

SENSING PROPERTIES OF DEFORMED GRAPHENE NANORIBBON  
THROUGH WARPING

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Specially dedication to my parents, siblings and friends for their love and respect

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## ABSTRACT

Semiconductor materials have become the fundamental building block in the design of gas sensor. Previous researchers work on gas sensor that recognized its limitations such as low binding energy, low sensitivity and poor selectivity. As an alternative, considerable interests have generated in carbon based material like graphene to improve the sensing device performance. The purpose of this study is to investigate the sensing behaviour of deformed armchair graphene nanoribbon (AGNR) through warping using Extended-Huckel Theory (EHT) coupled with Non-Equilibrium Green Function (NEGF). The AGNR are warped upward and inward at angles of  $180^\circ$ ,  $270^\circ$  and  $360^\circ$ . The sensing properties are measured for oxygen ( $O_2$ ) and ammonia ( $NH_3$ ) molecules, particularly on their binding energy, charge transfer and sensitivity. Generally, inward warped AGNR shows better performance compared to upward deformation. Three AGNR configurations have been investigated;  $3m$ ,  $3m+1$  and  $3m+2$ , where  $m$  known as an integer. Performance of each AGNR configuration behaves differently with the applied warping. Simulation results have indicated that for upward  $3m$  configuration, the warping AGNR exhibit 98% enhancement in binding energy when warped at  $360^\circ$  for  $NH_3$  molecule. While for  $3m+1$  and  $3m+2$  the binding energy exhibit 72% and 64% improvement respectively. The same trend of observation is achieved for  $O_2$  molecules. The results obtained also discovered a chemisorption in  $3m$ ,  $3m+1$  and  $3m+2$  configuration for both  $O_2$  and  $NH_3$ . For  $3m$  and  $3m+1$  configurations, the sensitivity has been observed at two-order of magnitude for higher warping angles, which most previous studies have not achieved. Meanwhile, negative sensitivity is observed in  $3m+2$  configuration. The marked improvement of the warped AGNR sensing properties is attributed from the combination of strain and curvature effect. The warping can also be an alternative method to minimize drawbacks in traditional gas sensors. The sensitivity of gas sensor could be enhanced by introducing the warped AGNR.

## ABSTRAK

Bahan semikonduktor telah menjadi asas pembangunan dalam reka bentuk sensor gas. Penyelidikan sebelum ini ke atas sensor gas mendapati kelemahan seperti tenaga ikatan yang lemah, kurang sensitif dan kepilihan. Sebagai alternatif, bahan berasaskan karbon seperti grafin digunakan untuk meningkatkan prestasi peranti penderiaan. Tujuan kajian ini adalah untuk mengkaji tingkah laku penderiaan dengan menggunakan *armchair graphene nanoribbon* (AGNR) yang telah diubah bentuk dengan penggunaan *Extended-Huckel Theory* (EHT) serta *Non-Equilibrium Green Function* (NEGF). AGNR diledingkan ke atas and ke bawah dengan sudut  $180^\circ$ ,  $270^\circ$  dan  $360^\circ$ . Sifat-sifat penderiaan diukur untuk mengkaji gas oksigen ( $O_2$ ) dan ammonia ( $NH_3$ ) kursusnyanya pada tenaga ikatan, pemindahan cas dan sensitiviti. Umumnya, AGNR yang meleding ke bawah menunjukkan prestasi penderiaan yang lebih baik berbanding dengan meleding ke atas. Tiga AGNR konfigurasi telah disiasat;  $3m$ ,  $3m+1$  dan  $3m+2$ , di mana  $m$  adalah integer. Prestasi setiap AGNR konfigurasi bertindak berbeza dengan ledingan. Keputusan, simulasi menunjukkan bahawa konfigurasi  $3m$  meleding ke atas, mempamerkan peningkatan 98% dalam tenaga ikatan ketika leding pada sudut  $360^\circ$  untuk molekul  $NH_3$ . Sementara,  $3m+1$  dan  $3m+2$  menunjukkan peningkatan tenaga ikatan masing-masing 72% dan 64%. Trend yang sama dicapai untuk molekul  $O_2$ . Hasil yang diperoleh juga menemui pengkimierapan bagi konfigurasi  $3m$ ,  $3m+1$  dan  $3m+2$  untuk kedua-dua  $O_2$  and  $NH_3$ . Untuk konfigurasi  $3m$  dan  $3m+1$ , sensitiviti telah diperhatikan pada dua tertib magnitud untuk sudut meleding yang lebih tinggi, di mana tidak dicapai dalam kebanyakan kajian sebelum ini. Sementara itu, sensitiviti negatif diperhatikan dalam konfigurasi  $3m+2$  untuk molekul  $O_2$ . Penambahbaikan yang ketara diperolehi dalam prestasi penderiaan daripada AGNR yang meleding dan ini adalah kesan daripada gabungan antara ketegangan dan keledingan. Keledingan ini juga boleh menjadi salah satu kaedah untuk meminimumkan kelemahan dalam sensor gas tradisional. Sensitiviti sensor gas dapat ditingkatkan dengan memperkenalkan ledingan pada AGNR

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**LIST OF ABBREVIATIONS**

1D	-	One-dimensional
2D	-	Two-dimensional
AGNR	-	Armchair Graphene Nanoribbon
ATK	-	Atomistic ToolKit
CNT	-	Carbon Nanotube
DFT	-	Density Functional Theory
DOS	-	Density of state
EHT	-	Extended Huckel Theory
GGA	-	Generalized Gradient Approximation
GNR/FET	-	Graphene Nanoribbon Field Effect Transistor
MOS	-	Metal oxide semiconductor
NEGF	-	Non Equilibrium Green Function
SCF	-	Self-consistency calculation
SV	-	Single Vacancy
SW	-	Stone-Wales
TB	-	Tight Binding
TS	-	Transmission Spectrum
VDW	-	Van der Waals
ZGNR	-	Zigzag Graphene Nanoribbon

**LIST OF SYMBOLS**

$CO$	-	Carbon monoxide
$CO_2$	-	Carbon dioxide
$E_a$	-	Binding energy
$NH_3$	-	Ammonia
$NiO$	-	Nickel oxide
$NO_2$	-	Nitrogen dioxide
$O_2$	-	Oxygen
$S$	-	Sensitivity
$SnO_2$	-	Tin dioxide
$TiO_2$	-	Titanium dioxide
$WO_3$	-	Tungsten trioxide
$\epsilon$	-	Strain
$\Gamma$	-	Gamma Function
$\pi$	-	Band Energy

## CHAPTER 1

### INTRODUCTION

#### 1.1 Background Study

Gas sensor is a device which detects the presence of various gases in a particular area. It has received much attention recently in both domestic and industrial application [1]. Sensing devices have wide application in various fields such as industrial, medical diagnosis, agricultural, chemical process and environmental monitoring safety. Based on studies done, the suitable competitor for the development of gas sensor for this wide range of application is solid state gas sensor [1, 2]. This solid state gas sensor owns its popularity based on its small size, high sensitivity for low concentration range of parts-per-million (ppm), low power consumption and cost. Besides that, it is also shown that solid state gas sensor has reversible interaction of gas with solid state material used [3]. These advantages have captured great interest of scientists and industries. In addition, this type of gas sensor has been used widely for chemical sensor. However, it suffers from stability for long term and measurement accuracy [1].

Capone *et al.* [1] discussed that semiconductor gas sensor is usually based on metal oxide semiconductor (MOS) such as tin dioxide ( $\text{SnO}_2$ ), titanium dioxide ( $\text{TiO}_2$ ), tungsten trioxide ( $\text{WO}_3$ ) and nickel oxide ( $\text{NiO}$ ). The MOS gas sensor has advantages in cost due to the simplicity in the architecture. This is desirable for the manufacturer to build MOS sensor. However, the disadvantages are that this MOS sensor has issue in power consumption and sensitivity to react with gas molecules. Using conventional material silicon has finally reached its limits and hence researchers are actively exploring other advanced material to replace silicon for better performance[4, 5]. By employing nanotechnology it provides enhanced solid state gas sensor performance, for example ultra-high sensitivity, fast response time and recovery and high specificity [6, 7].

Today's main focus in gas sensing aims to obtain high selectivity, sensitivity and low recovery time. Numerous studies have been established using carbon nanotubes (CNT) [8, 9], graphene [10, 11], graphene nanoribbon (GNR) [12, 13] which allow detection of various types of gases. Low-dimensional carbon structures have been the focus of extensive research since the discovery of fullerenes and (CNTs) [14]. These novel forms of carbon present unique opportunities to study low-dimensional physical phenomena. Depending on their chirality and diameter, CNT can behave as semiconducting or metallic. They are also known to have high chemical reactivity due to the strain present in their curved lattice. However, it has been found that the presence of chirality in CNT causes difficulty in the fabrication process. This is due to the fact that CNT is highly reactive, which easily reacts with other unintentionally contaminant that causes difficulty in the fabrication process.

Graphene is a planar allotrope of carbon where all of the carbon atoms form a covalent bond in a single plane. Its two-dimensional (2D) crystal with a large surface to volume ratio maximizes the effect of sensing [6, 15]. Any detection method is to accomplish a level of sensitivity that individual quanta of a measured entity can be resolved. In the case of chemical sensors, the quantum is one atom or molecule. Such resolution is beyond the reach of any detection technique, including solid-state gas sensors. Nano sensors that is made from graphene is competent of detecting individual events when a gas molecule binds or unbinds from the graphene's surface. The changes in current conductivity is due to the bind molecules that change the conductance in graphene one by one with electron. The achieved sensitivity is due to the fact that graphene is an exceptionally low-noise material electronically, which makes it a promising candidate for chemical sensor [8].

Though the potential of graphene as gas sensor has been demonstrated, additional modification still needs to be done in order to fully utilize this material. Graphene is known as zero band gap [16]. Therefore, graphene sensing properties are currently significantly lower than CNT and this has been shown to be directly linked to the low sensitivity [2]. Despite the fact, GNR has a finite band gap depending on the types of edge. There are two types of GNR which are called armchair GNR (AGNR) and zigzag GNR (ZGNR) as illustrated in Figure 1.1. AGNR has zigzag cross section at the edges of width, while ZGNR has armchair cross section. There are three types of dimer lines for AGNR know as  $3m$ ,  $3m+1$  and  $3m+2$ , where  $m$  is a positive integer. Earlier theoretical studies based on tight-binding (TB) approximation [17], found that GNRs are a relatively new class of nanomaterials that can behave as metallic or semiconducting character. ZGNR are all known as metallic regardless of



their widths. Meanwhile, for AGNR  $3m+2$  is metallic otherwise it is semiconducting [17, 18]. Thus, it is expected that gas molecule binding will have a higher effect on modifying the electronic properties of GNRs than graphene. They are currently being investigated to outstanding electrical, mechanical, thermal, optical, and quantum-mechanical properties [19].

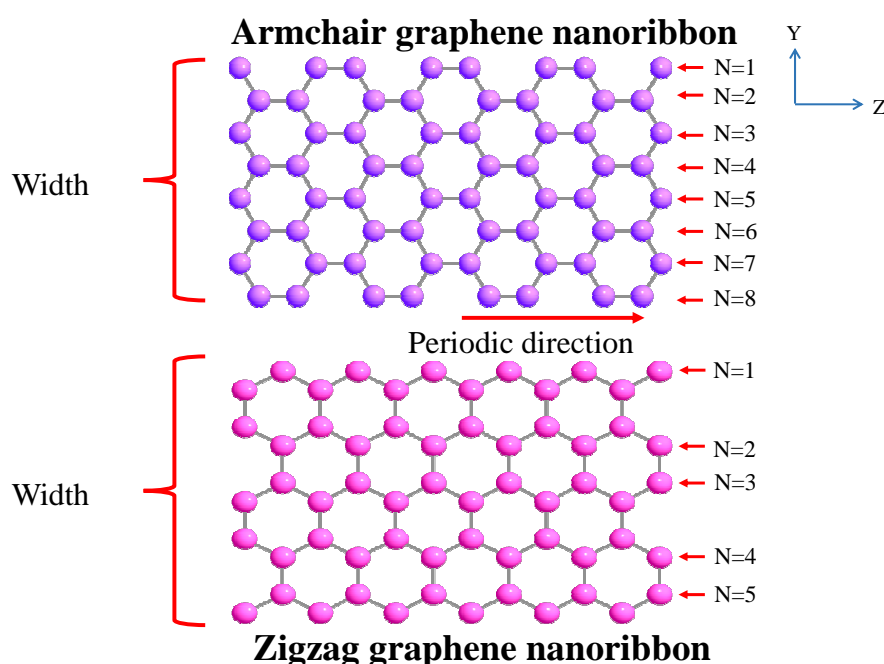


Figure 1.1: A illustration of GNR with AGNR (top) and ZGNR (bottom)

Recently, another promising approach has been introduced by using mechanical strain that has a substantial effect on the electronic band structure of AGNR [20, 21]. Due to the low reactivity of graphene, it requires harsh chemical treatment to allow functionalization to occur. Using conventional mechanical stress applied under uniaxial strain to the AGNR, the lattice becomes distorted and Dirac points relative to the allowed wavevector line in k-lines leading to the modulation of the electronic properties of AGNR [22]. Moreover, if the carbons are located in plane of graphene are generally expected to be relatively chemically inert due to the conjugation. Therefore, another technique such as non-planar deformation of graphene is initiated. This deformation includes scrolling, twisting, warping and buckling [23, 24]. This kind of deformation changes the AGNR to cylindrically geometry and

bond stretch is introduced by its own motion [25]. Although AGNR with various architectures and geometries have largely been successful, additional work is needed to address the deformation effects like warping on AGNR, which has been shown in Figure 1.2.

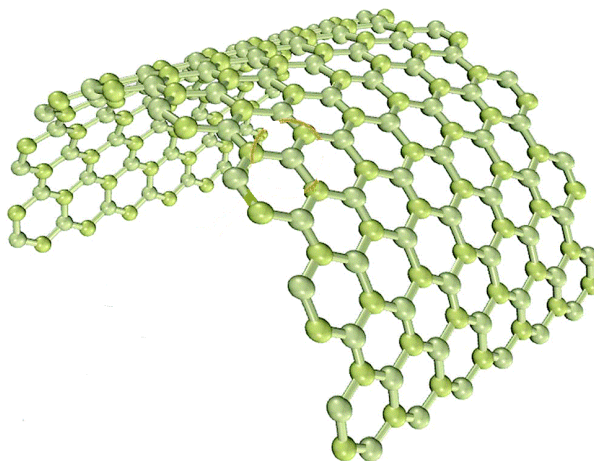


Figure 1.2: A Schematic Representation of Warping Graphene

The interesting combination of graphene and CNT behaviour in warping can give a rise to peculiar electronic properties [23]. This technique has been predicted to enhance the chemical reactivity as the carbon atoms are residing on the highly curved surface due to the diminished electronic delocalization [26]. It predicts that higher internal strain energy results in the formation of curvature in the device. In addition, warping deformation has more stable energy than planar structure and an extremely large surface to volume ratio, which favours the binding of gases on the structure and can probably further improve a sensor performance in sensitivity because the interaction between the gas molecule and material is higher [27]. This in turn can greatly manipulate its chemical reactivity allowing the tuning of sensing properties as needed. The effect of chemical functionalization on the mechanical properties of GNR still remain unexplored, although it is especially important for gas sensor application. It is a great interest to see the performance of warping AGNR as a gas sensor. The enhancement in gas sensing performance using warping is promising and its demand further studies. Therefore, it is important for future studies to understand the sensing performance of warping AGNR. It is worth studying since warping is a combination of graphene and CNT, which have originated from the same carbon material.

## 1.2 Problem Statement

Significant efforts have been made in using conventional material in gas sensor devices. Low selectivity, less sensitivity and high power consumption was observed in conventional material. Therefore, advanced carbon material is introduced such as CNT and graphene. A number of studies have demonstrated that low power consumption and high selectivity can be observed using this carbon material [5]. However, the sensitivity of graphene is less significant as compared to the sensitivity of CNT. Therefore, it has limited graphene application in sensing application.

The feasibility of using graphene deformation in the enhancement of graphene sensing behaviour is promising and demanding for further study as it is still in its early stage as compared to dopant and defect [12, 28, 29]. Although introducing dopant and defect improves the sensing performance, this technique deteriorates the crystallization and intrinsic of the graphene. There are many ways to deform graphene where one of the method is by warping [24, 30]. Most theoretical work on deformation graphene focuses on the electronic properties such as band structure, density of states, and thermal conductivity behaviour [30, 31]. There have been a few reports on the sensing capabilities such as binding energy, charge transfer and sensitivity. It is necessary to determine the binding energy because it determines how strongly the molecules bind with the material which later its affect charge transfer and sensitivity.

Motivated by these, this study aims to extend the idea to study the deformed AGNR on sensing properties through warping. As the warping will indirectly stretch the layers to form curvature which is almost similar in CNT. It is believed that the presence of curvature and bond stretch on the graphene layer will enhance its chemical reactivity. Therefore, the merit of using warping technique to form curvature and stretch on AGNR by varying warping angle is studied and is aimed to enhance the sensing properties. These fundamental properties deserve a closer look to fully exploit deformed graphene in electronic devices. Thus, this will be looked upon by means of computational study in this study.

## 1.3 Research Objective

This study focuses on the analysis of sensing behaviour of warping AGNR by computational study using Atomistic ToolKit (ATK) version of 2015.1. The objectives

are as follow:

1. To form deformed AGNR through warping method and to determine the sensing properties in term of binding energy, charge transfer and sensitivity.
2. To analyze and evaluate the sensing performance of inward and upward warping AGNR by the combination of curvature and strain effect
3. To investigate the gate biasing effect at highest warping angle between inward and upward warped AGNR.

#### **1.4 Research Scope**

1. This study will be conducted using ATK software version 2015.1 with three AGNR configurations which are  $3m$ ,  $3m+1$  and  $3m+2$ .
2. Properties evaluation include:
  - (a) Sensing properties :
    - i. Binding Energy, to obtain the binding capacity between analyte and substrate.
    - ii. Charge Transfer, to determine whether the analyte acts as donor or acceptor.
    - iii. Sensitivity, to measure how fast the material detect molecules.
  - (b) Electronic properties :
    - i. Band Structure, to obtain the energy gap of device either it behaves as semiconducting or metallic behavior even after warping is applied.
    - ii. Density of State (DOS), to determine whether the characteristic of unwarped device is sustained.
    - iii. Total Energy, to be used to calculate binding energy.
  - (c) Transport properties :
    - i. Current-voltage (I-V) characteristic, to observe the current conductivity is increase or decrease after warping applied and to use to obtain sensitivity.
    - ii. Transmission Spectrum (TS), to observe the pathway of electron flow.

3. AGNR deformation are limited to two types of warping which are inward and upward warping. While the warping angle of  $180^\circ$ ,  $270^\circ$  and  $360^\circ$  are used where the curvature increase with the warping angle.
4. Two types of gas molecules are chosen such as
  - (a) Ammonia ( $\text{NH}_3$ ): Ammonia is a toxic gas. Therefore, ammonia sensor is needed to prevent early possible leakage especially in industry.
  - (b) Oxygen ( $\text{O}_2$ ): Oxygen sensors are used in oxygen analysers which find a lot of use in medical applications such as anaesthesia monitors, respirators and oxygen concentration.

## 1.5 Research Contribution

The outcome of this study has two major contributions. The first contribution is the study of deformed AGNR through warping which are inward and upward at higher warping angle. This study investigate on three types of sensing properties which are binding energy, charge transfer and sensitivity. The second contribution is through simulation. It found that the sensing performance was enhanced in term of binding energy and sensitivity especially using  $3m+1$  AGNR configuration. The binding energy was improved by more than two fold while the sensitivity was enhanced by ten fold. Besides that, this study has been performed using computational simulation, which could help other experimental researcher to uncover critical areas in the fabrication process as well as optimize their time on designing.

## 1.6 Thesis Organization

This thesis begins with the introduction of gas sensor using advanced materials and background of the research study as outline in Chapter 1. Review study of gas sensing properties and deformation of GNR's electronic and mechanical properties which provides the foundation of the thesis has been presented in Chapter 2. In Chapter 3, computational details of the study have been presented. In addition details of warping behaviour, electronic and transport properties of warping AGNR have been also elaborated. The results achieved are shown in Chapter 4. Finally, a summary of this study with few recommendations for future study are presented in Chapter 5.

## REFERENCES

1. Capone, S., Forleo, A., Francioso, L., Rella, R., Siciliano, P., Spadavecchia, J., Presicce, D. S. and Taurino, A. M. Solid State Gas Sensors: State of the Art and Future Activities. *ChemInform*, 2004. 35(29). ISSN 0931-7597.
2. Basu, S. and Bhattacharyya, P. Recent developments on graphene and graphene oxide based solid state gas sensors. *Sensors and Actuators, B: Chemical*, 2012. 173: 1–21. ISSN 09254005.
3. Grattan, K. T. V. Physics, Chemistry & Technology of Solid State Gas Sensor Devices. *Journal of Modern Optics*, 1995. 42(7): 1553.
4. Llobet, E. Gas sensors using carbon nanomaterials: A review. *Sensors and Actuators, B: Chemical*, 2013. 179: 32–45. ISSN 09254005.
5. Varghese, S. S., Lonkar, S., Singh, K. K., Swaminathan, S. and Abdala, A. Recent advances in graphene based gas sensors. *Sensors and Actuators, B: Chemical*, 2015. 218(MAY): 160–183. ISSN 09254005.
6. Sharma, S. and Madou, M. A new approach to gas sensing with nanotechnology. *Philosophical transactions. Series A, Mathematical, physical, and engineering sciences*, 2012. 370(1967): 2448–73. ISSN 1364-503X.
7. Patra, M. K., Manzoor, K., Manoth, M., Negi, S. C., Vadera, S. R. and Kumar, N. Nanotechnology Applications for Chemical and Biological Sensors. *Science*, 2008. 58(5): 636–649. ISSN 0011748X.
8. Liu, J. and Li, G. A remote sensor for detecting methane based on palladium-decorated single walled carbon nanotubes. *Sensors (Basel, Switzerland)*, 2013. 13(7): 8814–8826. ISSN 14248220.
9. Collins, P. G., Bradley, K., Ishigami, M. and Zettl, A. Extreme Oxygen Sensitivity of Electronic Properties of Carbon Nanotubes. *Science*, 2000. 287(2000): 1801–1805. ISSN 00368075.
10. Schedin, F., Geim, A., Morozov, S., Hill, E., Blake, P., Katsnelson, M. and Novoselov, K. Detection of individual gas molecules adsorbed on graphene. *Nature Materials*, 2007. 6(9): 652–655. ISSN 1476-1122.

11. Zhang, Y.-H., Chen, Y.-B., Zhou, K.-G., Liu, C.-H., Zeng, J., Zhang, H.-L. and Peng, Y. Improving gas sensing properties of graphene by introducing dopants and defects: a first-principles study. *Nanotechnology*, 2009. 20(18): 185504. ISSN 1361-6528.
12. Jamal, G. R. A., Chowdhury, M. M., Rahman, F., Rahman, M. A., Shabnaz, S. and Habiba, U. Simulation of Graphene Nanoribbon Based Gas Sensor. 2015. 1(2): 66–73.
13. Imani, K., Jafari, G. and Abolhasani, M. R. Electronic Structure Calculation of Adsorbate Gas Molecules on an Armchair Graphene Nanoribbon. *ISRN Condensed Matter Physics*, 2012. 2012.
14. Iijima, S. and Ichihashi, T. Single-shell carbon nanotubes of 1-nm diameter. *Nature*, 1993. 363(6430): 603–605.
15. Liu, X., Cheng, S., Liu, H., Hu, S., Zhang, D., Ning, H. and Engineering, I. A Survey on Gas Sensing Technology. 2012: 9635–9665.
16. Ando, T. The electronic properties of graphene and carbon nanotubes. *October*, 2009. 1(October 2005): 17–21. ISSN 1884-4049. doi:10.1038/10.1038/asiamat.2009.1.
17. Brey, L. and Fertig, H. A. Edge states and the quantized Hall effect in graphene. *Phys. Rev. B*, 2006. 73(19): 195408.
18. Barone, V., Hod, O. and Scuseria, G. E. Electronic Structure and Stability of Semiconducting Graphene Nanoribbons. *Nano Letters*, 2006. 6(12): 2748–2754.
19. Geim, A. K. and Novoselov, K. The rise of graphene. *Nature Mater.*, 2007. 6(3): 183–191. ISSN 1476-1122.
20. Rasuli, R., Rafii-Tabar, H. and Zad, A. I. Strain effect on quantum conductance of graphene nanoribbons from maximally localized Wannier functions. *Phys. Rev. B*, 2010. 81(12): 125409.
21. Lu, Y. and Guo, J. Band gap of strained graphene nanoribbons. *Nano Research*, 2010. 3(3): 189–199. ISSN 19980124.
22. Huang, M., Yan, H., Heinz, T. F. and Hone, J. Probing Strain-Induced Electronic Structure Change in Graphene by Raman Spectroscopy. *Nano Letters*, 2010. 10(10): 4074–4079.
23. Ivanovskaya, V. V., Wagner, P., Zobelli, A., Suarez-Martinez, I., Yaya, A. and Ewels, C. P. Graphene edge structures: Folding, scrolling, tubing, rippling and twisting. 2012: 1–6.

24. Shenoy, V. B., Reddy, C. D., Ramasubramaniam, A. and Zhang, Y. W. Edge-stress-induced warping of graphene sheets and nanoribbons. *Physical Review Letters*, 2008. 101(24). ISSN 00319007.
25. Glukhova, O. and Slepchenkov, M. Influence of the curvature of deformed graphene nanoribbons on their electronic and adsorptive properties: theoretical investigation based on the analysis of the local stress field for an atomic grid. *Nanoscale*, 2012. 4(11): 3335. ISSN 2040-3364.
26. Wu, Q., Wu, Y., Hao, Y., Geng, J., Charlton, M., Chen, S., Ren, Y., Ji, H., Li, H., Boukhvalov, D. W., Piner, R. D., Bielawski, C. W. and Ruoff, R. S. Selective surface functionalization at regions of high local curvature in graphene. *Chemical communications (Cambridge, England)*, 2013. 49(7): 677–9. ISSN 1364-548X.
27. Scuseria, G. E. Electromechanical Properties of Suspended Graphene Nanoribbons. 2009: 1–4.
28. Zhang, Y.-H., Zhou, K.-G., Xie, K.-F., Gou, X.-C., Zeng, J., Zhang, H.-L. and Peng, Y. Effects of Stone-Wales defect on the interactions between NH<sub>3</sub>, NO<sub>2</sub> and graphene. *Journal of nanoscience and nanotechnology*, 2010. 10: 7347–7350. ISSN 15334880.
29. Robinson, J. A., Snow, E. S., Badescu, S. C., Reinecke, T. L. and Perkins, F. K. Role of Defects in Single-Walled Carbon Nanotube Chemical Sensors. *Nano Letters*, 2006. 6(8): 1747–1751.
30. Tong, G.-p. Electronic Properties of Deformed Graphene Nanoribbons. 2013.
31. Yu, C. and Zhang, G. Impacts of length and geometry deformation on thermal conductivity of graphene nanoribbons. *Journal of Applied Physics*, 2013. 113(4). ISSN 00218979.
32. Korotcenkov, G. Metal oxides for solid-state gas sensors: What determines our choice? *Materials Science and Engineering B: Solid-State Materials for Advanced Technology*, 2007. 139(1): 1–23. ISSN 09215107.
33. BATZILL, M. and DIEBOLD, U. The surface and materials science of tin oxide. *Progress in Surface Science*, 2005. 79(2-4): 47–154. ISSN 00796816.
34. Wisitsoraat, A., Tuantranont, A. and Wlodarski, W. Gas sensing properties of CNT-SnO<sub>2</sub> Nanocomposite Thin Film Prepared by E-beam Evaporation. *Sensors And Actuators*, 2007: 550–553.
35. Lee, D.-D., Chung, W.-Y., Choi, M.-S. and Baek, J.-M. Low-power micro gas sensor. *Sensors and Actuators B: Chemical*, 1996. 33(1-3): 147–150.



ISSN 09254005.

36. Jaegle, M., Wöllenstein, J., Meisinger, T., Böttner, H., Müller, G., Becker, T. and Bosch-v.Braunmühl, C. Micromachined thin film SnO<sub>2</sub> gas sensors in temperature-pulsed operation mode. *Sensors and Actuators B: Chemical*, 1999. 57(1-3): 130–134. ISSN 09254005.
37. Jimenez-Cadena, G., Riu, J. and Rius, F. X. Gas sensors based on nanostructured materials. *Analyst*, 2007. 132(11): 1083–1099.
38. Iijima, S. Helical microtubules of graphitic carbon. *Nature*, 1991. 354(6348): 56–58. ISSN 0028-0836.
39. Thai, T. T., Yang, L., Dejean, G. R. and Tentzeris, M. M. Nanotechnology enables wireless gas sensing, 2011. doi:10.1109/MMM.2011.940594.
40. Majumdar, S., Nag, P. and Devi, P. S. Enhanced performance of CNT / SnO<sub>2</sub> thick film gas sensors towards hydrogen. 2014. 147: 79–85.
41. Zhao, J., Buldum, A., Han, J. and Lu, J. P. Gas molecule adsorption in carbon nanotubes and nanotube bundles. *Nanotechnology*, 2002. 13(2): 195–200. ISSN 0957-4484.
42. Novoselov, K. S. Electric Field Effect in Atomically Thin Carbon Films. 2014. 666(2004).
43. Mark A. Bissett, M. T. and Ago, H. Strain Engineering the Properties of Graphene and other Two-dimensional Crystals, 2014.
44. Bonaccorso, F., Sun, Z., Hasan, T. and Ferrari, A. C. Graphene Photonics and Optoelectronics. *Nature Photonics*, 2010. 4(9): 611–622. ISSN 1749-4885.
45. Wehling, T. O., Novoselov, K. S., Morozov, S. V., Vdovin, E. E., Katsnelson, M. I., Geim, A. K. and Lichtenstein, A. I. Molecular doping of graphene. *Nano Letters*, 2008. 8(1): 173–177. ISSN 15306984.
46. Novoselov, K. S., Jiang, D., Schedin, F., Booth, T. J., Khotkevich, V. V., Morozov, S. V. and Geim, A. K. Two-dimensional atomic crystals. *Proceedings of the National Academy of Sciences*, 2005. 102(30): 10451–10453. ISSN 0027-8424.
47. Leenaerts, O., Partoens, B. and Peeters, F. M. Adsorption of H<sub>2</sub>O, NH<sub>3</sub>, CO, NO<sub>2</sub>, and NO on graphene: A first-principles study. *Adsorption Journal Of The International Adsorption Society*, 2008. 77(12): 125416. ISSN 1098-0121.
48. Kong, L., Enders, A., Rahman, T. S. and Dowben, P. A. Molecular adsorption on graphene. *Journal of Physics: Condensed Matter*, 2014. 26(44): 443001.

ISSN 0953-8984.

49. Shao, L., Chen, G., Ye, H., Wu, Y., Qiao, Z., Zhu, Y. and Niu, H. Sulfur dioxide adsorbed on graphene and heteroatom-doped graphene: A first-principles study. *European Physical Journal B*, 2013. 86(2): 54. ISSN 14346028.
50. Dan, Y., Lu, Y., Kybert, N. J., Luo, Z. and Johnson, A. T. C. Intrinsic response of graphene vapor sensors. *Nano Letters*, 2009. 9(4): 1472–1475. ISSN 15306984.
51. Kumar, S., Kaushik, S., Pratap, R. and Raghavan, S. Graphene on paper: A simple, low-cost chemical sensing platform. *ACS Applied Materials and Interfaces*, 2015. 7(4): 2189–2194. ISSN 19448252.
52. Wang, W. and Zhao, G. First principles study of the electronic energy bands and state density of lithium-doped narrow armchair graphene nanoribbons. *Solid State Communications*, 2013. 166: 6–11. ISSN 00381098.
53. Son, Y. W., Cohen, M. L. and Louie, S. G. Energy gaps in graphene nanoribbons. *Physical Review Letters*, 2006. 97(21): 1–4. ISSN 00319007.
54. Huang, B., Li, Z., Liu, Z., Zhou, G., Hao, S., Wu, J., Gu, B. L. and Duan, W. Adsorption of gas molecules on graphene nanoribbons and its implication for nanoscale molecule sensor. *Journal of Physical Chemistry C*, 2008. 112(35): 13442–13446. ISSN 19327447.
55. Misra, A. Carbon nanotubes and graphene-based chemical sensors. *Current Science*, 2014. 107(3): 419–429. ISSN 00113891.
56. Mahmoudi, A., Troudi, M., Bergaoui, Y., Bondavalli, P. and Sghaier, N. Analysis of Simulated Output Characteristics of Gas Sensor Based on Graphene Nanoribbon. 2016. 2016.
57. Paulla, K. K., Hassan, A. J., Knick, C. R. and Farajian, A. A. Ab initio assessment of graphene nanoribbons reactivity for molecule adsorption and conductance modulation: nitrogen dioxide nanosensor. *RSC Advances*, 2014. 4(5): 2346–2354. ISSN 2046-2069.
58. Asai, M., Ohba, T., Iwanaga, T., Kanoh, H., Endo, M., Campos-Delgado, J., Terrones, M., Nakai, K. and Kaneko, K. Marked adsorption irreversibility of graphitic nanoribbons for CO<sub>2</sub> and H<sub>2</sub>O. *Journal of the American Chemical Society*, 2011. 133(38): 14880–14883. ISSN 00027863.
59. Travlou, N. A., Seredych, M., Rodríguez-Castellón, E. and Bandoz, T. J. Activated carbon-based gas sensors: effects of surface features on the sensing

- mechanism. *J. Mater. Chem. A*, 2015. 3(7): 3821–3831. ISSN 2050-7488.
60. Srivastava, P., Jaiswal, N. K. and Sharma, V. First-principles investigation of armchair boron nitride nanoribbons for sensing PH<sub>3</sub> gas molecules. *Superlattices and Microstructures*, 2014. 73: 350–358. ISSN 10963677.
61. Kliros, G. S. Effect of uniaxial strain on the current-voltage characteristics of graphene nanoribbon field-effect transistors. *Proceedings of the International Semiconductor Conference, CAS*, 2013. 1: 27–30. ISSN 1545-827X.
62. Chang, S.-L., Wu, B.-R., Yang, P.-H. and Lin, M.-F. Curvature effects on electronic properties of armchair graphene nanoribbons without passivation. *Physical Chemistry Chemical Physics*, 2012. 14(47): 16409. ISSN 1463-9076.
63. Frank, I. W., Tanenbaum, D. M., van der Zande, A. M. and McEuen, P. L. Mechanical properties of suspended graphene sheets. *Journal of Vacuum Science & Technology B: Microelectronics and Nanometer Structures*, 2007. 25(6): 2558. ISSN 10711023.
64. Shenoy, V. B., Reddy, C. D. and Zhang, Y. W. Spontaneous curling of graphene sheets with reconstructed edges. *ACS Nano*, 2010. 4(8): 4840–4844. ISSN 19360851.
65. Kosynkin, D. V., Higginbotham, A. L., Sinitskii, A., Lomeda, J. R., Dimiev, A., Price, B. K. and Tour, J. M. Longitudinal unzipping of carbon nanotubes to form graphene nanoribbons. *Nature*, 2009. 458(7240): 872–876. ISSN 0028-0836.
66. Jiao, L., Zhang, L., Wang, X., Diankov, G. and Dai, H. Narrow graphene nanoribbons from carbon nanotubes. *Nature*, 2009. 458(7240): 877–880. ISSN 0028-0836.
67. Elias, A. L., Botello-Mendez, A. R., Meneses-Rodriguez, D., Gonzalez, V. J., Ramirez-Gonzalez, D., Ci, L., Munoz-Sandoval, E., Ajayan, P. M., Terrones, H. and Terrones, M. Longitudinal cutting of pure and doped carbon nanotubes to form graphitic nanoribbons using metal clusters as nanoscalpels. *Nano Letters*, 2010. 10(2): 366–372. ISSN 15306984.
68. Haddon, R. C. Chemistry of the fullerenes: the manifestation of strain in a class of continuous aromatic molecules. *Science (New York, N.Y.)*, 1993. 261(4): 1545–1550. ISSN 0036-8075.
69. www.quantumwise.com. URL <http://quantumwise.com/>.
70. Stokbro, K., Petersen, D., Smidstrup, S., Blom, A., Ipsen, M. and Kaasbjerg,

- K. Semiempirical model for nanoscale device simulations. *Phys. Rev. B*, 2010. 82(7): 75420.
71. Raza, H. and Kan, E. C. An extended Huckel theory based atomistic model for graphene nanoelectronics. *Journal of Computational Electronics*, 2008. 7: 372–375. ISSN 15698025.
72. Zou, Y., Long, M., Li, M., Zhang, X., Zhang, Q. and Xu, H. Control of electronic transport in nanohole defective zigzag graphene nanoribbon by means of side alkene chain. *RSC Advances*, 2015. 5(25): 19152–19158. ISSN 2046-2069.
73. Gharbi, M., Sun, Z. H., Sharma, P. and White, K. The origins of electromechanical indentation size effect in ferroelectrics. *Applied Physics Letters*, 2009. 95(14). ISSN 00036951.
74. Xu, R., Wang, Y., Liu, B. and Fang, D. Mechanics Interpretation on the Bending Stiffness and Wrinkled Pattern of Graphene. *Journal of Applied Mechanics*, 2013. 80(4): 040910. ISSN 0021-8936.
75. Zhang, J., Ong, K. P. and Wu, P. The influence of Out-of-Plane deformation on the band Gap of graphene nanoribbons. *Journal of Physical Chemistry C*, 2010. 114(29): 12749–12753. ISSN 19327447.
76. Cartamil-Bueno, S. and S.Rodriguez-Bolivar. I-V characteristics of in-plane and out-of-plane strained edge-hydrogenated armchair graphene nanoribbons. *Journal of Applied Physics*, 2015. 117(24). ISSN 10897550.
77. Barone, V., Hod, O., Peralta, J. E. and Scuseria, G. E. Accurate prediction of the electronic properties of low-dimensional graphene derivatives using a screened hybrid density functional. *Accounts of Chemical Research*, 2011. 44(4): 269–279. ISSN 00014842.
78. Li, Y., Jiang, X., Liu, Z. and Liu, Z. Strain effects in graphene and graphene nanoribbons: The underlying mechanism. *Nano Research*, 2010. 3(8): 545–556. ISSN 19980124.
79. Castro Neto, A. H., Guinea, F., Peres, N. M. R., Novoselov, K. S. and Geim, A. K. The electronic properties of graphene. *Reviews of Modern Physics*, 2009. 81(1): 109–162. ISSN 00346861.
80. Hou, Z. and Yee, M. Electronic and transport properties of graphene nanoribbons. *2007 7th IEEE Conference on Nanotechnology (IEEE NANO)*. 2007. ISBN 1944-9399 VO -. 554–557.
81. Xu, Y., Gao, H., Li, M., Guo, Z., Chen, H., Jin, Z. and Yu, B. Electronic

- transport in monolayer graphene with extreme physical deformation: *ab initio* density functional calculation. *Nanotechnology*, 2011. 22(36): 365202. ISSN 0957-4484.
82. Chen, C. W., Hung, S. C., Yang, M. D., Yeh, C. W., Wu, C. H., Chi, G. C., Ren, F. and Pearton, S. J. Oxygen sensors made by monolayer graphene under room temperature. *Applied Physics Letters*, 2011. 99(24): 2012–2015. ISSN 00036951.
  83. Soldano, C., Talapatra, S. and Kar, S. *Carbon Nanotubes and Graphene Nanoribbons: Potentials for Nanoscale Electrical Interconnects*. vol. 2. 2013. ISBN 3905163985.
  84. Topsakal, M., Bagci, V. M. K. and Ciraci, S. Current-voltage (I-V) characteristics of armchair graphene nanoribbons under uniaxial strain. *Physical Review B - Condensed Matter and Materials Physics*, 2010. 81(20): 1–6. ISSN 10980121.
  85. Nguyen, C. V., Ilyasov, V. V. and Nguyen, H. N. Tuning the electronic properties of armchair graphene nanoribbons by strain engineering. *Physica Scripta*, 2015. 90(1): 015802. ISSN 0031-8949.
  86. Haddon, R. C. Hybridization and the orientation and alignment of  $\pi$ -orbitals in nonplanar conjugated organic molecules:  $\pi$ -orbital axis vector analysis (POAV2). *Journal of the American Chemical Society*, 1986. (12): 2837–2842. ISSN 0002-7863.
  87. De Andres, P. L. and Vergs, J. A. First-principles calculation of the effect of stress on the chemical activity of graphene. *Applied Physics Letters*, 2008. 93(17): 1–4. ISSN 00036951.
  88. Dai, J., Yuan, J. and Giannozzi, P. Gas adsorption on graphene doped with B, N, Al, and S: A theoretical study. *Applied Physics Letters*, 2009. 95(23): 10–13. ISSN 00036951.
  89. Orlov, A. V. and Ovid'ko, I. A. Mechanical properties of graphene nanoribbons: A selective review of computer simulations. *Reviews on Advanced Materials Science*, 2015. 40: 249–256.
  90. Perlstein, J. The Weak Hydrogen Bond In Structural Chemistry and Biology. *Journal of the American Chemical Society*, 2000. 123(1): 191–192. ISSN 0002-7863.
  91. Silva-Tapia, A. B., García-Carmona, X. and Radovic, L. R. Similarities and differences in O<sub>2</sub> chemisorption on graphene nanoribbon vs. carbon nanotube. *Carbon*, 2012. 50(3): 1152–1162. ISSN 00086223.

92. Ryu, S., Liu, L., Berciaud, S., Yu, Y. J., Liu, H., Kim, P., Flynn, G. W. and Brus, L. E. Atmospheric oxygen binding and hole doping in deformed graphene on a SiO<sub>2</sub> substrate. *Nano Letters*, 2010. 10(12): 4944–4951. ISSN 15306984.
93. Cusano, A., Consales, M., Crescitelli, A., Penza, M., Aversa, P., Veneri, P. D. and Giordano, M. Charge transfer effects on the sensing properties of fiber optic chemical nano-sensors based on single-walled carbon nanotubes. *Carbon*, 2009. 47(3): 782–788. ISSN 00086223.
94. Huang, P., Zhu, H., Jing, L., Zhao, Y. and Gao, X. Graphene covalently binding aryl groups: Conductivity increases rather than decreases. *ACS Nano*, 2011. 5(10): 7945–7949. ISSN 19360851.
95. Panchakarla, L. S., Subrahmanyam, K. S., Saha, S. K., Govindaraj, A., Krishnamurthy, H. R., Waghmare, U. V. and Rao, C. N. R. Synthesis, Structure and Properties of Boron and Nitrogen Doped Graphene. *Advanced Materials*, 2009. 21(46): 12.
96. Wei, D., Liu, Y., Wang, Y., Zhang, H., Huang, L. and Yu, G. Synthesis of n-doped graphene by chemical vapor deposition and its electrical properties. *Nano Letters*, 2009. 9(5): 1752–1758. ISSN 15306984.
97. McGuire, K., Gothard, N., Gai, P. L., Dresselhaus, M. S., Sumanasekera, G. and Rao, A. M. Synthesis and Raman characterization of boron-doped single-walled carbon nanotubes. *Carbon*, 2005. 43(2): 219–227. ISSN 00086223.
98. Denis, P. A., Faccio, R. and Momburu, A. W. Is it possible to dope single-walled carbon nanotubes and graphene with sulfur? *ChemPhysChem*, 2009. 10(4): 715–722. ISSN 14394235.
99. Romero, H. E., Shen, N., Joshi, P., Gutierrez, H. R., Tadigadapa, S. A., Sofo, J. O. and Eklund, P. C. n-type behavior of graphene supported on Si/SiO<sub>2</sub> substrates. *ACS Nano*, 2008. 2(10): 2037–2044. ISSN 19360851.
100. Subrahmanyam, K. S., Kumar, P., Maitra, U., Govindaraj, A., Hembram, K. P. S. S., Waghmare, U. V. and Rao, C. N. R. Chemical storage of hydrogen in few-layer graphene. *Proceedings of the National Academy of Sciences*, 2011. 108(7): 2674–2677. ISSN 0027-8424.
101. Cho, B., Hahm, M. G., Choi, M., Yoon, J., Kim, A. R., Lee, Y.-J., Park, S.-G., Kwon, J.-D., Kim, C. S., Song, M., Jeong, Y., Nam, K.-S., Lee, S., Yoo, T. J., Kang, C. G., Lee, B. H., Ko, H. C., Ajayan, P. M. and Kim, D.-H. Charge-transfer-based Gas Sensing Using Atomic-layer MoS<sub>2</sub>. *Scientific Reports*, 2015. 5(1): 8052. ISSN 2045-2322.

102. Zhou, M., Lu, Y.-H., Cai, Y.-Q., Zhang, C. and Feng, Y.-P. Adsorption of gas molecules on transition metal embedded graphene: a search for high-performance graphene-based catalysts and gas sensors. *Nanotechnology*, 2011. 22(38): 385502. ISSN 0957-4484.
103. Schierbaum, K. D., Weimar, U., Göpel, W. and Kowalkowski, R. Conductance, work function and catalytic activity of SnO<sub>2</sub>-based gas sensors. *Sensors and Actuators: B. Chemical*, 1991. 3(3): 205–214. ISSN 09254005.
104. Mortazavi Zanjani, S. M., Sadeghi, M. M., Holt, M., Chowdhury, S. F., Tao, L. and Akinwande, D. Enhanced sensitivity of graphene ammonia gas sensors using molecular doping. *Applied Physics Letters*, 2016. 108(3): 1–6. ISSN 00036951.
105. Lau, C. N., Bao, W. and Velasco, J. Properties of suspended graphene membranes. *Materials Today*, 2012. 15(6): 238–245. ISSN 13697021.