

RHEOLOGICAL, MECHANICAL, MORPHOLOGICAL AND THERMAL
PROPERTIES OF RECYCLED POLY (ETHYLENE TEREPHTHALATE)-
POLYETHYLENE FILLED MONTMORILLONITE

RIZUAN BIN MOHD ROSNAN

UNIVERSITI TEKNOLOGI MALAYSIA

RHEOLOGICAL, MECHANICAL, MORPHOLOGICAL AND THERMAL
PROPERTIES OF RECYCLED POLY (ETHYLENE TEREPHTHALATE)-
POLYETHYLENE FILLED MONTMORILLONITE

RIZUAN BIN MOHD ROSNAN

A thesis submitted in partial fulfillment of the
requirements for the award of the degree of
Master of Science (Polymer Technology)

Faculty of Chemical Engineering
Universiti Teknologi Malaysia

SEPTEMBER 2012

*Special dedications to my beloved wife and parents...
Thanks for the love and memories*

ACKNOWLEDGEMENT

In the name of Allah, the Most Beneficent, the Most Merciful. Thankful to Allah for His blessing and gracing that leads my thesis successfully completed.

The author wishes to thank Dr. Agus Bin Arsad for his support, guidance, and confidence in me, the result of which has lead to the completion of this dissertation. In addition, the author wishes to thank the following people; first and foremost his lovely wife, Farrah Bt A. Rahim for her enthusiastic support. His parents Mohd Rosnan Busri, Siti Aisah Lani, Sharifah Thalhon Sy Abu Bakar and Allahyarham A. Rahim Daud for their love and who shared with me their experiences of life and kept things in perspective.

All the members of the Polymer group past and present whose antics and advice has helped the time pass: Zahidfullah, Liyana, Ali Redza, Sarah, Hafiz, Ramli. To those in the Department of Polymer Engineering that have made the accomplishment of this feat easier: Assoc. Prof. Dr Mat Uzir Wahit, Assoc. Prof. Dr. Abdul Razak Rahmat, Dr. Zurina Mohamed. Together dedicated to lab teams which are really supporting me especially for weekend lab arrangement: Suhee Tan, Azri, Izad, Encik Nordin, Puan Zainab, Encik Said (Mediteg) and Encik Rahimi (Idemitsu-PS) and everyone else that has shown me support, which is a list too long to detail in full.

ABSTRACT

The objective of this research is to investigate the effect of incorporating nanofiller, montmorillonite (MMT) on mechanical, morphology, rheological and thermal properties of recycled poly(ethylene terephthalate) (rPET) and High density polyethylene (HDPE) nanocomposites. The MMT contents in 90:10 rPET/HDPE and 70:30 rPET/PE ranged from 1 to 5 wt.%. Blends based on rPET/HDPE nanocomposites were made through single extrusion and injection-molded into tensile and impact test samples. Samples underwent rheological test by using capillary rheometer, and the morphology of the nanocomposites was investigated by scanning electron microscopy (SEM). Thermal stability of organoclays and nanocomposites was tested by thermogravimetric analysis (TGA). The results pointed out that MMT displayed a higher compatibilizing act giving rise to a neat improvement of phase dispersion and interfacial adhesion in the blends. The maximum tensile strength was at 3 wt. % and 1 wt. % of MMT for 90:10 and 70:30 rPET/HDPE blends. However, tensile modulus decreased significantly with the incorporation of MMT. Impact strength for both blending 90:10 and 70:30 reached a maximum point at 3 wt. % and started to decrease beyond 3 wt. %. The incorporation of MMT increased the shear viscosity of 90:10 and 70:30 which reached a maximum value at 3 wt. % and 1 wt. %.

ABSTRAK

Objektif penyelidikan ini adalah untuk mengkaji kesan penambahan pengisi nano montmorillonite (MMT) terhadap pencirian morfologi, mekanikal, sifat reologi dan terma komposit bersaiz nano poli(etilena terephthalate) kitar semula (rPET) dan polietilena berketumpatan tinggi (HDPE). Komposisi komposit bersaiz nano dibahagikan kepada 90:10 dan 70:30 rPET/HDPE dengan kesan penambahan MMT dari 1 hingga 5 wt%. Komposisi komposit nano rPET/HDPE telah disediakan menggunakan mesin penyemperit berskru tunggal berpenyebati nano dan mesin acuan penyuntikan bagi membentuk sampel untuk ujian kekuatan regangan dan kekuatan hentaman. Ujian reologi dijalankan menggunakan reometer kapilari serta ciri-ciri morfologi komposit nano dilakukan menggunakan mikroskopi pengimbasan elektron (SEM). Kestabilan terma pengisi nano MMT dan komposit nano telah diuji menggunakan alat analisis termogravimetri (TGA). Keputusan menunjukkan penambahan MMT mampu menyumbang peningkatan keboleherakan fasa dan lekatan antara muka komposit. Kekuatan regangan maksima bagi komposisi komposit bersaiz nano 90:10 dan 70:30 rPET/HDPE dengan penambahan MMT adalah pada 3 wt. % dan 1 wt. %. Modulus regangan menurun secara ketara dengan penambahan MMT. Kekuatan hentaman kedua-dua komposisi komposit bersaiz nano 90:10 dan 70:30 mencapai nilai maksima pada penambahan MMT sebanyak 3 wt. % tetapi berkurang apabila melebihi 3 wt. %. MMT meningkat kelikatan ricih 90:10 dan 70:30 dengan mencapai nilai maksima pada penambahan 3 wt. % dan 1 wt. %.

TABLE OF CONTENTS

CHAPTER	TITLE	PAGE
	DECLARATION	ii
	DEDICATION	iii
	ACKNOWLEDGEMENT	iv
	ABSTRACT	v
	ABSTRAK	vi
	TABLE OF CONTENTS	vii
	LIST OF TABLES	x
	LIST OF FIGURES	xi
	LIST OF ABBREVIATIONS	xiv
1	INTRODUCTION	1
	1.1 Research Background	1
	1.2 Problem Statements	3
	1.3 Objectives	3
	1.4 Scope of Study	4
2	LITERATURE REVIEW	6
	2.1 Recycling of PET	6

2.2	Montmorillonite (MMT) - nanoclay filler	8
2.3	Recycled PET/HDPE blends	9
2.4	Recycled polymer/MMT nanocomposites	10
3	METHODOLOGY	13
3.1	Materials	13
3.1.1	Recycled PET (rPET)	13
3.1.2	High Density Polyethylene (HDPE)	14
3.1.3	Montmorillonite (MMT)	15
3.1.4	Compatibilizer (PE-g-MAH)	15
3.2	Blend Formulations	16
3.3	Preparation and treatment of waste PET flakes	17
3.4	Preparation of recycled PET/HDPE/MMT Nanocomposites	17
3.5	Characterization	18
3.5.1	Tensile Test	18
3.5.2	Izod Impact Test	18
3.5.3	Capillary Rheometer Test and Analysis	19
3.5.4	Scanning Electron Microscopy (SEM)	19
3.5.5	Thermogravimetric Analysis (TGA)	20
4	RESULT AND DISCUSSION	21
4.1	Result and Discussion	21
4.2	Tensile Properties	21
4.2.1	Tensile Strength	21
4.2.2	Tensile Modulus	23
4.2.3	Strain at Break	25
4.3	Impact Strength	26
4.4	Morphology Characterization of the Blends (SEM)	29
4.5	Rheological Properties	34

4.6	Thermogravimetric Behaviours	39
5	CONCLUSION AND RECOMMENDATION	41
5.1	Conclusion	41
5.2	Recommendation	45
	REFERENCES	46

LIST OF TABLES

TABLE NO.	TITLE	PAGE
3.1	Specification of HDPE resin	14
3.2	Specification of MMT clay	15
3.3	The typical properties of PE-g-MAH	16
3.4	Melt blending formulation	16

LIST OF FIGURES

FIGURE NO.	TITLE	PAGE
4.1	Effect of MMT concentration on different rPET/HDPE ratio on tensile strength	23
4.2	Effect of MMT concentration on different rPET/HDPE ratio on tensile modulus	24
4.3	Effect of MMT concentration on different rPET/HDPE ratio on elongation at break percentage	25
4.4	Effect of MMT addition on rPET/HDPE ratio on impact strength	27
4.5	Morphology of the fracture surface of the rPET/HDPE nanocomposites without MMT	30
4.6	Morphology of the fracture surface of rPET/HDPE with 1 phr MMT nanocomposites	31
4.7	Morphology of the fracture surface of rPET/HDPE with 3 phr MMT nanocomposites	32
4.8	Morphology of the fracture surface of rPET/HDPE with	33

5 phr MMT nanocomposites

4.9	Plots of shear viscosity as a function of shear rate for 100% rPET, and rPET/HDPE nanocomposites at different HDPE ratio.	35
4.10	Plots of shear viscosity as a function of shear rate for 90:10 rPET/HDPE nanocomposites at different loading of MMT	36
4.11	Plots of shear viscosity as a function of shear rate for 70:30 rPET/HDPE nanocomposites at different loading of MMT	37
4.12	TGA curves of 90:10 rPET/HDPE nanocomposites at 10°C/min in N ₂	39
4.13	TGA curves of 70:30 rPET/HDPE nanocomposites at 10°C/min in N ₂	41

LIST OF ABBREVIATIONS

ASTM	-	American Standard of Testing and Method
CO ₂	-	Carbon Dioxide
DSC	-	Differential Scanning Calorimetry
E-GMA	-	ethylene-glycidyl methacrylate copolymer
EMAA	-	ethylene-methacrylic acid random copolymer
EVOH	-	Ethylene vinyl alcohol
G'	-	Storage Modulus
HDPE	-	High density polyethylene
HDT	-	Heat Deflection Temperature
HDXLPE	-	High density cross-linked polyethylene
HMWPE	-	High molecular weight polyethylene
IUPAC	-	International Union of Pure and Applied Chemistry
LDPE	-	Low density polyethylene
LLDPE	-	Linear low density polyethylene
MDPE	-	Medium density polyethylene
MMT	-	Montmorillonite
MMT-IM	-	MMT treated with 1,2-dimethyl-3-octadecyl-1H-imidazol-3-ium chloride
MMT-IME	-	MMT treated with [3-(glycidyoxy)propyl] trimethoxysilane
MFI	-	Melt Flow Indexer
MT	-	Metric Tonne
PE	-	Polyethylene
PEAL	-	polyethylene and aluminium
PE-g-MAH	-	Polyethylene grafted with maleic anhydride

PET	-	Poly (ethylene terephthalate)
PETE	-	Poly (ethylene terephthalate)
PRS	-	PET-Recycling Switzerland
rPET-PE	-	recycled PET-Polyolefin
SEM	-	Scanning Electron Microscopy
TGA	-	Thermal Gravimetric Analysis
T_g	-	Glass Transition Temperature
T_c	-	Crystalline Temperature
T_m	-	Melting Temperature
UHMWPE	-	Ultra high molecular weight polyethylene
ULMWPE	-	Ultra low molecular weight polyethylene
USA	-	United States of America
VLDPE	-	Very low density polyethylene
X_c	-	Crystalline Fraction
XLPE	-	Cross-linked polyethylene

CHAPTER 1

INTRODUCTION

1.1 Background of Study

Poly (ethylene terephthalate) (PET) is the most major post consumer plastics wastes, following with polyethylene, polypropylene, polystyrene and polyvinyl chloride (Lin, 1998). In 2009, PET consumption was about 8% from the total of global polymer demand equal to 176 million MT overall (Plasticsnews, 2010). In Malaysia trend, the analysis has been done by Nexant ChemSystems and was highlighted in RMK-10 has shown the consumption about 20% of 2.6 million MT was contributed by PET. It was the second largest consumption after HDPE percentage in 2010 and expected to continue among the major polymer applications in Malaysia.

Desired final properties with desires specific purposes that are influenced by the processing parameter must be understood and this fundamental is the key to produce of new materials. Therefore, incorporation of the polymer with organoclay to nanocomposites will be the option to overcome their properties enhancement concern (Takekoshi *et al.*, 1996). The blends system of recycled PET-Polyolefin, rPET/HDPE is mostly preferred because of its low cost and the resistances of

polyolefin in contributing to high temperatures of processes needed for PET approximately between 230°C to 270°C under suitable conditions. High density Polyethylene, HDPE is one of the solutions to overcome the mechanical properties weaknesses of rPET. Because of its low glass transition temperature, it may have good performance as impact modifier to increase the impact strength of rPET. Blending postconsumer polymers can become an alternative solution to introduce a better properties and to broad the application potential of recycled plastics.

In this research, the effects of nanoclay and concentration, on the rheological and physical properties of rPET were investigated. There are very limited research had been published on recycled PET/clay nanocomposites regarding on the preparation and characterization (Pegoretti *et al.*, 2004). This future work aim is to provide guidelines for the rPET-based nanocomposites compounding with HDPE and organoclay. A scrapped PET sheet was used and the effects of organoclay addition on rheological properties of rPET/HDPE were investigated to enhance PET recycling recovery as the goal. Beside, mechanical tests were performed for characterizing the nanocomposites. The morphology was analyzed by Scanning Electron Microscopy (SEM) in order to study the extent of filler dispersed in the matrix. Thermogravimetric Analyzer (TGA) was used to study the thermal stability of the nanocomposites.

1.2 Problem Statement

PET represents one of the most successful and widespread examples of polymer recycling, mostly in the drinking bottles industry which has made PET the main target for plastic recycling. The way to ensure the rPET is ready to be used as virgin demands, the tailor-made of the enhanced properties should be achieved to overcome the drawbacks prior to the applications.

The research of rPET blended HDPE with MMT were answered several of problems.

- a) Does HDPE will be able to increase the impact resistance of rPET due to it incorporation?
- b) Does the addition of MMT will result the great contribution to rheological and mechanical strength by this blending?
- c) Does the MMT will probably help to improve the compatibility of rPET and HDPE?

1.3 Objective of Study

The objective of this study is to enhance the properties of rPET to become more stable in recycling process particularly, in the beverage industry which is the major of PET application as a drinking bottle for consumer. In this research, scrap of PET sheets was for blending with HDPE and the addition of organoclay, MMT was used as filler. Through this, the rPET/HDPE nanocomposites will be utilised into the packaging manufacturing.

Three objectives of this investigation are,

- a) To identify the effect of HDPE composition rPET/HDPE nanocomposites on rheological, mechanical and thermal stability of the nanocomposites.
- b) To investigate the effect of organoclay on the rheological properties, mechanical, morphological and thermal stability of rPET/HDPE nanocomposites.
- c) To study the effect of MMT into rPET/HDPE matrix for enhancement through morphology.

1.4 Scope of Study

i) Literature review

Elaboration of the current research and development trend of recycled PET/HDPE/MMT and related to the objective of the research study.

Sample preparation

ii) Preparation and treatment of rPET

- The sheets were washed, dried and crushed into a small flakes size before it ready to use.

iii) Preparation of rPET/HDPE nanocomposites

- a) The rPET/HDPE/MMT was prepared by melt blending method with single screw extruder.

- b) Injection moulding was for producing of testing sample for mechanical testing.

iv) Mechanical properties analysis

- a) Tensile test by using an Instron equipment.
 - To determine the elastic modulus and elongation of specimen
- b) Izod Impact test by using Toyoseiki impact machine.
 - To determine the impact strength of the nanocomposites

v) Rheological properties study by using capillary rheometer Gottfert Rheograph 2003.

-Capillary Rheometer – to determine the shear rate, shear stress and viscosity

vi) Morphology study by using SEM type JEOL JSM-6390 LV.

-Scanning Electron Microscopy (SEM) – to determine the surface and structure of specimen.

vii) Thermal properties study by using Perkin-Elmer TGA-7 instrument.

-Thermogravimetric Analysis (TGA)

-to determine the degradation of specimen

vii) Report Writing

REFERENCES

- Blumstein, A. (1961). Polymerization study of absorbed layer. *Bulletin de la Societe Chimique de France*, 889-906.
- Brandrup, J., Bittner, M., Michaeli, W. & Menges G. (1996). *Recycling and Recovery of Plastics*, Carl Hanser Verlag, N.Y.: Hanser Gardner Publication.
- Burillo G., Clough R.L., Czvikovszky T., Guven O., Le Moel A., Liu W., Singh A., Yang J. & Zaharescu T. (2002). Polymer recycling: potential application of radiation technology. *Radiation Physics. Chemistry*, 64, 41-51.
- Caldicott, R. J. (1999). *The basics of stretch blow molding PET containers*, Bridging the Millennia, NY: Society of Plastics Engineers, Inc.
- Campo, E. A. (2008). *Industrial Polymers*. Carl Hanser Verlag, M.: Hanser Gardner Publication.
- Chang, J. H., Kim, S. J., Joo, Y. L. and Im, S. (2004). Poly(ethylene terephthalate) nanocomposites by in situ interlayer polymerization: the thermo-mechanical properties and morphology of the hybrid fibers. *Polymer*, 45 (3), 919-926.

- Coran A. Y. And Patel, R. (1976). Predicting elastic moduli of heterogeneous Polymer Compositions. *Journal of Applied Polymer Science*, 20 (11), 3005-3016.
- Dimitrova, T. L., Mantia, F. P., Pilati, F., Toselli, M. Valenza, A. and Visco, A. (2000). On the compatibilization of PET/HDPE blends through a new class of copolyester. *Polymer*, 41, 4817-4824.
- Dodbiba, G. and Fujita, T. (2004). Progress in separating plastic materials for recycling. *Physical Separation in Science and Engineering*, 13(3-4), 165-182.
- Lopes, C. M. A., Felisberti, M. I., and Goncalves, M.C. (2007). Blends of poly(ethylene terephthalate) and low density polyethylene containing aluminium: A material obtained from packaging recycling. *Journal of Applied Polymer Science*, 106 (4), 2524-2535.
- Gilman, J. W., Morgan, A. B., Harris, R. H. Jr., Trulove, P. C., DeLong, H. C. & Sutto, T. E. (2000). Polymer layered silicate nanocomposites: thermal stability of organic cationic treatments. *Polymeric Materials Science and Engineering*, 83, 59-60.
- Greenland, D.J. (1963). Adsorption of polyvinyl alcohols by montmorillonite. *Journal of Colloid Science*, 18, 647-664.
- Jarukumjorn, K. And Chareunkvun, S. (2007) Compatibilization of recycled high density polyethylene (HDPE)/polyethylene terephthalate (PET) blends. *Journal of Science and Technology*, 14 (1), 1-8.
- Katbab, A, A. and Hamzehlou, S. (2007). Bottle-to-bottle recycling of PET via nanostructure formation by melt intercalation in twin screw compounder: Improved thermal, barrier, and microbiological properties. *Journal of Applied Polymer Science*, 106 (2), 1375-1382.

- Kathiravale, S., Mohd Fairus, A.F., Mohamad Azman, C.M.I., Norasalwa, Z., Khaironie, M.T. and Rohyiza B. (2009). "Waste Management-The Obligation and Opportunities in Malaysia". Retrieved on February 4, 2011, from <http://www.mint.gov.my/conference/w2w/frmprogram.html>
- Ke, Y. C., Long, C, and Qi, Z. (1999). Crystallization, properties, and crystal and nanoscale morphology of PET –clay nanocomposites. *Journal of Applied Polymer Science*, 71, 1139-1146.
- Kracalik, M., Studenovsky, M., Mikesnova, J., Kovarova, J., Sikora, A., Thomann, R. and Friedrich, C. (2007). Recycled PET-Organoclay Nanocomposites with enhanced processing properties and thermal stability, *Journal of Applied Polymer Science*. 106, 2092-2100.
- Kracalik, M., Studenovsky, M., Mikesnova, J., Sikora, A., Thomann, R., Friedrich, C., Fortelny, I. and Simonik, J. (2007). Recycled PET Nanocomposites improved by Silanization of Organoclays, *Journal of Applied Polymer Science*. 106, 926-937.
- Kracalik, M., Mikesnova, J., Puffr, R., Baldrian, J., Thomann, R. and Friedrich, C. (2007). Effect of 3D structures on recycled PET/organoclay nanocomposites, *Polymer Bulletin*, 58, 313-319.
- Kracalik, M., Mikesova, J., Puffr, R., Baldrian, J., Thomann, R. and Friedrich, C. (2007). Effect of 3D structures on recycled PET/organoclay nanocomposites. *Polymer Bulletin*, 58, 313-319.
- Lei, Y., Wu, Q., Clemons, C.M. and Guo, W. (2009). Phase Structure and Properties of Poly(ethylene terephthalate)/High-Density Polyethylene Based on Recycled Materials. *Journal of Applied Polymer Science*, 113, 1710-1719.

- Lin, C.C. (1998). Recycling technology of poly(ethylene terephthalate) materials. *Macromolecular Symposia*, 135:129-135.
- Mei, L.H.I., Bizarria, M.T.M., Giraldi, A.L.F.M., Carvalho, C.M., Velasco, J.I., and Avila, A. (2008). Morphology and thermomechanical properties of recycled PET-organoclay nanocomposites. *Journal of Applied Polymer Science*, 104, 1839-1844.
- Meyer, J.M. (1998). PET recycling wins the battle. *L'Usine Nouvelle*, 2659, 38-42.
- Moet, A. And Carte, T. L. (1993). Morphology origin of super toughness in poly(ethylene terephthalate)/ polyethylene blends. *Journal of Applied Polymer Science*, 48, 611-624.
- Murakami, O. (2001). Technology for recycling plastics materials. *Mitsubishi Electric Advance Magazine, Environmental Technology Editions*, 96, 6–9, Retrieved on Jan 19, 2011, from http://global.mitsubishielectric.com/company/r_and_d/advance/2001.html.
- Nikles, D.E. and Farahat, M.S. (2005). New Motivation for the Depolymerization Products Derived from Poly(Ethylene Terephthalate) (PET) Waste: a Review. *Macromolecular Materials Engineering*, 290, 13–30.
- Okada, A., Fukoshima, Y., Inagaki, S., Usuki, A., Sugiyama, S., Kurashi, T. & Kamigaito, O. (1988) *U.S. Patent No. 4,739,007*. Aichi: JP. Kabushiki Kaisha Toyota Chou Kenkyusho.
- Okada, A., Kawasumi, M., Usuki, A., Kojima, Y., Kurauchi, T. and Kamigaito, O. (1990). Nylon 6-clay hybrid. *Materials Research Society Symposia Proceedings*, 171, 45-50.

- Patrik, G., Raymond, S. and Roland, T. (2007). PET-bottle-to-bottle-recycling – key success factors 6 years experience in Switzerland, *REDILO 2007 Congress Davos*.
- Pegoretti, A., Kalorik, J., Peroni, C. and Migliaresi, C. (2004). Recycled poly(ethylene terephthalate)/layered silicate nanocomposites: morphology and tensile mechanical properties, *Polymer*, 45(8), 2751-2759.
- Piringer, O.G. and Baner, A.L. (2008). *Plastic packaging: interactions with food and pharmaceuticals*. (2nd ed.) Weinheim: Wiley-VCH.
- Pracella, M., Rolla, L., Chianno, D. and Galeski, A. (2002). Compatibilization and properties of (ethylene terephthalate)/Polyethylene blends based on recycled materials. *Macromolecular Chemistry and Physics*, 203, 1473-1485.
- Retolaza, A., Eguiazabal, J. I. and Nazabal, J. J. (2003). Poly(ethylene-co-methacrylic acid) – lithium ionomer as a compatibilizer for poly(ethylene terephthalate)/linear low-density polyethylene blends. *Journal of Applied Polymer Science*, 87 (8), 1322-1328.
- Retolaza, A., Eguiazabal, J. I. and Nazabal, J. J. (2002). A lithium ionomer poly(ethylene-co-methacrylic acid) and high density polyethylene. *Journal of Polymer Engineering & Science*, 42 (11), 2072-2083.
- Rujnic, M., Sercer, M. and Pilipovic, A. (2008). PET bottles recycling waste – utilization and properties. *Archives of Materials Science*, 29, 1-2.
- Takekoshi, T., Khouri, F.F., Campbell, J.R., Jordan, T.C. & Dai, K.H. (1996). *U.S. Patent No. 5, 530, 052*. Schenectady NY: General Electric Co.

- Torres, N., Robin, J.J. and Boutevin, B. (2000). Study of thermal and mechanical properties of virgin and recycled poly(ethylene terephthalate) before and after injection molding. *European Polymer Journal*, 36, 2075-2080.
- Uskov, I.A. (1960). High molecular weight compounds. *Journal of Polymer Science*, 2, 926-930.
- Utracki, L.A. (2002). Compatibilization of polymers blends. *The Canadian of Chemical Engineering*, 80, 1008-1016.
- Wang, Y., Gao, J., Ma, Y. and Argawal, U.S. (2006). Study on mechanical properties, thermal stability and crystallization behavior of PET/MMT nanocomposites. *Composites Part B: Engineering*, 37 (6), 399-407.
- Yamada, K., Kunimune, N., Leong, Y.W., Thumsorn, S. and Hahada, H. (2011). Influence of the reactive processing of recycled poly(ethylene terephthalate) / poly(ethylene-co-glycidyl methacrylate) blends. *Journal of Applied Polymer Science*, 120 (1), 50-55.
- Yao, L. and Beatty, C. (1999). The in situ compatibilization of HDPE/PET blends. *Imaging and Image Analysis Applications for Plastics*, 89-95.
- Zanetti, M., Lomakin, S. and Camino, G. (2000). Polymer layered silicate nanocomposite. *Macromolecular Materials and Engineering*, 279 (1), 1-9.