

PRODUCTION AND CHARACTERIZATION OF PROTEASE FROM
HALOPHILIC *VIRGIBACILLUS* SPECIES CD6

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Specially dedicated to *my beloved family, future life partner, soulmates and friends*

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ABSTRACT

In enzyme production industries, the major challenges that hinder the efficient and economic commercial scale application of proteases are their stability in broad range of pH, temperature, salinity, as well as their optimal activity in the presence of metal ions, organic solvents and detergents. Moreover, the enzyme purification steps also contribute to the cost of production. To overcome this problem, characterization and production of crude protease with attractive properties from wild bacterial isolate could be an alternative as it is a more cost-effective way compared to production of protease that involves purification steps and protein engineering approach. Therefore, crude protease of *Virgibacillus* sp. CD6 isolated from salted-fish was characterized in this study using azocasein assay and bioinformatics tools. Protease production was found to be highest when using soybean meal and yeast extract as nitrogen source compared to other organic nitrogen sources. The protease exhibited vast range of stability with optimum activity at 10.0 % (w/v) NaCl, 60°C, pH 7 and 10, indicating its polyextremophilicity. The enzyme activity was enhanced by Mg^{2+} , Mn^{2+} , Cd^{2+} and Al^{3+} . Both PMSF and EDTA hindered protease activity, denoting the presence of serine protease and metalloprotease properties respectively. High protease stability (>80%) was demonstrated in presence of organic solvents and detergent constituents investigated, and surprisingly it is exceptionally compatible with commercial detergents. Phylogenetic analyses revealed that proteases of *Virgibacillus* sp. demonstrated far distance relationship with other species, which worth for further exploration. Attributes of this protease can actualize necessity of searching superlative enzymes from extremophiles for diverse applications, particularly in detergent industry.

ABSTRAK

Dalam industri penghasilan enzim, cabaran utama yang menghalang aplikasi komersial protease yang cekap dan ekonomi adalah ciri-ciri protease yang stabil dalam pelbagai pH, suhu, kadar garam serta aktiviti optimum dalam ion logam, pelarut organik, dan unsur detergen. Selain itu, proses penulenan enzim juga menyumbang kepada kos penghasilan. Bagi mengatasi masalah ini, pencirian dan penghasilan protease dari bakteria tanpa melibatkan proses penulenan boleh menjadi alternatif kerana ia adalah cara yang kos efektif berbanding dengan penghasilan protease yang melibatkan penulenan enzim dan kejuruteraan protein. Oleh itu, protease daripada *Virgibacillus* sp. CD6 yang dipencilkan daripada ikan masin telah dicirikan dalam kajian ini dengan penggunaan azocasein assay dan alat bioinformatik. Penghasilan protease didapati paling tinggi apabila menggunakan kacang soya dan ekstrak yis sebagai sumber nitrogen berbanding dengan sumber nitrogen organik yang lain. Protease tersebut mempamerkan luas kestabilan dengan aktiviti optimum pada 10.0% (w/v) NaCl, 60°C, pH 7 dan 10, menunjukkan ciri poli-ekstremofi. Aktiviti enzim telah dipertingkatkan oleh Mg^{2+} , Mn^{2+} , Cd^{2+} dan Al^{3+} . Kedua-dua PMSF dan EDTA didapati menghalang aktiviti protease, menandakan ciri protease serine dan metalloprotease masing-masing. Kestabilan protease yang tinggi (>80%) telah ditunjukkan dalam pelarut organik dan unsur detergen, serta amat serasi dengan bahan pencuci komersial. Analisis filogenetik menunjukkan bahawa protease daripada *Virgibacillus* sp. mempunyai hubungan yang jauh dengan spesies lain, bernilai untuk penerokaan selanjutnya. Sifat-sifat protease ini boleh merealisasi keperluan mencari enzim cemerlang dari eskstremofi untuk pelbagai aplikasi, terutamanya dalam industri detergen.

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LIST OF SYMBOLS

| | | |
|--------------------|---|--------------------------|
| A_{420} | - | Absorbance at 420 nm |
| A_{750} | - | Absorbance at 750 nm |
| α | - | Alpha |
| \approx | - | Approximately |
| β | - | Beta |
| $^{\circ}\text{C}$ | - | Degree celcius |
| D | - | Diameter |
| = | - | Equal |
| γ | - | Gamma |
| g | - | Gram |
| g/L | - | Gram per liter |
| > | - | Greater than |
| h | - | Hour |
| kPa | - | Kilo Pascal |
| < | - | Less than |
| L | - | Liter |
| \log_{10} | - | Logarithm to base 10 |
| mg/ml | - | Milligram per milliliter |
| μl | - | Microliter |
| mg | - | Milligram |
| mg/L | - | Milligram per liter |
| ml | - | Milliliter |
| mm | - | Millimeter |
| mM | - | Millimolar |
| M | - | Molar mass |
| nm | - | Nanometer |

| | | |
|-------------------|---|---------------------------|
| - | - | Negative |
| n | - | Number |
| OD ₆₀₀ | - | Optical density at 600 nm |
| / | - | Or |
| % | - | Percent |
| cm ⁻¹ | - | Per centimeter |
| M ⁻¹ | - | Per molar |
| π | - | Pi |
| ± | - | Plus-minus |
| + | - | Positive |
| ® | - | Registered trademark |
| ² | - | Square |
| × | - | Times |
| ™ | - | Trademark |
| U/mg | - | Units per milligram |
| U/ml | - | Units per volume |
| v/v | - | Volume per volume |
| w/v | - | Weight per volume |

LIST OF ABBREVIATIONS

| | | |
|---|---|---|
| A | - | Alanine |
| Al ³⁺ | - | Aluminum ion |
| Al ₂ (SO ₄) ₃ | - | Aluminum sulfate |
| APC | - | Activated protein C |
| ATP | - | Adenosine triphosphate |
| BLASTp | - | Protein-protein Basic Local Alignment Search Tool |
| BSA | - | Bovine serum albumin |
| C | - | Cysteine |
| C ₆ H ₅ Na ₃ O ₇ | - | Trisodium citrate |
| C ₆ H ₅ Na ₃ O ₇ ·2H ₂ O | - | Trisodium citrate dihydrate |
| Ca ²⁺ | - | Calcium ion |
| CaCl ₂ | - | Calcium chloride |
| Cd ²⁺ | - | Cadmium ion |
| Cd(NO ₃) ₂ | - | Cadmium nitrate |
| Cl ⁻ | - | Chloride ion |
| Co ²⁺ | - | Cobalt ion |
| CoCl ₂ | - | Cobalt chloride |
| Cu ²⁺ | - | Copper (II) ion |
| CuSO ₄ | - | Copper (II) sulfate |
| CuSO ₄ ·5H ₂ O | - | Copper (II) sulfate pentahydrate |
| D | - | Aspartic acid |
| DMSO | - | Dimethyl sulfoxide |
| DNA | - | Deoxyribonucleic Acid |
| DTT | - | Dithiothreitol |
| EC | - | Enzyme commission |
| EDTA | - | Ethylene Diamine Tetraacetic Acid |

| | | |
|--------------------------------------|---|---|
| <i>et al.</i> | - | And friends |
| F | - | Phenylalanine |
| FDA | - | Food and Drug Administration |
| Fe ³⁺ | - | Ferum (III) ion |
| FeCl ₃ | - | Ferum (III) chloride |
| G | - | Glycine |
| Glu, E | - | Glutamic acid |
| H-bond | - | Hydrogen bond |
| H ⁺ | - | Hydrogen ion |
| H ₂ O ₂ | - | Hydrogen peroxide |
| HCl | - | Hydrochloric acid |
| His, H | - | Histidine |
| I | - | Isoleucine |
| IAA | - | Iodoacetic acid |
| ID | - | Identifier |
| K | - | Lysine |
| K ⁺ | - | Potassium ion |
| K ₂ HPO ₄ | - | Dipotassium hydrogen phosphate |
| KCl | - | Potassium chloride |
| KH ₂ PO ₄ | - | Potassium dihydrogen phosphate |
| KNO ₃ | - | Potassium nitrate |
| L | - | Leucine |
| M | - | Methionine |
| MEGA 7.0 | - | Molecular Evolutionary Genetic Analysis version 7.0 |
| Mg ²⁺ | - | Magnesium ion |
| MgCl ₂ | - | Magnesium chloride |
| MgSO ₄ .7H ₂ O | - | Magnesium sulfate heptahydrate |
| Mn ²⁺ | - | Manganese ion |
| MnCl ₂ | - | Manganese chloride |
| N | - | Asparagine |
| Na ⁺ | - | Sodium ion |
| Na ₂ CO ₃ | - | Sodium carbonate |
| NaCl | - | Sodium chloride |

| | | |
|--------------------|---|---|
| NaHCO ₃ | - | Sodium bicarbonate |
| NaNO ₂ | - | Sodium nitrite |
| NaOH | - | Sodium hydroxide |
| NH ₄ Cl | - | Ammonium chloride |
| Ni ²⁺ | - | Nickel ion |
| NiSO ₄ | - | Nickel sulfate |
| OH ⁻ | - | Hydroxide ion |
| P | - | Proline |
| PHB | - | Polyhydroxybutyrate |
| PMSF | - | Phenylmethylsulfonyl fluoride |
| pI | - | Isoelectric point |
| PSI-BLAST | - | Position-Specific Iterated Basic Local Alignment Search Tool |
| Q | - | Glutamine |
| R | - | Arginine |
| rcf | - | Relative centrifugal force |
| rpm | - | Rotary per minute |
| rRNA | - | Ribosomal ribonucleic acid |
| SAPS | - | Statistical Analysis of Protein Sequences |
| SD | - | Standard deviation |
| SDS | - | Sodium dodecyl sulfate |
| SDS-PAGE | - | Sodium dodecyl sulfate polyacrylamide gel electrophoresis |
| Ser, S | - | Serine |
| sp. | - | Species (singular) |
| spp. | - | Species (plural) |
| T | - | Threonine |
| t-PA | - | Tissue plasminogen activator |
| TCA | - | Trichloroacetic acid |
| Tris | - | 2-Amino-2-(hydroxymethyl)propane-1,3-diol |
| u-PA | - | Urokinase type plasminogen activator |
| USA | - | United States of America |
| USD | - | United States dollar |

| | | |
|-------------------|---|--------------------|
| UV | - | Ultraviolet |
| V | - | Valine |
| W | - | Tryptophan |
| X, Xaa | - | Unknown amino acid |
| Y | - | Tyrosine |
| Zn ²⁺ | - | Zinc ion |
| ZnSO ₄ | - | Zinc sulfate |

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CHAPTER 1

INTRODUCTION

1.1 Background of study

Halophilic bacteria has been recognized as one of the extremophiles that has valuable applications in industry and environment (Oren, 2010; Edbeib *et al.*, 2016; Yin *et al.*, 2015). They are found in natural saline and hypersaline habitats such as seawater, salt marshes and lagoon. Occurrence of halophiles can be from seawater to brines (Brock, 1979), some habitats include Dead Sea between Israel and Jordan and also Great Salt Lake in Utah (Oren, 2006). Besides that, salty environments inhabited by halophilic and halotolerant bacteria include food products such as salted fish and fermented food (Enache *et al.*, 2012), and these type of foods are commonly found in Malaysia.

Well-adapted strategies in saline environments utilized by halophilic bacteria made them useful in industrial applications. These halophilic bacteria has been used for production of valuable metabolites and solutes such as stress protectants (DasSarma and DasSarma, 2006), saline wastewater treatments (Shivanand and Mugeraya, 2011) and biodegradation of organic pollutants in environmental biotechnology (Le Borgne *et al.*, 2008). Halophilic bacteria can be classified under different phyla. Under different phylum, halophilic bacteria have different physiological requirements such as compatible solute used and salt concentration required. This diversity makes the halophilic bacteria as one of the source of opportunity and abundance, including industrial enzymes.

One of the enzymes produced by halophilic bacteria is protease, which is a type of hydrolase. Protease can be produced from animal, plant and microbial source. Protease from microbial source has been extensively used in various application especially in detergent industry since 1960 (Rao *et al.*, 1998) due their effectiveness in removing protein stains (Karn and Kumar, 2015). Until today, proteases contributed approximately 60% of the global industrial enzymes market (Anithajothi *et al.*, 2014). While from this amount, microbial proteases constitute 40% of total enzyme production (Raval *et al.*, 2014) which applied in various industries. The largest market undeniable is detergent industry, as this industry contributed to production of 13.5 billion tons per year (Adrio and Demain, 2014).

Apart from that, use of eco-friendly protease recovered from industrial sludge for bio-conversion of proteinaceous waste material into value-added products has become an increasingly concern due to it is a cost effective process (Karn and Kumar, 2015). And also, protease has been engineered using rational design and directed evolution approach to improve its properties and functions to be applied as therapeutic agents and in food processing (Li *et al.*, 2013). Based on huge demand of protease market and its application, new candidate of protease remained a worth for further discovery.

1.2 Problem statement / significance of study

Halophilic bacteria produce polyextremophilic enzymes that may have useful application in various biotechnological field. For instance, protease can act as fibrinolytic agent and also removing protein based stains such as blood and sweat effectively (Karn and Kumar, 2015). Most of the commercial bacterial proteases used in detergent industry are produced from *Bacillus* sp. (Gupta *et al.*, 2002b), lesser investigation on protease from *Virgibacillus* sp., and until today, no commercial protease is originated from genus *Virgibacillus* as well. Furthermore, expenditure cost in detergent industry such as purification, production (Niyonzima and More, 2015b) and protein engineering to increase protease efficiency (Li *et al.*, 2013) are expensive. To sort out these problems, a single step of production with the use of crude enzyme is required (Niyonzima and More, 2015a), a more cost effective way compared to purification. Moreover, exploration on novel enzymes with extraordinary properties from extremophiles is always in demand and continuously in research field. Therefore, this study was conducted to characterize extracellular protease produced from a halophilic bacterium, *Virgibacillus* sp. strain CD6 that is potentially to be applied in various industries, especially in detergent formulation.

1.3 Objectives of study

The objectives of this research are:

- i. To select the best nitrogen source for protease production.
- ii. To assess the effect of physico-chemical factors on the activity and stability of protease from *Virgibacillus* sp. CD6.
- iii. To analyze extracellular protease sequences encoded for *Virgibacillus* sp.

1.4 Scope of study

The previously isolated halophilic bacteria, *Virgibacillus* sp. strain CD6 was initially screened for extracellular protease activity by using qualitative approaches, (skim milk agar and gelatin liquefaction). After that, medium for protease production was formulated and effect of nitrogen sources on protease production was investigated. The optimum conditions of protease activity and its stability in terms of pH, temperature and salt concentration were determined. Then, protease stability in presence of metal ions, inhibitors, detergent constituents and organic solvent was assessed. Compatibility of protease with commercial detergents and substrate specificity of protease were also investigated. Lastly, annotated protein sequences of extracellular proteases of *Virgibacillus* sp. were analyzed using bioinformatics approach and phylogenetic protein tree was constructed.

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