EFFECTS OF HEAVY METALS AND SELECTED CARBON COMPOUNDS ON THE GROWTH OF BACTERIUM CA

HAKAR MOHAMMED SALEH

UNIVERSITI TEKNOLOGI MALAYSIA

EFFECTS OF HEAVY METALS AND SELECTED CARBON COMPOUNDS ON THE GROWTH OF BACTERIUM CA

HAKAR MOHAMMED SALEH

A dissertation submitted in partial fulfillment of the requirements for the award of the degree of Master of Science (Biotechnology)

Faculty of Biosciences and Medical Engineering
Universiti Teknologi Malaysia

Dedicated to my beloved parents, and my siblings

ACKNOWLEDGEMENT

I thank Allah for made this work easier than I thought, achievable and made it real.

I am grateful to Dr. Chong Chun Shiong, whose supervision was positive, he has guided, hinted and motivated me throughout the work. I wish to thank and appreciate him for his patience and all his efforts. I shall not forget to thank my colleagues at Enzyme Research lab, T02, UTM and the helpful lab assistant.

I would have never done this if my parents would not have made it possible through their support, being positive and their patience, I can only thank you and this must be consider as achievement for you as well.

ABSTRACT

Nowadays environment has been suffering from the contamination by heavy metals and organic compounds. Modernization, urbanization and industrialization further increase the levels of contamination. In this work, the ability of bacterium CA previously isolated from textile waste water was investigated to tolerate to aluminium, copper, cobalt, zinc and manganese at different concentrations. The MIC were found to be 400, 200, 40, 60, 300 mg/L, respectively, the growth rate at MIC were 0.1732, 0.417, 0.0885, 0.2824 and 0.4046 h⁻¹, respectively, and the doubling time values at MIC were 4.005, 1.66187, 7.83, 2.4539 and 1.7128 h⁻¹, respectively. Different organic compounds were supplied as sole carbon sources at chemically defined media. The bacterium CA could grow on the CDM provided with casamino acids, casamino acids (as both carbon and nitrogen source), glucose and glycerol with growth rate of 0.0297, 0.0317, 0.04 and 0.0492 h⁻¹, respectively. Whereas the bacterium failed to grow when sulfanilic acids, acrylamide and benzene were only source of carbon. Bacterium CA was found to be sensitive to tetracycline hydrochloride and kanamycin sulfate with clear zones of 2.8 and 2.6 cm, respectively, while it was resistant to XY-12 and ampicillin trihydrate. Moreover, bacterium CA was found to be closest with Bacillus thuringiensis with accession number NR_102506 via 16S rRNA gene analysis. In conclusion, bacterium CA was positive to tolerate to heavy metals, results were significant in case of aluminium, manganese and copper. Bacterium CA was able to grow on different organic compounds that indicated the presence of different mechanisms of carbon source up take and metabolism in the bacterium system. The obtained results suggested the bacterium CA could be potentially used for the treatment of wastewater contamination containing heavy metals and organic compounds.

ABSTRAK

Pada masa kini persekitaran telah mengalami pencemaran logam berat dan sebatian organik. Pemodenan, perbandaran dan pengindustrian meningkatkan lagi aras pencemaran. Dalam kajian ini keupayaan bakteria CA yang sebelum ini diasingkan daripada air sisa tekstil untuk bertolak ansur-dengan aluminium, kuprum, kobalt, zink dan mangan pada kepekatan yang berbeza. disiasat dan kepekatan renjatan minimum (MIC) adalah 400 200, 40, 60, 300 mg/L, masing-masing. pertumbuhan bakteria pada MIC adelah 0.1732, 0.417, 0.0885, 0.2824 dan 0.4046 h-1, masing-masing, dan doubling masa nilai pada MIC adalah 4.005, 1.662, 7.83, 2.4539 dan 1.7128 h⁻¹, masing-masing. Sebatian organik yang berbeza dibekalkan sebagai sumber karbon yang tunggal di media CDM. Bakteria CA dapat bertumbuh di CDM yang dibekalkan dengan asid casamino, asid casamino (bertindak sebagai sumber karbon dan nitrogen), glukosa dan glycerol dengan kadar pertumbuhan sebanyak 0.0297, 0.04, 0.0317 dan 0.0492 h⁻¹, masing-masing. Manakala bakteria gagal tumbuh di media CDM apabila asid sulfanilic, acrylamide dan benzene dibekalkan sebagai sumber karbon. Bakteria CA telah didapati sensitif kepada tetracycline hydrochloride dan kanamycin sulfat dengan zon-zon yang jelas 2.8 cm dan 2.6 cm masing-masing. Bakteria CA didapati tidak sensitif terhadap XY-12 dan ampicillin trihydrate. Selain itu, bakteria CA telah dikenalpasti sebagai Bacillus thuringiensis (NR_102506) melalui kaedah analisis 16S rRNA gen. Kesimpulannya, bakteria CA positif untuk bertolak ansur untuk logam berat terutamanya, aluminium, mangan dan tembaga. Bakteria CA ini telah berjaya bertumbuh dengan menggunakan sebatian organik yang berbeza sebagai sumber karbon menunjukkan kehadiran mekanisma yang berbeza di dalam sistem bakteria. Keputusan yang diperolehi mencadangkan bakteria CA berpotensi untuk digunakan untuk rawatan pencemaran logam berat dan sebatian organik.

TABLE OF CONTENTS

CHAPTER	TITLE DECLARATION		
	DED	DICATION	iii
	ACK	NOWLEDGEMENT	iv
	ABS	TRACT	V
	ABS	TRAK	vi
	TABLE OF CONTENTS LIST OF TABLES LIST OF FIGURES LIST OF ABBREVIATION		
	LIST	T OF APPENDIX	xvii
1	INTRODUCTION		
	1.0	Background	1
	1.1	Significance of the Work / Problem Statement	2
	1.2	Objectives	3
	1.3	Study scope	3
2	LITERATURE REVIEW		
	2.1	Heavy Metals Contamination: Global Scenario	4
		2.1.1 Heavy Metals Present in Textile	
		Industry	5
		2.1.2 Toxicity of Heavy Metals	5
		2.1.3 Heavy Metal Contamination In Soil	6

		2.1.4	Contamin	nation of Heavy Metals at	
			Aquatic I	Environment	7
	2.2	Organi	ic Compou	nd Contamination	9
	2.3	Source	es of H	eavy Metals and Organic	
		Contar	minants		11
	2.4	Metho	ds Involve	d In Removal of Heavy Metals	
		and Or	ganic Com	pounds	14
		2.4.1	Physical	Treatments	14
			2.4.1.1	Activated Carbon Adsorption	14
			2.4.1.2	Membrane Filtration	15
		2.4.2	Chemical	1 Treatments	16
			2.4.2.1	Chemical Precipitation	16
			2.4.2.2	Ion Exchange	17
			2.4.2.3	Chemical Advanced	
				Oxidation	17
		2.4.3	Biologica	al Treatments	18
			2.4.3.1	Phytoremediation	19
			2.4.3.2	Biosorption	20
	2.5	Biosor	ption of He	eavy Metals Using Bacteria	21
	2.6	Heavy	Metal Tol	erant Bacteria	22
	2.7	The A	pplications	of Heavy Metal Tolerant and	
		Organi	ic Compou	nds Degrading Bacteria	23
3	MAT	ERIAL	S AND MI	ETHODS	24
	3.1	Design	of Work		24
	3.2	Mediu	m Preparat	ion	26
		3.2.1	Nutrient	Broth	26
		3.2.2	Nutrient	Agar	26
		3.2.3	Chemica	lly Defined Medium	26
		3.2.4	Heavy M	letal Stock Solution Preparation	27
		3.2.5	Bacterial	Culture Preparation	28
		3.2.6	Antimicr	obial Reagent Preparation	28
	3.3	Growt	h Profile		28

	3.4	Heavy	Metals Tolerance Test	29
	3.5	Carbon	Utilization Experiment	29
	3.6	16S Fu	Il Sequence PCR Identification	29
		3.6.1	Genomic DNA Extraction	29
		3.6.2	Gel Electrophoresis	31
		3.6.3	Polymerase Chain Reaction	31
		3.6.4	PCR Product Clean Up	32
		3.6.5	The 16S rRNA Gene Sequencing	
			Analysis	32
	3.7	Antimi	crobial Reagent Test	33
4	RESU	JLTS AN	ND DISCUSSION	34
	4.1	Growth	Profiling	34
	4.2	Heavy	Metal Tolerance Test	36
		4.2.1	Effect of Cobalt Sulfate (CoSO ₄ .7H ₂ O)	
			on the Growth of CA Strain	36
		4.2.2	Effect of Zinc Sulfate (ZnSO ₄ .7H ₂ O) on	
			the Growth of CA Strain	38
		4.2.3	Effect of Aluminium Potassium Sulfate	
			(AlK (SO ₄).12H ₂ O) on the Growth of	
			CA Strain	40
		4.2.4	Effect of Manganese Chloride	
			(MnCl ₂ .4H ₂ O) on the Growth of CA	
			Strain	42
		4.2.5	Effect of Copper Sulfate (CuSO ₄ .5H ₂ O)	
			on the Growth of CA Strain	45
	4.3	Effect	of Carbon Source on the Growth of	
		Bacteriu	um CA	47
		4.3.1	Proteinaceous Compounds as Carbon	
			and Nitrogen Source	52
	4.4	Antimi	crobial Reagent Test	53
	4.5	The 169	S rRNA Analysis	56
		4.5.1	Genomic DNA Extraction	56

		4.5.2	PCR	57
		4.5.3	Analyse of Sequences	58
5	CON	CLUSIC	ON AND RECOMMENDATIONS	61
	5.1	Conclu	asion	61
	5.2	Recom	nmendations	63
REFEREN	NCES			64
Appendice	s A-C			74-80

LIST OF TABLES

TABLE NO	TITLE	PAGE
2.1	Source of heavy metals	11
2.2	Source of organic pollutants	13
3.2	Composition of chemically defined medium	27
4.1	The growth of bacterium CA in CDM with different organic compounds used as carbon source and casamino acids as carbon and nitrogen source.	48

LIST OF FIGURES

FIGURE NO	TITLE	PAGE
3.1	Description of the steps carried out throughout the work	25
4.1	Growth pattern of bacterium CA in nutrient broth for ten hours at 37°C	35
4.2	Growth of bacterium CA under different concentrations of cobalt sulfate, at 37°C	36
4.3	The specific growth rate of bacterium CA at 0 mg/L, 20 mg/L, 40 mg/L and 60 mg/L of cobalt sulfate	37
4.4	Growth of bacterium CA under different concentrations of zinc sulfate, at 37°C	38
4.5	The specific growth rate of bacterium CA at 0 mg/L, 40 mg/L, 60 mg/L, 80 mg/L and 100 mg/L of zinc sulfate	39
4.6	Growth of bacterium CA under different concentrations of aluminium potassium sulfate	40

4.7	mg/L, 100 mg/L, 200 mg/L, 300 mg/L and 400	
	mg/L of aluminium potassium sulfate	41
4.8	Growth of bacterium CA under different concentrations of manganese chloride	43
4.9	The specific growth rate of bacterium CA under 0 mg/L, 80 mg/L, 100 mg/L, 200 mg/L, 300 mg/L and 400 mg/L of manganese chloride	43
4.10	Growth of bacterium CA under different concentrations of copper sulfate	465
4.11	The specific growth rate of bacterium CA under 0 mg/L, 60 mg/L, 80 mg/L, 100 mg/L, 200 mg/L and 300 mg/L of copper sulfate	45
4.12	The minimal inhibitory concentration (MIC) of cobalt sulfate, zinc sulfate, aluminium potassium sulfate, manganese chloride and copper sulfate on the growth of bacterium CA	47
4.13	Growth of bacterium CA at CDM provided with different organic compounds as sole carbon sources	49
4.14	The specific growth rate values of bacterium CA on casamino acids, glucose and glycerol	50
4.15	Growth of bacterium CA at CDM with casamino acids as carbon and/or nitrogen sources	52

4.16	Antimicrobial agents' effect on bacterium CA at 37°C, 20 hours incubation	54
4.17	Clear zone prevention of growth of bacterium CA formed by tetracycline hydrochloride	54
4.18	Clear zone prevention of growth of bacterium CA formed by Kanamycin sulfate	55
4.19	Extracted DNA band and 1kb ladder on agarose gel	57
4.20	Bands of 1kb DNA ladder and PCR product on the agarose gel	58
4.21	Blastn sheet result of bacterium CA	61
4.22	Phylogenetic tree of bacterium CA done as Neighbour-joining bootstrap method	60

LIST OF ABBREVIATIONS

AH - Aliphatic Hydrocarbon

Bacterium CA - An isolated bacterium from textile waste water

Cd - Cadmium

CDM - Chemically Defined Media

CO₂ - Carbon dioxide

Cr - Chromium

Cu - Copper

DDTs - Dichloro Diphenyl Trichloroethanes

DNA - Deoxyribonucleic acid

EDTA - Ethylenediaminetetraacetic acid

EMP - Embden–Meyerhof–Parnas

EOC - Emerging Organic Contaminant

GAC - Granular Activated Carbon

HBC - Hexachlorobenzene

Hg - Mercury

HMP - Hexose monophosphate

k - Specific growth rate

MCL - Maximum Contaminant Level

MEGA - Molecular Evolutionary Genetics Analysis

MIC - Minimal inhibitory concentration

BLASTn - nucleotide Basic Local Alignment Search Tool

NCBI - National Center for Biotechnology Information

OD - Optical Density

PAC - Powder Activated Carbon

PAH - Polycyclic Aromatic Hydrocarbon

Pb - Lead

PCB - Polychlorinated Biphenyl

PCR - Polymerase Chain Reaction

rRNA - Ribosomal Ribonucleic acid

STP - Sewage Treatment Plant

TAE - Tris-acetate-EDTA

t_d - Doubling time (Generation time)

USEPA - United States Environmental Protection Agency

UV - Ultraviolet

WHO - World Health Organization

Zn - Zinc

LIST OF APPENDICES

APPENDIX	TITLE	PAGE
A	Single colony of bacterium CA	74
В	Data gained from OD 600nm	75
С	Partial sequence of 16S rRNA of bacterium CA	80

CHAPTER 1

INTRODUCTION

1.0 Background

Urbanisation and modernisation have given human beings a number of advantages. These include the improvement of the quantity and quality of food, clothes and entertainment products. Despite of their positive features, such developments pose negative impacts on ecosystem through releasing of effluents and they are reason behind the increase of proportion of contaminants like heavy metals, organic compounds and others in the environment (Nagajyoti *et al.*, 2010; Chopra *et al.*, 2009; Pérez-López *et al.*, 2003).

Contaminants like heavy metals and organic compounds are introduced into environment through industrial effluents and these compounds give negative impacts to the environments and living organisms. Hence the innovation of removal treatments such as physical, chemical and biological approaches is urgent. The examples of physical treatments are activated carbon adsorption and membrane filtration, Chemical treatments are chemical precipitation, ion exchange and chemical advanced oxidation, and Biological treatments are biosorption, phytoremediation and bioleaching. Biological means are preferred over physical and chemical means because biological treatments are eco-friendly, cost effective and promising (Vargas-Garcia *et al.*, 2012;

Sutton *et al.*, 2013; Szulc *et al.*, 2014; Umrania, 2006; Fosso-Kankeu and Mulaba-Bafubiandi, 2013; Colin *et al.*, 2012).

Bacteria have been applied in bioremediation processes such as in bioadsorption treatment process in layer by layer fattening strategy (*Pseudomonas sp.* as biosorbent) to remove heavy metals from industrial effluents (Luo *et al.*, 2014) and in membrane bioreactor treatment to remove micro contaminants including organic compounds (Luo *et al.*, 2014). Bacteria have numerous mechanisms, various kind of enzymes and have unique structure to tolerate and remove the contaminants of different kinds including heavy metals and organic compounds.

In this study, bacterium CA previously isolated from textile wastewater was further characterized through assessment of its ability to tolerate certain heavy metals at determined concentrations, then its capability to utilize different organic compounds as carbon source, in addition, the effects of antimicrobial agents on the growth of bacterium CA was studied. Eventually it was identified via 16S rRNA method.

1.1 Significance of the Work / Problem Statement

Bioremediation that uses microorganisms to clean up an environment is relatively cost-effective compared to conventional physical or chemical processes (Abdel-El-Haleem, 2003). The bacterium CA was able to tolerate some heavy metals even at high concentrations and was positive to grow on some organic compounds, hence such results postulate that bacterium CA could be a potential option to remove of heavy metals and organic compounds from the contaminated area.

1.2 Objectives

- a) To investigate the tolerance of the bacterium CA isolated from textile wastewater towards heavy metals.
- b) To screen for the ability of the bacterium CA to use the selected organic compounds as carbon source.
- c) To test the effect of antimicrobial agents on the growth of bacterium CA.
- d) To identify the bacterium CA based on 16S rRNA gene analysis.

1.3 Study Scope

Bacterium CA was firstly examined to tolerate (aluminium, cobalt, zinc, copper and manganese). Its ability was screened to utilize the organic compounds (casamino acids, sulfanilic acids, glucose, glycerol, acrylamide and benzene) as sole carbon source for growth. Effects of antibiotics on the growth bacterium CA were examined. Finally, the bacterium CA was identified based on the 16S rRNA gene analysis.

REFERENCES

- Abdel-El-Haleem, D. (2004). Acinetobacter: Environmental and Biotechnological Applications. *African Journal of Biotechnology*, 2(4), 71-74.
- Abdolali, A., Guo, W. S., Ngo, H. H., Chen, S. S., Nguyen, N. C. and Tung, K. L. (2013). Typical Lignocellulosic Wastes and By-Products for Biosorption Process in Water and Wastewater Treatment: A Critical Review. *Bioresource technology*, 160, 57-66.
- Abril, G. A., Wannaz, E. D., Mateos, A. C., Invernizzi, R., Plá, R. R. and Pignata, M. L. (2014). Characterization of Atmospheric Emission Sources of Heavy Metals and Trace Elements through a Local-Scale Monitoring Network Using *T. Capillaris*. *Ecological Indicators*, 40, 153-161.
- Agarwal, S. K. (2009). Heavy Metal Pollution (Vol. 4). APH Publishing.
- Alexandratos, S. D. (2008). Ion-exchange Resins: A Retrospective From Industrial And Engineering Chemistry Research. *Industrial & Engineering Chemistry Research*, 48(1), 388-398.
- Ali, I. (2014). Water Treatment By Adsorption Columns: Evaluation At Ground Level. *Separation & Purification Reviews*, 43(3), 175-205.
- Al-Jeshi, S. and Neville, A. (2008). An Experimental Evaluation of Reverse Osmosis Membrane Performance in Oily Water. *Desalination*, 228(1), 287-294.
- Arslan-Alaton, I. and Alaton, I. (2007). Degradation of Xenobiotics Originating From the Textile Preparation, Dyeing, and Finishing Industry Using Ozonation and Advanced Oxidation. *Ecotoxicology and environmental safety*, 68(1), 98-107.
- Aydinalpi, C. and Marinova, S. (2003). Distribution and Forms of Heavy Metals in Some Agricultural Soils. *Polish Journal of Environmental Studies*, 12(5), 629-634.

- Cai, Q. Y., Mo, C. H., Wu, Q. T., Katsoyiannis, A. and Zeng, Q. Y. (2008). The Status of Soil Contamination by Semivolatile Organic Chemicals (SVOCs) in China: A Review. *Science of the Total Environment*, 389(2), 209-224.
- Checa-Moreno, R., Manzano, E., Mirón, G. and Capitan-Vallvey, L. F. (2008). Comparison between Traditional Strategies and Classification Technique (SIMCA) in the Identification of Old Proteinaceous Binders. *Talanta*, 75(3), 697-704.
- Chen, G., Cheng, K. Y., Ginige, M. P. and Kaksonen, A. H. (2012). Aerobic Degradation of Sulfanilic Acid using Activated Sludge. *Water research*, 46(1), 145-151.
- Cheng, S. (2003). Heavy Metal Pollution in China: Origin, Pattern and Control. Environmental Science and Pollution Research, 10(3), 192-198.
- Chien, C., Kuo, Y., Chen, C., Hung, C., Yeh, C. and Yeh, W. (2008). Microbial Diversity of Soil Bacteria in Agricultural Field Contaminated with Heavy Metals. *Journal of Environmental Sciences*, 20(3), 359-363.
- Chopra, A. K., Pathak, C., and Parasad, G. (2009). Scenario of Heavy Metal Contamination in Agricultural Soil and Its Management. *Journal of Applied and Natural Sciences*, 1(1), 99-108.
- Colin, V. L., Villegas, L. B. and Abate, C. M. (2012). Indigenous microorganisms as potential bioremediators for environments contaminated with heavy metals. *International Biodeterioration & Biodegradation*, 69, 28-37.
- Comstock, S. E. and Boyer, T. H. (2014). Combined Magnetic Ion Exchange and Cation Exchange for Removal of DOC And Hardness. *Chemical Engineering Journal*, 241, 366-375.
- Das, N. (2010). Recovery of Precious Metals through Biosorption—A Review. Hydrometallurgy, 103(1), 180-189.
- Dash, H. R., Mangwani, N., Chakraborty, J., Kumari, S. and Das, S. (2013). Marine Bacteria: Potential Candidates for Enhanced Bioremediation. *Applied Microbiology and Biotechnology*, 97(2), 561-571.
- Djedidi, Z., Bouda, M., Souissi, M. A., Cheikh, R. B., Mercier, G., Tyagi, R. D. and Blais, J. F. (2009). Metals Removal from Soil, Fly Ash and Sewage Sludge Leachates by Precipitation and Dewatering Properties of the Generated Sludge. *Journal of Hazardous Materials*, 172(2), 1372-1382.

- Dubey, N. K. (Ed.). (2011). Natural Products in Plant Pest Management. CABI.
- Emmanuel Joshua Jebasingh, S., Lakshmikandan, M., Rajesh, R. P. and Raja, P. (2013). Biodegradation of Acrylamide and Purification of Acrylamidase from Newly Isolated Bacterium *Moraxella osloensis* MSU11. *International Biodeterioration & Biodegradation*, 85, 120-125.
- Erdem, M., Ucar, S., Karagöz, S. and Tay, T. (2013). Removal of Lead (II) Ions from Aqueous Solutions onto Activated Carbon Derived from Waste Biomass. *The Scientific World Journal*, 2013.
- Eskelinen, K., Särkkä, H., Kurniawan, T. A. and Sillanpää, M. E. (2010). Removal of Recalcitrant Contaminants from Bleaching Effluents in Pulp and Paper Mills using Ultrasonic Irradiation and Fenton-like Oxidation, Electrochemical Treatment, and/or Chemical Precipitation: A Comparative Study. *Desalination*, 255(1), 179-187.
- Esplugas, S., Bila, D. M., Krause, L. G. T. and Dezotti, M. (2007). Ozonation and Advanced Oxidation Technologies to Remove Endocrine Disrupting Chemicals (EDCs) and Pharmaceuticals And Personal Care Products (Ppcps) In Water Effluents. *Journal of Hazardous Materials*, *149*(3), 631-642.
- Farag, S. and Zaki, S. (2010). Identification Of Bacterial Strains From Tannery Effluent And Reduction Of Hexavalent Chromium.
- Farhadian, M., Vachelard, C., Duchez, D. and Larroche, C. (2008). In situ Bioremediation of Monoaromatic Pollutants In Groundwater: A Review. *Bioresource Technology*, *99*(13), 5296-5308.
- Foo, K. Y. and Hameed, B. H. (2009). An Overview Of Landfill Leachate Treatment Via Activated Carbon Adsorption Process. *Journal of Hazardous Materials*, 171(1), 54-60.
- Fosso-Kankeu, E. and Mulaba-Bafubiandi, A. F. (2013). Implication of Plants and Microbial Metalloproteins in the Bioremediation of Polluted Waters: A Review. *Physics and Chemistry of the Earth, Parts A/B/C*, 67, 242–252.
- Gadd, G. M. (2004). Microbial Influence on Metal Mobility and Application for Bioremediation. *Geoderma*, 122(2), 109-119.
- Garbarino, J. R., Hayes, H. C., Roth, D. A., Antweiler, R. C., Brinton, T. I. and Taylor,H. E. (1996). Heavy Metals in the Mississippi River. US Geological SurveyCircular USGS CIRC, 53-72.

- Gullinkala, T., Digman, B., Gorey, C., Hausman, R. and Escobar, I. C. (2010). Desalination: Reverse Osmosis and Membrane Distillation. *Sustainability Science and Engineering*, *2*, 65-93.
- Gunathilake, K. D. P. P., Yu, L. J. and Rupasinghe, H. P. (2014). Reverse Osmosis as a Potential Technique to Improve Antioxidant Properties of Fruit Juices used for Functional Beverages. *Food chemistry*, *148*, 335-341.
- He, J. and Chen, J. P. (2014). A Comprehensive Review on Biosorption of Heavy Metals by Algal Biomass: Materials, Performances, Chemistry, and Modelling Simulation Tools. *Bioresource Technology*, 160, 67-78.
- Horáková, M., Klementová, Š, Kříž, P., Balakrishna, S. K., Špatenka, P., Golovko, O., Hájková, P. and Exnar, P. (2013). The Synergistic Effect of Advanced Oxidation Processes to Eliminate Resistant Chemical Compounds. Surface and Coatings Technology, 241, 154-158.
- Huuha, T. S., Kurniawan, T. A. and Sillanpää, M. E. (2010). Removal of Silicon from Pulping Whitewater Using Integrated Treatment of Chemical Precipitation and Evaporation. *Chemical Engineering Journal*, 158(3), 584-592.
- Jadhav, J. P., Kalyani, D. C., Telke, A. A., Phugare, S. S. and Govindwar, S. P. (2010).
 Evaluation of the Efficacy of a Bacterial Consortium for the Removal of Color,
 Reduction of Heavy Metals, and Toxicity From Textile Dye Effluent.
 Bioresource Technology, 101(1), 165-173.
- Jaffé, R. (1991). Fate of Hydrophobic Organic Pollutants in the Aquatic Environment: A Review. *Environmental Pollution*, 69(2), 237-257.
- Järup, L. (2003). Hazards of Heavy Metal Contamination. *British Medical Bulletin*, 68(1), 167-182.
- Jernberg, J., Pellinen, J. and Rantalainen, A. L. (2013). Identification of Organic Xenobiotics in Urban Aquatic Environments Using Time-Of-Flight Mass Spectrometry. *Science of the Total Environment*, 450, 1-6.
- Jiang, J. J., Lee, C. L. and Fang, M. D. (2014). Emerging Organic Contaminants in Coastal Waters: Anthropogenic Impact, Environmental Release and Ecological Risk. *Marine pollution bulletin*.
- Kalantari, A., Bina, B., Taleby, M. and Loloee, M. (2006). The Relationship Between Polycyclic Aromatic Hydrocarbons And Heavy Metals. *Journal of Research in Health Sciences*, 6(1), 14-17.

- Kavamura, V. N. and Esposito, E. (2010). Biotechnological strategies applied to the decontamination of soils polluted with heavy metals. *Biotechnology advances*, 28(1), 61-69.
- Kim, B. H. and Gadd, G. M. (2008). *Bacterial physiology and metabolism* (p. 553). Cambridge: Cambridge university press.
- Krewski, D., Yokel, R. A., Nieboer, E., Borchelt, D., Cohen, J., Harry, J., Kacew, S., Lindsay, J. Mahfouz, A. M. and Rondeau, V. (2007). Human health risk assessment for aluminium, aluminium oxide, and aluminium hydroxide. *Journal of Toxicology and Environmental Health, Part B*, 10(S1), 1-269.
- Kříbek, B., Majer, V., Knésl, I., Nyambe, I., Mihaljevič, M., Ettler, V. and Sracek, O. (2014). Concentrations of arsenic, copper, cobalt, lead and zinc in cassava (*Manihot esculenta Crantz*) growing on uncontaminated and contaminated soils of the Zambian Copperbelt. *Journal of African Earth Sciences*.
- Kucuksezgin, F., Pazi, I., Yucel-Gier, G., Akcali, B. and Galgani, F. (2013). Monitoring of heavy metal and organic compound levels along the Eastern Aegean coast with transplanted mussels. *Chemosphere*, *93*(8), 1511-1518.
- Kumar, B., Sharma, D., Sharma, P., Katoch, V. M., Venkatesan, K. and Bisht, D. (2013). Proteomic analysis of Mycobacterium tuberculosis isolates resistant to kanamycin and amikacin. *Journal of proteomics*, 94, 68-77.
- Kurniawan, T. A., Lo, W. H. and Chan, G. (2006). Physico-chemical treatments for removal of recalcitrant contaminants from landfill leachate. *Journal of hazardous materials*, 129(1), 80-100.
- Kusada, H., Hanada, S., Kamagata, Y. and Kimura, N. (2014). The effects of N-acylhomoserine lactones, β-lactam antibiotics and adenosine on biofilm formation in the multi-β-lactam antibiotic-resistant bacterium *Acidovorax* sp. strain MR-S7. *Journal of Bioscience and Bioengineering*.
- Leung, H. M., Leung, A. O. W., Wang, H. S., Ma, K. K., Liang, Y., Ho, K. C., Cheung, K.C., Tohidi, F. and Yung, K. K. L. (2014). Assessment of heavy metals/metalloid (As, Pb, Cd, Ni, Zn, Cr, Cu, Mn) concentrations in edible fish species tissue in the Pearl River Delta (PRD), China. *Marine pollution bulletin*, 78(1), 235-245.

- Liang, X., Ning, X. A., Chen, G., Lin, M., Liu, J. and Wang, Y. (2013). Concentrations and speciation of heavy metals in sludge from nine textile dyeing plants. *Ecotoxicology and environmental safety*, *98*, 128-134.
- Lin, M., Ning, X. A., Liang, X., Wei, P., Wang, Y. and Liu, J. (2014). Study of the heavy metals residual in the incineration slag of textile dyeing sludge. *Journal of the Taiwan Institute of Chemical Engineers*.
- Liu, Z. H., Kanjo, Y. and Mizutani, S. (2009). Removal mechanisms for endocrine disrupting compounds (EDCs) in wastewater treatment—physical means, biodegradation, and chemical advanced oxidation: a review. *Science of the Total Environment*, 407(2), 731-748.
- Lofrano, G., Meriç, S., Zengin, G. E. and Orhon, D. (2013). Chemical and biological treatment technologies for leather tannery chemicals and wastewaters: A review. *Science of the Total Environment*, 461, 265-281.
- Luo, S., Li, X., Chen, L., Chen, J., Wan, Y. and Liu, C. (2014). Layer-by-layer strategy for adsorption capacity fattening of endophytic bacterial biomass for highly effective removal of heavy metals. *Chemical Engineering Journal*, 239, 312-321.
- Luo, Y., Guo, W., Ngo, H. H., Nghiem, L. D., Hai, F. I., Zhang, J., Liang, S. and Wang, X. C. (2014). A review on the occurrence of micropollutants in the aquatic environment and their fate and removal during wastewater treatment. *Science of The Total Environment*, 473, 619-641.
- Marcovecchio, J. E., Botté, S. E., and Freije, R. H. (2007). Heavy metals, major metals, trace elements. *Handbook of water analysis*, 275-311.
- Megharaj, M., Ramakrishnan, B., Venkateswarlu, K., Sethunathan, N. and Naidu, R. (2011). Bioremediation approaches for organic pollutants: a critical perspective. *Environment International*, *37*(8), 1362-1375.
- Meunier, N., Drogui, P., Montané, C., Hausler, R., Mercier, G. and Blais, J. F. (2006). Comparison between electrocoagulation and chemical precipitation for metals removal from acidic soil leachate. *Journal of hazardous materials*, *137*(1), 581-590.
- Mohan, D., and Pittman Jr, C. U. (2007). Arsenic removal from water/wastewater using adsorbents—a critical review. *Journal of hazardous materials*, *142*(1), 1-53.

- Momodu, M. A. and Anyakora, C. A. (2010). Heavy metal contamination of ground water: The Surulere case study. *Res. J. Environ. Earth Sci*, 2(1), 39-43.
- Mulligan, C. N., Yong, R. N. and Gibbs, B. F. (2001). Remediation technologies for metal-contaminated soils and groundwater: an evaluation. *Engineering* geology, 60(1), 193-207.
- Nagajyoti, P. C., Lee, K. D. and Sreekanth, T. V. M. (2010). Heavy metals, occurrence and toxicity for plants: a review. *Environmental Chemistry Letters*, 8 (3), 199-216.
- Nguyen, H. D. N. and Yuk, H. G. (2013). Changes in resistance of Salmonella Typhimurium biofilms formed under various conditions to industrial sanitizers. *Food Control*, 29(1), 236-240.
- Oncel, M. S., Muhcu, A., Demirbas, E. and Kobya, M. (2013). A comparative study of chemical precipitation and electrocoagulation for treatment of coal acid drainage wastewater. *Journal of Environmental Chemical Engineering*, 1(4), 989-995.
- Ong, Y. K., Li, F. Y., Sun, S. P., Zhao, B. W., Liang, C. Z. and Chung, T. S. (2014). Nanofiltration hollow fiber membranes for textile wastewater treatment: Labscale and pilot-scale studies. *Chemical Engineering Science*, 114, 51-57.
- Pelaez, A. I., Lores, I., Sotres, A., Mendez-Garcia, C., Fernandez-Velarde, C., Santos, J. A., Gallego, J. L. R. and Sanchez, J. (2013). Design and field-scale implementation of an "on site" bioremediation treatment in PAH-polluted soil. *Environmental Pollution*, 181, 190-199.
- Perelo, L. W. (2010). Review: In situ and bioremediation of organic pollutants in aquatic sediments. *Journal of hazardous materials*, 177(1), 81-89.
- Pérez-López, M., Alonso, J., Nóvoa-Valiñas, M. C., and Melgar, M. J. (2003). Assessment of heavy metal contamination of seawater and marine Limpet, Patella vulgata L., from Northwest Spain. *Journal of Environmental Science and Health, Part A*, 38(12), 2845-2856.
- Pulford, I. D. and Watson, C. (2003). Phytoremediation of heavy metal-contaminated land by trees—a review. *Environment international*, 29(4), 529-540.
- Raja, C. E., Selvam, G. S., and Omine, K. I. Y. O. S. H. I. (2009). Isolation, identification and characterization of heavy metal resistant bacteria from

- sewage. In Int Joint Symp on Geodisaster Prevention and Geoenvironment in Asia (pp. 205-211).
- Rajbanshi, A. (2009). Study on heavy metal resistant bacteria in Guheswori sewage treatment plant. *Our Nature*, 6(1), 52-57.
- Rengaraj, S., Yeon, K. H. and Moon, S. H. (2001). Removal of chromium from water and wastewater by ion exchange resins. *Journal of hazardous materials*, 87(1), 273-287.
- Saritha, P., Aparna, C., Himabindu, V. and Anjaneyulu, Y. (2007). Comparison of various advanced oxidation processes for the degradation of 4-chloro-2 nitrophenol. *Journal of hazardous materials*, *149*(3), 609-614.
- Secondes, M. F. N., Naddeo, V., Belgiorno, V. and Ballesteros Jr, F. (2014). Removal of emerging contaminants by simultaneous application of membrane ultrafiltration, activated carbon adsorption, and ultrasound irradiation. *Journal of hazardous materials*, 264, 342-349.
- Sethuraman, V., Haq, B., Chezhian, A., Shanker, S., and Selvan, D. S. (2011). Changes due to the effect of the Heavy Metals (Hgcl2 and Znso4) concentration on the marine fish, Tilapia Mossambica (Peters, 1852). *Archives of Applied Science Research*, *3*(6), 333-341.
- Sharma, V. K., Triantis, T. M., Antoniou, M. G., He, X., Pelaez, M., Han, C., Song, W., O'Shea, K. E., de la Cruz, A. A., Kaloudis, T., Hiskia, A. and Dionysiou, D. D. (2012). Destruction of microcystins by conventional and advanced oxidation processes: a review. Separation and Purification Technology, 91, 3-17.
- Siripornadulsil, S. and Siripornadulsil, W. (2013). Cadmium-tolerant bacteria reduce the uptake of cadmium in rice: Potential for microbial bioremediation. *Ecotoxicology and environmental safety*, 94, 94-103.
- Stefan, D. S. and Meghea, I. (2014). Mechanism of simultaneous removal of Ca²⁺, Ni²⁺, Pb²⁺ and Al³⁺ ions from aqueous solutions using Purolite[®] S930 ion exchange resin. *Comptes Rendus Chimie*, 17(5), 496-502.
- Suda, T., Hata, T., Kawai, S., Okamura, H. and Nishida, T. (2012). Treatment of tetracycline antibiotics by laccase in the presence of 1-hydroxybenzotriazole. *Bioresource technology*, 103(1), 498-501.

- Sutton, N. B., Grotenhuis, T. and Rijnaarts, H. H. (2013). Impact of organic carbon and nutrients mobilized during chemical oxidation on subsequent bioremediation of a diesel-contaminated soil. *Chemosphere*, 97, 64-70.
- Szulc, A., Ambrożewicz, D., Sydow, M., Ławniczak, Ł., Piotrowska-Cyplik, A., Marecik, R. and Chrzanowski, Ł. (2014). The influence of bioaugmentation and biosurfactant addition on bioremediation efficiency of diesel-oil contaminated soil: Feasibility during field studies. *Journal of environmental management*, 132, 121-128.
- Tan, Y., Kilduff, J. E., Kitis, M. and Karanfil, T. (2005). Dissolved organic matter removal and disinfection byproduct formation control using ion exchange. *Desalination*, 176(1), 189-200.
- Umrania, V. V. (2006). Bioremediation of toxic heavy metals using acidothermophilic autotrophes. *Bioresource technology*, *97*(10), 1237-1242.
- Valverde, A., González-Tirante, M., Medina-Sierra, M., Santa-Regina, I., García-Sánchez, A. and Igual, J. M. (2011). Diversity and community structure of culturable arsenic-resistant bacteria across a soil arsenic gradient at an abandoned tungsten—tin mining area. *Chemosphere*, 85(1), 129-134.
- van Deventer, J. (2011). Selected ion exchange applications in the hydrometallurgical industry. *Solvent Extraction and Ion Exchange*, 29(5-6), 695-718.
- Vargas-García, M. D. C., López, M. J., Suárez-Estrella, F. and Moreno, J. (2012).
 Compost as a source of microbial isolates for the bioremediation of heavy metals: *In vitro* selection. *Science of the Total Environment*, 431, 62-67.
- Verbych, S., Hilal, N., Sorokin, G. and Leaper, M. (2005). Ion exchange extraction of heavy metal ions from wastewater. Separation science and technology, 39(9), 2031-2040.
- Vijayaraghavan, K. and Yun, Y. S. (2008). Bacterial biosorbents and biosorption. *Biotechnology advances*, 26(3), 266-291.
- Wibisono, Y., Cornelissen, E. R., Kemperman, A. J. B., van der Meer, W. G. J. and Nijmeijer, K. (2014). Two-phase flow in membrane processes: A technology with a future. *Journal of Membrane Science*, 453, 566-602.
- Willey, J. M., Sherwood, L. and Woolverton, C. J. (2008). *Prescott's principles of microbiology*. McGraw-Hill Higher Education.

Zhang, Y. F., He, L. Y., Chen, Z. J., Wang, Q. Y., Qian, M. and Sheng, X. F. (2011). Characterization of ACC deaminase-producing endophytic bacteria isolated from copper-tolerant plants and their potential in promoting the growth and copper accumulation of *Brassica napus*. *Chemosphere*, 83(1), 57-62.