

INVESTIGATION ON MORPHOLOGY AND STRUCTURAL PROPERTIES OF
2D CARBON NANOSTRUCTURE GROWN VIA 150 MHz PECVD

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“Demi agama dan negara tercinta”

“For science and humanity”

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ABSTRACT

A 150 MHz very high frequency plasma enhanced chemical vapor deposition (150 MHz VHF-PECVD) system was utilized to fabricate two-dimensional carbon nanostructure from the mixture of CH₄ and H₂. Morphology and structural properties of the grown nanostructure were investigated by means of microscopic imaging, Raman spectroscopy and X-ray diffraction technique. FESEM imaging had revealed two different carbon nanowalls (CNW), namely wavy-like and dense structure. A significant change in the film density and wall size were observed when H₂ flow rate and substrate temperature were varied. It was found that a suitable intermixing of H₂ and CH₄ is necessary for synthesizing good quality CNW. A limited or excessive amount of H₂ flow produced CNW having high defects density and poor surface coverage due to variation in the concentration of H radicals. In addition, a drastic change in film morphology was observed at growth temperature between 750 °C to 850 °C due to high rate of surface reactions. The growth of CNW was found to be more efficient at smaller electrode spacing due to better flux of hydrocarbon radicals towards the substrate surface. Typical characteristics of CNW were observed from strong D band, narrow bandwidth of G band and single broad peak of 2D band of Raman spectra indicating the presence of disordered nanocrystalline graphite structure with high degree of graphitization. The occurrence of strong peak at [002] plane with interplanar distance of 0.34 nm confirmed the growth of 2D highly graphitized CNW. It can be concluded that a capacitively coupled 150 MHz VHF-PECVD is a promising alternative technique for CNW fabrication due to its capability to dissociate CH₄ to CH_x and H radicals more efficiently.

ABSTRAK

Satu sistem pengendapan wap kimia berfrekuensi sangat tinggi 150 MHz, telah diguna pakai untuk menghasilkan karbon berstruktur nano dua dimensi daripada percampuran gas CH_4 dan H_2 . Ciri permukaan serta sifat struktur nano yang tumbuh telah dikaji menggunakan teknik pengimejan mikroskop (FESEM), spektroskopi Raman dan teknik pembelauan sinar X. Imej FESEM telah mendedahkan dua jenis sintesis tembok nano karbon (CNW) yang berlainan, iaitu berombak dan tumpat. Perubahan ketumpatan saput dan saiz tembok yang ketara telah diperolehi semasa pertumbuhan pada kadar aliran H_2 dan suhu substrat yang berbeza. Didapati, percampuran gas yang sesuai di antara H_2 dan CH_4 adalah perlu bagi menghasilkan sintesis CNW yang berkualiti. Kadar aliran H_2 yang terhad atau berlebihan akan menghasilkan saput CNW yang berketumpatan kecacatan tinggi serta kawasan liput permukaan yang buruk akibat daripada perubahan kepekatan radikal H. Pada suhu pertumbuhan antara $750\text{ }^\circ\text{C}$ ke $850\text{ }^\circ\text{C}$, permukaan saput telah mengalami perubahan drastik akibat daripada kadar tindak balas permukaan yang tinggi. Pertumbuhan CNW didapati lebih cekap pada jarak elektrod yang dekat, disebabkan oleh aliran fluks radikal hidrokarbon kepada permukaan substrat yang lebih elok. Ciri-ciri khusus spektra Raman bagi CNW yang telah diperhatikan iaitu keamatan jalur D yang tinggi, lebar jalur G yang sempit dan puncak 2D berjalur lebar, membuktikan kehadiran struktur nano-hablur grafit yang bercelaru dengan darjah penggrafitan yang tinggi. Kemunculan puncak satah [002] yang tinggi dengan jarak antara satah 0.34 nm , mengesahkan kewujudan struktur grafit dua dimensi. Kesimpulannya, sistem pengendapan wap kimia berfrekuensi sangat tinggi 150 MHz, merupakan satu kaedah alternatif yang memberansangkan bagi pertumbuhan CNW hasil daripada kemampuannya untuk menguraikan CH_4 kepada radikal CH_x dan H dengan lebih berkesan.

TABLE OF CONTENTS

CHAPTER	TITLE	PAGE
	DECLARATION	ii
	DEDICATION	iii
	ACKNOWLEDGEMENTS	iv
	ABSTRACT	v
	ABSTRAK	vi
	TABLE OF CONTENTS	vii
	LIST OF TABLES	xii
	LIST OF FIGURES	xiv
	LIST OF ABBREVIATIONS	xvii
	LIST OF SYMBOLS	xix
	LIST OF APPENDICES	xxii
1	INTRODUCTION	1
	1.1 Section Overview	1
	1.2 Background of the Study	2

1.2.1	Brief Description of Renowned Carbon Nanostructure	2
1.2.2	General Characteristic and Application of 1D and 2D Carbon Nanostructures	3
1.2.3	General Approach and Bottom-up Synthesis of Carbon Nanostructures	4
1.2.4	Motivation to Conduct Research on 2D Carbon Nanostructures	5
1.3	Statement of Problems	6
1.4	Objectives of the Study	7
1.5	Scope of the Study	7
1.6	Significance of the Study	8
1.7	Thesis Layout	10
2	LITERATURE REVIEW	11
2.1	Introduction	11
2.2	Brief History of Carbon Nanomaterials	11
2.3	Morphological Characteristics of Carbon Nanowalls	14
2.4	Research Exploration on Carbon Nanowalls	15
2.4.1	Microwave Plasma Enhanced Chemical Vapor Deposition	17
2.4.2	Inductively Coupled Plasma Enhanced Chemical Vapor Deposition	21
2.4.3	Capacitively Coupled Plasma-Enhanced Chemical Vapor Deposition with Radical Injection	24

2.4.4	Electron Beam Excited Plasma Enhanced Chemical Vapor Deposition	28
2.4.5	Hot Filament Chemical Vapor Deposition	30
2.5	150 MHz VHF-PECVD Setup and System Design	36
3	METHODOLOGY	40
3.1	Introduction	40
3.2	Description of Methodology	41
3.3	CNW Thin Film Deposition	43
3.3.1	Material and Sample Preparation	43
3.3.2	Catalyst Preparation	44
3.3.3	Growth Parameters of CNW	45
3.3.3.1	Influence of H ₂ Flow Rate	46
3.3.3.2	Influence of Substrate Temperature	47
3.3.3.3	Influence of Electrode Distance	48
3.3.3.4	Influence of Deposition Time	49
3.4	Morphology and Structural Characterization of CNW	49
3.4.1	Study of Morphology by FESEM	50
3.4.2	Structural Analysis by Raman Spectroscopy	51
3.4.3	Elemental Composition Study by Grazing Incidence X-Ray Diffraction	51

4	RESULTS AND DISCUSSION	53
4.1	Introduction	53
4.2	Morphological Study of Carbon Nanowalls	54
4.2.1	Estimation of Average CNW Density and Size	55
4.2.2	Influence of H ₂ Flow Percentage on CNW Morphology	55
4.2.3	Influence of Substrate Temperature on CNW Morphology	59
4.2.4	Influence of Electrode Distance on CNW Morphology	62
4.2.5	Effect of Deposition Time on CNW Morphology	65
4.2.6	Growth of CNW on Ni-Catalyzed Single Crystal Si and Silica Glass	68
4.3	Structural Characterization of CNW by Raman Spectroscopy	70
4.3.1	Raman Scattering for Growth of Varying H ₂ Flow Rate	71
4.3.2	Raman Scattering at Different Substrate Temperature	74
4.3.3	Raman Scattering for Growth at Varying Electrode Distance	76
4.3.4	Raman Scattering for Growth at Different Deposition Time	79
4.3.5	Spectral Interpretation of Raman Band in Graphitic Structure	82
4.3.6	Discussion on CNW Structural Properties Grown by 150 MHz VHF-PECVD	85
4.3.6.1	Effect of H ₂ Flow Rate	85
4.3.6.2	Effect of Substrate Temperature	87
4.3.6.3	Effect of Electrode Distance	87

4.3.7	Raman Scattering of CNW Grown on Catalyzed Silica Glass and Si	88
4.4	GIXRD Analysis of Grown CNW	91
5	CONCLUSION AND RECOMMENDATIONS	93
5.1	Conclusion	93
5.2	Recommendations and Future Work	95
	REFERENCES	97
	Appendices A-C	105-114

LIST OF TABLES

TABLE NO.	TITLE	PAGE
1.1	Summary of chosen experimental parameters	8
2.1	Simplified fabrication methods taken from various literatures	33
2.2	150 MHz VHF-PECVD Subsystem and Components	38
3.1	Experimental run for varying H ₂ flow rate	47
3.2	Experimental run for varying substrate temperature	48
3.3	Experimental run for varying electrode distance	48
3.4	Experimental run for varying deposition time	49
4.1	Summary of peak intensity and intensity ratio of fitted Raman spectra during growth at various H ₂ flow rate	72
4.2	Summary of bandwidth for G, D' and 2D band and crystallite size estimation of fitted Raman spectra during growth at various H ₂ flow rate	72
4.3	Summary of peak intensity and intensity ratio of fitted Raman spectra during growth at different substrate temperature	75

4.4	Summary of bandwidth for G, D' and 2D band and crystallite size estimation of fitted Raman spectra during growth at different substrate temperature	75
4.5	Summary of peak intensity and intensity ratio of fitted Raman spectra during growth utilizing different electrode separation	78
4.6	Summary of bandwidth for G, D' and 2D band and crystallite size estimation of fitted Raman spectra during growth utilizing different electrode separation	78
4.7	Summary of peak intensity and intensity ratio of fitted Raman spectra during growth at varying deposition time	81
4.8	Summary of bandwidth for G, D' and 2D band and crystallite size estimation of fitted Raman spectra during growth at varying deposition time	81
4.9	Typical reported Raman peak of CNW from various literatures	83
4.10	Summary of responses obtained from Lorentzian fit for growth of CNW with the presence of Ni catalyst	90

LIST OF FIGURES

FIGURE NO.	TITLE	PAGE
2.1	Arrangement and shape of carbon nanostructures	12
2.2	Example of SEM images of carbon nanowalls taken from literatures. Petal-like sheets and Carbon nanowalls grown via MPECVD	13
2.3	Schematic diagram of NIRIM-type microwave plasma-enhanced CVD system and ASTeX-type microwave plasma-enhanced CVD system	17
2.4	SEM images of carbon nanostructures grown at different H ₂ /CH ₄ flow rate ratios by using NIRIM-MWPCVD reactor	19
2.5	SEM images of diamond films, aligned carbon nanotubes, and carbon nanoflakes synthesized by ASTeX-MWPCVD reactor	21
2.6	Graphical assembly of inductively coupled plasma (ICP) of planar spiral and helical spiral inductive coil	22
2.7	Reactor set-up of inductively coupled plasma used to study the synthesis of carbon nanowalls	23
2.8	SEM images of carbon nanowall grown on RF-ICP by utilizing CH ₄ /Ar, CH ₄ /H ₂ and CH ₄ /N ₂ mixture at various growth time	24

2.9	Experimental set-up of RF-CCP CVD equipment facilitate with H radical injector for growth of carbon nanowalls	26
2.10	SEM images on the growth of nanowalls on Si substrate utilizing RF-CCP CVD with and without H radical injection	27
2.11	Simplified diagram of electron beam excited plasma enhanced chemical vapor deposition	28
2.12	SEM images of carbon nanowalls grown by EBEP-CVD at different growth time	29
2.13	SEM images of carbon nanowalls detached from substrate	30
2.14	Schematic diagram of hot filament chemical vapor deposition (HFCVD)	31
2.15	Interconnection of anode to RF generator	36
2.16	Interconnection of cathode to temperature controller	37
2.17	Flow diagram of vacuum system for VHF-PECVD	37
2.18	Flow diagram of gas supply system for VHF-PECVD	38
2.19	Schematic diagram of 150MHz VHF-PECVD system	39
3.1	Flow diagram of experimental works	42
3.2	Flow diagram of standard preliminary procedure for CNW growth	46
4.1	Photograph of several post-growth samples	54
4.2	SEM images of CNW grown at various H ₂ flow rate at 20000 times magnification	56
4.3	Relationship of wall density and size towards varying H ₂ flow percentage	57

4.4	SEM images of CNW grown at different substrate temperature at 20000 times magnification	60
4.5	Relationship of wall density and size towards varying growth temperature	61
4.6	SEM images of CNW grown at different electrode separation at 20000 times magnification	63
4.7	Relationship of wall density and size towards varying electrode separation	64
4.8	SEM images of CNW grown at increasing deposition time at 20000 times magnification	66
4.9	Relationship of wall density and size towards varying deposition time	67
4.10	Comparison of SEM images of CNW grown on different type of substrate; on silica glass coated with Ni catalyst and single crystal p-type Si coated with Ni catalyst	69
4.11	Acquired Raman spectra for experimental set of varying H ₂ flow percentage	71
4.12	Acquired Raman spectra for experimental set of varying substrate temperature	74
4.13	Acquired Raman spectra for experimental set of varying electrode distance	77
4.14	Acquired Raman spectra for experimental set of varying deposition time.	80
4.15	Acquired Raman spectra for the synthesis of CNW on silica glass-coated with Ni thin film at different deposition time	88
4.16	Acquired Raman spectra for the synthesis of CNW on a single crystal Si (sc-Si) sputtered with Ni thin film	89
4.17	XRD pattern for CNW obtained for growth on silica glass at different growth time and presence of catalyst	91

LIST OF ABBREVIATIONS

<i>0D</i>	-	Zero dimensional
<i>1D</i>	-	One dimensional
<i>2D</i>	-	Two dimensional
<i>2D-CN</i>	-	Two dimensional carbon nanostructures
<i>AFM</i>	-	Atomic force microscopy
<i>ASTex</i>	-	Applied Science and Technology Inc.
<i>CCD</i>		Central composite design
<i>CCP</i>	-	Capacitively coupled plasma
<i>CNF</i>	-	Carbon nanoflakes
<i>CNR</i>	-	Carbon nanorods
<i>CNS</i>	-	Carbon nanosheets
<i>CNT</i>	-	Carbon nanotubes
<i>CNW</i>	-	Carbon nanowalls
<i>CVD</i>	-	Chemical vapor deposition
<i>DC-PECVD</i>	-	Direct current plasma enhanced chemical vapor deposition
<i>DLC</i>	-	Diamond-like carbon
<i>EBE-PECVD</i>	-	Electron beam excited plasma enhanced chemical vapor deposition
<i>FC</i>	-	Filamentous carbon
<i>FESEM</i>	-	Field emission scanning electron microscope
<i>FET</i>	-	Field effect transistor

<i>GIXRD</i>	-	Grazing incidence x-ray diffraction
<i>HFCVD</i>	-	Hot filament chemical vapor deposition
<i>ICP</i>	-	Inductively coupled plasma
<i>LR</i>	-	Laboratory grade
<i>MFC</i>	-	Mass flow controller
<i>MPECVD</i>	-	Microwave plasma enhanced chemical vapor deposition
<i>MSDS</i>	-	Material safety data sheet
<i>MWCNT</i>	-	Multiwall carbon nanotubes
<i>nc-graphite</i>	-	Nanocrystalline graphite
<i>NIRIM</i>	-	National Institute of Research of Inorganic Materials, Japan
<i>PECVD</i>	-	Plasma enhanced chemical vapor deposition
<i>RF</i>	-	Radio frequency
<i>RF-CCP</i>	-	Radio frequency capacitively coupled
<i>RI-CCPCVD</i>	-	Radical injection capacitively coupled plasma enhanced chemical vapor deposition
<i>RF-ICP</i>	-	Radio frequency inductively coupled plasma
<i>sc-Si</i>	-	Single crystal Silicon
<i>SEM</i>	-	Scanning electron microscope
<i>SWCNT</i>	-	Single wall carbon nanotubes
<i>TEM</i>	-	Transmission electron microscope
<i>VHF-PECVD</i>	-	Very high frequency plasma enhanced chemical vapor deposition

LIST OF UNITS AND SYMBOLS

<i>cm</i>	-	Centi-meter (unit of length)
<i>mm</i>	-	Mili-meter
<i>nm</i>	-	Nano-meter
μm	-	Micro-meter
A/cm^2	-	Ampere per centimeter square (unit of current density)
<i>min</i>	-	Minutes
<i>s</i>	-	Seconds
<i>sccm</i>	-	Standard cubic centimeter (unit of flow rate)
<i>mTorr</i>	-	Mili-Torr (unit of pressure)
<i>V</i>	-	Volt (unit of electrical energy)
<i>kV</i>	-	Kilo-volt
$V/\mu m$	-	Volt per micro-meter
<i>L/s</i>	-	Liter per seconds (unit of flow rate/pumping speed)
cm^{-1}	-	One per centi-meter (unit of wavenumber)
°	-	Degree (unit of angle)
°C	-	Degree celcius (unit of temperature)
<i>W</i>	-	Watt (unit of power)
Ω	-	Ohm (unit of electrical resistance)
%	-	Percentage
<i>MHz</i>	-	Mega-Hertz (unit of frequency)

<i>GHz</i>	-	Giga-Hertz
<i>Ar</i>	-	Argon gas
<i>Al₂O₃</i>	-	Aluminium Oxide or alumina
<i>C₂</i>	-	Carbon radicals
<i>CH₃</i>	-	Hydrocarbon radicals of CH ₃
<i>CH₄</i>	-	Methane gas
<i>CH_x</i>	-	Hydrocarbon radicals of CH (with x =1,2,3)
<i>C₂H₂</i>	-	Acetylene gas
<i>C₂F₆</i>	-	Hexafluoroethane gas
<i>C₄F₆</i>	-	Hexafluoro-2-butyne gas
<i>CF₄</i>	-	Tetrafluoromethane gas
<i>CFH₃</i>	-	Fluoromethane free radicals
<i>Co</i>	-	Cobalt catalyst
<i>Cu-Kα</i>	-	Copper-K alpha X-ray source
<i>E_D</i>	-	Electrode distance
<i>Fe</i>	-	Iron catalyst
<i>G_t</i>	-	Growth time
<i>H</i>	-	Hydrogen atom or radicals
<i>H₂</i>	-	Hydrogen gas
<i>I_D</i>	-	Intensity of D band
<i>I_G</i>	-	Intensity of G band
<i>I_{2D}</i>	-	Intensity of 2D band
<i>I_D/I_G</i>	-	Intensity ratio of D to G band
<i>I_{D'}/I_G</i>	-	Intensity ratio of D' to G band
<i>I_{2D}/I_G</i>	-	Intensity ratio of 2D to G band
<i>MnO₂</i>	-	Manganese Oxide
<i>N₂</i>	-	Nitrogen gas
<i>NH₃</i>	-	Ammonia
<i>Ni</i>	-	Nickel catalyst
<i>NiFe</i>	-	Nickel-Ferrite catalyst
<i>SiO₂</i>	-	Silicon dioxide
<i>SiO_x</i>	-	Silicon oxide (x=integer)
<i>SiN_x</i>	-	Silicon nitride (x=integer)

sp^2	-	sp^2 molecular bonding or orbitals
sp^3	-	sp^3 molecular bonding or orbitals
T_s	-	Substrate temperature
W_D	-	Bandwidth of D band
W_G	-	Bandwidth of G band
W_{2D}	-	Bandwidth of 2D band
λ_α	-	Wavelength of X-ray beam

LIST OF APPENDICES

APPENDIX	TITLE	PAGE
A1	Magnified SEM images of CNW grown at various H ₂ flow rate	105
A2	Magnified SEM images of CNW grown at different substrate temperature	109
A3	Magnified SEM images of CNW grown at varying electrode distance	112
A4	Magnified SEM images of CNW grown at different deposition time	115
B1	Estimation of Average CNW Density and Size by FESEM Image Analysis	119
B2	Estimation of average CNW density and size for sample growth at varying H ₂ flow rate percentage	123
B3	Estimation of average CNW density and size for sample growth at varying substrate temperature	124
B4	Estimation of average CNW density and size for sample growth at varying electrode separation	125
B5	Estimation of average CNW density and size for sample growth at varying deposition time	126
B6	Estimation of average CNW density and size for sample growth on silica and single crystal Si with presence of Ni catalyst.	127

C1	Lorentzian fit of Raman spectra for samples growth at various H ₂ flow	128
C2	Lorentzian fit of Raman spectra for samples growth at different substrate temperature	132
C3	Lorentzian fit of Raman spectra for samples growth at varying electrode distance	135
C4	Lorentzian fit of Raman spectra for samples growth at various deposition time	138

CHAPTER 1

INTRODUCTION

1.1 Section Overview

This thesis presents the investigative report on two-dimensional carbon nanostructure (2D-CN) fabrication and structural characterizations; focusing on carbon nanowalls (CNW), grown by very high frequency RF plasma enhanced chemical vapour deposition (VHF-PECVD). The thesis starts with a brief review of the most popular one-dimensional (1D) and 2D carbon nanostructure and their potential applications, characteristics of graphite-based nanomaterial, bottom-up approach of depositing thin film for carbon nanostructure, structural properties characterization, experimental results and finally the conclusion and suggestions. Toward the end of this chapter, the aim and outline of the thesis will be discussed.

1.2 Background of the Study

1.2.1 Brief Description of Renowned Carbon Nanostructures

Low dimensional carbon material popularity has been increased since the groundbreaking discovery by Iijima (1991), where he first reported on the growth of helical graphitic microtubules or as nowadays widely known as carbon nanotubes (CNT), which has been the pioneer of bottom-up fabrication of carbon nanostructures. Thirteen years later, another great experimental discovery by Novoselov *et al.* (2004) regarding the electrical properties of graphene has made carbon nanostructure as the most promising material for future application in nanoelectronics, photovoltaic and energy storage. Since then graphite-based material has been in the highlight and intensively studied.

Carbon nanostructures are originated from manipulation of graphene honeycomb atomic lattice, expressed in various shapes and dimensions. 1D carbon nanostructures such as CNT, filamentous carbon (FC) and carbon nanorods (CNR) are labeled as 1D because of their one-direction growth (Hiramatsu and Hori, 2010). For example, CNT is composed of manipulation of graphene sheets into rolled or hollow cylindrical form. A single wall CNT (SWCNT) is only consists of one roll of graphene sheet, meanwhile a multi-wall CNT (MWCNT) possesses two or more rolled-graphene sheets to form tubular graphitic structure.

A two-dimensional carbon nanostructures (2D-CN) such as carbon nanosheets (CNS), carbon nanoflakes (CNF) and carbon nanowalls (CNW) are a self-organized of stacked flat graphene sheets forming graphitic building block of walls or petal-like structure. The structure exhibits sharp edges with high aspect ratio that is typically in the range of few nanometer to tens of nanometer. For example, a CNW is composed of combination of planar graphene layers, stacking towards each

other to form almost vertical graphitic walls. The sharp edges of the wall determine its aspect ratio.

1.2.2 General Characteristic and Application of 1D and 2D Carbon Nanostructures

Bulk carbon material has contributed toward many industries such as polymer, composite, firefighting, electrochemical sensors and energy production. As the rapid emergence of nanotechnology, there has been a lot of interest in producing carbon nanomaterials, which characteristics and properties are expected to surpass their bulk material. For example, diamond has been considered as the hardest bulk material on earth, indebted to its sp^3 carbon units of zinc-blende atomic lattice. However, CNT is predicted to overtake diamond in hardness (Kumar and Ando, 2010), as it possesses sp^2 carbon bond units of honeycomb lattice, which is approximately 56 times stronger than steel wire. In summary, graphite-based nanostructures exhibit interesting mechanical, physical and chemical properties due to the existence of honeycomb crystal lattice.

1D and 2D carbon nanostructure are blessed with excellent mechanical properties, high carrier mobility, large and sustainable current density and high absorption of light in the visible range. Most of these features are highly demanded in nanoelectronics, electron field emission, and blackbody-like coating. For example, the exploitation of excellent mechanical and electrical properties of CNT has proven to be beneficial in scanning probe microscopy. In a report by Ye *et al.* (2004), they had successfully fabricated a large-scale production of CNT cantilevers for atomic force microscopy (AFM) imaging with good image capturing characteristics. Besides that, high current density of CNT (10^9 - 10^{10} A/cm²) has made it possible to adapt with the continual decreasing size of Si integrated circuits. This has been proven by the fabrication of CNT as field effect transistors (FET) as reported by Wind *et al.* (2002);

Javey *et al.* (2004). In pursuit of producing a high efficiency solar cell, high light absorption material is needed to collect maximum amount of solar energy. SWCNT forest has been proven to behave like a blackbody material, as it is capable of absorbing a wide range of light. Works by Mizuno *et al.* (2009) have shown that vertically aligned SWCNT forest is capable to absorb light of wide spectral range (0.2 – 200 μm).

As for 2D-CN, its application may vary depending on its morphology and film quality. According to Hiramatsu and Mori (2010), 2D-CN with sharp edges, aligned and medium spacing of walls are demanded for application in field emission. In contrast, high-density film with less aligned walls structure is more suitable for gas storage application. In a report by Krivchenko *et al.* (2013), densest CNW film with minimal edges size possessed the best optical absorption behavior and they exhibit higher light absorption by one order of magnitude compared to CNT forest. On the other hand, the most promising application for 2D-CN such as CNW is to be employed as templates for growth of nanostructure and this has been proven fruitful for application in energy storage devices. For example, in the works of Hassan *et al.* (2014), they had managed to fabricate MnO_2 thin film with CNW/Ni as templates and found that dense and sharp edges CNW film with minimal defects lead to fast electron and ion transport and stable electrochemical cyclic activity thus providing a unique capacitive behavior.

1.2.3 General Approach and Bottom-up Synthesis of Carbon Nanostructures

Conventional thin film deposition method such as CVD has been renowned of its versatility in fabricating numerous nanostructures including carbon. There have been numerous reports regarding the growth of carbon nanostructure using modified CVD method such as DC-PECVD, RF-PECVD, Microwave PECVD (MPECVD) and electron beam excited PECVD (EBEPECVD). Typically, precursor gas such as

methane and acetylene are employed as the carbon source. For the growth of CNT, arc-discharge method and catalytic thermal CVD have been commonly employed in tremendous reports while PECVD is often employed in purpose to increase deposition rate and to lower the growth temperature. However, the history of 2D-CN growth is involved with accidental event prior to fabricating CNT. The earliest report on 2D-CN can be found in the work of Ando *et al.* (1996), where he observed both CNT and other petal-like graphite sheets using hydrogen arc discharge method. Meanwhile, in 2002, Wu *et al.* (2002) had reported the growth of CNW on catalyzed substrates using MPECVD. This can be considered as the first report to use the term “carbon nanowalls”. Currently, 2D-CN is fabricated by using various PECVD with modification, for example PECVD with hydrogen radical injection.

1.2.4 Motivation to Conduct Research on 2D Carbon Nanostructures

As has been briefly reviewed previously, 2D carbon nanostructures exhibit unique mechanical, physical and chemical properties. Hence, intensive research on its fabrication method and film properties can be fruitful as a stepping-stone toward providing alternative in many fields of applications such as renewable energy and nanoelectronics.

To date, the understanding regarding growth mechanism of 1D and 2D carbon nanostructures are still premature. For 1D carbon nanostructure such as CNT, two growth mechanisms; i) tip-growth model and ii) base-growth model have been widely accepted, however these growth mechanisms are only applicable for CNT growth on catalyzed substrate. Still, there have been reports on CNT growth on non-catalyzed substrate (Rumelli *et al.*, 2011.). For 2D-CN, there have been a number of reports on its growth mechanism and is proven direct yet random. Briefly, the growth of 2D-CN is initialized with the nucleation stage of precursor gas followed by

random assembly of nanoflakes on top of agglomerated nanoislands (Kondo *et al.*, 2009).

As conclusion, the understanding on growth mechanism model of carbon nanostructures is proven beneficial in order to further control and manipulate their growth properties for different application. It is hoped that the result of this study can be used to provide clarification on 2D-CN growth properties. Furthermore, the fabrication of 2D-CN using VHF-PECVD is less to be found, thus this study will help to provide a report on the ability of VHF-PECVD to grow CNW. It must be noted that VHF-PECVD is simpler to setup compared to radical injection PECVD and MPECVD.

1.3 Statement of Problems

All previously mentioned superior physical properties of 2D-CN however, currently exhibits some difficulties especially in material preparation and processing requirement, which are unique compare to the well-established Si processing. To date very few researchers have achieved to produce CNW with tunable morphology since its application is highly dependent on film density and structural arrangement of the walls. Besides that, current researches on the fabrication of CNW often employed a highly modified PECVD system such as MPECVD and capacitively-coupled PECVD with hydrogen radical injection. The conventional PECVD with 13.56 MHz RF plasma source is simpler to assemble compared to MPECVD and PECVD with hydrogen radical injection, which employed a two-stage plasma source to operate.

1.4 Objectives of Study

The objectives of this study are as followed;

- 1) To optimize the growth condition of CNW by varying H₂-to-CH₄ gas flow rate ratio, substrate temperature, electrode separation and deposition time.
- 2) To determine the influence of H₂ flow rate percentage, substrate temperature and electrode separation on CNW growth.
- 3) To characterize the surface morphology, crystallinity and degree of graphitization of grown CNW.

1.5 Scope of Study

In order to meet the research objectives, the experimental works must be thoroughly outlined and highlighted. The focus of this research can be categorized into two different components, which includes the fabrication of CNW using VHF-PECVD and characterization of its surface morphology and structural properties.

In the fabrication of CNW, growth parameters played a crucial role in determining the film quality. Parameters investigated in this study are summarized in Table 1.1.

Table 1.1 : Summary of chosen experimental parameters.

No.	Parameter	Range/ value	Unit
1.	CH ₄ -to-H ₂ flow rate ratio	17:0 – 17:20	Standard cubic centimeter, (sccm)
2.	Substrate temperature	700 - 850	Celcius, (°C)
3.	Electrode separation	25 – 55	Milimeter, (mm)
4.	RF power	25	Watt, (W)

Meanwhile, the substrates chosen for this study were quartz glass and single crystal Si. Typically, the growth of CNW does not require the presence of catalyst, however in this study some samples will be equipped with thin film of Ni catalyst in order to see its influence on film morphology.

For characterization of CNW film, properties such as; i) surface morphology, ii) crystallinity, and iii) degree of graphitization were chosen. In order to study the surface morphology, microscopic imaging such as field emission scanning electron microscopy (FESEM) was employed. The crystallinity and structural properties of film such as defect and stacking of graphene layers were investigated using grazing incidence x-ray diffraction (GIXRD) and Raman spectroscopy.

Finally, the growth parameters and structural characterization were correlated in order to determine the optimum growth condition of CNW using VHF-PECVD. It will also reveal the ability and feasibility of 150 MHz VHF-PECVD in synthesizing 2D carbon nanostructure with tunable morphology.

1.6 Significance of Study

As has been mentioned earlier in this chapter, carbon nanostructures exhibit unique and promising material properties for future application in variety of fields. It is hoped that this study will bring more enlightenment regarding the fabrication technique and film properties of CNW.

To date, most fabrication approach of CNW involve highly modified and complex system of PECVD. Thus, this study is hoped to evaluate the potential of simpler PECVD system (VHF-PECVD) to fabricate CNW. There are also reports on the utilization of high frequency PECVD such as in the works of Dikosnimos *et al.* (2005), however to the best of our knowledge, there are no reports on CNW film grown via 150 MHz VHF plasma source. Microwave PECVD can be classified as ultrahigh frequency plasma source at typical value of 0.915 and 2.45 GHz but it always involves the use of magnetron, which require very high voltage to operate.

In most cases, the growth of CNW does not require any catalyst, thus it is less time-consuming procedure, as there is no substrate and surface plasma treatment needed. Furthermore, it has become possible to grow CNW in various type flat surface such as Si wafer, stainless steel and glass. Another important feature of CNW is its blackbody-like behavior. Combination of both features will provide the solar cell and energy storage industry an alternative of new coating material in purpose to improve solar cell efficiency.

1.7 Thesis Layout

This thesis consists of five chapters. Each of the chapters will be briefly discussed and they are interrelated to each other bounded by the scope of study.

Chapter 1 deals with the overview of 1D and 2D carbon nanostructures, and their potential applications, recent fabrication technique and motivation to provide the need of study by highlighting its drawbacks. Therefore, an alternative is proposed concerning the specific problem, which has been addressed in the objectives of the study.

Chapter 2 is a detail review on the types of morphology of CNW and recent reports that cover recent fabrication technique and its influence on growth properties of CNW film. It also discusses the experimental results of recent literature regarding its structural properties. The parameters that affect the process were addressed in this chapter.

Chapter 3 describes the methods, technique and characterization preferred to be implemented in this study. Samples were taken for analysis and all the analytical methods are described in this chapter.

Chapter 4 is the full result of this study. The parameters that contribute and give impact to the study will be discussed. Each of the result will be transformed into simplified graphical representation for further clarification and discussion..

Chapter 5 is the conclusion obtained from the study. It gives an insight on how the objectives have been met. The chapter ends with suggestions that were made for future research.

REFERENCES

- Ando, Y., Zhao, X. & Ohkohchi, M. 1997. Production of Petal-Like Graphite Sheets by Hydrogen Arc Discharge. *Carbon*, 35, 153-158.
- Arai, Y., Sato, S. & Wagatsuma, K. 2014. Comparative Study on the Emission Spectrometric Determination of Manganese Using Nitrogen-Oxygen Okamoto-Cavity Microwave Induced Plasma and Argon Radio-Frequency Inductively-Coupled Plasma. *Microchemical Journal*, 116, 135-141.
- Banerjee, D., Mukherjee, S. & Chattopadhyay, K. K. 2011. Synthesis of Amorphous Carbon Nanowalls by DC-PECVD on Different Substrates and Study of Its Field Emission Properties. *Applied Surface Science*, 257, 3717-3722.
- Bora, B., Bhuyan, H., Favre, M., Wyndham, E., Chuaqui, H. & Wong, C. S. 2013. Measurements of Plasma Parameters in Capacitively Coupled Radio Frequency Plasma from Discharge Characteristics: Correlation with Optical Emission Spectroscopy. *Current Applied Physics*, 13, 1448-1453.
- Butoi, C. I., Mackie, N. M., Williams, K. L., Capps, N. E. & Fisher, E. R. 2000. Ion and Substrate Effects on Surface Reactions of CF_2 Using C_2F_6 , $\text{C}_2\text{F}_6/\text{H}_2$ and Hexafluoropropylene Oxide Plasmas. *Journal of Vacuum Science & Technology A: Vacuum, Surfaces, and Films*, 18, 2685-2698.
- Bystrov, K., Van De Sanden, M. C. M., Arnas, C., Marot, L., Mathys, D., Liu, F., Xu, L. K., Li, X. B., Shalpegin, A. V. & De Temmerman, G. 2014. Spontaneous Synthesis of Carbon Nanowalls, Nanotubes and Nanotips Using High Flux Density Plasmas. *Carbon*, 68, 695-707.
- Cho, H. J., Kondo, H., Ishikawa, K., Sekine, M., Hiramatsu, M. & Hori, M. 2014. Density Control of Carbon Nanowalls Grown by CH_4/H_2 Plasma and Their Electrical Properties. *Carbon*, 68, 380-388.
- Chuang, A. T. H., Boskovic, B. O. & Robertson, J. 2006. Freestanding Carbon Nanowalls by Microwave Plasma-Enhanced Chemical Vapour Deposition. *Diamond and Related Materials*, 15, 1103-1106.

- Cott, D. J., Verheijen, M., Richard, O., Radu, I., Gendt, S. D., Elshocht, S. V. & Vereecken, P. M. 2013. Synthesis of Large Area Carbon Nanosheets for Energy Storage Applications. *Carbon*, 58, 59-65.
- Davami, K., Shaygan, M., Kheirabi, N., Zhao, J., Kovalenko, D. A., Rummeli, M. H., Opitz, J., Cuniberti, G., Lee, J.-S. & Meyyappan, M. 2014. Synthesis and Characterization of Carbon Nanowalls on Different Substrates by Radio Frequency Plasma Enhanced Chemical Vapor Deposition. *Carbon*, 72, 372-380.
- Dikonimos, T., Giorgi, L., Giorgi, R., Lisi, N., Salernitano, E. & Rossi, R. 2007. Dc Plasma Enhanced Growth of Oriented Carbon Nanowall Films by HFCVD. *Journal of Diamond & Related Materials*, 16, 1240.
- Evlashin, S., Svyakhovskiy, S., Suetin, N., Pilevsky, A., Murzina, T., Novikova, N., et al. (2014). Optical and IR absorption of multilayer carbon nanowalls. *Carbon*, 70(0), 111-118.
- Ferrari, A. C. 2007. Raman Spectroscopy of Graphene and Graphite: Disorder, Electron-Phonon Coupling, Doping and Nonadiabatic Effects. *Solid State Communications*, 143, 47-57.
- French, B. L., Wang, J. J., Zhu, M. Y. and Holloway, B. C. (2006). Evolution of structure and morphology during plasma-enhanced chemical vapor deposition of carbon nanosheets. *Thin Solid Films*, 494(1-2), 105-109.
- Gavare, Z., Svagere, A., Zinge, M., Revalde, G. & Fyodorov, V. 2012. Determination of Gas Temperature of High-Frequency Low-Temperature Electrodeless Plasma Using Molecular Spectra of Hydrogen and Hydroxyl-Radical. *Journal of Quantitative Spectroscopy and Radiative Transfer*, 113, 1676-1682.
- Hassan, S., Suzuki, M., Mori, S. & El-Moneim, A. A. 2014. MnO₂/Carbon Nanowall Electrode for Future Energy Storage Application: Effect of Carbon Nanowall Growth Period and MnO₂ Mass Loading. *RSC Advances*, 4, 20479-20488.
- Hiramatsu, M. & Hori, M. 2006. Fabrication of Carbon Nanowalls Using Novel Plasma Processing. *Japan Society of Applied Physics*, 45, 5522-5527.
- Hiramatsu, M. & Hori, M. 2010. *Carbon Nanowalls*, Germany, Springer-Verlag/Wien.

- Hiramatsu, M., Shiji, K. & Hori, M. 2004. Fabrication of Vertically Aligned Carbon Nanowalls Using Capacitively Coupled Plasma-Enhanced Chemical Vapor Deposition Assisted by Hydrogen Radical Injection. *applied physics*, 4708.
- Hiramatsu, M., Lau, C. H., Bennett, A. & Foord, J. S. 2002. Formation of Diamond and Nanocrystalline Diamond Films by Microwave Plasma Cvd. *Thin Solid Films*, 407, 18-25.
- Hiramatsu, M., Deguchi, T., Nagao, H. & Hori, M. 2007. Area-Selective Growth of Aligned Single-Walled Carbon Nanotube Films Using Microwave Plasma-Enhanced Cvd. *Diamond and Related Materials*, 16, 1126-1130.
- Hiramatsu, M., Kato, K., Lau, C. H., Foord, J. S. & Hori, M. 2003a. Measurement of C₂ Radical Density in Microwave Methane/Hydrogen Plasma Used for Nanocrystalline Diamond Film Formation. *Diamond and Related Materials*, 12, 365-368.
- Hiramatsu, M., Ito, K., Lau, C. H., Foord, J. S. & Hori, M. 2003b. Fabrication of Vertically Aligned Carbon Nanostructures by Microwave Plasma-Enhanced Chemical Vapor Deposition. *Diamond and Related Materials*, 12, 786-789.
- Hiramatsu, M., Nagao, H., Taniguchi, M., Amano, H., Ando, Y. & Masaru 2005. High-Rate Growth of Films of Dense, Aligned Double-Walled Carbon Nanotubes Using Microwave Plasma-Enhanced Chemical Vapor Deposition. *The Japan Society of Applied Physics*, 44, 693-695.
- Iijima, S. 1991. Helical Microtubules of Graphitic Carbon. *Nature*, 354, 56-58.
- Itoh, T., Nakanishi, Y., Ito, T., Vetushka, A., Ledinský, M., Fejfar, A., et al. (2012). Electrical properties of carbon nanowall films. *Journal of Non-Crystalline Solids*, 358(17), 2548-2551.
- Jain, H. G., Karacuban, H., Krix, D., Becker, H.-W., Nienhaus, H. & Buck, V. 2011. Carbon Nanowalls Deposited by Inductively Coupled Plasma Enhanced Chemical Vapor Deposition Using Aluminum Acetylacetonate as Precursor. *Journal of Carbon*, 4987-4995.
- Kim, S. Y., Choi, W. S., Lee, J.-H. & Hong, B. 2014. Substrate Temperature Effect on the Growth of Carbon Nanowalls Synthesized Via Microwave PECVD. *Materials Research Bulletin*, 58, 112-116.
- Kobayashi, K., Tanimura, M., Nakai, H., Yoshimura, A., Yoshimura, H., Kojima, K. & Tachibana, M. 2007. Nanographite Domains in Carbon Nanowalls. *journal of Applied Physics*, 101, 094306.

- Kondo, S., Hori, M., Yamakawa, K., Den, S., Kano, H. & Hiramatsu, M. 2008. Highly Reliable Growth Process of Carbon Nanowalls Using Radical Injection Plasma-Enhanced Chemical Vapor Deposition. *Journal of Vacuum Science and Technology*, 26, 1294.
- Kondo, S., Kawai, S., Takeuchi, W., Yamakawa, K., Den, S., Kano, H., Hiramatsu, M. & Hori, M. 2009. Initial Growth Process of Carbon Nanowalls Synthesized by Radical Injection Plasma-Enhanced Chemical Vapor Deposition. *Journal of Applied Physics*, 106.
- Krivchenko, V. A., Evlashin, S. A., Mironovich, K. V., Verbitskiy, N. I., Nefedov, A., Wöll, C., Kozmenkova, A. Y., Suetin, N. V., Svyakhovskiy, S. E., Vyalikh, D. V., Rakhimov, A. T., Egorov, A. V. & Yashina, L. V. 2013. Carbon Nanowalls: The Next Step for Physical Manifestation of the Black Body Coating. *Sci. Rep.*, 3.
- Kumar, M. & Ando, Y. 2010. Chemical Vapor Deposition of Carbon Nanotubes: A Review on Growth Mechanism and Mass Production. *Journal of Nanoscience and Nanotechnology*, 10, 3739–3758.
- Kurita, S., Yoshimura, A., H., K., Uchida, T., Tkojima, K., Tachibana, M., Molina-Morales, P. & Nakai, H. 2005. Raman Spectra of Carbon Nanowalls Grown by Plasma-Enhanced Chemical Vapor Deposition. *journal of applied physics*, 104320.
- Lei, M. K., Liu, Y. & Li, Y. P. 2011. Controllable Wettability of Poly(Ethylene Terephthalate) Film Modified by Oxygen Combined Inductively and Capacitively Coupled Radio-Frequency Plasma. *Applied Surface Science*, 257, 7350-7358.
- Li, Z. Q., Lu, C. J., Xia, Z. P., Zhou, Y. and Luo, Z. (2007). X-ray diffraction patterns of graphite and turbostratic carbon. *Carbon*, 45(8), 1686-1695.
- Lin, C. R., Su, C. H., Chang, C. Y., Hung, C. H. and Huang, Y. F. (2006). Synthesis of nanosized flake carbons by RF-chemical vapor method. *Surface and Coatings Technology*, 200(10), 3190-3193.
- Lisi, N., Giorgi, R., Re, M., Dikonimos, T., Giorgi, L., Salernitano, E., Gagliardi, S. & Tatti, F. 2011. Carbon Nanowall Growth on Carbon Paper by Hot Filament Chemical Vapour Deposition and Its Microstructure. *Carbon*, 49, 2134-2140.

- Mackie, N. M., Dalleska, N. F., Castner, D. G. & Fisher, E. R. 1997. Comparison of Pulsed and Continuous-Wave Deposition of Thin Films from Saturated Fluorocarbon/H₂ Inductively Coupled Rf Plasmas. *Chem Mat*, 9, 349-362.
- Malesevic, A., Vizireanu, S., Kemps, R., Vanhulsel, A., Haesendonck, C. V. & Dinescu, G. 2007. Combined Growth of Carbon Nanotubes and Carbon Nanowalls by Plasma-Enhanced Chemical Vapor Deposition. *Carbon*, 45, 2932-2937.
- Masaru, H. & Toshio, G. 2006. Progress of Radical Measurements in Plasmas for Semiconductor Processing. *Plasma Sources Science and Technology*, 15, S74.
- Milev, A., Wilson, M., Kannangara, G. S. K. and Tran, N. (2008). X-ray diffraction line profile analysis of nanocrystalline graphite. *Materials Chemistry and Physics*, 111(2-3), 346-350.
- Mizuno, K., Ishii, J., Kishida, H., Hayamizu, Y., Yasuda, S., Futaba, D. N., Yumura, M. & Hata, K. A Black Body Absorber from Vertically Aligned Single-Walled Carbon Nanotubes. Proceedings of the National Academy of Sciences of the United States of America, 29 January 2009 2009. Highwire Press, 6044.
- Mori, T., Hiramatsu, M., Yamakawa, K., Takeda, K. & Hori, M. 2008. Fabrication of Carbon Nanowalls Using Electron Beam Excited Plasma-Enhanced Chemical Vapor Deposition. *Diamond and Related Materials*, 17, 1513-1517.
- Naito, S., Ikeda, M., Ito, N., Hattori, T. & Gotto, T. 1993. Effect of Rare Gas Dilution on CH₃ Radical Density in Rf-Discharge CH₄ Plasma. *Japan Journal Applied Physics*, 32, 5271.
- Ni, Z. H., Fan, H. M., Fan, X. F., Wang, H. M., Zheng, Z., Feng, Y. P., Wu, Y. H. & Shen, Z. X. 2007. High Temperature Raman Spectroscopy Studies of Carbon Nanowalls. *Journal of Raman Spectroscopy*, 38, 1449-1453.
- Novoselov, K. S., Geim, A. K., Morozov, S. V., Jiang, D., Zhang, Y., Dubonos, S. V., I.V.Grigorieva & Firsov, A. A. 2004. Electric Field Effect in Atomically Thin Carbon Films. *Journal of Science*, 306, 666.
- Rummeli, M. H., Bachmatiuk, A., Borrnert, F., Schauffel, F., Ibrahim, I., Cendrowski, K., Martynkova, G. S., Placha, D., Palen, E. B., Cuniberti, G. & Buchner, B. 2011. Synthesis of Carbon Nanotubes with and without Catalyst Particles. *Nanoscale Research Letters*, 6, 303.

- Sahoo, S. C., Mohapatra, D. R., Lee, H.-J., Jejurikar, S. M., Kim, I., Lee, S.-C., Park, J.-K., Baik, Y.-J. & Lee, W.-S. 2014. Carbon Nanoflake Growth from Carbon Nanotubes by Hot Filament Chemical Vapor Deposition. *Carbon*, 67, 704-711.
- Sato, K., Saito, R., Oyama, Y., Jiang, J., Cançado, L. G., Pimenta, M. A., et al. (2006). D-band Raman intensity of graphitic materials as a function of laser energy and crystallite size. *Chemical Physics Letters*, 427(1-3), 117-121.
- Seiya, S., Anindita, C., Chia-Liang, C. and Masamichi, Y. (2011). Effect of Hydrogen on Carbon Nanowall Growth by Microwave Plasma-Enhanced Chemical Vapor Deposition. *Japanese Journal of Applied Physics*, 50(1S1), 01AF08.
- Seo, D. H., Kumar, S. and Ostrikov, K. (2011). Control of morphology and electrical properties of self-organized graphenes in a plasma. *Carbon*, 49(13), 4331-4339.
- Shang, N. G., Au, F. C. K., Meng, X. M., Lee, C. S., Bello, I. & Lee, S. T. 2002. Uniform Carbon Nanoflake Films and Their Field Emissions. *Chemical Physics Letters*, 358, 187-191.
- Shih, W.-C., Jeng, J.-M., Huang, C.-T. & Lo, J.-T. 2010. Fabrication of Carbon Nanoflakes by Rf Sputtering for Field Emission Applications. *Vacuum*, 84, 1452-1456.
- Shiji, K., Hiramatsu, M., Enomoto, A., Nakamura, M., Amano, H. & Hori, M. 2005. Vertical Growth of Carbon Nanowalls Using Rf Plasma-Enhanced Chemical Vapor Deposition. *Diamond and Related Materials*, 14, 831-834.
- Shimabukuro, S., Hatakeyama, Y., Takeuchi, M., Itoh, T. & Nonomura, S. 2008. Effect of Hydrogen Dilution in Preparation of Carbon Nanowall by Hot-Wire Cvd. *Thin Solid Films*, 516, 710-713.
- Shiomi, T., Nagai, H., Kato, K., Hiramatsu, M. & Nawata, M. 2001. Detection of C₂ Radicals in Low-Pressure Inductively Coupled Plasma Source for Diamond Chemical Vapor Deposition. *Diamond and Related Materials*, 10, 388-392.
- Shin, C., Iftiqar, S. M., Park, J., Kim, Y., Baek, S., Jang, J., Kim, M., Jung, J., Lee, Y., Kim, S. & Yi, J. 2013. Optimization of Intrinsic Hydrogenated Amorphous Silicon Deposited by Very High-Frequency Plasma-Enhanced Chemical Vapor Deposition Using the Relationship between Urbach Energy

- and Silane Depletion Fraction for Solar Cell Application. *Thin Solid Films*, 547, 256-262.
- Srivastava, S. K., Shukla, A. K., Vankar, V. D. & Kumar, V. 2005. Growth, Structure and Field Emission Characteristics of Petal Like Carbon Nano-Structured Thin Films. *Thin Solid Films*, 492, 124-130.
- Tzeng, Y., Chen, W. L., Wu, C., Lo, J.-Y. & Li, C.-Y. 2013. The Synthesis of Graphene Nanowalls on a Diamond Film on a Silicon Substrate by Direct-Current Plasma Chemical Vapor Deposition. *Carbon*, 53, 120-129.
- Wang, J. J., Zhu, M. Y., Outlaw, R. A., Zhao, X., Manos, D. M., Holloway, B. C. & Mammana, V. P. 2004. Freestanding Subnanometer Graphite Sheets. *Journal of Applied Physics*, 85, 1265.
- Wang, Y., Li, J. & Song, K. 2014. Study on Formation and Photoluminescence of Carbon Nanowalls Grown on Silicon Substrates by Hot Filament Chemical Vapor Deposition. *Journal of Luminescence*, 149, 258-263.
- Wang, Z. J., Tian, X. B., Gong, C. Z., Shi, J. W., Yang, S. Q., Fu, R. K. Y. & Chu, P. K. 2012. Plasma Immersion Ion Implantation into Cylindrical Bore Using Internal Inductively-Coupled Radio-Frequency Discharge. *Surface and Coatings Technology*, 206, 5042-5045.
- Wind, S. J., Appenzeller, J., Martel, R., Derycke, V. & Avouris, P. 2002. Vertical Scaling of Carbon Nanotube Field-Effect Transistors Using Top Gate Electrodes. *Journal of Applied Physics Letters*, 80, 3817.
- Wu & Yang 2002. Effects of Localized Electric Field on the Growth of Carbon Nanowalls. *Nano Letters*, 2, 355-359.
- Wu, Y., Qiao, P., Chong, T. & Shen, Z. 2002a. Carbon Nanowalls Grown by Microwave Plasma Enhanced Chemical Vapor Deposition. *Advanced Materials*, 14, 64-67.
- Wu, Y., Yang, B., Zong, B., Sun, H., Shenc, Z. & Fengc, Y. 2003. Carbon Nanowalls and Related Materials. *Journal of materials chemistry*, 14, 469-477.
- Wu, Y. H., Yang, B. J., Han, G. C., Zong, B. Y., Ni, H. Q., Luo, P., Chong, T. C., Low, T. S. & Shen, Z. X. 2002b. Fabrication of a Class of Nanostructured Materials Using Carbon Nanowalls as the Templates. *Advanced Functional Materials*, 12, 489-494.

- Xu, X., Ge, H., Wang, S., Dai, Z., Wang, Y. & Zhu, A. 2009. Influence of the Low-Frequency Source Parameters on the Plasma Characteristics in a Dual Frequency Capacitively Coupled Plasma Reactor: Two Dimensional Simulations. *Progress in Natural Science*, 19, 677-684.
- Ye, Q., Cassell, A. M., Liu, H., Chao, K.-J., Han, J. & Meyyappan, M. 2004. Large-Scale Fabrication of Carbon Nanotube Probe Tips for Atomic Force Microscopy Critical Dimension Imaging Applications. *Journal of Nano Letters*, 7, 1301.
- Xiong, G., Hembram, K. P. S. S., Zakharov, D. N., Reifenberger, R. G. and Fisher, T. S. (2012). Controlled thin graphitic petal growth on oxidized silicon. *Diamond and Related Materials*, 27–28(0), 1-9.
- Yoshimura, H., Yamada, S., Yoshimura, A., Hirose, I., Kojima, K. and Tachibana, M. (2009). Grazing incidence X-ray diffraction study on carbon nanowalls. *Chemical Physics Letters*, 482(1–3), 125-128.