

TENSILE, BARRIER, AND DEGRADATION PROPERTIES OF LOW DENSITY
POLYETHYLENE/EPOXIDIZED NATURAL RUBBER NANOCOMPOSITE
FILMS

SEYED AHMAD ATTARAN

A thesis submitted in fulfilment of the
requirements for the award of the degree of
Doctor of Philosophy (Polymer Engineering)

Faculty of Chemical Engineering
Universiti Teknologi Malaysia

AUGUST 2015

To my beloved father, mother, sister, and brother

ACKNOWLEDGEMENT

First and foremost, I praise God, the almighty for providing me this opportunity and granting me the capability to proceed successfully. This thesis appears in its current form due to the assistance and guidance of several people. I would like to acknowledge forever my supervisor, Prof. Dr. Azman Hassan, for his kind encouragement, guidance and inspiration throughout this research. Very special thanks goes out to my co-supervisor, Assoc. Prof. Dr. Mat Uzir Wahit, without whose guidance, motivation, suggestions and encouragement, I would not have considered a graduate career in psychological research.

I must also acknowledge to all lecturers from department of Polymer Engineering who have directly and indirectly contributed towards the success of my research project. I wish also express my deep appreciation to all technicians of laboratories of Polymer Engineering and group of AMTEC members for their helps and assistantships in my research. My appreciation also goes out to my fellow colleagues in Enhanced Polymer Research Group (EnPRO) and friends.

Last but not the least, I would like to deep thank to my family specially my parents and my sister for their continues loves and emotional encouragements, for always believing in me and never failing to provide me any supports. My thanks goes to my dear twin brother for his emotional supports and accompany during my study in Malaysia.

ABSTRACT

The demand for the development of high-performance packaging films and the equally growing environmental awareness have called for intensive research in the field of eco-friendly films with good mechanical and barrier properties. The present study investigates the use of epoxidized natural rubber (ENR) and organo modified montmorillonite (OMMT) in the development of low density polyethylene (LDPE) nanocomposites films. The films were successfully prepared by first melt blending at 190 °C in a twin-screw extruder and then blown via single screw machine. Linear low density polyethylene-grafted-maleic anhydride (LLDPE-g-MA), was used as compatibilizer to obtain better dispersion of nanoclay in the system. The nanocomposite films were prepared at different amounts of rubber from 2.5-10 wt%. The obtained nanocomposites were analyzed for tensile, thermal, morphology, gas permeability and degradation properties. The study of tensile properties of films was done via Lloyd 2.5 kN according to ASTM D882 in machine direction. Thermal analysis was carried out via differential scanning calorimeter (DSC) and thermogravimetric analysis (TGA). Atomic force microscope (AFM) was used to probe the morphology of phases, and X-ray diffraction (XRD) was applied to examine the interlayer distance of OMMT. Chemical characterization and interaction between materials were studied by Fourier transform infrared spectroscopy (FTIR). The barrier property of films was determined by constant pressure/variable volume type permeation cell. Degradation and biodegradation were studied via Q-SUN Xenon test chamber and soil burial test, respectively. The study has shown the presence of OMMT enhanced the tensile modulus and barrier property of LDPE film by 46% and 80%, respectively. In addition, incorporation of ENR improved the Young's modulus and barrier properties of compatibilized nanocomposite films. Improvement in the degradation onset temperature ($T_{10\%}$) of nanocomposite films was observed with addition of 6wt% OMMT and ENR contents. The intercalation of OMMT regarding addition of LLDPE-g-MA into nanocomposites was confirmed by XRD. FTIR analysis has shown a formation of (C-O) ester bond; the driving force for intercalation between MA group of LLDPE-g-MA and epoxy group of ENR. Significant degradation took place through addition of OMMT and ENR as samples were highly fragile and broken. Biodegradation of films increased through addition of ENR. The most significant finding from this research is the development of eco-friendly nanocomposite film formulation with enhanced barrier properties. Based on tensile modulus, barrier and biodegradation properties, compatibilized nanocomposites film containing 7.5 wt% ENR (LDPE/OMMT/LLDPE-g-MA/ENR_{7.5}) was the optimum formulation. The unique combination of tensile modulus, barrier and biodegradation properties for nanocomposite films has shown that this nanocomposite is a potential candidate for a variety of food packaging applications.

ABSTRAK

Permintaan yang sentiasa meningkat terhadap pembangunan filem pembungkusan yang berprestasi tinggi di samping perkembangan tentang kesedaran alam sekitar yang setara telah menarik perhatian untuk menggiatkan penyelidikan dalam pengeluaran filem yang bersifat mesra alam dengan sifat-sifat mekanikal dan rintangan yang baik. Penyelidikan ini mengkaji kemungkinan untuk menggunakan getah asli terepoksi (ENR) dan organo montmorilonit yang diubahsuai (OMMT) secara bersama dalam membangunkan filem berasaskan polietilena berketumpatan rendah (LDPE). Filem telah berjaya disediakan melalui pengadunan leburan pada suhu 190°C dengan menggunakan penyemperitan skru berkembar diikuti oleh peniupan filem melalui mesin skru tunggal. Polietilena linear berketumpatan rendah tercantum malik anhidrida (LLDPE-g-MA) digunakan sebagai penserasi untuk mendapatkan penyerakan tanah liat bersaiz nano yang lebih baik di dalam sistem. Filem komposit nano telah disediakan dengan kandungan getah yang berbeza (2.5-10) peratus berat. Komposit nano yang diperolehi dianalisis untuk sifat-sifat regangan, terma, morfologi, kebolehtelapan gas dan penguraian. Kajian regangan filem dilakukan menggunakan Lloyd 2.5 kN mengikut ASTM D882 di dalam arah mesin (MD). Sifat-sifat terma dan kestabilan telah dilakukan melalui analisa kalorimetri pengimbasan pembezaan (DSC) dan analisa termogravimetri (TGA). Mikroskop daya atom (AFM) telah digunakan untuk menyiasat fasa morfologi dan pembelauan sinar-X (XRD) telah digunakan untuk memeriksa jarak di antara lapisan OMMT. Ciri-ciri kimia dan interaksi di antara bahannya dikaji menggunakan Fourier infra merah (FTIR). Sifat rintangan filem ditentukan melalui sistem tekanan malar dan ketelapan sel jenis isipadu bolehubah. Penguraian dan bio-penguraian masing-masing telah diperiksa melalui ujian kebuk Q-SUN Xenon dan ujian penanaman dalam tanah. Kajian menunjukkan kehadiran OMMT meningkatkan modulus regangan dan sifat rintangan filem LDPE sebanyak 46.25% dan 80%. Masing-masing tambahan pula, penambahan ENR meningkatkan modulus Young serta sifat rintangan filem komposit nano berpenserasi. Peningkatan dalam suhu pemulaan degradasi ($T_{-10\%}$) filem komposit nano telah diperhatikan dengan penambahan 6 peratus berat OMMT dan kandungan ENR. Interkalasi OMMT disebabkan penambahan LLDPE-g-MA dalam filem komposit nano telah disahkan oleh XRD. Analisis FTIR menunjukkan pembentukan ikatan C-O ester, daya penggerak untuk interkalasi antara kumpulan MA daripada LLDPE-g-MA dan kumpulan epoksi daripada ENR. Penguraian telah berlaku secara ketara melalui penambahan OMMT dan ENR dengan sampel menjadi terlalu rapuh dan mudah patah. Bio-penguraian filem telah meningkat dengan penambahan ENR. Penemuan yang terpenting dalam kajian ini ialah pembangunan filem komposit nano yang mesra alam dengan sifat rintangan yang dipertingkatkan. Berdasarkan sifat modulus regangan, rintangan dan bio-pegeuraian, filem komposit nano yang mengandungi 7.5 peratus berat ENR (LDPE/OMMT/LLDPE-g-MA/ENR_{7.5}) merupakan formulasi yang optimum. Gabungan unik sifat modulus regangan, rintangan dan bio-pegeuraian untuk filem komposit nano telah menunjukkan komposit nano ini merupakan pilihan untuk pelbagai pembungkusan makanan.

TABLE OF CONTENTS

CHAPTER	TITLE	PAGE
	DECLARATION	ii
	DEDICATION	iii
	ACKNOWLEDGEMENT	iv
	ABSTRACT	v
	ABSTRAK	vi
	TABLE OF CONTENTS	vii
	LIST OF TABLES	xv
	LIST OF FIGURES	xvii
	LIST OF ABBREVIATIONS	xxv
	LIST OF SYMBOLS	xxx
	LIST OF APPENDICES	xxxii
1	INTRODUCTION	1
1.1	Background	1
1.2	Problem Statements	4
1.3	Objectives of Research	6
1.4	Scopes of Research	6
1.5	Thesis Outline	8
2	LITERATURE REVIEW	9
2.1	Polyolefin	9
2.1.1	Polyethylene	10
2.2.1	Low-Density Polyethylene	13
2.3	Compatibilizer	15
2.4	Nanoclay	23

2.4.1	Montmorillonite	24
2.5	Natural Rubber	30
2.6	Epoxidized Natural Rubber	32
2.7	Classification of Polymer Blend	35
2.7.1	Rubber-Plastic Blends	36
2.7.1.1	ENR/Thermoplastic Blend	37
2.7.1.2	LDPE/Rubber Blend	40
2.8	Polyolefin Nanocomposite	42
2.8.1	Classification of Nanocomposite	43
2.8.1.1	Elastomeric Nanocomposite	43
2.8.1.2	Thermoset Nanocomposite	45
2.8.1.3	Thermoplastic Nanocomposite	45
2.8.2	Polyethylene Nanocomposites	46
2.8.3	Rubber/LDPE Nanocomposite	49
2.9	Methods Used for the Synthesis of Polymer/Clay Nanocomposites	50
2.9.1	In-situ Polymerization	50
2.9.2	Intercalation from Polymer Solution	52
2.9.3	Melt Compounding	53
2.10	Polyolefin Film	55
2.10.1	Plastic Film	55
2.11	Degradable Polymer	56
2.11.1	Classification of Degradable Polymers	58
2.11.1.1	Biodegradable Polymer	59
2.11.1.2	Photo-degradable Polymer	60
2.12	Mechanism of Degradation	61
2.12.1	Biodegradation	61
2.12.2	Photo-Degradation	63
2.13	Application of Biodegradable Polymer	64
2.14	Factors Affecting Photo-Degradation	65
2.14.1	Wavelength Sensitivity of Photo-degradation	65
2.14.2	Influence of Temperature and Moisture	66
2.15	Gas Barrier Property	67
2.15.1	Parameters Affecting the Barrier Property	71

2.15.1.1 Polarity	72
2.15.1.2 Crystallinity	73
2.15.1.3 Liquid Crystallinity	73
2.15.1.4 Stiff Molecules	74
2.15.1.5 Close Packing	74
2.15.1.6 Cross-Linking	74
2.15.1.7 Fillers and Blend	74
2.15.1.8 Orientation	75
2.16 Methods Used to Improve Barrier Property	75
2.16.1 Lamination	76
2.16.2 Co-Extrusion	76
2.16.3 Polymer Layered Nanocomposite	77
2.17 Factors Affecting Permeability	81
2.17.1 Permeation	82
2.17.2 Polymer Characteristics	82
2.17.3 Penetrant	83
2.17.4 Environment	83
2.17.4.1 Temperature	83
2.17.4.2 Humidity	84
3 METHODOLOGY	86
3.1 Materials	86
3.1.1 Low-Density Polyethylene	86
3.1.2 Organo Modified Montmorillonite	87
3.1.3 Linear Low Density Polyethylene Grafted Maleic Anhydride	87
3.1.4 Epoxidized Natural Rubber (ENR)	88
3.2 Designing and Composition of Materials	89
3.3 Sample Preparation	91
3.3.1 Nanocomposite Film Preparation	91
3.3.1.1 Two Roll Mill	91
3.3.1.2 Extrusion	92
3.3.1.3 Blown Film Extrusion	92
3.3.2 Test Specimens Preparation	93

3.4	Sample Characterization	93
3.4.1	Tensile Test	93
3.4.2	Fourier Transform Infrared Spectroscopy	94
3.4.3	Characterization Test	94
3.4.3.1	X-Ray Diffraction	94
3.4.4	Morphological Test	95
3.4.4.1	Atomic Force Microscopy	95
3.4.5	Thermal Behavior Test	96
3.4.5.1	Differential Scanning Calorimetry	96
3.4.5.2	Thermogravimetric Analysis	97
3.4.6	Oxygen Permeability Test	97
3.4.6.1	Constant-Pressure System and Soap Bubble Flow Meter	97
3.4.7	Degradation Test	99
3.4.7.1	Q-SUN Xenon Test Chamber	99
3.4.8	Biodegradation Test	100
3.4.8.1	Soil Burial Test	100
4	RESULTS AND DISCUSSION	101
4.1	Tensile Properties	101
4.1.1	Tensile Properties of LDPE Film	102
4.1.1.1	Effect of OMMT	102
4.1.1.2	Effect of ENR Content	103
4.1.2	Tensile properties of LDPE/ENR blend	106
4.1.2.1	Effect of OMMT	106
4.1.2.2	Effect of LLDPE-g-MA	108
4.1.3	Tensile properties of LDPE/LLDPE-g-MA/ENR	111
4.1.3.1	Effect of OMMT	111
4.1.4	Tensile Properties of LDPE/OMMT	114
4.1.4.1	Effect of LLDPE-g-MA	114
4.1.4.2	Effect of ENR Content	115
4.1.5	Tensile Properties of LDPE/OMMT/ENR	118
4.1.5.1	Effect of LLDPE-g-MA	118
4.1.6	Tensile Properties of LDPE/OMMT/LLDPE-g-MA	120

4.1.6.1 Effect of ENR Content	120
4.2 Thermal Properties of Films via Differential Scanning Calorimeter	122
4.2.1 Thermal Properties of Neat LDPE	123
4.2.1.1 Effect of OMMT	123
4.2.1.2 Effect of ENR Content	125
4.2.2 Thermal Properties of LDPE/ENR Blends	128
4.2.2.1 Effect of OMMT	128
4.2.2.2 Effect of LLDPE-g-MA	129
4.2.3 Thermal Behavior of LDPE/LLDPE-g-MA/ENR	131
4.2.3.1 Effect of OMMT	131
4.2.4 Thermal Behavior of LDPE/OMMT	132
4.2.4.1 Effect of LLDPE-g-MA	132
4.2.4.2 Effect of ENR Content	135
4.2.5 Thermal Behavior of LDPE/OMMT/ENR	137
4.2.5.1 Effect of LLDPE-g-MA	137
4.2.6 Thermal Behavior of LDPE/OMMT/LLDPE-g-MA	138
4.2.6.1 Effect of ENR Content	138
4.3 Thermal Properties of Films via Thermogravimetric Analysis	140
4.3.1 Thermal Stability of Neat LDPE	141
4.3.1.1 Effect of OMMT	141
4.3.1.2 Effect of ENR Content	143
4.3.2 Thermal Properties of LDPE/ENR Blends	145
4.3.2.1 Effect of OMMT	145
4.3.2.2 Effect of LLDPE-g-MA	146
4.3.3 Thermal Properties of LDPE/LLDPE-g-MA/ENR	147
4.3.3.1 Effect of OMMT	147
4.3.4 Thermal Properties of LDPE/OMMT	148
4.3.4.1 Effect of LLDPE-g-MA	148
4.3.4.2 Effect of ENR Content	150
4.3.5 Thermal Properties of LDPE/OMMT/ENR	151
4.3.5.1 Effect of LLDPE-g-MA	151
4.3.6 Thermal Properties of LDPE/OMMT/LLDPE-g-MA	152

4.3.6.1 Effect of ENR	152
4.4 Structural Characterization by X-Ray Diffraction	153
4.4.1 Characterization of LDPE Film	154
4.4.1.1 Effect of OMMT	154
4.4.2 Characterization of LDPE/OMMT	155
4.4.2.1 Effect of LLDPE-g-MA	155
4.4.2.2 Effect of ENR Content	157
4.4.3 Characterization of LDPE/OMMT/ENR	
Nanocomposite Films	159
4.4.3.1 Effect of LLDPE-g-MA	159
4.4.4 Characterization of LDPE/OMMT/LLDPE-g-MA	
Nanocomposite Films	161
4.4.4.1 Effect of ENR	161
4.5 Chemical Structure Analysis by Fourier Transform	
Infrared Spectroscopy	163
4.5.1 FTIR Study of Neat LDPE Film	163
4.5.1.1 Effect of OMMT	163
4.5.1.2 Effect of ENR Content	165
4.5.2 FTIR Study of LDPE/ENR Blends	168
4.5.2.1 Effect of OMMT	168
4.5.2.2 Effect of LLDPE-g-MA	170
4.5.3 FTIR Study of LDPE/OMMT	172
4.5.3.1 Effect of LLDPE-g-MA	172
4.5.3.2 Effect of ENR Content	173
4.5.4 FTIR Study of Compatibilized LDPE/OMMT	
Nanocomposite Film	175
4.5.4.1 Effect of ENR	175
4.6 Morphological Analysis by Atomic Force Microscope	176
4.6.1 Morphology of Neat LDPE Film	176
4.6.1.1 Effect of OMMT	177
4.6.1.2 Effect of ENR Content	178
4.6.2 Morphology of LDPE/ENR Film	180
4.6.2.1 Effect of OMMT	180
4.6.2.2 Effect of LLDPE-g-MA	182

4.6.3 Morphology of LDPE/LLDPE-g-MA/ENR Film	183
4.6.3.1 Effect of OMMT	183
4.6.4 Morphology of LDPE/OMMT Nanocomposite Film	185
4.6.4.1 Effect of LLDPE-g-MA	185
4.6.4.2 Effect of ENR Content	185
4.6.5 Morphology of LDPE/OMMT/ENR Nanocomposite Film	188
4.6.5.1 Effect of LLDPE-g-MA	188
4.6.6 Morphology of Compatibilized LDPE Nanocomposite	189
4.6.6.1 Effect of ENR Content	189
4.7 Gas Permeability of Films	191
4.7.1 Permeability of Neat LDPE	192
4.7.1.1 Effect of OMMT	192
4.7.1.2 Effect of ENR Content	192
4.7.2 Permeability of LDPE/ENR Blends	194
4.7.2.1 Effect of OMMT	194
4.7.3 Permeability of LDPE/LLDPE-g-MA/ENR	196
4.7.3.1 Effect of OMMT	196
4.7.4 Permeability of LDPE/OMMT	197
4.7.4.1 Effect of LLDPE-g-MA	197
4.7.4.2 Effect of ENR	197
4.7.5 Permeability of LDPE/OMMT/ENR	199
4.7.5.1 Effect of LLDPE-g-MA	199
4.7.6 Permeability of LDPE/OMMT/LLDPE-g-MA	200
4.7.6.1 Effect of ENR Content	200
4.7.7 Balanced barrier and mechanical properties	201
4.8 Degradation	202
4.8.1 Photo-Degradation	202
4.9 Biodegradation	204
4.9.1 Biodegradation of Neat LDPE	204
4.9.1.1 Effect of OMMT	205
4.9.1.2 Effect of ENR	205
4.9.2 Biodegradation of LDPE/ENR	206

4.9.2.1 Effect of OMMT	206
4.9.2.2 Effect of LLDPE-g-MA	206
4.9.3 Biodegradation of LDPE/LLDPE-g-MA/ENR	209
4.9.3.1 Effect of OMMT	209
4.9.4 Biodegradation of LDPE/OMMT	211
4.9.4.1 Effect of LLDPE-g-MA	211
4.9.4.2 Effect of ENR	212
4.9.5 Biodegradation of LDPE/OMMT/ENR	212
4.9.5.1 Effect of LLDPE-g-MA	212
4.9.6 Biodegradation of LDPE/OMMT/LLDPE-g-MA	213
4.9.6.1 Effect of ENR	213
5 CONCLUSION AND RECOMMENDATIONS	214
5.1 Conclusion	214
5.1.1 Effect of OMMT	215
5.1.2 Effect of LLDPE-g-MA	215
5.1.3 Effect of ENR	216
5.2 Recommendations	217
REFERENCES	219
Appendix A	260

LIST OF TABLES

TABLE NO.	TITLE	PAGE
2.1	Approaches for achieving miscible blend and to compatible the separated blend phases	16
2.2	Potential commercial uses of ENR	35
2.3	Plastic production in kilograms per capita in 1993	56
2.4	Solar irradiances	66
2.5	Oxygen transmit rate of polymers	85
3.1	Properties of LDPE	87
3.2	Properties of OMMT	87
3.3	Characteristic features of compatibilizer	88
3.4	Blend formulations	90
4.1	Tensile properties of neat LDPE, LDPE nanocomposites, and LDPE/ENR blends	102
4.2	Thermal properties of neat LDPE, LDPE/OMMT, and LDPE/ENR blends	124
4.3	Effect of OMMT on thermal properties of LDPE/ENR blends	128
4.4	Effect of LLDPE-g-MA on thermal properties of LDPE/ENR blends	129
4.5	Effect of OMMT on thermal properties of LDPE/LLDPE-g-MA/ENR	132
4.6	Thermal properties of un-compatibilized nanocomposite blended with different amount of rubber	135

4.7	Effect of LLDPE-g-MA on thermal properties of LDPE/OMMT/ENR nanocomposites	137
4.8	Thermal properties of compatibilized nanocomposites blended with different amount of rubber	138
4.9	TGA results for neat LDPE film and incorporated with OMMT, and different amount of ENR	141
4.10	Effect of OMMT on thermal property of LDPE/ENR blends	146
4.11	Effect of LLDPE-g-MA on thermal property of LDPE/ENR blends	147
4.12	Effect of OMMT on thermal property of LDPE/LLDPE-g-MA/ENR blends	148
4.13	Effect of LLDPE-g-MA and rubber contents on thermal property of nanocomposite	149
4.14	Effect of LLDPE-g-MA on thermal property of LDPE/OMMT/ENR nanocomposites	152
4.15	Effect of ENR on thermal property of compatibilized nanocomposite	152
4.16	Effect of LLDPE-g-MA on d-spacing of nanocomposite	155
4.17	Effect of ENR on d-spacing of LDPE/OMMT nanocomposite	159
4.18	Effect of LLDPE-g-MA on d-spacing of LDPE/OMMT/ENR nanocomposites	160
4.19	Effect of ENR on d-spacing of compatibilized nanocomposite	162
4.20	Characteristic bands for OMMT	164
4.21	FTIR observation of ENR ₂₅	168
4.22	Gas permeability values of various LDPE films	192
4.23	Biodegradation of PE films by means of weight loss after certain time	204

LIST OF FIGURES

FIGURE NO.	TITLE	PAGE
2.1	Plastics consumption in U.S.A in 1997	10
2.2	PE structure	11
2.3	Common packaging plastics in U.S. municipal solid waste stream, 1998	12
2.4	The polyethylene family	13
2.5	Diagrammatic of supposed conformation of some compatibilizers include (a) diblock, (b) triblock, (c) multi-graft, (d) single-graft copolymers at the interface of a heterogeneous polymer blend	18
2.6	The modified schematic representation of the dispersion process of the nanoclay in the polymer matrix with the aid of compatibilizer	19
2.7	Maleic anhydride reaction with hydroxyl group of clay	21
2.8	Clay morphology of PE/MMT with different dispersing additive	22
2.9	Nanoclay structure	26
2.10	Possibility of arrangement of clay in matrix	27
2.11	Substructures of nanocomposite	28
2.12	Intercalated, intercalated-and-flocculated and exfoliated structure	28
2.13	Global production and consumption of natural rubber	31
2.14	Structure of rubber, (a) NR, cis-1,4-polyisoprene, (b) formation of peroxy formic acid, and (c) ENR production	34

2.15	Scheme of in-situ polymerization	51
2.16	Scheme of swelling	51
2.17	Scheme of different steps for polymer solution	52
2.18	Intercalation from polymer solution	53
2.19	Scheme of different steps for melt compounding	54
2.20	Scheme of melt intercalation	54
2.21	Classification of environmentally degradable bio-based polymeric materials	57
2.22	Norish type I and type II reactions	61
2.23	Biodegradation process of biodegradable polymers	62
2.24	Mechanism of photo-degradation	64
2.25	An anticipated effects of climatic change factors on the degradation of materials by solar ultraviolet radiation	67
2.26	General mechanism of gas or vapor permeation through a plastic film	69
2.27	Formation of tortuous path in (a) conventional composites (b) polymer layered silicate nanocomposites	70
2.28	Factors affect to enhance the barrier property	72
2.29	The blocking of a diffusing oxygen molecule caused by hydrogen bonds	72
2.30	Typical structure of the random liquid crystalline copolyester, vectra® B and N correspond, respectively, to the monomers: p-Hydroxy benzoic acid (HBA) and 2-hydroxy-6-naphthoic acid(HNA)	73
2.31	Polymer matrix region (grey area bonded to a MMT particle)	75
2.32	Schematic representation of the gas path in a nanocomposites and in a pure polymer	78
2.33	Effect of delamination on tortuosity and aspect ratio of nanoplatelets	79

2.34	Permeability of blend as a function of morphology	80
3.1	Scheme of LLDPE-g-MA	88
3.2	Blown film extrusion machine schematic	92
3.3	X-ray diffraction principal	94
3.4	Scheme of a typical permeation cell through continues flow method	98
3.5	Constant-pressure system and soap bubble flow meter schematic	99
3.6	Q-SUN Xenon test chamber	100
4.1	Scheme of cross-linking formation of ENR via ring opened epoxide group	104
4.2	Effect of OMMT on Young's modulus of LDPE/ENR	106
4.3	Effect of OMMT on tensile strength of LDPE/ENR	107
4.4	Effect of OMMT on EB of LDPE/ENR	108
4.5	Effect of LLDPE-g-MA on Young's modulus of LDPE/ENR	109
4.6	Effect of LLDPE-g-MA on tensile strength of LDPE/ENR	110
4.7	Effect of LLDPE-g-MA on EB of LDPE/ENR	110
4.8	Proposed grafting mechanism in LDPE/LLDPE-g-MA/ENR	111
4.9	Effect of OMMT on Young's modulus of LDPE/LLDPE-g-MA/ENR	112
4.10	Effect of OMMT on tensile strength of LDPE/LLDPE-g-MA/ENR	113
4.11	Effect of OMMT on EB of LDPE/LLDPE-g-MA/ENR	113
4.12	Proposed interactions in LDPE nanocomposite	114
4.13	Effect of ENR on Young's modulus of LDPE/OMMT nanocomposite	116
4.14	Effect of ENR on tensile strength of LDPE/OMMT nanocomposites	117

4.15	Effect of ENR on EB of LDPE/OMMT nanocomposites	117
4.16	Effect of LLDPE-g-MA on Young's modulus of LDPE/OMMT/ENR nanocomposites	119
4.17	Effect of LLDPE-g-MA on tensile strength of LDPE/OMMT/ENR nanocomposites	119
4.18	Effect of LLDPE-g-MA on EB of LDPE/OMMT/ENR nanocomposites	120
4.19	Effect of ENR on Young's modulus of compatibilized nanocomposite	121
4.20	Effect of ENR on tensile strength of compatibilized nanocomposite	121
4.21	Effect of ENR on EB of compatibilized nanocomposite	122
4.22	Effect of OMMT on LDPE melting point	124
4.23	Effect of ENR on LDPE crystalline point	125
4.24	Effect of ENR on melting point of LDPE blend	127
4.25	Effect of rubber content on crystalline point of LDPE blend	127
4.26	Effect of LLDPE-g-MA on melting point of LDPE/ENR blends	130
4.27	Effect of LLDPE-g-MA on crystalline point of LDPE/ENR blends	131
4.28	Effect of LLDPE-g-MA on melting point of nanocomposite	134
4.29	Effect of LLDPE-g-MA on crystalline point of nanocomposite	134
4.30	Effect of ENR on melting point of nanocomposite	136
4.31	Effect of ENR on crystalline point of nanocomposite	136
4.32	Effect of ENR on melting point of compatibilized nanocomposite	139
4.33	Effect of ENR on crystalline point of compatibilized nanocomposite	140

4.34	Thermal property of LDPE film by incorporation of OMMT	142
4.35	Thermal property of LDPE film by incorporation of various amounts of rubber	144
4.36	Effect of LLDPE-g-MA on thermal property of LDPE nanocomposite	150
4.37	Effect of different amount of ENR on thermal stability of LDPE/OMMT nanocomposite	151
4.38	Effect of different amount of ENR on thermal stability of LDPE/OMMT/LLDPE-g-MA nanocomposite	153
4.39	XRD of OMMT, LDPE/OMMT and compatibilized LDPE nanocomposite	154
4.40	Reaction between clay and MA	156
4.41	Effect of ENR content on XRD of LDPE nanocomposite	157
4.42	XRD diffractograms for compatibilized nanocomposite blended with different amounts of ENR	162
4.43	FTIR study of OMMT	164
4.44	Comparing the FTIR of LDPE and LDPE/OMMT	165
4.45	FTIR spectrum of raw ENR ₂₅	166
4.46	Effect of different amount of ENR on FTIR spectrum of LDPE	167
4.47	Effect of OMMT on FTIR spectrum of LDPE/ENR _{2.5}	169
4.48	Effect of OMMT on FTIR spectrum of LDPE/ENR ₁₀	169
4.49	Effect of LLDPE-g-MA on FTIR spectrum of LDPE/ENR _{2.5}	171
4.50	Effect of LLDPE-g-MA on FTIR spectrum of LDPE/ENR ₁₀	171
4.51	Comparing the FTIR of LDPE and LDPE nanocomposite	172
4.52	Effect of 2.5 wt% ENR on FTIR of LDPE/OMMT nanocomposite	174

4.53	Effect of 10 wt% ENR on FTIR of LDPE/OMMT nanocomposite	174
4.54	Effect of ENR content on FTIR of compatibilized nanocomposite	175
4.55	AFM images for LDPE film	177
4.56	AFM images for LDPE/OMMT nanocomposite	178
4.57	AFM images for LDPE/ENR _{2.5} blend film	179
4.58	AFM images for LDPE/ENR ₅ blend film	179
4.59	AFM images for LDPE/ENR _{7.5} blend film	179
4.60	AFM images for LDPE/ENR ₁₀ blend film	180
4.61	Effect of OMMT on AFM images of LDPE/ENR _{2.5} blend	181
4.62	Effect of OMMT on AFM images of LDPE/ENR ₁₀ blend	181
4.63	AFM images for LDPE/LLDPE-g-MA/ENR _{2.5} blend	183
4.64	AFM images for LDPE/LLDPE-g-MA/ENR ₁₀ blend	183
4.65	Effect of OMMT on AFM images of LDPE/LLDPE-g-MA/ENR _{2.5}	184
4.66	Effect of OMMT on AFM images of LDPE/LLDPE-g-MA/ENR ₁₀	184
4.67	AFM images for compatibilized LDPE/OMMT nanocomposite	185
4.68	Effect of 2.5 wt% ENR on AFM images of LDPE/OMMT nanocomposite	186
4.69	Effect of 10 wt% ENR on AFM images of LDPE/OMMT nanocomposite	186
4.70	Effect of LLDPE-g-MA on AFM images of LDPE/OMMT/ENR _{2.5}	188
4.71	Effect of LLDPE-g-MA on AFM images of LDPE/OMMT/ENR ₁₀	189
4.72	Effect of 2.5 wt% ENR on AFM images of compatibilized nanocomposite	190

4.73	Effect of 5 wt% ENR on AFM images of compatibilized nanocomposite	190
4.74	Effect of 7.5 wt% ENR on AFM images of compatibilized nanocomposite	191
4.75	Effect of 10 wt% ENR on AFM images of compatibilized nanocomposite	191
4.76	Proposed mechanism of the strain-induced crystallization in LDPE/ENR film during the blown film process; (a) before stretching, (b) at an extension of more than 40% the stretched chains of LDPE act as the nucleus of crystallites (yellow part) and LDPE crystallites induced the crystallization of ENR (pink part) and (c) the extension at 200%, (MD: machine direction, TD: transverse direction)	194
4.77	Effect of OMMT on gas permeability of LDPE/ENR blends	195
4.78	Effect of LLDPE-g-MA on gas permeability of LDPE/ENR blends	195
4.79	Effect of OMMT on gas permeability of LDPE/LLDPE-g-MA/ENR	196
4.80	Effect of ENR on gas permeability of un-compatibilized nanocomposite	198
4.81	Effect of LLDPE-g-MA on gas permeability of LDPE/OMMT/ENR	200
4.82	Effect of ENR on gas permeability of compatibilized nanocomposite	201
4.83	Balanced nanocomposites properties based on tensile modulus and permeability	202
4.84	Effect of ENR on biodegradation of LDPE	205
4.85	Effect of LLDPE-g-MA on biodegradation of LDPE/ENR _{2.5} blend	207
4.86	Effect of LLDPE-g-MA on biodegradation of LDPE/ENR ₅ blend	207
4.87	Effect of LLDPE-g-MA on biodegradation of LDPE/ENR _{7.5} blend	208

4.88	Effect of LLDPE-g-MA on biodegradation of LDPE/ENR ₁₀ blend	208
4.89	Effect of OMMT on biodegradation of LDPE/ENR _{2.5} blend	209
4.90	Effect of OMMT on biodegradation of LDPE/ENR ₅ blend	210
4.91	Effect of OMMT on biodegradation of LDPE/ENR _{7.5} blend	210
4.92	Effect of OMMT on biodegradation of LDPE/ENR ₁₀ blend	211
4.93	Effect of LLDPE-g-MA on biodegradation of LDPE/OMMT	212

LIST OF ABBREVIATIONS

ASTM	- American society for testing and materials
ATM	- Atmosphere
BUR	- Blow up ratio
CEC	- Cation exchange capacity
CI	- Carbonyl index
CPE	- Chlorinated polyethylene
CPP	- Chlorinated polypropylene
DCP	- Dicumyl peroxide
DMA	- Dynamic mechanical analysis
DSC	- Differential scanning calorimeter
EB	- Elongation at break
EMA	- Ethylene-co-methyl acrylate
ENR	- Epoxidized natural rubber
EUV	- Extreme ultraviolet
FTIR	- Fourier transform infrared spectroscopy
FUV	- Far ultraviolet
GDI	- Gasoline direct injection
HDPE	- High density polyethylene
ICI	- Imperial chemical industries

IIR	-	Butyl rubber
iPP	-	Isotactic polypropylene
IR	-	Ionizing radical
ISO	-	International organization for standardization
KH-MT	-	Alkylamine modified MMT
LCPs	-	Liquid crystal polymers
LDPE	-	Low density polyethylene
LDPE-g-MA	-	Low density polyethylene grafted maleic anhydride
LLDPE	-	Linear low density polyethylene
LLDPE-g-MA	-	Linear low density polyethylene grafted maleic anhydride
LUV	-	Low ultraviolet
MD	-	Machine direction
mEPDM	-	Maleated Ethylene-propylene-diene Rubber
mEPR	-	Ethylene–propylene random copolymer grafted with maleic anhydride
MFI	-	Melt flow index
MMT	-	Montmorillonite
mPOE	-	Maleated ethylene-octene copolymer
MRB	-	Malaysian Rubber Board
mSEBS	-	Ethene-co-butenelstyrene grafted with maleic anhydride
MUV	-	Middle ultraviolet
NO _X	-	Oxides of nitrogen
NR	-	Natural rubber
NBR	-	Acrylonitrile butadiene rubber

NUV	-	Near ultraviolet
OMMT	-	Organic modified montmorillonite
OTR	-	Oxygen transmission rate
PA	-	Polyamide
PA6	-	Polyamide 6
PA12	-	Polyamide 12
PC	-	Polycarbonate
PCL	-	Poly caprolactone
PCNs	-	Polymer/clay nanocomposites
PDMS	-	Poly (dimethyl siloxane) rubber
PE	-	Polyethylene
PEA	-	Poly(ethylene coacrylic acid)
PE-g-MA	-	Polyethylene grafted maleic anhydride
Pema-Zn	-	Poly(ethylene-co-methacrylic acid)
PEO	-	Poly(ethylene oxide)
PET	-	Poly(ethylene terephthalate)
PETG	-	Polyethylene terephthalate glycol
PEU	-	Poly(ester-urethane)
PHAs	-	Poly(hydroxy alkanoates)
PHBV	-	Poly(3-hydroxybutyrate-co-3-hydroxyvalerate) with 12% mole of hydroxyvalerate content
Ph-PP	-	Phenolic modified PP
PLA	-	Poly(lactic acid)
PLLA	-	Poly(L-lactic acid)
PLSNs	-	Polymer layered silicate nanocomposite

PLST	-	Plasticized starch
PMMA	-	Poly(methyl methacrylate)
POO	-	Peroxide
POOH	-	Hydro peroxide
PP-b-PPG	-	Polypropylene monobutyl ether
PPC	-	Poly(propylene carbonate)
PPCH	-	Polypropylene-clay hybrid
PS	-	Polystyrene
Psi	-	Pound per square inch
PU	-	Polyurethane
PVA	-	Poly(vinyl alcohol)
PVDC	-	Poly(vinylidene chloride)
RR	-	Reclaimed rubber
SBR	-	Styrene butadiene rubber
SEM	-	Scanning electron microscopy
Si-69	-	Silane coupling agent
SO _X	-	Oxides of sulphur
SR	-	Silicone rubber
SUV	-	Supper ultraviolet
T ₋₁₀	-	Onset temperature, evaluated as 10% weight loss temperature
T ₋₅	-	The 5% weight loss temperature
T _C	-	Crystallization temperature
TD	-	Transverse direction
TEM	-	Transition electron microscopy

T _g	-	Glass transition temperature
TGA	-	Thermogravimetric analysis
T _m	-	Melting temperature
TPOs	-	Thermoplastic olefins
TPU	-	Thermoplastic polyurethane
TPVs	-	Thermoplastic vulcanizates
UFNBRP	-	Ultra-fine fully-vulcanized acrylonitrile butadiene rubber powder
UHMWPE	-	Ultra high molecular weight polyethylene
ULDPE	-	Ultra low density polyethylene
UP	-	Unsaturated polyester
USCC	-	U.S. composting council
UTS	-	Ultimate tensile strength
UV	-	Ultraviolet
UVA	-	Ultraviolet A
UVB	-	Ultraviolet B
UVC	-	Ultraviolet C
VOC	-	Volatile organic compounds
VUV	-	Vacuum ultraviolet
X _C	-	Degree of crystallinity
XRD	-	X-ray diffraction

LIST OF SYMBOLS

Al	-	Aluminum
AL ³⁺	-	Aluminum cation
Å	-	Angstrom
CC	-	Cubic centimeter
CH ₄	-	Methane
CO ₂	-	Carbon dioxide
°C	-	Degree of centigrade
d	-	Diameter
eV	-	Electron volt
°F	-	Fahrenheit
hrs	-	Hours
J	-	Joule
kg	-	Kilogram
kN	-	Kilo Newton
L	-	Length
lb	-	Pound
m	-	Meter
m ²	-	Square meter
mA	-	Miliampere

Mg	-	Magnesium
mm	-	Millimeter
min	-	Minute
MPa	-	Mega Pascal
nm	-	Nano meter
rpm	-	Revolution per minute
s	-	Second
Si^{+4}	-	Silica cation
V	-	Volume
P	-	Permeability
ΔH_f	-	Enthalpy of fusion for 100% crystalline
ΔH_m	-	Melting enthalpy
Δm	-	Flow of a gas
Δp	-	Pressure difference
$\tan \delta$	-	Mechanical loss factor
θ	-	Degree
μ	-	Micron
μm	-	Micro meter

LIST OF APPENDICES

APPENDIX	TITLE	PAGE
A	List of publications	260

CHAPTER 1

INTRODUCTION

1.1 Background

Environmental concerns and petroleum resource limitations have drawn ever increasing attention over the past two decades to the development of polymers from renewable resources, particularly in the field of plastic industries. Characteristic features of petroleum-produced plastics such as light weight, high strength, chemical inertness, and low cost have led to their replacement of normal packaging plastics which include glass, metals, and ceramics. However, despite their versatilities and performances, their uses in the food packaging industry, in particular, are limited due to inherent permeability to gases and non-degradability that lead to many environmental problems due to their disposal including harm to the environment ecosystem, water supplies and sewer systems as well as to rivers and streams.

Food packaging is one of the largest growing sectors within the plastic packaging market domain. Food needs to be packaged with a strong material that can keep it safe from contamination from the environment as well as be durable enough to keep the food safe from the time of packaging up to delivery to the consumer. The primary function of food packaging is to preserve the food from penetration by harmful or dangerous substances (oxygen, moisture, light, microbes, etc.) during storage and distribution.

Traditional disposal methods include recycling, incineration and burying in landfills. Recycling has remained the most viable method of waste disposal mainly since it reduces pollution and damage to the environment. However, recycling of waste into new products after processing requires prior collection and segregation, which increases the cost of processing and thus the price of recycled products (Roy *et al.*, 2011). In addition, these products tend to have inferior long-term properties even after proper stabilization, thereby limiting their market ability.

Non-renewability, instability and rising prices due to depletion of petroleum resources, and inherent permeability of product are among the main reasons that motivate researchers to seek alternative packaging materials in place of fossil fuels to overcome the drawbacks associated with conventional plastics. Drawbacks include renewability and/or biodegradability (Gross and Kalra, 2002, Imam *et al.*, 2008), and permeability (Arora and Padua, 2010).

Organic modified layered silicate clay such as montmorillonite (OMMT) has received much attention due to low cost, availability and eco-friendly substance. Polymer/OMMT offers enormous improvement in the range of physical and engineering properties at low filler loading. Modifying the properties of various composites is greatly contributed to some characteristic parameters such as filler loading, size and shape and affinity towards matrix material. Nanosized layered structure has a large surface area providing sufficient interfacial regions in polymer nanocomposites for enhancement in mechanical and barrier properties of polymer matrix (Durmus *et al.*, 2007, Gatos and Karger-Kocsis, 2007, Goettler *et al.*, 2007, Khalili *et al.*, 2013). In addition, OMMT as nano-particles in packaging industry has been focused recently due to low cost, attainability and availability, process ability and good performance (Majeed *et al.*, 2013).

Bio-polymers are potential materials for packaging applications as they have many advantages over synthetic polymers including renewability, recyclability and cost effectiveness (Okhamafe, 2009). Regarding advancement in nanocomposite science and technology, compounding of polymers with bio-polymer/nanoclay is a technique that can complement the drawbacks of conventional polymers and provide

promise of a new product featuring stronger, higher barrier, short life, disposal, and environmentally compatible packaging materials (Yu *et al.*, 2006). It is intended that the use of bio-polymer/nanoclay based materials will contribute to sustainability through a reduction in environmental hazards and a reduction in the accumulation associated with disposal of synthetic polymer-based packaging materials and plastic litter, both in the sea and on land.

It is discovered that incorporation of OMMT caused to increase and accelerate the degradation of polymer (Qin *et al.*, 2003, Shah and Paul, 2006, Kumanayaka *et al.*, 2010, Ammala *et al.*, 2011). The experiment result indicates that the photo-oxidative degradation of PE/OMMT nanocomposite is faster than that of pure PE upon UV exposure due to the effect of OMMT and ammonium ion, in which the effect of ammonium ion is primary. Azlina *et al.* (2011) investigated the permeability of TPNR nanocomposite using OMMT nanoclay and the result indicated that oxygen barrier property of TPNR increased two-fold by adding only 2 wt% OMMT. Moreover, preparation of LDPE/OMMT nanocomposite by Arunvisut *et al.* (2007) shows gas permeability of LDPE film for O₂ gas decreased when clay contents were increased in PE/clay nanocomposites. From neat PE to 7 wt % of clay, oxygen permeability slightly decreased by 24%; from 3.8×10^{-6} cc.m/s.atm.m² for neat LDPE to $2.9 \times 3 \times 10^{-6}$ cc.m/s.atm.m² for 7 wt % of clay in PE/clay nanocomposites.

Epoxidized natural rubber (ENR) is a modified NR formed typically from poly(cis-1,4-isoprene), employing peracetic acid (Baker and Gelling, 1987) with an estimated annual world production of 12.5 million metric tons. ENR is one of the most promising resources and plays an important role in fulfilling the specific requirements of the composites as engineered materials. ENR has received attention from scientists and researchers owing to its advantages over most other polymers. Effective properties included bio-based material, susceptibility for living organisms, nontoxic, low cost, and biocompatibility which enabled ENR to be utilized for producing heat resistance, impact resistance, chemical resistance, tear resistance, eco-friendly and naturally degradable composite blends (Linos *et*

al., 2000, Chapman, 2007, Balakrishnan *et al.*, 2012a, Balakrishnan *et al.*, 2012b, Pichaiyut *et al.*, 2012a, Yikmis and Steinbüchel, 2012, Zakaria *et al.*, 2013).

This research presents a novel nanocomposite film, which is partially biodegradable, based on LDPE/OMMT nanocomposite modified with ENR, an environmentally friendly biobased polymer. The physicochemical characterizations of films are analyzed by Fourier transform infrared (FTIR) spectrum, X-ray diffraction (XRD) and atomic force microscope (AFM). Furthermore, thermal stability and behavior of films are studied by thermogravimetric analysis (TGA), and differential scanning calorimeter (DSC), respectively. Permeability, mechanical property, degradation and biodegradation behavior of films are also studied using constant pressure and a soap bubble flow meter, tensile testing machine, Q-SUN Xenon test chamber, and soil burial test, respectively.

1.2 Problem Statements

In recent years, there is an increased requirement in the plastic industry for producing packaging films with high barrier and great mechanical properties compared with those already on the market (Lange and Wyser, 2003). The latest researches on the properties of an inventive class of composite material investigated as polymer/clay nanocomposite (PCN) indicate that it provides the answer to improve the properties (Xie *et al.*, 2012, Khalili *et al.*, 2013). Properties such as super mechanical strength (Santamaría and Eguiazabal, 2012) as well as improved barrier properties against oxygen (Zhong *et al.*, 2007, Dadbin *et al.*, 2008, Dadfar *et al.*, 2011) are achievable with this process.

Currently, polyolefins especially LDPE is the largest market section and is used widely in packaging industry due to flexibility, low cost, and good processing property; as a result, improvements in both the mechanical and barrier properties seem important. Nevertheless, LDPE as one of hydrophobic polymers is resistant to biodegradation, which is another main attribute in packaging.

Environmental problems caused by disposal films including harm to the environment eco-system, water supplies and sewer systems as well as to rivers, and streams have drawn considerable attention of scientists to produce and utilize alternative material with environmentally friendly property. Recently, most scientists and researchers have focused their studies on the production of biodegradable films in addition to meeting the desired purpose in improving mechanical and barrier properties (Henriette, 2009, Cho and Kim, 2011).

In order to reduce the waste pollution as a result of packaging production attempt has been done to improve the biodegradability of film. ENR known as bio-based polymer with characteristic property has been blended to nanocomposites. Due to existence of ether group in ENR's backbone, it has enhanced and accelerated the rate of degradation mechanism that can promote degradability. Moreover, respect to biodegradability characteristic and susceptibility for living organisms, the biodegradability of film has been improved.

Furthermore, despite low barrier property of LDPE, incorporation of ENR has been enhanced the barrier property of film due to impermeability characteristic feature during stretching process and its polarity, which promises to expand the production of high barrier film in addition to biodegradable and environmentally friendly material.

In present study, novel LDPE/OMMT/ENR nanocomposites are developed. The nanocomposites with various ENR contents (0-10 wt%) are melt-compounded and blown to films. OMMT nanoclay has been used as impermeable layers and high aspect ratio to promote the barrier property and improve tensile properties. ENR as one of the environmental friendly bio-based polymers and barrier modifier has been blended with LDPE/OMMT nanocomposite to produce biodegradable nanocomposite packaging films alongside desired barrier property. Compatibilizer known as linear low-density polyethylene grafted maleic anhydride has been employed to improve the clay distribution and reduce the interfacial tension between phases.

1.3 Objectives of Research

The present study investigates the effect of ENR concentration on barrier and biodegradation of films. The knowledge obtained from this investigation will be useful in designation of film with improved barrier property and environmentally friendly degradable films. The overall main aims of the present study are to make environmentally compatible packaging film based on LDPE by enhancing its rate of biodegradation without scarifying its barrier.

The specific objectives of this research are:

- i) To examine the effect of 6 wt% of OMMT on tensile, thermal, morphology, barrier properties of neat LDPE, and LDPE/ENR nanocomposites films
- ii) To study the effect of LLDPE-g-MA as compatibilizer on the tensile, thermal, morphology, barrier and biodegradation properties of neat LDPE and LDPE/ENR nanocomposites films
- iii) To carry out the effect of ENR content on the tensile, thermal, morphology, barrier, and biodegradation properties of neat LDPE, uncompatibilized and compatibilized LDPE nanocomposites films

1.4 Scopes of Research

In this project, LDPE, OMMT, ENR and LLDPE-g-MA have been used as raw materials to produce nanocomposite films. ENR as one of environmental friendly bio-based polymers and barrier modifier has been blended with LDPE/OMMT nanocomposites to produce biodegradable nanocomposites packaging film alongside desired barrier property. OMMT has been used as nanofiller, and

LLDPE-g-MA has been applied with reference to compatibilizer because of chemical interaction between the polar groups of ENR and OMMT and miscibility with LDPE.

To obtain LDPE nanocomposite the twin screws extruder has been used for compounding the raw materials, and single screw blown film machine has been applied to produce nanocomposite film. In addition, following tests are carried out:

- i) The tensile properties of LDPE blend and nanocomposite films are determined by tensile test
- ii) The influence of nanoclay on film characterization is examined by using X-ray diffraction (XRD)
- iii) The chemical interactions of blends and nanocomposites are determined by Fourier transform infrared spectroscopy (FTIR)
- iv) Thermal behavior properties of blends and nanocomposite films are studied by differential scanning calorimeter (DSC) and thermogravimetric analysis (TGA)
- v) Permeability of the films through O₂ is examined via constant-pressure system and a soap bubble flow meter
- vi) Morphology and distribution of ENR and OMMT is observed by using atomic force microscope (AFM)
- vii) Degradation progress of different films is analyzed via Q-SUN Xenon test chamber
- viii) Biodegradability of films is examined through soil burial test

The study is highly relevant from applied viewpoint of nanotechnology and sustainability for development of advanced and environmental friendly packaging material having higher performance properties and enhanced biodegradability. It is expected this study will serve as millstone in achieving our goals toward development of environmentally friendly product without sacrificing barrier property.

1.5 Thesis Outline

Chapter 1 presents background of the study supported with an overview of discovering previous research have been conducted to packaging, nanotechnology and nanocomposites. In addition, problem statements, objectives and scope of study have been reported.

Chapter 2 begins with some general characteristic of polyolefin being used in food packaging in particular LDPE, along with its particular applications. Properties of compatibilizer, nanoclay, epoxidized natural rubber, in general, and discussion regarding their potential to develop high performance nanocomposites are presented as well. In the literature survey, some information regarding degradation and classification of degradable polymer also is reported. The key finding of previous studies done by other researchers for barrier property of film and parameters affecting the property is described as well.

Chapter 3 describes the properties of material used in present study. A detailed formulation of all prepared samples and experimental procedure to prepare film is also reported. Different utilized techniques and their operating conditions are also reported.

Chapter 4 reports the influence of OMMT which is being used frequently in nanocomposite on tensile, thermal, morphology and barrier properties. This chapter also discusses the role of LLDPE-g-MA on the properties of LDPE/ENR blends and LDPE/OMMT/ENR nanocomposites. Influence of adding ENR also on properties of un-compatibilized and compatibilized nanocomposites are discussed. Result and discussion about varying contents of ENR and influence on tensile, thermal, morphology, barrier and biodegradation is presented in details.

Chapter 5 concludes the results presented in chapter 4. In addition, suggestions for future work are also included.

REFERENCES

- Abd El-Rehim, H. A., Hegazy, E.-S. A., Ali, A. M. and Rabie, A. M. (2004). Synergistic Effect of Combining UV-Sunlight–Soil Burial Treatment on the Biodegradation Rate of LDPE/Starch Blends. *Journal of Photochemistry and Photobiology A: Chemistry* 163, 547-556.
- Abdel-Bary, E. M. (ed.) (2003). *Handbook of Plastic Films*, UK: Rapra Technology Limited.
- Abdul Wahab, M. K., Ismail, H. and Othman, N. (2012). Compatibilization Effects of PE-G-MA on Mechanical, Thermal and Swelling Properties of High Density Polyethylene/Natural Rubber/Thermoplastic Tapioca Starch Blends. *Polymer-Plastics Technology and Engineering* 51, 298-303.
- Abdullah, I. and Dahlan, M. (1998). Thermoplastic Natural Rubber Blends. *Prog. Polym. Sci.*, 23, 665-706.
- Adedeji, A., Lyu, S. and Macosko, C. W. (2001). Block Copolymers in Homopolymer Blends: Interface Vs Micelles. *Macromolecules*, 34, 8663-8668.
- Ahmad, A., Mohd, D. H. and Abdullah, I. (2004). Mechanical Properties of Filled NR/LLDPE Blends. *Iranian Polymer Journal*, 13, 173-178.
- Ahmad, I. and Fern, L. P. (2006). Effect of PE-G-MA Compatibilizer on the Morphology and Mechanical Properties of 70/30 HDPE/ENR Blends. *Polymer-Plastics Technology and Engineering*, 45, 735–739.
- Ahn, Y. C. and Paul, D. R. (2006). Rubber Toughening of Nylon 6 Nanocomposites. *Polymer*, 47, 2830-2838.
- Albertsson, A. C. (1980). Microbial and Oxidative Effects in Degradation of Polyethene. *J. Appl. Polym. Sci.*, 25, 1655-1671.
- Albertsson, A. C., Andersson, S. O. and Karlsson, S. (1987). The Mechanism of Biodegradation of Polyethylene. *Polymer Degradation and Stability*, 18, 73-87.

- Albertsson, A. C. and Karlsson, S. (1990). The Influence of Biotic and Abiotic Environments on the Degradation of Polyethylene *Progress in Polymer Science*, 15, 177-192.
- Alexandre, M. and Dubois, P. (2000). Polymer-Layered Silicate Nanocomposites: Preparation, Properties and Uses of a New Class of Materials. *Materials Science and Engineering R-Reports*, 28, 1-63.
- Alwaan, I. M. and Hassan, A. (2013). The Effects of Magnesium Oxide on the Thermal, Morphological, and Crystallinity Properties of Metallocene Linear Low-Density Polyethylene/Rubbers Composite. *Journal of Polymer Engineering* 33, 229-238.
- Ammala, A., Bateman, S., Dean, K., Petinakis, E., Sangwan, P., Wong, S., Yuan, Q., Yu, L., Patrick, C. and Leong, K. H. (2011). An Overview of Degradable and Biodegradable Polyolefins. *Progress in Polymer Science*, 36, 1015-1049.
- Anastasiadis, S. H., Gancarz, I. and Koberstein, J. T. (1989). Compatibilizing Effect of Block Copolymers Added to the Polymer/Polymer Interface. *Macromolecules*, 22, 1449-1453.
- Andrade, A. L. (2003). *Plastics and the Environment*. Hoboken, New Jersey: Wiley Interscience.
- Andrade, A. L., Hamid, H. S. and Torikai, A. (2003). Effects of Climate Change and UV-B on Materials. *Photochemical and Photobiological Sciences*, 2, 68-72.
- Archer, B. L. and Audley, B. G. (1973). Phytochemistry. Miller, L. M. (Ed.) *Rubber Gutta Percha and Chicle* New York: Van Nostrand Reinhold Co.
- Arnaud, R., Dabin, P. J. L., Al-Malaika, S., Chohan, S., Coker, M. and Scott, G. (1994). Photooxidation and Biodegradation of Commercial Photodegradable Polyethylenes. *Polymer Degradation and Stability*, 46, 211-224.
- Arora, A. and Padua, G. W. (2010). Review: Nanocomposites in Food Packaging. *Journal of Food Science*, 75, R43-R49.
- Arroyo, M., Lopez-Manchado, M. A., Valentin, J. L. and Carretero, J. (2007). Morphology/Behaviour Relationship of Nanocomposites Based on Natural Rubber/Epoxydized Natural Rubber Blends. *Composites Science and Technology*, 67, 1330-1339.
- Arunvisut, S., Phummanee, S. and Somwangthanaroj, A. (2007). Effect of Clay on Mechanical and Gas Barrier Properties of Blown Film LDPE/Clay Nanocomposites. *Journal of Applied Polymer Science*, 106, 2210-2217.

- Arutchelvi, J., Sudhakar, M., Arkatkar, A., Doble, M., Bhaduri, S. and Uppara, P. V. (2008). Biodegradation of Polyethylene and Polypropylene. *Indian Journal of Biotechnology*, 7, 9-22.
- As'habi, L., Jafari, S. H., Khonakdar, H. A. and Baghaei, B. (2011). Morphological, Rheological and Thermal Studies in Melt Processed Compatibilized PA6/ABS/Clay Nanocomposites. *Journal of Polymer Research* 18, 197-205.
- Asaletha, R., Kumaran, M. G. and Thomas, S. (1998). Thermal Behaviour of Natural Rubber/Polystyrene Blends: Thermogravimetric and Differential Scanning Calorimetric Analysis. *Polym. Degrad. Stab.* , 61, 431-439.
- Asaletha, R., Kumarana, M. G. and Thomas, S. (1999). Thermoplastic Elastomers from Blends of Polystyrene and Natural Rubber: Morphology and Mechanical Properties. *European Polymer Journal*, 35, 253-271.
- Attaran, S. A. (2010). *Mechanical, Thermal and Morphological of Epoxidized Natural Rubber Toughened Polypropylene Nanocomposite*. MS.c. Thesis. Universiti Teknologi Malaysia, Skudai.
- Austin, J. R. and Kontopoulou, M. (2006). Effect of Organoclay Content on the Rheology, Morphology, and Physical Properties of Polyolefin Elastomers and Their Blends with Polypropylene. *Polym. Eng. Sci.*, 46, 1491-1501.
- Azlina, H. N., Sahrim, H. A. and Rozaidi, R. (2011a). Enhanced Tensile and Dynamic Mechanical Properties of Thermoplastic Natural Rubber Nanocomposites. *Polymer-Plastics Technology and Engineering*, 50, 1383-1387.
- Azlina, H. N., Sahrim, H. A. and Rozaidi, R. (2012). Effect of Nanoclay on the Microstructure and the Properties of Thermoplastic Natural Rubber (TPNR)/OMMT Nanocomposites. *Journal of Thermoplastic Composite Materials*, 25, 351-362.
- Azlina, H. N., Sahrim, H. A., Rozaidi, R., Bahri, A. R. S., Yamamoto, Y. and Kawahara, S. (2011b). Oxygen Barrier Properties of New Thermoplastic Natural Rubber Nanocomposites. *Polymer-Plastics Technology and Engineering*, 50, 1564-1569.
- Azlina, N. H., Sahrim, H. A., Rozaidi, R., Bahri, A. R. S., Yamamoto, Y. and Kawahara, S. (2011c). Oxygen Barrier Properties of New Thermoplastic Natural Rubber Nanocomposites. *Polymer-Plastics Technology and Engineering*, 50, 1564-1569.

- Baghaei, B., Jafari, S. H., Khonakdar, H. A., Rezaeian, I., As'habi, L. and Ahmadian, S. (2009). Interfacially Compatibilized LDPE/POE Blends Reinforced with Nanoclay: Investigation of Morphology, Rheology and Dynamic Mechanical Properties. *Polym. Bull.* , 62, 255-270.
- Baker, C. S. L. and Gelling, I. R. (1987). Epoxidized Natural Rubber. Gelling, I. R. (Ed.) *Development of Rubber Technology*. London, UK: Elsevier Applied Science.
- Baker, C. S. L., Gelling, I. R. and Newell, R. (1985). Epoxidized Natural Rubber. *Rubber Chem. Technol.*, 58, 67-85.
- Balakrishnan, H., Attaran, S. A., Imran, M., Hassan, A. and Wahit, M. U. (2012a). Epoxidized Natural Rubber–Toughened Polypropylene/Organically Modified Montmorillonite Nanocomposites. *Journal of Thermoplastic Composite Materials*, 27, 233–250.
- Balakrishnan, H., Nematzadeh, N., Wahit, M. U., Hassan, A. and Imran, M. (2012b). Epoxidized Natural Rubber Toughened Polyamide6/Organically Modified Montmorillonite Nanocomposites. *Journal of Thermoplastic Composite Materials*, 26, 1-18.
- Beckman, J. A., Crane, G., Key, E. L. and Laman, J. R. (1974). Scrap Tire Disposal. *Rubber Chem. Technol.* , 47, 597-625.
- Bergstrom, C., Honkanen, A. and Villanen, M. (1979). The Influence Of "Two-Phase" Synthesis Conditions on Blown Film Extrusion of LDPE. *European Polymer Journal*, 15, 301-305.
- Bhattacharyya, A. R., Ghosh, A. K. and Misra, A. (2001). Reactively Compatibilised Polymer Blends: A Case Study on PA6/EVA Blend System. *Polymer*, 42, 9143-9154.
- Bhowmick, A. K., Bhattacharya, M., Mitra, S., Kumar, K. D., Maji, P. K., Choudhury, A., George, J. J. and Basak, G. C. (2011). Morphology-Property Relationship in Rubber-Based Nanocomposites: Some Recent Developments. Heinrich, G. (Ed.) *Advanced Rubber Composites*. Berlin Heidelberg: Springer.
- Bhowmick, A. K., Heslop, J. and White, J. R. (2001a). Effect of Stabilizers in Photodegradation of Thermoplastic Elastomeric Rubber-Polyethylene Blends - A Preliminary Study. *Polymer Degradation and Stability*, 74, 513-521.

- Bhowmick, A. K., Heslop, J. and White, J. R. (2001b). Photodegradation of Thermoplastic Elastomeric Rubber- Polyethylene Blends. *J Appl Polym Sci.*, 86, 2393-2402.
- Bhowmick, A. K., Heslop, J. and White, J. R. (2002). Thermal, UV- and Sunlight Ageing of Thermoplastic Elastomeric Natural Rubber-Polyethylene Blends. *Material science*, 37, 5141-5151.
- Bikiaris, D., Prinos, J., Koutsopoulos, K., Vouroutzis, N., Pavlidou, E., Frangi, N. and Panayiotou, C. (1998). LDPE/Plasticized Starch Blends Containing PE-G-MA Copolymer as Compatibilizer. *Polymer Degradation and Stability*, 59, 287-291.
- Bitinis, N., Verdejo, R., Cassagnau, P. and Lopez-Manchado, M. A. (2011). Structure and Properties of Polylactide/Natural Rubber Blends. *Materials Chemistry and Physics*, 129, 823-831.
- Blumstein, A. (1965). Polymerization of Adsorbed Monolayers. I. Preparation of the Clay-Polymer Complex. *Journal of Polymer Science Part A: General Papers*, 3, 2653-2664.
- Boer, J. D. and Pennings, A. J. (1983). *Coll. Polym. Sci.*, 261, 750.
- Borah, J. S. and Chaki, T. K. (2011). Dynamic Rheological, Morphology and Mechanical Properties of Compatibilized LLDPE/EMA Blends. *Journal of Polymer Research* 18, 907-916.
- Bovey, F. A. and Winslow, F. H. (eds.) (1979). *Macromolecules: An Introduction to Polymer Science*, London: Academic Press Inc.
- Cantor, K. (2006). *Blown Film Extrusion: An Introduction*. USA: Hanser Gardner Publishers, Inc.
- Carlsson, D. J. and Wiles, D. M. (1976). The Photooxidative Degradation of Polypropylene. Part I. Photooxidation and Photoinitiation Processes. *J. Macromol. Sci. Rev. Macromol. Chem.* , C14, 65-106.
- Carone, J., Kopcak, U., Gonçalves, M. C. and Nunes, S. P. (2000). In Situ Compatibilization of Polyamide 6/Natural Rubber Blends with Maleic Anhydride. *Polymer*, 41, 5929-5935.
- Carvalho, J. W. C., Sarantopoulos, C. and Innocentini-Mei, L. H. (2010). Nanocomposites-Based Polyolefins as Alternative to Improve Barrier Properties. *Journal of Applied Polymer Science*, 118, 3695-3700.

- Chapman, A. V. (2007). Natural Rubber and Nr-Based Polymers: Renewable Materials with Unique Properties. *24th international H.F. Mark-symposium. 'Advances in the field of elastomers & thermoplastic elastomers'*. 15-16 November. Vienna, Austria.
- Chen, D. Z., Yang, H. Y., He, P. S. and Zhang, W. (2005). Rheological and Extrusion Behavior of Intercalated High-Impact Polystyrene/Organomontmorillonite Nanocomposites. *Composites Science and Technology*, 65, 1593-1600.
- Chiellini, E. and Solaro, R. (1996). Biodegradable Polymeric Materials. *Advanced Materials*, 8, 305-313.
- Child, C. L., Clarke, R. B. F. F. and Habgood, b. J. (1942). 544,359. UK: Br. Pat.
- Chinellato, A. C., Vidotti, S. E., Hu, G. H. and Pessan, L. A. (2010). Compatibilizing Effect of Acrylic Acid Modified Polypropylene on the Morphology and Permeability Properties of Polypropylene/Organoclay Nanocomposites. *Composites Science and Technology*, 70, 458-465.
- Chiu, F. C., Lai, S. M., Chen, Y. L. and Lee, T. H. (2005). Investigation on the Polyamide 6/Organoclay Nanocomposites with or without a Maleated Polyolefin Elastomer as a Toughener. *Polymer*, 46, 11600-11609.
- Cho, T. W. and Kim, S. W. (2011). Morphologies and Properties of Nanocomposite Films Based on a Biodegradable Poly(Ester)Urethane Elastomer. *Journal of Applied Polymer Science*, 121, 1622–1630.
- Choudalakis, G. and Gotsis, A. D. (2009). Permeability of Polymer/Clay Nanocomposites: A Review. *European Polymer Journal*, 45, 967-984.
- Chow, W. S. and Lok, S. K. (2008). Effect of EPM-G-MAH on the Flexural and Morphological Properties of Poly(Lactic Acid)/Organic-Montmorillonite Nanocomposites. *Journal of Thermoplastic Composite Materials*, 21, 265-277.
- Chu, L. H., Guo, S., Chiu, W. and Tseng, H. (1993). Viscosity-Morphology-Compatibility Relationship of Polymer Blend. *Journal of Applied Polymer Science*, 49, 1791-1797.
- Ciardelli, F., Coiai, S., Passaglia, E., Pucci, A. and Ruggeri, G. (2008). Nanocomposites Based on Polyolefins and Functional Thermoplastic Materials. *Polymer International*, 57, 805-836.

- Cole, Q. P. (1959). *U.S. Patent 2912410*. New York: U.S. Patent, Quinton P. Cole, Pittsfield, Mass., assignor to General Electric Company.
- Craig, I. H., White, J. R., Shyichuk, A. V. and Syrotynska, I. (2005). Photo-Induced Scission and Crosslinking in LDPE, LLDPE, and HDPE. *Polymer Engineering and Science*, 45, 579-587.
- Cruz-Pinto, J. J. C., Carvalho, M. E. S. and Ferreira, J. F. A. (1994). The Kinetics and Mechanism of Polyethylene Photo-Oxidation. *Angew. Makromol. Chem.*, 216, 113-133.
- Cundell, A. M. and Mulcock, A. P. (1975). The Biodegradation of Vulcanized Rubber. *Div. Ind. Microbiol.*, 16, 88-96.
- Dadbin, S., Noferesti, M. and Frounchi, M. (2008). Oxygen Barrier LDPE/LLDPE/Organoclay Nano-Composite Films for Food Packaging. *Macromolecular Symposia*, 274, 22-27.
- Dadfar, S. M. A., Alemzadeh, I., Dadfar, S. M. R. and Vosoughi, M. (2011). Studies on the Oxygen Barrier and Mechanical Properties of Low Density Polyethylene/Organoclay Nanocomposite Films in the Presence of Ethylene Vinyl Acetate Copolymer as a New Type of Compatibilizer. *Materials and Design*, 32, 1806-1813.
- Dahlan, H. M., Zaman, M. D. K. and Ibrahim, A. (2002a). Liquid Natural Rubber (LNR) as a Compatibiliser in NR/LLDPE Blends - II: The Effects of Electron-Beam (EB) Irradiation. *Radiation Physics and Chemistry*, 64, 429-436.
- Dahlan, H. M., Zaman, M. D. K. and Ibrahim, A. (2002b). The Morphology and Thermal Properties of Liquid Natural Rubber (LNR)-Compatibilized 60/40 NR/LLDPE Blends. *Polymer Testing*, 21, 905-911.
- Dasari, A., Yu, Z. Z. and Mai, Y. W. (2005). Effect of Blending Sequence on Microstructure of Ternary Nanocomposites. *Polymer*, 46, 5986-5991.
- Datta, S. and Lohse, D. J. (1996). *Polymeric Compatibilizers: Uses and Benefit in Polymer Blend*. New York: Hanser.
- Davis, A. and Sims, D. (1983). *Weathering of Polymers*. New York: Elsevier Applied Science Publishers.
- Davis, S. R., Brough, A. R. and Atkinson, A. (2003). Formation of Silica/Epoxy Hybrid Network Polymers. *Journal of Non-Crystalline Solids*, 315, 197-205.

- DeArmitt, C. and Hancock, M. (2003). Filled Thermoplastics. Rothon, R. N. (Ed.) *Particulate-Filled Polymer Composites*. UK: Rapra Technology Limited.
- Dedecker, K. and Groeninckx, G. (1998). Reactive Compatibilisation of a/(B/C) Polymer Blends - Part 1. Investigation of the Phase Morphology Development and Stabilisation. *Polymer*, 39, 4985-4992.
- Dedecker, K. and Groeninckx, G. (1999). Reactive Compatibilization of the Polyamide 6/Poly(Phenylene Oxide) Blend by Means of Styrene-Maleic Anhydride Copolymer. *Journal of Applied Polymer Science*, 73, 889-898.
- Dennis, H. R., Hunter, D. L., Chang, D., Kim, S., White, J. L., Cho, J. W. and Paul, D. R. (2001). Effect of Melt Processing Conditions on the Extent of Exfoliation in Organoclay-Based Nanocomposites. *Polymer*, 42, 9513-9522.
- Dintcheva, N. T., Filippone, G., La Mantia, F. P. and Acierno, D. (2010). Photo-Oxidation Behaviour of Polyethylene/Polyamide 6 Blends Filled with Organomodified Clay: Improvement of the Photo-Resistance through Morphology Modification. *Polymer Degradation and Stability* 95, 527-535.
- Dintcheva, N. T., La Mantia, F. P. and Malatesta, V. (2009). Photo-Oxidation Behaviuor of Polyethylene/Multi-Wall Carbon Nanotube Composite Films. *Polymer Degradation and Stability*, 94, 162-170.
- Dintcheva, N. T., La Mantia, F. P. and Malatesta, V. (2011). Effect of Different Dispersing Additives on the Morphology and the Properties of Polyethylene-Based Nanocomposite Films. *Express Polymer Letters*, 5, 923-935.
- Drzal, L. T., Rich, M. J. and Koenig, M. F. (1983). Adhesion of Graphite Fibers to Epoxy Matrices: Ii. The Effect of Fiber Finish. *Journal of Adhesion*, 16 133-152.
- Dumont, M. J., Reyna-Valencia, A., Emond, J. P. and Bousmina, M. (2007). Barrier Properties of Polypropylene/Organoclay Nanocomposites. *Journal of Applied Polymer Science*, 103, 618-625.
- Durmus, A., Woo, M., Kasgoez, A., Macosko, C. W. and Tsapatsis, M. (2007). Intercalated Linear Low Density Polyethylene (LLDPE)/Clay Nanocomposites Prepared with Oxidized Polyethylene as a New Type Compatibilizer: Structural, Mechanical and Barrier Properties. *European Polymer Journal*, 43, 3737-3749.

- Egharevba1, O., Okieimen, F. E., Okwu1, U. N. and Malomo, D. (2011). Preparation and Properties of PVC/ELNR-30 Blends. *Materials Sciences and Applications*, 2, 196-199.
- Ehsani, M., Zeynali, M. E., Abtahi, M. and Harati, A. A. (2009). LDPE/EPDM Blends as Electrical Insulators with Unique Surface, Electrical and Mechanical Properties. *Iranian Polymer Journal*, 18, 37-47.
- Elias, L., Fenouillot, F., Majeste, J. C., Alcouffe, P. and Cassagnau, P. (2008a). Immiscible Polymer Blends Stabilized with Nano-Silica Particles: Rheology and Effective Interfacial Tension. *Polymer*, 49, 4378-4385.
- Elias, L., Fenouillot, F., Majeste, J. C. and Cassagnau, P. (2007). Morphology and Rheology of Immiscible Polymer Blends Filled with Silica Nanoparticles. *Polymer*, 48, 6029-6040.
- Elias, L., Fenouillot, F., Majeste, J. C., Martin, G. and Cassagnau, P. (2008b). Migration of Nanosilica Particles in Polymer Blends. *Journal of Polymer Science Part B-Polymer Physics*, 46, 1976-1983.
- Elliott, D. J. (1990). Natural Rubber-Polypropylene Blends. De, S. K. and Bhowmick, A. K. (Eds.). *Thermoplastic Elastomer from Rubber-Plastic Blends*. Chichester, UK: Ellis Horwood.
- Erlandsson, B., Karlsson, S. and Albertsson, A. C. (1997). The Mode of Action of Corn Starch and a Pro-Oxidant System in LDPE: Influence of Thermo-Oxidation and UV-Irradiation on the Molecular Weight Changes. *Polymer Degradation and Stability*, 55, 237-245.
- European-Commission (2007). Reference Document on Best Available Techniques in the Production of Polymers. Edificio Expo, c/ Inca Garcilaso, s/n, E-41092, Sevilla, Spain, European-Commission: 1-288.
- Febrianto, F., Yoshioka, M., Nagai, Y., Mihara, M. and Shiraishi, N. (1999). Composites of Wood and Trans-1,4-Isoprene Rubber I: Mechanical, Physical, and Flow Behavior. *Journal of Wood Science*, 45, 38-45.
- Fengge, G. (2004). Clay/Polymer Composites: The Story. *Materials Today*, 7, 50-55.
- Fontanella, S., Bonhomme, S., Koutny, M., Husarova, L., Brusson, J.-M., Courdavault, J.-P., Pitteri, S., Samuel, G., Pichon, G., Lemaire, J. and Delort, A.-M. (2010). Comparison of the Biodegradability of Various Polyethylene Films Containing Pro-Oxidant Additives. *Polymer Degradation and Stability*, 95, 1011-1021.

- Fornes, T. D., Yoon, P. J., Keskkula, H. and Paul, D. R. (2001). Nylon 6 Nanocomposites: The Effect of Matrix Molecular Weight. *Polymer*, 42, 9929-9940.
- Fu, X. and Qutubuddin, S. (2001). Polymer-Clay Nanocomposites: Exfoliation of Organophilic Montmorillonite Nanolayers in Polystyrene. *Polymer*, 42, 807-813.
- Gahleitner, M. (2001). Melt Rheology of Polyolefins. *Progress in Polymer Science*, 26, 895-944.
- Galgali, G., Ramesh, C. and Lele, A. (2001). A Rheological Study on the Kinetics of Hybrid Formation in Polypropylene Nanocomposites. *Macromolecules*, 34, 852-858.
- Gan, S.-N. and Abdul Hamid, Z. (1997). Partial Conversion of Epoxide Groups to Diols in Epoxidized Natural Rubber. *Polymer*, 38, 1953-1956.
- Gatos, K. G. and Karger-Kocsis, J. (2007). Effect of the Aspect Ratio of Silicate Platelets on the Mechanical and Barrier Properties of Hydrogenated Acrylonitrile Butadiene Rubber (HNBR)/Layered Silicate Nanocomposites. *Eur Polym J* 43, 1097-1104.
- Gattin, R., Copinet, A., Bertrand, C. and Couturier, Y. (2002). Biodegradation Study of a Starch and Poly(Lactic Acid) Co-Extruded Material in Liquid, Composting and Inert Mineral Media. *International Biodeterioration & Biodegradation*, 50, 25-31.
- Gelling, I. R. (1985). Modification of Natural Rubber Latex with Peracetic Acid. *Rubber Chem. Technol.*, 58, 86-96.
- Gelling, I. R. (1991). Epoxidized Natural Rubber. *J. Nat. Rubb. Res.*, 6, 184-205.
- Gelling, I. R. and Smith, F. (1979). *International Rubber Conference*. Oct. . Kuala Lumpur. Venice: Rubber Research Institute of Malaysia, 140-149.
- George, J., Joseph, R., Thomas, S. and Varughese, K. T. (2003). High Density Polyethylene/Acrylonitrile Butadiene Rubber Blends: Morphology, Mechanical Properties, and Compatibilization. *Journal of Applied Polymer Science*, 57, 449-465.
- George, S., Varughese, K. T. and Thomas, S. (2000). Thermal and Crystallisation Behaviour of Isotactic Polypropylene/Nitrile Rubber Blends. *Polymer* 41, 5485-5503.

- Georgea, S., Varugheseb, K. T. and Thomasa, S. (2000). Thermal and Crystallisation Behaviour of Isotactic Polypropylene/Nitrile Rubber Blends. *Polymer* 41, 5485–5503.
- Giesse, R. and De Paoli, M.-A. (1989). Photo-Degradable Polymer Films Derived from Low Density Polyethylene. *Polymer Degradation and Stability*, 23, 201-207.
- Goettler, L. A., Lee, K. Y. and Thakkar, H. (2007). Layered Silicate Reinforced Polymer Nanocomposites: Development and Applications. *Polymer Reviews*, 47, 291-317.
- Golebiewski, J., Rozanski, A., Dzwonkowski, J. and Galeski, A. (2008). Low Density Polyethylene-Montmorillonite Nanocomposites for Film Blowing. *European Polymer Journal*, 44, 270-286.
- Gonzalez, I., Eguiazabal, J. I. and Nazabal, J. (2005). Compatibilization Level Effects on the Structure and Mechanical Properties of Rubber-Modified Polyamide-6/Clay Nanocomposites. *Journal of Polymer Science Part B-Polymer Physics*, 43, 3611-3620.
- Gopakumar, T. G., Lee, J. A., Kontopoulou, M. and Parent, J. S. (2002). Influence of Clay Exfoliation on the Physical Properties of Montmorillonite/Polyethylene Composites. *Polymer*, 43, 5483-5491.
- Goritz, D., Horst Muller, F. and Sietz, W. (1977). Temperature Induced and Stress-Induced Crystallization in Oriented Polymers. Fischer, E. W., Horst Müller, F. and Kausch, H. H. (Eds.). *Mehrphasige Polymersysteme*. Steinkopff.
- Goritz, D., Muller, F. H. and Sietz, W. (1997). Temperature Induced and Stress-Induced Crystallization in Oriented Polymers. *Colloid Polym. Sci.*, 62, 114.
- Grassie, N. and Scott, G. (1985). *Polymer Degradation and Stabilization*. New York: Cambridge University Press.
- Gross, R. A. and Kalra, B. (2002). Biodegradable Polymers for the Environment. *Science*, 297, 803-807.
- Guduri, B. R. and Luyt, A. S. (2008). Structure and Mechanical Properties of Polycarbonate Modified Clay Nanocomposites. *J Nanosci Nanotechnol*, 8, 1880-1885.
- Guillet, J. E. (2002). Plastics and the Environment Scott, G. (Ed.) *Degradable Polymers: Principles and Applications*. 2nd Ed. Dordrecht: Kluwer Academic Publishers.

- Gulmine, J. V., Janissek, P. R., Heise, H. M. and Akcelrud, L. (2002). Polyethylene Characterization by Ftir. *Polymer Testing*, 21, 557–563.
- Hage, E., Walsh W, K. and Murayama, T. (1985). Poly(Acrylourethane)-Polyepoxide Semi-Interpenetrating Networks Formed by Electron-Beam Curing. *Multicomponent Polymer Materials*. American Chemical Society.
- Halimatuddahliana and Ismail, H. (2004). Thermoplastic Elastomer Based on PP/EPDM/ENR25 and PP/EPDM/NR Blends. *Polymer-Plastics Technology and Engineering*, 43, 357-368.
- Hamid, H., Amin, M. and Maadhah, A. (1992). *Handbook of Polymer Degradation*. New York.: Maecel dekker Inc.
- Hasegawa, N., Okamoto, H., Kawasumi, M. and usuki, A. (1999). Preparation and Mechanical Properties of Polystyrene-Caly Hybrids. *Journal of Applied Polymer Science*, 74, 3359-3364.
- Hedenqvist, M. S. (2005). Barrier Packaging Materials. *Handbook of Environmental Degradation of Materials*.
- Heinisch, K. F. (1974). *Dictionary of Rubber*. Applied Science Publishers.
- Hemati, F. and Garmabi, H. (2011). Compatibilised LDPE/LLDPE/Nanoclay Nanocomposites: I. Structural, Mechanical, and Thermal Properties. *Canadian Journal of Chemical Engineering*, 89, 187-196.
- Henriette, M. C. d. A. (2009). Nanocomposites for Food Packaging Applications. *Food Research International*, 42, 1240-1253.
- Heping, Y., Sidong, Y. and Zheng, P. (1999). Preparation and Study of Epoxidized Natural Rubber. *Journal of analysis and calorimetry*, 58, 293-299.
- Hernandez, R. J. (2002). Plastics in Packaging. *Handbook of Plastics, Elastomers & Composites*. New York : McGraw-Hill.
- Hernandez, R. J., Selke, S. E. and Culter, J. D. (2000). *Plastic Packaging, Properties, Processing, Applications and Regulations*. Hanser.
- Holst, O., Stenberg, B. and Christiansson, M. (1998). Biotechnological Possibilities for Waste Tyre-Rubber Treatment. *Biodegradation* 9, 301-310.
- Hong, S.-I. and Rhim, J.-W. (2012). Preparation and Properties of Melt-Intercalated Linear Low Density Polyethylene/Clay Nanocomposite Films Prepared by Blow Extrusion. *Lwt-Food Science and Technology*, 48, 43-51.
- Hotta, S. and Paul, D. R. (2004). Nanocomposites Formed from Linear Low Density Polyethylene and Organoclays. *Polymer*, 45, 7639-7654.

- Hu, W. C., Koberstein, J. T., Lingelser, J. P. and Gallot, Y. (1995). Interfacial-Tension Reduction in Polystyrene/Poly(Dimethylsiloxane) Blends by the Addition of Poly(Styrene-B-Dimethylsiloxane). *Macromolecules*, 28, 5209-5214.
- Huang, C. I., Chang, C. P., Shimizu, K. and Han, C. C. (2004). Phase Behavior and Crystallization Analysis in Binary Crystalline Blends of Syndiotactic Polypropylene and Ethylene–Propylene Random Copolymer. *J Polym Sci Part B Polym Phys* 42, 2995–3005.
- Huang, J.-C., Shetty, A. S. and Wang, M.-S. (1990). Biodegradable Plastics: A Review. *Advances in Polymer Technology*, 10, 23-30.
- Huang, L.-P., Zhou, X.-P., Cui, W., Xie, X.-L. and Tong, S.-Y. (2008). Toughening Effect of Maleic Anhydride Grafted Linear Low Density Polyethylene on Linear Low Density Polyethylene. *Journal of Material Science*, 43, 4290–4296.
- Hui, S., Chaki, T. K. and Chattopadhyay, S. (2008). Effect of Silica-Based Nanofillers on the Properties of a Low-Density Polyethylene/Ethylene Vinyl Acetate Copolymer Based Thermoplastic Elastomer. *Journal of Applied Polymer Science*, 110, 825-836.
- Hwang, S. W., Ryu, H. C., Kim, S. W., Park, H. Y. and Seo, K. H. (2012). Grafting Maleic Anhydride onto EVA and Effect on the Physical and Rheological Properties of PETG/EVA-G-MAH Blends. *Journal of Applied Polymer Science*, 125, 2732-2739.
- Imam, S., Glenn, G., Chiou, B.-S., Shey, J., Narayan, R. and Orts, W. (2008). Types, Production and Assessment of Biobased Food Packaging Materials. Chiellini, E. (Ed.) *Environmentally Compatible Food Packaging*. Cambridge, England: Woodhead
- International standard (2007). Space Environment (Natural and Artificial) — Process for Determining Solar Irradiances. Switzerland.
- IRSG (2014). Natural Rubber Statistics. International Rubber Study Group.
- Isik, F. (2005). *Nanocomposites Based on Blends of Polyethylene*. MS.c. Thesis. Middle East Technical University, Ankara.
- Ismail, H., Galpaya, D. and Ahmad, Z. (2009). The Compatibilizing Effect of Epoxy Resin (EP) on Polypropylene (PP)/Recycled Acrylonitrile Butadiene Rubber (NBRR) Blends. *Polymer Testing*, 28, 363-370.

- Ismail, H. and Suryadiansyah (2002). Thermoplastic Elastomers Based on Polypropylene/Natural Rubber and Polypropylene/Recycle Rubber Blends. *Polymer Testing*, 21, 389-395.
- Jaafar, J., Ismail, A. F. and Matsuura, T. (2009). Preparation and Barrier Properties of SPEEK/Cloisite 15a (R)/TAP Nanocomposite Membrane for DMFC Application. *Journal of Membrane Science*, 345, 119-127.
- Jakisch, L., Komber, H. and Bohme, F. (2003). Multifunctional Coupling Agents: Synthesis and Model Reactions. *Journal of Polymer Science Part a-Polymer Chemistry*, 41, 655-667.
- Jakisch, L., Komber, H., Haussler, L. and Bohme, F. (2000). A New Bifunctional Coupling Agent: Synthesis and Model Reactions. *Macromolecular Symposia*, 149, 237-243.
- Jalali-Arani, A., Katbab, A. A. and Nazockdast, H. (2003). Preparation of Thermoplastic Elastomers Based on Silicone Rubber and Polyethylene by Thermomechanical Reactive Blending: Effects of Polyethylene Structural Parameters. *Journal of Applied Polymer Science*, 90, 3402-3408.
- Jana, R. N., Bhattacharya, A. K., Nando, G. B. and Gupta, B. R. (2002). Compatibilized Blends of Low Density Polyethylene and Polydimethylsiloxane Rubber-Rheological Behaviour. *kautschuk und gummi kunststoffe*, 55, 660-664.
- Janaa, R. N., Mukundab, P. G. and Nandoa, G. B. (2003). Thermogravimetric Analysis of Compatibilized Blends of Low Density Polyethylene and Poly(Dimethyl Siloxane) Rubber. *Polymer Degradation and Stability*, 80, 75-82.
- Javadi, A., Pilla, S., Gong, S. and TurngLih-Sheng (2011). Biobased and Biodegradable PHBV-Based Polymer Blends and Biocomposites: Properties and Applications. Pilla, S. (Ed.) *Handbook of Bioplastics and Biocomposites Engineering Applications*. Canada: Wiley.
- Jendrossek, D., Tomasi, G. and Kroppenstedt, R. M. (1997). Bacterial Degradation of Natural Rubber: A Privilege of Actinomycetes. *FEMS Microbiol Lett*, 150, 179-188.
- Jeon, H. G., Jung, H. T., Lee, S. W. and Hudson, S. D. (1998). *Polym Bull*, 41, 107.

- Johns, J. and Rao, V. (2008). Characterization of Natural Rubber Latex/Chitosan Blends. *International Journal of Polymer Analysis and Characterization*, 13, 280-291.
- Johnson, T. and Thomas, S. (1999). Nitrogen/Oxygen Permeability of Natural Rubber, Epoxidised Natural Rubber and Natural Rubber/Epoxydised Natural Rubber Blends. *Polymer*, 40, 3223-3228.
- Joly, S., Garnaud, G., Ollitrault, R., Bokobza, L. and Mark, J. E. (2002). Organically Modified Layered Silicates as Reinforcing Fillers for Natural Rubber. *Chemistry of Materials*, 14, 4202-4208.
- Jose, S., Aprem, A. S., Francis, B., Chandy, M. C., Werner, P., Alstaedt, V. and Thomas, S. (2004). Phase Morphology, Crystallisation Behaviour and Mechanical Properties of Isotactic Polypropylene/High Density Polyethylene Blends. *Eur Polym J* 40, 2105-2115.
- Jose, S., Thomas, P. S., Thomas, S. and Karger-Kocsis, J. (2006). Thermal and Crystallisation Behaviours of Blends of Polyamide 12 with Styrene-Ethylene/Butylene-Estyrene Rubbers. *Polymer*, 47, 6328-6336.
- Kaczmarek, H. and Oldak, D. (2006). The Effect of UV-Irradiation on Composting of Polyethylene Modified by Cellulose. *Polymer Degradation and Stability* 91, 2282-2291.
- Kahar, A. W. M., Ismail, H. and Othman, N. (2012). Morphology and Tensile Properties of High-Density Polyethylene/Natural Rubber/Thermoplastic Tapioca Starch Blends: The Effect of Citric Acid-Modified Tapioca Starch. *Journal of Applied Polymer Science*, 125, 768-775.
- Kallitsis, J. K. and Kalfoglou, N. K. (1987). Miscibility of Chlorinated Polymers with Epoxidized Natural Rubber II. Blends with Chlorinated Polypropylene and with Poly(Vinylidene Chloride-Alkyl Acrylate) Copolymer. *Die Angewandte Makromolekulare Chemie*, 148, 103-117.
- Kanai, T. and Campbell, G. A. (1999). *Film Processing*. Hanser Publishers, Munich.
- Karger-Kocsis, J. and Wu, C. M. (2004). Thermoset Rubber/Layered Silicate Nanocomposites. Status and Future Trends. *Polymer Engineering and Science*, 44, 1083-1093.
- Kato, M., Okamoto, H., Hasegawa, N., Tsukigase, A. and A., U. (2003). Preparation and Properties of Polyethylene-Clay Hybrids. *Polymer Engineering and Science*, 43 1312-1316.

- Kato, M., Usuki, A. and Okada, A. (1997). Synthesis of Polypropylene Oligomer-Clay Interaction Compound. *J. Appl. Polym. Sci.*, 66, 1781-1785.
- Kawasumi, M., Hasegawa, N., Kato, M., Usuki, A. and Okada, A. (1997). Preparation and Mechanical Properties of Polypropylene-Clay Hybrids. *Macromolecules*, 30, 6333-6338.
- Ketan, K. M. (2002). *A Literature Survey on Nanocomposites*. MS.c. Thesis. University of Massachusetts Lowell, Massachusetts.
- Khabbaz, F. and Albertsson, A. C. (2000). Great Advantages in Using a Natural Rubber Instead of a Synthetic SBR in a Pro-Oxidant System for Degradable LDPE. *Biomacromolecules*, 1, 665-673.
- Khalili, S., Masoomi, M. and Bagheri, R. (2013). The Effect of Organo-Modified Montmorillonite on Mechanical and Barrier Properties of Linear Low-Density Polyethylene/Low-Density Polyethylene Blend Films. *Journal of Plastic Film & Sheeting*, 29, 39-55.
- Khan, J. H. and Hamid, S. H. (1998). Natural and Controlled Photo Aging of LDPE: Changes in Structural, Molecular and Surface Characteristics. *Journal of polymer materials*, 15, 177-184.
- Khatua, B. B., Lee, D. J., Kim, H. Y. and Kim, J. K. (2004). Effect of Organoclay Platelets on Morphology of Nylon-6 and Poly(Ethylene-Ran-Propylene) Rubber Blends. *Macromolecules*, 37, 2454-2459.
- Khosrokhavar, R., Bakhshnadeh, G. R., Ghoreishy, M. H. R. and Naderi, G. (2009). Pp/Epdm Blends and Their Developments up to Nanocomposites. *Journal of Reinforced Plastics and Composites*, 28, 613-639.
- Kietzke, T., Neher, D., Kumke, M., Montenegro, R., Landfester, K. and Scherf, U. (2004). A Nanoparticle Approach to Control the Phase Separation in Polyfluorene Photovoltaic Devices. *Macromolecules*, 37, 4882-4890.
- Kilwon, C., Fengkui, L. and Jaesung, C. (1999). Crystallization and Melting Behaviour of Blends Polypropylene and Maleated Polypropylene Blends. *Polymer*, 40, 1719-1729.
- Kim, C. K., Kim, J. J. and Paul, D. R. (1994). Phase-Stability of Ternary Blends. *Polymer Engineering and Science*, 34, 1788-1798.
- Kim, D. H., Fasulo, P. D., Rodgers, W. R. and Paul, D. R. (2007). Effect of the Ratio of Maleated Polypropylene to Organoclay on the Structure and Properties of

- Tpo-Based Nanocomposites. Part I: Morphology and Mechanical Properties. *Polymer*, 48, 5960-5978.
- Kim, J. K., Hu, C. G., Woo, R. S. C. and Sham, M. L. (2005). Moisture Barrier Characteristics of Organoclay-Epoxy Nanocomposites. *Composites Science and Technology*, 65, 805-813.
- Kim, J. K., Kim, S. and Park, C. E. (1997). Compatibilization Mechanism of Polymer Blends with an in-Situ Compatibilizer. *Polymer*, 38, 2155-2164.
- Kim, J. R., Jamieson, A. M., Hudson, S. D., Manas-Zloczower, I. and Ishida, H. (1998). Influence of Exothermic Interaction on the Morphology and Droplet Coalescence of Melt-Mixed Immiscible Polymer Blends Containing a Block Copolymer. *Macromolecules*, 31, 5383-5390.
- Kirk-Othmer (1993). Encyclopedia of Chemical Technology. 4 th ed.: John Wiley and Sons, Inc., New York.
- Knonakdar, H. A., Morshedian, J., Mehrabzadeh, M., Wagenknecht, U. and Jafari, S. H. (2003). Thermal and Shrinkage Behaviour of Stretched Peroxide Crosslinked High-Density Polyethylene. *Eur. Polym. J.*, 39, 1729-1734.
- Koklas, S. N., Sotiropoulou, D. D., Kallitsis, J. K. and Kalfoglou, N. K. (1991). Compatibilization of Chlorinated Polyethylene/Poly(Vinyl Chloride) Blends with Epoxidized Natural Rubber. *Polymer*, 32, 66-72.
- Kolybaba, M., Tabil, L. G., Panigrahi, S., Crerar, W. J., Powell, T. and Wang, B. (2003). Biodegradable Polymers: Past, Present, and Future. *An American Society of Agricultural Engineers (ASAE) Meeting Presentation*. 3-4 OCTOBER. 2003 CSAE/ASAE Annual Intersectional Meeting, Sponsored by the Red River Section of ASAE Quality Inn & Suites 301 3rd Avenue North Fargo, North Dakota, USA. The Society for Engineering in Agriculture, Food and Biological Systems.
- Koning, C., Duin, M. V., Pagnoulle, C. and Jerome, R. (1998). Strategies for Compatibilization of Polymer Blends. *Prog. Polym. Sci.*, 23, 707-757.
- Kornmann, X. (2001). *Synthesis and Characterization of Thermoset-Layered Silicate Nanocomposites*. Ph.D. Thesis. Lulea University of Technology, Lulea, Sweden.
- Kornmann, X. t., Berglund, L. A., Sterte, J. and Giannelis, E. P. (1998). Nanocomposites Based on Mmt and Unsaturated Polyester. *Polymer Engineering and Science*, 38, 1351-1358.

- Koros, W. J. (1990). *Barrier Polymers and Structures: Overview in Barrier Polymers and Structures*. Washington, DC: American chemical society.
- Koshy, A. T., Kuriakose, B., Thomas, S. and Varghese, S. (1992). Effect of Silica on Mechanical Properties and Degradation of Natural Rubber/Ethylene-Vinyl Acetate Copolymer Blends. *Kautsch Gummi Kunstst*, 45, 852–855.
- Koshy, A. T., Kuriakose, B., Thomas, S. and Varghese, S. (1993). Studies on the Effect of Blend Ratio and Crosslinking System on Thermal, X-Ray and Dynamic Mechanical Properties of Blends of Natural Rubber and Ethylene-Vinyl Acetate Copolymer. *Polymer*, 34, 3428-3436.
- Koszinowski, J. (1986). Diffusion and Solubility of N-Alkanes in Polyolefines. *Journal of Applied Polymer Science*, 32, 4765-4786.
- Krishnamoorti, R., Ren, J. and Silva, A. S. (2001). Shear Response of Layered Silicate Nanocomposites. *Journal of Chemical Physics*, 114, 4968–4973.
- Krohn, J., Tate, R. and Jordy, D. (1997). Factors Affecting the Permeability of PE Blown Films. *Journal of Plastic Film and Sheeting*, 13, 327-335.
- Kumanayaka, T. O. (2010). *Photo-Oxidation and Biodegradation of Polyethylene Nanocomposites*. Ph.D. Thesis. RMIT,
- Kumanayaka, T. O., Parthasarathy, R. and Jollands, M. (2010). Accelerating Effect of Montmorillonite on Oxidative Degradation of Polyethylene Nanocomposites. *Polymer Degradation and Stability*, 95, 672-676.
- Kumar, C. R., George, K. E. and Thomas, S. (1996). Morphology and Mechanical Properties of Thermoplastic Elastomers from Nylon-Nitrile Rubber Blends. *Journal of Applied Polymer Science*, 61, 2383-2396.
- Kurian, T. and Mathew, N. M. (2011). Natural Rubber: Production, Properties and Applications. *Biopolymers*. John Wiley & Sons, Inc.
- Kusmono, Ishak, Z. A. M., Chow, W. S., Takeichi, T. and Rochmadi (2008). Influence of SEBS-G-MA on Morphology, Mechanical, and Thermal Properties of PA6/PP/Organoclay Nanocomposites. *European Polymer Journal*, 44, 1023-1039.
- La Mantia, F. P., Dintcheva, N. T., Filippone, G. and Acierno, D. (2006). Structure and Dynamics of Polyethylene/Clay Films. *Journal of Applied Polymer Science*, 102, 4749-4758.

- Lange, J. and Wyser, Y. (2003). Recent Innovations in Barrier Technologies for Plastic Packaging - a Review. *Packaging Technology and Science*, 16, 149-158.
- Lee, D. C. and Jang, L. W. (1996). Preparation and Characterization of Pmma-Clay Hubrid Composite by Emulsion Polymerization. *Journal of Applied Polymer Science*, 61, 1117-1122.
- Lee, D. S., Yam, K. L. and Piergiovanni, L. (2008a). *Food Packaging Science and Technology*. CRC Press, Taylor & Francis Group.
- Lee, H. S., Fasulo, P. D., Rodgers, W. R. and Paul, D. R. (2005a). Tpo Based Nanocomposites. Part 1. Morphology and Mechanical Properties. *Polymer*, 46, 11673-11689.
- Lee, J.-H., Jung, D., Hong, C.-E., Rhee, K. Y. and Advani, S. G. (2005b). Properties of Polyethylene-Layered Silicate Nanocomposites Prepared by Melt Intercalation with a Pp-G-Ma Compatibilizer. *Composites Science and Technology*, 65, 1996-2002.
- Lee, S. Y., Kang, L. A., Dohl, G. H., Kim, W. J., Kim, J. S., Yoon, H. G. and Wu., Q. (2008b). Thermal, Mechanical and Morphological Properties of Polypropylene/Clay/Wood Flour Nanocomposites. *Express Polymer Letters*, 2, 78-87.
- Leffler, M., Neumann, W., Straube, E. and G., S. (1995). Microbial Surface Desulfurization of Scrap Rubber Crumb - Contribution Towards Material Recycling of Scrap Rubber. *Kautsch. GUL11mi Kunstst.*, 48, 454-457.
- Lew, C. Y., Murphy, W. R. and McNally, G. M. (2004). Preoaration and Properties of Polyolefin-Clay Nanocomposites. *Polymer Engineering and Science*, 44, 1027-1035.
- Li, C., Kong, Q., Zhao, J., Zhao, D., Fan, Q. and Xia, Y. (2004). Crystallization of Partially Miscible Linear Low-Density Polyethylene/Poly(Ethylene-Co-Vinylacetate) Blends. *Materials Letters*, 58, 3613-3617.
- Li, J. X., Wu, J. S. and Chan, C. M. (2000). Thermoplastic Nanocomposite. *Polymer*, 41, 6935-6937.
- Li, Y. M. (2001). *Mechanical Behavior of Multi-Phase Clay-Modified Polypropylene Blend Systems*. Ph.D. Thesis. A&M University, Texas.
- Liang, G., Xu, J., Bao, S. and Xu, W. (2004). Polyethylene/Maleic Anhydride Grafted Polyethylene/Organic-Montmorillonite Nanocomposites. I.

- Preparation, Microstructure, and Mechanical Properties. *Journal of Applied Polymer Science*, 91, 3974-3980.
- Lim, J. W., Hassan, A., Rahmat, A. R. and Wahit, M. U. (2006). Morphology, Thermal and Mechanical Behavior of Polypropylene Nanocomposites Toughened with Poly(Ethylene-Co-Octene). *Polymer International* 55, 204-215.
- Lim, S. T., Hyun, Y. H., Lee, C. H. and CHOI, H. J. (2003). Preparation and Characterization of Microbial Biodegradable Poly(3-Hydroxybutyrate)/Organoclay Nanocomposite. *Journal of Materials Science Letters*, 22, 299– 302.
- Linos, A., Berekaa, M. M., Reichelt, R., Keller, U., Schmitt, J., Flemming, H.-C., Kroppenstedt, R. M. and Steinbüchel, A. (2000). Biodegradation of Poly(Cis-1,4-Polyisoprene) Rubbers by Distinct Actinomycetes: Microbial Strategies and Detailed Surface Analysis. *Appl Environ Microbiol*, 66, 1639-1645.
- Liu, N. C. and Baker, W. E. (1994). Basic Functionalization of Polypropylene and the Role of Interfacial Chemical Bonding in Its Toughening. *Polymer*, 35, 988-994.
- Liu, N. C. and Baker, W. E. (1997). Modification of Polymer Melts by Oxazolines and Their Use for Interfacial Coupling Reactions with Other Functional Polymers. Al-Malaika, S. (Ed.) *Reactive Modifiers for Polymers*. 1st ed. London: Chapman and Hall.
- Lizymol, P. P. and Thomas, S. (1993). Thermal Behaviour of Polymer Blends: A Comparison of the Thermal Properties of Miscible and Immiscible Systems. *Polym Degrad Stab*, 41, 59-64.
- Lopez-Manchado, M. A., Arroyo, M., Herrero, B. and Biagiotti, J. (2003). Vulcanization Kinetics of Natural Rubber-Organoclay Nanocomposites. *Journal of Applied Polymer Science*, 89, 1-15.
- Lu, H., Xu, X., Li, X. and Zhang, Z. (2006). Morphology, Crystallization and Dynamic Mechanical Properties of PA66/Nano-Sio₂ Composites. *Bulletin of Materials Science*, 29, 485-490.
- Lu, X., Qian, R. and Brown, N. (1995). The Effect of Crystallinity on Fracture and Yielding of Polyethylenes. *Polymer*, 36, 4239-4244.

- Lukin , K. (2005). *Braggs Law*, <Http://Www.Eserc.Stonybrook.Edu/Projectjava/Bragg>. [Accessed 2005].
- Luyt, A. S., Molefi, J. A. and Krump, H. (2006). Thermal, Mechanical and Electrical Properties of Copper Powder Filled Low-Density Band Linear Low-Density Polyethylene Composites. *Polymer Degradation and Stability*, 91, 1629-1636.
- Macosko, C. W., Guegan, P., Khandpur, A. K., Nakayama, A., Marechal, P. and Inoue, T. (1996). Compatibilizers for Melt Blending: Premade Block Copolymers. *Macromolecules*, 29, 5590-5598.
- Magaraphan, R., Skuarriya, R. and Kohjiya, S. (2007). Morphological Study of LLDPE-NR Reactive Blending with Maleic Anhydride. *Journal of Applied Polymer Science*, 105, 1914-1921.
- Magaraphan, R., Thajaroen, W. and Lim-Ochakun, R. (2003). Structure and Properties of Natural Rubber and Modified Montmorillonite Nanocomposites. *Rubber Chemistry and Technology*, 76, 406-418.
- Mahapram, S. and Poompradub, S. (2011). Preparation of Natural Rubber (NR) Latex/Low Density Polyethylene (LDPE) Blown Film and Its Properties. *Polymer Testing*, 30, 716-725.
- Mahmoudian, S. (2012). *Regenerated Cellulose/Montmorillonite and Graphene Nanocomposite Via Ionic Liquids*. Ph.D. Thesis. Universiti Teknologi Malaysia, Skudai.
- Majeed, K., Jawaid, M., Hassan, A., Abu Bakar, A., Abdul Khalil, H. P. S., Salema, A. A. and Inuwa, I. (2013). Potential Materials for Food Packaging from Nanoclay/Natural Fibres Filled Hybrid Composites. *Materials and Design*, 46, 391-410.
- Margaritis, A. G., Kallitsis, J. K. and Kalfoglou, N. K. (1987). Miscibility of Chlorinated Polymers with Epoxidized Natural Rubber: 3. Blends with Chlorinated Polyethylenes. *Polymer*, 28, 2122-2129.
- Mark, H. F., Bikales, N. M., Overberger, C. G. and Menges, G. (1985). Encyclopedia of Polymer Science and Engineering. In: Kroschwitz, J. I. (ed.) second ed.: John Wiley and sons New York.
- Martuscelli, E. (1990). *Thermoplastic Elastomers from Rubber-Plastic Blends* New York: Ellis Horwood.

- Massey, L. K. (2002). *Permeability Properties of Plastics and Elastomers: A Guide to Packaging and Barrier Materials*.
- Massey, L. K. (2004). *Film Properties of Plastics and Elastomers*. United States of America by Plastics Design Library / William Andrew, Inc.
- Mathew, A. P., Packirisamy, S. and Thomas, S. (2001). Studies on the Thermal Stability of Natural Rubber/Polystyrene Interpenetrating Polymer Networks: Thermogravimetric Analysis. *Polymer Degradation and Stability* 72, 423-439.
- Mehrabzadeh, M. and Kamal, M. R. (2002). Polymer-Clay Nanocomposites Based on Blends of Polyamide-6 and Polyethylene. *The Canadian Journal of Chemical Engineering*, 80, 1083-1092.
- Messermith, P. B. and Giannelis, E. P. (1995). Synthesis and Barrier Properties of Poly (E-Caprolactone)-Layered Silicates Nanocomposites. *Journal of Applied Polymer Science Part A*, 33, 1047-1057.
- Mirzadeh, A. and Kokabi, M. (2007). The Effect of Composition and Draw-Down Ratio on Morphology and Oxygen Permeability of Polypropylene Nanocomposite Blown Films. *European Polymer Journal*, 43, 3757-3765.
- Mishra, J. K., Chang, Y.-W. and Kim, D.-K. (2007). Green Thermoplastic Elastomer Based on Polycaprolactone/Epoxydized Natural Rubber Blend as a Heat Shrinkable Material. *Materials Letters*, 61, 3551-3554.
- Mishra, J. K., Chang, Y.-W. and Kim, W. (2011). The Effect of Peroxide Crosslinking on Thermal, Mechanical, and Rheological Properties of Polycaprolactone/Epoxydized Natural Rubber Blends. *Polymer Bulletin*, 66, 673-681.
- Mittal, V. (2008). Mechanical and Gas Permeation Properties of Compatibilized Polypropylene-Layered Silicate Nanocomposites. *Journal of Applied Polymer Science*, 107, 1350-1361.
- Mittal, V. (2012). Mechanical and Gas Barrier Properties of Polypropylene Layered Silicate Nanocomposites: A Review. *The Open Macromolecules Journal*, 6, 37-52.
- Mohamad, N., Zainol, N. S., Rahim, F. F., Maulod, H. E. A., Rahim, T. A., Shamsuri, S. R., Azam, M. A., Yaakub, M. Y., Abdollah, M. F. B. and Manaf, M. E. A. (2013). Mechanical and Morphological Properties of

- Polypropylene/Epoxydized Natural Rubber Blends at Various Mixing Ratio. *Procedia Engineering*, 68, 439-445.
- Mohamad, Z., Ismail, H. and Thevy, R. C. (2006). Characterization of Epoxydized Natural Rubber/Ethylene Vinyl Acetate (ENR-50/EVA) Blend: Effect of Blend Ratio. *Journal of Applied Polymer Science*, 99, 1504-1515.
- Mohammad, Z. B. (2007). *Characterization and Properties of Epoxydized Natural Rubber (ENR-50) / Ethylene Vinylene Acetate (EVA) Blends*. Ph.D. Thesis. USM, Pinang.
- Mohanty, S., Mukunda, P. G. and Nando, G. B. (1995). Thermal-Analysis of Blends of Poly(Ethylene Coacrylic Acid) (PEA) and Epoxydized Natural-Rubber (ENR). *Polymer Degradation and Stability*, 50, 21-28.
- Moly, K. A., Radusch, H. J., Androsh, R., Bhagavan, S. S. and Thomas, S. (2005). Nonisothermal Crystallisation, Melting Behaviour and Wide Angle X-Ray Scattering Investigations on Linear Low Density Polyethylene (LLDPE)/Ethylene Vinyl Acetate (EVA) Blends: Effects of Compatibilisation and Dynamic Crosslinking. *Eur Polym J*, 41, 1410-1419.
- Mooibroek, H. and Cornish, K. (2000). Alternative Sources of Natural Rubber. *Appl Microbiol Biotechnol* 53, 355-365.
- Morales, A. R., Cruz, C. V. M., Peres, L. and Ito, E. N. (2010). PEAD/PEBDL Composites: Evaluation of the Exfoliation of Organophilic Clay Using the Nielsen Model and of the Mechanical, Optical and Permeability Properties. *Polímeros*, 20, 39-45.
- Morawiec, J., Pawlak, A., Slouf, M., Galeski, A., Piorkowska, E. and Krasnikowa, N. (2005). Preparation and Properties of Compatibilized LDPE/Organomodified Montmorillonite Nanocomposites. *European Polymer Journal*, 41, 1115-1122.
- Moyle, A. I. (1942). *Bibliography and Collected Abstracts on Rubber Producing Plants : Other Than Species of Hevea*. College Station, Tex.: Texas Agricultural Experiment Station.
- Naderi, G., Lafleur, P. G. and Dubois, C. (2007). Dynamically Vulcanized Nanocomposite Thermoplastic Elastomers Based on Epdm/Pp (Rheology and Morphology). *International Polymer Processing*, 22, 284-292.

- Nair, K. G. and Dufresne, A. (2003a). Crab Shell Chitin Whisker Reinforced Natural Rubber Nanocomposites. 1. Processing and Swelling Behavior. *Biomacromolecules*, 4, 657-665.
- Nair, K. G. and Dufresne, A. (2003b). Crab Shell Chitin Whisker Reinforced Natural Rubber Nanocomposites. 2. Mechanical Behavior. *Biomacromolecules*, 4, 666-674.
- Nakamura, Y., Watanabe, A., Mori, K., Tamura, K. and Inagaki, M. (1986). Co-Crosslinking Blend of Incompatible Polymers Vi: Poly(Vinyl Chloride) Toughened with Polyethylene Via a Co-Crosslinking Technique. *Journal of Materials Science*, 21, 4485-4488.
- Nakason, C., Jamjinno, S., Kaesaman, A. and Kiatkalmjornwong, S. (2008a). Thermoplastic Elastomer Based on High-Density Polyethylene/Natural Rubber Blends: Rheological, Thermal, and Morphological Properties. *Polymers for Advanced Technologies*, 19, 85-98.
- Nakason, C., Jarnthong, M., Kaesaman, A. and Kiatkalmjornwong, S. (2008b). Thermoplastic Elastomers Based on Epoxidized Natural Rubber and High-Density Polyethylene Blends: Effect of Blend Compatibilizers on the Mechanical and Morphological Properties. *Journal of Applied Polymer Science*, 109, 2694-2702.
- Nakason, C., Jarnthong, M., Kaesaman, A. and Kiatkalmjornwong, S. (2009). Influences of Blend Proportions and Curing Systems on Dynamic, Mechanical, and Morphological Properties of Dynamically Cured Epoxidized Natural Rubber/High-Density Polyethylene Blends. *Polymer Engineering and Science*, 49, 281-292.
- Nakason, C., Kaesaman, A. and Supasanthitkul, P. (2004a). The Grafting of Maleic Anhydride onto Natural Rubber. *Polym Test*, 23, 35–41.
- Nakason, C., Kaewsakul, W. and Kaesaman, A. (2011). Thermoplastic Natural Rubbers Based on Blending of Ethylene-Vinyl Acetate Copolymer with Different Types of Natural Rubber. *Journal of elastomers and plastics*, 0, 1-25.
- Nakason, C., Kaewsakul, W. and Kaesaman, A. (2012). Thermoplastic Natural Rubbers Based on Blending of Ethylene-Vinyl Acetate Copolymer with Different Types of Natural Rubber. *Journal of elastomers and plastics*, 44, 89-111.

- Nakason, C., Narathichat, M., Kummerloewe, C. and Vennemann, N. (2013). Thermoplastic Natural Rubber Based on Polyamide-12 Blended with Various Types of Natural Rubber: Effect of Processing Oils and Plasticizer. *Journal of elastomers and plastics*, 45, 47-75.
- Nakason, C., Nuansomr, K., Kaesaman, A. and Kiatkamjornwong, S. (2006a). Dynamic Vulcanization of Natural Rubber/High-Density Polyethylene Blends: Effect of Compatibilization, Blend Ratio and Curing System. *Polymer Testing*, 25, 782-796.
- Nakason, C., Panklieng, Y. and Kaesaman, A. (2004b). Rheological and Thermal Properties of Thermoplastic Natural Rubbers Based on Poly(Methyl Methacrylate)/Epoxidized-Natural-Rubber Blends. *Journal of Applied Polymer Science*, 92, 3561-3572.
- Nakason, C., Saiwari, S. and Kaesaman, A. (2006b). Rheological Properties of Maleated Natural Rubber/Polypropylene Blends with Phenolic Modified Polypropylene and Polypropylene-G-Maleic Anhydride Compatibilizers. *Polymer Testing*, 25, 413-423.
- Nakason, C., Tobprakhon, A. and Kaesarnan, A. (2005). Thermoplastic Vulcanizates Based on Poly(Methyl Methacrylate)/Epoxidized Natural Rubber Blends: Mechanical, Thermal, and Morphological Properties. *Journal of Applied Polymer Science*, 98, 1251-1261.
- Nakason, C., Wannavilai, P. and Kaesaman, A. (2006c). Effect of Vulcanization System on Properties of Thermoplastic Vulcanizates Based on Epoxidized Natural Rubber/Polypropylene Blends. *Polymer Testing*, 25, 34-41.
- Nakason, C., Wannavilai, P. and Kaesaman, A. (2006d). Thermoplastic Vulcanizates Based on Epoxidized Natural Rubber/Polypropylene Blends: Effect of Compatibilizers and Reactive Blending. *Journal of Applied Polymer Science*, 100, 4729-4740.
- Nakason, C., Wannavilai, P. and Kaesaman, A. (2006e). Thermoplastic Vulcanizates Based on Epoxidized Natural Rubber/Polypropylene Blends: Effect of Epoxide Levels in Enr Molecules. *Journal of Applied Polymer Science*, 101, 3046-3052.
- Narathichat, M., Kummerloewe, C., Vennemann, N., Sahakaro, K. and Nakason, C. (2012). Influence of Epoxide Level and Reactive Blending on Properties of

- Epoxidized Natural Rubber and Nylon-12 Blends. *Advances in Polymer Technology*, 31, 118-129.
- Nematzadeh, N., Wahit, M. U., Hassan, A. and Mahmoudian, S. (2011). Mechanical and Thermal Properties of Polyamide 6 Nanocomposite Toughened with Epoxidised Natural Rubber-25. Sapuan, S. M., Mustapha, F., Majid, D. L., Leman, Z., Ariff, A. H. M., Ariffin, M. K. A., Zuhri, M. Y. M., Ishak, M. R. and Sahari, J. (Eds.). *Composite Science and Technology, Pts 1 and 2*.
- Ng, S. C. and Gan, L. H. (1981). *European Polymer Journal*, 17, 1073.
- Nguyen, Q. T. and Baird, D. G. (2006). Preparation of Polymer–Clay Nanocomposites and Their Properties. *Advances in Polymer Technology*, 25, 270-285.
- Nielsen, L. E. (1967). Models for the Permeability of Filled Polymer Systems. *Journal of macromolecular science, Part A: Chemistry*, 5, 929-942.
- Nikkhah, S. J., Ramazani S. A, A., Baniasadi, H. and Tavakolzadeh, F. (2009). Investigation of Properties of Polyethylene/Clay Nanocomposites Prepared by New in Situ Ziegler-Natta Catalyst. *Materials & Design*, 30, 2309-2315.
- Noor Azlina, H., Sahrim, H. A., Rozaidi, R., Bahri, A. R. S., Yamamoto, Y. and Kawahara, S. (2011). Oxygen Barrier Properties of New Thermoplastic Natural Rubber Nanocomposites. *Polymer-Plastics Technology and Engineering*, 50, 1564-1569.
- Nouparvar, H., Hassan, A., Mohamad, Z. and Wahit, M. U. (2012). Epoxidized Natural Rubber-50 Toughened Polyamide 6 Nanocomposites: The Effect of Epoxidized Natural Rubber-50 Contents on Morphological Characterization, Mechanical and Thermal Properties. *Journal of Elastomers & Plastics*, 1-15.
- Nwabunma, D. (2007). Overview of Polyolefin Composites. Nwabunma, D. and Kyu, T. (Eds.). *Polyolefin Composites*. John Wiley & Sons, Inc.
- Obasi, H. C. and Igwe, I. O. (2014). Effects of Native Cassava Starch and Compatibilizer on Biodegradable and Tensile Properties of Polypropylene. *American Journal of Engineering Research (AJER)*, 3, 96-104.
- Okada, A. and Usuki, A. (1995). The Chemistry of Polymer-Clay Hybrids. *Mater. Sci. Eng. C3*, 109-115.
- Okhamafe, A. O. (2009). *Biopolymers in Drug Delivery: Recent Advances and Challenges*. Benin City, Nigeria: Bentham Science Publishers Ltd.

- Orr, C. A., Cernohous, J. J., Guegan, P., Hirao, A., Jeon, H. K. and Macosko, C. W. (2001). Homogeneous Reactive Coupling of Terminally Functional Polymers. *Polymer*, 42, 8171-8178.
- Osman, M. A., Rupp, J. E. P. and Suter, U. W. (2005). Effect of Non-Ionic Surfactants on the Exfoliation and Properties of Polyethylene-Layered Silicate Nanocomposites. *Polymer*, 46, 8202-8209.
- Othman, N., Hassan, A., Rahmat, A. R. and Wahit, M. U. (2007). Preparation and Characterisation of Polyethylene-Octene Grafted Maleic Anhydride-Toughened 70 : 30 PA6/PP/Mmt Nanocomposites. *Polymers & Polymer Composites*, 15, 217-227.
- Otsu, T., Tanaka, H. and Wasaki, H. (1979). Photodegradation of Chloromethyl Vinyl Ketone Polymer and Copolymers with Styrene and α -Methylstyrene. *Polymer*, 20, 55-58.
- Pandey, J. K., Reddy, K. R., Kumar, A. P. and Singh, R. P. (2005). An Overview on the Degradability of Polymer Nanocomposites. *Polymer Degradation and Stability*, 88, 234-250.
- Pang, Y., Jia, D., Hu, H., Hourston, D. and Song, M. (2000). Effects of a Compatibilizing Agent on the Morphology, Interface and Mechanical Behaviour of Polypropylene/Poly (Ethylene Terephthalate) Blends. *Polymer*, 41, 357-365.
- Papadopoulou, C. P. and Kalfoglou, N. K. (2000). Comparison of Compatibilizer Effectiveness for Pet/Pp Blends: Their Mechanical, Thermal and Morphology Characterization. *Polymer*, 41, 2543-2555.
- Passador, F. R., Ruvolo-Filho, A. C. and Pessan, L. A. (2013). Effects of Different Compatibilizers on the Rheological, Thermomechanical, and Morphological Properties of HDPE/LLDPE Blend-Based Nanocomposites. *Journal of Applied Polymer Science*, 130, 1726-1735.
- Paul, D. R. and Bucknall, C. B. (2000). *Polymer Blends*. New York: Wiley.
- Paul, D. R. and Robeson, L. M. (2008). Polymer Nanotechnology: Nanocomposites. *Polymer*, 49, 3187-3204.
- Pavliková, S., Thomann, R., Reichert, P., Mülhaupt, R., Marcinčin, A. and Borsig, E. (2003). Fiber Spinning from Poly(Propylene)-Organoclay Nanocomposite. *Journal of Applied Polymer Science*, 89, 604-611.

- Pegoretti, A., Dorigato, A. and Penati, A. (2007). Tensile Mechanical Response of Polyethylene-Clay Nanocomposites. *Express Polymer Letters*, 1, 123-131.
- Pereira de Abreu, D. A., Paseiro Losada, P., Angulo, I. and Cruz, J. M. (2007). Development of New Polyolefin Films with Nanoclays for Application in Food Packaging. *European Polymer Journal*, 43, 2229-2243.
- Persico, P., Ambrogi, V., Carfagna, C., Cerruti, P., Ferrocino, I. and Mauriello, G. (2009). Nanocomposite Polymer Films Containing Carvacrol for Antimicrobial Active Packaging. *Polymer Engineering and Science*, 49, 1447-1455.
- Picard, E., Vermogen, A., Gerard, J. F. and Espuche, E. (2008). Influence of the Compatibilizer Polarity and Molar Mass on the Morphology and the Gas Barrier Properties of Polyethylene/Clay Nanocomposites. *Journal of Polymer Science Part B-Polymer Physics*, 46, 2593-2604.
- Pichaiyut, S., Nakason, C., Kummerlowe, C. and Vennemann, N. (2012a). Thermoplastic Elastomer Based on Epoxidized Natural Rubber/Thermoplastic Polyurethane Blends: Influence of Blending Technique. *Polymers for Advanced Technologies*, 23, 1011-1019.
- Pichaiyut, S., Nakason, C. and Vennemann, N. (2012b). Thermoplastic Elastomers-Based Natural Rubber and Thermoplastic Polyurethane Blends. *Iranian Polymer Journal*, 21, 65-79.
- Pilar Villanueva, M., Cabedo, L., Maria Lagaron, J. and Gimenez, E. (2010). Comparative Study of Nanocomposites of Polyolefin Compatibilizers Containing Kaolinite and Montmorillonite Organoclays. *Journal of Applied Polymer Science*, 115, 1325-1335.
- Poh, B. T. and Khok, G. K. (2000). Tensile Property of Epoxidized Natural Rubber/Natural Rubber Blends. *Polym. Plast. Technol. Eng.*, 39, 151-161.
- Poh, B. T., Kwok, C. P. and Lim, G. H. (1995). Reversion Behaviour of Epoxidized Natural Rubber. *Eur. Polym. J.*, 31, 223-226.
- Poh, B. T. and Razai, M. J. B. (1999). Abrasion Property of Epoxidized Natural Rubber. *Polym. Plast. Technol. Eng.*, 38, 341-350.
- Polphat, R., Sukanya, U. and Jareerat, R. (2011). Relationship between Tensile Properties and Morphology of Epoxy Resin Modified by Epoxidised Natural Rubber. *Journal of Materials Science and Engineering* 5, 504-510.

- Pomposo, J. A., Calahorra, E., Eguiazabal, I. and Cortazar, M. (1993). Miscibility Behavior of Ternary Poly(Methyl Methacrylate) Poly(Ethyl Methacrylate) Poly(P-Vinylphenol) Blends. *Macromolecules*, 26, 2104-2110.
- Pongtanayuta, K., Thongpina, C. and Santawitee, O. (2013). The Effect of Rubber on Morphology, Thermal Properties and Mechanical Properties of PLA/NR and PLA/ENR Blends. *Energy Procedia* 34, 888-897.
- Pospisil, J., Horak, Z., Krulis, Z., Nespurek, S. and of, T. o. a. r. (1998). The Origin and Role of Structural Inhomogeneities and Impurities in Material Recycling of Plastics. *Macromol Symp* 35, 247-263.
- Pracella, M., Rolla, L., Chionna, D. and Galeski, A. (2002). Compatibilization and Properties of Poly(Ethylene Terephthalate)/Polyethylene Blends Based on Recycled Materials. *Macromol. Chem. Phys.*, 203, 1473-1485.
- Pukanszky, B. (1990). Influence of Interface Interaction on the Ultimate Tensile Properties of Polymer Composites. *Composites*, 21, 255-262.
- Pummerer R, Burkard P A and Kautschuk, U. (1922). *Ber. Dtsch. Chem. Ges.*, , 55, 3458.
- Qin, C., Yin, J. H. and Huang, B. T. (1990). Mechanical-Properties, Structure, and Morphology of Natural-Rubber Low-Density-Polyethylene Blends Prepared by Different Processing Methods. *Rubber Chemistry and Technology*, 63, 77-91.
- Qin, H., Zhang, S., Zhao, C., Feng, M., Yang, M., Shu, Z. and Yang, S. (2004a). Thermal Stability and Flammability of Polypropylene/Montmorillonite Composites. *Polym. Degrad. Stab.*, 85, 807-813.
- Qin, H., Zhang, Z., Feng, M., Gong, F., Zhang, S. and Yang, M. (2004b). The Influence of Interlayer Cations on the Photo-Oxidative Degradation of Polyethylene/Montmorillonite Composites. *J. Polym. Sci. Part B Polym. Phys.* , 42, 3006-3012.
- Qin, H. L., Zhao, C. G., Zhang, S. M., Chen, G. M. and Yang, M. S. (2003). Photo-Oxidative Degradation of Polyethylene/Montmorillonite Nanocomposite. *Polymer Degradation and Stability*, 81, 497-500.
- Raja Nazrul Hakim, B. R. N. (2009). *Preparation, Characterization and Properties of Organoclay Filled Natural Rubber Nanocomposites*. M.Sc. Thesis. Universiti Sains Malaysia, Pinang.

- Ranade, A., D Souza, N., Gnade, B., Thellen, C., Orroth, C., Froio, D., Lucciarini, J. and Ratto, J. A. (2004). Effect of Coupling Agent on the Dispersion of Petg Montmorillonite Nanocomposite Films. *Materials Research Society Symposium Proceedings*. Cambridge Univ Press, 791, 283-288.
- Ranade, A., Nayak, K., Fairbrother, D. and D'Souza, N. A. (2005). Maleated and Non-Maleated Polyethylene-Montmorillonite Layered Silicate Blown Films: Creep, Dispersion and Crystallinity. *Polymer*, 46, 7323-7333.
- Ratnam, C. T., Nasir , M., Baharin, A. and Zaman, K. (2000). Electron Beam Irradiation of Epoxidized Natural Rubber: Ftir Studies. *Polymer International*, 49, 1693-1701.
- Ray, I., Roy, S., Chaki, T. K. and Khastgir, D. (1994). Studies on Thermal Degradation Behaviour of EVA/LDPE Blend. *Elastomers and Plastics*, 26, 168-182.
- Ray, S., Quek, S. Y., Easteal, A. and Chen, X. D. (2006). The Potential Use of Polymer-Clay Nanocomposites in Food Packaging. *International Journal of Food Engineering*, 2, 1-11.
- Ray, S. S., Okamoto, K. and Okamoto, M. (2003). Structure-Property Relationship in Biodegradable Poly(Butylene Succinate)/Layered Silicate Nanocomposites. *Macromolecules*, 36, 2355-2367.
- Ray, S. S. and Okamoto, M. (2003). Polymer/Layered Silicate Nanocomposites: A Review from Preparation to Processing. *Progress in Polymer Science*, 28, 1539-1641.
- Razmjooei, F., Naderi, G. and Bakhshandeh, G. (2012). Preparation of Dynamically Vulcanized Thermoplastic Elastomer Nanocomposites Based on Lldpe/Reclaimed Rubber. *Journal of Applied Polymer Science*, 124, 4864-4873.
- Redhwi, H. H., Siddiqui, M. N., Andrade, A. L. and Hussain, S. (2013). Durability of Ldpe Nanocomposites with Clay, Silica, and Zinc Oxide-Part I: Mechanical Properties of the Nanocomposite Materials. *Journal of Nanomaterials*, 2013, 1-6.
- Roberts AD (1990). *Natural Rubber Science and Technology*. Oxford: Oxford science Publishers.

- Robertson, G. (2008). State-of-the-Art Biobased Food Packaging Materials. Chiellini, E. (Ed.) *Environmentally Compatible Food Packaging*. Cambridge, England: CRC.
- Robeson, L. M. (2007). *Polymer Blends: A Comprehensive Review*. Munich: Hanser Verlag.
- Robinson, J. R. (1978). *Sustained and Controlled Release Drug Delivery System*. New York: Marcel Dekker.
- Rodrigues, A., Carvalho, B. d. M., Pinheiro, L. A., Bretãs, R. E. S., Canevarolo, S. V. and Marini, J. (2013). Effect of Compatibilization and Reprocessing on the Isothermal Crystallization Kinetics of Polypropylene/Wood Flour Composites. *Polímeros*, 23, 312-319.
- Roy Choudhury, N., Chaki, T. K. and Bhowmick, A. K. (1991). Thermal Characterization of Thermoplastic Elastomeric Natural Rubber Polypropylene Blends. *Thermochimica Acta*, 176, 149-161.
- Roy Choudhury, N., Chaki, T. K., Dutta, A. and Bhowmick, A. K. (1989). Thermal, X-Ray and Dynamic Mechanical Properties of Thermoplastic Elastomeric Natural Rubber-Polyethylene Blends. *Polymer*, 30, 2047-2053.
- Roy, P. K., Hakkarainen, M., Varrna, L. K. and Albertsson, A. C. (2011). Degradable Polyethylene: Fantasy or Reality. *Environmental Science and Technology*, 45, 4217-4227.
- Roy, P. K., Surekha, P., Rajagopal, C. and Choudhary, V. (2006a). Effect of Cobalt Carboxylates on the Photo-Oxidative Degradation of Low-Density Polyethylene. Part-I. *Polymer Degradation and Stability*, 91, 1980-1988.
- Roy, P. K., Surekha, P., Rajagopal, C., Raman, R. and Choudhary, V. (2006b). Study on the Degradation of Low-Density Polyethylene in the Presence of Cobalt Stearate and Benzil. *Journal of Applied Polymer Science*, 99, 236-243.
- Rozik, N. N., Younan, A. F. and Messieh, S. L. A.-E. (2012). Preparation and Physical Studies of Nbr/Epdm Rubber Blends with Biir as Compatibilizer. *Kautschuk Und Gummi Kunststoffe*, 65, 30-35.
- Rubin, I. I. (1990). *Handbook of Plastic Materials and Technology*. New York: John Wiley and Sons Inc.
- Ruggeri, R. T. and Beck, T. R. (1980). Investigation of the Mass Transfer Characteristics of Polyurethane Paint. *Org. Coat. Plast. Chem.*, 43, 580-585.

- Saad, A. L. G. and El-Sabbagh, S. (2001). Compatibility Studies on Some Polymer Blend Systems by Electrical and Mechanical Techniques. *Journal of Applied Polymer Science*, 79, 60-71.
- Saleesung, T., Saeoui, P. and Sirisinha, C. (2010). Mechanical and Thermal Properties of Thermoplastic Elastomer Based on Low Density Polyethylene and Ultra-Fine Fully-Vulcanized Acrylonitrile Butadiene Rubber Powder (Ufnbrp). *Polymer Testing*, 29, 977-983.
- Salehabadi, A. and Abu Bakar, M. (2013). Epoxidized Natural Rubber-Organomodified Montmorillonite Nanohybrids; Interaction and Thermal Decomposition. *Materials Science Forum* 756, 119-126.
- Sam, S. T., Ismail, H. and Ahmad, Z. (2010a). Effect of Epoxidized Natural Rubber on the Processing Behavior, Tensile Properties, Morphology, and Thermal Properties of Linear-Low-Density Polyethylene/Soya Powder Blends. *Journal of Vinyl and Additive Technology*, 16, 238-245.
- Sam, S. T., Ismail, H. and Ahmad, Z. (2010b). Effect of Epoxidized Natural Rubber on the Processing Behavior, Tensile Properties, Morphology, and Thermal Properties of Linear-Low-Density Polyethylene/Soya Powder Blends. *Journal of Vinyl and additive technology*, 238-245.
- Sanchez, M. S., Ferrer, G. G., Cabanilles, C. T., Duenas, J. M. M., Pradas, M. M. and Ribelles, J. L. G. (2001). Forced Compatibility in Poly(Methyl Acrylate)/Poly(Methyl Methacrylate) Sequential Interpenetrating Polymer Networks. *Polymer*, 42, 10071-10075.
- Santamaria, P. and Eguiazabal, J. I. (2012). Structure and Mechanical Properties of Blown Films of Ionomer-Compatibilized LDPE Nanocomposites. *Polymer Testing*, 31, 367-374.
- Santra, R. N., Mukunda, P. G., Nando, G. B. and Chaki, T. K. (1993). Thermogravimetric Studies on Miscible Blends of Ethylene Methyl Acrylate Copolymer (EMA) and Polydimethylsiloxane Rubber (PDMS). *Thermochimica Acta*, 219, 283-292.
- Schadler, L. S. (2003). Nanocomposite Science and Technology. Ajayan, P. M., S., S. L. and Braun, P. V. (Eds.). *Polymer-Based and Polymer-Filled Nanocomposites*. Weinheim: Wiley-VCH.
- Schnabel, W. (1981). *Polymer Degradation, Principles and Practical Applications*. Munich: Hanser International.

- Schnecko, H. (1998). Rubber Recycling. *Macromolecular Symposium* 135, 327-343.
- Schroeder, P. A. (2002). Infrared Spectroscopy in Clay Science. *CMS Workshop Lectures, Vol. 11, Teaching Clay Science*. Colorado Aurora. The Clay Mineral Society, 11, 181-206.
- Scott, G. (1999). *Polymers and the Environment*. Cambridge: Royal Society of Chemistry.
- Selke, S. E. M., Culter, J. D. and Hernandez, R. J. (2004). *Plastics Packaging Properties, Processing, Applications, and Regulations*. Munich: Hanser Publisher.
- Seng, L. Y., Ahmad, S. H., Rasid, R., Noum, S. Y. E., Hock, Y. C. and Tarawneh, M. a. A. (2011). Effects of Liquid Natural Rubber (LNR) on the Mechanical Properties of Lnr Toughened Epoxy Composite. *Sains Malaysiana*, 40, 679-683.
- Sengupta, R., Chakraborty, S., Bandyopadhyay, S., Dasgupta, S., Mukhopadhyay, R., Auddy, K. and Deuri, A. S. (2007). A Short Review on Rubber/Clay Nanocomposites with Emphasis on Mechanical Properties. *Polymer Engineering and Science*, 47, 1956-1974.
- Shah, A. A., Hasan, F., Hameed, A. and Ahmed, S. (2008). Biological Degradation of Plastics: A Comprehensive Review. *Biotechnology Advances*, 26, 246-265.
- Shah, R. K., Krishnaswamy, R. K., Takahashi, S. and Paul, D. R. (2006). Blown Films of Nanocomposites Prepared from Low Density Polyethylene and a Sodium Ionomer of Poly(Ethylene-Co-Methacrylic Acid). *Polymer*, 47, 6187-6201.
- Shah, R. K. and Paul, D. R. (2006). Organoclay Degradation in Melt Processed Polyethylene Nanocomposites. *Polymer*, 47, 4075-4084.
- Sharif, J., Wan Yunus, W. M. Z., Mohd Dahlarb, K. Z. H. and Ahmad, M. H. (2005). Preparation and Properties of Radiation Crosslinked Natural Rubber/Clay Nanocomposite. *Polymer Testing*, 24, 211-217.
- Shen, Z., Simon, G. P. and Cheng, Y.-B. (2002). Comparison of Solution Intercalation and Melt Intercalation of Polymer-Clay Nanocomposites. *Polymer*, 43, 4251-4260.
- Shin, J., Kim, J.-C. and Chang, J.-H. (2011). Characterizations of Ultrahigh Molecular Weight Polyethylene Nanocomposite Films with Organomica. *Polymer Engineering and Science*, 51, 679-686.

- Shubhra, Q. T. H., Alam, A. K. M. M., Khan, M. A., Saha, M., Saha, D. and Gafur, M. A. (2010). Study on the Mechanical Properties, Environmental Effect, Degradation Characteristics and Ionizing Radiation Effect on Silk Reinforced Polypropylene/Natural Rubber Composites. *Composites Part a-Applied Science and Manufacturing*, 41, 1587-1596.
- Siddiqui, M. A. and Ahmed, Z. (2005). Mineralogy of the Swat Kaolin Deposits, Pakistan. *Arabian Journal for Science and Engineering*, 30, 195-218.
- Silvestre, C., Duraccio, D. and Cimmino, S. (2011). Food Packaging Based on Polymer Nanomaterials. *Progress in Polymer Science*, 36, 1766-1782.
- Singh, B. and Sharma, N. (2008). Mechanistic Implications of Plastic Degradation. *Polymer Degradation and Stability*, 93, 561-584.
- Siracusa, V. (2012). Food Packaging Permeability Behaviour: A Report. *International Journal of Polymer Science*, 2012, 1-11.
- Smith, R. (2005). *Biodegradable Polymers for Industrial Applications*. Cambridge, UK: Woodhead Publishing Limited and CRC Press LLC.
- Soares, R. M. D., Lima, A. M. F., Oliveira, R. V. B., Pires, A. T. N. and Soldi, V. (2005). Thermal Degradation of Biodegradable Edible Films Based on Xanthan and Starches from Different Sources. *Polymer Degradation and Stability*, 90, 449-454.
- Solomon, M. J., Almusallam, A. S., Seefeldt, K. F., Somwangthanaroj, A. and Varadan, P. (2001). Rheology of Polypropylene/Clay Hybrid Materials. *Macromolecules*, 34, 1864-1872.
- Standard Oil Development Co. (1937). Fr. Pat., 812,490.
- Suaysom, W. (2005). *Studies on Thermal, Ultraviolet Irradiation and Ozone Resistance of Natural Rubber for Outdoor Application* MS.c. Thesis. Kasetsart University, Thailand.
- Sundararaj, U. and Macosko, C. W. (1995). Drop Breakup and Coalence in Polymer Blends-the Effects of Concentration and Compatibilization. *Macromolecules*, 28, 2647-2657.
- Supri, A. G., Salmah, H. and Hazwan, K. (2008). Low Density Polyethylene-Nanoclay Composites: The Effect of Poly(Acrylic Acid) on Mechanical Properties, Xrd, Morphology Properties and Water Absorption. *Malaysian Polymer Journal (MPJ)* 3, 39-53.

- Taghizadeh, M. T. and Sabouri, N. (2013). Biodegradation Behaviors and Water Adsorption of Poly (Vinyl Alcohol)/Starch/Carboxymethyl Cellulose/Clay Nanocomposites. *International Nano Letters*, 3, 51.
- Tanrattanakul, V. and Udomkichdecha, W. (2001). Development of Novel Elastomeric Blends Containing Natural Rubber and Ultra-Low-Density Polyethylene. *Journal of Applied Polymer Science*, 82, 650-660.
- Gas Barrier Properties of Resins *Technical Bulletin No. 110*, 1-12.
- Teh, P. L., Mohd Ishak, Z. A., Hashim, A. S., Karger-Kocsis, J. and Ishiaku, U. S. (2004). Effects of Epoxidized Natural Rubber as a Compatibilizer in Melt Compounded Natural Rubber–Organoclay Nanocomposites. *European Polymer Journal*, 40, 2513-2521.
- Thitithammawong, A., Noordermeer, J. W. M., Kaesaman, A. and Nakason, C. (2008). Influence of Compatibilizers on the Rheological, Mechanical and Morphological Properties of Epoxidized Natural Rubber/Polypropylene Thermoplastic Vulcanizates. *Journal of Applied Polymer Science*, 107, 2436-2443.
- Tjong, S. C. (2006). Structural and Mechanical Properties of Polymer Nanocomposites. *Materials Science and Engineering R-Reports*, 53, 73-197.
- Tjong, S. C., Bao, S. P. and Liang, G. D. (2005). Polypropylene/Montmorillonite Nanocomposites Toughened with SEBS-G-MA: Structure-Property Relationship. *Journal of Polymer Science Part B-Polymer Physics*, 43, 3112-3126.
- Tokia, S., Fujimakib, T. and Okuyamab, M. (2000). Strain-Induced Crystallization of Natural Rubber as Detected Real-Time by Wide-Angle X-Ray Diffraction Technique. *Polymer*, 41, 5423-5429.
- Ton-That, M. T., Perrin-Sarazin, F., Cole, K. C., Bureau, M. N. and Denault, J. (2004). Polyolefin Nanocomposites: Formulation and Development. *Polymer Engineering and Science*, 44, 1212-1219.
- Torres, N., Robin, J. L. and Boutevin, B. (2001). Study of Compatibilization of Hdpe-Pet Blends by Adding Grafted or Statistical Copolymers. *J. Appl. Polym. Sci.*, 81, 2377-2386.
- Tosaka, M., Murakami, S., Poompradub, S. and Kohjiya, S. (2004). Orientation and Crystallization of Natural Rubber Network as Revealed by WAXD Using Synchrotron Radiation. *Macromolecules* 37, 3299-3309.

- Trossarelli, L. and Brunella, V. (2003). Polyethylene: Discovery and Growth. *UHMWPE Meeting*. 19 September. University of Torino, Italy.
- Tsai, T. Y. (2000). Templated Synthesis of Nanoparticles, Nanoporous Materials and Nanowires. *Chemical Information Monthly*, 14, 59-69.
- Tseng, F. P., Lin, J. J. and Chang, F. C. (2001). Poly(Oxypropylene)-Amide Grafted Polypropylene as Novel Compatibilizer for PP and PA6 Blends. *Polymer* 42, 713-725.
- Tsuchii, A. and Tokiwa, Y. (1999). Microbial Degradation of Natural Rubber. Steinbüchel, A. (Ed.) *Biochemical Principles and Mechanisms of Biosynthesis and Biodegradation of Polymers*. Germany: Wiley-VCH.
- Turro, N. J. (1991). *Modern Molecular Photochemistry*. University Science Books.
- Uddin, F. (2008). Clays, Nanoclays, and Montmorillonite Minerals. *Metallurgical and Materials Transactions A-Physical Metallurgy and Materials Science*, 39A, 2804-2814.
- Ugel, E., Giuliano, G. and Modesti, M. (2011). Poly(Ethylene-Co-Vinyl Acetate)/Clay Nanocomposites: Effect of Clay Nature and Compatibilising Agents on Morphological Thermal and Mechanical Properties. *Soft Nanoscience Letters*, 1, 105-119.
- USEPA (2000). Municipal Solid Waste Generation, Recycling and Disposal in the United States: Facts and Figures for 1998. *EPA 530-F-00-024*. Office of Solid Waste and Emergency Response, Washington, DC, US Environmental Protection Agency: 16-17.
- Utracki, L. A. (2002). *Polymer Blends Handbook*. Dordrecht: Kluwer Academic Publisher.
- Utracki, L. A. (2004). *Clay-Containing Polymeric Nanocomposites*. UK: Rapra Techn Ltd.
- Vaia, R. A., Ishii, H. and Giannelis, E. P. (1993). Synthesis and Properties of 2-Dimensional Nanostructures by Direct Intercalation of Polymer Melts in Layered Silicates. *Chemistry of Materials*, 5, 1694-1696.
- Van Zyl, A. J. P., Graef, S. M., Sanderson, R. D., Klumperman, B. and Pasch, H. (2003). Monitoring the Grafting of Epoxidized Natural Rubber by Size-Exclusion Chromatography Coupled to FTIR Spectroscopy. *Journal of Applied Polymer Science*, 88, 2539-2549.

- Varughese, K. T., Nando, G. B., De, S. K. and Sanyal, S. K. (1989). Tensile and Tear Failure of Plasticized Poly(Vinyl Chloride)/Epoxidized Natural Rubber Miscible Blends. *Journal of Materials Science* 24, 3491-3496.
- Vergnes, B. and Lertwimolnun, W. (2008). Impact of Processing Conditions on the Morphology, Structure and Properties of Polymer-Organoclay Nanocomposites. Thomas, S. and Zaikov, G. E. (Eds.). *Polymer Nanocomposite Research Advances*. New York: Nova Science Publishers, Inc.
- Vogt, N. B. and Kleppe, E. A. (2009). Oxo-Biodegradable Polyolefins Show Continued and Increased Thermal Oxidative Degradation after Exposure to Light. *Polymer Degradation and Stability*, 94, 659-663.
- Vu, Y. T., Mark, J. E., Pham, L. H. and Engelhardt, M. (2001). Clay Nanolayer Reinforcement of Cis-1,4-Polyisoprene and Epoxidized Natural Rubber. *Journal of Applied Polymer Science*, 82, 1391-1403.
- Wahab, M. K. A., Ismail, H. and Othman, N. (2012). Effects of Dynamic Vulcanization on the Physical, Mechanical, and Morphological Properties of High-Density Polyethylene/(Natural Rubber)/(Thermoplastic Tapioca Starch) Blends. *Journal of Vinyl and Additive Technology*, 18, 192-197.
- Wahit, M. U., Hassan, A., Ishak, Z. A. M. and Czigany, T. (2009). Ethylene-Octene Copolymer (POE) Toughened Polyamide 6/Polypropylene Nanocomposites: Effect of Poe Maleation. *Express Polymer Letters*, 3, 309-319.
- Wan, C. Y., Qiao, X. Y., Zhang, Y. and X., Z. Y. (2003). Effect of Different Clay Treatment on Morphology and Mechanical Properties of Pvc-Clay Nanocomposites. *Polymer Testing*, 22, 453-461.
- Wan, N. Y., Chin, K. P. and Mt Saad, C. S. (2010). Comparison of Epoxidised Natural Rubber (ENR) 37.5 and ENR 25/ ENR 50 Physical Blend: Specialty Polymer for ‘Green Tyre’ Application. *Materials Science and Engineering*, 11.
- Wang, K. H., Choi, M. H., Koo, C. M., Choi, Y. S. and Chung, I. J. (2001). Synthesis and Characterization of Maleated Polyethylene/Clay Nanocomposites. *Polymer*, 42, 9819-9826.
- Wang, K. H., Choi, M. H., Koo, C. M., Xu, M. Z., Chung, I. J., Jang, M. C., Choi, S. W. and Song, H. H. (2002). Morphology and Physical Properties of

- Polyethylene/Silicate Nanocomposite Prepared by Melt Intercalation. *Journal of Polymer Science Part B-Polymer Physics*, 40, 1454-1463.
- Wang, K. K., Koo, C. M. and Chung, I. J. (2003a). Physical Properties of Polyethylene/Silicate Nanocomposite Blown Films. *Journal of Applied Polymer Science*, 89, 2131-2136.
- Wang, S. F., Hu, Y., Qu, Z. K., Wang, Z. Z., Chen, Z. Y. and Fan, W. C. (2003b). Preparation and Flammability Properties of Polyethylene/Clay Nanocomposites by Melt Intercalation Method from Na⁺ Montmorillonite. *Materials Letters* 57, 2675-2678.
- Wang, S. F., Hu, Y., You, F., Song, L., Chen, Z. Y. and Fan, W. C. (2003c). Self Assembly of Polycarbonate/Acrylonitrile-Butadiene-Styrene/Montmorillonite Nanocomposites. *Journal of Applied Polymer Science*, 90, 1445-1446.
- Wang, X., Jia, D., Chen, M. and Ieee (2008). Structure and Properties of Epoxidized Nature Rubber/Organoclay Nanocomposites. *2nd IEEE International Nanoelectronics Conference (INEC)*. Mar 24-27, 2008. Shanghai, PEOPLES R CHINA. 1-3, 335-339.
- Wang, Y., Chen, F. B., Li, Y. C. and Wu, K. C. (2004). Melt Processing of Polypropylene/Clay Nanocomposites Modified with Maleated Polypropylene Compatibilizers. *Composites Part B-Engineering*, 35, 111-124.
- Wang, Y. and Chiao, S.-M. (2013). *Green Polylactide Blends for Durable Applications*. Society of Plastics Engineers (SPE), Plastics Research Online. Available: www.4spepro.org [Accessed June 2014].
- Warner, W. C. (1994). Methods of Devulcanization. *Rubber Chem. Technol.*, 67, 559-566.
- Weinkauf, D. H. and Paul, D. R. (1990). Effect of Structural Order on Barrier Properties. Koros, W. J. (Ed.) *Barrier Polymers and Structures*. Washington, DC: American Chemical Society.
- Wiederhorn, S., Fields, R., Low, S., Bahng, G.-W., Wehrstedt, A., Hahn, J., Tomota, Y., Miyata, T., Lin, H., Freeman, B., Aihara, S., Hagihara, Y. and Tagawa, T. (2006). Mechanical Properties. Czichos, H., Saito, T. and Smith, L. E. (Eds.). *Hand Book of Materials Measurment Methods*. Berlin Heidelberg: Springer.
- Wilfong, D. L., Hiltner, A. and Baer, E. J. (1986). Toughening of Polyester Resins through Blending with Polyolefins. *J. Mater. Sci.*, 21, 2014-2026.

- Wu, S. (1987). Formation of Dispersed Phase in Incompatible Polymer Blends: Interfacial and Rheological Effects. *Polymer Engineering and Science*, 27, 335-343.
- Wu, Z., Zhou, C., Qi, R. and Zhang, H. (2002). Synthesis and Characterization of Nylon 1012/Clay Nanocomposite. *Journal of Applied Polymer Science*, 83, 2403-2410.
- Xie, B., Yang, M., Li, S., Li, Z. and Feng, J. (2003). Studies on Polyamide-6/Polyolefin Blend System Compatibilized with Epoxidized Natural Rubber. *J. Appl. Polym. Sci.*, 88, 398-403.
- Xie, L., Lv, X. Y., Han, Z. J., Ci , J.-H., Fang, C. Q. and Ren, P. G. (2012). Preparation and Performance of High-Barrier Low Density Polyethylene/Organic Montmorillonite Nanocomposite. *Polymer-Plastics Technology and Engineering*, 51, 1251-1257.
- Xie, W., Gao, Z., Pan, W. P., Hunter, D., Singh, A. and Vaia, R. (2001). Thermal Degradation Chemistry of Alkyl Quaternary Ammonium Montmorillonite. *Chemistry of Materials*, 13, 2979–2990.
- Yang, J., He, G. S., Zhang, B., Luo, W. A., Chen, X. D., Fu, R. W. and Zhang, M. Q. (2011). Interfacial Adhesion of Nanoparticles in Polymer Blends by Intrinsic Fluorescence Spectra. *Express Polymer Letters*, 5, 799-808.
- Yano, A., Usuki, A., Okada, A., Karauchi, T. and Kamigaito, O. (1993). Synthesis and Properties of Polyimide-Clay Hybrid. *Journal of Polymer Science Part A: Polymer Chemistry*, 31, 2493-2499.
- Yeniova, C. E. and Yilmazer, U. (2010). Characteristics of Impact Modified Polystyrene/Organoclay Nanocomposites. *Polymer Composites*, 31, 1853-1861.
- Yikmis, M. and Steinbüchel, A. (2012). Historical and Recent Achievements in the Field of Microbial Degradation of Natural and Synthetic Rubber. *Applied and Environmental Microbiology* 78, 4543-4551.
- Yıldız, S., Karaağac, B. and Ozkoc, G. (2013). Toughening of Poly(Lactic Acid) with Silicon Rubber. *Polymer Engineering and Science*, 2029-2036.
- Yoksan, R. (2008). Epoxidized Natural Rubber for Adhesive Applications *Kasetsart J (Nat Sci)*, 42, 325-332.
- Yoo, S., Lee, S., Jeon, M., Lee, H. and Kim, W. (2013). Effects of Compatibilizers on the Mechanical, Morphological, and Thermal Properties of

- Poly(Propylene Carbonate)/Poly(Methyl Methacrylate) Blends. *Macromolecular Research*, 21, 1182-1187.
- Yu, L., Dean, K. and Li, L. (2006). Polymer Blends and Composites from Renewable Resources. *Progress in Polymer Science*, 31, 576-602.
- Zakaria, Z., Islam, M. S., Hassan, A., Mohamad Haafiz, M. K., Arjmandi, R., Inuwa, I. M. and Hasan, M. (2013). Mechanical Properties and Morphological Characterization of PLA/Chitosan/Epoxy Natural Rubber Composites. *Advances in Materials Science and Engineering*, 2013, 1-7.
- Zanetti, M., Bracco, P. and Costa, L. (2004). Thermal Degradation Behaviour of PE/Clay Nanocomposites. *Polymer Degradation and Stability* 85, 657-665.
- Zanetti, M., Camino, G., Reichert, P. and Mu"lhaupt, R. (2001a). Thermal Behavior of Polypropylene Layered Silicate Nanocomposites. *Macromol Rapid Commun*, 22, 176-180.
- Zanetti, M., Camino, G., Thomann, R. and Mu"lhaupt, R. (2001b). Synthesis and Thermal Behaviour of Layered Silicate EVA Nanocomposites. *Polymer* 42, 4501-4507.
- Zerda, A. S. and Lesser, A. J. (2001). Intercalated Clay Nanocomposite: Morphology, Mechanics and Fracture Behavior. *Journal of Polymer Science: Part B: Polymer Physics*, 39, 1137-1146.
- Zhai, H., Xu, W., Guo, H., Zhou, Z., Shen, S. and Song, Q. (2004). Preparation and Characterization of PE and PE-G-MAH/Montmorillonite Nanocomposites. *European Polymer Journal*, 40, 2539-2545.
- Zhang, C., Man, C., Pan, Y., Wang, W., Jiang, L. and Dan, Y. (2011). Toughening of Polylactide with Natural Rubber Grafted with Poly(Butyl Acrylate). *Polymer International*, 60, 1548-1555.
- Zhang, L., Wan, C. and Zhang, Y. (2009). Investigation on Morphology and Mechanical Properties of Polyamide 6/Maleated Ethylene-Propylene-Diene Rubber/Organoclay Composites. *Polymer Engineering and Science*, 49, 209-216.
- Zhang, M. and Sundararaj, U. (2006). Thermal, Rheological, and Mechanical Behaviors of LLDPE/PEMA/Clay Nanocomposites: Effect of Interaction between Polymer, Compatibilizer, and Nanofiller. *Macromolecular Materials and Engineering*, 291, 697-706.

- Zhang, Z. (1998). *Transport and Mechanical Property Studies of Barrier Plastic Food Packaging Materials*. Ph.D. Thesis. University of Guelph, Canada.
- Zhao, C., Qin, H., Gong, F., Feng, M., Zhang, S. and Yang, M. (2005). Mechanical, Thermal and Flammability Properties of Polyethylene/Clay Nanocomposites. *Polymer Degradation and Stability* 87, 183-189.
- Zhong, Y. and De Kee, D. (2005). Morphology and Properties of Layered Silicate-Polyethylene Nanocomposite Blown Films. *Polymer Engineering and Science*, 45, 469-477.
- Zhong, Y., Janes, D., Zheng, Y., Hetzer, M. and De Kee, D. (2007). Mechanical and Oxygen Barrier Properties of Organoclay-Polyethylene Nanocomposite Films. *Polymer Engineering and Science*, 47, 1101-1107.
- Zurina, M., Ismail, H. and Ratnam, C. T. (2006). Characterization of Irradiation-Induced Crosslink of Epoxidised Natural Rubber/Ethylene Vinyl Acetate (Enr-50/EVA) Blend. *Polymer Degradation Stability*, 91, 2723-2730.