

PARTICLE SIZE ANALYSIS ON GINGER ESSENTIAL OIL NANOEMULSIONS

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DEDICATION

“My dearest mother, father, husband and son”

This is for all of you

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ABSTRACT

Ginger essential oil has various biological properties such as antioxidant and anti-inflammatory. However, ginger essential oils has its own weaknesses, such as high volatility, low absorption, and poor water-solubility. Transdermal drug delivery is one of the alternatives to transport ginger essential oil into the body. The transdermal drug delivery system which is a nanoemulsion was introduced to overcome the weakness of essential oil. The droplet size of a nanoemulsion is an important property which determines the stability and ease of penetration. In this research, the nanoemulsions were prepared using a spontaneous emulsification method. The effect of preparation conditions and system composition on particle size of ginger essential oil nanoemulsions were examined. In organic phase, ginger essential oil and surfactants which are Tween 40, 60 and 80 were used. In aqueous phase, distilled water and co-solvent (glycerol) were used. For nanoemulsion formation, the organic phase was slowly added into the aqueous phase while being stirred at 500 rpm. The surfactant type had a major effect on particle size, where the smallest droplets particle size were formed by using Tween 80 (15.40 nm). The surfactant concentration also shows a great impact on particle size, where at surfactant-to-emulsion ratio (SER) 25 %, the smallest droplets were formed (11.3 nm). By increasing the temperature of organic phase and stirring speed, the particle size was reduced with the smallest droplets being formed at 90 °C (11.16 nm) and at 800 rpm (11.23 nm). Co-solvent addition also had shown an impact on particle size where at 10 % of co-solvent concentration, the smallest droplets were formed (11.22 nm). For thermodynamic stability, nanoemulsions with SER 15 %, 20 %, 25 % and 30% had shown a great stability with no phase and size separation. For storage stability, the droplets particle size were increased by 23 % throughout the two months of storage. In conclusion, a smaller droplet particle (< 15 nm) can be formed by optimizing the system composition and homogenization conditions of nanoemulsions.

ABSTRAK

Minyak pati halia mempunyai pelbagai sifat-sifat biologi seperti antioksidan dan anti-radang. Walau bagaimanapun, minyak pati halia mempunyai kelemahan tersendiri seperti pemeruwapan yang tinggi, penyerapan yang rendah, dan kelarutan air yang lemah. Penghantaran dadah secara transdermal adalah salah satu alternatif untuk menghantar minyak pati halia ke dalam badan. Sistem penghantaran dadah transdermal iaitu nanoemulsi diperkenalkan untuk mengatasi kelemahan minyak pati. Saiz titisan nanoemulsi adalah sifat penting iaitu yang menentukan kestabilan dan penembusan yang lebih mudah. Dalam kajian ini, nanoemulsi telah disediakan menggunakan kaedah pengemulsian spontan. Kesan syarat penyediaan dan komposisi sistem pada saiz zarah nanoemulsi minyak halia telah disiasat. Dalam fasa organik, minyak halia dan surfaktan iaitu Tween 40, 60 dan 80 digunakan. Dalam fasa akueus, air suling dan pelarut bersama (gliserol) digunakan. Untuk pembentukan nanoemulsi, fasa organik ditambah perlahan-lahan ke dalam fasa akueus sambil dikacau pada 500 rpm. Jenis surfaktan memberi pengaruh pada saiz zarah, di mana saiz titisan terkecil saiz zarah dibentuk dengan menggunakan Tween 80 (15.40 nm). Kepekatan surfaktan juga menunjukkan kesan besar pada saiz zarah, di mana pada nisbah surfaktan-ke-emulsi (SER) 25%, titisan terkecil terbentuk (11.3 nm). Dengan meningkatkan suhu fasa organik dan kelajuan pengacau, saiz zarah dikurangkan dengan titisan terkecil dibentuk pada 90 °C(11.16 nm) dan pada 800 rpm (11.23 nm). Penambahan pelarut bersama juga menunjukkan kesan pada saiz zarah di mana pada 10% kepekatan pelarut bersama, titisan terkecil dibentuk (11.22 nm). Untuk kestabilan termodinamik, nanoemulsi dengan SER 15%, 20%, 25% dan 30% telah menunjukkan kestabilan yang baik tanpa pemisahan fasa dan saiz. Untuk kestabilan penyimpanan, saiz zarah titisan meningkat 23% sepanjang dua bulan penyimpanan. Kesimpulannya, zarah titisan yang lebih kecil (< 15 nm) dapat dibentuk dengan mengoptimimumkan komposisi sistem dan keadaan homogenisasi nanoemulsi.

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LIST OF ABBREVIATION

AFM	-	Atomic force microscope
API	-	Active pharmaceutical ingredient
BAC	-	Benzalkonium chloride
BHA	-	Butylated hydroxyanisole
BZT	-	Benzethonium chloride
CMC	-	Critical micelle concentration
CTAB	-	Cetyl trimethylammonium bromide
CO ₂	-	Carbon dioxide
DPPH	-	2, diphenyl – picrylhydrazyl
EPI	-	Emulsion phase inversion
FTC	-	Ferric thiocyanate
GC-MS	-	Gas chromatography mass spectrometry
HLB	-	Hydrophile – lipophile balance
NF- κ B	-	Nuclear factor kappa light chain enhancer of activated B cells
NLC	-	Nanostructured lipid carriers
NSAIDs	-	Nonsteroidal anti-inflammatory drugs
O/W	-	Oil-in-water
O/W/O	-	Oil-in-water-in-oil
PAMAM	-	Polyamidoamine
PDI	-	Polydispersity Index
PIC	-	Phase inversion composition
PIT	-	Phase inversion temperature
PFOA	-	Perfluorooctanoate
PFOS	-	Perfluorooctanesulfonate
ROS	-	Reactive oxygen species
SDS	-	Sodium dodecyl sulphate
SE	-	Spontaneous emulsification
SER	-	Surfactant-to-emulsion ratio
SLES	-	Sodium lauryl ether sulphate
SLN	-	Solid lipid nanoparticles

STM	-	Scanning tunneling microscope
TBA	-	Thiobarturic acid
W/O	-	Water-in-oil
W/O/W	-	Water-in-oil-in-water

LIST OF SYMBOLS

atm	-	Standard atmosphere
°C	-	Temperature
D	-	Diameter
Φ	-	Phase volume ratio
%	-	Percentage
eV	-	Electron volt
Hz	-	Hertz
min	-	minute
nm	-	Nanometre
rpm	-	Revolutions per minute
U	-	Unified atomic mass unit

CHAPTER 1

INTRODUCTION

1.1 Background of Study

From the past century, it is proven that nanotechnology has the great potential to bring benefits to pharmaceutical industry for its ability in formulation development and drug delivery system (Amiji, 2007). Nanopharmaceuticals have become huge industry because of its ability to overcome solubility and stability issues in drug delivery. With rising interest in nanopharmaceutical, researchers have produce great number of improvement and widening the range of drug delivery systems. These systems can improve the stability, absorption, and therapeutic concentration of the drug within the target tissue and allow the long-term release of the drug at the target site (Chaudhari, 2012; Kaiser *et al.*, 2005). These advancement in drug delivery have facilitated the targeting of specific tissues. The most common drug delivery routes are oral and parenteral routes. The oral route has the advantage of pre-determined doses, portability and patient self-administration. Therefore, the oral route has become the most convenient in delivering the medications (Brambilla *et al.*, 2014). However, there is some limitations to oral route which are large amount of drug is lost in the vicinity of the target organ, drugs are exposed to the first-past effect and highly dependent on patient compliance (Prausnitz and Langer, 2008; Rastoga and Yadav, 2012). That limitations can possibly be overcome by advanced drug delivery methodologies such as transdermal drug delivery.

Transdermal drug delivery is one of the systems lying under the category of controlled drug delivery, with the objective to deliver the drug through the skin in a predetermined and controlled rate. It has various advantages such as prolonged therapeutic effect, improved bioavailability, reduced side-effects, better patient compliance and easy termination of drug therapy (Rastoga and Yadav, 2012). There

are two main route of drug penetration which are intercellular and transcellular routes. There are some factors that need to be considered while using this delivery system such as biological, physicochemical and environmental factors.

Nowadays, the type of nanostructures that is being used have shown a significant increased. There are wide range of drug delivery systems that are available for various type of drug. Liposomes, polymeric micelles, dendrimers and nanoemulsions are some of potential nanostructures used for drug delivery. Liposomes are lipid bilayer system that can accommodate hydrophilic drug inside the core and lipophilic drug between the bilayer (Escobar-Chavez, 2012). Its nontoxic characteristic and it remain in the bloodstream for a long period of time make it to be one of the best alternatives for drug delivery system (Bakowsky *et al.*, 2008). Polymeric micelles are macromolecular assembly that form from synthetic block copolymer and has a spherical inner core and outer core (Yokoyama, 2011). It has a large solubilisation power, higher stability in bloodstream, more loading capacity and longevity (Ahmad, *et al.*, 2014). Dendrimers are nanostructures that are made up of a series of branches around an inner core. They were chosen to be drug delivery system because of their functionalization, ease of preparation, and the ability to exhibit multiple versions of surface groups for biological identification process.

Nanoemulsions are one of the nanostructures and part of the drug delivery system. Nanoemulsion are isotropic dispersed systems of two immiscible liquids, usually containing of an oily system dispersed in an aqueous system (oil-in-water), or an aqueous system dispersed in an oily system (water-in-oil) which forming droplet (Escobar-Chávez, 2012). There are three types of nanoemulsions. The types are oil-in-water (O/W) nanoemulsions, water-in-oil (W/O) nanoemulsions and also bicontinuous nanoemulsions (W/O/W) (Mishra *et al.*, 2014; Singh *et al.*, 2017). The types of nanoemulsions formed can be predicted by the type of surfactants used weather it is water soluble or oil soluble (Singh *et al.*, 2017). There are three main components in nanoemulsions which are oils, surfactants and aqueous phase. The oil is used to form internal phase of the emulsion. The oil phase can carry the drugs compound in soluble form. Usually, the oils used are mineral oil, vegetable oil, medium-chain triglycerides, and squalene (Lee *et al.*, 2010). Surfactants are used as

the stabilizer in nanoemulsions by reducing interfacial tension and prevent droplet aggregation (Singh *et al.*, 2017). Tweens 20, 40, 60, and 80 (Polyoxyethylene sorbitan monolaurate), Spans 20, 40, 60, 80 (Sorbitan monolaurate), and cremophor EL (Polyoxyl-35 castor oil) are example of surfactants that have been used to produce nanoemulsions (Singh *et al.*, 2017). For the aqueous phase, a purified water are usually used (Lee *et al.*, 2010).

Nanoemulsions can be fabricate by using various methods. The methods are divided into two categories which are high energy methods and low energy methods. In high energy methods, it involve the use of mechanical devices to produce intense disruptive forces which lead to smaller droplets size (Solan and Solé, 2012). Most commonly used high energy methods for producing nanoemulsions are high pressure homogenization, ultrasonification and microfluidization (Mishra *et al.*, 2014; Singh *et al.*, 2017; Saberi *et al.*, 2013). In low energy methods, the formation of droplets particle are depends on the surfactant-oil-water mixture in specific condition and environment. Some of low energy methods that have been introduced for fabrication of nanoemulsions are spontaneous emulsification, phase inversion temperature and emulsion phase inversion (McClements and Rao, 2011; Solan and Solé, 2012).

There are more than 3000 types of essential oils that have been produced but only 300 types of essential oils that is important for a commercial purpose (Bakkali *et al.*, 2008). Essential oils have been used for many application such as pharmaceutical, perfumes, sanitary products, agriculture, dentistry, as food preservative and flavour additives, cosmetics and natural remedies (Burt, 2004; Bakkali *et al.*, 2008). Essential oils are a volatile, natural, have a pungent odour and produce from a different parts of the plant (flowers, barks, buds, seeds, leaves, wood, fruits and roots) (Perricone *et al.*, 2015; Burt, 2004). They are normally stored in different anatomical parts of the plant such as oil ducts, resin ducts, and grandular trichomes (Baser and Demirci, 2007; El Asbahani *et al.*, 2015). An individual essential oils contain 20-100 components with different concentrations and some of it are the main compound with higher concentration (Perricone *et al.*, 2015; Bakkali *et al.*, 2008). The components are separated into several groups which are terpenes and terpenoid, aromatic and aliphatic

constituents, and phenylpropanoids (Bakkali *et al.*, 2008). The biological properties are determined by the main compound in the essential oils.

Ginger has been used worldwide as a cooking spice and home remedy for thousands of years. In traditional usage, ginger has been used to treat nausea, digestive aids, rheumatism, reduce cholesterol, and to fight arthritis. Ginger or also known as *Zingiber officinale* is belong to the Zingiberaceae family. The ginger family is cultivated in a tropical climate especially in Southeast Asia like Malaysia and Indonesia (Banerjee *et al.*, 2011). Ginger essential oils are extracted from its rhizomes. The ginger rhizome contains 60–70% carbohydrates, 9% protein, 8% ash, 3–8% crude fiber, 3–6% fatty oil and 2–3% volatile oil. The volatile oils that contain in ginger are only one to three percentage of its weight (Srinivasan, 2017). The main organic compound in ginger essential oil is α -zingiberene which represent the special aroma of ginger. The ginger essential oil can be extracted using various techniques such as steam distillation, microwave application methods and hydrodistillation (Kamalirroosta *et al.*, 2013). Studies has shown that ginger essential oils has broad range of pharmacological activities including anti-inflammatory, antioxidant, antibacterial, and insecticidal (Ficker *et al.*, 2003).

1.2 Problem Statement

Essential oils such as ginger essential oil contain a several of biological properties which include antioxidant, antimicrobial and anti-inflammatory. With these functional properties of ginger essential oil, it is attracted the pharmaceutical and health industries to use the essential oils to replace the used of synthetic drugs as their active pharmaceutical ingredient (API). However, it has some weakness which are highly volatile, low absorption, and it has poor water-solubility which the ability of essential oil constituents to move through water-based blood stream and cellular target is weak. A suitable colloidal delivery system need to be introduced to overcome the weaknesses. Transdermal drug delivery is one of the alternative to transport drug into the body. The drug is delivered through the skin at controlled rate. The use of transdermal drug delivery can prolonged therapeutic effect, improved bioavailability,

and reduced side-effects. Therefore, transdermal nanocarrier such as nanoemulsion can be used to transport the ginger essential oil because it can reduce volatility, improve the absorption mechanism and improve the bioefficacy of ginger essential oil towards the targeting tissue. Nanoemulsions is a suitable drug delivery system because of their unique properties such as small droplet size, high physical stability, high bioavailability and optical transparency compared to other conventional emulsions. The droplet size of is very important property because it determine the stability, bioavailability, and transparency of the nanoemulsions. In transdermal drug delivery, the molecular size is important because the smaller the size, the easier the drug to penetrate into the skin. Therefore, a study about the factors affecting particle size of ginger essential oil nanoemulsions are proposed.

1.3 Objectives

The main objectives of this research are:

1. To identify the factors for particle size optimization of ginger essential oil nanoemulsions.
2. To evaluate the thermodynamic and storage stability of ginger essential oil nanoemulsions.

1.4 Scopes of Study

1.4.1 The Emulsion Preparation Using Spontaneous Emulsification

Method

The formulation of nanoemulsions will be prepared by using spontaneous emulsification method. In this method, an organic phase (containing ginger essential oil and surfactant) are poured into an aqueous phase (distilled water) while

magnetically stirred. The nanoemulsions will be stirred at 500 rpm and the temperature for the organic phase and aqueous phase is at room temperature of 25 °C.

1.4.2 The Factors for Particle Size Optimization of Nanoemulsions

The nanoemulsions will be formulate by manipulating the surfactant type's usage, the concentration of surfactant, temperature of organic phase, stirring speed, and addition of co-solvent. For manipulating the surfactant type, the surfactants that will be used are Tween 40, Tween 60 and Tween 80. Concentration of surfactant will be manipulated by varying the surfactant-to-emulsion (SER) percentage from 10 % to 30 % while maintaining the oil content. The temperature of organic phase will be manipulated by varying the temperature from 0 to 90 °C and maintaining the temperature of aqueous phase. For stirring speed, it will be manipulated by changing the speed at three different speed which are 200 rpm, 500 rpm and 800 rpm. Lastly, for addition of co-solvent, the co-solvent that will be used is glycerol. The co-solvent will be added to the aqueous phase. The addition of co-solvent will be manipulated by varying the concentration of co-solvent from 0 % to 50 %.

1.4.3 The Stability of Ginger Essential Oil Nanoemulsions

The stability of ginger essential oil nanoemulsions will be examine using two parameters which are thermodynamic stability and storage stability. Thermodynamic stability will be analysed using heating-cooling cycle method and Freeze-Thaw stress method. For Heating-Cooling cycle, the nanoemulsions will be keep at 40 °C and 4 °C alternatively each temperature for 48 hours. For Freeze-Thaw stress, the nanoemulsions were kept at -21 °C and 25 °C for 48 hours for each temperature. For storage stability, the ginger essential oil nanoemulsions will be analysed by storing the nanoemulsions for two month and the droplet size of the nanoemulsions will be taken at day one, first month and second month to see any changes occur during the storage time. The changes in droplet size of the nanoemulsions will determine the stability of the nanoemulsions.

1.5 Significance of Study

Ginger essential oil has a broad biological properties such as anti-inflammatory, antioxidant and antiviral but it is highly volatile, poor water-solubility, and low absorption. The ginger essential oil can be transport into the body using transdermal drug delivery because it prolonged therapeutic effect, improved bioavailability, and reduced side-effects. To overcome this disadvantages, a colloidal transdermal drug delivery system which is nanoemulsions is introduce. Nanoemulsions can reduce the volatility, improve absorption and improve the efficacy of ginger essential oil. Droplet particle size is one of the functional properties of nanoemulsions which is very important because it decide the stability, efficacy, bioavailability, and appearance of emulsions. In transdermal drug delivery, the molecular size determine the penetration rate into the skin. The smaller the molecular size, the easier the drug to penetrate. Therefore, determining the factors that effecting particle size of nanoemulsions is perform.

1.6 Thesis Outline

There are five chapters in this thesis that covered a detailed explanation of the research study. The first chapter in this thesis are covering the introduction of study, problem statement, objectives, scopes, significance of study and also the outline of the thesis. The problem background discussed a problem facing by the essential oil and the importance of particle size nanoemulsions. In Chapter 2, more detailed in literature review on nanotechnology in drug delivery, essential oil industry and current researches that are being studied to search solutions for the problem facing. Chapter 2 also reviewed the details about properties, structures and characteristics of nanoemulsions.

Chapter 3 reported the chemicals used in this study and also the experimental methods that were applied in this study. Chapter 4 was discussed the experimental results and findings which were achieved according to the objectives and scopes that were designed for this research. Finally Chapter 5 presented the appropriate

conclusion for the findings of this research. A few recommendations were suggested to improve the stability of essential oil nanoemulsions. Lastly, the future studies for the next researchers were also discussed.

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