# THE PERFORMANCE OF PROGRESSIVE FREEZE CONCENTRATION FOR WATER PURIFICATION USING ROTATING CYLINDRICAL CRYSTALLISER WITH ANTI-SUPERCOOLING HOLES

### FARAH HANIM BINTI AB.HAMID

A thesis submitted in fulfilment of the requirements for the award of the degree of Doctor of Philosophy (Chemical Engineering)

Faculty of Chemical and Energy Engineering Universiti Teknologi Malaysia

MAY 2016

### **DEDICATION**

Specially dedicated to my best partner, my wonderful husband, Amiro Iqbal Mohd Aminudin, my beautiful and fabulous mother, Rohana Sulaiman, forever in memory, my adored late father, Ab.Hamid Mohamed, my baby boo, Aakif Uthman Amiro Iqbal my beloved siblings, my cute overloaded nephews and nieces, and not to forget my family in laws. Thanks for your continuous love, support and prayers.

#### ACKNOWLEDGEMENT

Alhamdulillah all praise to ALLAH SWT for His blessing for health and opportunity to complete this thesis for my PhD. This thesis could not have been completed without the assistance and support provided by many people. There are no proper words to convey my deep gratitude and respect for my thesis and research supervisor, Dr. Mazura Jusoh. This work would not have been possible without her guidance, support and encouragement. Under her guidance I successfully overcame many difficulties and learned a lot. And not to forget, my co-supervisor Associate Proffesor Adnan Ripin.

Special thanks to my wonderful husband, Amiro Iqbal Mohd Aminudin for always being there through my ups and downs, for always being there to cheer me up and standing by me through the good times and bad and for always being my biggest supporter. Not to forget, my beloved and beautiful mom, Rohana Sulaiman. You are the reason for what I have been now. My sibling and family in laws, thank you for your continuous supports and prayers. I love you all from the bottom of my heart.

During this work, I have collaborated with many colleagues for whom I have great regards, and I wish to extend my warmest thanks to all those who have helped me with my work especially to my research group members, Nor Zanariah Safiei, Nurul Aini Amran, Shafirah Samsuri, Noor Naimah Mohamed Noor and Norshafika Yahya. Thanks for yourhelp and ideas contribution thoughout the study. I would also like to thank other staffs in UTM who have helped me in one way or another.

### ABSTRACT

Progressive freeze concentration (PFC) is a freeze concentration method which forms ice crystals as a layer or a block on a cooled surface, which can be applied in purification of seawater to obtain pure water in ice form and leave the impurity behind. The aim of this study is to design a rotating cylindrical crystalliser with anti-supercooling holes in order to prevent supercooling phenomenon as well as to improve the solution movement in the crystalliser to increase productivity of the PFC system. Prevention of supercooling is important as it will affect the purity of ice produced. The performance analysis was carried out by using saline solution as simulated seawater, and was evaluated by the value of effective partition constant (K), desalination rate ( $R_d$ ), efficiency (E %) and solute recovery ( $R_s$ ). The system performance was found to be at its best at 300 rpm of rotation speed, four hours of rotation time, coolant temperature at -8 °C and 35 g/L of initial concentration with K value, E%, R<sub>d</sub> and R<sub>s</sub> of 0.376, 62.37%, 35.71% and 0.672, respectively. Low K value, high E%, high  $R_d$  and high  $R_s$  represent the best performance due to higher purity of ice crystal produced. Response surface methodology (RSM) in STATISTICA software was employed for optimisation process to obtain the optimum conditions in producing the best K value and  $R_d$ . Due to the limitation of optimization process by RSM offered by STATISTICA, a hybrid Artificial Neural Network and Genetic Algorithm in MATLAB was implemented for multiple response optimisations, where the best K value and  $R_d$  predicted were 0.26 and 49%, respectively. A mathematical heat transfer model in predicting ice crystal growth at different coolant temperature was successfully developed with an Average Absolute Relative Deviation of 5.56% and  $R^2$  value of 0.873. This newly designed crystalliser was found to be capable of producing ice crystals with high efficiency and productivity.

### ABSTRAK

Pemekatan pembekuan progresif (PFC) adalah satu kaedah pemekatan pembekuan yang membentuk kristal ais sebagai satu lapisan atau bongkah di atas permukaan sejuk, yang mana telah digunakan dalam penulenan air laut bagi mendapatkan air tulen dalam bentuk ais dan menyisihkan bendasing. Tujuan kajian ini adalah untuk membangunkan satu silinder penghablur berputar bersama lubang anti-penyejukan lampau bagi menangani fenomena penyejukan lampau dan juga memperbaiki pergerakan larutan di dalam penghablur berkenaan untuk meningkatkan produktiviti sistem PFC. Pencegahan masalah penyejukan lampau adalah penting kerana ia akan menjejaskan ketulenan ais yang terhasil. Analisa prestasi telah dijalankan dengan menggunakan larutan garam sebagai air laut simulasi, dan telah dinilai oleh nilai pemalar pemisahan berkesan (K), kadar penyahgaraman ( $R_d$ ), kecekapan (E%) dan dapatan bahan larut ( $R_s$ ). Prestasi sistem yang terbaik ditemui pada kelajuan putaran 300 rpm, masa pusingan selama empat jam, suhu penyejuk pada -8 °C dan kepekatan larutan awal pada 35 g/L dengan nilai K, E%, R<sub>d</sub> dan R<sub>s</sub> masing-masing adalah 0.376, 62.37%, 35.71% dan 0.672. Nilai K yang rendah, E% yang tinggi,  $R_d$  yang tinggi dan  $R_s$  yang tinggi mewakili prestasi terbaik kerana ketulenan kristal ais yang dihasilkan adalah lebih tinggi. Kaedah gerak balas permukaan (RSM) dalam perisian STATISTICA telah digunakan untuk proses pengoptimuman bagi mendapatkan keadaan optimum dalam menghasilkan nilai K dan  $R_d$  yang terbaik. Disebabkan oleh keterbatasan proses pengoptimuman oleh RSM yang dapat diberikan oleh STATISTICA, satu gabungan Rangkaian Neural Tiruan dan Algoritma Genetik dalam MATLAB telah dilaksanakan untuk pengoptimuman beberapa tindak balas untuk mendapatkan keadaan optimum, di mana nilai K dan  $R_d$ terbaik yang diramalkan masing-masing adalah 0.26 dan 49%. Satu model matematik pemindahan haba dalam meramalkan kadar pertumbuhan kristal ais pada suhu penyejuk yang berbeza telah berjaya dibangunkan dengan Purata Sisihan Mutlak Relatif adalah 5.56% dan  $R^2$  adalah 0.873. Penghablur dengan reka bentuk baru berkebolehan menghasilkan kristal ais yang mempunyai keberkesanan dan produktiviti yang tinggi.

# TABLE OF CONTENTS

CHAPTER	TITLE	PAGE
	DEDICATION	iii
	ACKNOWLEDGEMENT	iv
	ABSTRACT	v
	ABSTRAK	vi
	TABLE OF CONTENTS	vii
	LIST OF TABLES	xiii
	LIST OF FIGURES	XV
	LIST OF SYMBOLS	XX
	LIST OF ABBREVIATIONS	xxiii
	LIST OF APPENDICES	xxiv
1	INTRODUCTION	1
	1.1 Research Background	1
	1.2 Problem Statement	3
	1.3 Objective of Study	4
	1.4 Scope of Study	4
	1.5 Significance of Study	5
	1.6 Thesis Outline	6
2	LITERATURE REVIEW	9
	2.1 Introduction	9
	2.2 Methods of Concentrating Solution	9

	2.2.1	Evaporation process	10
	2.2.2	Reverse osmosis	11
	2.2.3	Freeze Concentration	13
2.3	Catego	ories of Freeze Concentration	19
	2.3.1	Suspension Freeze Concentration (SFC)	19
	2.3.2	Progressive Freeze Concentration (PFC)	22
2.4	Ice Cr	ystallisation Process	23
	2.4.1	Ice Nucleation	24
	2.4.2	Ice Crystal Growth	26
2.5	Progre	ssive Freeze Concentration Design	26
	2.5.1	Vertical Vessel with Cooling Plate Holes	27
	2.5.2	Tube Ice with Bubble Flow Circulator	28
	2.5.3	Ultrasonic Irradiation	30
	2.5.4	Square Pillar Shape	32
	2.5.5	A Heidolph Rotating Evaporator with Balloon Flask	34
	2.5.6	Radial Freezing with Stirring Effects	35
	2.5.7	Tubular Ice System	37
	2.5.8	Evaporator plates	38
	2.5.9	A dynamic layer melt crystallisation	40
	2.5.10	Shaker type freeze concentrator	42
	2.5.11	Multi Plate Freeze Concentrator	44
	2.5.12	Spiral Finned Crystalliser	45
2.6	The Pl	nenomenon of Supercooling	52
2.7	Detern	ninant Parameters of Progressive Freeze	
	Conce	ntration	55
	2.7.1	Effective Partition Constant (K)	55
	2.7.2	Ice Purity	56
	2.7.3	Desalination Rate	57
	2.7.4	Efficiency of Concentration	57
	2.7.5	Solute Yield	58
	2.7.6	Water Recovery	59
2.8	Metho	d of Controlling Progressive Freeze Concentration	59
	2.8.1	Effect of Flow Rate	60
	2.8.2	Effect of Initial Concentration	60

	2.8.3	Effect of	of Operation Time	61
	2.8.4	Effect of	of Coolant Temperature	61
	2.8.5	Effect of	of Advance Speed of Ice Front	62
2.9	Desig	n of Expe	eriment (DOE) and Optimisation	63
	2.9.1	Respon	se Surface Methodology (RSM)	63
	2.9.2	Central	Composite Design (CCD)	65
	2.9.3	Analysi	s of Variance (ANOVA)	66
2.10	Multi	ple Respo	onse Optimisations via ANN	67
2.11	Fluid	Mechani	с	69
2.12	Frictio	on study		70
2.13	Heat 7	Fransfer I	Model in Freeze Concentration	72
2.14	Sumn	nary of Li	iterature Review and Research Gap	77
ME	THOD	OLOGY		79
3.1	Introd	luction		79
3.2	Desig	n of Crys	talliser	81
	3.2.1	Prelimi	nary Design	81
	3.2.2	Actual	design of rotating crystalliser	86
	3.2.3	Constru	action material	87
	3.2.4	Anti-su	percooling Holes	88
	3.2.5	Solution	n Movement	90
		3.2.5.1	Introduction of Motor	90
		3.2.5.2	A Set of Baffle	91
	3.2.6	Ice visi	bility and Sampling Process	92
	3.2.7	Temper	ature Profiling Features	93
	3.2.8	Geomet	trical Analysis	94
		3.2.8.1	Fluid Mechanics Analysis	94
		3.2.8.2	Friction Study	95
		3.2.8.3	Holes Test	95
3.3	Mater	ial		96
3.4	Labor	atory Equ	uipment	97
3.5	Exper	imental S	Setup	99
3.6	Adapt	ation for	Real Application	100
3.7	Exper	imental F	Procedure	101

3

3.8	Investigations on the Effects of the Operating Condit	ions 102
3.9	Evaluation Method	104
	3.9.1 Effective Partition Constant	104
	3.9.2 Desalination Rate	104
	3.9.3 Efficiency of Concentration	105
	3.9.4 Solute Recovery	105
	3.9.5 Water Recovery	106
3.10	Mass Balance Calculation for Salt Content	106
3.11	Optimisation	108
	3.11.1 Variable selection	109
	3.11.2 Design of Experiment	109
	3.11.3 Evaluation of the Fitted Model	110
	3.11.4 Determination of the Optimal Conditions	113
3.12	Multiple Response Optimizations	113
3.13	Development of Heat Transfer Model	115
PER	FORMANCE OF ROTATING CRYSTALLISER	
IN W	VATER PURIFICATION PROCESS	117
4.1	Introduction	117
4.2	Geometrical Analysis	117
	4.2.1 Holes Test	118
	4.2.2 Fluid Mechanics Analysis	119
	4.2.3 Friction Study	120
4.3	Efficiency of PFC in Water Purification Process	122
	4.3.1 Effect of Rotation Time	123
	4.3.2 Coolant Temperature	127
	4.3.3 Rotation Speed of the Crystalliser	132
	4.3.4 Initial Concentration	135
4.4	Mass Balance of Salt Content	138
ОРТ	TIMISATION OF PARAMETERS FOR PFC	138
5.1	Introduction	138
5.2	Experimental Design	141
5.3	Optimisation of the Effective Partition Constant (K)	141

	5.3.1	Model	Adequacy Check	143
	5.3.2	Respon	se Surface Contour Plots Analysis	149
		5.3.2.1	Interaction Effects of Rotation Time $(X_1)$	
			and Coolant Temperature (X <sub>2</sub> )	149
		5.3.2.2	Interaction Effects of Rotation time (X <sub>1</sub> )	
			and Rotation Speed (X <sub>3</sub> )	151
		5.3.2.3	Interaction Effects of Rotation Time $(X_1)$	
			and Initial Concentration (X <sub>4</sub> )	153
		5.3.2.4	Interaction Effects of Coolant Temperature	re
			$(X_2)$ and Rotation Speed $(X_3)$	155
		5.3.2.5	Interaction Effect of Coolant Temperature	ļ.
			(X <sub>2</sub> ) and Initial Concentration (X <sub>4</sub> )	156
		5.3.2.6	Interaction Effect of Rotation Speed (X <sub>3</sub> )	
			and Initial Concentration (X <sub>4</sub> )	158
	5.3.3	Optimu	m Operating Condition	160
5.4	Optim	nisation o	f the Desalination Rate (R <sub>d</sub> )	161
	5.4.1	Model	Adequacy Checking	162
	5.4.2	Respon	se Surface Contour Plots Analysis	168
	5.4.3	Optimu	m Operating Condition	176
5.5	Multi	ple Respo	onse Optimisation	176
HE	AT TRA	ANSFER	MODEL IN PREDICTING	
ICE	CRYS	TAL GR	OWTH	182
6.1	Introd	uction		182
6.2	Temp	erature P	rofiling	183
6.3	Mode	l Develoj	oment	184
	6.3.1	Fundan	nental of Heat Transfer Model	184
	6.3.2	Rationa	le of the model	187
	6.3.3	Convec	tive Heat Transfer Coefficient	188
	6.3.4	Heat Ti	ansfer Area	190
	6.3.5	Develo	pment of Final Heat Transfer Model	191
6.4	Predic	ction of I	ce Crystal Growth	193
6.5	Mode	l Validat	ion	195

6

7 CO		CONCLUSIONS AND RECOMMENDATIONS		
	7.1	Conclusions	198	
	7.2	Recommendations for Future Research	201	
REFERENCE	ËS		203	
Appendices A-	·F		216-238	

# LIST OF TABLES

TABLE NO.	TITLE	PAGE
2.1	The summary of advantages and disadvantages for	
	concentrating solution technologies	17
2.2	The summary for previous design	48
3.1	Summary of rotating crystalliser dimension	86
3.2	Thermal conductivity for different material	88
3.3	Freezing point of ethylene glycol at different volume	
	percentage	96
3.4	Specification for salinometer	97
35	Specification for cooling bath	98
3.6	Specification for PicoLog recorder.	98
3.7	Value of varied and constant parameter	103
3.8	Experimental range and level coded.	109
3.9	Experiment combination using CCD.	111
3.10	ANOVA for fitted mathematical model to an experimental	
	data set	112
3.11	Lower and upper bounds for GA input variables.	114
4.1	Effect of holes number towards the system performance	118
4.2	Reynolds number for varied rotation speed.	120
4.3	Friction factor and friction loss	121
4.4	Summary of mass balance calculation (salt content)	139
5.1	The experimental K value for each run.	142
5.2	Observed value and predicted value of K.	143
5.3	ANOVA table for the model of K value.	146
5.4	Regression analysis for K.	147

Summary of optimum range of condition.	160
The experimental $R_d$ values	162
Observed value and predicted value of $R_d$	163
ANOVA analysis on $R_d$ value	166
Regression analysis for $R_d$ .	167
Listed model for effect and the interaction between two	
variables.	168
Summary of optimum range of condition	176
Optimum condition for both responses	177
Process variables corresponding to each Pareto-optimal	
solutions.	180
Optimal conditions of K value and $R_d$	180
Convective heat transfer coefficient for seawater solution	193
Convective heat transfer coefficient for ethylene glycol	194
Calculation of ice crystal growth rate	194
Comparison between predicted value and experimental	
value of ice crystal growth, dm/dt.	196
	The experimental $R_d$ values Observed value and predicted value of $R_d$ ANOVA analysis on $R_d$ value Regression analysis for $R_d$ . Listed model for effect and the interaction between two variables. Summary of optimum range of condition Optimum condition for both responses Process variables corresponding to each Pareto-optimal solutions. Optimal conditions of K value and $R_d$ Convective heat transfer coefficient for seawater solution Convective heat transfer coefficient for ethylene glycol Calculation of ice crystal growth rate Comparison between predicted value and experimental

# LIST OF FIGURES

FIGURE NO.	TITLE	PAGE
2.1	Schematic of multi-stage flash desalination process	11
2.2	Schematic of reverse osmosis process	12
2.3	Two methods for freeze concentration	19
2.4	Schematic suspension freeze concentration system	21
2.5	Freeze concentration set-up	23
2.6	Various kinds of nucleation	25
2.7	A vertical cylindrical vessel with cooling plate holes	28
2.8	A structure of the cooling plate holes	28
2.9	Schematic diagram for experimental apparatus by using	
	tube ice with bubble flow circulator	29
2.10	Experimental apparatus of freeze concentration with	
	ultrasonic irradiation	31
2.11	Schematic system flow diagram of freeze wastewater	
	treatment	32
2.12	Schematic model of ice maker structure: agitation (left),	
	aeration (right)	33
2.13	Laboratory scale setup	35
2.14	Radial freezing device	36
2.15	Tubular ice systems for scale up of PFC	38
2.16	Schematic layout for evaporator plates	39
2.17	A dynamic layer melt crystallisation set-up	41
2.18	Shaker typed of freeze concentrator	43
2.19	Schematic layout of multiplate freeze concentrator	45
2.20	Experimental setup for spiral finned crystalliser	46
2.21	Schematic diagram of cooling curve	53

2.22	Cooling curves for the solution and the cooling surface with $(A)$ and without $(B)$ halos in the cooling plate	51
2.22	with (A) and without (B) holes in the cooling plate	54
2.23	Some profiles of surface response generated from a	
	quadratic model in the optimization of two variables. (a)	
	maximum, (b) plateau, (c) maximum outside	
	theexperimental region, (d) minimum, and (e) saddle surfaces	65
2.24	Central composite designs for the optimization of: (a) two	
	variables and (b) three variables	66
2.25	Segment heat and mass transfer	73
2.26	Temperature profile	74
3.1	Flow chart of research methodology.	80
3.2	Explode view of rotating crystalliser	82
3.3	Main body of rotating crystalliser	83
3.4	The combination of lid and gear	83
3.5	Illustration of rotating bearing	84
3.6	Illustration of baffle	84
3.7	The illustration of crystalliser dimension	85
3.8	A complete set of rotating crystalliser	87
3.9	Close up of anti-supercooling holes	89
3.10	Mechanism of ice nucleation	90
3.11	An illustration of rotation direction	91
3.12	A real set of baffle	92
3.13	The lid equipped with the gear	93
3.14	A thermocouple path (top view of the lid)	94
3.15	Experimental setup	99
3.16	Actual experimental setup	100
3.17	Illustration of adaptation for real application	101
3.18	Flow diagram of mass balance calculation for salt content	107
3.19	Flow chart of STATISTICA steps	108
4.1	A closed up of ice layer formed	122
4.2	Effect of rotation time towards K value, R <sub>s</sub> and E	124
4.3	Desalination rate, water recovery and ice salinity versus	

	rotation time	126
4.4	Ice thickness versus rotation time	126
4.5	Relationship between coolant temperature and K value,	
	R <sub>s</sub> and E %	128
4.6	Temperature profile	129
4.7	Desalination rate, water recovery and ice salinity versus	
	coolant temperature	131
4.8	Effect of rotation speed on K value, $R_s$ and E %	133
4.9	Desalination rate, water recovery and ice salinity versus	134
	coolant temperature	
4.10	Effect of initial concentration on K value, $R_s$ and E %	136
4.11	Effect of initial concentration towards desalination rate	
	$(\mathbf{R}_d)$ , water recovery, and ice salinity.	137
5.1	Observed versus predicted values of the response K	144
5.2	Normal probability plots of residuals	145
5.3	Predicted values against raw residuals	146
5.4	Standardize Pareto chart of variable effects on K value	149
5.5	A surface plot of rotation time and coolant temperature	
	towards K value	150
5.6	A contour plot of K as a function of rotation time and	
	coolant temperature	151
5.7	An interaction between rotation time and rotation speed	
	towards K value	152
5.8	A contour plot of K as a function of rotation time and	
	rotation speed	153
5.9	Surface plot of K value against rotation time and initial	
	concentration	154
5.10	A contour plot of rotation time against initial	
	concentration	154
5.11	Surface effect of coolant temperature versus rotation	155
	speed	
5.12	A contour plot of coolant temperature against rotation	
	speed	156

5.13	Surface plot of K as a function of coolant temperature and	
	initial concentration	157
5.14	Contour plot of coolant temperature against initial	
	concentration	158
5.15	Surface plot of K value against rotation speed and initial	
	concentration	159
5.16	Contour plot of rotation speed against initial concentration	160
5.17	Experimental versus predicted values of the response	164
5.18	Normal probability plots of residuals	164
5.19	Plot of residuals versus fitted values	165
5.20	Standardize Pareto chart of variable effects on $R_d$ value	167
5.21	Effect of rotation time and coolant temperature towards $R_d$	
	value	169
5.22	Effect of rotation time and rotation speed towards $R_{\rm d}$	
	value	169
5.23	Effect of rotation time and initial concentration towards	
	R <sub>d</sub> value	170
5.24	Effect of coolant temperature and rotation speed towards	
	R <sub>d</sub> value	170
5.25	Effect of coolant temperature and initial concentration	
	towards R <sub>d</sub> value	171
5.26	Effect of rotation speed and initial concentration towards	
	R <sub>d</sub> value	171
5.27	Contour plot rotation time against coolant temperature	173
5.28	Contour plot rotation time against rotation speed	173
5.29	Contour plot rotation time against initial concentration	174
5.30	Contour plot coolant temperature against rotation speed	174
5.31	Contour plot coolant temperature against initial	
	concentration	175
5.32	Contour plot rotation speed against initial concentration	175
5.33	Experimental data against predicted data for K value	178
5.34	Experimental data against predicted data for $R_d$ value	178
5.35	Pareto optimal set of solution of K value and $R_d$	179

6.1	Snapshot of temperature profile for solution temperature	183
6.2	Schematic diagram of heat transfer in PFC	184
6.3	Graph of ice thickness against coolant temperature	192
6.4	Comparison between predicted value and experimental	
	value	196

### LIST OF SYMBOLS

Κ effective partition constant \_  $\mathbf{R}_d$ desalination rate \_ Y solute yield -E % efficiency of concentration -Uo overall heat transfer coefficient -CL solute of concentrations in the solution phases concentration of ice  $\mathbf{C}_i$ solute concentrations in the ice  $C_{S}$ - $C_{o}$ the solute concentration at the beginning in the solution phase -V<sub>L</sub> concentrate volume -Vo initial volume of the solution mass in the concentrated  $m_s$ mass in the initial solution  $m_{\rm o}$ solute mass in the liquid phase m<sub>s liq</sub> initial solute mass  $m_{s0}$ -Q amount of heat transferred -Α area for heat transfer - $\Delta T_{min}$  effective temperature difference R total resistance inside surface area of tube Ai -

A <sub>o</sub>	-	outside surface area of tube
$\mathbf{h}_{\mathbf{i}}$	-	heat transfer coefficient for solution
$h_{o}$	-	heat transfer coefficient for coolant
Х	-	thickness of ice layer
Κ	-	thermal conductivity
$A_{m}$	-	logarithmic mean area
Ko	-	limiting partition coefficient
Y	-	kinematic viscosity
μ	-	dynamic viscosity
Р	-	density of fluids
$R_e$	-	Reynolds number
V	-	velocity
D	-	internal diameter of pipe
$h_{f}$	-	head loss
F	-	appropriate dimensionless factor
G	-	acceleration of gravity
L	-	length of straight pipe
R <sub>s</sub>	-	solute recovery
So	-	initial salinity of the solution
$\mathbf{S}_{\mathbf{i}}$	-	salinity of ice produced
$\mathbf{S}_l$	-	concentration of NaCl in the concentrated solution
$T_i$	-	temperature of the ice layer interface
To	-	temperature of inside wall
$T_{\rm w}$	-	temperature of outside wall
$\lambda_i$	-	heat conductivity coefficient of the ice
δ	-	thickness of the wall
$\lambda_{\mathrm{w}}$	-	heat conductivity coefficient of the wall
L	-	latent heat of fusion, J/kg

- $dm_i/dt$  ice growth rate
- $\alpha_s$  heat transfer coefficient of seawater solution
- $N_{Nu}\$   $\ \ Nusselt$  number
- $N_{Pr}$  Prandtl number
- $\mu_b$  fluid viscosity at bulk average temperature
- $\mu_w$  viscosity at wall temperature

## LIST OF ABBREVIATIONS

- RO reverse osmosis
- FC freeze concentration
- SFC suspension freeze concentration
- PFC progressive freeze concentration
- COD Chemical oxygen demand
- RSM Response Surface Methodology
- ANN Artificial Neural Network
- MSF multi stage flash
- HTS heat transfer surface
- FBHE fluidized bed heat exchanger
- SSHE scraped surface heat exchanger
- STC stirred tank crystalliser
- ANOVA- analysis of variance
- MOO multi objective optimization
- GA genetic algorithm
- CCD central composite design

# LIST OF APPENDICES

APPENDIX	TITLE	PAGE
A1	Moody chart	217
B1	Overall results of efficiency study	219
C1	F Test table	221
C2	Center composite design with observed and predicted	
	response	222
D1	Density of seawater	224
D2	Kinematic viscosity of seawater	225
D3	Dynamic viscosity of seawater	226
D4	Thermal conductivity of seawater	227
D5	Specific heat of seawater	228
D6	Prandt number of seawater	229
<b>E1</b>	Density of ethylene glycol	231
E2	Thermal conductivity of ethylene glycol	232
E3	Specific thermal capacity of ethylene glycol	233
E4	Dynamic viscosity of ethylene glycol	234
E5	Prandtl number of ethylene glycol	235
F1	Convective heat transfer coefficient for seawater solution	237
F2	Convective heat transfer coefficient for ethylene glycol	237
F3	Data summary for heat transfer model	238

### **CHAPTER 1**

### **INTRODUCTION**

### 1.1 Research Background

Water purification is categorized as an extremely important process to ensure that the water is safe to use and free from any contaminants that can give bad effect to health. Water purification can be achieved through concentration method. Several concentration methods have been developed since the last few decades in order to meet the growing demand in the industry. Among these, three methods have found their commercial applications which are evaporation, reverse osmosis (RO) and freeze concentration (FC).

In evaporation method, due to high latent heat, a vast amount of fuel and thermal energy is consumed in order to evaporate water. As a result, this process has higher operating costs and environmental problems. Meanwhile, in RO process, the membranes are sensitive to pH, oxidizers, a wide range of organics, algae, bacteria and particulates (Miller, 2003). As a consequence, the cost of RO is high due to the maintenances of membrane. In addition, the osmotic pressure applied in this method also leads to high energy consumption which contributes to operation costs increment (Rodriguez *et al.*, 2000).

However, among all the technologies introduced before, freeze concentration is thought to present a splendid opportunity to be applied because in terms of energy, it is more favourable than evaporation (Englezos, 1994). Freeze concentration is one of the methods to concentrate solution by producing pure ice in the solution. It crystallises out the water from liquid solution in the form of ice crystals and leaves a highly concentrated solution behind (Okawa *et al.*, 2009). According to Melak *et al.* (2016), freeze desalination has a lower energy requirement compared to other thermal processes. Moreover, the latent heat of water solidification (334 J/g) is much lower than heat of vaporization (2257 J/g) (Nakagawa *et al.*, 2010), where the quantity of latent heat of solidification is about 1/7 of the latent heat of vaporization in the same condition (Gao *et al.*, 2014). Due to a low temperature operation, where corrosion is not significant, the freeze concentration process has lower production costs, requires less maintenance and with less precipitation (Luo *et.al.*, 2010). There are two methods in freeze concentration, which are suspension freeze concentration (SFC) and progressive freeze concentration (PFC).

SFC is a method to produce small ice particles. Eventhough it achieves high efficiency of concentration, but the ice particles formed in this method are difficult to be separated from the concentrated solution (Okawa *et al.*, 2009). In addition, the size of ice crystal is still limited in this conventional method (Gu *et al.*, 2005). This process is unfavourable because it needs many equipments in order to complete the process, involving separation of ice and mother liquor. The steps involved make the whole process very expensive.

PFC has been introduced in order to amend the weaknesses of SFC. In this process, the ice will be formed as a large single ice crystal instead of small particles. The ice crystal will be formed on the surface of the heat conducting material where the cooling is supplied. The separation of ice crystal from its mother liquor is much easier to be handled compared to SFC as only a single crystal is formed. Due to this reason, this process requires a lower cost since it needs no series of equipments. However, despite of it being easy to handle, the productivity of PFC is discovered to be much lower than the conventional SFC.

Therefore, there are several designs of the crystalliser that have been developed in order to improve the productivity of freeze concentration system and in the meantime, to reduce the high investment costs. Several designs have been introduced, designed, constructed and operated under different conditions in order to amend the weaknesses and reduce the cost from previous design. Tubular ice system was introduced in 2005 by Miyawaki *et al.* (2005) while cylindrical vessel in vertical was introduced by Ramos *et al.* (2005) which is altering the previous design from Liu *et al.* (1998). From time to time, the crystalliser design has been developed in order to adapt with the demand of quality and quantity of the product.

#### **1.2 Problem Statement**

In PFC, solution movement is very important because it will affect the product quality and productivity. The problem of low productivity is basically due to low distribution of heat from the solution to the coolant which is essentially influenced by solution movement. Therefore, solution movement plays a main role to provide optimum heat distribution. From previous studies, several problems of solution movement have been detected. For instance, a typical stirrer introduced by Liu et al. (1998) and Ramos et al. (2005) face the problem of topical movement, where the solution will only have a strong and fast circulation near the rotating blade. In other words, the movement is only focused at one place, resulting in uneven distribution of the solute, thus will affect the quality of ice produced. For falling film introduced by Sánchez et al. (2011) and Moreno et al. (2014a), the disadvantage of this type of solution movement is it is affected by ambient temperature. Meanwhile, for pumping type (Jusoh et al., 2009; Miyawaki et al., 2005), both designs face a problem of flow pattern since it has bends, which contribute to producing uneven flow. This phenomenon can give an affect to the consistency of ice purity produced. Therefore, it is important to come up with a new approach to provide solution movement.

Apart from the disadvantages of existing designs of PFC which include low productivity and poor ice quality, the phenomenon of supercooling is a serious problem that should be concerned on (Liu *et al.*, 1998). Ice seeding process needs to be done to address this problem if there is no special design feature introduced to the crystalliser. The main problem of the ice seeding process is it consumes longer operation time. In addition, with this extra requirement, the procedure will involve too many steps which will lead to the process becoming more complicated. Therefore, it is essential to develope a new design of crystalliser with a feature to overcome the supercooling problem.

### **1.3** Objective of Study

In order to complete this research successfully, several objectives have to be achieved. The objectives of this research are as follows:

- 1. To design a rotating cylindrical crystalliser with anti-supercooling holes as the main component in PFC system.
- To evaluate the performance of the newly designed crystalliser for PFC system at various operating conditions which are coolant temperature, rotation time, rotation speed and initial concentration, and to determine the optimum conditions.
- 3. To develop a mathematical heat transfer model on prediction of ice crystal growth.

### **1.4** Scope of Study

To achieve the objectives of this study, these following scopes were applied:

- A new crystalliser with cylindrical shape was designed taking into consideration the aspects of construction material, solution movement and design feature in preventing supercooling problem. The additional features for ice sampling and temperature profiling purpose were also introduced. The study of fluid mechanic in term of flow pattern and friction study of the PFC system was also investigated. In addition, the holes capability to freeze out water molecules was determined through some additional experiments. These experiments were carried out to examine the influence of holes numbers towards ice purity produced.
- 2. The efficiency of PFC system was determined based on the effect of operating parameters which are rotation speed of crystalliser, rotation time, initial concentration and coolant temperature. The range for rotation speed is 150 rpm to 350 rpm. Meanwhile the range for rotation time is an hour to five hours, initial concentration at 25g/L to 45g/L and coolant temperature at -11°C to -7°C. The performance of the system was measured by observing the changes in effective partition constant, (K), desalination rate (R<sub>d</sub>), efficiency of concentration (E%), water recovery and solute recovery (R<sub>s</sub>). Saline solution was used as target liquid. Optimization for this system was analysed by using Response Surface Methodology (RSM) via Statistica software and Artificial Neural Network (ANN)
- 3. A new model of heat transfer was developed in order to predict the growth of ice formed based on a modification of convective heat transfer coefficient according to the system condition.

#### **1.5** Significance of Study

In this research, this newly developed crystalliser design is believed to be able to improve the quality and productivity of the PFC system. The cylindrical shape is equipped with anti-supercooling holes to prevent supercooling phenomenon and improvement of the productivity could also be attained by introducing a new approach for solution movement. This research is able to provide the fundamental in designing a crystalliser with high performance and better quality and productivity. In addition, this crystalliser is focused to be applied for water purification process which could be an important new source of water supply, since water demand is dramatically increasing over the world recently in areas with fresh water supply scarcity. High quality water supply is expected to be produced, which is safe and compatible with the standard of World Health Organisation. Furthermore, this research can also be a reference for the industries since PFC can also be applied in other industrial processes such as food concentration, wastewater treatment and pharmaceutical.

### 1.6 Thesis Outline

This thesis is written in 7 chapters and the chronology for each chapter is briefly elucidated below.

The first chapter is Chapter 1 which presents the research background of research, the problem statement and the research objectives. The scopes of the study are also introduced as a guideline in order to achieve the objectives of the study.

This is followed by Chapter 2 where a review on previous studies on freeze concentration and specifically on progressive freeze concentration is elaborated. The previous technology on water purification is reviewed and the advantages and disadvantages for each method are listed. Furthermore, the design for PFC setup is also discovered where the improvement for each design can be seen from times to times. A brief review about optimisation process via Response Surface Methodology (RSM) and multi response optimisation via Artificial Neural Networks (ANN) is also presented in this chapter. Finally, a fundamental of heat transfer model in PFC is explicated at the end of this chapter.

Then, the procedures involved which starts with designing the crystalliser, experimental set up, experimental works including evaluation method are demonstrated in Chapter 3. A flow chart of research methodology is also provided in this chapter. Elaboration on the details in developing a crystalliser named rotating crystalliser with anti-supercooling holes is presented. This includes the procedures in material selection, determination of dimension, and introduction of special features. The procedure of optimisation process is elucidated in detail, followed by the development of heat transfer model in predicting ice crystal growth.

All the results and discussion on the geometrical analysis such as fluid mechanics, friction study and holes test and the effect of all the four parameters including rotation speed, coolant temperature, rotation time and initial concentration is covered in Chapter 4. Five determinant parameters were employed in order to determine the performance of the system which are effective partition constant (K), desalination rate ( $R_d$ ), solute recovery ( $R_s$ ), water recovery and concentration efficiency (E%).

Chapter 5 expounds the optimisation procedures in order to determine the optimum condition for the rotating crystalliser with anti-supercooling holes system by implementing Response Surface methodology via Statistica software. Furthermore, the optimum level of each variable and the effect of their interactions on the response which is K value and desalination rate ( $R_d$ ) as a function of two variables were studied by plotting three dimensional response surface curves known as contour plots. In addition, the contour plots were employed in aiding visualization of the interaction. However, a set of optimum condition is required to simultaneously obtain a low value of K and a high value of desalination rate (Rd). Therefore, a multiple objective optimisation was implemented in order to find out an experimental optimal point for both responses. For this reason, a hybrid of ANN-GA was used to achieve the optimisation objective.

The development of heat transfer model is explained in Chapter 6. A new model was developed in order to predict the ice crystal growth for the system. By considering the new system of rotating crystalliser with anti-supercooling holes, a

modification of convective heat transfer coefficient is done according to the condition of the new system such as type of flow and heat transfer area. A fundamental heat transfer model and several assumptions were employed in developing the new model. Furthermore, the validation of the model is also provided at the end of the chapter in order to determine the model appropriateness.

Finally, Chapter 7 presents the conclusion for each of the study objective. A brief summary for the results and discussions has been made and presented in this chapter. Furthermore, several suggestions and recommendations are advocated for future research study.

### REFERENCES

- Aguiar, I. B., Miranda, N. G. M., Gomes, F. S., Santos, M. C. S., Freitas, D. d. G. C., Tonon, R. V. and Cabral, L. M. C. (2012). Physicochemical and Sensory Properties of Apple Juice Concentrated by Reverse Osmosis and Osmotic Evaporation. *Innovative Food Science and Emerging Technologies*. (0).
- Aider, M. and De-Halleux, D. (2008). Passive and Microwave-Assisted Thawing in Maple Sap Cryoconcentration Technology. *Journal of Food Engineering*. 85, 65–72.
- Al-Subaie, K. Z. (2007). Precise Way to Select a Desalination Technology. Desalination. 206, 29-35.
- Alarcon-Rodriguez, A., Ault, G. and Galloway, S. (2010). Multi-Objective Planning of Distributed Energy Resources: A Review of the State-of-the-Art. *Renewable and Sustainable Energy Reviews*. 14 (5), 1353-1366.
- Altınkaya, H., Orak, İ. M. and Esen, İ. (2014). Artificial Neural Network Application for Modeling the Rail Rolling Process. *Expert Systems with Applications*. 41 (16), 7135-7146.
- Álvarez, S., Riera, F. A., Álvarez, R. and Coca, J. (1998). Permeation of Apple Aroma Compounds in Reverse Osmosis. *Separation and Purification Technology*. 14 (1–3), 209-220.
- Auleda, J. M., Raventós, M., Sánchez, J. and Hernández, E. (2011). Estimation of the Freezing Point of Concentrated Fruit Juices for Application in Freeze Concentration. *Journal of Food Engineering*. 105 (2), 289-294.
- Baños, R., Manzano-Agugliaro, F., Montoya, F. G., Gil, C., Alcayde, A. and Gómez, J. (2011). Optimization Methods Applied to Renewable and Sustainable Energy: A Review. *Renewable and Sustainable Energy Reviews*. 15 (4), 1753-1766.

- Baş, D. and Boyacı, İ. H. (2007). Modeling and Optimization I: Usability of Response Surface Methodology. *Journal of Food Engineering*. 78 (3), 836-845.
- Betiku, E. and Taiwo, A. E. (2015). Modeling and Optimization of Bioethanol Production from Breadfruit Starch Hydrolyzate Vis-À-Vis Response Surface Methodology and Artificial Neural Network. *Renewable Energy*. 74 (0), 87-94.
- Bezerra, M. A., Santelli, R. E., Oliveira, E. P., Villar, L. S. and Escaleira, L. A. (2008). Response Surface Methodology (Rsm) as a Tool for Optimization in Analytical Chemistry. *Talanta*. 76 (5), 965-977.
- Bhatnagar, B. S., Cardon, S., Pikal, M. J. and Bogner, R. H. (2005). Reliable Determination of Freeze-Concentration Using Dsc. *Thermochimica Acta*. 425 (1-2), 149-163.
- Bhatti, M. S., Kapoor, D., Kalia, R. K., Reddy, A. S. and Thukral, A. K. (2011). Rsm and Ann Modeling for Electrocoagulation of Copper from Simulated Wastewater: Multi Objective Optimization Using Genetic Algorithm Approach. *Desalination*. 274 (1–3), 74-80.
- Borissova, A. (2009). General Systems Modeling of Multi-Phase Batch Crystallization from Solution. *Chemical Engineering and Processing: Process Intensification*. 48 (1), 268-278.
- Cengel, Y. A. and Ghajar, A. J. (2010). *Heat and Mass Transfer: Fundamentals and Applications*. (Fourth ed.). New York, United States of America: McGraw-Hill.
- Chen, C. R. and Ramaswamy, H. S. (2002). Modeling and Optimization of Variable Retort Temperature (Vrt) Thermal Processing Using Coupled Neural Networks and Genetic Algorithms. *Journal of Food Engineering*. 53 (3), 209-220.
- Cheok, C. Y., Chin, N. L., Yusof, Y. A., Talib, R. A. and Law, C. L. (2012). Optimization of Total Phenolic Content Extracted from Garcinia Mangostana Linn. Hull Using Response Surface Methodology Versus Artificial Neural Network. *Industrial Crops and Products*. 40 (0), 247-253.
- Clifford, M. (2002). Section 5 Basic Fluid Mechanics Aeronautical Engineer's Data Book. (pp. 76-95). Oxford: Butterworth-Heinemann.

- Cook, D. F., Ragsdale, C. T. and Major, R. L. (2000). Combining a Neural Network with a Genetic Algorithm for Process Parameter Optimization. *Engineering Applications of Artificial Intelligence*. 13 (4), 391-396.
- Cook, K. L. K. and Hartel, R. W. (2010). Mechanisms of Ice Crystallization in Ice Cream Production. *Comprehensive Reviews in Food Science and Food Safety*. 9 (2), 213-222.
- Ellenberger, J. P. (2010). Chapter 4 Piping and Pipeline Sizing, Friction Losses, and Flow Calculations Piping and Pipeline Calculations Manual. (pp. 35-55). Boston: Butterworth-Heinemann.
- Englezos, P. (1993). The Freeze Concentration Process and Its Applications. Vancouver: Department of Chemical Engineering, University of Brifsh Columbia
- Englezos, P. (1994). The Freeze Concentration Process and Its Applications. Developments in Chemical Engineering and Mineral Processing. 2 (1), 3-15.
- Eren, İ. and Kaymak-Ertekin, F. (2007). Optimization of Osmotic Dehydration of Potato Using Response Surface Methodology. *Journal of Food Engineering*. 79 (1), 344-352.
- Erin, R. G. (2005). Factors Effecting the Optimization of Lipase-Catalyzed Palm-Based Esters Synthesis. PhD thesis, Universiti Putra Malaysia, Serdang.
- Flesland, O. (1995). Freeze Concentration by Layer Crystallization. Drying Technology. 13 (8-9), 1713-1739.
- Fujioka, R., Wang, L. P., Dodbiba, G. and Fujita, T. (2013). Application of Progressive Freeze-Concentration for Desalination. *Desalination*. 319 (0), 33-37.
- Gao, P., Guo, Z., Zhang, D., Zhou, X. and Zhou, G. (2014). Performance Analysis of Evaporation-Freezing Desalination System by Humidity Differences. *Desalination*. 347 (0), 215-223.
- Gao, W. and Shao, Y. (2009). Freeze Concentration for Removal of Pharmaceutically Active Compounds in Water. *Desalination*. 249 (1), 398-402.
- Gay, G., Lorain, O., Azouni, A. and Aurelle, Y. (2003). Wastewater Treatment by Radial Freezing with Stirring Effects. *Water Research*. 37 (10), 2520-2524.
- Geankoplis, C. J. (2003). *Transport Processes and Separation Process Principles*. (Fourth ed.). United States of America: Pearson Education, Inc.

- Ghiasi, M. M., Esmaeili-Jaghdan, Z., Halali, M. A., Lee, M., Abbas, A. and Bahadori, A. (2015). Development of Soft Computing Methods to Predict Moisture Content of Natural Gases. *Journal of the Taiwan Institute of Chemical Engineers*. 55, 36-41.
- Gosukonda, R., Mahapatra, A. K., Liu, X. and Kannan, G. (2015). Application of Artificial Neural Network to Predict Escherichia Coli O157:H7 Inactivation on Beef Surfaces. *Food Control.* 47 (0), 606-614.
- Goto, S., Yamamoto, Y., Sugi, T., Yasunaga, T., Ikegami, Y. and Nakamura, M. (2010). Construction of Simulation Model for Spray Flash Desalination System. *Electrical Engineering in Japan*. 170 (4).
- Gu, X., Suzuki, T. and Miyawaki, O. (2005). Limiting Partition Coefficient in Progressive Freeze-Concentration. *Journal of Food Science*. 70 (9), E546-E551.
- Gu, X., Watanabe, M., Suzuk, T. and Miyawaki, O. (2008). Limiting Partition Coefficient in a Tubular Ice System for Progressive Freeze-Concentration. *Food Science and Technology*. 14, 249-252.
- Guan, X. and Yao, H. (2008). Optimization of Viscozyme L-Assisted Extraction of Oat Bran Protein Using Response Surface Methodology. *Food Chemistry*. 106 (1), 345-351.
- Gulfo, R., Auleda, J. M., Raventós, M. and Hernández, E. (2014). Calculation Process for the Recovery of Solutes Retained in the Ice in a Multi-Plate Freeze Concentrator: Time and Concentration. *Innovative Food Science & Emerging Technologies*. 26, 347-359.
- Gunathilake, M., Dozen, M., Shimmura, K. and Miyawaki, O. (2014). An Apparatus for Partial Ice-Melting to Improve Yield in Progressive Freeze-Concentration. *Journal of Food Engineering*. 142 (0), 64-69.
- Gurak, P. D., Cabral, L. M. C., Rocha-Leão, M. H. M., Matta, V. M. and Freitas, S.
  P. (2010). Quality Evaluation of Grape Juice Concentrated by Reverse Osmosis. *Journal of Food Engineering*. 96 (3), 421-426.
- Habib, B. and Farid, M. (2006). Heat Transfer and Operating Conditions for Freeze Concentration in a Liquid–Solid Fluidized Bed Heat Exchanger. *Chemical Engineering and Processing: Process Intensification*. 45 (8), 698-710.
- Habib, B. and Farid, M. (2008). Freeze Concentration of Milk and Saline Solutions in a Liquid–Solid Fluidized Bed: Part Ii. Modelling of Ice Removal.

Chemical Engineering and Processing: Process Intensification. 47 (4), 539-547.

- Hartel, R. W. (2001). Crystallization in Foods. Gaithersburg: Aspen Publishers.
- He, T. X. and Yan, L. J. (2009). Application of Alternative Energy Integration Technology in Seawater Desalination. *Desalination*. 249 (1), 104-108.
- Hernández, E., Raventós, M., Auleda, J. M. and Ibarz, A. (2009). Concentration of Apple and Pear Juices in a Multi-Plate Freeze Concentrator. *Innovative Food Science & amp; Emerging Technologies*. 10 (3), 348-355.
- Hernández, E., Raventós, M., Auleda, J. M. and Ibarz, A. (2010). Freeze Concentration of Must in a Pilot Plant Falling Film Cryoconcentrator. *Innovative Food Science & amp; Emerging Technologies*. 11 (1), 130-136.
- Huige, N. J. J. and Thijssen, H. A. C. (1972). Production of Large Crystals by Continuous Ripening in a Stirrer Tank. *Journal of Crystal Growth*. 13–14 (0), 483-487.
- Incropera, F. P., Dewitt, D. P., Bergman, T. L. and Lavine, A. S. (2007). Fundamentals of Heat and Mass Transfer. (Sixth ed.). John Wiley and Sons (Asia) Pte Ltd.
- Iritani, E., Katagiri, N., Okada, K., Cao, D.-Q. and Kawasaki, K. (2013). Improvement of Concentration Performance in Shaking Type of Freeze Concentration. *Separation and Purification Technology*. 120 (0), 445-451.
- Istadi and Saidina Amin, N. A. (2005). A Hybrid Numerical Approach for Multi-Responses Optimization of Process Parameters and Catalyst Compositions in Co2 Ocm Process over Cao-Mno/Ceo2 Catalyst. *Chemical Engineering Journal*. 106 (3), 213-227.
- Istadi, I. and Saidina Amin, N. A. (2007). Modelling and Optimization of Catalytic– Dielectric Barrier Discharge Plasma Reactor for Methane and Carbon Dioxide Conversion Using Hybrid Artificial Neural Network—Genetic Algorithm Technique. *Chemical Engineering Science*. 62 (23), 6568-6581.
- Jesus, D. F., Leite, M. F., Silva, L. F. M., Modesta, R. D., Matta, V. M. and Cabral, L. M. C. (2007). Orange (Citrus Sinensis) Juice Concentration by Reverse Osmosis. *Journal of Food Engineering*. 81 (2), 287-291.
- Jiao, B., Cassano, A. and Drioli, E. (2004). Recent Advances on Membrane Processes for the Concentration of Fruit Juices: A Review. *Journal of Food Engineering*. 63 (3), 303-324.

- Joyce, A., Loureiro, D., Rodrigues, C. and Castro, S. (2001). Small Reverse Osmosis Units Using Pv Systems for Water Purification in Rural Places. *Desalination*. 137 (1–3), 39-44.
- Jusoh, M., Mohd Yunus, R. and Abu Hassan, M. A. (2009). Performance Investigation on a New Design for Progressive Freeze Concentration System. *Journal of Applied Sciences*. 9 (17), 3171-3175.
- Kagitani, K. and Hayakawa, K. (2006). Retrieved on from
- Kamari, A., Sattari, M., Mohammadi, A. H. and Ramjugernath, D. (2015). Reliable Method for the Determination of Surfactant Retention in Porous Media During Chemical Flooding Oil Recovery. *Fuel.* 158 (0), 122-128.
- Kawasaki, K., Matsuda, A. and Kadota, H. (2006). Freeze Concentration of Equal Molarity Solutions with Ultrasonic Irradiation under Constant Freezing Rate: Effect of Solute. *Chemical Engineering Research and Design*. 84 (2), 107-112.
- Kenari, S. L. D., Alemzadeh, I. and Maghsodi, V. (2011). Production of L-Asparaginase from Escherichia Coli Atcc 11303: Optimization by Response Surface Methodology. *Food and Bioproducts Processing*. 89 (4), 315-321.
- Kenny, R., Gorgol, R. G. and Prahacs, S. (1992). Freeze Crystallization of Temcell's Bctmp Effluent. *Pulp Paper Can.* 93 (10), 55-58.
- Keshani, S., Luqman Chuah, A., Nourouzi, M. M., A.R., R. and Jamilah, B. (2010). Optimization of Concentration Process on Pomelo Fruit Juice Using Response Surface Methodology (Rsm). *International Food Research Journal*. 17, 733-742.
- Khawaji, A. D., Kutubkhanah, I. K. and Wie, J.-M. (2008). Advances in Seawater Desalination Technologies. *Desalination*. 221 (1-3), 47-69.
- Khodaiyan, F., Razavi, S. H. and Mousavi, S. M. (2008). Optimization of Canthaxanthin Production by Dietzia Natronolimnaea Hs-1 from Cheese Whey Using Statistical Experimental Methods. *Biochemical Engineering Journal*. 40 (3), 415-422.
- Kimizuka, N., Viriyarattanasak, C. and Suzuki, T. (2008). Ice Nucleation and Supercooling Behavior of Polymer Aqueous Solutions. *Cryobiology*. 56 (1), 80-87.
- Kiss, I., Vatai, G. and Bekassy-Molnar, E. (2004). Must Concentration Using Membrane Technology. *Desalination*. 162, 295-300.

- Kobayashi, A., Shirai, Y., Nakanishi, K. and Matsuno, R. (1996). A Method for Making Large Agglomerated Ice Crystals for Freeze Concentration. *Journal* of Food Engineering. 27 (1), 1-15.
- König, A. and Schreiner, A. (2001). Purification Potential of Melt Crystallisation. *Powder Technology*. 121 (1), 88-92.
- Lee, W. C., Yusof, S., Hamid, N. S. A. and Baharin, B. S. (2006). Optimizing Conditions for Enzymatic Clarification of Banana Juice Using Response Surface Methodology (Rsm). *Journal of Food Engineering*. 73 (1), 55-63.
- Liesebach, J., Rades, T. and Lim, M. (2003). A New Method for the Determination of the Unfrozen Matrix Concentration and the Maximal Freeze-Concentration. *Thermochimica Acta*. 401 (2), 159-168.
- Liew Abdullah, A. G., Sulaiman, N. M., Aroua, M. K. and Megat Mohd Noor, M. J. (2007). Response Surface Optimization of Conditions for Clarification of Carambola Fruit Juice Using a Commercial Enzyme. *Journal of Food Engineering*. 81 (1), 65-71.
- Liu, L., Fuji, T., Hayakawa, K. and Miyawaki, O. (1998). Prevention of Initial Supercooling in Progressive Freeze Concentration. *Bioscience, Biotechnology* and Biochemistry. 62 (12), 2467-2469.
- Liu, L., Miyawaki, O. and Nakamura, K. (1997). Progressive Freeze Concentration of Model Liquid Food. *Food Science and Technology International Tokyo*. 3 (4), 348-352.
- Lorain, O., Thiebaud, P., Badorc, E. and Aurelle, Y. (2001). Potential of Freezing in Wastewater Treatment: Soluble Pollutant Applications. *Water Research*. 35 (2), 541-547.
- Lucas, S., Calvo, M. P., Palencia, C. and Cocero, M. J. (2004). Mathematical Model of Supercritical Co2 Adsorption on Activated Carbon: Effect of Operating Conditions and Adsorption Scale-Up. *The Journal of Supercritical Fluids*. 32 (1–3), 193-201.
- Luo, C. S., Chen, W. W. and Han, W. F. (2010). Experimental Study on Factors Affecting the Quality of Ice Crystal During the Freezing Concentration for the Brackish Water. *Desalination*. 260 (1-3), 231-238.
- Madaeni, S. S. and Zereshki, S. (2010). Energy Consumption for Sugar Manufacturing. Part I: Evaporation Versus Reverse Osmosis. *Energy Conversion and Management*. 51 (6), 1270-1276.

- Marler, R. T. and Arora, J. S. (2004). Survey of Multi-Objective Optimization Methods for Engineering. *Structural and Multidisciplinary Optimization*. 26 (6), 369-395.
- Matsuda, A., Kawasaki, K. and Kadota, H. (1999). Freeze Concentration with Supersonic Radiation under Constant Freezing Rate Effect of Kind and Concentration of Solutes. *Journal of Chemical Engineering of Japan.* 32 (5), 569-572.
- Mazaheri, H., Lee, K. T., Bhatia, S. and Mohamed, A. R. (2010). Subcritical Water Liquefaction of Oil Palm Fruit Press Fiber in the Presence of Sodium Hydroxide: An Optimisation Study Using Response Surface Methodology. *Bioresource Technology*. 101 (23), 9335-9341.
- McCabe, W. L., Smith, J. C. and Harriott, P. (2005). Unit Operations of Chemical Engineering (7th ed. ed., pp. 1140). Boston: McGraw-Hill.
- Melak, F., Du Laing, G., Ambelu, A. and Alemayehu, E. (2016). Application of Freeze Desalination for Chromium (Vi) Removal from Water. *Desalination*. 377, 23-27.
- Mezher, T., Fath, H., Abbas, Z. and Khaled, A. (2011). Techno-Economic Assessment and Environmental Impacts of Desalination Technologies. *Desalination*. 266 (1–3), 263-273.
- Miller, J. E. (2003). Review of Water Resources and Desalination Technologies
- Minhas, M. B., Jande, Y. A. C. and Kim, W. S. (2014). Combined Reverse Osmosis and Constant-Current Operated Capacitive Deionization System for Seawater Desalination. *Desalination*. 344 (0), 299-305.
- Miyawaki, O., Kato, S. and Watabe, K. (2012). Yield Improvement in Progressive Freeze-Concentration by Partial Melting of Ice. *Journal of Food Engineering*. 108 (3), 377-382.
- Miyawaki, O., Liu, L., Shirai, Y., Sakashita, S. and Kagitani, K. (2005). Tubular Ice System for Scale-up of Progressive Freeze-Concentration. *Journal of Food Engineering*. 69 (1), 107-113.
- Mohammad Fauzi, A. H., Amin, N. A. S. and Mat, R. (2014). Esterification of OleicAcid to Biodiesel Using Magnetic Ionic Liquid: Multi-ObjectiveOptimization and Kinetic Study. *Applied Energy*. 114 (0), 809-818.
- Montgomery, D. C. (2005). *Design and Analysis of Experiments*. (6th ed.). New York, USA: John Wiley and Sons.

- Montgomery, D. C., Bert Keats, J., Perry, L. A., Thompson, J. R. and Messina, W. S. (2000). Using Statistically Designed Experiments for Process Development and Improvement: An Application in Electronics Manufacturing. *Robotics* and Computer-Integrated Manufacturing. 16 (1), 55-63.
- Moreno, F. L., Hernández, E., Raventós, M., Robles, C. and Ruiz, Y. (2014a). A Process to Concentrate Coffee Extract by the Integration of Falling Film and Block Freeze-Concentration. *Journal of Food Engineering*. 128 (0), 88-95.
- Moreno, F. L., Raventós, M., Hernández, E. and Ruiz, Y. (2014b). Block Freeze-Concentration of Coffee Extract: Effect of Freezing and Thawing Stages on Solute Recovery and Bioactive Compounds. *Journal of Food Engineering*. 120 (0), 158-166.
- Nakagawa, K., Maebashi, S. and Maeda, K. (2010). Freeze-Thawing as a Path to Concentrate Aqueous Solution. *Separation and Purification Technology*. 73 (3), 403-408.
- Nandi, S., Mukherjee, P., Tambe, S. S., Kumar, R. and Kulkarni, B. D. (2002). Reaction Modeling and Optimization Using Neural Networks and Genetic Algorithms: Case Study Involving Ts-1-Catalyzed Hydroxylation of Benzene. *Industrial & Engineering Chemistry Research*. 41 (9), 2159-2169.
- Nourbakhsh, H., Emam-Djomeh, Z., Omid, M., Mirsaeedghazi, H. and Moini, S. (2014). Prediction of Red Plum Juice Permeate Flux During Membrane Processing with Ann Optimized Using Rsm. *Computers and Electronics in Agriculture*. 102 (0), 1-9.
- Okawa, S., Ito, T. and Saito, A. (2009). Effect of Crystal Orientation on Freeze Concentration of Solutions. *International Journal of Refrigeration*. 32 (2), 246-252.
- Paul, E. L., Tung, H.-H. and Midler, M. (2005). Organic Crystallization Processes. *Powder Technology*. 150 (2), 133-143.
- Peñate, B. and García-Rodríguez, L. (2012). Current Trends and Future Prospects in the Design of Seawater Reverse Osmosis Desalination Technology. *Desalination*. 284 (0), 1-8.
- Perry, R. H. and Green, D. W. (1997). *Perry's Chemical Engineers' Handbook*. (7th ed.). USA: McGraw-Hill.
- Petzold, G. and Aguilera, J. M. (2013). Centrifugal Freeze Concentration. *Innovative Food Science & Emerging Technologies*. 20 (0), 253-258.

- Petzold, G., Niranjan, K. and Aguilera, J. M. (2013). Vacuum-Assisted Freeze Concentration of Sucrose Solutions. *Journal of Food Engineering*. 115 (3), 357-361.
- Prakash Maran, J., Sivakumar, V., Thirugnanasambandham, K. and Sridhar, R. (2013). Artificial Neural Network and Response Surface Methodology Modeling in Mass Transfer Parameters Predictions During Osmotic Dehydration of Carica Papaya L. *Alexandria Engineering Journal*. 52 (3), 507-516.
- Prawitwong, P., Takigami, S. and Phillips, G.-O. (2007). Phase Transition Behavior of Sorbed Water in Konjac Mannan. *Food Hydrocolloids*. 21 (8), 1368–1373.
- Qin, X., Chen, L. and Sun, F. (2008). Performance of Real Absorption Heat-Transformer with a Generalized Heat Transfer Law. Applied Thermal Engineering. 28 (7), 767-776.
- Rafigh, S. M., Yazdi, A. V., Vossoughi, M., Safekordi, A. A. and Ardjmand, M. (2014). Optimization of Culture Medium and Modeling of Curdlan Production from Paenibacillus Polymyxa by Rsm and Ann. *International Journal of Biological Macromolecules*. 70 (0), 463-473.
- Ramos, F. A., Delgado, J. L., Bautista, E., Morales, A. L. and Duque, C. (2005). Changes in Volatiles with the Application of Progressive Freeze-Concentration to Andes Berry (Rubus Glaucus Benth). *Journal of Food Engineering*. 69 (3), 291-297.
- Ramteke, R. S., Singh, N. I., Rekha, M. N. and Eipeson, W. E. (1993). Methods for Concentration of Fruit Juices: A Critical Evaluation. *Journal of Food Science and Technology*. 30, 391–402.
- Randall, D. G. and Nathoo, J. (2015). A Succinct Review of the Treatment of Reverse Osmosis Brines Using Freeze Crystallization. *Journal of Water Process Engineering*. 8, 186-194.
- Rane, M. V. and Padiya, Y. S. (2011). Heat Pump Operated Freeze Concentration System with Tubular Heat Exchanger for Seawater Desalination. *Energy for Sustainable Development*. 15 (2), 184-191.
- Rich, A., Mandri, Y., Mangin, D., Rivoire, A., Abderafi, S., Bebon, C., Semlali, N.,Klein, J.-P., Bounahmidi, T., Bouhaouss, A. and Veesler, S. (2012). SeaWater Desalination by Dynamic Layer Melt Crystallization: Parametric Study

of the Freezing and Sweating Steps. *Journal of Crystal Growth*. 342 (1), 110-116.

- Rodriguez, M., Luque, S., Alvarez, J. and Coca, J. (2000). A Comparative Study of Reverse Osmosis and Freeze Concentration for the Removal of Valeric Acid from Wastewaters. *Desalination*. 127 (1), 1-11.
- Roos, A. C., Verschuur, R. J., Schreurs, B., Scholz, R. and Jansens, P. J. (2003). Development of a Vacuum Crystallizer for the Freeze Concentration of Industrial Waste Water. *Chemical Engineering Research and Design.* 81 (8), 881-892.
- Ruby-Figueroa, R., Cassano, A. and Drioli, E. (2012). Ultrafiltration of Orange Press Liquor: Optimization of Operating Conditions for the Recovery of Antioxidant Compounds by Response Surface Methodology. *Separation and Purification Technology*. 98 (0), 255-261.
- Sachindra, N. M. and Mahendrakar, N. S. (2005). Process Optimization for Extraction of Carotenoids from Shrimp Waste with Vegetable Oils. *Bioresource Technology*. 96 (10), 1195-1200.
- Sahasrabudhe, A. B., Desai, R. R. and Jabade, S. K. (2012). Modeling and Simulation of a Freeze Concentration Technique for Sugarcane Juice Concentration. *Applied Mechanics and Materials*. 110-116, 2768-2773.
- Samsuri, S., Amran, N. A. and Jusoh, M. (2015). Spiral Finned Crystallizer for Progressive Freeze Concentration Process. *Chemical Engineering Research* and Design. 104, 280-286.
- Sánchez, J., Hernández, E., Auleda, J. M. and Raventós, M. (2011). Freeze Concentration of Whey in a Falling-Film Based Pilot Plant: Process and Characterization. *Journal of Food Engineering*. 103 (2), 147-155.
- Sánchez, J., Ruiz, Y., Auleda, J. M., Hernández, E. and Raventós, M. (2009). Review. Freeze Concentration in the Fruit Juices Industry. *Food Science and Technology International*. 15 (4), 303-315.
- Semiat, R. (2008). Review: Energy Issues in Desalination Processes. Environmental Science and Technology. 42 (22), 8193-8201.
- Sharqawy, M. H., John, H. L. V. and Syed, M. Z. (2010). The Thermophysical Properties of Seawater: A Review of Existing Correlations and Data. *Desalination and Water Treatment*. 16, 354–380.

- Shen, C., Wang, L. and Li, Q. (2007). Optimization of Injection Molding Process Parameters Using Combination of Artificial Neural Network and Genetic Algorithm Method. *Journal of Materials Processing Technology*. 183 (2–3), 412-418.
- Shirai, Y., Wakisaka, M., Miyawaki, O. and Sakashita, S. (1998). Conditions of Producing an Ice Layer with High Purity for Freeze Wastewater Treatment. *Journal of Food Engineering*. 38 (3), 297-308.
- Shirai, Y., Wakisaka, M., Miyawaki, O. and Sakashita, S. (1999). Effect of Seed Ice on Formation of Tube Ice with High Purity for a Freeze Wastewater Treatment System with a Bubble-Flow Circulator. *Water Research*. 33 (5), 1325-1329.
- Sin, H. N., Yusof, S., Sheikh Abdul Hamid, N. and Rahman, R. A. (2006). Optimization of Enzymatic Clarification of Sapodilla Juice Using Response Surface Methodology. *Journal of Food Engineering*. 73 (4), 313-319.
- Sinha, K., Chowdhury, S., Saha, P. D. and Datta, S. (2013). Modeling of Microwave-Assisted Extraction of Natural Dye from Seeds of Bixa Orellana (Annatto) Using Response Surface Methodology (Rsm) and Artificial Neural Network (Ann). *Industrial Crops and Products*. 41 (0), 165-171.
- Sobati, M. A. and Abooali, D. (2015). Molecular Based Models for Estimation of Critical Properties of Pure Refrigerants: Quantitative Structure Property Relationship (Qspr) Approach. *Thermochimica Acta*. 602 (0), 53-62.
- Sundqvist, Å., Sellgren, A. and Addie, G. (1996a). Pipeline Friction Losses of Coarse Sand Slurries. Comparison with a Design Model. *Powder Technology*. 89 (1), 9-18.
- Sundqvist, Å., Sellgren, A. and Addie, G. (1996b). Slurry Pipeline Friction Losses for Coarse and High Density Industrial Products. *Powder Technology*. 89 (1), 19-28.
- Tan, I. A. W., Ahmad, A. L. and Hameed, B. H. (2008). Preparation of Activated Carbon from Coconut Husk: Optimization Study on Removal of 2,4,6-Trichlorophenol Using Response Surface Methodology. *Journal of Hazardous materials*. 153 (1–2), 709-717.
- Teng, L. Y., Chin, N. L. and Yusof, Y. A. (2011). Rheological and Textural Studies of Fresh and Freeze-Thawed Native Sago Starch–Sugar Gels. I. Optimisation

Using Response Surface Methodology. *Food Hydrocolloids*. 25 (6), 1530-1537.

- Tirmizi, S. H. and Gill, W. N. (1989). Experimental Investigation of the Dynamics of Spontaneous Pattern Formation During Dendritic Ice Crystal Growth. *Journal* of Crystal Growth. 96, 277-292.
- Trinh, T. K. and Kang, L. S. (2011). Response Surface Methodological Approach to Optimize the Coagulation–Flocculation Process in Drinking Water Treatment. *Chemical Engineering Research and Design*. 89 (7), 1126-1135.
- Wakisaka, M., Shirai, Y. and Sakashita, S. (2001). Ice Crystallization in a Pilot-Scale Freeze Wastewater Treatment System. *Chemical Engineering and Processing: Process Intensification*. 40 (3), 201-208.
- Wu, D., Li, Y., Shi, Y., Fang, Z., Wu, D. and Chang, L. (2002). Effects of the Calcination Conditions on the Mechanical Properties of a Pcomo/Al2o3 Hydrotreating Catalyst. *Chemical Engineering Science*. 57 (17), 3495-3504.
- Youssef, P. G., Al-Dadah, R. K. and Mahmoud, S. M. (2014). Comparative Analysis of Desalination Technologies. *Energy Procedia*. 61 (0), 2604-2607.
- Zhang, Z. and Hartel, R. W. (1996). A Multilayer Freezer for Freeze Concentration of Liquid Milk. *Journal of Food Engineering*. 29 (1), 23-38.