

BIODEGRADATION OF REMAZOL BLACK B BY BACTERIAL
CONSORTIUM NAR-2

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*Specially dedicated to my beloved Dad and Mom, Reza
Kardí and María Hadíghí.*

*To my adorable husband
Níma
And
My granny*

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ABSTRACT

The ability of the bacterial consortium NAR-2 consisting of A1, C1 and L17 to degrade the model azo dye Remazol Black B (RBB) was studied in batch and in continuous systems. Continuous decolourisation was performed in a borosilicate glass column (12 mm x 20 mm) packed with Surfactant Modified Clinoptilolite immobilised with bacterial consortium NAR-2. In batch studies, 90.79% decolourisation of RBB was achieved under microaerophilic condition within 80 minutes by inoculating 10% (v/v) of bacterial consortium NAR-2 at a 1:1:1 ratio. This was achieved in modified P5 medium pH 7 and incubated at 45°C under microaerophilic condition. In column bioreactor studies, decolourisation was observed at 45°C and carried out by varying the flow rates and dye concentrations. Flow rate at 0.2, 0.4, 0.6, 0.8, and 1.0 ml/min were tested and dye concentration of 0.1, 0.3, 0.5, 0.7, and 1.0 g/L were used. Almost 95.87% decolourisation of 0.1 g/L RBB was achieved at the flow rate 0.2 ml/min. By fixing 0.2 ml/min as default flow rate, varying concentrations of RBB were examined. Above 90% decolourisation was achieved with 0.1, 0.3 and 0.5 g/L RBB but at 0.7 and 1.0 g/L the percentage drop to 36 and 28%, respectively. Decolourisation percentage began to drop at higher dye concentration. Biomass leached out from the column was determined using viable cell count. From both flow rate and dye concentration experiments, it can be seen that C1 cell wash out was the highest as compared to A1 and L17. Analyses of decolourized and biodegradation products of RBB using total aromatic amines (TAA) showed that reduction of RBB resulted in the formation of aromatic amines. Further aerobic degradation for 15 days showed the amines concentration reduced from an initial of 18 mg/L to 2 mg/L following aerobic treatment in batch whereas in column experiment, the amines concentration dropped significantly from 34 mg/L to 11 mg/L.

ABSTRAK

Keupayaan konsortium bakteria NAR-2 terdiri daripada A1, C1 dan L17 untuk menyahwarnakan model azo pewarna Remazol Black B (RBB) telah dikaji dalam kelompok dan dalam sistem lengkap berterusan. Penyahwarnaan lengkap berterusan dilakukan dengan menggunakan kolum kaca borosilika (12 mm x 20 mm) dimampatkan dengan konsortium bakteria NAR-2 yang disekat gerak di atas clinoptilolite dengan permukaan yang telah diubah suai dengan surfaktan. Dalam eksperimen kelompok, 90.79% penyahwarnaan RBB telah dicapai di bawah keadaan mikroaerofilik dalam tempoh 80 minit dengan menginokulasi 10% (v/v) konsortium bakteria NAR-2 pada nisbah 1:1:1. Ini telah dicapai dalam medium P5 terubah suai pada pH 7 dan dieram pada 45°C di bawah keadaan mikroaerofilik. Dalam eksperimen kolum bioreaktor, penyahwarnaan telah diperhatikan pada 45°C dan dijalankan dengan mengubah kadar alir dan kepekatan pewarna. Kadar alir 0.2, 0.4, 0.6, 0.8, dan 1.0 ml/min, dan kepekatan pewarna 0.1, 0.3, 0.5, 0.7, dan 1.0 g / L telah dikaji. Hampir 95.87% penyahwarnaan 0.1 g/L RBB telah dicapai pada kadar alir 0.2 ml/min. Dengan menetapkan 0.2 ml/min sebagai kadar alir tentu awal, RBB pada kepekatan berbeza diperiksa. Lebih daripada 90% penyahwarnaan dicapai dengan 0.1, 0.3 dan 0.5 g/L RBB tetapi pada kepekatan 0.7 dan 1.0 g/L, peratusan menurun kepada 36 dan 28%, masing-masing. Peratusan penyahwarnaan mula berkurangan pada kepekatan pewarna yang lebih tinggi. Biomas yang terlarut lesap dari kolum ditentukan dengan menggunakan kiraan sel berdaya hidup. Berdasarkan kedua-dua eksperimen kadar alir dan kepekatan pewarna, dapat dilihat bahawa sel C1 yang terlarut resap adalah yang tertinggi berbanding A1 dan L17. Analisis produk ternyahwarna dan biodegradasi RBB menggunakan jumlah amina aromatik (TAA) mengesahkan bahawa penyahwarnaan RBB menghasilkan amina aromatik. Lanjutan degradasi aerobik selama 15 hari menunjukkan kepekatan amina menurun daripada 18 mg/L kepada 2 mg/L dalam eksperimen kelompok manakala dalam eksperimen kolum, kepekatan amina menurun dengan ketara daripada 34 mg/L hingga 11 mg/L.

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LIST OF ABBREVIATIONS

A1	-	<i>Citrobacter</i> sp. A1
C1	-	<i>Enterococcus</i> sp. C1
CMC	-	Critical micelle concentration
EPS	-	Exopolysaccharide
HCL	-	Hydrochloric acid
HDTMA-Br	-	Hexadecyltrimethylammonium bromide
KH ₂ PO ₄	-	Potassium dihydrogen phosphate
K ₂ HPO ₄	-	Dipotassium hydrogen phosphate
L17	-	<i>Citrobacter</i> sp. L17
LPS	-	Lipopolysaccharide
NA	-	Nutrient agar
NAOH	-	Sodium hydroxide
RBB	-	Remazol Black B
SEM	-	Scanning electron microscope
SMC	-	Surfactant modified clinoptilolite
TAA	-	Total aromatic amines
TAHNDS	-	Triaminohydroxynaphthalene disulphanilic acid
TAHNDSDP ₂	-	7-amino-8-hydroxy-1,2 naphthaquinone 3,6-disulphonate-1,2-diimine
TAHNDSDP ₃	-	5,6-Dihydroxy-3-imino-4-oxo-3,4-dihydronaphthalene- 2,7-disulphonic acid
TAHNDSDP ₄	-	4,6-Dihydroxy-3,5-dioxo-3-dihydronaphthalene-2,7- disulphonic acid

LIST OF SYMBOLS

cfu/mL	-	Colony forming units per mL
g/L	-	gram per litre
L	-	Litre
mg/L	-	milligram per Litre
mL	-	millilitre
mm	-	millimeter
mM	-	millimolar
M	-	Molar
µm	-	micrometer
nm	-	nanometer
OD600	-	Optical density at 600 nm
rpm	-	revolution per minute
v/v	-	volume per volume
w/v	-	weight per volume

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CHAPTER 1

INTRODUCTION

1.1 Introduction and Background

A variety of commercial applications of dyes generate effluent in huge amount that requires professional treatment. Different kinds of dyes are utilized in companies such as paper, textile, pulp, paints, carpet, and printing (Aksu *et al.*, 2005). The emissions which contain dyes are enormously coloured and are among the main causes of ecological pollution (Zouari-Mechichi *et al.*, 2006). Up to 50% of the dyes are vanished after the dyeing process in the textile manufacturing industry, and approximately 10–15% of them are discharged in the effluents (Vaidya and Datye, 1982). Dyes are grouped as triphenylmethane, azo, heterocyclic, anthraquinone, and polymeric dyes depending on the chemical structure of the chromophoric category. The largest and most relevant of the dyes used are azo dyes. They are carcinogenic and mutagenic material which may cause cancer to human being and are not fully eliminate through traditional wastewater treatment mechanism (Wong and Yuen, 1996). Azo dyes are branded by the existence of one or more azo category (-N=N-) and negative sulfonate family (-SO₃⁻). Azo dyes itself is dangerous; though when the azo bonds are condensed, it produces amines which are more carcinogenic than the parent arrangement (Hong *et al.*, 2007).

The physical and chemical treatment of azo dyes in wastewater entails chemical corrosion, electrochemical active carbon adsorption, reverse osmosis, anion exchange resins, irradiation, and ozonation (Selcuk, 2005). Nevertheless, many of these techniques are expensive and time-consuming hence cannot be regarded as practical alternative for treating huge waste streams.

Treating dyes with the adoption of biological methods making use of microorganisms to decolourise and biodegrade azo dyes in anaerobic, aerobic or joint anaerobic-aerobic preparation mechanism have proven to be productive. Among the most reliant means which own the ability to decolourise synthetic dyes is the adoption of microorganisms containing bacterial strains, fungi and yeast (Ferreira *et al.*, 2000; Saratale *et al.*, 2011). Several authors have described proper decolourising potential by microbial blend culture compared to pure culture (Chan *et al.*, 2011; Chan *et al.*, 2012; Joshi *et al.*, 2008; Khehra *et al.*, 2005).

Earlier authors have indicated that azo dyes experience two chronological processes; anaerobic that generates amines and a successive step of aerobic conduct (Hong *et al.*, 2007). Amines are considered as more poisonous and recalcitrant when compared to the family compound. These can be eliminated under aerobic condition with the use of custom made microorganisms. Hence, it is of great relevance to prepare dye-comprising waste before disposal and discharge.

1.2 Statement of Problem

Unethical release of textile dye waste matter containing azo dyes and their metabolites in aqueous environment is aesthetically obnoxious and result to a decrease in sunlight infiltration, which in turn reduces photosynthetic processes, dissolved oxygen absorption, and quality of water, and had acute poisonous impact on aquatic flora and fauna, causing severe ecological damages globally (Vandevivere *et al.*, 1998).

Numerous synthetic azo dyes and their metabolites are lethal, carcinogenic, and mutagenic (Saratale *et al.*, 2011; Singh and Arora, 2011). Furthermore, plentiful information point to the fact that textile dyes and emissions have poisonous effects on the germination rates and biomass of many plant species which have essential environmental roles, such as been a source of livelihood to living things, soil erosion prevention and organic material provision that is so important to soil fertility (Ghodake *et al.*, 2009). In biological treatment under anaerobic circumstances, bacteria can diminish azo bonds by azoreductase enzyme and discharge ingredient amines that are susceptible to aerobic biodegradation. Acclimatised ecologically friendly and economical decolourising bacteria generate more efficient methods of lowering dye pollution. For that reason, handling of manufacturing effluents containing azo dyes and their metabolites is essential prior to their final release to the environment.

1.3 Objectives of the Study

The general objectives of this research were:

1. To utilize bacteria *Citrobacter* sp.A1, *Citrobacter* sp.L17, and *Enterococcus* sp.C1 in consortium to degrade Remazol Black B in batch effectively.
2. To modify surface of clinoptilolite with surfactant HDTMA-Br for immobilisation of bacterial consortium for continuous decolourisation study.
3. To assess capacity of decolourization by the selected consortium in a continuous up flow packed bed column and followed by aerobic degradation.

1.4 Scope of Study

In this research, the aim is to use a bacterial consortium capable of decolourising model azo dye Remazol Black B (RBB) in batch and in continuous system. The capacity of decolourisation of chosen consortium designated as NAR-2 were evaluated both in batch and continuous systems. The degree of decolourisation and percentage for the two systems were examined and compared. As was earlier highlighted, the decrease of azo dyes yielded aromatic amines which are more dangerous than the parent compound. Hence, the concentration of amines following decolourisation and degradation were examined with the use of ammoniacal nitrogen test analysis (TAA). The concentration of amines should be lowered after aerobic degradation of decolourised RBB, indicating NAR-2 consortium's capability to degrade or even mineralise amines.

1.5 Significance of Study

This study focus on biological treatment in batch with suspended mix culture and as continuous system utilizing biofilm formation of NAR-2 on cheap and robust support material, clinoptilolite in the degradation of textile dye, RBB under sequential microaerophilic-aerobic environment. It is hoped that from this study, the efficiency of this system in treating real textile wastewater can be improved. The use of support material such as clinoptilolite has several advantages over suspended culture, because the support can be reused over many cycles and provide a less toxic environment to the bacteria. Consequently, environmental pollution can be reduced with an ecofriendly system as such this. It is hoped that this study can offer an economic wastewater treatment alternative to the textile industry.

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