

**Study of N₂ adsorption on metal oxides modified Na-Y zeolite:
Equilibrium and kinetic**

Chieng, Y. Y.¹, Kamarudin, K. S. N.¹, Mat, H. B.¹, and Hamdan, H.²

¹Advanced Process Engineering (APEN) Research Group,
Faculty of Chemical and Natural Resources Engineering,

²Zeolite and Porous Material Group,
Ibnu Sina Institute for Fundamental Science Studies,
Universiti Teknologi Malaysia,
81310 Skudai, Johor,
Malaysia.

Email: yychieng@yahoo.com

ABSTRACT

Adsorption equilibrium and kinetic study of gas N₂ on metal oxides modified Na-Y zeolite have been undertaken in order to elucidate the fundamental mechanism of adsorbate-adsorbent interactions. Studies were carried out by measuring both the adsorption isotherm and the FTIR spectra. Results indicate that the electronegativity and valent variations of metal cations greatly affect the gas N₂ adsorption capacity as well as kinetic transport. The isosteric heat of adsorption that measured and N₂ adsorption bands that observed in FTIR spectra also demonstrate that N₂ interact strongly with metal oxides modified Na-Y zeolite.

Keywords: N₂ adsorption; equilibrium; kinetic; metal oxides; Na-Y zeolite.