SPECTROSCOPIC STUDIES OF ERBIUM DOPED ZINC TELLURITE GLASS

EMBEDDED WITH GOLD NANOPARTICLES

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То

My beloved family....

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ABSTRACT

Series of glasses based on (70-x)TeO₂ - 30ZnO - 0.5Er₂O₃ - xAu, where $0.0 \le x \le 0.5$ mol % were prepared using melt-quenching technique. The nature of the glass was confirmed using X-ray diffraction and the size of Au NPs (GNPs) was estimated by using transmission electron microscope image. It was found that the glass was amorphous in nature with an average diameter size of GNPs of about 2.6 ± 1.6 nm. The thermal stability of the glass was determined by means of differential thermal analysis (DTA). From the DTA curve, the glass transition temperature (T_g) , crystallization temperature (T_c) and melting temperature (T_m) was identified and thermal stability was calculated. The structural characteristic was investigated using Fourier transform infra red spectroscopy in the range of 400 cm⁻¹ to 3500 cm⁻¹. The spectra of the glass shows two absorption peaks around 646 cm⁻¹ and 748 cm⁻¹ corresponding to the TeO₄ and TeO₃ structural units, respectively. The absorption properties of these samples were obtained by using UV-Visible spectrometer which showed six absorption peaks. The values of optical band gap (E_{opt}) and Urbach energy (E_u) were calculated from the absorption peak and were found to be in the range of 2.57 - 2.64 eV and increased as the concentration of GNPs was increased, while (E_u) was found in the range 0.39 - 0.41 eV. The emission characteristic of this glass was characterized using photoluminescence spectroscopy. It was observed that there were two distinctive emissions centered at 650 nm and 845 nm which are assigned to ${}^{4}F_{9/2}$ - ${}^{4}I_{15/2}$ and ${}^{4}I_{13/2}$ - ${}^{4}I_{15/2}$ transition, respectively under 488 nm excitation wavelength. It was observed that sample of 0.3 mol % GNPs exhibits the highest enhancement due to energy transfer from GNPs to Er^{3+} ion.

ABSTRAK

Beberapa siri kaca berdasarkan (70-x)TeO₂ – 30ZnO – 0.5Er₂O₃ – xAu, dengan $0.0 \le x \le 0.5$ mol % telah berjaya disediakan dengan menggunakan teknik pelindapan leburan. Sifat kaca telah disahkan dengan menggunakan analisis pembelauan sinar-X dan saiz zarah emas, Au dianggarkan menggunakan imej mikroskop imbasan elektron. Hasil menunjukkan bahawa kaca adalah bersifat amorfos dengan purata diameter saiz zarah emas adalah 2.6 ± 1.6 nm. Kestabilan terma kaca telah ditentukan dengan menggunakan analisis pembezaan terma (DTA). Melalui analisis lengkuk pembezaan terma ini parameter seperti suhu transisi kaca (Tg), suhu penghabluran kaca (Tc), suhu lebur (Tm) telah dikenalpasti dan kestabilan terma juga telah dihitung. Ciri struktur kaca pula dikaji menggunakan spektroskopi infra merah tansformasi Fourier dalam julat 400 cm⁻¹ hingga 3500 cm⁻¹. Spektrum kaca ini menunjukkan dua puncak sekitar 646 cm⁻¹ dan 748 cm⁻¹ yang masingmasing mewakili unit struktur TeO₄ dan TeO₃. Sifat penyerapan kaca pula diperoleh dengan menggunakan spekroskopi UV-Vis yang memberikan enam puncak serapan. Nilai bagi jurang tenaga optik (E_{opt}) dan tenaga Urbach (E_u) telah dihitung daripada puncak serapan dan didapati berada dalam julat 2.57 - 2.64 eV dan meningkat dengan pertambahan kepekatan zarah nano emas manakala nilai E_u berada dalam julat 0.39 - 0.41 eV. Ciri pancaran kaca ini pula dianalisis menggunakan spektroskopi fotoluminesen. Terdapat dua puncak pancaran yang ketara masingmasing pada 645 nm dan 845 nm, yang merujuk kepada transisi ${}^{4}F_{9/2}$ - ${}^{4}I_{15/2}$ dan ${}^{4}I_{13/2}$ - ${}^{4}I_{15/2}$, di bawah pengujaan panjang gelombang 488 nm. Sampel yang mengandungi zarah nano emas sebanyak 0.3 mol % mempunyai keamatan pancaran tertinggi disebabkan perpindahan tenaga dari GNPs ke ion Er^{3+} .

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LIST OF SYMBOLS

T _c	Crystallization Temperature
T _g	Transformation Temperature
n	Refractive index
d	Atomic distance
λ	Wavelength
θ	Angle
ω	The angular frequency
α	The absorption coefficient
E _{opt}	Optical band gap energy
E _u	Urbach energy
ħ	Plank constant
ΔΕ	Width of band tail

LIST OF ABREVIATIONS

GNPs	Gold nanoparticles	
XRD	X-ray diffraction	
FT-IR	Fourier transformed infrared	
TEM	Transmission electron microscopy	
UV	Ultra violet	
Vis	Visible	
ASTM	American Society for Testing Material	
PL	Photoluminescence	
NIR	Near infra red	
MIR	Middle Infrared	
RE	Rare-earth	
Тр	Trigonal pyramid	
Tbp	Trigonal bypyramid	
Er ₂ O ₃	Erbium oxide	
ZnO	Zinc oxide	
NR	Non radiatively	
CET	Cooperative energy transfer	
ET	Energy transfer	
LSPR	Localized surface plasmon resonance	
ESA	Energy state absorption	

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CHAPTER 1

INTRODUCTION

1.1 Introduction

Glasses are one of the oldest as well as one of the newest materials in the world that find a variety of used in everyday life. Glass is an amorphous material that can be formed by some techniques such as sol gel processing of solution, vapor deposition, melt quenching or by neutron irradiation of crystalline material (Doremus and Narottam, 1986) but by all these techniques, cooling from the liquid states is the most important and most widely used (Uhlmann and Kreidel,1983). Glass material is not new. It has been used since 4000 years ago in Egypt. It is different from crystalline material because it does not possess long range order and has no periodic arrangement. The interesting fact about glass is it acquires many unique properties such transparency, clear and not easily corroded.

Tellurite glasses are finding widespread applications in the field of photonics like chalcogenide glasses due to some characteristics features such as low phonon energy, high refractive index and excellent infrared transmission (Zakery, *et al.*, 2007). It has been reported that, these glasses are good candidates for hosting rareearth (RE) ions because of their low-phonon energy environment, good optical properties and chemical durability (Sun, 1998) (Erariah and Bull, 2010). The erbiumdoped tellurite glass also have shown optical and chemical properties suitable for photonic applications such as optical limiter, laser light modulator and are thermally stable for fiber drawing.

Recently, metallic nanostructures is a subject of considerable interest because they are endowed with unique optical properties and functionalities contrast to their bulk counterparts as reported by Som et al. (2010). In addition, tellurite glass embedded with metallic nanoparticles (NPs) recently has been the subject of intensive investigations because of their ability to enhance the luminescence of tellurite glass. Previous studies have been shown that glasses doped with RE ions and metallic NPs have the ability to enhance the luminescence and improved nonlinear optical properties of tellurite glass They are becoming promising for practical applications in photonics as optical amplifier, optical recording, laser active media and infrared-to-visible converters (Carmo *et al.*, 2009). Nanoparticles (NPs) may originate changes of the material's luminescence characteristics as well as enhancement of the nonlinear optical properties (Kassab *et al.*, 2008).

For enhanced efficiency and improved up-conversion, absorption crosssection has to be increased. Most concepts for this enhancement rely on energy transfer (ET) from a species with a large absorption cross-section to RE ion. Use of two or more rare-earth ions together and energy transfer between them (Kumar, 2007) or use of the metallic NPs with RE ions has been found a successful way to enhance luminescence of the glass (Kassab, 2011) (Piasecki *et,al.*, 2010). In order to get enhancement on optical characteristics in these nano-composite glasses, the concentration of RE ions should be low enough to avoid the quenching effect. Therefore, glasses containing metallic NP doped with low concentration of RE ions are particularly of interest to us.

In spite of few spectroscopic studies, the role of NPs and enhancement of optical properties of tellurite glass is far from being understood. The spectroscopic characterizations of this tellurite nanocomposite are essential for the optimization and applications that needs careful sample preparation, fabrication and spectroscopic investigations. In this study, tellurite glass has been used as a host while erbium oxide as a dopant and will be embedded with gold nanoparticles (GNPs) in order to investigate the effects of concentration of metallic nanoparticles by means of spectroscopic characteristics. The relation between the concentration and spectroscopic characteristic used to determine the optimum composition of GNPs to prepare a high effiency of tellurrite glass as a host especially in laser system and photonic field.

1.2 Problem Statements

Rare-earth doped tellurite glasses embedded with metallic nanoparticles has been receiving special interest due to their ability to enhance the performance of tellurite glass for their applications. Even though there are numbers of research on tellurite glass has been done, yet the characteristics of Er^{3+} ions doped glass embedded with gold nanoparticles (GNPs) has not been fully investigated. In addition, in spite of few experiments on tellurite doped GNPs , the clear explanation about the role played by GNPs are still lacking. Few studies have been done in this system but are limited to certain properties. Therefore, in the present study we will investigate the role played by GNPs on optical properties by spectroscopic techniques which are UV-Vis, Photoluminescence and FTIR spectrometer.

1.3 Research Objectives

The objectives of this work are:

- i) To prepare a series of glass samples based on by melt quenching technique.
- ii) To determine the existence of NPs.
- iii) To determine the thermal parameters.
- iv) To characterize the vibrational spectroscopy of the glass
- v) To characterize the optical properties of the glass.
- vi) To characterize the emission behavior of the glass.

1.4 Scope of Study

In order to achieve the objectives, this research has been carried out within the stated scope as follows:

- i) The preparations of $(70-x)Te_2O 30ZnO 0.5Er_2O_3 xAu$ glass where x = (0.1, 0.2, 0.3, 0.4, 0.5) mol% by melts quenching technique. The composition of glass is chosen due to the high stability and highest glass forming ability as investigated Sahar et.al., 2013.
- The determination of NPs using Transmission Electron Microscopy (TEM).
- iii) The determination of thermal parameters using Diffrential Thernal Analysis (DTA).

- iv) The characterization of optical properties using UV-Vis Spectroscopy.
- v) The characterization of vibrational spectrum using Fourier Transform Infra Red (FTIR) Spectroscopy.
- vi) The characterization of emission behavior using Photoluminescence (PL) Spectroscopy.

1.5 Significance of study

This study is fundamentally important to explain the role played by gold nanoparticels inside tellurite glass by analyzing the absorption, emission, and FTIR spectra of the glass. Hence, from this systematic experimental of fabrication and spectroscopic studies of this sample it would give better suggestions about how to increase the efficiency of tellurite glass for their applications.

REFERENCES

- Akshaya, K., Rai, D.K and Rai, S.B. (2002). Optical studies of Eu³⁺ ions doped in tellurite glass. *Spectrochimica Acta Part A*. 58, 2115-2125.
- Almeida, R., Silva, D.M., Luciana R.P. Kassab, L.R.P (2011). Eu³⁺ luminescence in tellurite glasses with gold nanostructure. *Optics Communications*. 281, 108– 111
- Arlt, G. and Schweppe, H. (1968). Paratellurite. A new piezoelectric maternal. *Journal* of Solid State Communications. 6, 783-784.
- Balda, R., Al-Saleh, M., Miguel, A., Navarro, J.M., Fernandez, J (2011). *Optical Materials*. 34, 481-486.
- Banijamali, S., Eftekhari, B., and Aghaei, A.R (2012). The effect of ionic and metallic silver on the crystalline phases developed in CaO-Al₂O₃-TiO₂-P₂O₅ glasses. *Journal of Non-Cryst. Solids.* 358, 303- 310.
- Burger, H., Kniepp, K., Hobert, H., and Vogel, W. (1992). Glass formation, properties and structure of glasses in the TeO₂- ZnO system. *Journal of Non-Crystalline Solids*. 151, 134-142.
- Cai, G.F.W., C. Kan, C. Li, C.K.C., Zhang, L (2003). Controllable optical properties of Au/SiO₂ nanocomposite induced by ultrasonic irradiation and thermal annealing. *Journal of Applied. Physics*. 83,36-39.

- Carmo, A. P., Bell. M. J. V., Anjos, V., Almeida, Ricardo de., Silva, Davinson M. da., And Kassab, L. R. P. (2009). Thermo-optical properties of tellurite glasses doped with Eu³⁺ and Au nanoparticles. *Journal of Applied Physics*. 42.
- Doremus, R.H., AND Narottam P.B. (1986). *Handbook of Glass Properties*. Newyork: Academic Press,: 501.

Doremus, R.H., (1994). Glass Science. 2nd edition. New York: Willey Interscience.

Dousti, M. R., Sahar, M.R., Ghoshal, S.K., Amjad, R. J., Samavati, A. R (2013). Effect of AgCl on spectroscopic properties of erbium doped zinc tellurite glass. *Journal of molecular Structure*. 1035, 6-12.

El-Mallawany, R.A. (2002). Tellurite Glass Handbook. Florida : Press LLC.

- El-Mallawany, R.E., Patra, A., Christopher, S., Kapoor, R., Prasad, P.N (2004). Study of luminescence of properties Er³⁺-ions in new tellurite glasses. *Optical Materials*. 26, 267–270.
- El- Mallawany, R.E., Abdalla, M.D., Ahmed, I.A (2008). *Materials Chemistry and Physics*. 109, 291-296.
- Erariah, B., and Bull (2010). Optical properties of tellurite glass doped with samarium trioxide. *Journal of Nonlinear Optical Physics and Material*. 33, 391-394.
- Fortes, Luis M., Santos, Luis F., Clara Goncalves, M.and Almeida, Rui M. (2006). The Effects of ZnCl₂ and ErCl₃ on the vibrational spectra and structure of tellurite glasses. *Journal of Non-Crystalline Solids*. 352, 690-694.

Giancolli, D. C (1985). Physics: Principle with applications. New Jersey: Princle Hall.

- Golberg, P. (1996). *Luminescence of Inorganic Solids*. New York and London: Academic Press.
- Heiman, D (2004). Photoluminescence Spectroscopy. Northeastern University.
- Hou, Z., Xue, Z., Wang, S., Hu., X., Zhou., Y., and Niu, C (2012). Thermal stability and and structure of tellurite glass. *Engineering material*. 512-515. 994-997.
- Hu, L. and Jiang, Z (1996). Properties and structures TeO₂ based glasses containing Ferroelectric components. *Phy. Chem. Glasses.* 371, 19-21.
- Jaba, N., Mermet, A., Duval, E., and Champagnon, B. (2005). Raman spectroscopy studies of Er³⁺-doped zinc tellurite glasses. *Journal of Non-Crystalline Solids*. 351.
- Kamalaker, V., Upender, G., Prasad, M., and Mouli, V. C (2010). Infrared, ESR and Optical absorption studies of Cu²⁺ ions doped in TeO₂- ZnO-NaF glass system. *Indian Journal of Pure and Applied Physics*. 48, 709-715.
- Kassab, L.R.P., Camilo, M.E., Amancio, C.T., Silva, D.M., Martinelli, J.R (2008). Effects of gold nanoparticles in the green and red emissions of TeO₂–PbO–GeO₂ glasses doped with Er³⁺– Yb³⁺. *Optical Materials*. 33, 1948–1951.
- Khatir. S., Romain. F., Portier. J., Rossignol. S., Tanguy. B., Videau. J.J, and Turrell. S. Study of Optical Properties of TeO₂-PbO-TiO₂ Glass System. *Journal* of Molecular Structure. 298 (1993) 13.
- Kumar, R (2007). Frequency Up-Conversion in rare-earth ions. *Applied Physics*. 30. 1-17.

- Laksiminarayana, G., Qiu, J. (2009). Enhancement of Pr³⁺ Luminescence in TeO₂ ZnO
 -Nb₂O₅-MO₃ Glasses Containing Silver Nanoparticles. Journal of Alloys an
 Compounds. 478: 630-635.
- Liu, J. G., Mason, P., and Hilton, F (1996). Detection of rapid erosion in SE spain. *Journal of Remote Sensing*. 17, 1005-1018.
- Mazzuca, M., Portier, J., Tanguy, B., Romain, F., Fadli, A., and Turrell, S. (1995).
 Raman scattering in tellurium-metal oqyde glasses. *Journal of Molecular Structure*. 349, 41
- Mott, N. F., and Davis, E.A. (1979). Electronic process in Non-Crystalline Materials. Oxford: Claredon.
- Mott, N.F. (1993). Superconductivity in chalchogenide glases. *Journal of Non-Crystalline Solids*. 164-166 (Part 2), 1177-1178.
- O' Donnel, M. (2004). Tellurite And Flurotellurite Glasses For A Ative And Passive Fiberoptic Waveguide. Doctor of Philosophy, University of Nottingham.
- Podmaniczky, A. (1976). Some properties of TeO₂ light deflectors with small interaction length. *Opt.Commun.* 16(1), 161-165.
- Prasad, P.N., Kapoor, K., Jain, H., Toulouse, J., and Pattnaik, R.K. Crystallization of TeO₂ – Nb₂O₅ glasses and their network structure. Journal of material science. 5597, 136-142.
- Rivera, V.A.G., Ledemi, Y., Osorio, S.P.A., Manzani, D., Messaddeq, Y., Nunes, L. A.
 O., and Marege, E. Jr. (2011). Growth of silver nano-particle embedded in tellurite glass: Interaction between localized surface plasmon resonance and Er³⁺ ions. *Journal of optical material*. 33, 888-892.

- Rivera, V.A.G., Ledemi, Y., Osorio, S.P.A., Manzani, D., Messaddeq, Y., Nunes, L. A.
 O., and Marege, E. Jr. (2011). *Plamonic coupling in Er³⁺ : Au tellurite glass*.
 Brazil: Springer.
- Rivera, V.A.G., Ledemi, Y., Osori, S.P.A., Marega Jr, E (2012). Efficient plasmonic Coupling between Er3+:(Ag/Au) in tellurite glasses. *Journal of Non-Crystalline Solids*. 358 399–405.
- Rosmawati, S., Sidek, H.A.A., Zainal, A. T., and Mohd, Z. T (2007). IR and UV spectral Studies of zinc tellurite glasses. *Journal of Applied Sciences*. 7(20), 3051-3056.
- Sahar, M.R., Jehbu A.K., Karim, M.M (1997). TeO₂- ZnO-ZnCl₂ glasses for IR transmission. *Journal of Non-Crystalline Solids*. 213 & 214, 164-167.

Sahar, M. R. (1998). Sains Kaca. Penerbit Universiti Teknologi Malaysia.

- Sahar, M.R., Sulhadi.K., Rohani, M.S. (2007). Spectroscopic studies of TeO₂ ZnO-Er₂O₃ glass system. *Journal of Matter Science*. 42, 824-827.
- Sahar, M.R., Sulhadi, K., and Rohani, M. S. (2008). The preparation and structural studies in the (80-x) TeO₂ – 20ZnO-(x)Er₂O₃ glass system. *Journal of Non-Crystalline Solids*. 354, 1179- 1181.
- Sahar., M.R., Dousti., M.R., Goshal, S.K., Arifin, R., and Amjad, R.J (2013). Plasmonic Enhanced luminescence in Er³⁺: Ag co-doped tellurite glass. *Journal of Molecular Structure*. 1033, 79-83.
- Sidek, H.A.A., Rosmawati, S., Talib, Z. A., Halimah, M. K and Daud, W. M (2009). Synthesis and optical properties of ZnO-TeO₂ glass system. *American Journal of Applied Sciences*. 6 (8): 1489-1494.

- Sekiya. T., Mochida. N., and Soejima. A. Raman spectra of binary tellurite glasses containing tri- or tetra-valent cations. *Journal of Non-Cryst. Solid.* 191 (1995) 115.
- Setsuhisa Tanabe, S., Hayashi, H., Hanada, T., Onodera, N (2002). Fluorescence properties of Er³⁺ ions in glass ceramics containing LaF₃ nano,crystal. *Optical Materials* 19, 343–349.
- Shelby, J. E. (2005). *Introduction to Glass Science and Technology*. 2nd ed. Cornwall UK. The Royal Society of Chemistry.
- Sidebottom, D. L., Hruschka, M. A., Potter, B.G., Brow, R.K. (1997). Structure and optical properties of rare earth-doped zinc oqyhalide tellurite glasses. *Journal of Non-Crystalline Solids*. 222, 282-289.
- Silva, A.P., Carmo, A.P., Anjos, V., Bell, M. J. V. Kassab, L. R. P., Pinto, and Ricardo de Almeida (2011). Temperature coefficient of optical path of tellurite glasses doped with gold nanoparticles. *Optical Materials*. 34, 239-243.
- Singh, S. K., Giri, N. K., Rai, D. K., and Rai, S. B. (2010). Enhanced upconversion emission in Er³⁺-doped tellurite glass containing silver nanoparticles. *Solid State Sciences*. 12, 1480-1483.
- Sulhadi, Sahar, M. R., Rohani, M. S., and Arifin, R (2007). Thermal stability and structural studies in the TeO₂ –ZnO-MgO-Li₂O-Er₂O₃ glass system. *Solid State Sciences and Technology*. 15, 116-121.
- Som, Tirtha., and Karmakar, Basudeb (2010). Surface plasmon resonance and enhanced Fluorescence application of single-step synthesized elliptical nano gold-embedded antimony glass dichroic nanocamposite. *Plamonic*. 5, 149-159.

- Som, Tirtha., and Karmakar, Basudeb (2009). Enhancement of Er³⁺ Upconverted luminescence in Er³⁺: Antimony glass dichroic nanocamposites containing hexagonal Au nanoparticles. J.Opt.Soc.Am.B 26:12.
- Souza, J.R. and Olivares, R. A (2002). Analysis of power transients in erbium-doped fiber amplifier, with application to wavelength routed optical network. Journal of Microwaves and Optoelectronics. 2, 21-33.
- Thomas, R.L., Vasuja., Hari, M., Radhaksihnan.P (2011). Optical non-linearity in ZnO doped TeO₂ glasses. *Journal of Optoelectronics and advanced materials*. 13(5), 523-527.
- Ueda, Jumpei., Tanabe, Setsuhisa., and Ishida, Akito (2009). Surface plasmon excited infra-red- to-visible upconversion in Er³⁺-doped transparent glass ceramics. *Journal of Non- Crystalline Solids*. 355, 1912-1915.
- Warner, A., White, D., and Bonther, V. (1972). Acoustic-Optic light deflectors using optical activity in paratellurite. *Journal of Applied Physics*. 43(11), 4489-4495.
- Z, Dutton. And Cooper (1966). Introduction to tellurite glasses. El-Malawany. *Tellurite Glasses Handbook*. (pp.1). Florida. Press:LLC
- Zaki, H. M., (2005). AC conductivity and frequency dependence of dielectric properties for copper doped magnetite. *Physica B*. 363, 232-244.