NOVEL UV LED PHOTOPOLYMERIZATION AND CHARACTERIZATION FOR POLYACRYLAMIDE AND POLY(*N*-ISOPROPYLACRYLAMIDE) HYDROGELS

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To my families, supervisor and friends, thank you for all your support along the way

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ABSTRACT

In this study, polyacrylamide (PAAm) and poly(N-isopropylacrylamide) (PNIPAAm) hydrogels were synthesized via ultraviolet light-emitting diode (UV LED) $(\lambda \sim 365 \text{ nm})$ photopolymerization system. UV LED technology has offered better alternative than UV mercury (Hg) system for curing technology especially for temperature-sensitive polymeric hydrogels as it can be operated without heat generation and with fast curing response. The control experiment using commercial photoinitiators has shown that UV LED system was suitable to polymerize hydrogels provided that; photoinitiator has the overlap emission with UV LED spectra. However, most commercial photoinitiators have limited solubility in water. Thus, suitable water soluble photoinitiator (WSPI) for UV LED system (i.e. $\lambda \rightarrow 330-365$ nm) was prepared in order to synthesize UV LED curable hydrogel in purely water formulation. The water soluble photoinitiator was obtained from complexation of 2,2-dimethoxy-2phenylacetophenone (DMPA) and methylated-β-cyclodextrin (MβCD). According to the results presented in this work, high monomer conversion (> 90 %) was achieved with WSPI-initiated hydrogels. The non-responsive and responsive behavior of PAAm and PNIPAAm hydrogels towards temperature were demonstrated by swelling and rheological measurements. In addition, swelling and rheological methods gave good correlation for determination of mesh sizes. From rheological measurements, the elastic modulus (G') was higher than the loss modulus (G'') and both parameters were independent to the measured frequency window. It has shown that UV LED cured hydrogels possessed ideal rubber characteristic. Tensile properties of the hydrogels showed similar trend curve as commercial contact lenses reported in the previous study. Clearly, this study has revealed that UV LED system is a good tool to synthesize hydrogels by using the excellent choice of photoiniator.

ABSTRAK

Dalam kajian hidrogel poliakrilamida (PAAm) poli(Nini, dan isopropilakrilamida) (PNIPAAm) telah disintesis melalui sistem fotopempolimeran sinaran ultraungu diod pemancar cahaya (UV LED) ($\lambda \sim 365$ nm). UV LED teknologi telah memberikan alternatif yang lebih baik berbanding sistem sinaran ultraungu merkuri (UV Hg) kerana teknologi pempolimerannya lebih cepat dan sistemnya tidak menjana haba. Ciri-ciri ini amat sesuai untuk hidrogel polimer yang sensitif terhadap haba. Eksperimen kawalan menggunakan foto pemula komersial telah menunjukkan bahawa sistem UV LED sesuai untuk pempolimeran hidrogel dengan syarat foto pemula tersebut mempunyai pertindihan spektra dengan sistem UV LED. Walau bagaimanapun, kebanyakan foto pemula komersial tidak larut dalam air. Oleh itu, foto pemula larut air (WSPI) yang sesuai untuk sistem UV LED (iaitu $\lambda \rightarrow 330-365$ nm) telah disediakan bagi membolehkan sintesis formulasi hidrogel yang berasaskan air menggunakan teknik pempolimeran UV LED. Foto pemula larut air telah diperoleh melalui pengkompleksan 2,2-dimetoksi-2-fenilasetofenon (DMPA) dan metil-βsiklodekstrin (MβCD). Berdasarkan hasil kajian, penukaran monomer yang tinggi (> 90%) telah diperoleh bagi formulasi hidrogel yang menggunakan WSPI. Sifat tidak responsif dan responsif hidrogel PAAm dan PNIPAAm terhadap suhu telah dipamerkan melalui darjah pembengkakan dan pengukuran reologi, Di samping itu, darjah pembengkakan dan pengukuran reologi ini juga memberi korelasi yang baik untuk menentukan saiz jaringan. Melalui pencirian pengukuran reologi hidrogel, modulus elastik (G') adalah lebih tinggi daripada modulus kehilangan (G'') dengan kedua-dua parameter ini tidak bergantungan pada ukuran frekuensi. Ini menunjukkan hidrogel yang diperoleh mempunyai ciri getah yang ideal. Sifat tegangan bagi hidrogel pula menunjukkan keputusan lengkuk yang sama seperti kajian yang dilaporkan sebelum ini. Secara umumnya, kajian ini telah membuktikan bahawa sistem UV LED berpotensi tinggi dalam mensintesis hidrogel dengan menggunakan foto pemula yang sesuai.

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LIST OF ABBREVIATIONS

3D	-	Three dimensional
AAm	-	Acrylamide
APS	-	Ammonium persulfate
BDMM	-	2-benzyl-2-(dimethylamino)-1-(4-morpholinophenyl)- 1-butanone
Chivacure 300	-	Oligo[2-hydroxy-2-methyl-1-[4-(1-ethylvinyl)phenyl] propanone]
Darocur 1173	-	2-hydroxy-2-methyl-1-phenyl propan-1-one
DMPA	-	2,2-Dimethoxy-2-phenylacetophenone
FTIR	-	Fourier Transform Infrared Spectroscopy
Irgacure 2959	-	1-[4-(2-hydroxyethoxy)-phenyl]-2-hydroxy-2-methyl- 1-propane-1-one
LCST	-	Lower Critical Solution Temperature
MBAAm	-	N,N'-methylenebisacrylamide
MBS	-	Sodium 4-[2-(4-morpholino)benzoyl-2-dimethylamino] butylbenzenesulfonate
NIPAAm	-	N-isopropylacrylamide
NMR	-	Nuclear Magnetic Resonance
WSPI	-	Water soluble photoinitiator
PAAm	-	Polyacrylamide
PET	-	Poly(ethylene terephthalate)
PHEMA	-	Poly-2-hydroxyethylmethacrylate
PMMA	-	Poly(methylmethacrylate)

PNIPAAm	-	Poly(N-isopropylacrylamide)
SP	-	Smart Polymer
SRH	-	Stimuli Responsive Hydrogels
TEMED	-	N,N,N',N'-tetramethylethylenediamine
TOC	-	Total Organic Carbon
UV	-	Ultraviolet
UV LED	-	Ultraviolet light-emitting diode
UV Hg	-	Ultraviolet mercury based lamp
UV-Vis	-	Ultraviolet visible spectroscopy
VOC	-	Volatile Organic Compounds

LIST OF SYMBOLS

C_N	-	Characteristic ratio
G'	-	Elastic modulus
G''	-	Loss modulus
l	-	Length
L_B	-	Length at break
L_O	-	Original length
Μ	-	Molar mass
M_{c}	-	Crosslinking points
Ν	-	Newton
N _A	-	Avogadro's number
Q	-	Degree of swelling
Т	-	Temperature
\mathbf{W}_{d}	-	Mass of dried samples
W _s	-	Mass of swollen samples
wt	-	Weight
γ	-	Strain amplitude
λ	-	Wavelength
ξ	-	Mesh sizes
ω	-	Angular frequency

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CHAPTER 1

INTRODUCTION

1.1 Research Background

In recent years, the development of polymer based hydrogels has increased worldwide. Hydrogels are hydrophilic polymers which built up of three dimensional polymeric networks [1]. Nowadays, the development of polymer based hydrogels has attracted great attention especially in biomedical applications, such as drug carriers, tissue engineering and actuators [2]. This is attributed to the characteristics of hydrogels that similar to biological tissue and at the same time compatible with the human body [1, 3].

Hydrogels have soft consistency and high water content [3] which tend to swell and retained water in its structure due to the crosslinking structure with the presence of hydrophilic groups [4]. According to their swelling behavior, hydrogels can be divided into two categories; i.e., conventional and stimuli responsive hydrogels (SRH) [5]. In this study, conventional polyacrylamide (PAAm) hydrogels and SRH poly(N-isopropylacrylamide) (PNIPAAm) hydrogels were chosen as both UV hydrogels were derivatives and usually synthesized for curing photopolymerization.

Free radical photopolymerization is one of the conventional methods used to polymerize hydrogels. Photopolymerization converts monomer into polymeric hydrogels with the help of photoinitiators [6]. Some of the researchers preferably chose photopolymerization due to its several advantages; i.e., it can be created in situ, fast curing rates as well as temporal and spatial control over the polymerization process [6, 7].

Ultraviolet light-emitting diode (UV LED) system is a green technology and environmental friendly system. It can offer fast curing rate, reduction in down time associated in maintenance and cost effectiveness. UV LED system produced less energy than UV mercury (UV Hg) system which cause no temperature builds up and very little heat is generated [8]. Up to date, UV LED system has been used for few materials, such as adhesives and coating technology [8, 9]. Hence, this light source is envisioned to be effective for curing hydrogels as well since UV LED was successfully used for coating technology.

UV LED system emits monochromatic radiation. Typical UV LED emission wavelengths are 365, 385 and 405 nm. Oligo(2-hydroxy-2-methyl-1-[4-(1-methylvinyl)phenylpropanone) under the trade name Chivacure 300 and 2,2-dimethoxy-2-phenylacetophenone (DMPA) are examples of commercially available photoinitiators. These photoinitiators have only moderate solubility in water.

From recent studies, new water soluble photoinitiators can be synthesized by several methods; i.e., introduction of hydrophilic groups or attachment of ionic groups [10-12]. Self-assembly between monomer and photoinitiator is one of the method that have been widely used due to their effectiveness and rapidness in producing modified water soluble photoinitiator [11].

In this study, PAAm and PNIPAAm hydrogels was prepared using water soluble photoinitiator via UV LED photopolymerization. Studies on physical properties of hydrogels are also very limited since hydrogels were known to have low mechanical strength. Therefore, the efficiency of modified water soluble photoinitiator towards the physical and tensile properties of hydrogels was analyzed.

1.2 Problem Statement

Recently, a considerable amount of research reported on PAAm and PNIPAAm hydrogels and most of it proposed the use of UV Hg curing system. However, several limitations of using UV Hg system were encountered; for example high energy consumption, heat generation and takes longer time to warm up.

No study has been reported yet on photopolymerization of hydrogels using UV light from LED source. Elimination of the harmful mercury and ozone extraction and reduction in down time associated in maintenance are some of the benefits that UV LED can offer [8]. Thus, development of UV LED system for hydrogels curing is a promising technology to replace UV Hg system.

The success of photopolymerization technology depends on the availability and action of appropriate photoinitiators. Successful of photopolymerization process does not only depending on the UV systems chosen, but also the efficiency of photoinitiator [13]. For such applications, 1-[4-(2-hydroxyethoxy)-phenyl-2hydroxy-2-methyl-1-propane-1-one (Irgacure 2959, $\lambda \sim 280$ nm) is the most commonly used photoinitiator, by virtue of its moderate water solubility. On the contrary, this initiator has an absorption wavelength that far below the wavelength of UV LED light ($\lambda \sim 365$ nm); make it inefficient for UV LED system.

For certain applications, water soluble photoinitiator was required to create free organic solvents condition. There is no significant published research with regard to hydrogels formulation using water soluble photoinitiator via UV LED system [12]. Thus, novel water soluble photoinitiator, WSPI with the absorption wavelength near 365 nm was produced in order to suit the UV LED system.

Water soluble photoinitiator was synthesized by several methods such as selfassembly, sulfonation process and addition of quaternary group. Self-assembly was simple and effective method among these studies [14]. This is because self-assembly method allowed for straightforward complexation to obtain new photoinitiator from the commercial photoinitiator with similar performance.

In brief, UV LED system is envisioned to be the most effective UV source for synthesis and curing of PAAm and PNIPAAm hydrogels. Suitable and efficient water soluble photoinitiator was prepared. It is expected that desired final properties of resulting hydrogels with high monomer conversion was obtained. Hydrogels with high monomer conversion shows a good integration hydrogels and very useful in many biomedical applications.

1.3 Objectives of the study

This study revolves on the development of PAAm and PNIPAAm hydrogels using the UV LED system with. Specifically, the aims are:

- a) To synthesize and characterize PAAm hydrogels using UV LED system based on optimized photopolymerization conditions obtained for UV Hg system using commercial photoinitiators.
- b) To synthesize and characterize WSPI for UV LED curable PAAm and PNIPAAm hydrogels system.
- c) To synthesize and characterize the monomer conversion, swelling, mesh sizes, rheological and tensile properties of PAAm and PNIPAAm hydrogels using WSPI via UV LED system.

1.4 Scope of the study

In this study, the first task was to synthesize PAAm and PNIPAAm hydrogels from acrylamide (AAm) and *N*-isopropylacrylamide (NIPAAm) monomers, respectively with *N*,*N'*-methylenebisacrylamide (MBAAm) as a crosslinker monomer. Polymerization of PAAm and PNIPAAm hydrogels were conducted via UV LED systems. The conditions for polymerization were adopted from the optimized photopolymerization conditions of UV Hg system in the previous study. The polymerizations were initiated using three different commercial photoinitiators; Irgacure 2959, Chivacure 300 and DMPA.

In the second task, WSPI was prepared through complexation of DMPA and methylated- β -cyclodextrin (M β CD). WSPI obtained was further tested with fourier transform infrared spectroscopy (FTIR), UV-Visible spectroscopy (UV-VIS), nuclear magnetic resonance (NMR) and solubility test.

PAAm and PNIPAAm hydrogels were prepared using synthesized WSPI via UV LED systems. Proper tuning on photopolymerization conditions has to be achieved for completion of monomer conversion. This includes UV time and photoinitiator concentrations. Various amount of photoinitiators concentration were added to the hydrogel formulation, varying from 1-5 wt.% relative to AAm/NIPAAm and UV times (2.5 - 20 minutes).

The scope of the work also included characterization of physical and structural properties of resulting bulk hydrogels. After polymerization, monomer conversion was determined. For further hydrogels characterization, swelling measurements in pure water as function of temperature were performed. Mechanical properties of cured hydrogels were studied by using oscillatory rheometer and texture analyzer. Using rheology, photopolymerized bulk PAAm and PNIPAAm hydrogels were characterized to determine their viscoelasticity, temperature responsiveness of the hydrogels and microstructure. Mesh sizes of the hydrogels were calculated from degree of swelling and rheology measurement.

- Kojima, K., Ito, M., Morishita, H., and Hayashi, N. A Novel Water-Soluble Photoinitiator for the Acrylic Photopolymerization Type Resist System. *Journal of Chemistry Material*, 1998, 10: 3429-3433.
- Wang, Y., Jiang, X., and Yin, J. A Water-Soluble Supramolecular-Structured Photoinitiator between Methylated b-Cyclodextrin and 2,2-Dimethoxy-2phenylacetophenone. *Journal of Applied Polymer Science*, 2007, 105: 3817-3823.
- Li, S. J., Wu, F. P., Li, M. Z., and Wang, E. J. Host/guest Complex of Me-B-CD/2,2-dimethoxy-2-phenyl acetophenone for Initiation of Aqueous Photopolymerization: Kinetics and Mechanism. *Polymer*, 2005, 46: 11934-11939.
- Ikemura, K. and Endo, T. A Review of The Development of Radical Photopolymerization Initiators Used for Designing Light-Curing Dental Adhesives and Resin Composites. *Dental Materials Journal* 2010, 29(5): 481-501.
- Jiang, X. and Yin, J. Photoinitiated Synthesis of Polymer Brush from Dendritic Photoinitiator Electrostatic Self-Assembly. *Journal of Chemical Communication*, 2005: 4927-4928
- Gulrez, S. K. H., Al-Assaf, S., and Phillips, G. O., *Hydrogels: Methods of Preparation, Characterization and Applications*, in *Progress in Molecular and Environmental Bioengineering - From Analysis and Modeling to Technology Applications*, P.A. Carpi, Editor 2011, Intech: United Kingdom. p. 117-150.
- Deligkaris, K., Tadele, T. S., Olthuis, W., and Berg, A. v. d. Hydrogel-Based Devices for Biomedical Applications. *Journal of Sensors and Actuators B* 2010, 147 765-774.
- Yang, Q., Adrus, N., Tomicki, F., and Ulbricht, M. Composites of Functional Polymeric Hydrogels and Porous Membranes. *Journals of Materials Chemistry*, 2011, 21: 2783-2811.
- Peppas, N. A., Bures, P., Leobandung, W., and Ichikawa, H. Hydrogels in Pharmaceutical Formulations. *European Journal of Pharmaceutics and Biopharmaceutics* 2000, 50: 27-46.
- 19. Paleos, G. A., What are Hydrogels?, 2012, Pittsburgh Plastics Manufacturing

- Ebara, M., Kotsuchibashi, Y., Uto, K., Aoyagi, T., Kim, Y.-J., Narain, R., Idota, N., and Hoffman, J. M., *Smart Hydrogels*, in *Smart Biomaterials*2014, Springer Japan. p. 9-65.
- Drury, J. L. and Mooney, D. J. Hydrogels for Tissue Engineering: Scaffold Design Variables and Applications. *Journal of Biomaterials* 2003, 24: 4337-4351.
- 22. Adrus, N. B., Stimuli-Responsive Hydrogels and Hydrogel Pore-Filled Composite Membranes, in Department of Chemistry2012, Universität Duisburg-Essen,: German. p. 1-148.
- Okay, O., *General Properties of Hydrogels*. Hydrogel Sensors and Actuators, ed. G. Gerlach and K.F. Arndt2009, Berlin: Springer-Verlag Berlin Heidelberg. 1-14.
- 24. Adrus, N. and Ulbricht, M. Rheological Studies on PNIPAAm Hydrogel Synthesis via in Situ Polymerization and on Resulting Viscoelastic Properties. *Journal of Reactive and Functional Polymers*, 2013, 73: 141-148.
- Gudeman, L. F. and Peppas, N. A. PH-Sensitive Membranes from Poly (Vinyl Alcohol) / Poly (Acrylic Acid) Interpenetrating Networks. *Journal of Membrane Science*, 1995, 107: 239-248.
- Fanger, C., Wack, H., and Ulbricht, M. Macroporous Poly(*N*isopropylacrylamide) Hydrogels with Adjustable Size "Cut-off" for the Efficient and Reversible Immobilization of Biomacromolecules. *Journal of Macromolecular Bioscience*, 2006, 6: 393-402.
- Neamtu, A. P. Chiriac, and Nita, L. E. Characterization of Poly(Acrylamide) as Temperature-Sensitive Hydrogel. *Journal of Optoelectronics and Advanced Materials*, 2006, 8(5): 1939 - 1943.
- Kumar, A., Srivastava, A., Galaev, I. Y., and Mattiasson, B. Smart Polymers: Physical Forms and Bioengineering Applications. *Journal of Progress Polymer Science*, 2007, 32: 1205–1237.
- Junzhang Song, Yu, R., Wang, L., Zheng, S., and Li, X. Poly(N-Vinylpyrrolidone)-Grafted Poly(N-Isopropylacrylamide) Copolymers: Synthesis, Characterization and Rapid Deswelling and Reswelling Behavior of Hydrogels. *Journal of Polymer* 2011, 52: 2340-2350.
- 30. Qin, L., He, X. W., Zhang, W., Li, W. Y., and Zhang, Y. K. Macroporous Thermosensitive Imprinted Hydrogel for Recognition of Protein by Metal

Coordinate Interaction. *Journal of Analytical Chemistry*, 2009, 81: 7206-7216.

- Zhang, X. Z., Wu, D.-Q., and Chu, C. C. Effect of the Crosslinking Level on the Properties of Temperature-Sensitive Poly(N-isopropylacrylamide) Hydrogels. *Journal of Polymer Science*, 2003, 41(Part B: Polymer Physics): 582-593.
- Muniz, E. C. and Geuskens, G. Compressive Elastic Modulus of Polyacrylamide Hydrogels and Semi-IPNs with Poly(Nisopropylacrylamide). *Macromolecules* 2001, 34: 4480-4484.
- Singh, D., Kuckling, D., Choudhary, V., Adler, H. J., and Koul, V. Synthesis and Characterization of Poly(Nisopropylacrylamide) Films By Photopolymerization. *Journal of Polymer Advance Technology* 2006, 17: 186-192.
- Tokuyama, H., Ishihara, N., and Sakohara, S. Effects of Synthesis-Solvent on Swelling and Elastic Properties of Poly(N-Isopropylacrylamide) Hydrogels. *European Polymer Journal* 2007, 43: 4975-4982.
- Xue, W., Champ, S., Huglin, M. B., and Jones, T. G. J. Rapid Swelling and Deswelling in Cryogels Of Crosslinked Poly(N-Isopropylacrylamide-Co-Acrylic). *European Polymer Journal* 2004, 40: 703-712.
- Zuidema, J. M., Rivet, C. J., Gilbert, R. J., and Morrison, F. A. A protocol for rheological characterization of hydrogels for tissue engineering strategies. J Biomed Mater Res B Appl Biomater, 2014, 102(5): 1063-73.
- 37. Dinu, M. V., Schwarz, S., Dinu, I. A., and Drăgan, E. S. Comparative rheological study of ionic semi-IPN composite hydrogels based on polyacrylamide and dextran sulphate and of polyacrylamide hydrogels. *Colloid and Polymer Science*, 2012, 290(16): 1647-1657.
- Abdurrahmanoglu, S., Can, V., and Okay, O. Design of high-toughness polyacrylamide hydrogels by hydrophobic modification. *Polymer*, 2009, 50(23): 5449-5455.
- Ramazani-Harandi, M. J., Zohuriaan-Mehr, M. J., Yousefi, A. A., Ershad-Langroudi, A., and Kabiri, K. Rheological determination of the swollen gel strength of superabsorbent polymer hydrogels. *Polymer Testing*, 2006, 25(4): 470-474.

- Adrus, N. and Ulbricht, M. The Rubber Elasticity of Poly(Nisopropylacrylamide) Hydrogel Networks. *Journal of Advanced Materials Research* 2013, 812: 210-215.
- 41. Webber, R. E. and Shull, K. R. Strain Dependence of the Viscoelastic Properties of Alginate Hydrogels. *Macromolecules* 2004, 37: 6153-6160.
- 42. Drury, J. L., Dennis, R. G., and Mooney, D. J. The tensile properties of alginate hydrogels. *Biomaterials*, 2004, 25(16): 3187-99.
- 43. Gildner, C. D., Lerner, A. L., and Hocking, D. C. Fibronectin matrix polymerization increases tensile strength of model tissue. *Am J Physiol Heart Circ Physiol*, 2004, 287(1): H46-53.
- 44. Liu, L., Wang, N., Xu, L., Yu, X., Zhang, R., and Wang, T. A novel method of determining wax cohesiveness by using a texture analyzer. *Journal of Texture Studies*, 2015: 1 33.
- 45. Hurler, J., Engesland, A., Poorahmary Kermany, B., and Škalko-Basnet, N. Improved texture analysis for hydrogel characterization: Gel cohesiveness, adhesiveness, and hardness. *Journal of Applied Polymer Science*, 2012, 125(1): 180-188.
- Wang, J. and Ugaz, V. M. Using In Situ Rheology to Characterize The Microstructure In Photopolymerized Polyacrylamide Gels for DNA Electrophoresis. *Electrophoresis*, 2006, 27: 3349-3358.
- 47. Righetti, P. G. and Gelfi, C. Electrophoresis Gel Media: The State of The Art. *Journal of Chromatography B*, 1996, 699: 63-75.
- Menter, P., Acrylamide Polymerization A Practical Approach, 2000, Bio Rad Laboratory.
- 49. Tanaka, T. Phase transitions in gels and a single polymer. *Polymer*, 1979, 20: 1404-1412.
- Lilian C. Lopergolo, Lugao, A. B., and Catalani, L. H. Direct UV Photocrosslinking of Poly(N-Vinyl-2-Pyrrolidone) (PVP) to Produce Hydrogels. *Polymer* 2003, 44 6217-6222.
- 51. Zubricky, G. A. and Middlemass, K., *LED Curing of Light -Curable Materials, Unraveling The Myths and Realities*, 2010.
- 52. Rob Karsten, Larson, B., and Miller, K. *Characterizing the Efficiency of UV LED Curing*. 23 August 2013; 1-8]. Available from:

http://www.phoseon.com/uploads/pdfs/characterizing-the-efficiency-of-uvled-curing.pdf.

- Bircher, P. F. UV-LED Curing in Industrial Printing. 2009; Available from: <u>http://www.swissphotonics.net/libraries.files/UV-</u> <u>LEDcuringinindustrialprinting2009-03-181.pdf</u>.
- 54. Jaranilla-Tran, E., *UV/LED Curing in Graphic Arts Applications*, R. Group, Editor 2012.
- 55. Ye, Q., Park, J., Topp, E., and Spencer, P. Effect of photoinitiators on the in vitro performance of a dentin adhesive exposed to simulated oral environment. *Dent Mater*, 2009, 25(4): 452-8.
- 56. Giorgi, M. C. C., Aguiar, F. H. B., Soares, L. E. S., Martin, A. A., Liporoni, P. C. S., and Paulillo, L. A. M. S. Does an Additional UV LED Improve The Degree of Conversion and Knoop Hardness of Light-Shade Composite Resins? *European Journal of Dentistry*, 2012, 6: 396-401.
- Mucci, V. and Vallo, C. Efficiency of 2,2-Dimethoxy-2-phenylacetophenone for the Photopolymerization of Methacrylate Monomers in Thick Sections. *Journal of Applied Polymer Science*, 2012, 123: 418-425
- 58. Kumar, C. V., Gururaj, M., Paul, J., Krishnaprasada, L., and Divyashree, R. A Comparative Evaluation of Curing Depth and Compressive Strength of Dental Composite Cured with Halogen Light Curing Unit and Blue Light Emitting Diode: An In Vitro Study. *Journal of Contemporary Dental Practice*, 2012, 13(6): 834-837.
- 59. Jansen, B., *Identification of Unknown Photo-initiators in Offset UV-inks and Prints*, 2012, University of Amsterdam.
- 60. Halliday, R., Key Benefits of Next-Gen UV LED Technology, 2010, LUMEX.
- 61. Richards, A. *Reflected Ultraviolet Imaging for Forensics Applications*. 2010.
- 62. Ortyl, J. and Popielarz, R. New Photoinitiators for Cationic Polymerization. *Journal of Polimery*, 2012, 57: 7-8.
- Fairbanks, B. D., Schwartz, M. P., Bowman, C. N., and Anseth, K. S. Photoinitiated Polymerization of PEG-Diacrylate with Lithium Phenyl-2,4,6-Trimethylbenzoylphosphinate: Polymerization Rate and Cytocompatibility. *Journal of Biomaterials* 2009, 30: 6702-6707.
- 64. Liska, R. Photoinitiators with Functional Groups. V. New Water-Soluble Photoinitiators Containing Carbohydrate Residues and Copolymerizable

Derivatives Thereof. Journal of Polymer Science: Part A: Polymer Chemistry, 2002, 40: 1504-1518.

- 65. Chitect. Chivacure 300 Photoinitiator for UV Radiation Curing Systems.
 2010; 1-2]. Available from: <u>http://www.foresight-chem.com/upload/file/170%20CHIVACURE%20300%20DS%20US.pdf</u>.
- Visconti, M. and Cattaneo, M. A Highly Efficient Photoinitiator for Water-Borne UV-Curable Systems. *Progress in Organic Coatings*, 2000, 40: 243-251.
- Kaczmarek, H., Gałka, P., and Kowalonek, J. Influence of a photoinitiator on the photochemical stability of poly(methyl methacrylate) studied with fourier transform infrared spectroscopy. *Journal of Applied Polymer Science*, 2010, 115(3): 1598-1607.
- Farahat, M. S. and Nikles, D. E. On the UV Curability and Mechanical Properties of Novel Binder Systems Derived from Poly(ethyleneterephthalate) (PET) Waste for Solventless Magnetic Tape Manufacturing, Methacrylated Oligoesters. *Journal of Macromolecule Material Engineering*, 2002, 287(5): 353-362.
- Qiu, J. and Wei, J. Water-Soluble and Polymerizable Thioxanthone Photoinitiator Containing Imidazole. *Journal of Applied Polymer Science*, 2014: 1-6.
- Ortega, J. A. T., Sulfonation/Sulfation Processing Technology for Anionic Surfactant Manufacture, 2012, intechopen: Universidad de La Salle Colombia. p. 269-294.
- 71. Foster, N. C., Sulfonation and Sulfation Processes, 1997, Chemithon. p. 1-36.
- 72. Kojima, K., Itoh, M., Morishita, H., and Hayashi, N. Characterization of Water-Soluble Acrylic Resist Using a Novel Photoinitiator. *Journal of Photopolymer Science and Technology*, 1998, 11(1): 165-170.
- 73. Srinivasan, U., Liepmann, D., and Howe, R. T. Microstructure to Substrate Self-Assembly using Capillary Forces. *Journal of Microelectromechanical Systems*, 2001, 10(1).
- 74. Szillat, F., Schmidt, B. V. K. J., Hubert, A., Barner-Kowollik, C., and Ritter,
 H. Redox-Switchable Supramolecular Graft Polymer Formation via Ferrocene–Cyclodextrin Assembly. *Journal of Macromolecular Rapid Communications*, 2014: 1-8.

- 75. Alupei, I. C., Alupei, V., and Ritter, H. Cyclodextrins in Polymer Synthesis: Photoinitiated Free-Radical Polymerization of *N*-Isopropylacrylamide in Water Initiated by a Methylated β-Cyclodextrin/2-Hydroxy-2-methyl-1phenylpropan-1-one Host/Guest Complex. *Macromolecular Rapid Communications*, 2002, 23: 55-58.
- McKenzie, M., Betts, D., Suh, A., Bui, K., Kim, L. D., and Cho, H. Hydrogel-Based Drug Delivery Systems for Poorly Water-Soluble Drugs. *Molecules*, 2015, 20(11): 20397-408.
- 77. Ahadian, S., Sadeghian, R. B., Salehi, S., Ostrovidov, S., Bae, H., Ramalingam, M., and Khademhosseini, A. Bioconjugated Hydrogels for Tissue Engineering and Regenerative Medicine. *Bioconjug Chem*, 2015, 26(10): 1984-2001.
- Domingues, R. M., Silva, M., Gershovich, P., Betta, S., Babo, P., Caridade, S. G., Mano, J. F., Motta, A., Reis, R. L., and Gomes, M. E. Development of Injectable Hyaluronic Acid/Cellulose Nanocrystals Bionanocomposite Hydrogels for Tissue Engineering Applications. *Bioconjug Chem*, 2015, 26(8): 1571-81.
- Ganji, F., Farahani, S. V., and Farahani, E. V. Theoretical Description of Hydrogel Swelling: A Review. *Iranian Polymer Journal*, 2010, 19(5): 375-398.
- Nicolson, P. C. and Vogt, J. Soft Contact Lens Polymers: An Evolution. Biomaterials 2001, 22: 3273-3283.
- Tranoudis, I. and Efron, N. Tensile properties of soft contact lens materials. Contact Lens & Anterior Eye, 2004, 27(4): 177-91.
- Sun, J.-Y., Zhao, X., Illeperuma, W. R. K., Chaudhuri, O., Oh, K. H., David J. Mooney, Vlassak, J. J., and Suo, Z. Highly Stretchable and Tough Hydrogels. *Journal of Nature*, 2012, 489: 133-136.
- Beebe, D. J., Moore, J. S., Bauer, J. M., Yu, Q., Liu, R. H., Devadoss, C., and Jo, B.-H. Functional Hydrogel Structures for Autonomous Flow Control Inside Microfluidic Channels. *Journal of Nature*, 2000, 404: 588-590.
- Peppas, N. A., Hilt, J. Z., Khademhosseini, A., and Langer, R. Hydrogels in Biology and Medicine: From Molecular Principles to Bionanotechnology. *Journal of Advance Materials*, 2006, 18: 1345-1360.

- 85. Lee, K. Y. and Mooney, D. J. Hydrogels for Tissue Engineering. *Journal of Chemical Reviews*, 2001, 101(7).
- Vlierberghe, S. V., Dubruel, P., and Schacht, E. Biopolymer-Based Hydrogels As Scaffolds for Tissue Engineering Applications: A Review. *Biomacromolecules* 2011, 12: 1387-1408.
- 87. Hutmacher, D. W. Scafolds in Tissue Engineering Bone and Cartilage. *Journal of Biomaterials* 2000, 21: 2529-2543.
- Gong, J. P., Katsuyama, Y., Kurokawa, T., and Osada, Y. Double-Network Hydrogels with Extremely High Mechanical Strength. *Advanced Materials*, 2003, 15: 1155-1158.
- Okay, O. and Oppermann, W. Polyacrylamide-Clay Nanocomposite Hydrogels: Rheological and Light Scattering Characterization. *Journal of Macromolecules*, 2007, 40: 3378-3387.
- Zhang, X. Z. and Chu, C. C. Preparation of Thermosensitive PNIPAAm Hydrogels with Superfast Response. *Journal of Chemistry Communication*. 2004: 350-351.
- Wu, X. S., Hoffman, A. S., and Yager, P. Synthesis and Characterization of Thermally Reversible Macroporous Poly (*N*-lsopropylacrylamide) Hydrogels. *Journal of Polymer Science: Part A Polymer Chemistry*, 1992, 30: 2121-2129.
- Kwok, A. Y., Qiao, G. G., and Solomon, D. H. Synthetic Hydrogels. 1.
 Effects of Solvent on Poly(Acrylamide) Networks. *Polymer*, 2003, 44(20): 6195-6203.
- 93. Adrus, N. and Ulbricht, M. Rheological Studies on PNIPAAm Hydrogel Synthesis via In Situ Polymerization and on Resulting Viscoelastic Properties. *Reactive and Functional Polymers*, 2013, 73(1): 141-148.
- 94. Alonso, R. C. B., Brandt, W. C., Souza-Junior, E. J. C., Puppin-Rontani, R. M., and Sinhoreti, M. A. C. Photoinitiator concentration and modulated photoactivation: influence on polymerization characteristics of experimental composites. *Applied Adhesion Science* 2014, 2(10): 1-11.
- 95. Zhang, X. Z., Xu, X. D., Cheng, S. X., and Zhuo, R. X. Strategies to Improve The Response Rate of Thermosensitive PNIPAAm Hydrogels. *Journal of Soft Matter*, 2008, 4: 385-391.

- Magin, C. M., Finlay, J. A., Clay, G., Callow, M. E., Callow, J. A., and Brennan, A. B. Antifouling Performance of Cross-linked Hydrogels: Refinement of an Attachment Model. *Biomacromolecules*, 2011, 12: 915-922.
- Kristi S. Anseth, Bowman, C. N., and Brannon-Peppas, L. Mechanical properties of hydrogels and their experimental determination. *Biomoterials*, 1996, 17: 1647-1657.
- Calvet, D., Wong, J. Y., and Giasson, S. Rheological Monitoring of Polyacrylamide Gelation: Importance of Cross-Link Density and Temperature. *Journal of Macromolecules* 2004, 37: 7762-7771.
- 99. Nesrinne, S. and Djamel, A. Synthesis, characterization and rheological behavior of pH sensitive poly(acrylamide-co-acrylic acid) hydrogels. *Arabian Journal of Chemistry*, 2013.
- Hamidi, M., Azadi, A., and Rafiei, P. Hydrogel Nanoparticles in Drug Delivery. Advanced Drug Delivery Reviews 2008, 60: 1638-1649.
- Huan, S., Liu, G., Han, G., Cheng, W., Fu, Z., Wu, Q., and Wang, Q. Effect of Experimental Parameters on Morphological, Mechanical and Hydrophobic Properties of Electrospun Polystyrene Fibers. *Materials*, 2015, 8(5): 2718-2734.
- 102. Costa, A. M. S. and Mano, J. F. Extremely strong and tough hydrogels as prospective candidates for tissue repair – A review. *European Polymer Journal*, 2015, 72: 334-364.