NR	-	Nonradiative
PL	-	Photoluminescence
QEs	-	Quantum efficiency
R	-	Radiative
RE	-	Rare earth
SAED	-	Selected area electron diffraction
SPR	-	Surface plasmon resonance
tbp	-	Trigonal bipyramidal
TEM	-	Transmission electron microscope
tp	-	Trigonal pyramidal
UV-Vis	-	Ultra violet visible
VSM	-	Vibrating sample magnetometer
XRD	-	X-ray diffraction

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## CHAPTER 1

### INTRODUCTION

#### 1.1 Background of Study

Glasses are unique materials that have been used for thousands of years. Glass is defined as material that is produced through melting process and then being cooled to be a solid without going through a crystallization process [1]. Glass has been extensively investigated due to its high temperature resistance, high dielectric constant and good mechanical strength [2]. Furthermore, glass is not only known because of their excellent thermal and mechanical properties but they are potentially to become a good medium for luminescence due to the enhancement of the absorption efficiency of rare earth ions [3]. This excellent property has motivate researcher to further the study in optimizing luminescent and become more suitable material in the development of laser and solid state device.

Over the past few decades, tellurite glasses have gained so much interest over conventional silicate due to their high thermal expansion coefficients, excellent transmission in the visible as well as IR wavelength regions and low melting temperature [4]. These interesting properties making it feasible to be prepared at low temperature. Khafagy *et al.* [5] reported that the structure of tellurite glasses have two basic structural units, i.e as  $\text{TeO}_3$  and  $\text{TeO}_4$  units. The  $\text{TeO}_3$  units dominated in the glass structural network but as the  $\text{TeO}_2$  content increase,  $\text{TeO}_3$  units transform into  $\text{TeO}_{3+1}$  then to  $\text{TeO}_4$  units [6]. An intensive study on  $\text{TeO}_2$  containing glasses has been conducted because of their properties

such as exhibit high refractive indices, good chemical durability, better corrosion resistance and good mechanical strength [7].

Recently, tellurite glasses doped with boron oxide have received great scientific interest because these oxides significantly are able to change the properties of tellurite glasses [8][9]. Additionally, borate is one of the most attracting materials where their structural properties have been studied extensively [10]. Conversely, borate based glasses are prospective due to their flexible random network structure consisting of tetrahedral  $\text{BO}_4$  and trigonal boron  $\text{BO}_3$  units [11]. Besides, due to the excellent rare earth solubility, good infrared transmission and high thermal stability, borate glasses are attractive for the development of new optical devices [12]. However, the strong hygroscopic nature of borate glasses limits their applications [13]. This drawback can be surmounted by stabilizing the borate network with  $\text{TeO}_2$  incorporation, which may offer improved chemical durability via the structural modifications of the tetrahedral boron networks [14].

Previous study by Wang *et al.* [15] claimed that  $\text{TeO}_2$  is a conditional glass former which explain the incapability of that compound to form glass on its own. As a result, the introduction of borate into tellurite glass network simultaneously enhances the ability of glass formation. The combination of  $\text{TeO}_2$  and  $\text{B}_2\text{O}_3$  is an intrinsically interesting subject of study due to the stability of borotellurite (BT) compound [16]. Further, BT glass needs another element known as glass modifier such as alkaline earth metal oxide and transition metal oxide to improve the network connectivity then produce a stable BT glass [17] with increasing non-bridging oxygen (NBO). The addition of such modifiers would modify and increase the NBO, consequently open up the glass structure [18]. The substitution of network modifier such as  $\text{MgO}$  would produce stable BT glass [19].

To this day, rare earth ions (REIs) doped glass material becomes an interesting topic in luminescence material. The synthesis and characterizations of REIs doped binary and ternary glasses are intensively performed due to their advantages in developing efficient photonic devices [20]. Dehelean *et al.* [21] acknowledged that REIs doped glasses exhibit high brightness and improved

efficiency thus are very prospective for broad array of technological applications. Trivalent  $\text{Eu}^{3+}$  ion is a well-known activator with simple electronic transitions. The  $\text{Eu}^{3+}$  ions possess prominent laser emissions in the orange or red region [22] and narrow band emission [23] with longer lifetime. Thus, BT glass has emerged as a favorable host for accommodating large amount of REIs. Maheshvaran *et al.* [3] reported that  $\text{Eu}^{3+}$  doped BT glass has potential for red-emitting glass due to excellent luminescent properties and can be used as optical materials. Hence,  $\text{Eu}^{3+}$  doped glass has drawn much interest in technological applications especially for optoelectronic materials [24].

Despite many studies, the Judd-Ofelt (J-O) theory has not been applied uniformly to characterize the spectroscopic properties of  $\text{Eu}^{3+}$  ion doped glasses. Bo and Teruto [25] used  ${}^5\text{D}_0 \rightarrow {}^7\text{F}_2, {}^7\text{F}_4$  and  ${}^7\text{F}_6$  emission transitions of the  $\text{Eu}^{3+}$  ions to obtain the J-O parameters. Van Deun *et al.* [26] studied the absorption spectra of  $\text{Eu}^{3+}$  ion involving  ${}^7\text{F}_0 \rightarrow {}^5\text{D}_2, {}^5\text{D}_4$  and  ${}^5\text{L}_6$  transitions, where the calculated J-O parameters of  $\text{Eu}^{3+}$  ions exhibits good agreement with the measured oscillator strengths. However, the study of J-O for glass containing magnetic NPs accompanied with REIs has not much been reported or been discussed in literature. In the present study the method of Bo and Teruto [25] to calculate the J-O parameters of  $\text{Eu}^{3+}$  ions doped inside BT glasses containing  $\text{Mn}_3\text{O}_4$  NPs will be applied.

Currently, the development of multifunctional metallic NPs is in full swing [27] due to the feasibility of sundry applications. The research on the modifications in the structural, magnetic and optical properties of REIs doped glasses by embedding various metallic NPs is however still in progress. In this view, several attempts have been made to improve the optical properties of REIs doped glasses by incorporating various metallic NPs (magnetic and nonmagnetic). Lately, various magnetic NPs embedded glass systems are prepared such as Ni NPs incorporated samarium doped zinc phosphate glass [28] and  $\text{Fe}_3\text{O}_4$  NPs in erbium doped zinc phosphate glass [29]. The impact of  $\text{Ni}^{2+}$  and  $\text{Fe}^{2+}$  magnetic NPs on the physical, structural and magneto-optic properties of REIs doped binary glasses have accrescent interest. Literatures hinted that these types of glasses with tailored properties are

greatly potentials for the advancement of magneto-optic devices including isolators, switches and sensors [30].

Synthesis and characterization of magnetic (Fe, Co, Ni and Mn) NPs have ever-growing interest. Amongst all, Mn ions have been frequently used to improve the structural, electrical and magnetic properties of vitreous systems [31]. Manganese ions exist in different valence states in glassy matrices [32]. For example,  $\text{Mn}^{3+}$  ions in borate glasses exist with octahedral coordination whereas in silicate and germanate glasses they exist as  $\text{Mn}^{2+}$  ions with both octahedral and tetrahedral coordination [33]. Moreover, these well-known paramagnetic ions ( $\text{Mn}^{2+}$  and  $\text{Mn}^{3+}$ ) are identified as strong luminescence activators [34]. The incorporation of  $\text{Mn}_3\text{O}_4$  in glass has paramount importance due to their excellent physical and structural properties [31]. These transition metal ions contribute multi-valence states in the glass network and remarkably influence the properties. The appearance of super-paramagnetic behaviour, softness and large surface area of manganese nanoparticles make them highly potential magnetic materials [35] like manganese zinc ferrites useful for high density magnetic storage devices [36]. High saturation magnetization and magnetocrystalline anisotropy of  $\text{Mn}_3\text{O}_4$  NPs are useful for creating excellent traps for excited electrons in the glass host. Such electron trapping are advantageous to surmount data corruption in magnetic data storage [37].

Despite much research the impact of  $\text{Mn}_3\text{O}_4$  NPs on the structural properties of  $\text{Eu}^{3+}$  doped BT glass system is not yet explored. The incorporation of controlled concentrations of magnetic NPs in BT glass provides an opportunity to develop new stable magnetic glasses useful for nonlinear optical sensor and electronics devices. Additionally, literature shows that not many efforts are dedicated towards the incorporation of  $\text{Mn}_3\text{O}_4$  NPs in BT glass system for determining their role in improving the structural and optical properties. This motivated to investigate the REIs doped glasses containing  $\text{Mn}_3\text{O}_4$  NPs. Therefore, a new series of  $\text{Mn}_3\text{O}_4$  NPs embedded and  $\text{Eu}^{3+}$  ions doped BT glass systems are prepared and evaluate the  $\text{Mn}_3\text{O}_4$  NPs concentration dependent on thermal, structural and optical properties. Synthesized glasses are characterized using X-ray Diffraction (XRD), High Resolution Transmission Electron Microscopy (HRTEM), Differential Thermal

Analysis (DTA), Raman Spectroscopy, UV-Visible Spectroscopy and Photoluminescence (PL) Spectroscopy measurements. Meanwhile, Vibrating Sample Magnetometer (VSM) and Electron Spin Resonance (ESR) will be used to study the magnetic moment of the local magnetic properties due to the nature of spin-spin interaction.

## 1.2 Problem Statement

Plasmonic nanoglass is a new research paradigm with numerous application possibilities. Metallic NPs (such as gold, silver, copper, etc.) embedded REIs doped glass became attractive during past two decades. The inclusion of gold (Au) and silver (Ag) NPs in glass containing REIs [27][38] are exploited to enhance the quantum efficiency to achieve high performance lasing glass system. Yusoff and Sahar [39] have proposed that the enhancement of photoluminescence intensity is deduced by energy transfer from NPs to REIs and large local field in the vicinity of the REIs. Although significant effort continues in the development of the metallic NPs regrettably not many studies are dedicated towards the dispersing magnetic NPs such as manganese, nickel, ferrite and cobalt into rare earth doped glass. To overcome this limitation, the investigation of RE-doped glasses incorporated with magnetic NPs elements are suggested and become interest for magnetic studies.

Research of NPs effect on spectroscopic properties in RE doped oxides glasses have been widely studied [40][41]. It is known that, NPs have been able to enhance quantum efficiency which consequently leads to improve the optical characteristics of oxide glasses [38]. However, there are limited reports found about the influence of  $\text{Mn}_3\text{O}_4$  NPs in BT glass doped with optimum concentration of europium. Yet, their careful synthesis, detailed characterisation and systematic Judd Ofelt analysis are not reported extensively. Therefore, this research is also devoted to discover laser efficiency of  $\text{Eu}^{3+}$  ions doped BT glasses with NPs embedment via Judd Ofelt calculation.

In addition to that, the growth and nucleation of magnetic NPs favor the alteration of the local structure of the glass system. Carefully controlled size of  $\text{Mn}_3\text{O}_4$  NPs through appropriate heat treatment is one of the promising methods to modify and improve the magnetic properties of synthesized glass. The incorporation of  $\text{Mn}_3\text{O}_4$  NPs in the BT glass with tunable size and their effects on magnetic properties are not extensively studied. However it is a challenging task due to easy crystallization of discovered glass host [42]. In spite of that, this has motivated the researcher to investigate the RE doped glasses containing  $\text{Mn}_3\text{O}_4$  NPs elements. The role of  $\text{Mn}_3\text{O}_4$  NPs in the europium doped magnesium BT glass in optical and magnetic properties is far from being understood and requires further attention. There are only some reports on the influence of  $\text{Eu}^{3+}$  ions doped magnesium BT glass towards optical properties [43]. Therefore, the effects of  $\text{Mn}_3\text{O}_4$  NPs concentration changes on the thermal, structural and optical properties will be evaluated. Then, the magnetic properties are accomplished from the magnetic moment of the local magnetic properties due to the nature of spin-spin interaction.

### 1.3 Objectives of Study

The objectives of this study are as follows:

- i. To optimize the concentration of  $\text{Mn}_3\text{O}_4$  NPs embedded into europium doped magnesium BT glass via melt-quenching method.
- ii. To determine the influence of  $\text{Mn}_3\text{O}_4$  NPs concentration on the thermal, structural, optical and magnetic properties of  $\text{Eu}^{3+}$  doped BT glass.
- iii. To complement the experimental spectroscopic data using JO theoretical calculation of intensity parameter.

- iv. To control the size of  $\text{Mn}_3\text{O}_4$  NPs by varying heat treatment durations and investigate its effects on spectral and magnetic features modifications in these glass system.

#### 1.4 Scope of Study

In this study, 3 series of glass of different composition were prepared by melt quenching technique:

- a. Series I :  $(59-x)\text{TeO}_2-30\text{B}_2\text{O}_3-10\text{MgO}-1\text{Eu}_2\text{O}_3-x\text{Mn}_3\text{O}_4$  where  $(0.0 \leq x \leq 2.0$  in mol %)
- b. Series II :  $(59-y)\text{TeO}_2-30\text{B}_2\text{O}_3-10\text{MgO}-y\text{Eu}_2\text{O}_3-1\text{Mn}_3\text{O}_4$  where  $(0.0 \leq y \leq 2.0$  in mol %)
- c. Series III :  $58\text{TeO}_2-30\text{B}_2\text{O}_3-10\text{MgO}-1\text{Eu}_2\text{O}_3-1\text{Mn}_3\text{O}_4$  glasses at different heat treatment durations

These glasses are chosen due to the optimal performance of  $\text{TeO}_2\text{-B}_2\text{O}_3\text{-MgO-Eu}_2\text{O}_3$  glass system [44]. As increasing the concentration of  $\text{TeO}_2$ , BT glass will be more transparent. Nevertheless, the BT glass phase will become opaque if the concentration of  $\text{B}_2\text{O}_3$  is richer than  $\text{TeO}_2$ . Burger *et al.* [45] suggested that the optimum concentration of  $\text{TeO}_2$  is in the range of 3.9 to 73.6 mol % able to form glass in BT system. Meanwhile, Yardımcı *et al.* [46] determined that the glass forming range for concentration of  $\text{B}_2\text{O}_3$  in  $\text{B}_2\text{O}_3\text{-TeO}_2$  is system about 5 to 30 mol % and as the concentration increase, the crystallization starts to occur. The addition of MgO up to 15 mol % in glass system will increase the structural compactness of glass network [18]. The incorporation of MgO as network modifiers in BT glass cause the formation of more non bridging oxygen (NBO) and 1.0 mol % of  $\text{Eu}^{3+}$  as a dopant is the optimized concentration and have potential for solid state applications [47].

However, the role played by  $\text{Mn}_3\text{O}_4$  NPs as magnetic NPs in the structural features of the magnesium BT glass and the interaction of this element in the glass network is still a subject under study. Therefore, in current study the special attention is given to the inclusion of  $\text{Mn}_3\text{O}_4$  NPs in glass matrix to modify the thermal, structural, optical and magnetic properties of glass. The phase of the glass will be determined by X-ray Diffraction (XRD) and the existence and size of  $\text{Mn}_3\text{O}_4$  NPs in glass matrix investigated using High Resolution Transmission Electron Microscopy (HRTEM). Then, controlling the size of  $\text{Mn}_3\text{O}_4$  NPs through heat treatment process with different heat treatment durations above glass transition temperature,  $T_g$  which is determined from Differential Thermal Analysis (DTA). In terms of structural and optical properties influence by  $\text{Mn}_3\text{O}_4$  NPs will be determined by Raman spectroscopy, UV-Visible spectroscopy and Photoluminescence (PL) spectroscopy. Additionally, the magnetic properties will be accomplished from Vibrating Sample Magnetometer (VSM) and Electron Spin Resonance (ESR).

## 1.5 Significance of Study

Study of BT glass host embedded with  $\text{Mn}_3\text{O}_4$  NPs provides useful information on the advancement of the glass knowledge. The rapid quench process is used to achieve well transparent glasses which are physically and chemically stable and  $\text{Eu}^{3+}$  doped BT glass will improve the luminescence properties. Furthermore, embedding  $\text{Mn}_3\text{O}_4$  NPs in glass network as an alternative for understand the magnetic properties in order to determine the optimum concentration of  $\text{Mn}_3\text{O}_4$  NPs. This study is great importance to understand and explain the magnetic behaviour of  $\text{Mn}_3\text{O}_4$  NPs in glass. This may help in the discovery of a new magneto-device material that may contribute towards the development of NPs embedded RE doped photonics glass.

## REFERENCES

1. Sahar, M.R. (2000). *Fizik Bahan Amorfus*. (1st ed.). UTM Skudai: DBP.
2. Gandhi, Y., Kityk, I. V., Brik, M.G., Rao, P.R., Veeraiah, N. (2010). Influence of tungsten on the emission features of  $\text{Nd}^{3+}$ ,  $\text{Sm}^{3+}$  and  $\text{Eu}^{3+}$  ions in  $\text{ZnF}_2\text{-WO}_3\text{-TeO}_2$  glasses. *J. Alloys Compd.* 508, 278-291.
3. Maheshvaran, K., Veeran, P.K., Marimuthu, K. (2013). Structural and optical studies on  $\text{Eu}^{3+}$  doped boro-tellurite glasses. *Solid State Sci.* 17, 54-62.
4. Stambouli, W., Elhouichet, H., Gelloz, B., Fe, M. (2013). Optical and spectroscopic properties of Eu-doped tellurite glasses and glass ceramics. *J. Lumin.* 138, 201-208.
5. Khafagy, A.H., El-Adawy, A.A., Higazy, A.A., El-Rabaie, S., Eid, A.S. (2008). The glass transition temperature and infrared absorption spectra of:  $(70-x)\text{TeO}_2 + 15\text{B}_2\text{O}_3 + 15\text{P}_2\text{O}_5 + x\text{Li}_2\text{O}$  glasses. *J. Non. Cryst. Solids.* 354, 1460-1466.
6. Neov, S., Kozhukharov, V., Gerasimova, I., Krezhov, K., Sidzhimov, B. (1979). A model for structural recombination in tellurite glasses. *J. Phys. C: Solid State Phys.* 12, 715-718.
7. Ahmmad, S. Kareem, Samee, M.A., Edukondalu, A., Rahman, S. (2012). Physical and optical properties of zinc arsenic tellurite glasses. *Results Phys.* 2, 85, 175-181.
8. Paz, E.C., Lodi, T.A., Gomes, B.R.A., Melo, G.H.A. (2016). Optical and spectroscopic investigation on Calcium Borotellurite glass system. *Solid State Sci.* 55, 106-111.
9. Patil, S.D., Jali, V.M. (2013). Optical properties of Neodymium doped Borotellurite glasses. *Int. J. Sci. Res.* 1, 317-320.

10. Yano, T., Kunimine, N., Shibata, S., Yamane, M. (2003). Structural investigation of sodium borate glasses and melts by Raman spectroscopy. I. Quantitative evaluation of structural units. *J. Non. Cryst. Solids*. 321, 137–146.
11. Gaafar, M.S., Marzouk, S.Y., Zayed, H.A., Soliman, L.I., Serag El-Deen, A.H. (2013). Structural studies and mechanical properties of some borate glasses doped with different alkali and cobalt oxides. *Curr. Appl. Phys.* 13, 152-158.
12. Lakshminarayana, G., Kaky, K.M., Baki, S.O., Lira, A. (2017). Physical, structural, thermal, and optical spectroscopy studies of  $\text{TeO}_2\text{-B}_2\text{O}_3\text{-MoO}_3\text{-ZnO-R}_2\text{O}$  (R = Li, Na, and K)/MO (M = Mg, Ca, and Pb) glasses. *J. Alloys Compd.* 690, 799-816.
13. Said Mahraz, Z.A., Sahar, M.R., Ghoshal, S.K. (2014). Band gap and polarizability of boro-tellurite glass: Influence of erbium ions. *J. Mol. Struct.* 1072, 238-241.
14. Wang, J. S., Vogel, E.M. and Snitzer, E. (1994). 1.3 $\mu$  Emission of Neodymium and Praseodymium in Tellurite-Based Glasses, *J. Non-Cryst. Solids*. 178 109-113
15. Sudhakar Reddy, B., Hwang, H.-Y., Jho, Y.-D., Seung Ham, B. (2015). Optical properties of  $\text{Nd}^{3+}$ -doped and  $\text{Er}^{3+}\text{-Yb}^{3+}$  codoped borotellurite glass for use in NIR lasers and fiber amplifiers. *Ceram. Int.* 41, 3684-3692.
16. Azlan, M.N., Halimah, M.K., Shafinas, S.Z., Daud, W.M. (2015). Electronic polarizability of zinc borotellurite glass system containing erbium nanoparticles. *Mater. Express.* 5, 211-218.
17. Doweidar, H., El-Damrawi, G., Mansour, E., Fetouh, R.E. (2012). Structural role of MgO and PbO in  $\text{MgO-PbO-B}_2\text{O}_3$  glasses as revealed by FTIR; A new approach. *J. Non. Cryst. Solids*. 358, 941–946.
18. Smith, C.E., Brow, R.K. (2014). The properties and structure of zinc magnesium phosphate glasses. *J. Non. Cryst. Solids*. 390, 51-58.
19. Judd, B.R. (1962). Optical absorption intensities of rare-earth ions. *Phys. Rev.* 127, 750-761.
20. Dehelean, A., Culea, E. (2009). Magnetic behaviour of europium ions in some tellurite glasses obtained by the sol-gel method. *J. Phys. Conf. Ser.* 182, 12064.
21. Pisarski, W.A., Pisarska, J., Maczka, M., Ryba-Romanowski, W. (2006).

- Europium-doped lead fluoroborate glasses: Structural, thermal and optical investigations. *J. Mol. Struct.* 93, 207-211.
22. Sazali, E.S., Sahar, M.R., Ghoshal, S.K. (2013). Influence of Europium Ion on Structural, Mechanical and Luminescence Behavior of Tellurite Nanoglass. *J. Phys. Conf. Ser.* 431, 12008.
  23. Akamatsu, H., Fujita, K., Nakatsuka, Y., Murai, S., Tanaka, K. (2013). Magneto-optical properties of  $\text{Eu}^{2+}$ -containing aluminoborosilicate glasses with ferromagnetic interactions. *Opt. Mater.* 35, 1997-2000.
  24. Bo., P., Teturo, I., (1994). The Fluorescence Properties of  $\text{Eu}^{3+}$  in Various Glasses and the Energy Transfer Between  $\text{Eu}^{3+}$  and  $\text{Sm}^{3+}$  in Borosilicophosphate Glass. *Rev. Laser Eng.* 22, 16-27
  25. Van Deun, R., Binnemans, K., Görller-Walrand, C., Adam, J.L. (1999). Judd-Ofelt intensity parameters of trivalent lanthanide ions in a  $\text{NaPO}_3$ - $\text{BaF}_2$  based fluorophosphate glass. *J. Alloys Compd.* 283, 59-65.
  26. Ashur Said Mahraz, Z., Sahar, M.R., Ghoshal, S.K., Dousti, M.R., Amjad, R.J. (2013). Silver nanoparticles enhanced luminescence of  $\text{Er}^{3+}$  ions in borotellurite glasses. *Mater. Lett.* 112, 136-138.
  27. Azmi, S.A.M., Sahar, M.R. (2015). Optical response and magnetic characteristic of samarium doped zinc phosphate glasses containing nickel nanoparticles. *J. Magn. Magn. Mater.* 393, 341-346.
  28. Anigrahawati, P., Sahar, M.R., Ghoshal, S.K. (2015). Influence of  $\text{Fe}_3\text{O}_4$  nanoparticles on structural, optical and magnetic properties of erbium doped zinc phosphate glass. *Mater. Chem. Phys.* 152, 155-161.
  29. Malakhovskii, A. V., Edelman, I.S., Radzyner, Y., Yeshurun, Y. (2003). Magnetic and magneto-optical properties of oxide glasses containing  $\text{Pr}^{3+}$ ,  $\text{Dy}^{3+}$  and  $\text{Nd}^{3+}$  ions. *J. Magn. Magn. Mater.* 263, 161-172.
  30. Ramesh Babu, A., Rajyasree, C., Vinaya Teja, P.M., Yusub, S., Krishna Rao, D. (2012). Influence of manganese ions on spectroscopic and dielectric properties of  $\text{LiF-SrO-B}_2\text{O}_3$  glasses. *J. Non. Cryst. Solids.* 358, 1391-1398.
  31. Manzan, R.S., Donoso, J.P., Magon, C.J., Silva, I. d'Anciães A. (2015). Optical and Structural Studies of  $\text{Mn}^{2+}$  Doped  $\text{SbPO}_4$ - $\text{ZnO}$ - $\text{PbO}$  Glasses. *J. Braz. Chem. Soc.* 26, 2607-2614.

32. Kiran, N., Kesavulu, C.R., Suresh Kumar, A., Rao, J.L. (2011). Spectral studies on Mn<sup>2+</sup> ions doped in sodiumlead borophosphate glasses. *Phys. B Condens. Matter.* 406, 3816-3820.
33. Sumalatha, B., Omkaram, I., Rajavardhana Rao, T., Linga Raju, C. (2013). The structural, optical and magnetic parameter of manganese doped strontium zinc borate glasses. *Phys. B Condens. Matter.* 411, 99-105.
34. Portehault, D., Cassaignon, S., Jolivet, J. (2009). Structural and morphological control of manganese oxide nanoparticles upon soft aqueous precipitation through MnOMn<sup>2+</sup> reaction. *J. Mater. Chem.* 19, 2407-2416.
35. Edelman, I., Ivanova, O., Ivantsov, I., Velikanov, D., Pesktrakovskaja, E., Artemenko, A., Curely, J. (2011). Magnetic properties and morphology of manganese ferrite nanoparticles in glasses. *Mater. Sci. Engin.* 25, 012017.
36. Laffont, L., Gibot, P. (2010). High resolution electron energy loss spectroscopy of manganese oxides: Application to Mn<sub>3</sub>O<sub>4</sub> nanoparticles. *Mater. Charact.* 61, 1268-1273.
37. Sazali, E.S., Sahar, M.R., Ghoshal, S.K., Arifin, R. (2014). Optical properties of gold nanoparticle embedded Er<sup>3+</sup> doped lead-tellurite glasses. *J. Alloys Compd.* 607, 85-90.
38. Yusoff, N.M., Sahar, M.R. (2015). The incorporation of silver nanoparticles in samarium doped magnesium tellurite glass: Effect on the characteristic of bonding and local structure. *Phys. B Condens. Matter.* 470-471, 6-14.
39. Dousti, M.R., Poirier, G.Y., Camargo, A.S.S. (2015). Structural and spectroscopic characteristics of Eu<sup>3+</sup>-doped tungsten phosphate glasses. *Opt. Mater.* 45, 185-190.
40. Awang, A., Ghoshal, S.K., Sahar, M.R., Dousti, M.R. (2013). Enhanced spectroscopic properties and Judd Ofelt parameters of Er-doped tellurite glass : Effect of gold nanoparticles. *Curr. Appl. Phys.* 13, 1813-1818.
41. Rehana, P., Ravi, O., Ramesh, B., Dillip, G.R. (2016). Photoluminescence studies of Eu<sup>3+</sup> ions doped calcium zinc niobium borotellurite glasses. *Adv. Mater. Lett.* 7, 170-174.
42. Maheshvaran, K., Marimuthu, K. (2012). Concentration dependent Eu<sup>3+</sup> doped boro-tellurite glasses-Structural and optical investigations. *J. Lumin.* 132,

- 2259-2267.
43. H. Bürger, W. Vogel, V. Kozhukharov, M. Marinov. (1984). Phase equilibrium, glass-forming, properties and structure of glasses in the  $\text{TeO}_2$ - $\text{B}_2\text{O}_3$  system. *J. Mater. Sci.* 19, 403-412.
  44. Yardimci, D., Çelikbilek, M., Ersundu, A.E., Aydin, S. (2013). Thermal and microstructural characterization and crystallization kinetic studies in the  $\text{TeO}_2$ - $\text{B}_2\text{O}_3$  system. *Mater. Chem. Phys.* 137, 999-1006.
  45. Das, S., Amarnath Reddy, A., Vijaya Prakash, G., Near white light emission from  $\text{K}^+$  ion compensated  $\text{CaSO}_4:\text{Dy}^{3+}, \text{Eu}^{3+}$  phosphors. *Ceram. Int.* 2012, 38, 5769-5773.
  46. Dimitriev, Y., Kashchieva, E. (1975). Immiscibility in the  $\text{TeO}_2$ - $\text{B}_2\text{O}_3$  system. *J. Mater. Sci.* 10, 1419-1424.
  47. Elkhoshkhany, N., El-Mallawany, R. (2015). Optical and kinetics parameters of lithium boro-tellurite glasses. *Ceram. Int.* 41, 3561-3567.
  48. Swapna, Upender, G., Prasad, M. (2016). Raman, FTIR, thermal and optical properties of  $\text{TeO}_2$ - $\text{Nb}_2\text{O}_5$ - $\text{B}_2\text{O}_3$ - $\text{V}_2\text{O}_5$  quaternary glass system. *J. Taibah Univ. Sci.*
  49. Kaur, N., Khanna, A. (2014). Structural characterization of borotellurite and alumino-borotellurite glasses. *J. Non. Cryst. Solids.* 404, 116–123.
  50. Biirger, H., Kneipp, K., Vogel, W., Kozhukharov, V., Neov, S. (1992). Glass formation, properties and structure of glasses in the. *J. Non. Cryst. Solids.* 151, 134–142.
  51. Wang, J.S., Vogel, E.M. (1994). Tellurite glass: a new candidate for fiber devices. *Opt. Mater.* 3, 187-203.
  52. Suthanthirakumar, P., Karthikeyan, P., Manimozhi, P.K., Marimuthu, K. (2015). Structural and spectroscopic behavior of  $\text{Er}^{3+}/\text{Yb}^{3+}$  co-doped boro-tellurite glasses. *J. Non. Cryst. Solids.* 410, 26–34.
  53. Saddeek, Y.B., Gaafar, M.S. (2009). Physical and structural properties of some bismuth borate glasses. *Mater. Chem. Phys.* 115, 280-286.
  54. Suresh, S., Prasad, M., Upender, G., Kamalaker, V., Mouli, V.C. (2009). ESR, IR, Raman and optical absorption studies of  $60\text{B}_2\text{O}_3+10\text{TeO}_2+5\text{TiO}_2+24\text{R}_2\text{O}:\text{1CuO}$  (where  $\text{R}=\text{Li}, \text{Na}, \text{K}$ ) quaternary glasses. *Indian J. Pure Appl. Phys.* 47,

163–169.

55. Maheshvaran, K., Veeran, P.K., Marimuthu, K., Structural and optical studies on  $\text{Eu}^{3+}$  doped boro-tellurite glasses. *Solid State Sci.* 2013, 17, 54–62.
56. Halimah, M.K., Daud, W.M., Sidek, H. A A, Zainal, A. S. (2007). Structural analysis of borotellurite glass. *Am. J. Appl. Sci.* 4, 323-327.
57. Bhat, M.H., Ganguli, M., Rao, K.J. (2004). Investigation of the mixed alkali effect in boro-tellurite glasses -The role of NBO-BO switching in ion transport. *Curr. Sci.* 86, 676–691.
58. Rajkumar, G., Rajendran, V., Aravindan, S. (2012). Role of MgO on the HAp forming ability in phosphate based glasses. *Ceram. Int.* 38, 3781-3790.
59. Hussin, R., Salim, M.A., Alias, N.S., Abdullah, M.S. (2009). Vibrational Studies of Calcium Magnesium Ultraphosphate Glasses. *Sci. York.* 5, 41-53.
60. Watts, S.J., Hill, R.G., O'Donnell, M.D., Law, R. V. (2010). Influence of magnesia on the structure and properties of bioactive glasses. *J. Non. Cryst. Solids.* 356, 517-524.
61. Babu, A.M., Jamalaih, B.C., Suhasini, T., Rao, T.S., Moorthy, L.R. (2011). Optical properties of  $\text{Eu}^{3+}$  ions in lead tungstate tellurite glasses. *Solid State Sci.* 13, 574-578.
62. Sailaja, S., Dhoble, S.J., Sudhakar Reddy, B. (2011). Synthesis and photoluminescence properties of  $\text{Sm}^{3+}$  and  $\text{Dy}^{3+}$  ions activated  $\text{Ca}_2\text{Gd}_2\text{W}_3\text{O}_{14}$  phosphors. *J. Mol. Struct.* 1003, 115-120.
63. Anand Pandarinath, M., Upender, G., Narasimha Rao, K., Suresh Babu, D. (2016). Thermal, optical and spectroscopic studies of boro-tellurite glass system containing ZnO. *J. Non. Cryst. Solids.* 433, 60-67.
64. Melo, G.H.A., Dias, J.D.M., Lodi, T.A., Barboza, M.J. (2016). Optical and spectroscopic properties of  $\text{Eu}_2\text{O}_3$  doped CaBAI glasses. *Opt. Mater.* 54, 98-103.
65. Graça, M.P.F., Valente, M.A., Silva, C.C., Peres, M. (2009). Synthesis and optical properties of a lithium niobosilicate glass doped with europium. *Mater. Sci. Eng. C.* 29, 894-898.
66. Arunkumar, S., Marimuthu, K. (2013). Structural and luminescence studies on  $\text{Eu}^{3+}$ :  $\text{B}_2\text{O}_3$ - $\text{Li}_2\text{O}$ -MO-LiF (M=Ba, Bi<sub>2</sub>, Cd, Pb, Sr<sub>2</sub> and Zn) glasses. *J. Lumin.*

- 139, 6-15.
67. Awang, A., Ghoshal, S.K., Sahar, M.R., Arifin, R., Nawaz, F. (2014). Non-spherical gold nanoparticles mediated surface plasmon resonance in  $\text{Er}^{3+}$  doped zinc-sodium tellurite glasses: Role of heat treatment. *J. Lumin.* 149, 138-143.
  68. Amjad, R.J., Dousti, M.R., Sahar, M.R., Shaukat, S.F. (2014). Silver nanoparticles enhanced luminescence of  $\text{Eu}^{3+}$ -doped tellurite glass. *J. Lumin.* 154, 316-321.
  69. Said Mahraz, Z.A., Sahar, M.R., Ghoshal, S.K. (2015). Enhanced luminescence from silver nanoparticles integrated  $\text{Er}^{3+}$ -doped boro-tellurite glasses: Impact of annealing temperature. *J. Alloys Compd.* 649, 1102–1109.
  70. Reza Dousti, M., Sahar, M.R., Ghoshal, S.K., Amjad, R.J., Samavati, A.R. (2013). Effect of AgCl on spectroscopic properties of erbium doped zinc tellurite glass. *J. Mol. Struct.* 1035, 6-12.
  71. Mohan Babu, A., Jamalalah, B.C., Sasikala, T., Saleem, S.A., Rama Moorthy, L. (2011). Absorption and emission spectral studies of  $\text{Sm}^{3+}$ -doped lead tungstate tellurite glasses. *J. Alloys Compd.* 509, 4743-4747.
  72. Zhang, J.Z., Noguez, C. (2008). Plasmonic optical properties and applications of metal nanostructures. *Plasmonics.* 3, 127-150.
  73. Awang, A., Ghoshal, S.K., Sahar, M.R., Arifin, R., Nawaz, F. (2014). Non-spherical gold nanoparticles mediated surface plasmon resonance in  $\text{Er}^{3+}$  doped zinc-sodium tellurite glasses: Role of heat treatment. *J. Lumin.* 149, 138–143.
  74. Jiménez, J. A., Sendova, M. (2012). In situ isothermal monitoring of the enhancement and quenching of  $\text{Sm}^{3+}$  photoluminescence in Ag co-doped glass. *Solid State Commun.* 152, 1786-1790.
  75. Amjad, R.J., Dousti, M.R., Sahar, M.R., Shaukat, S.F. (2014). Silver nanoparticles enhanced luminescence of  $\text{Eu}^{3+}$ -doped tellurite glass. *J. Lumin.* 154, 316–321.
  76. Bose, V.C., Biju, V. (2015). Optical, electrical and magnetic properties of nanostructured  $\text{Mn}_3\text{O}_4$  synthesized through a facile chemical route. *Phys. E Low-dimensional Syst. Nanostructures.* 66, 24-32.
  77. Widanarto, W., Sahar, M.R.R., Ghoshal, S.K.K., Arifin, R. (2013). Effect of natural  $\text{Fe}_3\text{O}_4$  nanoparticles on structural and optical properties of  $\text{Er}^{3+}$  doped

- tellurite glass. *J. Magn. Magn. Mater.* 326, 123-128.
78. Azmi, S.A.M., Sahar, M.R. (2015). Optical response and magnetic characteristic of samarium doped zinc phosphate glasses containing nickel nanoparticles. *J. Magn. Magn. Mater.* 393, 341-346.
  79. Ghoshal, S. K., Zake, N. S. M., Arifin, R. (2015). Optical and structural behavior of  $\text{Sm}^{3+}$  doped tellurite glass containing  $\text{Mn}_3\text{O}_4$  NPs. *Solid State Sci. Technol.* 23, 82-89.
  80. Pike, J., Hanson, J., Zhang, L., Chan, S. (2007). Synthesis and redox behavior of nanocrystalline Hausmannite ( $\text{Mn}_3\text{O}_4$ ). *Chem. Mater.* 42, 5609-5616.
  81. Silva G. C., Almeida, F. S., Ferreira A. M., Ciminellia. V, S. T. (2012). Preparation and Application of a Magnetic Composite ( $\text{Mn}_3\text{O}_4/\text{Fe}_3\text{O}_4$ ) for Removal of As (III) from Aqueous Solutions. *Mater. Res.* 15, 403-408.
  82. Yusub, S., Srinivasa Rao, P., Krishna Rao, D. (2016). Ionic conductivity, dielectric and optical properties of lithium lead borophosphate glasses combined with manganese ions. *J. Alloys Compd.* 663, 708-717.
  83. Hashim, S. P. H. S., Sidek, H. A. A., Halimah, M. K., Matori, K.A., Yusoff, W.M.D.W. (2011). Physical properties of borotellurite glass doped with Manganese. *Solid State Sci. Technol.* 19, 342-347
  84. Nurbaisyatul, E.S., Azman, K., Azhan, H., Razali, W.A.W., Noranizah, A. (2014). The structural properties of trivalent rare earth ions ( $\text{Er}^{3+}$ ) doped borotellurite glass. *J. Teknol. Sciences Eng.* 69, 97-100.
  85. Kole, A. K., Kumbhakar, P. (2012). Effect of manganese doping on the photoluminescence characteristics of chemically synthesized zinc sulfide nanoparticles. *Appl. Nanosci.* 2, 15-23.
  86. Tang, J., Albrecht, A.C. (1970). *Raman Spectroscopy*. (1st ed.) New York: Springer, Boston, MA.
  87. Kundu, R.S., Dhankhar, S., Punia, R., Nanda, K., Kishore, N. (2014). Bismuth modified physical, structural and optical properties of mid-IR transparent zinc boro-tellurite glasses. *J. Alloys Compd.* 587, 66-73.
  88. Kesavulu, C.R., Chakradhar, R.P.S., Jayasankar, C.K., Rao, J.L. (2010). EPR, Optical, photoluminescence studies of  $\text{Cr}^{3+}$  ions in  $\text{Li}_2\text{O}-\text{Cs}_2\text{O}-\text{B}_2\text{O}_3$  glasses- An evidence of mixed alkali effect. *J. Mol. Struct.* 975, 93-99.

89. Dimitrov, V., Sakka, S. (1996). Electronic oxide polarizability and optical basicity of simple oxides. I. *J. Appl. Phys.* 79, 1736.
90. Amjad, R.J., Sahar, M.R., Ghoshal, S.K., Dousti, M.R. (2012). Enhanced infrared to visible upconversion emission in Er<sup>3+</sup> doped phosphate glass: Role of silver nanoparticles. *J. Lumin.* 132, 2714-2718.
91. Vázquez-Olmos, A., Redón, R., Rodríguez-Gattorno, G., Mata-Zamora, M.E. (2005). One-step synthesis of Mn<sub>3</sub>O<sub>4</sub> nanoparticles: Structural and magnetic study. *J. Colloid Interface Sci.* 291, 175-180.
92. Reisfeld, R., Pietraszkiewicz, M., Saraidarov, T., Levchenko, V. (2009). Luminescence intensification of lanthanide complexes by silver nanoparticles incorporated in sol-gel matrix. *J. Rare Earths.* 27, 544-549.
93. Rivera, V.A.G., El-Amraoui, M., Ledemi, Y., Messaddeq, Y., Marega, E. (2014). Expanding broadband emission in the near-IR via energy transfer between Er<sup>3+</sup>-Tm<sup>3+</sup> co-doped tellurite-glasses. *J. Lumin.* 145, 787-792.
94. Martínez, P. L. H., Govorov, A., Demir, H. V. (2017). *Understanding and Modeling Förster-type Resonance Energy Transfer (FRET)*. (2nd ed.) Singapore: Springer Singapore.
95. William, M. Y., Shionoya, S. Yamamoto, H. (2007). *Fundamentals of phosphors*. (1st ed.). New York: Taylor & Francis Group, LLC.
96. Sreedhar, V.B., Basavapoornima, C., Jayasankar, C.K. (2014). Spectroscopic and fluorescence properties of Sm<sup>3+</sup>-doped zincfluorophosphate glasses. *J. Rare Earths.* 32, 918-926.
97. Sreedhar, B., Sumalatha, C., Kojima, K. (1995). EPR and optical absorption spectra of some paramagnetic ions in lithium fluoroborate glasses. *J. Non. Cryst. Solids.* 192-193, 203-206.
98. Luo, W., Liao, J., Li, R., Chen, X. (2010). Determination of Judd-Ofelt intensity parameters from the excitation spectra for rare-earth doped luminescent materials. *Phys. Chem. Chem. Phys.* 12, 3276-3282.
99. Nawaz, F., Sahar, M.R., Ghoshal, S.K., Awang, A., Ahmed, I. (2014). Concentration dependent structural and spectroscopic properties of Sm<sup>3+</sup>/Yb<sup>3+</sup> co-doped sodium tellurite glass. *Phys. B Condens. Matter.* 433, 89-95.
100. Jorgensen, C.K., Reisfeld, R. (1983). Judd-Ofelt parameters and chemical

- bonding. *J. Less-Common Met.* 93, 107-112.
101. Ofelt, G.S. (1962). Intensities of crystal spectra of rare-earth ions. *J. Chem. Phys.* 37, 511-520.
  102. Vijayakumar, R., Marimuthu, K. (2016). Luminescence studies on Ag nanoparticles embedded Eu<sup>3+</sup> doped boro-phosphate glasses. *J. Alloys Compd.* 665, 294-303.
  103. Qiao, X., Luo, Q., Fan, X., Wang, M. (2008). Local vibration around rare earth ions in alkaline earth fluorosilicate transparent glass and glass ceramics using Eu<sup>3+</sup> probe. *J. Rare Earths.* 26, 883-888.
  104. Ferhi, M., Bouzidi, C., Horchani-Naifer, K., Elhouichet, H., Ferid, M. (2015). Judd-Ofelt analysis of spectroscopic properties of Eu<sup>3+</sup> doped KLa(PO<sub>3</sub>)<sub>4</sub>. *J. Lumin.* 157, 21-27.
  105. Arunkumar, S., Venkata Krishnaiah, K., Marimuthu, K. (2013). Structural and luminescence behavior of lead fluoroborate glasses containing Eu<sup>3+</sup> ions. *Phys. B Condens. Matter.* 416, 88-100.
  106. Selvaraju, K., Marimuthu, K., Seshagiri, T.K., Godbole, S. V. (2011). Thermal, structural and spectroscopic investigations on Eu<sup>3+</sup> doped boro-tellurite glasses. *Mater. Chem. Phys.* 131, 204-210.
  107. Hervault, A., Thanh, N.T.K. (2014). Magnetic nanoparticle-based therapeutic agents for thermo-chemotherapy treatment of cancer. *Nanoscale.* 6, 11553-11573.
  108. Widanarto, W., Sahar, M.R., Ghoshal, S.K., Arifin, R. (2013). Effect of natural Fe<sub>3</sub>O<sub>4</sub> nanoparticles on structural and optical properties of Er<sup>3+</sup> doped tellurite glass. *J. Magn. Magn. Mater.* 326, 123-128.
  109. Weckhuysen, B. M., Heidler, R., Schoonheydt, R.A. (2004). *Electron Spin Resonance Spectroscopy.* (1st ed.) New York: Springer, Berlin, Heidelberg.
  110. Ardelean, I., Mureşan, N., Păşcuţă, P. (2007). EPR and magnetic susceptibility studies of manganese ions in 70TeO<sub>2</sub>-25B<sub>2</sub>O<sub>3</sub>-5SrO glass matrix. *Mater. Chem. Phys.* 101, 177-181.
  111. Du, J., Gao, Y., Chai, L., Zou, G. (2006). Hausmannite Mn<sub>3</sub>O<sub>4</sub> nanorods: synthesis, characterization and magnetic properties. *Nanotechnology.* 17, 4923-4928.

112. Yusub, S., Srinivasa Rao, P., Krishna Rao, D. (2016). Ionic conductivity, dielectric and optical properties of lithium lead borophosphate glasses combined with manganese ions. *J. Alloys Compd.* 663, 708-717.
113. Massera, J., Sevrette, B., Petit, L., Koponen, J. (2014). Effect of partial crystallization on the thermal, optical, structural and  $\text{Er}^{3+}$  luminescence properties of silicate glasses. *Mater. Chem. Phys.* 147, 1099-1109.
114. Turba, T., Norton, M.G., Niraula, I., McIlroy, D.N. (2009). Ripening of nanowire-supported gold nanoparticles. *J. Nanoparticle Res.* 11, 2137–2143.
115. Viswanatha, R., Sarma, D., Rao, C., Müller, N. (2007). Growth of nanocrystals in a solution, *Nanomater. Chem.* 18, 139-170.
116. Shirsath, S.E., Mane, M.L., Yasukawa, Y., Liu, X., Morisako, A. (2014). Self-ignited high temperature synthesis and enhanced super-exchange interactions of  $\text{Ho}^{3+}$ - $\text{Mn}^{2+}$ - $\text{Fe}^{3+}$ - $\text{O}^{2-}$  ferromagnetic nanoparticl. *Phys. Chem. Chem. Phys.* 16, 2347-2357.
117. Wendlandt, W. W. (1986). *Thermal Analysis*. (3rd ed.). New York: WileyInterscience.
118. Gayathri Pavani, P., Sadhana, K., Chandra Mouli, V. (2011). Optical, physical and structural studies of boro-zinc tellurite glasses. *Phys. B Condens. Matter.* 406, 1242-1247.
119. Gazzali, P.M., Kanimozhi, V., Priyadharsini, P., Chandrasekaran, G. (2014). Structural and Magnetic properties of Ultrafine Magnesium Ferrite Nanoparticles. *Adv. Mater. Res.* 938, 128-133.
120. Djerdj, I., Arcon, D., Jaglicic, Z., Nedererberger, M. (2007). Nonaqueous Synthesis of Manganese Oxide Nanoparticles, Structural Characterization, and Magnetic Properties. *J. Phys. Chem C.* 111, 3614-3623.
121. Yang, F., Zhao, M., Sun, Q., Qiao, Y. (2015). A novel hydrothermal synthesis and characterisation of porous  $\text{Mn}_3\text{O}_4$  for supercapacitors with high rate capability. *RSC Adv.* 5, 9843-9847.
122. Yusoff, N.M., Sahar, M.R., Ghoshal, S.K. (2015).  $\text{Sm}^{3+}$ :Ag NPs assisted modification in absorption features of magnesium tellurite glass. *J. Mol. Struct.* 1079, 167-172.
123. Vemasevana Raju, K., Sailaja, S., Nageswara Raju, C., Sudhakar Reddy, B.

- (2011). Optical characterization of  $\text{Eu}^{3+}$  and  $\text{Tb}^{3+}$  ions doped cadmium lithium alumino fluoro boro tellurite glasses. *Spectrochim. Acta -Part A Mol. Biomol. Spectrosc.* 79, 87-91.
124. Ravi, O., Reddy, C.M., Manoj, L., Prasad, B.D. (2012). Structural and optical studies of  $\text{Sm}^{3+}$  ions doped niobium borotellurite glasses. *J. Mol. Struct.* 1029, 53–59.
125. Amjad, R.J., Sahar, M.R., Dousti, M.R., Ghoshal, S.K., Jamaludin, M.N. (2013). Surface enhanced Raman scattering and plasmon enhanced fluorescence in zinc-tellurite glass. *Opt. Express.* 21, 14282-90.
126. Davis, E.A., Mott, N.F. (2017). Conduction in non-crystalline systems V. Conductivity, optical absorption and photoconductivity in amorphous semiconductors. *J. Mol. Struct.* 179, 0903-0922.
127. Faznny, M.F., Halimah, M.K., Azlan, M.N. (2016). Effect of Lanthanum Oxide on Optical Properties of Zinc Borotellurite Glass System. *J. Optoelectron. Biomed. Mater.* 8, 49-59.
128. Maheshvaran, K., Arunkumar, S., Venkata Krishnaiah, K., Marimuthu, K. (2014). Investigations on luminescence behavior of  $\text{Er}^{3+}/\text{Yb}^{3+}$  co-doped borotellurite glasses. *J. Mol. Struct.* 1079, 130-138.
129. Lakshminarayana, G., Kaky, K.M., Baki, S.O., Lira, A. (2017). Physical, structural, thermal, and optical spectroscopy studies of  $\text{TeO}_2\text{-B}_2\text{O}_3\text{-MoO}_3\text{-ZnO-R}_2\text{O}$  ( $\text{R} = \text{Li, Na, and K}$ )  $\text{MO}$  ( $\text{M} = \text{Mg, Ca, and Pb}$ ) glasses. *J. Alloys Compd.* 690, 799-816.
130. Azmi, S.A.M., Sahar, M.R., Ghoshal, S.K., Arifin, R. (2015). Modification of structural and physical properties of samarium doped zinc phosphate glasses due to the inclusion of nickel oxide nanoparticles. *J. Non. Cryst. Solids.* 411, 53-58.
131. Selvi, S., Marimuthu, K., Murthy, N.S., Muralidharan, G. (2016). Red light generation through the lead boro telluro phosphate glasses activated by  $\text{Eu}^{3+}$  ions. *J. Mol. Struct.* 1119, 276–285.
132. Selvi, S., Marimuthu, K., Muralidharan, G. (2015). Structural and luminescence behavior of  $\text{Sm}^{3+}$  ions doped lead boro-telluro-phosphate glasses. *J. Lumin.* 159, 207-218.

133. Nurhafizah, H., Rohani, M.S., Ghoshal, S.K. (2016). Er<sup>3+</sup>:Nd<sup>3+</sup> concentration dependent spectral features of lithium-niobate-tellurite amorphous media. *J. Non. Cryst. Solids*. 443, 23-32.
134. Bandi, V.R., Grandhe, B.K., Woo, H.-J., Jang, K. (2012). Luminescence and energy transfer of Eu<sup>3+</sup> or/and Dy<sup>3+</sup> co-doped in Sr<sub>3</sub>AlO<sub>4</sub>F phosphors with NUV excitation for WLEDs. *J. Alloys Compd.* 538, 85-90.
135. Zhu, C., Chaussedent, S., Liu, S., Zhang, Y. (2013). Composition dependence of luminescence of Eu and Eu/Tb doped silicate glasses for LED applications. *J. Alloys Compd.* 555, 232-236.
136. Silva, G.H., Anjos, V., Bell, M.J. V, Carmo, A.P. (2014). Eu<sup>3+</sup> emission in phosphate glasses with high UV transparency. *J. Lumin.* 154, 294-297.
137. Tanko, Y.A., Ghoshal, S.K., Sahar, M.R. (2016). Ligand field and Judd-Ofelt intensity parameters of samarium doped tellurite glass. *J. Mol. Struct.* 1117, 64-68.
138. Ferhi, M., Bouzidi, C., Horchani-Naifer, K., Elhouichet, H., Ferid, M. (2014). Judd-Ofelt analysis and radiative properties of LiLa(1-x)Eux(PO<sub>3</sub>)<sub>4</sub>. *Opt. Mater.* 37, 607-613.
139. Xiangping Li, Baojiu Chen, Rensheng Shen, Haiyang Zhong. (2011). Fluorescence quenching of <sup>5</sup>D<sub>J</sub> (J = 1, 2 and 3) levels and Judd-Ofelt analysis of Eu<sup>3+</sup> in NaGdTiO<sub>4</sub> phosphors. *J. Phys. D : Appl. Phys.* 44, 1-6.
140. Sreedhar, V.B., Basavapoornima, C., Jayasankar, C.K. (2014). Spectroscopic and fluorescence properties of Sm<sup>3+</sup>-doped zincfluorophosphate glasses. *J. Rare Earths*. 32, 918-926.
141. Wan, M.H., Wong, P.S., Hussin, R., Lintang, H.O., Endud, S. (2014). Structural and luminescence properties of Mn<sup>2+</sup> ions doped calcium zinc borophosphate glasses. *J. Alloys Compd.* 595, 39-45.
142. Vijayakumar, R., Maheshvaran, K., Sudarsan, V., Marimuthu, K. (2014). Concentration dependent luminescence studies on Eu<sup>3+</sup> doped telluro fluoroborate glasses. *J. Lumin.* 154, 160-167.
143. Rivera, V. A G., Ledemi, Y., El-Amraoui, M., Messaddeq, Y., Marega, E. (2014). Green-to-red light tuning by up-conversion emission via energy transfer in Er<sup>3+</sup> -Tm<sup>3+</sup>-codoped germanium-tellurite glasses. *J. Non. Cryst.*

- Solids*. 392-393, 45-50.
144. Maheshvaran, K., Linganna, K., Marimuthu, K. (2011). Composition dependent structural and optical properties of  $\text{Sm}^{3+}$  doped boro-tellurite glasses. *J. Lumin.* 131, 2746-2753.
  145. Zhang, L., Zhang, Y. (2009). Fabrication and magnetic properties of  $\text{Fe}_3\text{O}_4$  nanowire arrays in different diameters. *J. Magn. Magn. Mater.* 321, 15-20.
  146. Berkowitz, A.E., Rodriguez, G.F., Hong, J.I., An, K. (2008). Monodispersed  $\text{MnO}$  nanoparticles with epitaxial  $\text{Mn}_3\text{O}_4$  shells. *J. Phys. D: Appl. Phys.* 41, 134007.
  147. Buckelew, B.A., Galun-mascarós, J.R., Dunbar, K.R. (2002). Facile Conversion of the Face-Centered Cubic Spinel Oxide  $\text{Mn}_3\text{O}_4$  at the Solid/Water. *Adv. Mater.* 14, 1646-1648.
  148. Tadić, M., Panjan, M., Marković, D., Milošević, I., Spasojević, V. (2011). Unusual magnetic properties of  $\text{NiO}$  nanoparticles embedded in a silica matrix. *J. Alloys Compd.* 509, 7134-7138.
  149. Liang, X., Shi, H., Jia, X., Yang, Y., Liu, X. (2011). Dispersibility, Shape and Magnetic Properties of Nano- $\text{Fe}_3\text{O}_4$  Particles. *Mater. Sci. Appl.* 2, 1644-1653.
  150. Issa, B., Obaidat, I.M., Albiss, B.A., Haik, Y. (2013). Magnetic nanoparticles: Surface effects and properties related to biomedicine applications. *Int. J. Mol. Sci.* 14, 21266-21305.
  151. Wakde, G.C., Kakde, A.S., Gaikar, P.S., Dudhe, C.M., Arjunwadkar, P.R. (2016). Influence of Aluminum Doping on the Magnetic Properties of Li-Zn Spinel Ferrite. *Appl. Sci. Engin. Technol.* 4, 311-316.
  152. Edelman, I.S., Ivanova, O.S., Petrakovskaja, E. A., Velikanov, D. A. (2015). Formation, characterization and magnetic properties of maghemite  $\gamma\text{-Fe}_2\text{O}_3$  nanoparticles in borate glasses. *J. Alloys Compd.* 624, 60-67.
  153. Djerdj, I., Arcon, D., Jaglicic, Z., Nedererberger, M. (2007). Nonaqueous Synthesis of Manganese Oxide Nanoparticles, Structural Characterization, and Magnetic Properties. *J. Phys. Chem C.* 111, 3614-3623.
  154. Singh, V., Chakradhar, R.P.S., Rao, J.L., Kim, D.K. (2008). EPR and luminescence properties of combustion synthesized  $\text{LiAl}_5\text{O}_8\text{:Mn}$  phosphors. *Mater. Chem. Phys.* 110, 43-51.

155. Sambasivam, S., Li, G.J., Jeong, J.H., Choi, B.C. (2012). Structural, optical, and magnetic properties of single-crystalline  $Mn_3O_4$  nanowires. *J. Nanopart Res.* 14, 1138
156. Srisittipokakun, N., Kedkaew, C., Kaewkhao, J. (2009). Electron Spin Resonance (ESR) and Optical Absorption Spectra of a Manganese Doped Soda-Lime-Silicate Glass System. *Kasetsart J. (Nat. Sci.)*. 43, 360-364.
157. Sreekanth Chakradhar, R.P., Sivaramaiah, G., Rao, J.L., Gopal, N.O. (2005). EPR and optical investigations of manganese ions in alkali lead tetraborate glasses. *Spectrochim. Acta - Part A Mol. Biomol. Spectrosc.* 62, 761-768.
158. Augustin, M., Fenske, D., Bardenhagen, I., Westphal, A. (2015). Manganese oxide phases and morphologies: A study on calcination temperature and atmospheric dependence. *Beilstein J. Nanotechnol.* 6, 47-59.
159. Naseri, M.G., Saion, E.B., Hashim, M., Shaari, A.H., Ahangar, H.A. (2011). Synthesis and characterization of zinc ferrite nanoparticles by a thermal treatment method. *Solid State Commun.* 151, 1031-1035.
160. Nicolae, E., Pascuta, P., Pustan, M., Tamas-gavrea, D.R. (2015). Effects of Eu : Ag codoping on structural, magnetic and mechanical properties of lead tellurite glass ceramics. *J. Non. Cryst. Solids.* 408, 18-25.
161. Wang, M.-C., Cheng, H.-Z., Lin, H.-J., Wang, C.-F., Hsi, C.-S. (2013). Crystallization and magnetic properties of a  $10Li_2O-9MnO_2-16Fe_2O_3-25CaO-5P_2O_5-35SiO_2$  glass. *Mater. Chem. Phys.* 140, 16-23.
162. Said Mahraz, Z.A., Sahar, M.R., Ghoshal, S.K. (2016). Impact of annealing time on silver nanoparticles growth assisted spectral features of erbium-zinc-boro-tellurite glass. *J. Lumin.* 180, 1-7.
163. Hu, P., Zhang, S., Wang, H., Pan, D. (2011). Heat treatment effects on  $Fe_3O_4$  nanoparticles structure and magnetic properties prepared by carbothermal reduction. *J. Alloys Compd.* 509, 2316-2319.
164. Soltani, I., Hraiech, S., Horchani-Naifer, K., Elhouichet, H. (2016). Growth of silver nanoparticles stimulate spectroscopic properties of  $Er^{3+}$  doped phosphate glasses: Heat treatment effect. *J. Alloys Compd.* 686, 556-563.
165. Mahraz, Z.A.S., Sahar, M.R., Ghoshal, S.K., Reduction of non-radiative decay rates in boro-tellurite glass via silver nanoparticles assisted surface plasmon

- impingement: Judd Ofelt analysis. *J. Lumin.* 2017, 190, 335-343.
166. Langar, A., Bouzidi, C., Elhouichet, H., Férid, M. (2014). Er-Yb codoped phosphate glasses with improved gain characteristics for an efficient 1.55  $\mu\text{m}$  broadband optical amplifiers. *J. Lumin.* 148, 249-255.
167. Qi, J., Xu, T., Wu, Y., Shen, X. (2013). Ag nanoparticles enhanced near-IR emission from  $\text{Er}^{3+}$  ions doped glasses. *Opt. Mater.* 35, 2502-2506.
168. Sun, H., Ge, F., Zhao, J., Cai, Z. (2016). Template-directed synthesis of hierarchically mesoporous superparamagnetic carbon-coated nickel nanoplates. *Mater. Lett.* 164, 152-155.
169. Abbas, M., Nazrul Islam, M. and C.K. (2014). *High Magnetization Superparamagnetic Core-Shell Nanoparticles*. P.Sabbas, N. (Ed.) *Magnetic Nanoparticles: Synthesis, Physicochemical Properties and Role in Biomedicine* (pp.113-116). United States of America: Nova Science Publisher, Inc.
170. Alagiri, M., Muthamizhchelvan, C., Ponnusamy, S. (2011). Structural and magnetic properties of iron, cobalt and nickel nanoparticles. *Synth. Met.* 161, 1776-1780.
171. Ai, L., Jiang, J. (2010). Influence of annealing temperature on the formation, microstructure and magnetic properties of spinel nanocrystalline cobalt ferrites. *Curr. Appl. Phys.* 10, 284-288.
172. Rajendran, M., Pullar, R.C., Bhattacharya, A.K., Das, D. (2001). Magnetic properties of nanocrystalline  $\text{CoFe}_2\text{O}_4$  powders prepared at room temperature: Variation with crystallite size. *J. Magn. Magn. Mater.* 232, 71-83.
173. Roy, S., Dubenko, I., Edoth, D.D., Ali, N. (2004). Size induced variations in structural and magnetic properties of double exchange  $\text{La}_{0.8}\text{Sr}_{0.2}\text{MnO}_3$  nano-ferromagnet. *J. Appl. Phys.* 96, 1202-1208.
174. Roca, A.G., Niznansky, D., Poltierova-Vejpravova, J., Bittova, B. (2009). Magnetite nanoparticles with no surface spin canting. *J. Appl. Phys.* 88, 105-118
175. Ajroudi, L., Mliki, N., Bessais, L., Madigou, V. (2014). Magnetic, electric and thermal properties of cobalt ferrite nanoparticles. *Mater. Res. Bull.* 59, 49-58.
176. Battistini, L., Benasciutti, R., Tassi, A. (1994). Effects of heat treatment on crystallographic and magnetic properties of magnetic steels. *J. Magn. Magn.*

*Mater.* 133, 603-606.

177. Bretcanu, O., Verné, E., Cöisson, M., Tiberto, P., Allia, P. (2006). Temperature effect on the magnetic properties of the coprecipitation derived ferrimagnetic glass-ceramics. *J. Magn. Magn. Mater.* 300, 412-417.
178. Shankwar, N., Kothiyal, G.P., Srinivasan, A. (2014). Understanding the magnetic behavior of heat treated CaO-P<sub>2</sub>O<sub>5</sub>-Na<sub>2</sub>O-Fe<sub>2</sub>O<sub>3</sub>-SiO<sub>2</sub> bioactive glass using electron paramagnetic resonance studies. *Phys. B Condens. Matter.* 448, 132-135.
179. Bhowmik, R.N., Satya, A.T., Bharathi, A. (2013). Synthesis of Co<sub>1.5</sub>Fe<sub>1.5</sub>O<sub>4</sub> spinel ferrite with high magnetic squareness and study its magnetic property by annealing the chemical routed sample at different temperatures. *J. Alloys Compd.* 559, 134-141.
180. Yang, L., Wang, Z., Zhai, B., Shao, Y. (2013). Magnetic properties of Eu<sup>3+</sup> lightly doped ZnFe<sub>2</sub>O<sub>4</sub> nanoparticles. *Ceram. Int.* 2013, 39, 8261-8266.