

Synthesis and Luminescence Properties of Doped Magnesium Boro-Tellurite Ceramics

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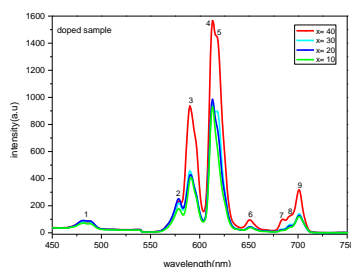
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Graphical abstract



Abstract

Glass has been widely utilized in the field of lighting, telecommunication and spectroscopy. Boro-tellurite is one of the suitable glasses used for solid state lighting and laser application. The investigation on the luminescence properties of rare earth doped ceramic is rarely used due to the opacity. In this paper boro-tellurite prepared in ceramic can show the better luminescence with the less advantage. The aim of this paper is to present the effect and advantages in luminescence results of boro-tellurite ceramics doped with the constant amount of rare earth. Doped magnesium boro-tellurite with Eu^{3+} and Dy^{3+} ceramic have been prepared using solid state reaction method with the compositions of $x\text{TeO}_2-(70-x)\text{B}_2\text{O}_3-30\text{MgO}$ with $10 \leq x \leq 40$, and have been doped with Eu_2O_3 (1mol%) and Dy_2O_3 (1mol%). The characterizations of the samples have been investigated by means of X-Ray diffraction, Raman, Infrared and Photoluminescence spectroscopy. From the X-ray diffraction results, two phases are assigned to MgTe_2O_5 and $\text{Mg}_2\text{B}_2\text{O}_5$. Raman spectroscopy showed strong bands observed in the vicinity of 140, 175, 220, 266, 332, 403, 436, 646, 694, 723, 757 and 806 cm^{-1} . FTIR spectra showed bands located in the range between $400-800 \text{ cm}^{-1}$ are assigned to the bending mode of Te-O-Te, TeO_3 and TeO_4 . In the range of $800-1400 \text{ cm}^{-1}$, the bands are associated with B-O, B-O-B, BO_3 and BO_4 bonds. The emission transition $^5\text{D}_0-^7\text{F}_2$ corresponded to the red emission (612 nm) was found to be the most intense in all the Eu^{3+} -doped magnesium boro-tellurite ceramics.

Keywords: Magnesium boro-tellurite ceramic, luminescence, doped Eu^{3+} and Dy^{3+}

Abstrak

Kaca telah digunakan secara meluas dalam bidang pencahayaan, telekomunikasi dan spektroskopi. Boro-tellurite adalah satu kaca yang sesuai digunakan dalam pencahayaan keadaan pepejal dan aplikasi laser. Penyelidikan luminesens bagi nadir bumi dop seramik adalah jarang digunakan disebabkan oleh kelegapan. Dalam kertas ini boro-tellurite seramik menunjukkan luminesens yang baik dengan kelebihan kurang. Matlamat kertas ini adalah untuk menunjukkan kesan dan kelebihan dalam keputusan luminesens bagi seramik boro-tellurite dop dengan amoun nadir bumi yang tetap. Dop magnesium boro-tellurite dengan Eu^{3+} and Dy^{3+} seramik disediakan dengan menggunakan kaedah tindakbalas keadaan pepejal yang berkompposisi $x\text{TeO}_2-(70-x)\text{B}_2\text{O}_3-30\text{MgO}$ dengan $10 \leq x \leq 40$, dan dop Eu_2O_3 (1mol%) and Dy_2O_3 (1mol%). Pencirian sampel disiasat oleh spektroskopi pembelauan sinar-X, Raman, infra merah and Fotoluminesens. Dari keputusan XRD, dua fasa dilihat sebagai MgTe_2O_5 and $\text{Mg}_2\text{B}_2\text{O}_5$. Spektroskopi Raman menunjukkan jalur kuat dilihat sekitar 140, 175, 220, 266, 332, 403, 436, 646, 694, 723, 757 dan 806 cm^{-1} . Spektra FTIR menunjukkan jalur terletak dalam lingkungan $400-800 \text{ cm}^{-1}$ adalah mod lentur bagi Te-O-Te, TeO_3 and TeO_4 . Dalam lingkungan $800-1400 \text{ cm}^{-1}$, jalur yang berkaitan adalah B-O, B-O-B, BO_3 and BO_4 . Peralihan pancaran $^5\text{D}_0-^7\text{F}_2$ yang merujuk kepada pancaran merah (612 nm) didapati menjadi yang paling tinggi dalam semua seramik Eu^{3+} -dop magnesium boro-tellurite.

Kata kunci: Seramik magnesium boro-tellurite, luminesens, dop Eu^{3+} and Dy^{3+}

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1.0 INTRODUCTION

Ceramic based on boro-tellurite host matrices doped with rare earth have tremendous applications for lasers, optical amplifier, photo sensitivity, optical storage, and bio-ceramics materials [1-8]. Currently, a great deal of research has been focused on rare earth

(RE) doped glasses owing to their extensive applications [9-10]. But, the investigation on the luminescence properties of rare earth doped ceramic is rarely used due to the opacity and hence, limited their applications. Eu^{3+} ion doped sodium-aluminum-tellurite (Eu^{3+} : NAT) was prepared in opaque ceramics compound [11].

The host material with low phonon energy that can reduce the non-radiative loss, play important role for obtaining highly efficient luminescent properties via multiphonon relaxations and thus achieve strong luminescence [12]. In this study, tellurite oxide based has been utilized due to their desirable physical properties, such as high refractive index, excellent infrared transmittance and high dielectric constant, good chemical durability and low melting temperature. On the other hand, borate has important physical properties such as high transparency, low melting point, high thermal stability, good rare earth ions solubility, resistance to

2.0 EXPERIMENTAL

High purity of H_3BO_3 (99.99%), TeO_2 (99.99%), MgO (99.99%), Eu_2O_3 (99.99%) and Dy_2O_3 (99.99%) were used as the raw materials. Two types of sample were prepared for the study; magnesium boro-tellurite ceramic and magnesium boro-tellurite with Eu^{3+} and Dy^{3+} ceramic. The samples with the compositions of $x\text{TeO}_2-(70-x)\text{B}_2\text{O}_3-30\text{MgO}$ with $10 \leq x \leq 40$ mol%, doped with Eu_2O_3 (1mol%) and Dy_2O_3 (1mol%) have been prepared using solid state reaction method. Analytical grade reagents of H_3BO_3 , TeO_2 , MgO , Eu_2O_3 and Dy_2O_3 powders in appropriate amounts (mol%) were thoroughly mixed in agate mortar. The mixtures were pressed into pellets using hydraulic press with 10 ton/cm^2 of pressure. Pelletized samples were then heat treated at 650°C for 6 hours. The dried pellet was grounded into fine powder for further characterizations.

Several experiments have been setup to determine the physical nature and characteristics of the prepared samples. The examination of the structure was examined using X-ray diffraction (XRD) using Siemens Diffraction D500 diffractometer with $\text{CuK}\alpha$ radiation [13]. Fourier transform infrared spectroscopy (FTIR) was examined using Perkin-Elmer spectrometer (Spectrum 100) [13]. Raman spectroscopy was examined using Bruker model RFS 100/S FT-Raman spectrometer. The emission spectra were obtained using Jasco Fluorescence spectrophotometer, Model FP8000 series, at room temperature.

3.0 RESULTS AND DISCUSSION

X-ray diffraction technique was used to identify the crystalline phase of the doped magnesium boro-tellurite with Eu^{3+} and Dy^{3+} ceramics. The XRD patterns, Figure 1, show the influence of varying composition of the $x\text{TeO}_2-(70-x)\text{B}_2\text{O}_3-30\text{MgO}$ with $10 \leq x \leq 40$ mol%. The x-ray diffractogram showed well-defined peaks which indicate the crystalline and phase formation of the synthesized compounds. There are two main phases; namely MgTe_2O_5 (ICDD: 01-073-3922) and $\text{Mg}_2\text{B}_2\text{O}_5$ (ICDD: 01-073-2107). The intensity increased with B_2O_3 and decrease with TeO_2 , ranging from $x = 10$ to 40. MgTe_2O_5 became a dominant phase in the sample followed by $\text{Mg}_2\text{B}_2\text{O}_5$ phase. The majority phase of MgTe_2O_5 is an evident that the TeO_2 contribute to the formation of the phase. The percentage of the crystalline phase can be estimated by comparing the highest peak heights summarized in the Table 1. From the estimation of Table 1, it can be concluded that the MgTe_2O_5 was the dominant phase.

chemicals and can enhance mechanical durability [3]. Addition of alkali oxide (MgO) as network modifier has also been employed since the utilization of modifier doped with boro-tellurite based are rarely reported.

The aim of this work is to prepare doped magnesium boro-tellurite ceramics via solid state reaction route and to study the behavior through X-Ray Diffraction (XRD), Infrared (IR), Raman and Photoluminescence (PL) spectroscopic for both materials. The concentration of Eu^{3+} and Dy^{3+} will remain constant.

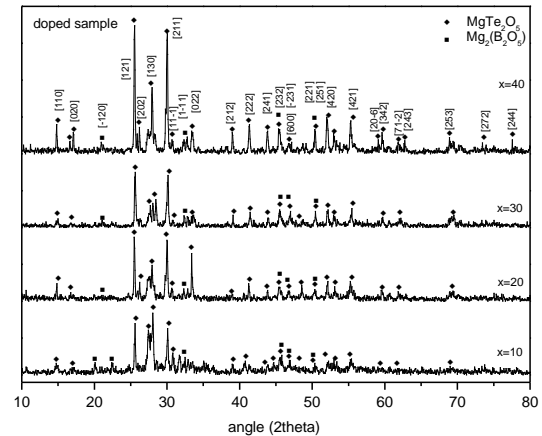


Figure 1 X-ray diffraction pattern of $x\text{TeO}_2-(70-x)\text{B}_2\text{O}_3-30\text{MgO}$ doped Eu^{3+} and Dy^{3+} ceramic ($10 \leq x \leq 40$ mol%)

Table 1 The estimation of percentage of the crystalline phase of $x\text{TeO}_2-(70-x)\text{B}_2\text{O}_3-30\text{MgO}$ doped Eu^{3+} and Dy^{3+} ceramic ($10 \leq x \leq 40$ mol%)

x	MgTe_2O_5 phase (%)	$\text{Mg}_2\text{B}_2\text{O}_5$ phase (%)
10	58.03	41.97
20	62.94	37.05
30	67.94	32.06
40	76.00	23.99

Raman spectrum of $x\text{TeO}_2-(70-x)\text{B}_2\text{O}_3-30\text{MgO}$ are presented in Figure 2. Raman spectroscopy show the strong band are observed at 140, 175, 220, 266, 332, 403, 436, 646, 694, 723, 757 and 806 cm^{-1} . Raman spectrum shows that besides the expected TeO_4 , TeO-Te , and BO_3 (peak 7 to 12) existent, the vibrations corresponding to the metal ions are also present in ceramics (peak 1 to 6), respectively.

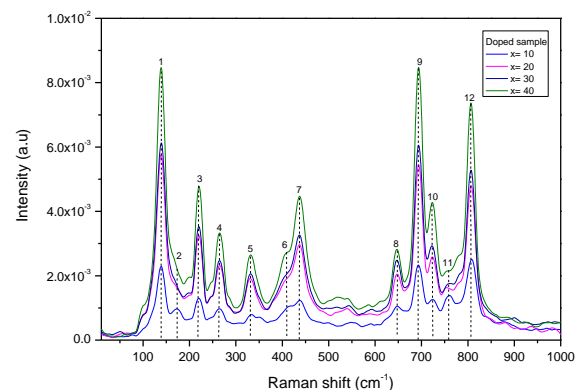


Figure 2 Raman spectra of $x\text{TeO}_2-(70-x)\text{B}_2\text{O}_3-30\text{MgO}$ doped Eu^{3+} and Dy^{3+} ceramic ($10 \leq x \leq 40$ mol%)

The infrared spectrum of $x\text{TeO}_2-(70-x)\text{B}_2\text{O}_3-30\text{MgO}$ with $10 \leq x \leq 40$ mol% in the range $1400-4000\text{cm}^{-1}$ are presented in the Figure 3. There are four bands in this range in which the peaks represent OH- and H- units are at around 3215cm^{-1} , 2920cm^{-1} and 2261cm^{-1} . The peaks observed between 3375cm^{-1} and 4000cm^{-1} are the hydroxyl groups due to OH- stretching and the peaks observed between $(2800-2900)\text{cm}^{-1}$ are the hydrogen bond [9]. Based on the Figure 3, the peaks around 1500cm^{-1} are due to the B-O stretching of BO_3 units which are observed and the band became smaller ($x=40$), indicating a decrease in TeO_2 amounts and increasing B_2O_3 .

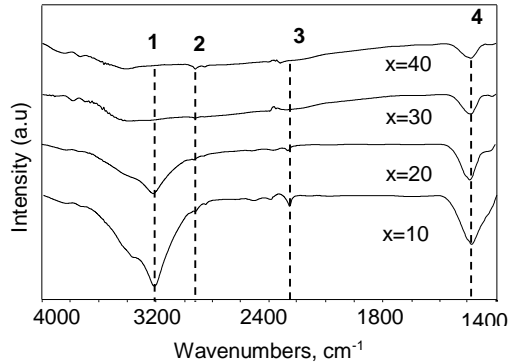


Figure 3 IR spectra of $x\text{TeO}_2-(70-x)\text{B}_2\text{O}_3-30\text{MgO}$ doped Eu^{3+} and Dy^{3+} ceramic ($10 \leq x \leq 40$ mol%)

In Figure 4, the IR spectra of $x\text{TeO}_2-(70-x)\text{B}_2\text{O}_3-30\text{MgO}$ with $10 \leq x \leq 40$ mol% in the range $(800-1400)\text{cm}^{-1}$ is presented. The bands located in the range between $(400-800)\text{cm}^{-1}$ are assigned to the bending mode of Te-O-Te , TeO_3 and TeO_4 . In the range of $(800-1400)\text{cm}^{-1}$, the B-O bond of B-O-B , BO_3 and BO_4 also appeared. The bands located around $(1331-1362)\text{cm}^{-1}$ are assigned to BO_3 units [15, 21] and in the range between $(881-1200)\text{cm}^{-1}$ which is the BO_4 units [10, 14, 16]. The bands positions around 431cm^{-1} , 564cm^{-1} , $610-680\text{cm}^{-1}$ and $700-780\text{cm}^{-1}$ showed the Te-O-Te , TeO_2 and TeO_3 respectively. All of these assignments are in good agreement with the reported literature [10, 17-18]. In Figure 4, the band gradually increased with TeO_2 . The intensity of the band in the range of $431-780\text{cm}^{-1}$ increased gradually with TeO_2 and the intensity of the band slightly decreased with B_2O_3 in the range of $881-1362\text{cm}^{-1}$.

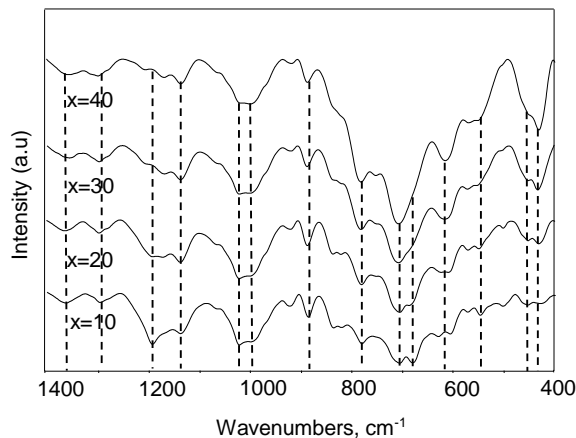


Figure 4 IR spectra of $x\text{TeO}_2-(70-x)\text{B}_2\text{O}_3-30\text{MgO}$ doped Eu^{3+} and Dy^{3+} ceramic ($10 \leq x \leq 40$ mol%)

The emission spectrum of doped $x\text{TeO}_2-(70-x)\text{B}_2\text{O}_3-30\text{MgO}$ with $10 \leq x \leq 40$ mol% is shown in Figure 5. The emission transitions

$^5\text{D}_0 \rightarrow ^7\text{F}_0$, $^5\text{D}_0 \rightarrow ^7\text{F}_0$, $^5\text{D}_0 \rightarrow ^7\text{F}_1$, $^5\text{D}_0 \rightarrow ^7\text{F}_2$, $^5\text{D}_0 \rightarrow ^7\text{F}_2$, $^5\text{D}_0 \rightarrow ^7\text{F}_3$, $^5\text{D}_0 \rightarrow ^7\text{F}_4$, $^5\text{D}_0 \rightarrow ^7\text{F}_4$ and $^5\text{D}_0 \rightarrow ^7\text{F}_4$ corresponded to the band position at 484.58 , 578.17 , 589.89 , 612.82 , 618.38 , 651.23 , 683.48 , 691.53 and 700.85nm , respectively. It can be noted that a series of Eu^{3+} characteristic emission lines are between 540nm and 780nm , which includes the most prominent peak at 612nm and corresponding to the $^5\text{D}_0 \rightarrow ^7\text{F}_J$ ($J=0, 1, 2, 3, 4$) transitions [19]. The luminescence spectrum was found to be lowest for the sample $x=10$ followed by $x=20, 30$ and 40 . The $^5\text{D}_0 \rightarrow ^7\text{F}_2$ transition is more intense than the other transitions as reported in the other literatures [20-22].

The emission peaks of Dy^{3+} were observed at 484.58nm and 578.17nm (peak 1 and 2) similar with the other reported values [23]. Figure 5 shows the emission peaks of Eu^{3+} are more dominant with high intensity compared the peaks of Dy^{3+} . The Dy^{3+} emission peaks are not present as dominant peak which indicates that Dy^{3+} acts as trap centers that cause long afterglow characteristics, rather than the luminescent centers in the host lattice [24].

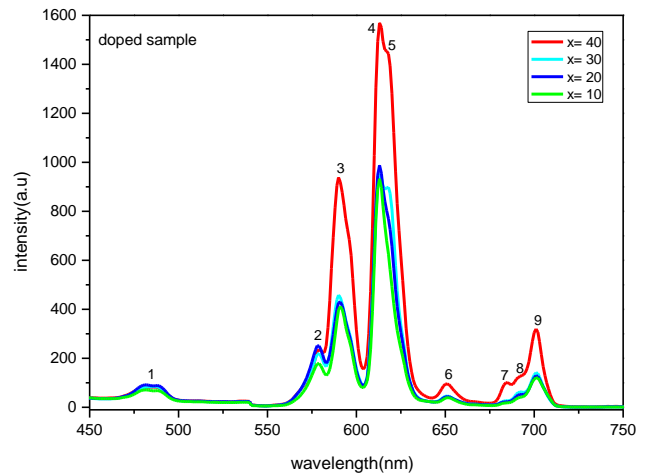


Figure 5 Emission spectrum of $x\text{TeO}_2-(70-x)\text{B}_2\text{O}_3-30\text{MgO}$ doped Eu^{3+} and Dy^{3+} ceramic ($10 \leq x \leq 40$ mol%)

4.0 CONCLUSION

The samples of $x\text{TeO}_2-(70-x)\text{B}_2\text{O}_3-30\text{MgO}$ with $10 \leq x \leq 40$ mol% doped with Eu_2O_3 (1 mol%) and Dy_2O_3 (1 mol%) have been prepared using solid state route. The structural studies of the samples have been investigated using X-ray diffraction (XRD), Infrared and Raman spectroscopy. The XRD profiles show that MgTe_2O_5 became a dominant phase in the sample followed by $\text{Mg}_2\text{B}_2\text{O}_5$ phase. The IR spectrum shows two series of band regions which obtained in $4000-1400\text{cm}^{-1}$ and the second region of $1400-400\text{cm}^{-1}$. The ceramics doped with Eu^{3+} shows a bright red emission at 612nm which belongs to the electric dipole ($^5\text{D}_0 \rightarrow ^7\text{F}_2$) transition of Eu^{3+} ions. There are two groups of emission at 484nm and 578nm occurring in the emission spectrum of the ceramics, which probably can be applied for white LEDs application.

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