

SYNTHESIS OF SUPRAMOLECULAR SELF-ASSEMBLED BENZENE-1,3,5-TRICARBOXAMIDES AND THEIR LIQUID CRYSTALLINE PROPERTIES

Juan Matmin¹, Leny Yuliati², Salasiah Endud¹, Hendrik O. Lintang^{*2}

¹Department of Chemistry, Faculty of Science, Universiti Teknologi Malaysia, 81310 Johor Bahru, Johor, Malaysia ²Ibnu Sina Institute for Fundamental Science Studies, Universiti Teknologi Malaysia, 81310 Johor Bahru, Johor, Malaysia *Corresponding author: <u>hendrik@ibnusina.utm.my</u>

Supramolecular self-assembly of soft matter have received many attention for various applications such as liquid crystals, sensors, catalysts, and optics due to its precise control and easily tunable characteristics. Particularly, benzene-1,3,5-tricarboxamides (BTAs) as π stacked organic motif were recognized for formation of liquid crystals materials; however, these BTAs were only reported bearing different short alkyl or aromatic side chains in order to have liquid crystalline properties. Herein, self-assemblies of BTAs (C_RNH-BTA, where R = 10, 12, 14, 16, and 18) via hydrogen bonding interactions have been demonstrated as mesophase liquid crystalline materials by simply tuning its long hydrophobic alkyl chains attached to amide. Series of BTAs were synthesized by a Schotten-Baumann reaction using 1,3,5-benzenetricarbonyl trichloride with respected alkyl amides in the presence of triethylamine in dry tetrahydrofuran (THF) under an inert condition for overnight. Nuclear magnetic resonance (NMR), matrix-assisted laser desorption/ionization time-of-flight mass (MALDI-TOF-MS) and Fourier transform infrared spectroscopy (FT-IR) have proved the successful isolation of the hydrophobic **BTAs** using column chromatography as powdery solids. Scanning electron microscope (SEM) measurements showed self-assembled aggregation of a tape-like morphology with different aspect to ratio lengths corresponding to different alkyl side-chains lengths. Likewise, FT-IR spectra showed vibration bands at 1540 and 1719 cm⁻¹ due to formation of N–H and C=O bondings, respectively, corresponding to the presence of hydrogen bonding interactions for all BTAs. Moreover, polarized optical microscopy (POM) showed spehrulitic-like texture upon isotropic melts at 140 °C monitored for 20 hours of C₁₂NH-BTA, strongly supports the characteristics of discotic liquid crystals for BTAs. In addition, differential scanning calorimetry (DSC) thermograms showed that longer alkyl side-chains shortened mesophase ranges due to the presence of higher steric hindrance of the alkyl amide chains.

P104

CHEMICAL CONSTITUENT FROM ROOT EXTRACT OF CASSIA SINGUEANA

Saidu Jibril*, Hasnah Mohd. Sirat*, Norazah Basar

Department of Chemistry, Faculty of Science, Universiti Teknologi Malaysia, 81310 Skudai, Johor, Malaysia *Corresponding author: saidu.jibril@fukashere.edu.ng, hasnah@kimia.fs.utm.my

Cassia singueana Linn. one of the species of *Caesalpiniaceae* is used for the treatment of various diseases by traditional medicine practitioners in Northern Nigeria. In this research, extraction of *Cassia singueana* root was carried out by soxhlet extraction technique and