First-principles calculations of antimony sulphide Sb$_2$S$_3$

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**Abstract**

The structural, electronic, as well as the optical properties of the Sb2S3, are investigated using full-potential (FP) linearized augmented plane wave (LAPW) method framed within density functional theory (DFT) via treating the exchange-correlation potential with Engel-Vosko generalized gradient approximation (EV-GGA). Electronic properties calculations were performed with and without taking into account the effects of spin-orbit coupling (SOC). From our results, we found that structural properties, the density of states and band structure are in good agreement with experimental results. The effects of SOC on electronic properties were found to be negligible. The calculated optical properties, namely, imaginary and real parts of the dielectric function, reflectivity, absorption coefficient, refractive index, extinction coefficient and energy loss function and their analysis showed; Sb2S3 metal chalcogenide is a promising material for solar cells.

**Keywords:** Density Functional Theory, LAPW, Antimony Sulphide, Spin-orbit coupling, Electronic properties, Optical properties

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**INTRODUCTION**

Metal chalcogenides materials such as cadmium telluride (CdTe) and copper indium gallium selenide (CIGS) have been known as an excellent candidate for high-performance photovoltaic devices like a solar cell. Though metal chalcogenides materials are highly efficient, the high cost, as well as high toxicity, limit their applications [1,2]. Therefore, low-cost, abundant and less toxic materials are needed [3]. A recent study showed that antimony sulphide (Sb$_2$S$_3$) hold great potential for photovoltaic application, due to its band gap having value in the range 1.65 to 2.4 eV and high absorption coefficient in the visible region [4], obtained through a change in structure dimension [5-6]. Sb$_2$S$_3$ semiconductor material belongs to metal chalcogenides group which is cheaper, abundant, and less toxic than any other compound in the group. Therefore, detailed knowledge of electronic and optical properties is essential for understanding its optoelectronic behavior such as band gap and light absorption. The band gap of a material determines its electrical conductivity [7] and optical properties that show the interaction between matter and electromagnetic radiation [8].

The role of the computational methods has been exploited to give an accurate prediction of the material properties in the numerous cases [9]. Several computational works have been reported on Sb$_2$S$_3$ using different exchange-correlation functionals for getting reliable band structure and band gap value, but none of them included the effect of spin-orbit coupling (SOC). As spin-orbit interaction effect in heavy elements is substantial, therefore inclusion is important [10-12]. In this paper, with the inclusion of SOC, the calculation of band structure, the density of state (DOS), partial density of states (PDOS) and optical properties are computed by employing FP-LAPW method [13], realized in WIEN2k code [14]. The optical properties such as refractive index, extinction coefficient, absorption coefficient, energy-loss, and reflectivity obtained from the calculation of dielectric function are found in better consistency with the reported experimental results.

**COMPUTATIONAL METHOD**

To compute the structural, electronic as well as the optical properties of Sb$_2$S$_3$ in the orthorhombic crystal structure, FP-LAPW method as realized in WIEN2k code is used along with the inclusion of fully relativistic and SOC effects. The simulated Sb$_2$S$_3$ crystallized structure in the orthorhombic crystal unit cell belonging to the space-group Pnma (62) is shown in Fig. 1. The initial structure parameters (lattice parameters and atomic positions) are used from the experimental study of Kyono and Kimata [15]. The electron exchange-correlation functional was treated by the Engel-Vosko generalized gradient approximation (EV-GGA) [16]. The muffin tin (MT) spheres radii of 2.0 and 2.43 a.u were chosen for S and Sb, respectively. The convergence parameter $R_{\text{MT}}$ was set to be 9.0 shown in Eq. 1 for controlling the size of the basis set. The value of another parameter, $G_{\text{max}}$ in Eq. 2 was set to 12, where $G_{\text{max}}$ is taken as the largest vector for...
the Fourier expansion of the charge density [17]. To perform self-consistent integration over the Brillouin zone (BZ), tetrahedron method was used with 28 k-points in the irreducible BZ corresponding to 250 k-points in the whole BZ. The self-consistent calculations were considered to be converged when the total energy was stabilized within 0.00001 Ry. For optical properties, a uniformly distributed denser k-point mesh of 1000 were used.

$$\Psi = \sum_{k_{\text{MAX}}} c_{k_n} e^{iK_n r}$$   \hspace{1cm} (1)

$$\rho(r) = \sum_{G_{\text{MAX}}} \rho_G e^{iG r}$$   \hspace{1cm} (2)

RESULTS AND DISCUSSION

Structural optimization
Investigation of the structural properties is the first step in DFT calculations to avoid a certain error for accurate predictions of other quantities. The lattice parameter $a$, $b$ and $c$ of Sb$_2$S$_3$ in orthorhombic structure were determined via energy optimization and is shown along with other previous experimental and theoretical values as can be seen in Table 1. Our calculated results reasonably comparable with the experimental measurements.

Electronic Structure
The band structure calculation of the Sb$_2$S$_3$ along high-symmetry points of the first Brillouin zone are plotted as shown in Fig. 2. The direct gap of Sb$_2$S$_3$ with and without SOC is of 1.64 eV and 1.60 eV respectively. Table 2 represents a summary of the band gap results obtained in this work along with available theoretical and experimental data from other researchers [4-5]. The obtained band gap with SOC is very close to experimental value [5] compared with other previous DFT calculations. Similarly, our results with EV-GGA approximation are much better than that the results obtained at the level of generalized gradient approximation by Perdew, Burke and. Ernzerhof (PBE-GGA) functional. However, our results are lower than the reported experimental measurement [6], this may be due to the limitation of the first-principles approach based on DFT [18] or experimental measurements have some issue as well because compare to the PBE-GGA, GGA functional constructed by Engel and Vosko usually show improvement in reproducing the band gap closer to the experimental band gap. The SOC normally reduces the band gap. Our obtained results with SOC are in line as reported by Pourghazi et. al for chalcogenide compound [19]. The results of total densities of states (DOS) helps to further elaborate the nature of band structure whereas the partial density of states (PDOS) gives information about the origin of bands. Our calculated DOS and partial DOS for Sb$_2$S$_3$ with the Fermi energy level represented by a reference line at zero energy is shown in Fig. 3. The results of DOS reveals that p-orbitals of Sb and S are the main contributors to the occupied states.

![Fig. 2 Bandstructure of Sb2S3 without (a) and with SOC (b)](image)

| Table 1 | Calculated equilibrium lattice constants of Sb$_2$S$_3$ |
|---|---|---|
| **Ref.** | $a$ (Å) | $b$ (Å) | $c$ (Å) |
| This work | 11.646 | 3.953 | 11.587 |

<table>
<thead>
<tr>
<th>Table 2</th>
<th>Calculated and experimental energy gaps of Sb$_2$S$_3$</th>
</tr>
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<tbody>
<tr>
<td><strong>Band gap energy (eV)</strong></td>
<td><strong>Ref.</strong></td>
</tr>
<tr>
<td>1.64</td>
<td>EV-GGA</td>
</tr>
<tr>
<td>1.60</td>
<td>EV-GGA (SOC)</td>
</tr>
<tr>
<td>1.20</td>
<td>PBE-GGA [20]</td>
</tr>
<tr>
<td>1.65</td>
<td>Exp [5]</td>
</tr>
<tr>
<td>2.24</td>
<td>Exp [6]</td>
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</table>
Optical properties

Investigation of the optical properties of materials is essential for understanding their optoelectronic nature. The electronic and ionic contributions to the real \( \varepsilon(\omega) \) and imaginary \( \varepsilon'(\omega) \) parts of the frequency dependent dielectric function, \( \varepsilon(\omega) = \varepsilon(\omega) + i\varepsilon'(\omega) \) as a function of incident photon can be determined from the band structure calculation via Wien2k code with the help of Kramers-Kronig relations [21]. With the knowledge of frequency dependent dielectric function, optical quantities, namely, absorption coefficient \( \alpha(\omega) \), energy loss function \( L(\omega) \), refractive index \( n(\omega) \), extinction coefficient \( k(\omega) \) and reflectivity \( R(\omega) \) can be obtained as can be seen in the following expressions:

\[
\varepsilon_1(\omega) = 1 + \frac{2}{\pi} \int_0^\infty d\omega' \frac{\omega'^2 \varepsilon_2(\omega')}{\omega'^2 - \omega^2} \tag{3}
\]

\[
n = \sqrt{\frac{\sqrt{\varepsilon_1^2(\omega) + \varepsilon_2^2(\omega)} + \varepsilon_1(\omega)}{2}} \tag{4}
\]

\[
k = \sqrt{\frac{\varepsilon_1^2(\omega) + \varepsilon_2^2(\omega) - \varepsilon_1(\omega)}{2}} \tag{5}
\]

\[
\alpha(\omega) = \frac{\omega}{c} \sqrt{2 \left( \varepsilon_1^2(\omega) + \varepsilon_2^2(\omega) - \varepsilon_1(\omega) \right)} \tag{6}
\]

\[
L(\omega) = \frac{\varepsilon_2(\omega)}{\varepsilon_1(\omega) + \varepsilon_2(\omega)} \tag{7}
\]

\[
R(\omega) = \left| \frac{\sqrt{\varepsilon(\omega) - 1} \varepsilon(\omega) + 1}{\sqrt{\varepsilon(\omega) + 1}} \right|^2 \tag{8}
\]

A single crystal of Sb\(_2\)S\(_3\) is the optically trixial system because it has orthorhombic structure crystal structure. Therefore, its linear dielectric tensor of the Sb\(_2\)S\(_3\) have three independent components. As shown in the Fig. 4(a) any of the directions has different value for dielectric function constant. The electronic at high frequency and ionic contributions of a non-polar system is contained in the static dielectric permittivity tensor \( \varepsilon(0) \). The dielectric constant \( \varepsilon(\omega) \) at high frequency obtained in x, y and z direction are 9.29434, 131.547, and 126.892, respectively. In the imaginary part of the dielectric function, the onset of the absorption edge \( \varepsilon'(\omega) \) called optical absorption edge occurred at 1.61 eV for x, y and z direction, which is due to the inter-band transition between conduction band minimum and valence band maximum states as can be seen in Fig. 4 (b). This value is corresponds to fundamental band gap shown in the Table 2. Moreover spectrum shape and optical gap value is found in quite good agreement with experimental results as well[4].

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Fig. 3 Calculated total density of states (DOS) with and without SOC (a) partial density of states (PDOS) of Sb\(_2\)S\(_3\) for Sb and S with SOC (b)

Fig 4 The calculated (a) real \( \varepsilon(\omega) \) and (b) imaginary \( \varepsilon'(\omega) \) parts of the complex dielectric constant of Sb\(_2\)S\(_3\).
The energy dependence absorption coefficient $\alpha(\omega)$, refractive index $n(\omega)$, extinction coefficient $k(\omega)$, energy-loss $L(\omega)$, and reflectivity $R(\omega)$ are given in Fig. 5 (a)-(e). These results are derived from the calculations of the $\varepsilon(\omega)$. From the Fig. 5 (a), the absorption coefficient for x, y, z-direction are 172.674 cm$^{-1}$, 152.248 cm$^{-1}$ and 167.867 cm$^{-1}$ respectively. Absorption coefficient along y-direction is close to the experimental measurement of 139.76 cm$^{-1}$. Fig. 5(b) displays the changes of refractive index with photon energy. As shown in Fig. 5 (b), the calculated $n(0)$ in x, y, z-direction are found to be 3.054, 3.638 and 3.566 respectively. The average value of $n(0)$ was found to be 3.29 which is higher as compared to the previously reported results, 2.087 [22], this may be due to the corresponding somewhat different DFT calculations. We observed that extinction coefficient $k(\omega)$ and imaginary part of dielectric function $\varepsilon_2(\omega)$ have a similar trend as can be seen in Fig 5(c). Fig. 5 (d) is a graph of electron energy loss function $L(\omega)$ with respect to the photon energy, this function describes energy loss of a fast moving electron through a material while moving from the top of a valence band to the bottom of the conduction band. The observed sharp maxima in the graph of the $L(\omega)$ can be linked to the existence of plasma oscillations [23]. The resonant energy loss is seen at 18.898 eV, 18.898 eV and 18.844 eV in x, y, and z-directions respectively. Optical reflectivity is calculated and shown in Fig. 5(e). The main peaks are roughly at 5.728 eV, 3.415 eV, 3.850 eV. From the figure, it can be seen that the reflectivity is high in the visible light region (1.65 - 3.1 eV) of the electromagnetic spectrum which makes the Sb$_2$S$_3$ compound a good material for photovoltaic applications.

![Graphs of optical constants](image)
CONCLUSION

In summary, the structural, electronic and optical properties of the Sb2S3 crystalline compound were calculated using the FP-LAPW approach at the level of the EV-GGA exchange-correlation functional. The calculated lattice parameters were found closer to the experimental measurements. The fully relativistic effect within spin-orbit coupling (SOC) was included in the calculations of the electronic and optical properties. The calculated band structure provides an excellent description of the semiconducting behavior of Sb2S3 with the band gap obtained the value (1.60 eV) in close agreement with experimental measurements. Projected density of states reveals that the p-orbitals of Sb and S atoms are major contributors for conduction band. Optical properties calculations indicate that the optical gap is almost the same as energy band gap from the bandstructure calculation. Optical gap of 1.61 eV shows that Sb2S3 is a promising material for solar cell application.

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REFERENCES