

SYNTHESIS OF EUGENYL BENZOATE CATALYZED BY *RHIZOMUCOR*
MIEHEI LIPASE IMMOBILIZED ON REINFORCED CHITOSAN-CHITIN
NANOWHISKERS SUPPORT

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DEDICATION

To my parents,

Abd Manan and Fatimah,

And all of my friends,

Without whom none of my success would be possible

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In the name of Allah, the Most Gracious and the Most Merciful

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ABSTRACT

The chemical route to produce esters has several drawbacks associated with the utilization of homogeneous acid catalysts that require high reaction temperatures, hence is energy intensive, not reusable, creates dissolved solids and involves laborious separation processes. Hence, a biotechnological route *via* enzymatic esterification has been proposed as an alternative way to synthesize the problematic anti-oxidant, eugenyl benzoate. This is because an enzyme-catalyzed synthesis offers favorable advantages such as the use of a more sustainable reaction process with high yields and purity, as well as the biocatalyst reusability. An ecofriendly support was prepared from chitosan-chitin nanowhiskers (CS-CNWs) for covalent immobilization of *Rhizomucor miehei* lipase (RML) to be used as the biocatalysts. Analyses on the RML-CS/CNWs using Fourier transform infrared spectroscopy (FTIR), thermogravimetric analysis-differential thermogravimetry (TGA-DTG), field emission scanning electron microscopy (FESEM), X-ray diffraction (XRD) and fluorescence microscopy affirmed the successful covalent immobilization of RML onto the surface of CS-CNWs. The resultant RML-CS/CNWs biocatalysts were studied for catalyzing synthesis of eugenyl benzoate for various reaction parameters. One-Variable-at-A-Time (OVAT) study revealed that under optimal experimental conditions of 50 °C at 250 rpm with catalyst loading of 3 mg/mL and 3:1 molar ratio of eugenol to benzoic acid, the maximum yield that reached 62.1% was attained after 5 h as compared to free RML (50.7%). The RML-CS-CNWs also demonstrated good operational stability, whereby the biocatalysts retaining 50% of its initial activity for up to eight successive esterification cycles. The present work also reports a response surface methodology (RSM) with Box-Behnken design (BBD) optimization process to synthesize eugenyl benzoate. The effects of four reaction parameters: reaction time, temperature, the substrate molar ratio of eugenol to benzoic acid and enzyme loading were assessed based on OVAT findings. Under optimum conditions, a maximum conversion yield of 66.8% was attained at 50 °C in 5 h using 3.75 mg/mL of the RML-CS-CNWs, and substrate molar ratio (eugenol:benzoic acid) of 3:1. Meanwhile, kinetic assessments revealed the RML-CS/CNWs catalyzed the reaction *via* a Ping-pong Bi Bi mechanism with eugenol inhibition, characterized by a V_{\max} of 3.83 mM min⁻¹ and turnover number (k_{cat}) of 40.39 min⁻¹ under an optimized experimental condition. Based on the findings, it can be concluded that the use of the RML-CS-CNWs biocatalysts was promising in affording relatively satisfactory yield of eugenyl benzoate within a reasonably short time. Aside from improving enzymatic operational activity and stability, the immobilization strategy can facilitate rapid and easy removal of the RML-CS-CNWs from the reaction mixture when completed.

ABSTRAK

Laluan kimia untuk menghasilkan ester mempunyai beberapa kelemahan yang dikaitkan dengan penggunaan mangkin asid homogen yang memerlukan suhu tindak balas yang tinggi, maka ianya intensif tenaga, tidak boleh diguna semula, menghasilkan pepejal terlarut dan melibatkan proses pemisahan yang menjerihkan. Maka, laluan bioteknologi melalui pengesteran berenzim telah dicadangkan sebagai suatu laluan alternatif untuk mensintesis anti-oksida yang bermasalah, eugenil benzoat. Ini kerana sintesis bermangkin enzim menawarkan kelebihan menguntungkan iaitu penggunaan proses tindak balas yang lebih lestari dengan hasil yang tinggi dan tulen, serta kebolehgunaan semula biomangkin tersebut. Suatu penyokong mesra alam telah disediakan daripada nanomisai kitin-kitosan (CS-CNWs) untuk pemegunan kovalen lipase *Rhizomucor miehei* (RML) untuk digunakan sebagai biomangkin. Analisis terhadap RML-CS-CNWs menggunakan spektroskopi inframerah transformasi Fourier (FTIR), analisis termogravimetri-termogravimetri pembezaan (TGA-DTG), mikroskopi imbasan elektron-medan pelepasan (FESEM), pembelauan sinar-X (XRD) dan mikroskopi pendarfluor membuktikan kejayaan pemegunan kovalen RML ke atas permukaan CS/CNWs. Biomangkin RML-CS-CNWs yang terhasil telah dikaji untuk memangkin sintesis eugenil benzoat untuk pelbagai parameter tindak balas. Kajian satu-pembolehubah-pada-satu-masa (OVAT) mendedahkan bahawa dalam keadaan optimum eksperimen iaitu 50 °C pada 250 rpm dengan muatan mangkin sebanyak 3 mg/mL dan nisbah molar eugenol kepada asid benzoik 3:1, hasil maksimum yang menjangkau 62.1% telah dicapai selepas 5 jam berbanding dengan RML bebas (50.7%). RML-CS-CNWs tersebut juga menunjukkan kestabilan operasi yang baik, di mana biomangkin mengekalkan 50% daripada aktiviti awal sehingga lapan kitaran pengesteran berturut-turut. Kajian ini turut melaporkan proses pengoptimuman menggunakan kaedah gerak balas permukaan (RSM) dengan reka bentuk Box-Behnken (BBD) untuk mensintesis eugenil benzoat. Kesan empat parameter tindak balas: masa tindak balas, suhu, nisbah molar substrat eugenol kepada asid benzoik dan muatan enzim dinilai berdasarkan keputusan OVAT. Dalam keadaan optimum, hasil penukaran maksimum sebanyak 66.8 % telah dicapai pada 50 °C dalam waktu 5 jam menggunakan 3.75 mg/mL RML-CS-CNWs, dan nisbah molar substrat (eugenol:asid benzoik) sebanyak 3:1. Sementara itu, penilaian kinetik mendedahkan RML-CS-CNWs memangkin tindak balas melalui mekanisma Ping-pong Bi Bi dengan perencatan eugenol, dicirikan oleh V_{max} 3.83 mM min⁻¹ dan nombor perolehan (k_{cat}) sebanyak 40.39 min⁻¹ di bawah keadaan eksperimen yang optimum. Berdasarkan keputusan kajian, dapat disimpulkan bahawa penggunaan biomangkin RML-CS-CNWs menjanjikan hasil eugenil benzoat yang relatif memuaskan di dalam jangka masa yang agak singkat. Selain daripada meningkatkan aktiviti operasi enzim dan kestabilan, strategi pemegunan membolehkan pengeluaran RML-CS-CNWs yang pantas dan mudah daripada campuran tindak balas setelah selesai.

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LIST OF ABBREVIATIONS

ANOVA	-	Analysis of Variance
BBD	-	Box-Behnken design
CNWs	-	Chitin nanowhiskers
CS	-	Chitosan
EDAC	-	1-Ethyl-3-[3-dimethylaminopropyl]carbodiimide hydrochloride
MES	-	2-(N-morpholino)ethanesulfonic acid
NHS	-	N-hydroxysuccinimide sodium salt
OVAT	-	One Variable at-a-Time
RML	-	<i>Rhizomucor miehei</i> lipase
RSM	-	Response surface methodology

LIST OF SYMBOLS

hr	-	hour
g	-	gram
M	-	molar (concentration)
mg	-	milligram
min	-	minute
mL	-	millilitre
mM	-	millimolar (concentration)
rpm	-	rotation per minute
s	-	second

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CHAPTER 1

INTRODUCTION

1.1 Background of Study

Eugenol derived from *Eugenia caryophyllata* is the principal chemical component of clove oil (76.8%) and the compound is popularly known for its excellent analgesic, anti-inflammatory and antibacterial effects (Moon *et al.*, 2011). Eugenol is a pale-yellow oil with a warm, pungent yet pleasing aroma and smell of bay leaves and clove. It is a familiar fragrance in dentistry as it is often mixed into a paste and used as a local antiseptic, dental cement, filler and restorative material (Devi *et al.*, 2010). Eugenol is usually used in its dimeric forms (i.e. eugenol esters) for increased bioactivity (Horchani *et al.*, 2010; Sadeghian *et al.*, 2008) while reducing allergic reactions and inflammatory effects when in contact with cells due to liberation of phenoxy radicals as well as quinone intermediates (Horchani *et al.*, 2010; Yadav and Yadav, 2012). In terms usage safety, the use of eugenol has been deemed safe by the Food and Drug Administration (Hemaiswarya and Doble, 2009).

The current chemical approach to synthesizing eugenol esters has several inherent drawbacks, among them is the use of strong acid catalysts that incurs tedious downstream processing which increases the overall costs of manufacturing process (Yadav and Yadav, 2012). To overcome such drawbacks, the biotechnological approach of employing enzymes in non-aqueous media may prove advantageous since all reactions are carried out under mild conditions (Horchani *et al.*, 2010). The feasibility of such method in employing enzyme as biocatalyst was

previously reported by several researchers (Chiaradia *et al.*, 2012; Horchani *et al.*, 2010; Mohamad *et al.*, 2015b). Herein, employment of lipases (triacylglycerol ester hydrolases, EC 3.1.1.3) to catalyze the synthesis of eugenol esters i.e. eugenyl benzoate is proposed.

Currently, lipase from *Rhizomucor miehei* (RML) has gained considerable attention and successfully employed in various synthesis reactions, especially esterification. Such preference of the scientific and manufacturing community towards RML may be attributable to numerous reports on its catalytic potential at different conditions of temperature, pressure, water content and substrates (Lorenzoni *et al.*, 2012; Skoronski *et al.*, 2014). Similar to other free forms of lipases, RML has a high tendency of deactivation in prolonged contact with high temperature and extreme pH, low activity in organic solvents and, hence insufficiently stable to withstand tough industrial processing conditions (Rodrigues and Fernandez-Lafuente, 2010). In this context, immobilization of RML onto a suitable solid support is one of the possible solutions that offers easy recovery and reusability of the biocatalysts for better productivity (Zou *et al.*, 2010), as well as improved stability and activity of enzymes (Mateo *et al.*, 2007; Zhao *et al.*, 2015).

In the present work, the organic polymers of chitosan (CS) and chitin nanowhiskers (CNWs) are chosen as the matrices for immobilization due to their excellent benefits. CS is a natural cationic biopolymer with the benefits of biocompatibility, non-toxicity as well as high mechanical strength, whereby the material can be easily fabricated into high quality films, fibres or bead forms. The versatility of this biomaterial is also due to the presence of amino and hydroxyl groups in its structure that facilitate attachment of enzymes *via* covalent bonding or crosslinking (Solanki *et al.*, 2009). Moreover, the performance of CS as the support for enzyme immobilization can be enhanced by introducing certain nanomaterials such as CNWs as filler. The use of CNWs has become increasingly popular in recent years due to their exceptional mechanical properties, large specific surface area, high aspect ratio, environmental benefits and low cost (Qin *et al.*, 2016). According to a recent report, the combination a polymer i.e. CS with the nano-filler CNWs to afford the CS/CNWs hybrid matrix can be regarded as a single polymer (polysaccharide)

composite (Šimkovic, 2013) whose mechanical parameters are strongly influenced by the origin of the components. The observably enhanced strength and stiffness of the CS/CNWs composite has been suggested to be a consequence of the high modulus of CNWs and additional interactions that occurred between both components (Kelnar *et al.*, 2015).

The process parameters for the RML-CS/CNWs-catalyzed esterification to synthesize eugenyl benzoate was statistically optimized using the method of response surface methodology (RSM). The software can predict the best reaction conditions that would maximize the yield of the ester *via* a statistically optimized model without requiring arduous and time-consuming experiments (Marzuki *et al.*, 2015a). RSM merges the experimental designs with interpolation by first or second-order polynomial equations in a sequential testing procedure (Pandiyan *et al.*, 2014) that allows good estimation of the optimized parameters in the RML-CS/CNWs-catalyzed esterification process. Herein, the present study aimed to model the RML-CS/CNWs-catalyze esterification of eugenol and benzoic acid using a three-level-four-factor Box-Behnken design (BBD).

This present study was aimed at investigating the optimized reaction parameters for the application of RML-CS/CNWs as potential sustainable biocatalysts. The developed RML-CS/CNWs beads was used as biocatalysts for the preparation of eugenyl benzoate from eugenol and benzoic acid. The esterification process was optimized using the OVAT and RSM by Box-Behnken Design (BBD) for relevant parameters that include incubation time, temperature, molar ratio of acid to alcohol and enzyme loading.

1.2 Statement of Problem

Currently, eugenyl benzoate is chemically synthesized as one of the important inhibitors of lipoxygenase, which is responsible in avoiding major cause of inflammation in asthma and allergic rhinitis (Sadeghian *et al.*, 2008). However, the

prevailing chemical route to produce eugenyl benzoate may harm the environment as well as requiring a lot of energy and time (Chaibakhsh *et al.*, 2012), development of new methods that are sustainable and cost effective to overcome such drawbacks may prove pertinent and timely. So far, attempts to produce eugenyl benzoate *via* the biotechnological route (Horchani *et al.*, 2010) remains lacking and the potential of such method to produce the ester has yet to be fully exploited. Although there are studies employing lipases as a biocatalyst in the synthesis of eugenyl benzoate, the outcome has been unsatisfactory due to the low conversion yield or the use of lipase derived from the pathogenic *Staphylococcus aureus*, in which the latter may pose adverse implications towards human health (Bartolomeu *et al.*, 2016; Hu *et al.*, 2012).

In this study, covalently bound RML onto CNWs reinforced chitosan (RML-CS/CNWs) will be used as biocatalysts for the lab scale synthesis of eugenyl benzoate. It was previously described that CS reinforced with CNWs using the biodegradable tannic acid as the crosslinker favorably altered the stability and the mechanical properties of the resultant CS/CNWs carrier support (Rubentheren *et al.*, 2015). It is hypothesized that the covalent attachment of RML onto CS/CNWs may confer certain benefits of biocompatibility of both supports to RML and additional structural integrity to the RML protein structure, thereby increasing operational stability of the nanobioconjugates for improved yield of eugenyl benzoate.

1.3 Objectives of the Study

The objectives of this study are:

- i. To characterize the morphology, physicochemical properties and stability of the RML-CS/CNWs.
- ii. To optimize the RML-CS/CNWs catalyzed synthesis of eugenyl benzoate.
- iii. To develop kinetic model for the RML-CS/CNWs catalyzed synthesis of eugenyl benzoate.

1.4 Scope of Study

The scopes of this project involve the preparation of chitin nanowhiskers using acid hydrolysis reaction in which the supports of CS/CNWs bead are prepared for the immobilization of the RML. The study subsequently assessed the morphological characteristics of the CS/CNWs and RML-CS/CNWs beads by:

- a) Fourier transform infrared (FTIR)
- b) Field emission scanning electron microscopy (FESEM)
- c) Thermal gravimetric analysis (TGA)
- d) X-ray diffraction (XRD)
- e) Fluorescence optical microscopy

Next, the study of the characterization of the physicochemical properties and optimize the RML-CS/CNWs assisted esterification of eugenol and benzoic acid to afford eugenyl benzoate was carried out using the method of OVAT for parameters temperature, amount of enzyme, incubation time, substrate molar ratio (alcohol:acid), stirring rate, reusability and thermal stability. The ANOVA on the results of the OVAT study is crucial to identify the four relevant parameters to be investigated in the subsequent RSM study.

The following part of the study is the optimization of the RML-CS/CNWs assisted synthesis of eugenyl benzoate using the method of RSM for four relevant parameters as the following: temperature, incubation time, amount of enzyme and molar ratio of acid to alcohol, according to the proposed conditions by the Design Expert 7.1.6 software utilizing the BBD method. The response of the reaction is determined in terms of the percentage yield of eugenyl benzoate. Lastly, the kinetic study for the RML-CS/CNWs catalyzed esterification of eugenol and benzoic acid based on different concentrations of the substrates were carried out to ascertain the mechanism that the developed biocatalysts and to identify the kinetic parameters of the lipase *viz.* V_{max} , K_m , K_{cat} and K_{eff} .

1.5 Significance of Study

The RML-CS/CNWs developed in this study offers a one-pot synthesis that promotes the use of the Green Chemistry philosophy in producing eugenyl benzoate while promoting lesser use of environmentally unfriendly chemicals and hazardous acids. In addition, the modification of the CS/CNWs polymer matrix by introducing tannic acid as the crosslinker is beneficial owing to its biodegradability, non-cytotoxicity and less expensive to produce (Rubentheren *et al.*, 2015). Moreover, an enzyme-assisted reaction is noteworthy carried out under mild conditions and the biocatalyst can be reused for several cycles of reaction, hence is prospectively cost-saving. Moreover, application of RML-CS/CNWs as the biocatalyst would make a significant cost reduction as the amount of enzyme utilized is low.

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