

POME POLISHING TREATMENT USING PALM KERNEL SHELL ACTIVATED
CARBON PRODUCED VIA INDUSTRIAL KILN

RUGAYAH BINTI ABD FATAH

UNIVERSITI TEKNOLOGI MALAYSIA

POME POLISHING TREATMENT USING PALM KERNEL SHELL
ACTIVATED CARBON PRODUCED VIA INDUSTRIAL KILN

RUGAYAH BINTI ABD FATAH

A thesis submitted in fulfilment of the
requirements for the award of the degree of
Master in Engineering (Chemical)

Faculty of Chemical and Energy Engineering
Universiti Teknologi Malaysia

JANUARY 2017

*To all my family who gave me encouragement, everlasting love and support
throughout my study years*

&

*To my husband Afiq, TQ for being my pillar of strength during my hard time and
picking up all my missing pieces. To Nuhaa, I Love You!*

ACKNOWLEDGMENT

بِسْمِ اللَّهِ الرَّحْمَنِ الرَّحِيمِ

I would like to express my gratefulness to Allah S.W.T for giving me strength and wisdom in my research work. In preparing this thesis, I was in contact with many people, researchers, academicians, technicians and practitioners. They all have contributed to my understanding and valuable thoughts during my research.

First and foremost, I would like to express my special thanks to my supervisor, Dr. Norzita Ngadi for her encouragement, guidance, ideas which enlighten my curiosity, suggestion, advice and friendship. To my field-supervisor, Dr Astimar Abdul Aziz from Malaysian Palm Oil Board (MPOB) and to MPOB for granting me generous financial support to complete this research study under *Graduate Student Assistantship Scheme (GSAS)* that enabling this research to be one successfully. I am gratefully expressing my thanks to my whole family who understand me and gave me the spirit and continuing support to finish this study.

To MPOB staff especially En. Abdul Hafiz Shukor, En. Mashuri, Pn. Nik Fadzillah Nik Mat, Pn. Sri Rahayu and Pn. Nor Ezzanee Jamal Nasir thanks a lot for making MPOB environment really warm for me.

My fellow colleagues who also should be recognized for their moral support, may Allah bless you all. Their view and tips are useful indeed, but it is not possible to list them all in this limited space. Thank you very much!

ABSTRACT

Malaysia is one of the largest producers and exporters of palm oil products in the Asian region. To manage the waste generated from palm oil processing is not easy as some of palm oil mills still fail to comply with the standard discharge limit set by Department of Environment even though biologically treatment unit has been applied. Hence, in this study polishing treatment unit was proposed using adsorption process that improved the 3-days biological oxygen demand (BOD_3), chemical oxygen demand (COD), suspended solid (SS) and colour removing efficiency so that the standard discharge limit can be achieved. The potential of palm kernel shell (PKS) based activated carbon (AC) as adsorbent in removal of organic matters in palm oil mill effluent (POME) was investigated. The precursor was prepared in a large scale environmental charcoal kiln by carbonization under controlled atmosphere and physically activated using large scale rotary kiln using the optimum parameter (steam injection at $5L\ h^{-1}$; temperature $900-1000\ ^\circ C$). The largest Brunauer-Emmett-Teller (BET) surface area of palm kernel shell activated carbon (PKS AC) obtained was $607.8\ m^2/g$ with the pore volume of $0.25\ cm^3/g$. The optimum condition to reduce BOD_3 to $17\ mgL^{-1}$ was $50\ g/250\ mL$ PKS AC, 24 h mixing time, 150 rpm and at pH 2 whereas COD, SS, and colour value obtained after polishing treatment were 203 mg/L, 145 mg/L and 2640 Pt-Co, respectively. The adsorption equilibrium was best represented by Langmuir and Freundlich isotherm model whilst the kinetics of adsorption was well described by pseudo-first order model. Regeneration study was carried out by thermal treatment. The adsorption capacity for organic matters was $6.08-4.36\ mg/g$, after five adsorption-regeneration cycles. This study has identified that PKS AC has a potential to be used as a precursor in the preparation of AC to achieve BOD_3 of less than $20\ mgL^{-1}$ in POME polishing treatment.

ABSTRAK

Malaysia merupakan salah satu pengeluar dan pengeksport terbesar produk minyak sawit di rantau Asia. Untuk menguruskan bahan buangan yang dihasilkan daripada pemprosesan minyak sawit bukanlah mudah memandangkan segelintir kilang minyak sawit masih gagal mematuhi had pelepasan standard yang ditetapkan oleh Jabatan Alam Sekitar walaupun unit rawatan biologi telah digunakan. Oleh itu, unit kajian rawatan penggilap menggunakan proses penjerapan telah dicadangkan untuk memperbaiki nilai keperluan oksigen biokimia 3-hari (BOD_3), keperluan oksigen kimia (COD), pepejal terampai (SS) dan kecekapan penyingkiran warna supaya had pelepasan standard boleh dicapai. Potensi karbon teraktif berasaskan tempurung kelapa sawit (PKS) sebagai penjerap untuk menyingkiran bahan organik di dalam bahan buangan sisa minyak sawit (POME) telah dikaji. Pelopor ini telah disediakan di dalam tanur arang mesra alam berskala besar di bawah atmosfera terkawal dan diaktifkan secara fizikal menggunakan tanur berputar skala besar di bawah parameter optimum (suntikan stim pada 5 Lh^{-1} ; suhu $900\text{-}1000 \text{ }^\circ\text{C}$). Luas permukaan Brunauer-Emmett-Teller (BET) terbesar karbon teraktif tempurung kelapa sawit (PKS AC) dapat dicapai sehingga $607.8 \text{ m}^2/\text{g}$ dengan jumlah liang $0.25 \text{ cm}^3/\text{g}$. Keadaan optimum untuk mengurangkan BOD_3 kepada 17 mgL^{-1} adalah $50 \text{ g}/250 \text{ ml}$ PKS AC, 24 jam masa pencampuran, 150 rpm dan pada pH 2 manakala bacaan akhir COD, SS dan nilai warna yang diperoleh selepas rawatan penggilap adalah masing-masing 203 mgL^{-1} , 145 mgL^{-1} dan 2640 Pt-Co. Keseimbangan penjerapan adalah terbaik dipadankan oleh kedua-dua model isoterma Langmuir dan Freudlich manakala kinetik penjerapan dihuraikan oleh model kinetik pseudo pertama. Kajian penjanaaan semula telah dilaksanakan dengan rawatan terma. Kapasiti penjerapan bahan organik didapati sebanyak $6.08\text{-}4.36 \text{ mg/g}$ walaupun selepas lima kitaran penjerapan-penjanaaan semula. Kajian ini telah mengenal pasti bahawa PKS AC mempunyai potensi untuk digunakan sebagai pelopor dalam penyediaan AC untuk mencapai BOD_3 kurang daripada 20 mgL^{-1} dalam rawatan penggilapan POME.

TABLE OF CONTENTS

CHAPTER	TITLE	PAGE
	DECLARATION	ii
	DEDICATION	iii
	ACKNOWLEDGEMENT	iv
	ABSTRACT	v
	ABSTRAK	vi
	TABLE OF CONTENTS	vii
	LIST OF TABLES	xi
	LIST OF FIGURES	xiii
	LIST OF SYMBOLS AND ABBREVIATIONS	xv
	LIST OF APPENDICES	xviii
1	INTRODUCTION	1
	1.1 Overview	1
	1.2 Background of Research	3
	1.3 Problem Statement	4
	1.4 Objectives of Research	5
	1.5 Scope of Research	5
	1.6 Significance of Research	6
	1.7 Thesis Outline	7
2	LITERATURE REVIEW	8
	2.1 Palm Oil Milling Process	8
	2.2 Palm Oil Mill Effluent (POME)	12
	2.3 POME Treatment System	17

2.3.1	Pretreatment Methods	17
2.3.1.1	Segregation (De-Oiling Tank and Sand Trap)	17
2.3.1.2	Cooling Pretreatment Unit	18
2.3.2	Primary Treatment Unit	18
2.3.2.1	Ponding Treatment System	18
2.3.2.2	Digester Tank and Aeration Pond	21
2.3.2.3	Extended Aeration System	22
2.3.3	Post-treatment Method: Land Application	22
2.4	Existing Technology for POME Treatment	23
2.5	Biomass	26
2.5.1	Biomass Chemical Compositions	27
2.5.2	Charcoal	32
2.5.3	The Stages in Charcoal Formation	34
2.5.4	Activated Carbon	35
2.5.4.1	Activated Carbon Preparation	37
2.6	Adsorption	39
2.6.1	Adsorption Equilibrium Models	40
2.6.2	Langmuir Adsorption Model	41
2.6.3	Freundlich Adsorption Model	44
2.6.4	BET (Brunauer, Emmett and Teller) Adsorption Isotherm	46
2.7	Regeneration	47
2.7.1	Thermal Regeneration	48
2.7.2	Current Activated Carbon Regeneration Studies	49
3	MATERIAL AND METHOD	52
3.1	Introduction	52
3.2	Materials and Equipment	54
3.2.1	PKS AC Preparation	55
3.2.2	POME Sample Collection and Preservation	57
3.2.3	PKS AC and Commercial AC Characterization	58

3.2.3.1	Surface Porosity Characterization (N ₂ Adsorption-Desorption)	59
3.2.3.2	Surface Morphology Scanning Surface Methodology (SEM)	59
3.2.3.3	Surface Functional Group Analysis	60
3.2.3.4	Thermal Analysis	60
3.3	Organic Adsorption Analyses	60
3.3.1	Adsorption Isotherm and Kinetic Study	63
3.4	Analytical Methods	64
3.4.1	Biological Oxygen Demand	65
3.4.2	Chemical Oxygen Demand – Reactor Digestion Method	66
3.4.3	Suspended Solid Measurement	66
3.4.4	Colour Measurement	67
3.5	Regeneration Study and Comparison of Regeneration Efficiency	67
4	RESULTS AND DISCUSSION	69
4.1	Introduction	69
4.2	PKS AC Preparation and Characterization	70
4.2.1	Chemical and Physical analyses of PKS, PKS charcoal and PKS AC	70
4.2.2	Thermal Analysis	72
4.2.3	Surface Functional Group Analysis	74
4.2.4	Nitrogen Adsorption Isotherm Analysis	77
4.2.5	Surface Morphology Analysis	80
4.3	Biologically Treated POME Characterization	82
4.4	Comparative Studies of PKS AC and CAC using Batch Adsorption Analysis	83
4.4.1	Effect of Adsorbent Dosage	83
4.4.2	Effect of Mixing Time	86
4.4.3	Effect of Mixing Rate	89
4.4.4	Effect of pH	91
4.4.5	Optimum Condition	94

	4.4.6 Adsorption Isotherm	94
	4.4.7 Adsorption Kinetics	99
	4.5 Saturation Point	102
	4.6 Regeneration and Adsorption Cycle Efficiency	104
5	CONCLUSIONS AND RECOMMENDATIONS	107
	5.1 Conclusion	107
	5.2 Recommendation	108
	REFERENCES	110
	APPENDICES A-I	123

LIST OF TABLES

TABLE NO.	TITLE	PAGE
2.1	Dry Basis of Calorific Value of Oil Palm Biomass (Yap, 2010)	11
2.2	Characteristic of Palm Oil Mill Effluents (Borja-Pardilla and Banks, 1994)	11
2.3	Chemical properties of Raw POME (Shahrakbah et al., 2004).	14
2.4	Environmental Regulations for Watercourse Discharge for POME (Malaysia, 1977)	15
2.5	Palm Oil Mill Effluent Discharge Standards (1978 -1984) (Malaysia, 1977)	16
2.6	Cellulose, Hemicellulose and Lignin Content in Various Types of Biomass	30
2.7	Proximate analysis of oil palm residues (Husain, Zainac and Abdullah, 2002)	32
2.8	Ultimate Analysis of Oil Palm Residues (Mahila et al., 2001)	32
2.9	Activated Carbon Pore Classification	37
2.10	Interpretations of IUPAC Classification for Adsorption Isotherms	41
2.11	Parameters Indicator of Langmuir Isotherm (Malik 2004)	42
3.1	Equipment Used in the Research Study	54
3.2	Reproducibility Data for All Parameters of Raw POME	58
4.1	PKS Charcoal composition and yield	70
4.2	Proximate and Ultimate analysis of Raw PKS, Coconut Shell, PKS Charcoal & PKS AC	72

4.3	FTIR spectra of raw PKS, PKS charcoal, and PKS AC	77
4.4	Single point BET Surface Area, Langmuir surface area, and pore volume of raw PKS, PKS AC and CAC	78
4.5	Characteristics of final treated POME	83
4.6	Comparison of percentage removal of BOD ₃ , COD, SS and colour on PKS AC and CAC adsorbent dose (10g/250mL), mixing time (1-120 hour), mixing rate (45-200 rpm) and pH (2-12)	94
4.7	Langmuir and Freundlich isotherm constant and separation factor (R_L) for adsorption of organic in terms of COD removal on PKS AC at 30, 40 and 50°C	98
4.8	Comparison of the of pseudo-first order and pseudo-second order model for different initial concentration of biologically treated POME at 30°C	102
4.9	Adsorption uptake and carbon yields of PKS AC and CAC regenerated five cycles	105
4.10	Adsorption isotherm parameters for the adsorption of organic matter onto virgin and PKS AC and CAC regenerated five cycles	106

LIST OF FIGURES

FIGURES NO.	TITLE	PAGE
2.1	Planted Area of Palm Oil and Production in Malayasia from 1975 to 2015	12
2.2	Chemical Structure of Cellulose	28
2.3	Sugar Components of Hemicellulose (Hashem et al., 2007)	29
2.4	Schematic Illustration of Lignin Building Units (Demirbas, 2008a)	30
2.5	Solid Palm Oil Mill Residue - Palm Kernel Shell	31
2.6	Palm Kernel Shell Charcoal	33
2.7	Pores of Palm Oil Shell Activated Carbon (Wan Nik, 2006)	37
2.8	The IUPAC Classification for Adsorption Isotherms (Webb and Orr, 1997)	41
2.9	Linearized Freundlich Isotherm for Basic Blue Dye Adsorption By Activated Palm Ash (Abdulbari et al., 2006)	45
3.1	Experimental design	53
3.2	Charcoal kiln system for PKS carbonization	55
3.3	Rotary kiln for PKS charcoal activation	56
3.4	Jar Test Apparatus	62
4.1	TGA profiles of raw PKS, PKS Charcoal and PKS AC	74
4.2	FTIR spectra of PKS, PKS charcoal and PKS AC	75

4.3	Nitrogen adsorption-desorption isotherm for (a) PKS charcoal (b) PKS AC and (c) CAC	79
4.4	SEM micrograph of raw PKS (6500 Magnification)	81
4.5	SEM micrograph of PKS charcoal (6500 Magnification)	81
4.6	SEM micrograph of PKS AC (6500 Magnification)	82
4.7	Effect of PKS AC and CAC dosage on (a) BOD ₃ , (b) COD, (c) SS and (d) colour removal	85
4.8	Colour reduction at different dosage using PKS AC	86
4.9	Effect of mixing time on (a) BOD ₃ , (b) COD, (c) SS and (d) colour removal using PKS AC and CAC on POME polishing treatment	87
4.10	Reduction percentage of (a) BOD ₃ , (b) COD, (c) SS and (d) colour versus mixing rate	90
4.11	Effect of pH in reduction of (a) BOD ₃ , (b) COD, (c) SS and (d) colour on adsorption of PKS AC and CAC	92
4.12	Langmuir isotherm for organic adsorption in terms of COD removal onto PKS AC	96
4.13	Freundlich isotherm for adsorption of organic in terms of COD removal onto PKS AC at 30, 40 and 50°C	97
4.14	Pseudo-first order kinetics for adsorption of organic matter on PKS AC at 30°C	100
4.15	Pseudo-second order kinetics for adsorption of organic matter on PKS AC at 30°C	100
4.16	Organic matter adsorption capacity vs. contact time at 30°C	101
4.17	Effect of BOD ₃ and COD value on POME treatment cycles on (a) PKS AC and (b) CAC	103

LIST OF SYMBOLS AND ABBREVIATIONS

ACs	-	Activated carbons
ASTM	-	American Society for Testing and Materials
BET	-	Brunauer, Emmett and Teller
BOD	-	Biochemical oxygen demand
BOD ₃	-	3-days Biochemical Oxygen Demand
CAC	-	Commercial Activated Carbon
Cd	-	Cadmium
Ce	-	equilibrium concentration
cm	-	centimeter
C ₀	-	initial concentration
CO ₂	-	Carbon dioxide
COD	-	Chemical oxygen demand
CPO	-	Crude palm oil
Cu	-	Cuprum
DOE	-	Department of Environmental
EFB	-	Empty Fruit Bunches
Fe	-	Ferum
FFA	-	Free fatty acids
FFB	-	Fresh fruits bunches
FTIR	-	Fourier Transform Infra-red
g	-	gram
H ₂ SO ₄	-	Sulphuric acid
H ₃ PO ₄	-	Phosphoric acid
HCl	-	Hydrochloric acid
HRT	-	Hydrolic retention time
IUPAC	-	International Union of Pure and Applied Chemistry

K	-	Kelvin
KOH	-	Potassium hydroxide
L	-	Liter
L/mg	-	liter per miligram
MB	-	Methylene blue
MDF	-	Medium density fiber board
mg L ⁻¹	-	Milligram per liter
mg/mg	-	miligram per miligram
Mn	-	Manganese
MO	-	Moringa Oleifera
mol / g	-	mol per gram
MPOB	-	Malaysian palm oil board
MT	-	Metric tonnes
NaOH	-	Sodium hydroxide
NTU	-	Nephelometric unit
NTU	-	Nephelometric Turbidity Unit
oC	-	Degree celcius
OPF	-	Oil palm frond
OPT	-	Oil palm trunk
PACl	-	Polyaluminum chloride
PAM	-	Polyacrilamide
PKS AC	-	Palm Kernel Shell Activated Carbon
PNP	-	P-nitrophenol
POME	-	Palm oil mill effluent
PU	-	Polyurethane
RO	-	Reverse osmosis
RSM	-	Response surface methodology
S _{BET}	-	BET surface area
SEM	-	Scanning electron microscopy
SEM	-	Scanning electron microscopy
SS	-	Suspended Solid
t	-	time
TGA	-	Thermal gravimetric analyzer

TSS	-	Total suspended solid
UF	-	Ultrafiltration
V _T	-	Total pore volume
wt%	-	Weight percent
ZnCl ₂	-	Zinc chloride

LIST OF APPENDICES

APPENDIX	TITLE	PAGE
A	Data of PKS AC and CAC dosage on BOD ₃ , COD, SS & Colour reduction	123
B	Data of mixing rate of PKS AC and CAC on BOD ₃ , COD, SS & Colour reduction	124
C	Data of mixing time of PKS AC and CAC on BOD ₃ , COD, SS & Colour reduction	125
D	Adsorption Equilibrium Data	127
E	Adsorption Kinetics Data	129
F	CAC Nitrogen Adsorption-Desorption for Surface Area	130
G	PKS AC Nitrogen Adsorption-Desorption for Surface Area Determination	132
H	FTIR spectra for RAW PKS, PKS CHARCOAL and PKS AC	134
I	Drum Arrangement inside Environmental Charcoal Kiln for Carbonization Process	136

CHAPTER 1

INTRODUCTION

1.1 Overview

We are proud that our country have brought oil palm trees from the small scale crop plantation to become the largest commodity product in the world. Malaysia has produced about 17.3 million tons of crude palm oil, and exported the bulk of its products and securing for the country RM44498 million in revenue by 2014 (Malaysia Palm Oil Board, 2014). Oil palm tree is the major crop plantation in Malaysia and the palm oil industry has had an exceptional growth for about four decades ago as the top agricultural industry. Presently, almost half of the agricultural land in Malaysia is under oil palm and the plantation area is still expanding from time to time. Today, 44% of world exports and 39% of world palm oil production were monopolized by Malaysia. Between 1990 and 2015, the palm oil production almost tripled from 6,094,622 to 19,666,953 tonnes per year, ranking Malaysia as the second biggest palm oil producer throughout the world after Indonesia (Malaysia Palm Oil Council, 2015). As the producer and exporter of palm oil and palm oil products, Malaysia plays an important role to fulfill the growing global needs for oils and fats in general.

Throughout the rapid development of the palm oil industry in the country, oil palm tree, its product and also the by-product produced have always been connected to the environment. In 2014, it was estimated that 93.38 million tonnes of empty fruit

bunches (EFB), 21.03 million tonnes of oil palm fibre (OPF), and 4.46 million tonnes of oil palm shell (OPS) all together totalled about 118.9 million tonnes of solid biomass were produced (Loh, 2016). The fast and rapid development both upstream and downstream would result to an unmanageable environmental pollution directly.

Currently, palm kernel shell (PKS) is less being exploited other than being used as a secondary fuel for boiler particularly during starting up of burning (Yap and Menon, 2012). The remaining of PKS is usually left abandoned at the landfill site near to the processing mill. For that reason, PKS which is less utilized in the current market is preferred as the raw material for the production of activated carbon.

Similar to other agricultural and industrial activities, oil palm cultivating, milling and processing are also regulated by environmental legislation that aims to conserve and protect the natural environment. These rules and regulations play an important role to minimize the degradation of the atmospheric, soil as well as water in the environment (Fairhurst and Mutert, 1999). To produce palm oil products, large amount of water is consumed, and consequently, it will generate tremendous volume of wastewater. Palm oil mills are the main source of palm oil wastewater other than palm oil refineries. The water pollution known as palm oil mill effluent (POME) is generated mainly from the sterilization process during crude oil production. POME water pollution is very synonymous to palm oil processing. Approximately, 30 million tonnes of POME are generated annually from more than 300 palm oil mills in Malaysia. An outstanding water pollution in the environmental system, enforcement must to be placed in its treatment system.

In 1978, the Environmental Quality Regulations which described the discharge standards for POME had been enacted. Several specifications for the discharge limit of POME had been implemented and one of the parameter was the Biochemical Oxygen Demand (BOD). Initially, the BOD discharge from the mill was from 25000 ppm of the untreated POME, the value was later reduced in the first

generation of discharge standard to 5000 ppm and was further reduced to the present BOD of 100 ppm (Malaysia, 1977).

1.2 Background of Research

The POME is brownish in colour and has a colloidal suspension containing 95-96% water, 4-5% total solids, and 0.6-0.7% oil and grease. It also contains high BOD and voluminous industrial waste. Over half of the total solids are dissolved solids, while the other half is a mixture of various form of inorganic and organic suspended solids. It is a thick liquid with a temperature of 80-90°C and acidic nature with a pH value around 4.0-5.0 during discharge in the cooling pond. With this characteristic, together with a high BOD value and low pH, makes this liquid waste not only difficult to treat by conventional methods but also highly polluting the environment. According to Bhatia et al., (2007b), the production of 985,603 tonnes of crude palm oil requires 1,477,595m³ of water to produce 38,797m³ of POME. Thus, effective POME treatment technologies are required to meet DOE discharge limits i.e. 400ppm TSS and 100ppm BOD to control water stream pollution.

Currently, there are several treatment methods that have been used by palm oil mill industries in Malaysia. Most of them are using the facultative anaerobic ponds, tank digestion and mechanical aeration, facultative ponds and tank digestion, facultative ponds and decanter, biological and physicochemical treatment. Apparently, this treatment method had also tried out several methods including membrane technology, up flow anaerobic sludge fixed film bioreactor, up flow anaerobic filtration, and aerobic sludge blanked. Presently, according to Vijayaraghavan et al., (2007) about 85% of the POME treatment method use the anaerobic and facultative ponding technology followed by the open tank digester together with aeration in the pond.

From previous research studies (Ahmad et al., 2003; Ahmad et al., 2005a, Ahmad et al., 2006; Ariffin et al., 2005; Mohamed and Lau, 1987; Ng et al., 1987; Bhatia et al., 2007b), it was reported that, most of the POME treatments used the physicochemical method (i.e. coagulation/flocculation method). However, most of the researches that were conducted focused more on the treatment of POME in the primary stage.

In the current situation, most of the treatment plants use the anaerobic digestion to treat POME followed by an aerobic oxidation in facultative and maturation or algae ponds. Thus, in this research study, it is intended to reduce the COD, BOD₃, SS and colour. The polishing part using activated carbon as an adsorbent will be used at the end of the treatment so that the value of the parameters could meet the Department of Environmental (DOE) standards. The polishing treatment is also aimed to bring the BOD₃ level to below 20 mgL⁻¹. The efficiency of the activated carbon adsorbent will be assessed by treating the final discharge of the effluent obtained from the treatment plant located at Labu, Negeri Sembilan.

1.3 Problem Statement

Through extensive research on the biological POME treatment technology, it was found that many treatment plants failed to achieve the standard discharge limits established by DOE. The final treated effluent discharged into the water courses still contains a high BOD₃, COD, SS and colour value which can certainly cause problems to the environment (Wu et al., 2009). For that reason, an alternative for POME treatment system is highly required to meet standard discharge limits. It is believed that tertiary treatment is able to improve the quality of the effluent since there is an attempt to enforce 20 mg/L BOD₃ discharge limit on crude palm oil (Liew et al., 2015). Therefore, in this study, the absorption process from activated carbon as

an option to treat biologically treated POME will be investigated. This study will explore the possibility of using activated carbon as a polishing step in POME treatment for reducing BOD₃ to below 20 mg/L. The production of PKS AC in a large scale using environmental kiln for carbonization and rotary kiln for activation will also be practiced.

1.4 Objectives of Research

This research project is designed to achieve the following objectives

- i. To synthesize and characterise AC from PKS produced from industrial kiln.
- ii. To study the effect of the synthesized PKS AC towards POME polishing treatment by assessing the removal of BOD₃, COD, SS and color.
- iii. To study the equilibrium and kinetic of the adsorption process.
- iv. To perform the regeneration study of the PKS AC.

1.5 Scope of Research

The steps and the scopes leading to the objectives are:

1. Characterization of raw PKS collected from Felcra Berhad Sungai Melikai Oil Palm Mill Mersing Johor, PKS charcoal and PKS AC on proximate (moisture, volatile matter, ash and fixed carbon), ultimate analysis (C, H, N, O, S content), lignocellulosic analysis (cellulose, hemicellulose and lignin), thermogravimetric analysis (TGA), Fourier

Transform Infrared Spectroscopy (FTIR), pore development, and morphology using Scanning Electron Microscopy (SEM).

2. To characterize biologically treated POME sample by finding the value of BOD₃, COD, SS and color for every different sample taken from the mill.
3. To investigate the optimum dosage (0-125g), mixing time (20-140 hours), mixing rate (45-200rpm), and pH (2-12) of PKS AC in reducing the pollutant load (BOD₃, COD, SS and color).
4. To analyse the equilibrium (Langmuir and Freundlich isotherm method) and kinetics study (Pseudo-first order & Pseudo-second order) of PKS AC sorption.
5. To compare the performance of PKS AC with commercial AC available in the market.
6. To perform saturation and regeneration studies (5 cycles) of PKS AC and compare to CAC.

1.6 Significance of Research

After the enforcement of the rules and regulation according to the standard discharge limit for crude palm oil industry under Environmental Quality (Prescribed Premises) (Crude Palm Oil) Regulations 1977, Malaysia and other crude palm oil industries are dealing with the challenge of sustainable development, palm oil economic feasibility, and most important is the balancing the environmental protection. Focusing on this country, the number of palm oil mills effluent discharge

failed to meet the required standard limit established by the DOE, even biological treatment methods has been employed. This research study is purposely done to make an up-grading of the existing wastewater treatment plants in order to comply with the standards stipulated by the authorities. By applying the adsorption process in the final part at the treatment plant is expected will significantly improve the treatment systems and thus reduce the liability of the effluent discharge to the environment.

1.7 Thesis Outline

This thesis is divided into 5 main chapters. Chapter 1 explains the introduction of this whole research work. This chapter consists of research background, problem statements, objectives, scopes and significance of this research. Literature Review are given in Chapter 2 which explains in detail about the previous researches that are related to POME treatment and production of activated carbon as well as researches that are relevant to this area. Chapter 3 has been described as the experimental procedures such as PKS AC preparation, POME characterization, analytical methods involved to evaluate the efficiency of PKS AC as well as the comparison between commercial AC and finally the regeneration of PKS AC. The next chapter is Chapter 4, where it explains in detail the results and discussions for the whole study. Finally, Chapter 5 concludes the findings and significance of this study. Recommendations are also suggested in assurance the positive outlook of this research area.

REFERENCES

- ABDUL AZIZ, A., DAS, K. HUSIN, M., AND MOKHTAR, A. (2002). Effects of Physical and Chemical Pre-treatment on Xylose and Glucose Production from Oil Palm Press Fibre. *Journal of Oil Palm Research*, 14(2):10-17.
- ABER, S. & SHEYDAEI, M. 2012. Removal of COD from Industrial Effluent Containing Indigo Dye Using Adsorption Method by Activated Carbon Cloth: Optimization, Kinetic, and Isotherm Studies. *CLEAN – Soil, Air, Water*, 40, 87-94.
- AHMAD, A. A. & HAMEED, B. H. 2009. Reduction of COD and color of dyeing effluent from a cotton textile mill by adsorption onto bamboo-based activated carbon. *Journal of Hazardous Materials*, 172, 1538-1543.
- AHMAD, A. L., ISMAIL, S., IBRAHIM, N. & BHATIA, S. 2003. Removal of suspended solids and residual oil from palm oil mill effluent. *Journal of Chemical Technology & Biotechnology*, 78, 971-978.
- AHMAD, A. L., SUMATHI, S. & HAMEED, B. H. 2005a. Adsorption of residue oil from palm oil mill effluent using powder and flake chitosan: Equilibrium and kinetic studies. *Water Research*, 39, 2483-2494.
- AHMAD, A. L., SUMATHI, S. & HAMEED, B. H. 2005b. Residual oil and suspended solid removal using natural adsorbents chitosan, bentonite and activated carbon: A comparative study. *Chemical Engineering Journal*, 108, 179-185.
- AHMAD, A. L., SUMATHI, S. & HAMEED, B. H. 2006. Coagulation of residue oil and suspended solid in palm oil mill effluent by chitosan, alum and PAC. *Chemical Engineering Journal*, 118, 99-105.
- AHMAD, M. A. & RAHMAN, N. K. 2011. Equilibrium, kinetics and thermodynamic of Remazol Brilliant Orange 3R dye adsorption on coffee husk-based activated carbon. *Chemical Engineering Journal*, 170, 154-161.

- AKTAŞ, Ö. & ÇEÇEN, F. 2009. Cometabolic bioregeneration of activated carbons loaded with 2-chlorophenol. *Bioresource Technology*, 100, 4604-4610.
- ÁLVAREZ, P. M., BELTRÁN, F. J., GÓMEZ-SERRANO, V., JARAMILLO, J. & RODRÍGUEZ, E. M. 2004. Comparison between thermal and ozone regenerations of spent activated carbon exhausted with phenol. *Water Research*, 38, 2155-2165.
- ARIFFIN, A., SHATAT, R. S. A., NIK NORULAINI, A. R. & MOHD OMAR, A. K. 2005. Synthetic polyelectrolytes of varying charge densities but similar molar mass based on acrylamide and their applications on palm oil mill effluent treatment. *Desalination*, 173, 201-208.
- ARAMI-NIYA, A., DAUD, W. M. A. W. & MJALLI, F. S. 2011. Comparative study of the textural characteristics of oil palm shell activated carbon produced by chemical and physical activation for methane adsorption. *Chemical Engineering Research and Design*, 89, 657-664.
- ARAMI-NIYA, A., WAN DAUD, W. M. A., S. MJALLI, F., ABNISA, F. & SHAFEEYAN, M. S. 2012. Production of microporous palm shell based activated carbon for methane adsorption: Modeling and optimization using response surface methodology. *Chemical Engineering Research and Design*, 90, 776-784.
- ARRIAGADA, R., GARCÍA, R., MOLINA-SABIO, M. & RODRIGUEZ-REINOSO, F. 1997. Effect of steam activation on the porosity and chemical nature of activated carbons from Eucalyptus globulus and peach stones. *Microporous Materials*, 8, 123-130.
- BAGREEV, A., RAHMAN, H. & BANDOSZ, T. J. 2002. Study of regeneration of activated carbons used as H₂S adsorbents in water treatment plants. *Advances in Environmental Research*, 6, 303-311.
- BANSODE, R. R., LOSSO, J. N., MARSHALL, W. E., RAO, R. M. & PORTIER, R. J. 2004. Pecan shell-based granular activated carbon for treatment of chemical oxygen demand (COD) in municipal wastewater. *Bioresource Technology*, 94, 129-135.
- BASIRON, Y., JALANI, B. S., CHAN, K. W. & MALAYSIA, L. M. S. 2000. *Advances in Oil Palm Research*, Malaysian Palm Oil Board, Ministry of Primary Industries, Malaysia.

- BHATIA, S., OTHMAN, Z. & AHMAD, A. L. 2007a. Coagulation–flocculation process for POME treatment using *Moringa oleifera* seeds extract: Optimization studies. *Chemical Engineering Journal*, 133, 205-212.
- BHATIA, S., OTHMAN, Z. & AHMAD, A. L. 2007b. Pretreatment of palm oil mill effluent (POME) using *Moringa oleifera* seeds as natural coagulant. *Journal of Hazardous Materials*, 145, 120-126.
- BORJA, R. & BANKS, C. J. 1994. Treatment of palm oil mill effluent by upflow anaerobic filtration. *Journal of Chemical Technology & Biotechnology*, 61, 103-109.
- BOUALIA, A., MELLAH, A., AISSAOUI, T. T., MENACER, K. & SILEM, A. 1993. Adsorption of organic matter contained in industrial H₃PO₄ onto bentonite: batch-contact time and kinetic study. *Applied Clay Science*, 7, 431-445.
- BRIDGWATER, A. V. 1994. Catalysis in thermal biomass conversion. *Applied Catalysis A: General*, 116, 5-47.
- BRIDGWATER, A. V. 1996. Production of high grade fuels and chemicals from catalytic pyrolysis of biomass. *Catalysis Today*, 29, 285-295.
- ÇALIŞKAN, E., BERMÚDEZ, J. M., PARRA, J. B., MENÉNDEZ, J. A., MAHRAMANLIOĞLU, M. & ANIA, C. O. 2012. Low temperature regeneration of activated carbons using microwaves: Revising conventional wisdom. *Journal of Environmental Management*, 102, 134-140.
- CATURLA, F., MOLINA-SABIO, M. & RODRÍGUEZ-REINOSO, F. 1991. Preparation of activated carbon by chemical activation with ZnCl₂. *Carbon*, 29, 999-1007.
- CAZETTA, A. L., JUNIOR, O. P., VARGAS, A. M. M., DA SILVA, A. P., ZOU, X., ASEFA, T. & ALMEIDA, V. C. 2013. Thermal regeneration study of high surface area activated carbon obtained from coconut shell: Characterization and application of response surface methodology. *Journal of Analytical and Applied Pyrolysis*, 101, 53-60.
- CHEN, X., JEYASEELAN, S. & GRAHAM, N. 2002. Physical and chemical properties study of the activated carbon made from sewage sludge. *Waste Management*, 22, 755-760.

- CHOI, G.-G., OH, S.-J., LEE, S.-J. & KIM, J.-S. 2015. Production of bio-based phenolic resin and activated carbon from bio-oil and biochar derived from fast pyrolysis of palm kernel shells. *Bioresource Technology*, 178, 99-107.
- CHUNGSIRIPORN, J., PRASERTSAN, S. & BUNYAKAN, C. 2006. Minimization of water consumption and process optimization of palm oil mills. *Clean Technologies and Environmental Policy*, 8, 151-158.
- CLIFFORD, D., CHU, P. & LAU, A. 1983. Thermal regeneration of powdered activated carbon (pac) and pac-biological sludge mixtures. *Water Research*, 17, 1125-1138.
- COELHO, C., OLIVEIRA, A. S., PEREIRA, M. F. R. & NUNES, O. C. 2006. The influence of activated carbon surface properties on the adsorption of the herbicide molinate and the bio-regeneration of the adsorbent. *Journal of Hazardous Materials*, 138, 343-349.
- DAUD, W. M. A. W. & ALI, W. S. W. 2004. Comparison on pore development of activated carbon produced from palm shell and coconut shell. *Bioresource Technology*, 93, 63-69.
- DEMIRBAS, A. 2008a. Heavy metal adsorption onto agro-based waste materials: A review. *Journal of Hazardous Materials*, 157, 220-229.
- DEMIRBAS, A. 2008b. The importance of bioethanol and biodiesel from biomass. *Energy Sources, Part B*, 3, 177-185.
- DEMIRBAŞ, A. 1999. Properties of charcoal derived from hazelnut shell and the production of briquettes using pyrolytic oil. *Energy*, 24, 141-150.
- DEVI, R., SINGH, V. & KUMAR, A. 2008. COD and BOD reduction from coffee processing wastewater using Avacado peel carbon. *Bioresource Technology*, 99, 1853-1860.
- DIEBOLD, J. P. & BRIDGWATER, A. V. 1997. Overview of Fast Pyrolysis of Biomass for the Production of Liquid Fuels. In: BRIDGWATER, A. V. & BOOCOOCK, D. G. B. (eds.) *Developments in Thermochemical Biomass Conversion*. Springer Netherlands.
- DO, D. D. 1998. *Adsorption Analysis: Equilibria and Kinetics*, Imperial College Press.
- DURÁN-VALLE, C. J., GÓMEZ-CORZO, M., PASTOR-VILLEGAS, J. & GÓMEZ-SERRANO, V. 2005. Study of cherry stones as raw material in

- preparation of carbonaceous adsorbents. *Journal of Analytical and Applied Pyrolysis*, 73, 59-67.
- FAIRHURST, T. & MUTERT, E. 1999. Introduction to oil palm production. *Better Crops International*, 13, 3-6.
- FEDERATION, W. E. & ASSOCIATION, A. P. H. 2005. Standard methods for the examination of water and wastewater. *American Public Health Association (APHA): Washington, DC, USA*.
- FOO, K. Y. & HAMEED, B. H. 2012a. A cost effective method for regeneration of durian shell and jackfruit peel activated carbons by microwave irradiation. *Chemical Engineering Journal*, 193–194, 404-409.
- FOO, K. Y. & HAMEED, B. H. 2012b. Microwave-assisted regeneration of activated carbon. *Bioresource Technology*, 119, 234-240.
- FOO, K. Y. & HAMEED, B. H. 2012c. A rapid regeneration of methylene blue dye-loaded activated carbons with microwave heating. *Journal of Analytical and Applied Pyrolysis*, 98, 123-128.
- FREEMAN, J. J. 1990. Active carbon Edited by R. C. Bansal, J.-B. Donnet and F. Stoeckli. Marcel Dekker, New York, 1988, pp. xiv + 482, US\$150.00. ISBN 0-8247-7842-1. *Journal of Chemical Technology & Biotechnology*, 48, 240-241.
- GIRGIS, B. S., YUNIS, S. S. & SOLIMAN, A. M. 2002. Characteristics of activated carbon from peanut hulls in relation to conditions of preparation. *Materials Letters*, 57, 164-172.
- GOMEZ-SERRANO, V., PASTOR-VILLEGAS, J., PEREZ-FLORINDO, A., DURAN-VALLE, C. & VALENZUELA-CALAHORRO, C. 1996. FT-IR study of rockrose and of char and activated carbon. *Journal of Analytical and Applied Pyrolysis*, 36, 71-80.
- GOUD, V. V., MOHANTY, K., RAO, M. S. & JAYAKUMAR, N. S. 2005. Phenol Removal from Aqueous Solutions by Tamarind Nutshell Activated Carbon: Batch and Column Studies. *Chemical Engineering & Technology*, 28, 814-821.
- GUO, J. & LUA, A. C. 2001. Kinetic study on pyrolytic process of oil-palm solid waste using two-step consecutive reaction model. *Biomass and Bioenergy*, 20, 223-233.

- GUO, J. & LUA, A. C. 2003. Textural and chemical properties of adsorbent prepared from palm shell by phosphoric acid activation. *Materials Chemistry and Physics*, 80, 114-119.
- HAGHSERESHT, F., NOURI, S., FINNERTY, J. J. & LU, G. Q. 2002. Effects of Surface Chemistry on Aromatic Compound Adsorption from Dilute Aqueous Solutions by Activated Carbon. *The Journal of Physical Chemistry B*, 106, 10935-10943.
- HASHIM, R., NADHARI, W. N. A. W., SULAIMAN, O., KAWAMURA, F., HIZIROGLU, S., SATO, M., SUGIMOTO, T., SENG, T. G. & TANAKA, R. 2011. Characterization of raw materials and manufactured binderless particleboard from oil palm biomass. *Materials & Design*, 32, 246-254.
- HAUCK, W., GREVILLOT, G. & LAMINE, A. S. 1997. Induktionsheizung: Anwendung auf die Regenerierung von beladenen Aktivkohle-Festbetten. *Chemie-Ingenieur-Technik*, 69, 1138-1142.
- HE, W., LÜ, G., CUI, J., WU, L. & LIAO, L. 2012. Regeneration of Spent Activated Carbon by Yeast and Chemical Method. *Chinese Journal of Chemical Engineering*, 20, 659-664.
- HILLIS, W. E. 1985. Wood and Biomass Ultrastructure. In: OVEREND, R. P., MILNE, T. A. & MUDGE, L. K. (eds.) *Fundamentals of Thermochemical Biomass Conversion*. Springer Netherlands.
- HU, Z. & SRINIVASAN, M. P. 2001. Mesoporous high-surface-area activated carbon. *Microporous and Mesoporous Materials*, 43, 267-275.
- HUSAIN, Z., ZAINAC, Z. & ABDULLAH, Z. 2002. Briquetting of palm fibre and shell from the processing of palm nuts to palm oil. *Biomass and Bioenergy*, 22, 505-509.
- IDRIS, M. A., JAMI, M. S. & MUYIBI, S. A. 2010. Tertiary treatment of biologically treated palm oil mill effluent (POME) using UF membrane system: effect of MWCO and transmembrane pressure. *International Journal of Chemical and Environmental Engineering*, 1, 108-112.
- JIA, Q. & LUA, A. C. 2008. Effects of pyrolysis conditions on the physical characteristics of oil-palm-shell activated carbons used in aqueous phase phenol adsorption. *Journal of Analytical and Applied Pyrolysis*, 83, 175-179.

- KIM, J. H., RYU, Y. K., HAAM, S., LEE, C. H. & KIM, W. S. 2001. Adsorption and steam regeneration of n-hexane, MEK, and toluene on activated carbon fiber. *Separation Science and Technology*, 36, 263-281.
- KUTTY, S., NGATENAH, S., JOHAN, N. & AMAT, K. Removal of Zn (II), Cu (II), chemical oxygen demand (COD) and colour from anaerobically treated palm oil mill effluent (POME) using microwave incinerated rice husk ash (MIRHA). *Int Conf Environ Ind Innov IPCBEE*, 2011. 90-94.
- LAINE, J., CALAFAT, A. & LABADY, M. 1989. Preparation and characterization of activated carbons from coconut shell impregnated with phosphoric acid. *Carbon*, 27, 191-195.
- LANGMUIR, I. 1918. THE ADSORPTION OF GASES ON PLANE SURFACES OF GLASS, MICA AND PLATINUM. *Journal of the American Chemical Society*, 40, 1361-1403.
- LARTEY, R., ACQUAH, F. & NKETIA, K. 1999. Developing national capability for manufacture of activated carbon from agricultural wastes. *Ghana Eng*, 19, 1-2.
- LATIF AHMAD, A., ISMAIL, S. & BHATIA, S. 2003. Water recycling from palm oil mill effluent (POME) using membrane technology. *Desalination*, 157, 87-95.
- LEE, K. M. & LIM, P. E. 2005. Bioregeneration of powdered activated carbon in the treatment of alkyl-substituted phenolic compounds in simultaneous adsorption and biodegradation processes. *Chemosphere*, 58, 407-416.
- Legal Research Board, 2008. Environmental Quality Act 1974 (Act 127) & Subsidiary Legislation. International Law Book Services, Malaysia, Petaling Jaya, pp. 1-62.
- LIM, J. L. & OKADA, M. 2005. Regeneration of granular activated carbon using ultrasound. *Ultrasonics Sonochemistry*, 12, 277-282.
- LIU, Y. 2008. New insights into pseudo-second-order kinetic equation for adsorption. *Colloids and Surfaces A: Physicochemical and Engineering Aspects*, 320, 275-278.
- LIEW, W. L., KASSIM, M. A., MUDA, K., LOH, S. K. & AFFAM, A. C. 2015. Conventional methods and emerging wastewater polishing technologies for palm oil mill effluent treatment: A review. *Journal of Environmental Management*, 149, 222-235.

- LU, P.-J., LIN, H.-C., YU, W.-T. & CHERN, J.-M. 2011. Chemical regeneration of activated carbon used for dye adsorption. *Journal of the Taiwan Institute of Chemical Engineers*, 42, 305-311.
- MAE, K., HASEGAWA, I., SAKAI, N. & MIURA, K. 2000. A New Conversion Method for Recovering Valuable Chemicals from Oil Palm Shell Wastes Utilizing Liquid-Phase Oxidation with H₂O₂ under Mild Conditions. *Energy & Fuels*, 14, 1212-1218.
- MALAYSIA (1977). *Environmental Quality (Prescribed Premises) (Crude Palm-Oil) Regulations 1977*. P.U. (A)342 1977.
- MAHLIA, T. M. I., ABDULMUIN, M. Z., ALAMSYAH, T. M. I. & MUKHLISHIEN, D. 2001. An alternative energy source from palm wastes industry for Malaysia and Indonesia. *Energy Conversion and Management*, 42, 2109-2118.
- MALIK, P. K. 2004. Dye removal from wastewater using activated carbon developed from sawdust: adsorption equilibrium and kinetics. *J Hazard Mater*, 113, 81-8.
- MAROTO-VALER, M. M., DRANCA, I., CLIFFORD, D., LUPASCU, T., NASTAS, R. & LEON Y LEON, C. A. 2006. Thermal regeneration of activated carbons saturated with ortho- and meta-chlorophenols. *Thermochimica Acta*, 444, 148-156.
- MATOS, J., NAHAS, C., ROJAS, L. & ROSALES, M. 2011. Synthesis and characterization of activated carbon from sawdust of Algarroba wood. 1. Physical activation and pyrolysis. *Journal of Hazardous Materials*, 196, 360-369.
- MATTSON, J. S. & MARK, H. B. *Activated Carbon: Surface Chemistry and Adsorption from Solution*, Books on Demand.
- MEIER, D., ANDERSONS, B., IRBE, I., CHIRKOVA, J. & FAIX, O. 2008. Preliminary Study on Fungicide and Sorption Effects of Fast Pyrolysis Liquids Used as Wood Preservative. *Progress in Thermochemical Biomass Conversion*. Blackwell Science Ltd.
- MENARD, D., PY, X. & MAZET, N. 2007. Activated carbon monolith of high thermal conductivity for adsorption processes improvement: Part B. Thermal

- regeneration. *Chemical Engineering and Processing: Process Intensification*, 46, 565-572.
- MOHAMMED, R. R., KETABCHI, M. R. & MCKAY, G. 2014. Combined magnetic field and adsorption process for treatment of biologically treated palm oil mill effluent (POME). *Chemical Engineering Journal*, 243, 31-42.
- MOHAMED ISMAIL ABDUL, K. & LAU LEH, H. 1987. The use of coagulating and polymeric flocculating agents in the treatment of palm oil mill effluent (POME). *Biological Wastes*, 22, 209-218.
- MOHAMED, M. M. 2004. Acid dye removal: comparison of surfactant-modified mesoporous FSM-16 with activated carbon derived from rice husk. *J Colloid Interface Sci*, 272, 28-34.
- MOHAMMED, R. R. & CHONG, M. F. 2014. Treatment and decolorization of biologically treated Palm Oil Mill Effluent (POME) using banana peel as novel biosorbent. *J Environ Manage*, 132, 237-49.
- MOHAN, D., PITTMAN, C. U. & STEELE, P. H. 2006. Pyrolysis of Wood/Biomass for Bio-oil: A Critical Review. *Energy & Fuels*, 20, 848-889.
- MALAYSIAN PALM OIL BOARD. *Overview of the Malaysian Oil Palm Industry 2014*. Available form:
http://bepi.mpob.gov.my/images/overview/Overview_of_Industry_2014.pdf
(accessed 2.27.16)
- MALAYSIA PALM OIL COUNCIL, 2015. *Malaysian Palm Oil Industry 2015*. Available form:
http://www.mpoc.org.my/Malaysian_Palm_Oil_Industry.aspx
(accessed 2.25.16)
- NABAIS, J. M. V., NUNES, P., CARROTT, P. J. M., RIBEIRO CARROTT, M. M. L., GARCÍA, A. M. & DÍAZ-DÍEZ, M. A. 2008. Production of activated carbons from coffee endocarp by CO₂ and steam activation. *Fuel Processing Technology*, 89, 262-268.
- NAHM, S. W., SHIM, W. G., PARK, Y.-K. & KIM, S. C. 2012. Thermal and chemical regeneration of spent activated carbon and its adsorption property for toluene. *Chemical Engineering Journal*, 210, 500-509.
- NARBAITZ, R. M. & MCEWEN, J. 2012. Electrochemical regeneration of field spent GAC from two water treatment plants. *Water Research*, 46, 4852-4860.

- NEOH, C., LAM, C., LIM, C., YAHYA, A. & IBRAHIM, Z. 2014. Decolorization of palm oil mill effluent using growing cultures of *Curvularia clavata*. *Environmental Science and Pollution Research*, 21, 4397-4408.
- NEWCOMBE, G. & DRIKAS, M. 1993. Chemical regeneration of granular activated carbon from an operating water treatment plant. *Water Research*, 27, 161-165.
- NG, W. J., GOH, A. C. C. & TAY, J. H. 1987. Palm oil mill effluent (POME) treatment—An assessment of coagulants used to aid liquid-solid separation. *Biological Wastes*, 21, 237-248.
- NOLL, K. E. 1991. *Adsorption Technology for Air and Water Pollution Control*, Taylor & Francis.
- OH, G. H. & PARK, C. R. 2002. Preparation and characteristics of rice-straw-based porous carbons with high adsorption capacity. *Fuel*, 81, 327-336.
- OLIVARES-MARÍN, M., FERNÁNDEZ-GONZÁLEZ, C., MACÍAS-GARCÍA, A. & GÓMEZ-SERRANO, V. 2012. Preparation of activated carbon from cherry stones by physical activation in air. Influence of the chemical carbonisation with H₂SO₄. *Journal of Analytical and Applied Pyrolysis*, 94, 131-137.
- PEREZ, M., ROMERO, L. I. & SALES, D. 2001. Organic Matter Degradation Kinetics in an Anaerobic Thermophilic Fluidised Bed Bioreactor. *Anaerobe*, 7, 25-35.
- QU, G. Z., LI, J., WU, Y., LI, G. F. & LI, D. 2009. Regeneration of acid orange 7-exhausted granular activated carbon with dielectric barrier discharge plasma. *Chemical Engineering Journal*, 146, 168-173.
- RIOS, R. V. R. A., MARTÍNEZ-ESCANDELL, M., MOLINA-SABIO, M. & RODRÍGUEZ-REINOSO, F. 2006. Carbon foam prepared by pyrolysis of olive stones under steam. *Carbon*, 44, 1448-1454.
- ROBERS, A., FIGURA, M., THIESEN, P. H. & NIEMEYER, B. 2005. Desorption of odor-active compounds by microwaves, ultrasound, and water. *AIChE Journal*, 51, 502-510.
- RODRÍGUEZ-REINOSO, F., MOLINA-SABIO, M. & GONZÁLEZ, M. T. 1995. The use of steam and CO₂ as activating agents in the preparation of activated carbons. *Carbon*, 33, 15-23.

- RUDZINSKI, W. & PLAZINSKI, W. 2009. On the applicability of the pseudo-second order equation to represent the kinetics of adsorption at solid/solution interfaces: a theoretical analysis based on the statistical rate theory. *Adsorption*, 15, 181-192.
- RYU, Y. K., KIM, K. L. & LEE, C. H. 2000. Adsorption and desorption of n-hexane, methyl ethyl ketone, and toluene on an activated carbon fiber from supercritical carbon dioxide. *Industrial and Engineering Chemistry Research*, 39, 2510-2518.
- SABIO, E., GONZÁLEZ, E., GONZÁLEZ, J. F., GONZÁLEZ-GARCÍA, C. M., RAMIRO, A. & GAÑAN, J. 2004. Thermal regeneration of activated carbon saturated with p-nitrophenol. *Carbon*, 42, 2285-2293.
- SHAHRAKBAH, Y., YOSHIHITO, S., YUNG-TSE, H. & MOHD ALI, H. 2004. Treatment of Palm Oil Wastewaters. *Handbook of Industrial and Hazardous Wastes Treatment*. CRC Press.
- SJÖSTRÖM, E. 1993. *Wood Chemistry: Fundamentals and Applications*, Academic Press.
- SOUZA, A. D., PINA, P. S., LEÃO, V. A., SILVA, C. A. & SIQUEIRA, P. F. 2007. The leaching kinetics of a zinc sulphide concentrate in acid ferric sulphate. *Hydrometallurgy*, 89, 72-81.
- SRIVASTAVA, V. C., MALL, I. D. & MISHRA, I. M. 2005. Treatment of pulp and paper mill wastewaters with poly aluminium chloride and bagasse fly ash. *Colloids and Surfaces A: Physicochemical and Engineering Aspects*, 260, 17-28.
- STUART, B. H. 2005. Spectral Analysis. *Infrared Spectroscopy: Fundamentals and Applications*. John Wiley & Sons, Ltd.
- SUN, K. & JIANG, J. C. 2010. Preparation and characterization of activated carbon from rubber-seed shell by physical activation with steam. *Biomass and Bioenergy*, 34, 539-544.
- TAMRIN, T., MUYIBI, S. A., JAMI, M. S. & AMOSA, M. K. 2013. Removal of Organics from Treated Palm Oil Mill Effluent (POME) Using Powdered Activated Carbon (PAC). *Proceedings of the 3rd International Conference on Biotechnology Engineering (ICBioE)*, 64-66.

- TCHOBANOGLIOUS, G., BURTON, F. L., STENSEL, H. D., METCALF & EDDY 2003. *Wastewater Engineering: Treatment and Reuse*, McGraw-Hill Education.
- TSAI, W. T., CHANG, C. Y., WANG, S. Y., CHANG, C. F., CHIEN, S. F. & SUN, H. F. 2001. Preparation of activated carbons from corn cob catalyzed by potassium salts and subsequent gasification with CO₂. *Bioresource Technology*, 78, 203-208.
- VERMA, V. & MISHRA, A. 2008. Removal of dyes using low cost adsorbents. *Indian Journal of Chemical Technology*, 15, 140.
- VIJAYARAGHAVAN, K., AHMAD, D. & EZANI BIN ABDUL AZIZ, M. 2007. Aerobic treatment of palm oil mill effluent. *Journal of Environmental Management*, 82, 24-31.
- WALTHER, H. J. 1988. S. D. Faust and O. M. Aly: Adsorption processes for water treatment. Guildford, Butterworth Scientific Ltd., 1987, 509 S., 204 Abb., 107 Tab., £ 75. –, ISBN 0-409-90000-1. *Acta hydrochimica et hydrobiologica*, 16, 572-572.
- WEBB, P., ORR, C. & MICROMERITICS INSTRUMENT, C. 1997. *Analytical methods in fine particle technology*, Norcross, Ga., Micromeritics Instrument Corp.
- WEBER JR, W. J., MCGINLEY, P. M. & KATZ, L. E. 1991. Sorption phenomena in subsurface systems: Concepts, models and effects on contaminant fate and transport. *Water Research*, 25, 499-528.
- WENZL, H. F. J. 1970. III - The Chemistry of Wood. In: WENZL, H. F. J. (ed.) *The Chemical Technology of Wood*. Academic Press.
- WU, T. Y., MOHAMMAD, A. W., JAHIM, J. M. & ANUAR, N. 2009. A holistic approach to managing palm oil mill effluent (POME): Biotechnological advances in the sustainable reuse of POME. *Biotechnology Advances*, 27, 40-52.
- WYMAN, C. 1996. *Handbook on Bioethanol: Production and Utilization*, Taylor & Francis.
- YANG, J., SHEN, Z. & HAO, Z. 2004. Preparation of highly microporous and mesoporous carbon from the mesophase pitch and its carbon foams with KOH. *Carbon*, 42, 1872-1875.

- YAP A.K.C, MENON R.N. 2012. Palm Biomass Utilisation in Palm Oil Mills. *Palm Oil Eng Bull*, 103. 9-13.
- YOUSSEF, A. M., RADWAN, N. R. E., ABDEL-GAWAD, I. & SINGER, G. A. A. 2005. Textural properties of activated carbons from apricot stones. *Colloids and Surfaces A: Physicochemical and Engineering Aspects*, 252, 143-151.
- ZAWANI, Z., LUQMAN, C. A. & CHOONG, T. S. Y. 2009. Equilibrium, kinetics and thermodynamic studies: Adsorption of remazol black 5 on the palm kernel shell activated carbon (PKS-AC). *European Journal of Scientific Research*, 37, 67-76.
- ZHANG, Y., YAN, L., QIAO, X., CHI, L., NIU, X., MEI, Z. & ZHANG, Z. 2008. Integration of biological method and membrane technology in treating palm oil mill effluent. *Journal of Environmental Sciences*, 20, 558-564.
- ZINATIZADEH, A., MOHAMED, A., NAJAFPOUR, G., ISA, M. H. & NASROLLAHZADEH, H. 2006. Kinetic evaluation of palm oil mill effluent digestion in a high rate up-flow anaerobic sludge fixed film bioreactor. *Process biochemistry*, 41, 1038-1046.