

DISSERTATION REPORT FOR MScDEGREE (MIX MODE)

SEMESTER I& SESSION 2012/2013

**SYNTHESIS OF ZEOLITE TEMPLATED GRAPHENE (ZTG) FROM
METHANOL VIA CHEMICAL VAPOR DEPOSITION (CVD) METHOD**

MOHD SYAFIQ BIN ELIAS

(MS112098)

ASSOC. PROF. DR. NOR AZIAH BUANG

**DEPARTMENT OF CHEMISTRY
FACULTY OF SCIENCE
UNIVERSITI TEKNOLOGI MALAYSIA**

2012

*For my respectful supervisor, my beloved mother and father, my siblings and my
best friend forever*

ACKNOWLEDGEMENT

I would like to express my foremost gratefulness to God for the strength and wisdom that are given to me while completing my research work. Also sincere thanks towards individuals who helped me in this project.

First and foremost, I would like to give special thanks to my beloved supervisors, Assoc. Prof Dr. Nor AziahBuang, for valuable experienced that been obtained throughout the semesters and also for the encouragement, help, guidance, constructive criticism and priceless suggestion for helping me to complete this project.

I also want to express my gratitude to all Advanced Carbon Nano Materials (ACNM) group members such as NurulAkmaliaHussin, Fatiha Ismail, FatirahFadil,JohariMd.Sallehand all the people that involved in my project for their courage, friendship, continued care and interest in helping me in till the end of this project.

Last but not least, my sincere appreciation also extends to my parents, my best friend, MohamadSaufiRosmiand never forgotten my beloved NurulAkmaliaHussinfor their prayers, moral support and indirectly involvement for helping me completed my project.

ABSTRACT

A new synthesis route in the production of graphene by template synthesis technique using zeolite as the host materials has successfully produced graphene. The highly regular ordered and highly crystalline structure of zeolite was successfully utilized for the formation of ordered sp^2 graphitized graphene structure in the zeolite porous framework. Graphitic carbon structure of zeolite template graphene (ZTG) has been synthesized via catalytic chemical vapor deposition (CVD) method from methanol as the carbon precursor. The influence factors of types of zeolite used in the template synthesis and CVD reaction temperatures have been investigated to obtain the optimum experimental condition for producing high quality of ZTG. The results show acid sites of the zeolite plays an important role in the synthesis of ZTG in porous framework of zeolite structure. CVD reaction temperatures at 500°C is considered as the best reaction temperature for the production of graphene using zeolite as template with high quality of carbon graphitic structure. UV- Visible spectroscopy and Raman spectroscopy analysis further proven the existence of sp^2 character of graphene structure with small amount of defect in the ZTG produced.

ABSTRAK

Kaedah sintesis baru untuk menghasilkan grafin dengan teknik sintesis acuan menggunakan zeolite sebagai bahan acuan telah berjaya menghasilkan grafin berkualiti tinggi dengan jumlah kuantiti yang mencukupi. Struktur yang sangat tersusun dan struktur hablur zeolite telah berjaya digunakan untuk membentuk struktur sp^2 karbon grafin yang tersusun di dalam struktur bingkai berliang zeolite. Struktur karbon grafin acuan zeolite (ZTG) telah disintesis dengan kaedah pemendapan wap kimia (CVD) daripada methanol sebagai sumber karbon. Faktor pengaruh jenis zeolite yang digunakan dalam teknik sintesis acuan dan suhu tindak balas CVD telah dikaji untuk mendapatkan keadaan eksperimen yang optimum untuk menghasilkan ZTG yang berkualiti tinggi. Keputusan eksperimen menunjukkan bahawa jenis zeolite memainkan peranan penting dalam sintesis ZTG di dalam struktur bingkai berliang zeolite. Suhu tindak balas CVD pada 500°C dianggap sebagai suhu tindak balas yang terbaik untuk menghasilkan grafin menggunakan zeolite sebagai bahan acuan dengan struktur karbon yang berkualiti tinggi. Analisis UV spektroskopidan Raman spektroskopimembuktikan lagi kewujudan kriteria sp^2 struktur grafin dengan sedikit kecacatan dalam ZTG yang dihasilkan.

TABLE OF CONTENTS

CHAPTER	TITLE	PAGE
	SUPERVISOR’S DECLARATION	ii
	AUTHOR’S DECLARATION	iii
	DEDICATION	iv
	AKNOWLEDGEMENT	v
	ABSTRACT	vi
	ABSTRAK	vii
	TABLE OF CONTENTS	viii
	LIST OF TABLES	xi
	LIST OF FIGURES	xii
	LIST OF ABBREVIATIONS	xiv
 I	 INTRODUCTION	
	1.1 Introduction	1
	1.2 Problem Statement	3
	1.3 Objective	3
	1.4 Scope of Study	4
	1.5 Significance of Study	5
	1.6 Outline of Research	6
 II	 LITERATURE REVIEW	
	2.1 Graphene: The New Allotrope of Carbon	7
	2.1.1 Properties of Graphene	8
	2.2 Synthesis of Graphene	10
	2.2.1 Chemical Reduction of Graphene Oxide (GO)	12

2.2.2	Thermal Reduction of Graphene Oxide (GO)	13
2.2.3	Epitaxial Growth on SiC	14
2.2.4	Chemical Vapor Deposition (CVD) Method	15
2.2.5	Liquid Phase Exfoliation by Organic Solvents	16
2.2.6	Solvothermal Method	17
2.2.7	Electrochemical Process	17
2.2.8	Unzipping of Carbon Nanotubes	18
2.2.9	Template Synthesis of Graphene	18
2.3	Characterization of Graphene	20
2.3.1	Fourier Transform Infrared Spectroscopy (FTIR)	20
2.3.2	UV-Visible Spectroscopy	21
2.3.3	Raman Spectroscopy	22
2.3.4	Photoluminescence (PL) Analysis	24
2.3.5	Atomic Force Microscopy (AFM)	24
2.4	Application of Graphene	25
2.4.1	Graphene as Transparent Conductive Electrode	25

III METHODOLOGY

3.1	General Instrument	27
3.2	Material	27
3.3	Procedure	28
3.3.1	Synthesis of Zeolite Template Graphene (ZTG)	28
3.3.2	Removal of Zeolite Framework in Zeolite Template Graphene (ZTG)	30
3.3.3	Dispersion Test of the Graphene Produced	30

IV	RESULTS AND DISCUSSION	
4.1	Zeolite as Host Material	31
4.2	Characterization of Zeolite Host Materials	32
4.2.1	X-Ray Diffraction Analysis of Zeolite Host Materials	32
4.2.2	Field Emission Scanning Electron Microscopy of Zeolite Host Materials	34
4.3	Synthesis of Zeolite Template Graphene (ZTG)	35
4.4	The Growth Mechanism of Zeolite Template Graphene (ZTG)	37
4.5	Thermogravimetric Analysis of Zeolite Template Graphene (ZTG)	42
4.6	Removal of Zeolite Framework in as- synthesized Zeolite Template Graphene (ZTG)	43
4.7	Dispersion Test of the Zeolite Template Graphene (ZTG)	45
4.8	Characterization of Zeolite Template Graphene (ZTG)	47
4.8.1	Fourier Transform Infrared (FTIR) Spectroscopy Analysis	47
4.8.2	UV-Visible Spectroscopy Analysis	50
4.8.3	Raman Spectroscopy Analysis	53
V	CONCLUSION	
5.1	Conclusion	58
5.2	Recommendations	59
	REFERENCES	61
	APPENDIX	68

LIST OF TABLES

TABLE NO.	TITLE	PAGE
3.1	Experimental conditions for the synthesis of ZTG via CVD reactor system	29
4.1	The Si/Al ratio in three different types of zeolites	39
4.2	Decomposition temperature of carbon materials formed in ZTG	42
4.3	Mass of carbon materials produced after removal of zeolite framework	44
4.4	The I_D/I_G ratio of the zeolite template graphene (ZTG) produced	57

LIST OF FIGURES

FIGURE NO.	TITLE	PAGE
1.1	The steps involved in synthesis of graphene in zeolite framework	2
1.2	The flow chart of the research	6
2.1	An assembled graphene/polyethylene terephthalate (PET) touch panel showing outstanding flexibility	8
2.2	FTIR spectrum of graphene oxide (GO)	21
2.3	UV-Vis spectra of graphene oxide (GO), Thermal Reduced Graphene (TRG) and Chemical Reduced Graphene (CRG)	22
2.4	Raman spectra of graphene nanosheets with different number of layers	23
3.1	Schematic diagram of the in-house built CCVD system	28
4.1	The XRD diffractogram of zeolite H-SAPO-34 (SA001) and zeolite H-SAPO-34 (SA002)	32
4.2	The XRD diffractogram of Zeolite Na-Y	33
4.3	FESEM micrograph of a) Zeolite Na-Y with 25000x magnification b) Zeolite H-SAPO-34 (SA001) with 100000x magnification and Zeolite H-SAPO-34 (SA002) with 100000 x magnifications	34
4.4	Temperature program of the synthesis of	36

	graphene process	
4.5	Formation of graphene in zeolite porous framework in the synthesis of ZTG	38
4.6	The pictures of zeolite template graphene (ZTG) solution a) after 1 hour sonication and b) after left for 2 weeks	46
4.7	FTIR Spectra of a) ZTG Y 400, b) ZTG Y 500, c) ZTG Y 600 and d) zeolite Na-Y	48
4.8	FTIR Spectra of a) ZTG H-SAPO-34 (SA001) 400, b) ZTG H-SAPO-34 (SA001) 500, c) ZTG H-SAPO-34 (SA001) 600 and d) zeolite ZTG H-SAPO-34 (SA001)	49
4.9	FTIR Spectra of a) ZTG H-SAPO-34 (SA002) 400, b) ZTG H-SAPO-34 (SA002) 500, c) ZTG H-SAPO-34 (SA002) 600 and d) zeolite ZTG H-SAPO-34 (SA002)	50
4.10	UV-Vis spectra of a) ZTG Y 600, b) ZTG Y 500 and c) ZTG Y 400	51
4.11	UV-Vis spectra of a) ZTG H-SAPO-34 (SA001) 600, b) ZTG H-SAPO-34 (SA001) 500 and c) ZTG H-SAPO-34 (SA001) 400	52
4.12	UV-Vis spectra of a) ZTG H-SAPO-34 (SA002) 600, b) ZTG H-SAPO-34 (SA002) 400 and c) ZTG H-SAPO-34 (SA002) 500	53
4.13	Raman spectra of a) ZTG Y 400, b) ZTG Y 500 and c) ZTG Y 600	54
4.14	Raman spectra of a) ZTG H-SAPO-34 (SA001) 400, b) ZTG H-SAPO-34 (SA001) 500 and c) ZTG H-SAPO-34	55

	(SA001) 600	
4.15	Raman spectra of a) ZTG H-SAPO-34 (SA002) 400, b) ZTG H-SAPO-34 (SA002) 500 and c) ZTG H-SAPO-34 (SA002) 600	56

LIST OF ABBREVIATIONS

ZTG	-	Zeolite Template Graphene
GO	-	Graphene Oxide
CVD	-	Chemical Vapor Deposition
AFM	-	Atomic Force Microscopy
FESEM	-	Field Emission Scanning Electron Microscopy
FTIR	-	Fourier Transform Infra-Red
TGA	-	Thermogravimetric Analysis
XRD	-	X-ray Diffractometer
XPS	-	X-ray Photoelectron Spectroscopy
TEM	-	Transmission Electron Microscopy
BET	-	Brunauer-Emmett-Teller
DME	-	Dimethyl Ether
NaOH	-	Sodium Hydroxide
HCl	-	Hydrochloric Acid

CHAPTER I

INTRODUCTION

1.1 Introduction

In the latest rising frontier materials in the area of materials science research since its discovery in 2004 [1], graphenes have attracted an intensive intention in recent years [2]. A novel two-dimensional [3] graphitic honeycomb lattice carbon structure [4] of graphene with sp^2 hybridization[5, 6] which only one atom thick [2, 3] possesses very fascinating electronic, thermal and mechanical properties [7]. Graphene has ultrahigh electron mobility[8], zero electronic band gap with [4], tunable band gap, quantum electronic transport[8], high thermal stability and conductivity[9], high elasticity, and electromechanical modulation[8].

In addition to its remarkable and outstanding sheets resistance, ballistic transport, fascinating flexibility[4] and transparency of graphene-based thin film[10] have raised the eyebrows of the researchers in applying this unique special material in advanced technologies such as nanoelectronics, transparent conducting electrodes, nanocomposites, supercapacitors, sensors, batteries, and other area of technologies [4]. The encouraging aspects of graphene is to use it in the touch panels in electronic devices, e-paper, organic light-emitting display (OLED), and solar cells[4].

From the discoveries of graphene until now, many researchers have done several attempts especially in preparing high quality of graphene and higher yield with low cost operation procedures [4]. In addition, the outstanding properties of graphene are hugely depending on the synthesis method, surface roughness and also graphene-

substrate interaction [8, 11]. Recently, the synthesis of graphene in large scale have been reported including liquid suspension graphene oxide (GO) followed by reduction by chemical treatment[3], thermal reduction of graphite oxide, epitaxial growth on SiC, liquid-phase exfoliation by organic solvents, epitaxial growth by chemical vapor deposition (CVD), solvothermal synthesis, and unzipping of carbon nanotubes [4, 12].

Besides that, despite the general top-down approach of preparation of graphene, some researchers also have introduced the bottom-up chemical approach towards the synthesis of the graphene-based film. It has been achieved by the thermal reaction of synthetic nano-graphene molecules of giant polycyclic aromatic hydrocarbons (PAHs) which are cross-linked together and fused further to form larger graphene sheets [13].

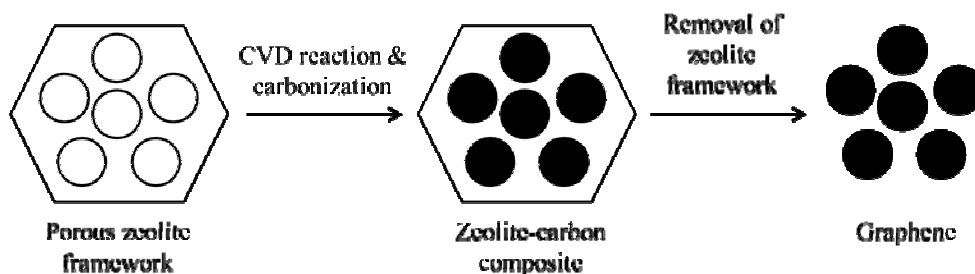


Figure 1.1: The steps involved in synthesis of graphene in zeolite framework

Herein different approach has been done in order to prepare the remarkable unique graphene materials. In this research, the graphene was synthesized inside some host materials using chemical vapor deposition (CVD) method from alcohol as the carbon precursor. Several host materials which is Zeolite H-Y type, Zeolite H-SAPO-34 (SA001) and Zeolite H-SAPO-34 (SA002) have been used to synthesis zeolite template graphene (ZTG) via CVD method. Those zeolite host materials act as the template for the growth of graphene as shown in Figure 1.1. Different CVD reaction temperatures that contribute to the preparation of graphene in zeolite pores have also been studies in this research.

1.2 Problem Statement

As describes before, the encouraging part of the use of graphene is its application in touch panel electronic devices. A transparent electrically conductive material is one of the critical components as electronic devices specifically in photoelectronic devices [9]. Presently, the state-of-the-art in transparent conductive electrode uses the indium tin oxide (ITO) due to its high electrical conductivity and optical transparency [10].

However, ITO appears to be increasingly problematic because of incompatibility with large scale productions[5]. Besides that, the ITO becomes increasingly expensive due to limited availability of the indium element on earth[6, 13]. Furthermore, the devices problematic like susceptible ion diffusion into the polymer layers and mechanical brittleness limit the applicability of ITO as the flexible photoelectronic devices[5]. Thus, thin graphene-based film is one of the best candidates which offer similar or better performances as ITO with cost-efficient method as the transparent conductive material.

Besides that, the problem statement of this research is the possibilities to synthesis zeolite template graphene (ZTG) in zeolite pores as the host materials via CVD method approach. Different host materials will leads to different physical and chemical properties of graphene produced. The CVD reaction parameters also play important roles in synthesizing graphene in the zeolite host materials. In this research, reaction parameters of CVD system which is the reaction temperatures have been studied to obtain better quality of graphene.

1.3 Objectives

The objectives of the research are:

1. To synthesis zeolite template graphene (ZTG) from different types of zeolite via chemical vapor deposition (CVD) method.

2. To investigate the effect of reaction temperature in the synthesis of zeolite template graphene (ZTG) during chemical vapor deposition (CVD) process.

These objectives will lead to find the best condition for producing high quality graphene sheets and transparent graphene film.

1.4 Scope of Study

Different types of zeolites were chosen as the host materials to synthesis the zeolite template graphene (ZTG) in the zeolite pores materials which is Zeolite H-Y type, Zeolite H-SAPO-34 (SA001) and Zeolite H-SAPO-34 (SA002). The synthesis of ZTG was then carried out in an in-house built chemical vapor deposition (CVD) reactor system from methanol as the carbon precursor. Reaction temperatures was varied and studied in order to obtain the ZTG with different quality and properties according to the different types of zeolite host materials. The ZTG produced was characterized using Thermal Gravimetry Analysis (TGA) to measure the different thermal range effect on the ZTG.

The ZTG produced then were undergone further chemical treatment with acid and base to remove the zeolite framework that act as the host for the production of graphene. The graphene obtained was dispersed in water as the medium to investigate the dispersion ability of the graphene produced. The dispersed solutions were than characterized using UV-spectrometer. The graphene obtained were characterized using Fourier-Transform Infrared Spectrophotometer (FT-IR) and Raman Spectroscopy.

1.5 Significance of Study

When the graphene sheets are successfully synthesized inside the zeolite pores as the host materials, it can open a new frontier of research regarding the synthesis approach of graphene sheets. Smaller size range of graphene sheets can be engineered

as the growth of graphene take place in the pores of zeolites which the size of graphene sheets formed will depend on the pore size of the zeolite as the host materials.

In addition, graphene that has been successfully synthesized in this research can be used to be applied in the fabrication of transparent conductive film. The transparent conductive film prepared from graphene is able to provide similar or better performances in conductivity as compared to indium tin oxide (ITO) which is widely used as the transparent electrode in touch panel devices. Besides that, graphene are the best candidates to replace ITO in the transparent electrode technology due to the low cost manufacturing of this remarkable material.

1.6 Outline of the Research

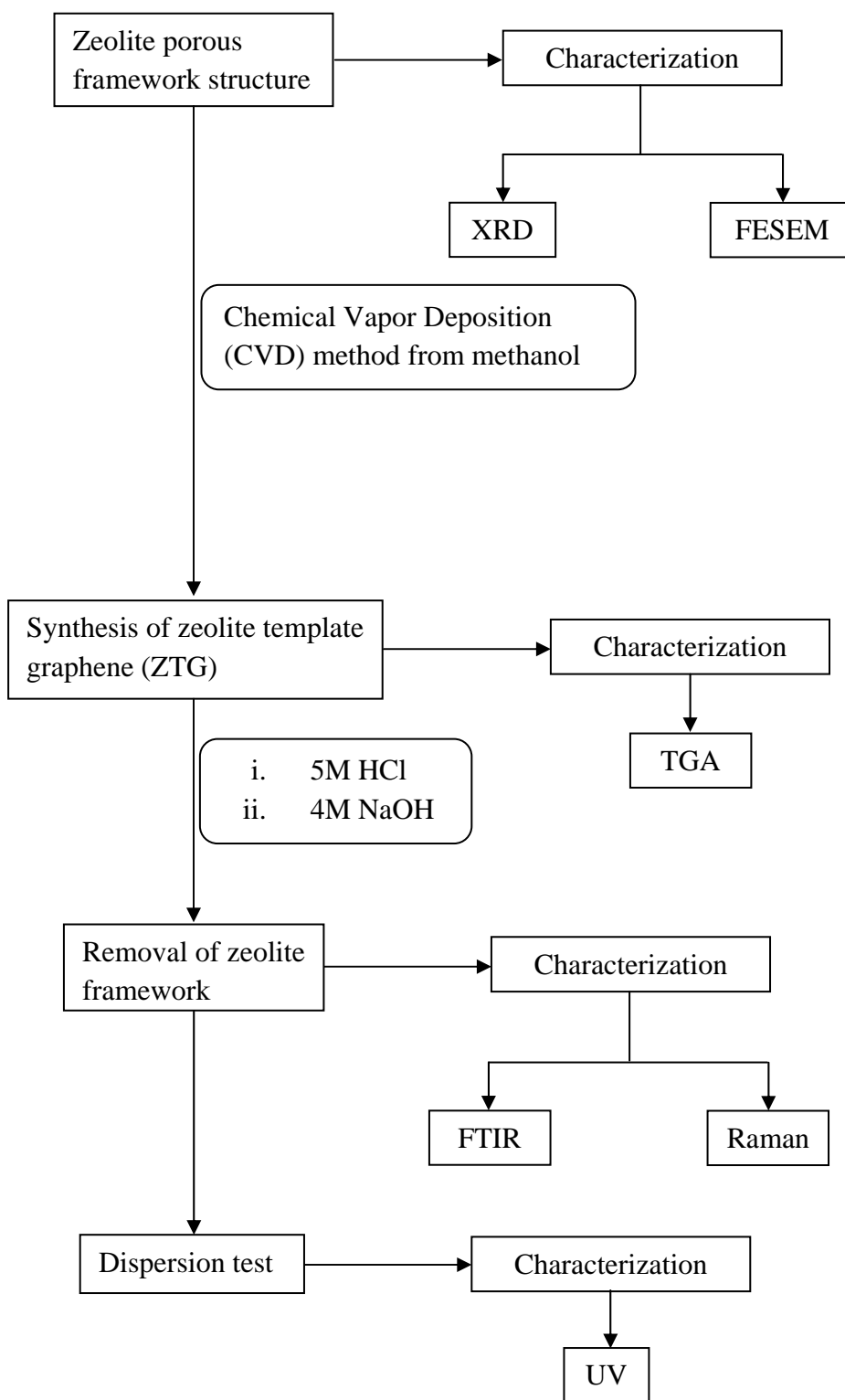


Figure 1.2: The flow chart of the research

REFERENCES

1. Rattana, T., Chaiyakun, S., Witit-anun, N., Nuntawong, N., Chindaudom, P., Oaew, S., Kedkeaw, C., and Limsuwan, P. (2012). Preparation and Characterization of Graphene Oxide Nanosheets. *Procedia Engineering*. **32**, 759-764.
2. Zhang, C., Tjiu, W.W., Fan, W., Yang, Z., Huang, S., and Liu, T. (2011). Aqueous Stabilization of Graphene Sheets Using Exfoliated Montmorillonite Nanoplatelets for Multifunctional Free-standing Hybrid Films via Vacuum-assisted Self-assembly. *Journal of Materials Chemistry*. **21**, 18011-18017.
3. Wan, L., Wang, S., Wang, X., Dong, B., Xu, Z., Zhang, X., Yang, B., Peng, S., Wang, J., and Xu, C. (2011). Room Temperature Fabrication of Graphene Films on Variable Substrates and Its Use as Counter Electrode for Dye-synthesized Solar Cells. *Solid States Sciences*. **13**, 468-475.
4. Oh, S. Y., Kim, S. H., Chi, Y. S., and Kang, T. J. (2012). Fabrication of Oxide Free Graphene Suspension and Transparent Thin Films using Amide Solvent and Thermal Treatment. *Applied Surface Science*. **258**, 8837-8844.
5. Kymakis, E., Stratakis, E., Stylianakis, M. M., Koudoumas, E., and Fotakis, C. (2011). Spin Coated Graphene Films as the Transparent Electrode in Organic Photovoltaic Devices. *Thin Solid Films*. **520**, 1238-1241.

6. Zheng, Q. B., Gudarzi, M. M., Wang, S. J., Geng, Y., LI, Z., and Kim, J. K. (2011). Improved Electrical and Optical Characteristics of Transparent Graphene Thin Films Produced by Acid and Doping Treatments. *Carbon*.**49**, 2905-2916.
7. Sun, J., Liu, H., Chen, X., David, G. E., Yang, W., and Duan, X. (2012). Synthesis of Graphene Nanosheets with Good Control over the Number of Layers within the Two-dimensional Galleries of Layered Double Hydroxides. *Chem. Commun.***48**, 8126-8128.
8. Cuong, T. V., Pham, V. H., Tran, Q. T., Hahn, S. H., Chung, J. S., Shin, E. W., and Kim, E. J. (2010). Photoluminescence and Raman Studies of Graphene Thin Films Prepared by Reduction of Graphene Oxide. *Materials Letters*.**64**, 399-401.
9. Wang, S. J., Geng, Y., Zheng, Q., and Kim, J. K. (2010). Fabrication of Highly Conducting and Transparent Graphene Films. *Carbon*.**48**, 1815-1823.
10. Xu, Y., Long, G., Huang, L., Huang, Y., Wan, X., Ma, Y., and Chen, Y. (2010). Polymer Photovoltaic Devices with Transparent Graphene Electrodes Produced by Spin Casting. *Carbon*.**48**, 3308-3311.
11. Ishigami, M., Chen, J. H., Cullen, W. G., Fuhrer, M. S., and Williams, E. D. (2007). Atomic Structure of Graphene on SiO₂. *Nano Letters*.**7**, 1643-1648.
12. Singh, V., Joung, D., Zhai, L., Das, S., Khondaker, S. I., and Seal, S. (2011). Graphene Based Materials: Past, Present and Future. *Progress in Materials Science*.**56**, 1178-1271.
13. Wang, X., Zhi, L., Tsao, N., Tomovic, Z., Li, J., and Mullen, K. (2008). Transparent Carbon Films as Electrode in Organic Sola Cells. *Nanostructures*.**47**, 2990-2992.

14. Venugopal, G., Krishnamoorthy, K., Mohan, R., and Kim, S. J.(2012). An Investigation of the Electrical Transport Properties of Graphene Oxide Thin Films. *Materials Chemistry and Physics*.**132**, 29-33.
15. Delgado, J. C., Kim, Y. A., Hayashi, T., Gomez, A. M., Hofmann, M., Muramatsu, H., Endo, M., Terrones, H., Shull, R.D., Dresselhaus, M. S., and Terrones, M. (2009). Thermal Stability Studies of CVD-grown Graphene Nanoribbons: Defect Annealing and Loop Formation.*Chemical Physics Letters*.**469**, 177-182.
16. Pu, N. W., Wang, C. A., Liu, Y. M., Sung, Y., Wang, D. S., Ger, and M. D. (2012). Dispersion of Graphene in Aqueous Solutions with Different Types of Surfactants and the Production of Graphene Films by Spray or Drop Coating. *Journal of the Taiwan Institute of Chemical Engineers*.**43**, 140-146.
17. Alanyalioglu, M., Segura, J. J., Oro-Sole, J., and Pastor, N. C. (2012). The Synthesis of Graphene Sheets with Controlled Thickness and Order Using Surfactant Assisted Electrochemical Processes. *Carbon*.**50**, 142-152.
18. Su, X., Wang, G., Li, W., Bai, J., and Wang, H. (2013). A Simple Method for Preparing Graphene Nano Sheets at Low Temperature. *Advanced Power Technology*. **24(1)**, 317-323.
19. Pei, S., and Cheng, H. M. (2012). The Reduction of Graphene Oxide. *Carbon*. **50**, 3210-3228.
20. Dhakate, S. R., Chaunan, N., Sharma, S., and Mathur, R.B. (2011). The Production of Multi-Layer Graphene Nanoribbons from Thermally Reduced Unzipped Multi-Walled Carbon Nanotubes. *Carbon*. **49**, 4170-4178.

21. Nguyen, S. T., Nguyen, H. T., Rinaldi, A., Nguyen, N. P. V., Fan, Z., and Duong, H. M. (2012). Morphology Control and Thermal Stability of Binder less-Graphene Aerogels from Graphite for Energy Storage Applications. *Colloids and Surfaces A: Physicochemical and Engineering Aspects*. **414**, 352-358.
22. Cataldo, F., Compagnini, G., Patane, G., Ursini, O., Angelini, G., Ribic, P. R., Margaritondo, G., Cricenti, A., Palleschi, G., and Valentini, F. (2010). Graphene Nanoribbons Produced by The Oxidative Unzipping of Single-Wall Carbon Nanotubes. *Carbon*. **48**, 2596-2602.
23. Chowdury, F. A., Morisaki, T., Otsuki, J., and Alam, M. S. (2012). Optoelectronic Properties of Graphene Oxide Thin Film Processed by Cost-Effective Route. *Applied Surface Science*. **259**, 460-464.
24. Jo, S. B., Park, J., Lee, W. H., Cho, K., and Hong, B. H. (2012). Large-area Graphene Synthesis and Its Application to Interface-Engineered Field Effect Transistors. *Solis State Communications*. **152**, 1350-1358.
25. Soldano, C., Mahmood, A., and Dujardin, E. (2010). Production, Properties and Potential of Graphene. *Carbon*. **48**, 2127-2150.
26. Cuong, T. V., Pham, V. H., Tran, Q. T., Chung, J. S., Shin, E. W., Kim, J. S., and Kim, E. J. (2010). Optoelectronic Properties of Graphene Thin Films Prepared by Thermal Reduction of Graphene Oxide. *Materials Letters*. **64**, 765-767.
27. Vadukumpully, S., Paul, J., Mahanta, N., and Valiyaveetil, S. (2011). Flexible Conductive Graphene/Poly(Vinyl Chloride) Composite Thin Films with High Mechanical Strength and Thermal Stability. *Carbon*. **49**, 198-205.

28. Dong, L. X., and Chen, Q. (2010). Properties, Synthesis, and Characterization of Graphene. *Frontier Material Science China*. **4**(1), 45-51.
29. Dideykin, A., Aleksenskiy, A. E., Kirilenko, D., Brunkov, P., Goncharov, V., Baidakova, M., Sakseev, D., and Vul, A. Y. (2011). Monolayer Graphene from Graphite Oxide. *Diamond and Related Materials*. **20**, 105-108.
30. Huang, L., Chang, Q. H., Guo, G. L., Liu, Y., Xie, Y. Q., Wang, T., Ling, B., and Yang, F. H. (2012). Synthesis of High Quality Graphene Films on Nickel Foils by Rapid Thermal Chemical Vapour Deposition. *Carbon*. **50**, 551-556.
31. Choi, Y. Y., Kang, S. J., Kim, H. K., Choi, W. M., and Na, S. I. (2012). Multilayer Graphene Films as Transparent Electrodes for Organic Photovoltaic Devices. *Solar Energy Materials & Solar Cells*. **96**, 281-285.
32. Wang, Y., Sun, H., Zhang, R., Yu, S., and Kong, J. (2012). Large Scale Templated Synthesis of Single-Layered Graphene with a High Electrical Capacitance. *Carbon*. In Press.
33. Chen, W., Yan, L., and Bangal, P. R. (2010). Preparation of Graphene by the Rapid and Mild Thermal Reduction of Graphene Oxide Induced by Microwave. *Carbon*. **48**, 1146-1152.
34. Geng, Y., Wang, S. J., and Kim, J. K. (2009). Preparation of Graphite Nanoplatelets and Graphene sheets. *Journal of Colloid and Interface Science*. **336**, 592-598.
35. Terrones, M., Mendez, A. R. B., Delgado, J. C., Urias, F. L., Cantu, Y. I. V., Macias, F. J. R., Elias, A. L., Sandoval, E. M., Marquez, A. G. C., Charlier, J. C., and Terrones, H. (2010). Graphene and Graphite

- Nanoribbons: Morphology, Properties, Synthesis, Defects and Applications. *Nano Today*. **5**, 351-372.
36. Wu, X., Hong, X., Nan, J., Luo, Z., Zhang, Q., Li, L., Chen, H., and Hui, K. S. (2012). Electrochemical Double-Layer Capacitor Performance of Novel Carbons Derived from SAPO Zeolite Templates. *Microporous and Mesoporous Materials*. **160**, 25-31.
37. Su, F., Zhao, X. S., Lv, L., and Zhou, Z. (2004), Synthesis and Characterization of Microporous Carbons Templated by Ammonium-form Zeolite Y. *Carbon*. **42**, 2821-2831.
38. Kawabuchi, Y., Oka, H., Kawano, S., Mochida I., and Yoshizawa, N. (1997). The Modification of Pore Size in Activated Carbon Fibers by Chemical Vapour Deposition and Its Effects on Molecular Sieve Selectivity. *Carbon*. **36**, 377-382.
39. Wu, J., Bai, s., Shen, X., and Jiang, L. (2010). Preparation and Characterization of Graphene/CdSNanocomposites. *Applied Surface Science*. **257**, 747-751.
40. Takai, K., Suzuki, T., Enoki, T, Nishihara, H., and Kyotani, T. (2010). Fabrication and Characterization of Magnetic Nanoporous Zeolite Templated Carbon. *Journal of Physics and Chemistry of Solids*. **71**, 565-568.
41. Munoz, T. A., Alvarez, C. M., and Sastre, E. (2012). Use of Different Templates on SAPO-34 Synthesis: Effect on the Acidity and Catalytic in the MTO Reaction. *Catalysis Today*. **179**, 27-34.
42. Su, F., Zeng, J., Yu, Y., Lv, L., Lee, J. Y., and Zhao, X. S. (2005). Template Synthesis of Microporous Carbon for Direct Methanol Fuel Application. *Carbon*. **43**, 2366-2373.

43. Chen, D., Moljord, K., and Holmen, A. (2012). A Methanol to Olefins Review: Diffusion, Coke Formation and Deactivation on SAPO Type Catalysts. *Microporous and Mesoporous Materials*. **164**, 239-250.
44. Liu, G., Tian, P., Li, J., Zhang, D., Zhou F., and Liu, Z. (2008). Synthesis, Characterization and Catalytic Properties of SAPO-34 Synthesized Using Diethylamine as a Template. *Microporous and Mesoporous Materials*. **111**, 143-149.
45. Marchese, L., Frache, A., Gianotti, E., Martra, G., Causa, M., and Coluccia, S. (1999). ALPO-34 and SAPO-34 Synthesized by using Morpholine as Templating Agent. FTIR and FT-Raman Studies of the Host-Guest and Guest-Guest Interactions within the Zeolitic Framework. *Microporous and Mesoporous Materials*. **30**, 145-153.