

REACTION CONTROL AND PROTEIN ENGINEERING OF *Bacillus lehensis* G1
MALTOGENIC AMYLASE FOR HIGHER MALTO-OLIGOSACCHARIDE
SYNTHESIS

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*Untuk Mak dan Abah tercinta
Jasa dan pengorbananmu tidak ternilai harganya*

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“Say, If the sea were ink for [writing] the words of my Lord, the sea would be exhausted before the words of my Lord were exhausted, even if We brought the like of it as a supplement.” (Qur’an 18:109)

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ABSTRACT

A multi-functional maltogenic amylase (MAG1) from alkaliphilic *Bacillus lehensis* G1 exhibited remarkable hydrolysis and transglycosylation activity to produce malto-oligosaccharides of various lengths. MAG1 demonstrated hydrolysis activity over wide range of substrates. Kinetic analysis revealed that the enzyme hydrolyzed small substrate more efficiently than the larger substrate. This was shown by lower Michaelis constant (K_m) value and higher turnover number (k_{cat}) and second order rate constant (k_{cat}/K_m) values for β -cyclodextrin compared to that of soluble starch. Malto-oligosaccharide synthesis by transglycosylation activity of MAG1 faces problem of product re-hydrolyzation due to the hydrolysis activity of the enzyme. An equilibrium-control reaction approach has been successfully employed to improve malto-oligosaccharides production by decreasing hydrolysis activity. A yield of 38% transglycosylation products was obtained with the presence of malto-oligosaccharides longer than maltoheptaose. The addition of organic solvents demonstrated an increase in the transglycosylation-to-hydrolysis ratio from 1.29 to 2.15. The transglycosylation activity of MAG1 was also successfully enhanced by using structure-guided protein engineering approach. A molecular modeling and substrate docking was performed to study the structure-function relationship for rational design. A unique subsite structure which has not been reported in other maltogenic amylases was revealed and the information was used to design mutants that have active sites with reduced steric interference and higher hydrophobicity properties to increase the transglycosylation activity. Mutations decreased the hydrolysis activity of the enzyme and caused various modulations in its transglycosylation property. W359F, Y377F and M375I mutations caused reductions in steric interference and alteration of subsite occupation. In addition, the mutations increased internal flexibility to accommodate longer donor/acceptor molecule for transglycosylation, resulted in increased transglycosylation to hydrolysis ratio of up to 4.0-fold. The increase of the active site hydrophobicity from W359F and M375I mutations reduced concentration of maltotriose used as donor/acceptor for transglycosylation to 100 mM and 50 mM, respectively compared to 200 mM of the wild-type. The improvement of the transglycosylation to hydrolysis ratio by 4.3-fold was also demonstrated by both mutants. Interestingly, reductions of both steric interference and hydrolysis by Y377F and W359F mutations caused a synergistic effect to produce malto-oligosaccharides with higher degree of polymerization than the wild-type. These findings showed that the transglycosylation activity of MAG1 was successfully improved by controlling water activity and modification of the active site structure. The high transglycosylation activity of MAG1 and mutants offers a great advantage for synthesizing malto-oligosaccharides and rare carbohydrates.

ABSTRAK

Amilase maltogenik pelbagai fungsi (MAG1) daripada bakteria alkalifilik, *Bacillus lehensis* G1 menunjukkan aktiviti hidrolisis dan pentransglukosilan tinggi untuk menghasilkan malto-oligosakarida yang mempunyai panjang rantai yang berbeza. MAG1 menunjukkan hidrolisis terhadap pelbagai jenis substrat. Analisis kinetik menunjukkan ia menjalankan hidrolisis lebih berkesan terhadap substrat kecil berbanding substrat besar. Ini dibuktikan melalui nilai pemalar Michaelis (K_m) yang lebih rendah dan nombor perolehan (k_{cat}) dan pemalar kadar tertib kedua (k_{cat}/K_m) yang lebih tinggi oleh β -siklodekstrin berbanding kanji larut. Sintesis malto-oligosakarida oleh amilase maltogenik sering berhadapan dengan hidrolisis semula produk disebabkan kebolehan enzim tersebut menjalankan aktiviti hidrolisis. Kaedah pengawalan keseimbangan tindakbalas telah digunakan untuk mengurangkan aktiviti hidrolisis justeru meningkatkan penghasilan malto-oligosakarida. Sebanyak 38% produk pentransglukosilan dengan kehadiran malto-oligosakarida lebih panjang daripada maltoheptosa berjaya dihasilkan. Penambahan pelarut organik meningkatkan nisbah pentransglukosilan kepada hidrolisis daripada 1.29 kepada 2.15. Aktiviti pentransglukosilan MAG1 juga ditingkatkan menggunakan kaedah kejuruteraan protein berpandukan struktur. Pemodelan molekul dan dok substrat dijalankan untuk mengkaji hubungkait struktur-fungsi untuk melaksanakan reka bentuk rasional. Struktur unik subtapak yang tidak pernah dilaporkan oleh amilase maltogenik lain telah dikenalpasti dan maklumat ini digunakan untuk mereka bentuk mutan yang mempunyai ciri tapak aktif yang kurang gangguan sterik dan tinggi hidrofobisiti untuk meningkatkan aktiviti pentransglukosilan enzim. Mutasi mengurangkan aktiviti hidrolisis enzim dan menyebabkan pelbagai perubahan pada ciri-ciri pentransglukosilan. Mutasi W359F, Y377F dan M375I mengurangkan gangguan sterik dan mengubah pendudukan subtapak. Mutasi turut meningkatkan fleksibiliti struktur dalaman untuk menampung molekul penderma/penerima yang lebih panjang dan meningkatkan nisbah pentransglukosilan kepada hidrolisis sebanyak 4.0 kali ganda. Peningkatan hidrofobisiti tapak aktif melalui mutasi W359F dan M375I menyebabkan pengurangan kepekatan maltotriosa yang digunakan sebagai molekul penderma/penerima untuk pentransglukosilan kepada masing-masing 100 mM dan 50 mM berbanding 200 mM oleh MAG1. Kenaikan nisbah pentransglukosilan kepada hidrolisis sebanyak 4.3 kali ganda juga ditunjukkan oleh kedua-dua mutan. Menariknya, pengurangan gangguan sterik dan aktiviti hidrolisis melalui mutasi Y377F dan W359F memberikan kesan sinergi untuk menghasilkan malto-oligosakarida yang lebih panjang daripada MAG1. Keputusan ini menunjukkan aktiviti pentransglukosilan MAG1 berjaya ditingkatkan dengan pengawalan aktiviti air dan pengubahsuaian struktur tapak aktif. Aktiviti pentransglukosilan yang tinggi ditunjukkan oleh MAG1 dan mutan menawarkan kelebihan yang besar untuk mensintesis malto-oligosakarida dan karbohidrat nadir.

TABLE OF CONTENTS

CHAPTER	TITLE	PAGE
	DECLARATION	ii
	DEDICATION	iii
	ACKNOWLEDGEMENT	iv
	ABSTRACT	v
	ABSTRAK	vi
	TABLE OF CONTENTS	vii
	LIST OF TABLES	xiii
	LIST OF FIGURES	xv
	LIST OF SYMBOLS AND ABBREVIATIONS	xix
	LIST OF APPENDICES	xxiv
1	INTRODUCTION	1
	1.1 Background: The catalyst of life	1
	1.2 Problem statement	3
	1.3 Objectives of the study	4
	1.4 Scopes of the study	5
	1.5 Rationale and novelty of the study	5
2	LITERATURE REVIEW	7
	2.1 Introduction	7
	2.2 Oligosaccharides as functional food	8
	2.3 Production of oligosaccharides	9
	2.3.1 Chemical synthesis of oligosaccharides	10

2.3.2	Enzymatic synthesis of oligosaccharides by glycosyl transferases	11
2.3.3	Enzymatic synthesis of oligosaccharides by glycosyl hydrolases	12
2.4	Strategies to improve transglycosylation of glycosyl hydrolases	15
2.4.1	Improved oligosaccharides synthesis by reaction equilibrium control	15
2.4.2	Improved oligosaccharides synthesis by protein engineering	18
2.5	Maltogenic amylase: a multitasking enzyme	22
2.5.1	Structure-function of maltogenic amylase	24
2.5.2	Mechanism of multiple activities	30
2.5.3	Maltogenic amylase from <i>Bacillus lehensis</i> G1 and physiological role	32
2.6	Characterization of maltogenic amylase	33
2.6.1	Biochemical characterization of maltogenic amylase	34
2.6.2	Substrate and product specificity for hydrolysis	36
2.6.3	Kinetic analysis of maltogenic amylase	37
2.6.4	Transglycosylation of maltogenic amylase	40
2.7	Maltogenic amylase engineering	42
2.7.1	Modulation of substrate specificity	42
2.7.2	Improvement of transglycosylation activity	45
3	MATERIALS AND METHODS	47
3.1	Operational framework	47
3.2	Bacteria strains and plasmid	49
3.3	Chemicals and biological enzymes	49

3.4	Bacteria culturing	50
3.5	Glycerol stock preparation	50
3.6	General DNA manipulation techniques	50
3.6.1	Agarose gel DNA electrophoresis	51
3.6.2	Extraction of plasmid DNA	51
3.6.3	Megaprimer polymerase chain reaction amplification	52
3.6.4	Full-length PCR amplification	54
3.6.5	Restriction enzyme digestion	54
3.6.6	Ligation	56
3.6.7	<i>E. coli</i> competent cells preparation	56
3.6.8	Transformation	56
3.6.9	Insert screening	57
3.6.10	DNA Quantification	57
3.7	Expression of MAG1	58
3.8	Purification of recombinant enzyme	58
3.9	General protein techniques	59
3.9.1	Sodium dodecyl sulfate-polyacrylamide gel electrophoresis	59
3.9.2	Western blot	60
3.9.3	Gel filtration chromatography	61
3.9.4	Protein concentration assay	62
3.9.5	Enzyme assay	63
3.10	Characterization of MAG1	63
3.10.1	Optimum temperature and pH	63
3.10.2	Thermal and pH stability	64
3.10.3	Effects of metal ions and additives	64
3.10.4	Substrate specificity	65

3.10.5	Enzyme kinetic studies	65
3.10.6	Hydrolysis activity of MAG1	66
3.10.7	Transglycosylation activity	67
3.11	Analysis of product formation	67
3.11.1	Thin-layer chromatography	67
3.11.2	High performance liquid chromatography	68
3.12	Homology modeling and docking of substrate	68
3.12.1	Protein sequence analysis	69
3.12.2	Template search and selection	69
3.12.3	MAG1 3D structure modeling	69
3.12.4	Structure refinement and validation	70
3.12.5	Analysis of 3D structure	70
3.12.6	Docking	71
4	CLONING, EXPRESSION AND CHARACTERIZATION OF THE RECOMBINANT MALTOGENIC AMYLASE (MAG1)	72
4.1	Introduction	72
4.2	Sequence analysis of MAG1	72
4.3	Heterologous expression and purification of recombinant MAG1	75
4.4	Oligomeric state determination of MAG1	77
4.5	Thermal inactivation analysis of MAG1	79
4.6	Substrate specificity and kinetic analysis	83
4.7	Hydrolysis product specificity of MAG1	87
4.8	Malto-oligosaccharide synthesis by MAG1	94
4.9	Optimization of transglycosylation by MAG1	96
4.10	Hydrolysis suppression by organic solvent	100

4.11	Malto-oligosaccharides with a high degree of polymerization	103
4.12	Conclusion	105
5	STRUCTURAL AND FUNCTIONAL PROPERTIES OF MAG1 AND RATIONAL DESIGN	106
5.1	Introduction	106
5.2	Homology modeling of MAG1	107
5.2.1	Template search and selection	107
5.2.2	Model building	110
5.2.3	Model refinement and energy minimization	111
5.3	Structural analysis of MAG1	117
5.3.1	The active site architecture	119
5.3.2	The aromatic platform for substrate recognition	126
5.4	Docking and subsite structure analysis	128
5.4.1	Docking of substrates into the MAG1 active site	129
5.4.2	Substrate binding and subsite structure analysis	131
5.5	Rational design of MAG1	148
5.6	Conclusions	151
5	MODULATION OF TRANSGLYCOSYLATION AND IMPROVED MALTO-OLIGOSACCHARIDE SYNTHESIS BY PROTEIN ENGINEERING	152
6.1	Introduction	152
6.2	Cloning and expression of the mutants	152
6.3	Characteristics and effects of mutations on hydrolysis activity and substrate specificity	155
6.4	Effects of mutations on transglycosylation activity of the enzyme	164

6.4.1	Donor/acceptor selectivity of mutants	165
6.4.2	The effect of mutation on donor/acceptor concentration for transglycosylation	175
6.4.3	Synthesis of malto-oligosaccharides with high degrees of polymerization	180
6.5	Conclusion	182
7	CONCLUSION AND RECOMMENDATIONS	187
7.1	Conclusion	187
7.2	Recommendations	188
	REFERENCES	190
	APPENDICES A-I	210 - 261

LIST OF TABLES

TABLE NO.	TITLE	PAGE
2.1	Biochemical properties of maltogenic amylase from various organisms.	35
2.2	Kinetic parameters of maltogenic amylases on various substrates.	38
3.1	<i>E. coli</i> strains used throughout the study.	49
3.2	External primers used for full-length MAG1 gene amplification.	52
3.3	Mutagenic primers sequences.	53
4.1	Purification table of MAG1 using nickel affinity chromatography.	75
4.2	The molecular weights (MW), retention volumes (V_e), partition coefficients (K_{av}) and log MW values of protein standards and MAG1 used in gel filtration chromatography.	78
4.3	The thermal deactivation constants (K_d) and half-lives ($T_{1/2}$) values of MAG1 at various temperatures.	80
4.4	Kinetic analysis of MAG1 on β -CD and soluble starch.	87
4.5	Product variations in response to the hydrolysis of various substrates by different amylolytic enzymes.	92
4.6	Effects of organic solvents on the proportion of hydrolysis, transglycosylation, and the product ratio.	101
4.7	Comparison of oligosaccharide production from various enzymes.	104
5.1	The alignment search results against the PDB database from various servers	109
5.2	The best model generated by Modeller 9.10 and evaluations.	111

5.3	Evaluation of MAG1 model (Model 4) before and after energy minimization	114
5.4	Intermolecular hydrogen bonds between the interface residues from both MAG1 subunits.	122
5.5	The binding energy and equilibrium dissociation constant of the best malto-oligosaccharides and β -CD conformation in the MAG1 active site from molecular docking using AutoDock Vina.	131
5.6	Subsite binding residues of MAG1 from docking study.	140
5.7	Subsite structures of MAG1 and related enzymes from different sources.	144
6.1	Specific activities for hydrolysis of wild-type and mutant enzymes.	155
6.2	Kinetic parameters for wild-type MAG1 and a W359F mutant.	158
6.3	Areas and volumes of the active site cavities of MAG1 and mutants as measured using KVFinder software.	160
6.4	Factors influencing substrate specificity, mutations carried out and the effects on maltogenic amylase and related enzymes.	162
6.5	The binding energy and equilibrium dissociation constant of MAG1 and mutants on maltotriose from molecular docking using AutoDock Vina.	174
6.6	Comparison of the effect of mutation to transglycosylation to hydrolysis ratio of maltogenic amylase and related enzyme from various sources.	180
6.7	Summary table of the effects to activity and structural modification resulted from the mutation of MAG1.	184

LIST OF FIGURES

FIGURE NO.	TITLE	PAGE
2.1	Mechanisms of hydrolysis and transglycosylation catalyzed by <i>Agrobacterium</i> sp. β -glucosidase.	14
2.2	Crystal structure of <i>Thermus</i> sp. maltogenic amylase.	25
2.3	Docking of β -CD (red stick) in the ThMA active site and maltose (blue stick) in the extra sugar-binding space.	26
2.4	The schematic diagram illustrates the active site architecture of CD-degrading enzyme that explains its substrate preference.	27
2.5	Stereoview of the residues in the active site of <i>Thermus</i> sp. maltogenic amylase that involve in binding of β -CD.	28
2.6	Stereoview of the residues in the active site of <i>T. vulgaris</i> R-47 α -amylase II that involved in binding of (a) maltohexaose and (b) β -CD.	29
2.7	The double displacement mechanism of retaining glycosyl hydrolases.	31
2.8	Physiological roles of cyclodextrin glucanotransferase (CGTase) and maltogenic amylase (MAase) in <i>B. lehensis</i> G1.	33
2.9	Proposed model of the substrate specificity for the different oligomeric states of <i>Thermus</i> sp. maltogenic amylase.	43
3.1	Operational framework of the study.	48
3.2	Schematic representations for the generation of MAG1 mutated gene using megaprimer PCR method.	55
3.3	The calibration curve for gel filtration column S200 10/300.	62

3.4	The Lineweaver-Burk plot and kinetic parameters determination (Lineweaver and Burk, 1934).	66
4.1	BLASTP result showing protein sequences producing significant alignments with MAG1 amino acid sequence.	73
4.2	A multiple sequence alignment of the active site center region of MAG1 with its homologous enzymes.	74
4.3	Ni ²⁺ affinity chromatogram of MAG1 purification exhibited by the curve plot.	76
4.4	Analysis of crude and purified MAG1.	77
4.5	Gel filtration chromatogram of MAG1 using Superdex 200 13/300 column and corresponding K _{av} versus log MW plot of protein standards.	78
4.6	Plots for the analysis of MAG1 thermal deactivation.	82
4.7	MAG1 activity on various substrates.	83
4.8	Partial multiple sequence alignment.	85
4.9	The separation of hydrolysis products by TLC.	88
4.10	The time course for the hydrolysis of β -CD by MAG1.	90
4.11	The separation of the transglycosylation products by TLC.	95
4.12	The optimization of the transglycosylation reaction.	98
4.13	The HPLC separation of the malto-oligosaccharides produced from MAG1 transglycosylation activity on maltotriose.	105
5.1	Secondary structure alignment between MAG1 and 1J0H.	108
5.2	Evaluation of MAG1 model energy (a) before and (b) after energy minimization using ANOLEA.	113
5.3	Ramachandran plot for the MAG1 model after energy minimization.	115
5.4	MAG1 model evaluation using ProSA-web.	116
5.5	Stereoview of the overall dimeric structure of MAG1 illustrated by program PyMOL.	117
5.6	Superimposition of MAG1 and 1J0H structures.	118
5.7	The structure of dimeric MAG1.	120
5.8	The surface structure of MAG1 and close-up active site structure.	121

5.9	Interface residues that involved in intermolecular hydrogen bond formation between MAG1 dimer subunits.	123
5.10	The locations of the extra sugar binding site in MAG1 structure as shown by the arrow.	125
5.11	The aromatic residues that form stacking interactions with the sugar ring for substrate recognition.	127
5.12	The aromatic platform in the MAG1 active site center.	128
5.13	Stereoview of the β -CD binding into the MAG1 active site.	132
5.14	A drawing of the binding of β -CD in the MAG1 active site.	133
5.15	A drawing of the binding of (a) maltopentaose, (b) maltotetraose, (c) maltotriose and (d) maltose in the MAG1 active site.	139
5.16	An illustration of the substrate (malto-oligosaccharides) occupation in MAG1 active site.	142
5.17	Mode of action of β -CD hydrolysis by MAG1.	146
5.18	Mechanisms of (a) hydrolysis and (b) transglycosylation reactions inside the active site of MAG1.	147
5.19	Positions of (a) water path entrance and (b) target amino acids for mutations.	150
6.1	Insert screening for cloning of MAG1 mutants.	153
6.2	SDS-PAGE analysis of the crude and purified mutant enzymes.	154
6.3	Native-PAGE analysis of MAG1 and mutant enzymes.	154
6.4	Hydrolysis activities of MAG1 and mutants on different substrates.	156
6.5	The difference in the distance between the centroid of the phenyl group of residue 359 and the glucose moiety of the substrate at subsite -2 for wild-type and the W359F mutant.	157
6.6	Temperature optimization of MAG1 and mutants.	163
6.7	pH optimization of MAG1 and mutants.	164
6.8	Schematic diagram of the transglycosylation reaction mechanism in maltogenic amylase.	165

6.9	Thin-layer chromatography analysis of enzyme selectivity toward different donor/acceptor molecules during the transglycosylation reaction.	167
6.10	Transglycosylation and hydrolysis levels of variants on different donor/acceptor molecules.	169
6.11	The subsite occupations of glucose units of maltopentaose molecules in the active sites of wild-type MAG1 and W359F mutant enzymes.	171
6.12	Comparison of (a) maltopentaose and (b) maltotriose occupations in W359F and MAG1 subsites.	172
6.13	Active site cavities of (a) MAG1 and (b) W359F mutants from KVFinder analysis.	173
6.14	The effect of maltotriose (donor/acceptor) concentration on the transglycosylation activity of (A) MAG1, (B) T380V, (C) W359F, (D) Y377F and (E) M375I mutant enzymes.	178
6.15	Thin-layer chromatography analysis of transglycosylation products of wild-type and mutant MAG1 enzymes.	181

LIST OF SYMBOLS AND ABBREVIATIONS

2-ME	-	2-mercaptoethanol
3D	-	three-dimensional
A280	-	absorbance at optical density of 280 nm
A600	-	absorbance at optical density of 600 nm
Ala or A	-	alanine
Al ³⁺	-	aluminium ion
Arg or R	-	arginine
Asn or N	-	asparagine
Asp or D	-	aspartic acid
<i>B. clausii</i>	-	<i>Bacillus clausii</i>
<i>B. lehensis</i>	-	<i>Bacillus lehensis</i>
<i>B. licheniformis</i>	-	<i>Bacillus licheniformis</i>
<i>B. stearothermophilus</i>	-	<i>Bacillus stearothermophilus</i>
<i>B. subtilis</i>	-	<i>Bacillus subtilis</i>
BLAST	-	Basic Local Alignment Search Tool
<i>B. adolescentis</i>	-	<i>Bifidobacterium adolescentis</i>
BSA	-	bovine serum albumin
C	-	carbon
Ca ²⁺	-	calcium ion
CD	-	cyclodextrin
Co ²⁺	-	cobalt ion
Cu ²⁺	-	cuprum ion
Cys or C	-	cysteine
CV	-	column volume
Da	-	Dalton
DNA	-	deoxyribonucleic acid
dNTP	-	deoxynucleotide

DMSO	-	dimethyl sulphoxide
<i>E. coli</i>	-	<i>Escherichia coli</i>
EC	-	Enzyme Commission
EDTA	-	ethylenediamine tetraacetic acid
Fe ²⁺	-	ferrous ion
Fe ³⁺	-	ferric ion
FPLC	-	Fast protein liquid chromatography
g	-	gram
GC	-	guanine and cytosine
<i>G. thermoleovorans</i>	-	<i>Geobacillus thermoleovorans</i>
GH	-	Glycosyl Hydrolase
Gln or Q	-	glutamine
Glu or E	-	glutamic acid
Gly or G	-	glycine
h	-	hour
H	-	hydrogen
Hg ²⁺	-	mercury ion
His or H	-	histidine
HPLC	-	high-performance liquid chromatography
I ²⁺	-	iodide ion
I or Ile	-	isoleucine
IPTG	-	isopropyl β-D-1-thiogalactopyranoside
K	-	Kelvin
K ⁺	-	potassium ion
kcal	-	kilo calorie
kDa	-	kilodalton
kJ	-	kilo joule
l	-	liter
<i>L. gasseri</i>	-	<i>Lactobacillus gasseri</i>
L or Leu	-	Leucine
LB	-	Luria-Bertani
Li ²⁺	-	lithium ion
Lys or K	-	lysine
m	-	mili

M	-	molar
M1	-	glucose
M2	-	maltose
M3	-	maltotriose
M4	-	maltotetraose
M5	-	maltopentaose
M6	-	maltohexaose
M7	-	maltoheptaose
MAG1	-	maltogenic amylase from <i>Bacillus lehensis</i> G1
Met or M	-	methionine
mg	-	milligram
Mg ²⁺	-	magnesium ion
MgSO ₄	-	magnesium sulphate
min	-	minute
ml	-	milliliter
mM	-	milimolar
Mn ²⁺	-	manganese ion
MW	-	molecular weight
Mx	-	malto-oligosaccharides longer than maltoheptaose
Na ⁺	-	sodium ion
NEB	-	New England Biolabs
NCBI	-	The National Center for Biotechnology Information
Ni ²⁺	-	nickel ion
nm	-	nanometer
O	-	oxygen
OH	-	hydroxyl
Pb ²⁺	-	lead ion
PCR	-	polymerase chain reaction
F or Phe	-	phenylalanine
pI	-	isoelectric point
PMSF	-	phenylmethylsulfonyl fluoride
Pro or P	-	proline
PSI-BLAST	-	Position-Specific Iterated BLAST
RMSD	-	root mean square deviation

rpm	-	revolutions per minute
s	-	second
SDS	-	sodium dodecyl sulphate
SDS-PAGE	-	sodium dodecyl sulfate-polyacrylamide gel electrophoresis
Ser or S	-	serine
sp.	-	species
<i>T. vulgaris</i>	-	<i>Thermoactinomyces vulgaris</i>
<i>T. pendens</i>	-	<i>Thermofilum pendens</i>
<i>T. volcanium</i>	-	<i>Thermoplasma volcanium</i>
<i>T. maritime</i>	-	<i>Thermotoga maritime</i>
<i>T. thermophiles</i>	-	<i>Thermus thermophiles</i>
Thr or T	-	threonine
TLC	-	thin-layer chromatography
Trp or W	-	tryptophan
Tyr or Y	-	tyrosine
U	-	unit
USD	-	United State Dollar
UV	-	ultraviolet
UV-VIS	-	ultraviolet-visible spectrophotometry
V	-	volt
V or Val	-	valine
v/v	-	volume per volume
w/v	-	weight per volume
w/w	-	weight per weight
<i>X. dendrorhous</i>	-	<i>Xanthophyllomyces dendrorhous</i>
Zn ²⁺	-	zinc ion
α	-	alpha
Å	-	Angstrom
β	-	beta
β-CD	-	β-cyclodextrin
°C	-	degree celcius
γ	-	gamma
μ	-	micro

μm	-	micro meter
-	-	minus
%	-	percent
π	-	pi
ϕ	-	phi
+	-	plus
Ψ	-	psi

LIST OF APPENDICES

APPENDIX	TITLE	PAGE
A	List of publications	210
B	Nucleotide sequence of <i>B. lehensis</i> G1's maltogenic amylase (including location of primers)	214
C	Plasmid maps and sequences	217
D	Scripts for computational work	219
E	Media and buffers preparation	226
F	Experimental data calculations	240
G	Example of experimental data calculation	244
H	Standard curves	250
I	Standard operational procedures for instruments	259

CHAPTER 1

INTRODUCTION

1.1 Background: The catalyst of life

A living cell may look tiny and insignificant, but inside, it is a sea of tremendous interrelated biochemical activities that sustain life. The biochemical activities inside the cell involved various chemical transformations to maintain growth. The rate of a biochemical reaction is far too slow to support metabolism of a cell if it occurs spontaneously. These metabolic reactions are therefore accelerated a million times faster by a biochemical catalyst called enzyme. The word 'enzyme' (from the Greek, meaning 'in yeast') was proposed by Wilhelm Kühne in 1878 to differentiate between the chemical substances having catalytic properties with 'ferments', the microbes (Barnett, 2003). A new science of biochemistry emerged in the late of the eighteenth century, after a chemist, Eduard Büchner discovered that a 'juice' that was extracted from yeast, is capable of converting sugar to alcohol and carbon dioxide. The discovery led to the award of a Nobel Prize and the birth of modern biochemistry (Manchester, 2000). Today, the knowledge has expanded that the use of enzymes for catalyzing various in vitro biochemical reactions has become so common in everyday applications.

When the human genome sequence was available, researchers were surprised by the fewer number of the protein encoded for such a very complex organism. Then, they discovered that many proteins or enzymes were actually multi-functional (Jeffery, 2004). Enzymes are highly specific to their substrates and catalyzing the

reactions. However, some enzymes are also capable of catalyzing more than one reaction. It was first observed in 1890s when Arthur Croft Hill found that the hydrolysis of maltose to glucose by yeast maltase was incomplete due to another reaction that took place at the same time. He discovered that in a concentrated solution, polymerization of glucose by a reversible process occurred along with hydrolysis until the equilibrium state was reached (Manchester 2000). This proved that some enzymes are indeed capable of catalyzing more than one reaction.

Due to its growing interest, numerous multi-functional enzymes are being discovered. Glycosyl hydrolase enzymes are one of the enzymes that can catalyze two reactions, which are hydrolysis that cleaves substrate to smaller products, or transglycosylation that joins two molecules to produce a larger or longer product. Maltogenic amylase (glucan-1,4- α -maltohydrolase EC 3.2.1.133) is an amylolytic enzyme from glycosyl hydrolase family 13 (GH 13) that exhibits multi-functional property. Unlike typical α -amylases, maltogenic amylase demonstrates multi-substrates specificity and prefers cyclodextrins (CDs) as a substrate. In addition to hydrolysis, maltogenic amylase demonstrates synthetic activity known as transglycosylation to produce sugar molecules with various lengths (Kim *et al.*, 1999b). This enzyme with multiple catalytic activities becomes an interesting subject to explore. It is important to elucidate the mechanism beneath its multi-functionality of performing both hydrolysis and transglycosylation. The enzyme can act as a biochemical switch which response is regulated by the changes of the surrounding condition (Jeffery, 2004). The switching point can be explained through structural basis and the understanding of this biological role will lead to a subsequent improvement of the existing biochemical catalyst.

Maltogenic amylase can be specified as a promiscuous enzyme characterized by having a catalytic domain that employs various substrates and executes multiple biochemical reactions. Extensive researches are being carried out on this valuable enzyme with substrate and catalytic promiscuity as it has the potential to be exploited in synthetic application and development of novel synthesis pathway through protein engineering (Hult and Berglund, 2007). The transglycosylation activity of maltogenic amylase offers a great advantage in the industry for the synthesis of

oligosaccharides. Oligosaccharides offer various health benefits and diverse applications in improving physicochemical properties of foods. Conventionally, oligosaccharides were synthesized using chemical methods. However, due to its laborious steps, high cost and low yield, enzymatic synthesis has become a preferred alternative. Glycosyl transferase and glycosyl hydrolase enzymes have been employed, but the latter is favored because it uses simpler and inexpensive substrates. However, the bottleneck is synthesis reaction competes with hydrolysis resulting in a low yield of oligosaccharides produced.

The increased market demand for oligosaccharides has heightened the need of an efficient biocatalyst. The major improvement in DNA technology and bioinformatics over the years has promoted the advances of enzyme or protein engineering. Now, the discovery of novel enzymes and the availability of complete crystal structure data have enabled researchers to tailor the existing biocatalyst to fit the reaction specifications (Bornscheuer *et al.*, 2012). The study of a crystal structure of maltogenic amylase has elucidated the structure-function relationship to explain the multi-substrate and multi-functional properties of the enzyme. The role of extra N-terminal residues in the formation of domain-swapped homodimer was responsible for the multi-substrate specificity of the enzyme (Kim *et al.*, 1999a; Lee *et al.*, 2002a). In addition, scientists have discovered an extra space that resides at the bottom of the active site cleft for accommodating small acceptor sugar molecules that involved in transglycosylation (Kim *et al.*, 1999a). This understanding has provided a crucial base in re-designing the existing maltogenic amylase (MAG1) from *Bacillus lehensis* G1 for improving transglycosylation property to achieve the objective of the current study.

1.2 Problem statement

Oligosaccharides can be synthesized chemically, but enzymatic synthesis has generally been the first choice because it employs milder conditions, involves simpler steps and eliminates the need for hydroxyl group protection (Hansson *et al.*,

2001). Glycosyl transferase (EC 2.4) and glycosyl hydrolase (EC 3.2) are the two candidate enzyme classes that have been used in oligosaccharides synthesis (Bucke, 1996). Glycosyl hydrolase is preferred for oligosaccharides production because this enzymes can use simple and inexpensive acceptor sugar molecules (Hansson *et al.*, 2001). However, its major shortcoming is the inevitable hydrolysis activity of the enzyme that causes the synthesized oligosaccharides to be hydrolyzed again (Hinz *et al.*, 2006). Various strategies have been employed to overcome the problem, including the control of thermodynamic equilibrium of the enzyme reaction by the elimination of water, which is a competing nucleophile, for transglycosylation. The incorporation of an organic medium into the reaction mixture has been shown to increase the synthesis of galacto-oligosaccharide by β -glycosidase (Cruz-Guerrero *et al.*, 2006). However, no study on the influence of organic solvents on malto-oligosaccharide synthesis by maltogenic amylase has been reported to date. Protein engineering approach has also become a common practice to improve the transglycosylation activity and decrease the hydrolysis activity of the enzyme especially when the use of high substrate concentration is not favorable for industrial application and organic solvent is often avoided when the product is targeted for food additive. Therefore, through protein engineering, the enzyme with desired properties can be obtained by re-constructing or re-designing the active site.

1.3 Objectives of the study

The objectives of this research are to improve the transglycosylation property of MAG1 by using reaction equilibrium control and protein engineering approach for production of malto-oligosaccharide and to study amino acids that are important for hydrolysis and transglycosylation.

1.4 Scopes of the study

This study focusses on the improvement of transglycosylation activity of MAG1 for malto-oligosaccharide production by appropriate strategies. Hence, the following scopes were outlined to achieve the objective:

- i. Cloning, expression, purification and characterization of MAG1 in *Escherichia coli* expression system.
- ii. Kinetic study of MAG1 hydrolysis activity on various substrates and products determination.
- iii. Reaction study of transglycosylation activity by MAG1 on various sugar donors and acceptors.
- iv. Reaction equilibrium control to reduce hydrolysis activity and increase transglycosylation activity by suppressing water activity.
- v. Construction of the 3D structure of MAG1, docking of substrates into MAG1 active site and subsite structure determination.
- vi. Rational design targeted for suppressing hydrolysis activity and improving transglycosylation activity of MAG1 for higher malto-oligosaccharide synthesis.
- vii. Mutant construction using site-directed mutagenesis, cloning, expression, purification, characterization and reaction studies (hydrolysis and transglycosylation) of the mutants.

1.5 Rationale and novelty of the study

Exploring new enzyme source is important because a different source of enzymes exhibits their own unique characteristic even though they have some similarities in terms of function, substrate they act upon or their structure. Moreover, continuous offering of data regarding this enzyme could be beneficial to the scientific community and provide more knowledge in understanding this type of the biocatalyst. The potential of maltogenic amylase for the synthesis of carbohydrate has been recognized by researchers. Many scientific studies dealing with this

enzyme for transglycosylation are only focusing on the production of transfer products from specific carbohydrate donor and acceptor. The production of malto-oligosaccharide by maltogenic amylase has a few been reported. However, the use of reaction equilibrium control approach to reduce the hydrolysis activity of maltogenic amylase for increased transglycosylation has yet to be reported in literature and little is known about the effects of organic solvents on the production of malto-oligosaccharide by maltogenic amylase. This report is the first to describe the optimization of reaction conditions and the incorporation of a water-miscible organic solvent to suppress hydrolysis activity during malto-oligosaccharide production by maltogenic amylase. The present study showed that the addition of an organic solvent could be used to produce malto-oligosaccharides with degrees of polymerization higher than maltoheptaose. The findings demonstrated that MAG1 is a promising candidate for carbohydrate synthesis applications.

Analysis of homology modeling and docking of malto-oligosaccharides in this study reveals a novel finding regarding the subsite structure of MAG1 which is different from the reported subsite structures of other maltogenic amylases. This finding leads to a proposed mode of action for β -cyclodextrin (β -CD) hydrolysis that describes how maltose is mainly produced from the hydrolysis. Protein engineering combined with site-directed mutagenesis has also been successfully employed in this study to shift the enzyme activity toward favoring transglycosylation and subsequently to increase the production of malto-oligosaccharide. Although the protein engineering approach has commonly been employed to improve transglycosylation activity of maltogenic amylase, the production of longer malto-oligosaccharides has not yet been reported. The mutants MAG1 have produced malto-oligosaccharides longer than that of the wild-type. Longer oligosaccharides are desirable as prebiotics because they are less fermentable and, therefore, can reach the most distal area of the colon (Voragen, 1998). Moreover, the understanding of the structural modifications generated from the mutation of specific amino acids in this study will also contribute knowledge for better understanding of the structure-function of maltogenic amylase and related enzymes.

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APPENDIX A: List of publications

1. Nor Hasmaliana Abdul Manas, Mohd Anuar Jonet, Abdul Munir Abdul Murad, Farah Diba Abu Bakar, Nor Muhammad Mahadi, Rosli Md. Illias (2016). Computational docking simulations and subsite structure analysis of a maltogenic amylase from *Bacillus lehensis* G1 provide insights into hydrolysis product specificity. *Journal of Molecular Graphics and Modeling*. Manuscript submitted.
2. Nor Hasmaliana Abdul Manas, Mohd Anuar Jonet, Abdul Munir Abdul Murad Nor Muhammad Mahadi and Rosli Md. Illias (2015). Modulation of transglycosylation and improved maltooligosaccharide synthesis by protein engineering of maltogenic amylase from *Bacillus lehensis* G1. *Process Biochemistry*. 50: 1572–1580
3. Nor Hasmaliana Abdul Manas, Samson Pachelles, Nor Muhammad Mahadi and Rosli Md. Illias (2014). The characterisation of an alkali-stable maltogenic amylase from *Bacillus lehensis* G1 and improved malto-oligosaccharide production by hydrolysis suppression. *PLoS ONE*. 9(9): e106481.
4. Rosli Md. Illias, Nor Hasmaliana Abdul Manas, Mohd Anuar Jonet and Abdul Munir Abdul Murad (2016). Improvement of transglycosylation activity of *Bacillus lehensis* G1 maltogenic amylase for higher malto-oligosaccharide synthesis. The 7th AFOB Regional Symposium 2016. January 28-30, 2016. Hue City, Vietnam.
5. Nor Hasmaliana Abdul Manas and Rosli Md Illias (2015). Structural Interpretation for Hydrolysis and Transglycosylation Characteristics of Maltogenic Amylase from *Bacillus lehensis* G1. The 20th Biological Sciences Graduate Congress (20th BSGC). December 9-11, 2015. Bangkok, Thailand.
6. Rosli Md Illias, Goh Kian Mau, Nor Hasmaliana Abdul Manas. (2014). Structural insight into the catalytic site of multifunctional starch degrading enzymes. *The 2nd International Symposium and Workshop on Functional Genomics and Structural Biology*. 21 – 24 January 2014. Selangor, Malaysia.

7. Nor Hasmaliana Abdul Manas, Samson Pachelles, Abdul Munir Abdul Murad, Nor Muhammad Mahadi and Rosli Md Illias. (2013). High Maltotriose-Producing Maltogenic Amylase from *Bacillus lehensis* G1 and Preliminary Investigation on its Multifunctionality. *International Congress of the Malaysian Society for Microbiology (ICMSM2013)*. 12 – 15 December 2013. Langkawi, Malaysia.
8. Nor Hasmaliana Abdul Manas, Samson Pachelles, Nor Muhammad Mahadi, Rosli Md. Illias. (2013). A Novel Hydrolysis Pattern of a High Maltotriose Producing Maltogenic Amylase from *Bacillus lehensis* G1. *Kolokium Biologi Struktur Ke-V*. 3 June 2013. Melaka, Malaysia.
9. Nor Hasmaliana Abdul Manas, Abdul Munir Abdul Murad, Nor Muhammad Mahadi, Rosli Md. Illias. (2010). Enhancement of Solubility in *Escherichia coli* and Characterization of Cold-Adapted α -Amylase from *Leucosporidium antarcticum* PI12. *Zamalah Graduate Colloquium*. 15 July 2010. Skudai, Malaysia.
10. Nor Hasmaliana Abdul Manas, Rosli Md Illias, Noor Azah Jema'on, Nor Muhammad Mahadi and Amir Rabu. (2010). Transformation of Na^+/H^+ Antiporter Gene from Alkaliphilic *Bacillus lehensis* G1 Increases Salt Tolerance to *Escherichia coli*. *3rd International Conference Southeast Asian Natural Resources and Environmental Management (SANREM 2010)*. 3 – 5 August 2010. Sabah, Malaysia.
11. Nor Hasmaliana Abdul Manas, Rosli Md Illias, Noor Azah Jema'on, Nor Muhammad Mahadi and Amir Rabu (2010). Expression and Functional Studies of Na^+/H^+ Antiporter from Alkaliphilic *Bacillus lehensis* G1. *11th Symposium of Malaysian Society of Applied Biology*. 13 – 15 June 2010. Kelantan, Malaysia.