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Current trends in additive manufacturing of selective laser melting for biomedical implant applications

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ABSTRACT

Additive manufacturing (AM) has obtained great attentions in the productions of customizable implants with complex shapes and good mechanical properties. AM fabricated implants of Ti alloy (Ti), stainless steel (SS), cobalt chromium (CoCr), iron (Fe) have been revolutionized in the productions of medical instruments, such as in artificial organs, including heart stents, drug delivery, scaffolds and orthopedic implants. However, high Young's modulus, superior mechanical characteristic, lack of surface integration, high corrosions, failed bone remodeling, and antibacterial performance are still incomparable to be used in anatomical structure. This review studied the issue of metallic implants designed by powder bed fusion of selective laser melting (SLM) techniques on the surface morphology, stress shielding effects, osseointegration, and antibacterial capabilities owing to implant failure. The rate of degradation and corrosion resistance of the metallic SLM implant, which would have an impact on cell proliferation and bone remodeling between the surrounding tissue, were discussed. The current review article also highlighted the various SLM parameters, like mechanical, chemical, and biological modifications, that might enhance the implant. Finally, applications, challenges, opportunities, and future developments of metallic implants using SLM techniques were addressed.

1. Introduction

The emerging technology known as additive manufacturing (AM) has the ability to produce significant results in the field of biomedical. This AM principle is a layer-by-layer manufacturing by utilizing pretooled (three-dimensional printing) 3D model computer-aided design (CAD) programming in a protective setting [1]. AM has the advantage of rapidly and accurately producing customized implants, complex parts at a reasonable cost with less waste being generated [2,3]. This AM technology has various potential applications in electronics, automotive, aeronautical, medicine and dentals. Among these, biomedical field has attracted the AM applications attributed to the freedom of fabrications and development of patients specific implants, such as prosthetics, stents, scaffolds, and drug delivery systems [4-6]. The widely use of AM in medical field attributed to their customizations and personalization of fabricating medical implants based on the fractured and infected sites. Based on the report by American Professional Association, the AM medical implants are estimated to rise by up to \$9.7 billion by the year of 2027 with 35% growth yearly specifically in dental application [7]. The use of AM implants revenue in medical industries are increasing to about 11% from the AM printed components whether on devices or implants such as orthopedic and dental [8]. The metallic biomaterial mechanical properties should include high strength with low modulus so it can function for a longer period to withstand high load of contract and relax of skeletal muscle [9]. The stress shielding effects or implant resorption occurs when there is a mismatch of mechanical properties of the metallic implant and native body. This effects lead to the implant failure and failed bone remodeling. Thus, the development of metallic materials by using AM technique is viable for the customizations of biomedical implants with porous structure and mechanical characteristics that match to the anatomical structure.

The surface of the implant plays a crucial role in cell adhesion or osteointegration which involves the microstructure, macrostructure as well as the nanostructure characteristics. The cell remodeling were fastens with a rougher AM surface as compared to the smoother surface. This alloy for the osteogenic cell attachment on the surface. Metallic

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alloy are commonly used in the field of biomedical part from composite and polymers, due to their higher biocompatibility and bioinert. The biocompatible implant materials include cobalt(Co), titanium (Yi), magnesium (Mg) and stainless steel (SS). These implant able to react with anatomical microenvironment without causing toxicity. The main properties of the implant should consist of: (a) porous and lattice structure, (b) bio activeness and biocompatibility, (c) good mechanical properties, (d) high corrosion resistance, (e) wear resistance, and (f) good osseointegration. These properties are mainly for the successful cell attachment and among these characteristic, pore is more prominent for the design of the metallic materials to allow the exchange of oxygen and minerals between the cells in the interconnected pores [5]. Apart from that, the mechanical characteristic should be matched to the neighboring tissue where the implant is placed so that it can withstand the stress shielding effects. Current metallic implants concerns are linked to the lack of corrosion resistance and low osseointegration which may compromise the function of the biomedical implants. The implants degradations may cause revision of surgeries, pain and failed implants.

For implants in the field of AM, there are seven techniques consist of binder jetting (BJ), directed energy deposition (DED), material extrusion (ME), material jetting (MJ), powder bed fusion (PBF), sheet lamination as well as vat photopolymerization (VPP). These AM techniques have been well recognized and identified in the ISO/ASTM 52900 Standard [10]. Nevertheless, not all these AM techniques can be utilized in medical devices and fabrication. As shown in Table 1, the benefit and drawbacks of these AM processes are summarized for the biomaterials fabrication. Furthermore, selecting the proper AM process for the fabrication of metallic materials is crucial for the development of biomaterials. From Table 1, metal powders are commonly used to produce implants by using powder bed fusion process because of their complex implant designs.

Currently, SLM biometallic implants are mostly focused on replacing tissue repair, dentistry, surgical tools as well as orthopedic. In upcoming future, the development of SLM printed metals will be further extended to form new biometallic applications for the implantable or metal bioelectronic [26]. SLM technique can print different types of materials, including metal, polymer, and metal combinations as well as metal-ceramic combinations. Due to its precision and near-density parts, it is widely used in medical applications to form complex structures implants based on anatomical structures. As SLM has becomes a main technology in the medical field, its stages in medical application is still in the initial phases prior to its clinical use [27]. This SLM technique is not as well established as those of the conventional method in the field of biomedical. The interaction between the metal powders, control of the residual stress, the external and internal defects and other technical problems still needed to be optimized to suit the microenvironment as well anatomic structure of human body [28]. The SLM accuracy is still insufficient as SLM adopted the layer-by-layer process, whereby the bond between each layer is compact, the defects affect the product density and strength, leading to a poor performance of the fabricated products.

Besides, in SLM fabrication, high quality raw materials are needed. These raw materials owing to various molding techniques either powdery or through filament to meet the specific requirement of the biomedical implants. These materials need to meet the requirement of implants in preventing any toxicity and biological risk. Then, for an implant to be used in clinical application, strict sterilization of implants is one of the crucial parts to ensure the success of implant. Sterilization of implant involves the placement of implant into the autoclave with high temperature of up to 130 °C. This is to ensure the implant is safe from the production of microorganisms, including bacteria, spores and fungi before implantation [29]. Commonly, the implant will be cleaned prior to the sterilization process. Thus, improper sterilization of the SLM implants will have a significant effect on the implant surface and properties. It is important to select the proper morphology and properties, especially to those of SLM implants with complex cellular structures [30,31].

This current review explained the previous strategies for the current trend of metallic SLM fabricated implant that have been used and identified their optimum parameters. Also, emphasized the most ideal alternative for the surface enhancement of the porous implant to be used in the medical field. The paper also presented the current challenges and limitations of AM manufacturing techniques *in vivo* and *in vitro* for the fabrication of AM implants. The successful AM implant would reduce stress shielding, enhance bone remodeling, and increase bacterial resistance to the antibiotic, which may be a major health concern after implantation.

Table 1

Tuble 1						
The materials	used a	and	clinical	application	of A	M

Method		Materials	Benefit	Drawback	Clinical Applications	Ref
Material Deposition	Material Extrusion (FDM)	Hydrogels, bioinks, ceramic, thermoplastic	Affordable, ease to find, biocomposite material, easily design	Time consuming, anisotropy, less resolution, high force of the nozzles impart	Bioprinting for cell culture, tissue and organ, rigid and soft models for surgical	[11,12]
	Directed Energy Deposition	Metallic metals	Fast, composite materials, high density	High cost, low resolution, needed post-processing	Limited use in clinical applications	[13,14]
	Material Jetting	Photopolymer, bio-inks	Good resolution, high cell proliferation, many cell/material deposition	Vey slow, high waste material, limited material selection, limited fabrication size	Bioprinting of soft tissue scaffolds for tissue and organ	[15–17]
Powder Based	Selective laser Sintering, Selective Laser Melting, Electron Beam Melting	Thermoplastics, metals powder, ceramic powders	High dense and strength parts, efficient, no needed of solvent	High cost, required post processing	Dental implant, orthopedic implant, wires, screw, rigid implant	[14, 18–20]
	Binder Jetting	Metal, polymer, ceramic,	Affordable, fast, comes in multicolor, no support	Low strength, required post processing, powders might cause hazard	Metallic implants of hard tissue	[21–23]
Liquid based	Stereolithography, Digital Light Processing	Ceramic resin, bio-resin, photopolymer,	High resolution, fast, high cell viability, no nozzle	Cause toxicity, limited material, uv might harm the DNA	Cell culture tissue and organ development	[12,24, 25]

2. Classification of additive manufacturing in metallic implants

AM has been used in fabricating biomedical implants attributed to their precision and ability to fabricate complex shaped implant altered to the anatomical structure of human body. This advantage has piqued the interest of industrial organizations and researchers in expanding the use of AM. AM involves layer-by-layer fabrications which allows the fabrication of metal, ceramic, and polymers. AM processes are divided into seven groups by the American Society for Testing/International Organization for Standardization (ASTM/ISO) 52,900:2015 standard which and Materials American Society for Testing and ISO. As illustrated in Fig. 1, The powder-based AM methods, such as DED (Fig. 1(a)), BJ (Fig. 1(b)), EBM (Fig. 1(c)), laser engineered net shaping (LENS) (Fig. 1 (d)) and SLM (Fig. 1(e)) are widely used in the fabrication of the packed and porous metallic implants as illustrated in. Nevertheless, other methods such as cold spraying, direct metal writing and diode-based processes have been proposed by researcher for the manufacture of metallic materials but are still under evaluation for their potential by



Fig. 1. Schematic illustration of current AM Processes (a) DED [32] (b) of binder jetting printing process [33] (c) electron beam melting process [34] (d) laser engineered net shaping (LENS) system [35] & selective laser melting [36].

ASTM to be recognized and classified in AM.

2.1. Directed energy deposition (DED)

DED is an AM process by transferring the energy between the fabricated material concurrently. DED is classified as electron beam AM due to its heat source is an electron beam. The fabricated materials are provided in the form of particles or wire, and a laser beam then melts the material on the desired surface in a vacuum chamber, as shown in Fig. 1 (a). The process commonly involves metal; however, it can still be used like ceramics. The type of gas utilized depends on the material, although argon is the most used to decrease oxidation in the molten pool. Compared to other AM methods, DED can be used as an addition or repairing the existing components. It also has the benefit of fabricating components with hybrid structures with of multiple materials of various compositions and structures [37]. DED has been previously employed for metallic implants, such as pure titanium (Cp-Ti) as well as porous structure for load bearing applications. The majority of the high cooling rates are attained during the DED process, which maintains the metastable β -phase of the Ti alloy intact. A study by Fisher et al. [38] found that these fast-cooling rates cause the laser deposited of beta-type Ti alloys of Ti-27.5Nb present an elastic modulus that almost similar to the cancellous bone. However, after the addition of oxygen content, the strength of the material decreases. This is due to the high reactivity of the oxygen which causes the martensitic phase transformations of Ti decreases with the increasing of oxygen content. Gonzalez et al. [32] fabricated DED of Ti-Nb and Ti-Zr-Nb and both of the sample demonstrated no apparent imperfections includes cracks or pores. However, there was formation of large, elongated columnar grain that grow parallelly to the building direction of the DED even after the sample has undergone polishing and etching with Kroll's reagent on the lower part (Fig. 2). These grains cause the DED sample to have a lower wear resistance. Another approach by Todaro et al. [39] utilized the high-intensity acoustic vibration in order to reduce the grain structure in the melt pool of DED-fabricated Ti6Al4V by stimulating the ultrasound. The microstructural analysis demonstrated the differences between the sample with and without ultrasound (Fig. 3). The sample without the ultrasound showed the formation of prior- β grains of ~ 0.5 mm in width, as shown in Fig. 3(a) and (c). The sample with the ultrasound, on the other hand, exhibits a fine structure of prior- β grains of $\sim 100 \mu m$ in Fig. 3 (b) and (d). The study concluded that the prior- β grain density increases with the ultrasound, indicating that the ultrasound enhances the nucleation during the DED solidification. The distribution of grain size and aspect ratio of prior- β grain on the sample drastically changed after ultrasound, demonstrating improved prior-B grain structure (Fig. 3(e) and (f)).



Fig. 2. Microstructure in the longitudinal XZ consist of elongated columnar prior beta grains parallel to the building directions [32].

2.2. Binder jet (BJ)

Binder Jet (BJ) printing is an AM method that printed liquid binding agent to generate a three-dimensional (3D) design on powder during manufacturing. It was first introduced in Massachusetts Institute of Technology (MIT) back in the early 90's [40]. The advantage of BJ is that it can be adapted to almost any type of metal power with a high production rate. It has broad range of technologies, including dynamic/powder interaction as well as post processing method. As shown in Fig. 1(b), BJ starts with applying a thin coating of powder to the printer bed. Next, the inkjet print head moves across the platform, depositing a droplet of binder in accordance with the data in the CAD file. Once the first layer has been printed out, the print head lowers the build platform, and the supply bed is lifted by one-layer thickness. However, this method demonstrates a reduction of the local temperature at a melt pool surface, which increases the presence of porosity due to the printing process that is performed at room temperature. This makes it impossible to produce metallic materials with high relative densities, leaving them unsuitable for use in biomedical implants for fracture fixation [41]. A previous study has reported on the use of BJ in fabricating biodegradable implant and scaffold, whereby they found that the BJ implant exhibit pores with positive benefits in in vitro degradation and cell adhesion. Chua et al. [42] combined the mixture of Fe-30 wt% with an open porosity of 36% that obtained through BJ. The corrosion rate of the implant sample was higher than the pure Fe by using potentiodynamic polarization test. The cytocompatibility showed an excellent cell infiltration on MC3T3-E1 pre-osteoblast cell line. These study were then followed by Hong et al. [43], whereby they utilized the Fe-35 wt% Mn with an addition of Calcium(Ca) and Magnesium(Mg). They confirmed the BJ sample showed that enhanced degradation of Ca was greater than that of the sintered compacted pellets. The MTT assays exhibited good live cells after 72 h of incubation.

2.3. Electron beam melting (EBM)

Electron beam melting (EBM) is an AM technology that holds potential in biomedical industries. The metal powder is melted by the electron beam in this powder bed fusion AM process to create complex parts. The materials fabricated in a vacuum chamber layer by layer, whereby the powder feedstock is supplied from the hopper in adjacent to the build chamber as shown in Fig. 1(c). EBM has the benefit of fabricating fine feature's part which involves mashes with internal structures. Typically, the direction of the lack-of-fusion faults is perpendicular to the build direction [44]. EBM showed pores on the surface of sample that caused from the leftover argon as trapped from the gas-atomization process that occurred in the powder during fabrications [44,45]. Researchers has been focused on the effect of EBM on the manufacturing, post processing as well the microstructure of the AM part. Szymczyk-Zio'łkowskaet et al. [46] demonstrated that the reduction of defects of EBM sample have a positive effects on the mechanical properties and their fatigue as well as elongation value once the EBM sample were treated by hot isostatic pressing (HIP). This is confirmed by Ahlfors et al. [47] that defects were reduced after HIP, whereby the sample showed 100% densities. This is attributed to the HIP process that allows the reduction of gas porosity as lack of fusion of the sample. Similar to Popov et al. [48] study's, whereby they demonstrated that the HIP treatment is able to close the pores on the sample. The result of the finalized sample demonstrated that the sample was free of unmelted surfaces due to the intense self-diffusion of the material at very high temperature of HIP. Therefore, it is crucial for EBM sample to undergo HIP treatment to ensure that the microstructure is appropriate for clinical applications.

2.4. LENS[™] technique

LENS is method of 3D components fabrications from CAD file like



Fig. 3. SEM image Grain refinement of Ti6Al4V fabricated by DED by high-intensity ultrasound. (a) Sample without ultrasound, (b) sample with ultrasound (c) columnar grains formed on surface (d) fine equiaxed grains (e) Histogram of prior- β grains (e) grain size and (f) aspect ratio [39].

other AM techniques. As shown in Fig. 1(d), LENS process is a system, whereby metal powder is injected directly into the melt pool of continuous wave. LENS uses argon gas in maintaining the moisture level to be < 10 parts per million in preventing the oxidation of molten metals [49]. For this reason, powder with specific high reactivity can only be used in the production process. Generally, the LENS operates based on the selective deposition of metallic or ceramic powders and are melted with 500 W high-power fiber laser simultaneously. In comparison to other AM techniques, LENS uses powder with high purity as well as homogenous in the shape of spherical [35]. Mitra et al. [50] fabricated dense and porous Ta-Ti alloy composition with nanotube surface modification to enhance the Ti biocompatibility. Study demonstrated the increasing addition of Ta fabricated by LENS result into microstructural variations due to presence of α -Ti and martensite alpha' which result to higher hardness. Under in vitro, the percentage of Ta of 10% and 25% reveled higher osteoblast formed closely similar to the nanotube modification where there was more bone tissue attached with no gaps on the sample. Narayanan et al. [51] characterized the corrosion behavior of biomedical grade Co-Cr-Mo by using LENS technique, whereby they tested it under the Hank solution. The result demonstrated that the LENS sample surface showed micro-galvanic cells as well as a new crack feature which can act as active site for the localized dissolution of the alloy. Attributed to this, more studies are needed on the corrosion behavior of the LENS products to adapt the microenvironment of anatomical body fluid.

2.5. Selective laser melting (SLM)

SLM is one of the most popular powder bed AM techniques for creating structural parts without any geometry restrictions. "Selective" is defined as the processing of single or partial powder, while "laser" refers to the laser that is used for processing, and "melting" means the melting of the designated materials. SLM has a benefit of short production time with minimal processing steps than other AM methods [45]. As shown in Fig. 1(e), SLM method involves the use of metallic powder that are uniform on the build platform. Then, the powder is selectively melted by the energy source. The main advantage of SLM is that it can fabricate a wide range of metallic materials, including clinical grade of stainless steel (SS), cobalt-chromium (Co–Cr), titanium (Ti), and their alloys [15]. This is attributed to the complex shapes of anatomical structures, such as the knee, hip, skull as well as dental that requires precise design and accuracy to match the anatomical structure.

SLM has the capabilities of creating fine porous structures making it preferred for implant usage. These structures improve the cell adhesion between the human body and the implant part. Kelly et al. [52] adopted the sheet-based microstructure in fabricating SLM implant with a high permeability as well as increased the compressive and tensile fatigue behavior of the implant. A similar study by Xie et al. [53] fabricated scaffold with thicker-walled gyroid which result to excellent tensile strength. The sample is in the microarchitecture similar to the trabecular bone. Research on the metallic implant of SLM has acquire good attentions. It has showed evident from a previous study of developed interbody fusing cage of Ti6Al4V with compressive modulus of 4 GPa, which almost similar to the compressive modulus of trabecular and compact bone [45,54,55]. Yin et al. [56] customized Mg-based composite for enhancement of corrosion as well as cytocompatibility. They found that the SLM Mg composited with KZ30 and bioactive glass of 45S5 showed cell viability under in vitro test as well as cytocompatibility. Li et al. [57] found that the SLM sample of β Ti35Zr28Nb alloy exhibited elastic modulus of ~ 1 GPa which is in the range of trabecular bone. After 4 weeks of in vitro with osteoblast cell, the cell formed informing on the surface. This result conformed the integration between the cell and bone.

3. Advance metallic materials in SLM technology

Around 70%–80% of implants in the medical field are made of biometallic materials [58]. These biometallic materials are crucial for the reconstruction of failed hard tissue structure as well as to enhance the quality of life. The demands for these biometallic materials are increasing rapidly due to the increasing number of populations which includes high risk accident and elderly people who are prone to higher rich of anatomical failure. This biological and mechanical biocompatibility still needed many improvements for further use in biomedical applications. Commonly, the practical metallic clinical fields include stainless steel (SS), cobalt-based alloys, titanium (Ti) and its alloy as tabulated in Table 2. As for the SLM fabricated metallic alloy, the composition of the mostly used metallic material is as shown in Table 3.

3.1. Stainless steel

Stainless steel has been widely used in biomedical implants in fabricating stents, bone implant fixations and other devices attributed to its malleability, resistance to corrosion and fatigue. As compared to other biomedical implants, stainless steel comes in cheaper price, however, they are non-MRI compatible. Stainless steel generally exhibits a higher ductility and cyclic twist strength as compared to Co-Cr and Ti alloys. Stiffness is greatest for Co-Cr alloys, while it is the lowest for Ti alloys [45]. A previous study on their ability for fabricating biomedical implant has gain attentions, especially for the SLM process. Dwivedi et al. [72] fabricated the 316L and they found that the wetting and water contact angle of the sample were in the range of 60°-90°. This confirmed that the SLM 316 L has good permeability, allowing diffusion for the implant once implanted in the human body. Tekder et al. [73] determined the ability of 316L in the corrosion behavior. The samples were oxidized with ceramic-based of $\mathrm{TiO}_{2\!.}$ The corrosion was tested on the SBF solution and the result obtained were found that the plasma oxidized samples had higher corrosion resistance than the untreated 316L samples. The highest corrosion was found when the temperatures were at average of 700 °C.

3.2. Cobalt-base alloy

Cobalt-based alloy is a nonmagnetic and heat resistant alloy which is widely used in wear and tear-related application inherent to their high strength and corrosion resistance. Their corrosion resistance in generally adaptable to different environments. However, due to their different alloys in cobalt-based, the corrosion is dependent on the concentration, temperature as well as the acidic condition of the media. Cobalt-based alloys are originated from the solid-solution effect of chromium and molybdenum. Nevertheless, they are very hard to be fabricated, making them very limited to be used in the medical and industrial field due to its excellent resistant to degradation. The surface degradation and corrosion is mainly cause by the H₂O₂ present on the surface and as the concentration increases below the corrosion potential shifts, it will form the passive film. Hu et al. [74] fabricated CoCrMo with face-centered cubic (FCC) structure and they found that the alloy of the XY plane showed a lower degeneration compared to XZ plane due to the presence of precipitates consisting of grain boundaries. However, as more H₂O₂ were passed into the sodium chloride (NaCl) solution, the thickness of passive film increases, making them increase the chance of degradation and corrosion. Similarly, Gong et al. [75] studied the CoCrMo corrosion behavior in lactic acid concentration, whereby they found that the corrosion resistance of XZ plane is better than the XY plane. This is also stated by Kajima et al. [76] previously that the built XZ plane released more metal ions as compared to XY plane due to the large are of corroded pool boundaries. However, it is crucial for the CoCrMo alloy to undergo post-processing as stated by Axinte et al. [77] and Jangtap et al. [78] in order to increase the biocompatibility of SLM CoCrMo implant.

Table 2

Properties of material develop for orthopedic implant.

Material	Melting Point (°C)	Yield Strength (MPa)	Tensile Strength (MPa)	Elastic Modulus (GPa)	Vickers hardness (HV)	Corrosion Resistance	Ref
SS	1325-1500	≈ 250	450	200	275	High	[59]
Iron	1150-1600	130	350	200	30-80	Good	[60-62]
CoCr Alloys	≈ 1600	480–580	≈800	≥ 150	380-430	Excellent	[63,64]
Ni	≈1450	≈600	≈1000	≈ 150	300	Good	[65-67]
Alloys							
Ti	1670	≈750	≈900	120	350	Average	[68,69]
Alloys							
Mg	650	≈ 150	≈250	≈ 40	100	Poor	[69]
Ta	2980	≈ 230	≈ 260	≈ 185	≈ 1000	Good	
Zn	420	50-120	100-200	≈ 100	30	Good	[70]

Table 3

The chemical content of the SLM metallic materials [71].

Materials	Al	V	Fe	Cr	Ni	Мо	Mn	Nb	Si	Р	С	Н	S	Ν	0	Р	Ti
SS 316L ASTM A276			Bal	16.00	10.00	2.00	2.00		1.00	0.045	<mark>0</mark> .03		<mark>0</mark> .03				
Iron (Fe) ASTM A564			Bal	15.00	3.00	3.00	1.00		0.07	<mark>0</mark> .15	0.07		0 .02	<mark>0</mark> .10	<mark>0</mark> .04	<mark>0.04</mark>	
Ni ASTM B446			5.00	20.00	Bal	8.00		3.15	0.50	0.015	<mark>0.1</mark> 0		0 .02			0.02	
Ti6Al4V ASTM B348	5.50	3.50	0.25								0.08	0.013		0.03	0.13		Bal

3.3. Titanium and titanium-base alloys

Ti alloys are currently an advancing material due to their predominant properties, exceptional mechanical characteristics, a high specific strength, outstanding corrosion resistance and good biocompatibility. Ti6Al4V is a desirable material for biomedical implants due of its distinct set of characteristics as it extensively used primarily in the aerospace and biomedical industries. Despite their high demand, Ti6Al4V is still expensive, large material waste as well as long lead time. Under these cases, AM offers the capability in the fabrication of Ti6Al4V with geometric complexities to be used in the medical field. Among the metallic biomaterials, Ti exhibits the highest biocompatibility, corrosion resistance and tensile strength as compared to stainless steel and cobalt chromium. Brandl et al. [79] stated previously that the wires deposited with Ti6Al4V achieved similar mechanical properties of wrought Ti6Al4V of AMS 4928. Suresh et al. [80] found that the elastic modulus of Ti-6Al-4V ELI is able to be reduced attributed to the porosity of \sim 15% and roughness. The cytocompatibility in the MC3T3-E1 showed no adherent toxicity and after 1 week of in vitro test, the proliferation rate was increased. Ji et al. [80] fabricated Ti-3Cu to determine the corrosion resistance and biocompatibility. The in vitro test in osteosarcoma cell MG63 showed good cell attachment. The antibacterial resistance under E.coli showed a stable highly-effective antibacterial rate with no cytotoxicity present.

3.4. Shape memory alloys - Ni-Ti alloy

Ni–Ti or also known as Nitinol is a type of alloy with composition combining of half nickel and Ti with composition of \sim 50% and they belong in shape memory alloys class. Attributed to their shape memory alloy, they are easily deformed at the low temperature as well as be able to retain to their initial shape even when exposed to high temperatures. This by the effect of its martensitic transformation on the surface of NiTi when expose at high temperatures. The Ni–Ti alloy is packed in an order of BCC structure which includes austenitic phase making them unable for twisting and bending. In biomedical field, Ni–Ti is commonly used in stent which can exhibit changes in diameter when inserted in human body, due to its austenitic phase at elevated temperature. Polozov et al. [81] fabricated NiTiNb shape alloy with a density of 99%. The yield strength obtained was up to 450 MPa with tensile strength of 706 MPa. The sample showed hysteresis of martensitic transformation once the temperature increased, demonstrating good mechanical properties. This study is similar to Yu et al. [82], whereby they found that SLM-NiTi showed ultra-high failure strength of up to 700 MPa and elongation of 10% under tensile conditions at room temperature.

3.5. Biodegradable metals – Mg based alloy

Magnesium (Mg) has gained attention in the medical field due to its appropriate mechanical properties, biocompatibility as well as lightweight. Mg is the highest positive charge in human body called cations, whereby mostly are stored in bone with composition up to 65% and soft tissues of 40% and mainly in muscle of 25% [83]. Mg plays important roles in synthetizing of protein from the mitochondria, known as the powerhouse of human body. The modulus of Mg \sim 40 GPa which is close to the cancellous bone modulus of 30 GPa [84,85]. These suitable properties make Mg a good choice for medical devices which preferred high compatibility of materials [86,87]. However, Mg exhibits the lowest corrosion resistance which attributed to the degradation of the implant after fixations, resulting to the premature loss and failed of mechanical properties [83]. The microenvironment of the human body fluid is between pH 7.4 and pH 7.6 that induce the local accumulation of ions, causing cytotoxic effect [88]. These effects are mainly attributed to the releasing of H₂ gases during degradation of Mg implant [89]. These issues are mainly the most primary issues for the limitation usage of Mg implants in clinical applications. Some study has focused on Mg doped with of rare elements that shows good resistance to corrosion [90]. Mg-Zn composites are much suitable for the biomedical field since Zn promotes bone mineralization as shown in a study by Shuai et al. [91], whereby they incorporated hydroxyapatite (HA) into Mg-Zn in order to slow down the degradation process. A good apatite layer was formed on the surface, encouraging good corrosion resistance. However, there was excessive HA content on the surface of the implant, leading to the pore formation. These pores eventually accelerated the bio-degradation. Nevertheless, SLM of Mg alloy is still it is challenge phase, largely owing to its high chemical activity and flammability even in bulk state [92].

3.6. Biodegradable metallic materials – zinc

Zinc (Zn) is a bioresorbable metal mostly designed to provide mechanical support for injured tissue [93]. Compared to Mg and Fe, Zn exhibits more biodegradation behavior to match the anatomical requirement. Zn is the second highest trace mineral in the human body which acts to maintain cardiomyopathy cardiac function and for the formation of new bone [94]. The lack of Zn in the human body is linked to the weaknesses and poor health of bone [95]. This allows zinc to be used as a new generation of orthopedic implant. However, zinc has a poorer mechanical strength and encourages localized deterioration which can lead to toxicity and implant failure. Thus, various studies have been done by alloying Zn with other elements. For example, Yang et al. [96] fused Zn element with Mg to improve the degradation and bone fracture healing. The study demonstrated that the Zn/Mg alloy showed ultimate tensile strength up to 650 ± 13 MPa, and in the *in vitro* study, the implant reduced its degradation behavior. Guillory et al. [97] evaluated the effect of Al element doped in the Zn on the increase activity of macrophages. Corrosion resistance of the Zn/Al implant after 20 months of in vivo in rat aorta and found that the degradation rate decreased with higher macrophages formed to engulf cells. A similar study was also reported by Drelish et al. [98] where they suggested that the pure Zn wire exhibits steady corrosion without any cytotoxicity. The bone remodeling rate increased, along with a decrease in inflammation in vivo. However, a buildup of passive corrosion on the implant was observed after 10 months of post-implantation. Recent study by Su et al. [99] found a uniform layer between the Zn and phosphate implant and the lumen tissue after 4 weeks on implantation in vivo of rabbit aorta as shown in Fig. 4. Connective tissues were observed between the material and host on the sample, demonstrating excellent biocompatibility to Zn implant to be used in clinical applications. However, the feasibility of SLM Zn in the medical field needs to be addressed before it is permanently used in anatomical structures to prevent its degradation.

3.7. A bioactive metal - Tantalum

Tantalum (Ta) is typically employed as an alloying element in Ti alloys to act as a beta-phase stabilizer, lowering the elastic modulus and preventing stress shielding effects. However, there has limited of study focusing on Ta as an alloying element due to its relatively expensive, high density and processing temperature (melting point of 3017 °C). Ti–Ta exhibit lower elastic modulus compared to the Ti6Al4V and easily detrimental to the surrounding bone tissue for long term implant applications. However, Brodie et al. [100] reported that Ti–25Ta and Ti–65Ta had the highest biocompatibility with bone, attributed to their lower elastic modulus (60–70 GPa) than conventional Ti6AL4V. Also, they demonstrated that the Ti–25Ta had the similar yield strength as the SLM Cp-Ti with a significant reduced in elastic modulus of 65 ± 5 GPa. There was a decrease in surface defects with randomly oriented α' on the surface [101]. In addition, combination Ti with 25% of Ta, Brodie et al. [100] showed mineralization in an *in vitro* study with increased alkaline

phosphatase activity. However, the lower Ta content exhibits an unpromising composition for an AM bone implant.

3.8. High entropy alloy – aluminum

Aluminum (Al) alloys have the merits of high specific strength, good casting capabilities, and low cost. However, their poor wear resistance, corrosion resistance, and incompatibility have hampered their application to be used in biomedical implants [102]. The high level of Al and their hypersensitivity may lead to a complicated health issue. Previously, Bowman et al. [103] used Al silicate to control bleeding in open reduction internal fixation surgery and found the Al was toxic at a minor level to the endothelial cell. Apart from the poor anti-corrosion and wide intrinsic friction coefficient, Al alloy can easily leach during a prolonged period. Thus, Al has barely been used in medical applications. Weng el al [104]. fabricated Al_2O_3 to enhance the biological response and antibacterial performance of Al. The antibacterial ratings were up to 99% for *S. aureus* with improved mechanical properties. However, as the duration of the immersion increases, the Al oxide is leaching out of the Al ions, increasing the sample corrosion rate.

4. Requirement for SLM printing in biomedical applications

4.1. Printability and structural performance

The implant that is fabricated using the SLM technique could vary in its microstructure, surface finish, and mechanical properties from the conventional techniques like casting and forging. The most noticeable features of these implants include the build direction of the morphology, embedded flaws, and high residual stress. It happens due to the nonuniform solidification in the melt pool [105], and the thermal stress distribution during the SLM process [106]. The rapid solidification (cooling rate at 105–106 K·s–1) can form a small grain size on the surface, which can change the functional properties and increase the requirement for the implant to undergo post-processing.

Despite biomedical implants attention to adopting SLM, it is crucial to overcome several problems, notably the formation of in-part defects [107]. For example, the lack of fusion can form voids due to the insufficient energy input during the fabrication process, whereby the prescribed volume of particle failed to fully melt [108]. Also, these inadequate amounts of energy tend to adhere irregular granules, which enclose the partially melted particles on the SLM sample. A study by Liverani et al. [109] fabricated Ti–Nb sample with melt pool of 2 µm, however there were defects formed on the surface caused by gas entrapment and void during the layer by layer process. Fig. 3 illustrates the schematic diagram of melt pool formation in the SLM process. As can be seen from Fig. 4, the adhered ((Fig. 5(a)) and partially melted particle (Fig. 5(b)) influenced the final surface roughness. Fig. 5(a) depicts a powder particle dispersed from a laser beam with surface damages generated by the staircase effect owing to layer-by-layer processing. The defect includes the balling, spatters or inadequate fusion that led to



Fig. 4. Cross sectional of backscattered electron image of Zn implant after in vivo implantation in aorta of a rat after 18 weeks [99].



Fig. 5. Schematic diagram of (a) surface defects during layer by layer scanning [111], (b) SEM image of cross section of AM Sample [112].

irregular surface morphology due to high heat-shock. A high surface roughness can deteriorate the performance of parts to be used in biomedical implants. Fig. 5(b) shows the surface defect containing of open pores, which consist of partially melted and adhered powder particles. The implant surface characteristic plays an important role in the osseointegration which the cell attachment relies on the microscopic, mesoscopic and the nano structures of the implants [110]. However, when the energy input is high, the metal vaporizes and eventually creates gas bubbles that trap inside the sample. This trapped gas causes a rounded-shaped pore to form on the fabricated sample.

Attributed to these, SLM has made a significant advancement in the optimization of the processing parameters. The process parameters known as "energy density" Ev, can be defined as laser power (P), scanning at velocity (v), hatch spacing (h) and layer thickness (l) based on Equation (1).

$$Ev = \frac{P}{v.h.l} \tag{1}$$

Several studies have highlighted the correlation between input process parameters and mechanical properties of SLM components. Shipley et al. [113] studied the effect of process parameters on mechanical properties and found a tolerable amount of compressive strength and acceptable microstructural composition. It also demonstrated that the energy density plays a role in having a higher relative density of the SLM Ti6Al4V. Taban et al. [114] reported that the higher energy density increases the density of the sample Ti–Nb and as the density lowered, there were balling, porosity, and cracking defects. However, Dhansay et al. [115] suggested that a high level of energy input needs to be maintained during the process to prevent unwanted porosity and residual stress. They also stated that the scan direction should be in the range of 90° to reduce the presence of martensite on the surface. Similarly Yadroitsev et al. [116] stated that the slope angle alignment should be generated parallel to scanning directions in obtaining SLM samples with less than 1% of void. Moreover, King et al. [117] utilized the processing parameters to prevent the formation of large pores on the surface of the sample. They found that the laser power has a greater effect on the development of voids, and with increasing laser power, the crater formed was more prominent. For example, by using one factor at a time (OFAT), Kasperovich et al. [118] suggested the correlation between input parameters and the void as a fracture domain of SLM Ti6Al4V. As shown in Fig. 6, they mentioned that the scanning speed is the most dominate factor (Fig. 6(a)), followed by laser power (Fig. 6(b)) and the most least sensitive was hatch distance (Fig. 6(c)).

Also, the Ev is used to control the product void. The increase in void is produced by the high energy density, which occurs in SLM processing at high power and low velocity [119-121]. As tabulated in Table 4, cumulatively it can be understood as intermediate energy densities which are shown to have void formation. In exploring this outcome, Xu et al. [122] performed the SLM process by changing the layer thickness while keeping the other parameters intact. It was determined that the optimum layer thickness of 30 µm contributed the finest columnar with alpha martensite microstructure in comparison to values of 60 µm and 90 μ m. The martensitic structure could lower the elongation, while α - β martensite eventually exhibits higher yield strength and improves tensile elongation to failure. However, a large powder particle of the martensitic structure could have the potential to produce large voids in the build sample. Salem et al. [123] fabricated lattice structure with various processing parameters and suggested that the laser power of 100 W with a scanning speed of 1600 mm/s reduced the void in the strut. Wen et al. [124] clarified processing parameters' impact on densification performance during the SLM technique. When power is set at 60 W, a lot of uneven pores are formed, which are the result of the unmelted material by the laser. On the other hand, Li et al. [125] came to the conclusion that a stronger laser power creates thermal strains in



Fig. 6. The effect of energy density on void following varies of (a) scanning speed (b) power and (c) hatch distance [118].

Table 4
The optimization of process parameter of SLM metallic sample.

Material	Power (P)	Scanning Speed (mm/s)	Layer thickness (um)	Hatch distance (um)	Remark	Reference
Ti6Al4V	200	1250	30	0.80	Surface roughness of SLM were lowered than the EBM caused by the initiation of fatigue crack	[126]
Ti6Al4V	280	1200	84	0.14	High laser power result to the increase in the porosity	[108]
Ti6Al4V	200	1100	30	0.18	Voids formed on the surface caused by the process overheating and insufficient fusion	[118]
316L	100	4000	50	0.13	Void formed from the vapor cavity caused by the evaporation of the 316L	[117]
316L	100	250	50	0.114	Sample of higher scanning speed demonstrated the highest densification and mechanical properties	[114]
316L	150	700	50	0.50	Laser power influenced the density, higher laser power yield higher relative density	[109]
Ti6Al4V	75	1000	250	0.77	Microcracks formed on the surface caused by the melted powder particle	[111]
Ti6Al4V	400	4000	30	-	400 W power showed the lowest internal porosity and defects on the structs design	[123]
Ti6Al4V	200	1000	50	0.12	The residual stress profile were affected by the process parameters	[121]

the printed material, which causes to crack since there is insufficient power to melt the powder.

The mechanical characteristics, such as elastic modulus, elongation, and tensile strength are also influenced by those defects [126–129]. The existence of defects that are perpendicular to the loading direction results in the lowering of the mechanical properties of fabricated products. However, most metal alloys in biomedical technology offer a high level of strength as compared to human bone. It tends to introduce a stress-shielding effect when shear stresses arise due to mechanical properties mismatch between the bone and the implant. The decrease in

bone density occurs as a consequence of an implant eliminating typical stress from the bone. In order to overcome this issue, many researchers have introduced porous structures or specially designed lattice scaffold to reduce the structural integrity of orthopedic implants. For example, Wang et al. [37] fabricated a medical implant using 316L stainless steel based on cubic, octet, and triply periodic minimal surface (TPMS) to determine their cytotoxicity via the SLM method. It was found that the pore sizes of these lattice struts ranged from 200 μ m to 500 μ m, resulting in Young's modulus comparable to anatomical bones. The cytotoxicity test showed a 30% cell viability enhancement on the surface with no

toxicity effect. Moreover, Wang et al. [38] mentioned that by modifying the SLM processing parameters, sintering temperature (1200 °C), and structural porosity ((between 3535 μ m and 160,160 μ m). The mechanical characteristics of 316L stainless steel were equivalent to cancellous bone. Also, Wally et al. [56] and Ran et al. [44] discovered that at a pore size of around 600 μ m, the compression stiffness of SLM Ti6Al4V equivalent to bone and outperformed in osseointegration.

4.2. Corrosion resistance

The excessive corrosion of the metallic implant such as Ti6Al4V when exposed to the medium or any environment forms passive film consisting of TiO₂, Al₂O₃, and V₂O₅ [130]. These films degrade the surface due to the abrasion and oxidation formed from the relative motions of the implant after implantation. Furthermore, vanadium, which exists in both elemental form and as oxides (V₂O₅), is hazardous. Implants based on the Ti6Al4V alloy are reported to exhibit a decline in fatigue strength as a result of surface corrosion pit formation to the dissolution of Ti²⁺ ions in body fluid. The toxic release of aluminium(Al) and vanadium ions can result in chronic disease including osteomalacia [131]. Whereas, Al ions on the surface of the implants during wear and tear might cause neurological disorders and Alzheimer's disease [132, 133]. In SLM, the rapid solidification and high rate of cooling during the

Table 5

The corrosion properties of the SLM metallic implants.

fusion powder bed process result in α' -martensite and fine grain development in the microstructure. This is distinct from traditional processing methods like forging and machining. The changes may have an impact on the corrosion resistance on end-product. Table 5 summarizes the investigations of the corrosion properties of SLM in metallic implants. For instance, Dai et al. [134] mentioned that SLM Ti6Al4V had a lower corrosion resistance than the conventional grade 5 alloys because it had more acicular α' phases and fewer β phases resultant in high cooling rate during the annealing processes in the melt pool bath. The α' martensite phase easily cracks in a sufficient amount of oxygen and lowering the hardness of the material. Guo et al. [135] found that by mixing of Cu in a base metal of Ti6Al4V powder, it could improve structural porosity of the product. Beside doubling the frequency of β phases in the microstructure, the inclusion of Cu itself could strengthen the corrosion behavior of the structure [136]. Chi et al. [137] demonstrated that the corrosion rate also increased with the increasing of surface roughness. The higher surface roughness of the sintered powder can affect the oxidation kinetics, causing the sample to be easily oxidized when exposed to the simulated body fluid (SBF) [138,139]. The irregulating surface, consisting of pit and valley initiation sites, favors trapping a liquid and drastically dropping the pH value. It leads to the formation of localized corrosion by destabilizing the protective layer [140]. Chiu et al. [141] confirmed that the corrosion can accelerate in SLM Ti6Al4V

Material	Assessment	Surface oxide/ precipitate/ ion	Remark	Ref	Year
Al-12Si	Effects of laser power on the corrosion resistance	He, Ar, N ₂	Pore clusters formed in the microstructure leading to corrosion	Wang et al. [144]	2014
CoCrMo	Electrochemical techniques and chemical analyses	Cr, O	Martensitic ε phase (hcp) formed on the surface increased the corrosion resistance	Hedberg et al. [145]	2014
Ti6Al4V	Corrosion resistance is determined in NaCl solution	TiO ₂	Corrosion resistance of the SLM caused by the high amount of acicular alpha' and low beta-Ti phase	Dai et al. [134]	2016
316 L stainless steel	Grain boundaries and correlation to corrosion	MnS, oxide film	The sample sowed relatively low value of corrosion current density due to the increased of Cr and Mo content and a decrease of Ni in the grain boundaries.	Zietala et al. [68]	2016
AlSiMg	Volta potential analysis	Al oxide films Al ₂ P	Larger and coarser microstructure regions showed significantly better potential difference between the phases.	Revilla et al. [146]	2016
Ti6Al4V-xCu	Effects of Cu content on phase constitution and corrosion resistance	Ti ₂ Cu	Trapped gas porosity exhibits a limited effect on the corrosion resistance.	Guo et al. [135]	2017
316 L stainless steel	Role of MnS in corrosion	MnS, TiN and Al2O3	Superior corrosion resistance relative to wrought form	Chao et al. [147]	2017
Ti6Al4V	Effect of heat treatment on the SLM sample	TiO ₂	Both horizontal and vertical specimens that have been heat- treated exhibit improved corrosion resistance	Chandramohan et al [142]	2017
316L Staipless	Different laser scan speed	MnS	The corrosion characteristics of the SLM printed 316L were also revealing higher pitting contentials compared with wrought 316L	Sander et al. [148]	2017
Ti6Al4V	Corrosive environment of Ringer's simulated body solution	VO^{2+} Ti ³⁺ OH ⁻ Cl ⁻	Corrosion were mainly caused by the formation of passive film during wear and tear	Chiu et al. [141]	2018
316L Stainless Steel	NaCl solution and SBF based on the protein albumin.	MnS	316L Stainless Steel showed higher corrosion compared to the wrought materials attributed to higher surface area covered by voids caused by the AM process	Stendhal et al. [149]	2018
CoCr	Effect of different laser power on corrosion resistance, and crystalline structure	Passive film, Cr ₂ O ₃ ,CrO Cr ³⁺ , Cr ²⁺	High laser powers facilitate the formation of complex oxide films, the fast cooling rate which result in the crystalline phase transformation from x to x .	Wang et al. [150]	2018
316L stainless steel	Effects of Phosphate buffered saline on corrosion resistance	Cr oxide film, Cl ⁻ , OH ⁻	SLM sample had a more passive region and higher charge transfer resistance (Rt) SLM 316L SS outperformed the conventional wrought 316L SS	Al-Mamun et al. [151]	2018
AlSi10Mg	Effect of HIP on the corrosion of SLM	Si, Cl ⁻	Untreated specimens showed superficial corrosion with	Rubben et al. [152]	2019
AlSi10Mg	Microstructure of the melt pools of	Si precipitate,	Grain size, Al-dendrite, and coarser eutectic-Si network developed melt pool caused by slower solidification rate	Fathi et al. [153]	2019
Ti6Al4V	The effect of 12 wt% HCl polish on	Oxide film, TiO ₂	Polishing the surface of the Ti–6Al–4 V alloy enhance the corrosion resistance attributed to the decrease in roughness	Chi et al. [137]	2020
7075 Al	Effect of heat treatment on the corrosion of SLM sample	Al ₂ Cu precipitate	Si phases and θ phases precipitated at the grain boundary has a significant effect to improve the hardness and corrosion properties	Liu et al. [154]	2020

Abbreviation: He-helium, Ar-argon, N2-nitrogen, Cr-chromium, O-oxide, TiO₂-titanium dioxide, MnS- manganese(II) sulfide, TiN- titanium nitride, Al_2O_3 - Aluminum oxide VO^{2+} - oxovanadium(IV) cation, OH^- -Hydroxide, Cl^- - chlorine ion Cr_2O_3 -chromium(III) oxide, γ -gamma, ε -dielectric.

samples, due to the mechanical attrition following the tribo-corrosion assessment in in vitro of simulated body fluid and chloride solutions. As a result, the rougher surface has the potential to severely increase the kinetics of corrosion and form cracks on the oxide layer that forms to protect titanium alloys. Furthermore, Chandramohan et al. [142] mentioned that the corrosion of the SLM fabricated Ti6Al4V sample was superior in the media solution containing chloride ion due to the less formation of vanadium oxide (V₂O₅) on the SLM sample. The insufficient resistance to corrosion of SLM Ti6Al4V was obtained when passive films of TiO₂, Al₂O₃, and V₂O₅ were formed as an exposure effect to a corrosive environment. These films degrade the surface, resultant of oxidation formed from the relative motions of the implant after fixation. However, the oxidation of these elements vary according to the phase-type of α elements (Al) and β elements (V) that diffuse into α and β phases, respectively, which causes an electrochemical difference in the sample. In conclusion, a higher $\alpha + \beta$ morphology can increase interfacial boundary regions between those phases and finally trigger the sample to corrode. A recent study by Ciliveri et al. [143] demonstrated that the Ti3Al2V produced by the reduction of Al and V in Ti6Al4V showed a reduction in compressive strength of up to 18%. This is attributed to the Al and V elements that formed the β -Ti phase in the Ti6Al4V sample.

One most employed biomaterial is cobalt-chromium(CoCr) alloy, consists of nickel (Ni) and molybdenum (Mo) [155]. The corrosion of CoCr are more prominent in the wet and high ions of the human body where the CoCr releases toxins into the body, forming of cancerous tumors [156]. The corrosion behavior of dental alloys mostly depends on numerous factors, including composition, pH level, post-processing treatment, and clinical circumstances simulation [157]. Like CoCrMo, the main constituent phases consist of γ -austenite and brittle ε -martensite, which are stable at both low and high temperatures; however, the stress interfacial area between these phases can facilitate crack initiation and corrosion [158,159]. Previously, Guoqing et al. [145] studied the corrosion behavior of SLM CoCrMo in PBS and they found that the metal ion released was lower than the cast sample regardless the duration of immersion time. The enhanced resistance to corrosion of SLM CoCrMo mainly caused by the presence of Mo's segregation to the fine cellular boundaries of mainly micron size carbon and the ε -martensite phase. Wang et al. [150] observed that SLM CoCr sample had a superior corrosion resistance after polishing the surface. This is due to the oxides on the inner and outer layer that play essential role in influencing the corrosion resistance.

The corrosion studies of SLM Al alloys (Al-Si alloys) mainly rely on features of surface texture, layer microstructure, residual stress and electrolyte concentrations [154,160,161]. The surface of SLM Al has various grain boundaries and passive film which increase the corrosion [162]. It is self-stable and has good corrosion resistance with neutral solutions; however, when exposed to open air containing chloride ions (pH < 7.0) of neutral body pH, the pitting polarization increased, making it susceptible to corrosion [163]. In general, SLM Al is better corrosion-resistant than the casting products. Several studies have focused on segregating the SLM Al via a coated sample with Mg, Si, and Fe, which are essential to cellular life and improve corrosion resistance. For example, Revilla et al. [146] coated the SLM Al and pure Al samples with silicon-magnesium and submerged them in 0.1 M of sodium chloride (NaCl) to assess their corrosion resistance. The SLM Al demonstrated less Si-Mg segregation than the pure Al counterpart. Similarly, Leon et al. [161] noticed a lower Fe–Mn segregation in the SLM sample; however, several local galvanic cell formations on the pure Al were obtained due to high corrosion of Al, inappropriate to be used in clinical application. According to Rubben et al. [152], the coarser grain that formed in SLM Al-Si sample offered superior corrosion resistance than the cast sample. Fathi et al. [153] evaluated the effect of laser parameters influence on the corrosion resistance of SLM AlSi10Mg by using a 3.5 wt% NaCl immersion similar to bodily fluid. They found that the increased laser energy density reduced the surface roughness by forming cyclic small-to-large melt-pool structures, as shown in Fig. 7. SLM



Fig. 7. Cyclic melt-pool structure[153].

AlSi10Mg corrosion was dependent on the rate of melt-pool solidification; surfaces with a higher solidification rate contain finer Si precipitates and Al dendrites, which reduces the occurrence of pitting corrosion. A recent study by Liu et al. [154] found a small amount precipitated at the grain boundary and these boundaries significantly improved the hardness and corrosion properties of SLM Al alloys, allowing to be used in biomedical implants.

SS application in biomedical industries has well address. The release of metallic ions of the SS can fasten the corrosion which may cause inflammatory reactions, severe ache and the repercussions of surgery [164]. Following SLM fabrication, austenitic stainless steels like 304L and 316L show an austenitic phase [165]. The high corrosion on SLM 316L was mainly cause by the absence of manganese sulfide (MnS) and the formation of carbon grain structure. Stainless steel is vulnerable to localized corrosion caused by decreased sulfur compounds and chloride ions [166]. Sander et al. [148] stated this pitting is due to the annihilation of MnS. MnS oxidized during pit initiation attracts the concentrated Cl ions. This ion can hasten pit growth and inhibit surface layer repassivation. Zietala et al. [68] and Chao et al. [147] stated that the presence of chloride and absence of MnS during the corrosion evaluation of the 316L sample in electrolyte causes the lower passive current density and higher pitting leading to the corrosion. Chao et al. [147] observed oxide/hydroxide layers in the passive film with higher charge transfer resistance in phosphate buffered saline causes by the absence of MnS. The passive film formation as reported by Al-Mamun et al. [151] found that higher corrosion potential (E_{corr}) breakdown potential in SLM SS 316L were three times higher than their wrought counterparts. A recent study by Karimi et al. [167] displayed that the SLM SS 316L can obtained superior corrosion resistance by tailoring the crystallographic texture at low energy density. This is supported by the study by Stendhal et al. [149] and Wang et al. [144] that the superior corrosion resistance in SLM SS 316L were attributed to the refined grains formed on the surface.

4.3. Osseointegration of SLM metallic implants

SLM has attracted various medical fields for the manufacturing of metallic structures as tabulated in Table 6. In bone implant, SLM has focused on 316L stainless steel [168–170], Ti6Al4V [171–179], Ti35Zr28Nb [57], Fe₈₁Ga₁₉ [180], Pure Zn [124], NiTi [181] and Co–Cr–Mo [182]. As for dental implant, AM focused on Ti6Al4V–5Cu [183], and Co–Cr [184,185].

Orthopedic implants are used for fixation, providing mechanical support and biological integration for the human body once osseointegration is achieved. For instance, stainless steel 316L implants are the most often utilized in the orthopedic industry for treating bone fractures and ruptures due of its affordability and minimal weight to form bone to implant contact. Wang et al. [168] studied the octet gyroid lattice

Table 6

Common techniques to enhance	the biological	l properties of t	the SLM metallic	implant from	earlier work.
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Material	Implant	Assessment	Yield Strength (Mpa)	In vitro/In Vivo	Remark	Ref
316L stainless	Bone implant	Porous structure	520.8	SBF	Porosity 160–35 µm and 58–28% match with those of cancellous hone	Xie et al. [169]
Co–Cr	Dental implant	Porous structure	300.5	X-ray testing	Presence of porosity on the sandblasted sample, demonstrating biocompatibility	Al Jabbari et al.
CoCrMo	End prosthetic ankle device	Process optimization	874.0	Stereophotogrammetric system	Custom-fit articular surfaces and able to contract as natural joint motion	Liverani et al. [182]
Ti6Al4V	Bone Implant	Compatibility	47.6	SBF	Apatite formed uniformly on surface after 14 days of immersion	Yan et al. [177]
Ti6Al4V	Bone implant	Porosity	252.1	<i>in vitro</i> cell and white rabbits	600 μm porosity showed highest bone to implant contact compared to 500 μm and 700 μm	Ran et al. [175]
316L SS	Bone implant	Compatibility	290.0	SBF	Poor performance and unstable during long- term experiments in simulated body fluid.	Kong et al. [170]
CoCr	Implant	Bone-replicating CoCr porous scaffolds	252.1	MG63	15–100 μm of irregularities and macro alterations were observed on the porous scaffolds	Caravaggi et al. [186]
Ti6Al4V	THR	Modulus of elasticity	122.8		The coated sample showed higher modulus than the cortical bone	Peng et al. [178]
Ti6Al4V	Dental	Porosity	204.0	MLO-A5 cell seeding	Pore size between 250 and 650 µm are favorable for cell attachment after 28 days.	Wally et al. [187]
Ti35Zr28Nb	Bone Implant	Scaffold porosity	200.0	SAOS-2	MTS assay revealed that SLM implant the scaffolds had cell viability Osteoblast adhered on the implant after 4 weeks of <i>in vitro</i> .	Li et al. [57]
Ti6Al4V and Ta	Bone Implant	Porosity	-	Rat distal femur	After 12 weeks in in vivo, there were higher osteoid present on the implant	Bandyopadhyay et al [179]
Ti6Al4V-5Cu	Dental implant	Microstructure and composition	-	Simulated saliva	The specific wear rate of Ti6Al4V–5Cu alloy was about 6.55 times better than that of the TC4	Zong et al. [183]
Ti6Al4V	Implant	Post processing	760.0	SAOS-2	Acid etched sample showed higher biological performance	Jamshidi et al. [174]
NiTi	Bone Implant	Modulus	170.5	SBF	NiTi structures etched with H ₂ O, HNO ₃ and HF exhibit elastic moduli that match that of bone tissue	Bartolemeu et al. [181]
Ti6Al4V	Bone Implant	Compatibility	-	MC3T3-E1	HA coated surface sample showed an increased in the metabolic activity on the	Fazel et al. [176]
Ti6AI4V	Bone Implant	Coating with HA and cell proliferation	200.0	rBMSCs	HA formed on surface, increase the proliferation and enhances cell differentiation	Pei et al. [173]
Co-Cr-Mo	Implant	Coating	180.0	In vitro femur Wistar rat	Enhanced osseointegration after 12 week of implantation	Latecola et al. [188]
Ti6Al4V	Bone Implant	Chemical Treatment	147.0	SBF	Apatite formed uniformly under SEM after 7 days of SBE immersion	Aufa et al. [171]
Ni–Ti	Implant	in vivo Compatibility	220.0	Pig	After 16 weeks <i>in vivo</i> , more bone to implant	Naujokat et al.
Co29Cr9W3Cu	Joint implant	Porosity		Rat	Early bone ingrowth into the bone contacts	Lu et al. [190]

Abbreviation: SBF-simulated body fluid, MG63-male osteosarcoma cell, MLO-A5- murine osteoblast to osteocyte-like cell, SAOS-2- female osteosarcoma cell, HNO₃nitric acid, HF-hydrofluoric acid, MC3T3-E1-murine calvarial cell, rBMSCs-mesenchymal stem cells, SEM-scanning electron microscopy.

structure made of 316L stainless steel to determine the appropriate bone cell mitigation and ingrowth via in vitro assessment. They demonstrated that this structure offered the highest osseointegration with a good degree of osteoblasts stability attached to the struts. The cytotoxicity test showed 25% cell viability, and the sample was safe even at a high toxicity test concentration (900 µg/mL). Similarly, Xie et al. [169] revealed that the struts structural performance of SS 316L with 28%-58% porosity matches that of the cancellous bone with increased osteoclast in the MC3T3-E1 cell. Kong et al. [170] fabricated SS 316L by using SLM and evaluated apatite formation following simulated body fluid and MC3T3-E1 cells. The F-actin cells, the main components of the cytoskeleton, were heavily spread on the surface sample under the fluorescent test and proliferated into osteoblast cells to form tiny bones. Also, the presence of chloride (Cl), calcium (Ca), and sodium (Na) were obtained after 288 h in the SBF immersion, proven the main build-up of apatite on the sample.

Several studies were conducted to investigate the osseointegration of SLM Ti6Al4V. Ran et al. [175] evaluated the cell attachment on SLM Ti6Al4V, which consists of a variety of circular pores by using *in vivo* rabbit femurs. It was demonstrated that the optimum structure for bone

ingrowth was a scaffold with pore sizes of 600 µm. Peng et al. [178] fabricated SLM Ti6Al4V with porosity ranging between 60% and 65%, suggesting that increasing porosity increases the bone-to-implant contact. In contrast, the diamond-shaped unit cell from Ti6Al4V materials was fabricated by Wally et al. [187] by using SLM. The compression stiffness of this novel structure was closely related to cancellous bone, and massive cell viability proliferation was also observed to the strut surfaces after 28 days. However, the implant did not support maximal calcium deposition due to the complex strut structure. Bandyopadhyay et al. [179] fabricated porous Ti and Ta with 30% volume fraction porosity, and after 12 weeks of in vivo in rat distal femur, there were higher osteoid formations on both implant surfaces. "The scanning electron microscopy result confirmed that the volume fraction porosity influenced the osseointegrations of the Ta and Ti implants. Furthermore, Jamshidi et al. [174] measured mechanical properties of SLM Ti6Al4V sample in x and y build up directions by using the universal testing machine. Due to the needle-like alpha' martensite structure, a higher Young's modulus and ultimate tensile strength in vertical direction were obtained. The cellular affinity of the sample was improved with increased bone to implant integration after post processing with etching,

as indicated in Fig. 8. An increased cellular activity on the surfaces were obtained, whereby the chemically etched surfaces demonstrated higher biological response with more SAOS-2 cells deposited after 14 days of immersion (Fig. 8(a) and (b)). Bandyopadhyay et al. [191] fabricated Ti3Al2V with reduced Al and V content with different orientation of 0°, 45° and 90° and they found that the compressive yield strength was reduced up to 18% which is 968 MPa compared to Ti6Al4V (1178 MPa) at 90° build orientation. The cyclic loading of Ti6Al4V fabricated at 90° on the other hand were the highest limit of 250 MPa which able to withstand the cycling multiaxial loading during the contraction and relaxation of the anatomical muscle.

Currently, Co-Cr is used in biomedical implants; however, the repetitive force and dynamic exerted on the teeth resulted in fatigue damage to these dental prostheses. For example, Latecola et al. [188] fabricated an implant from SLM Co-28Cr-6Mo and placed it in the rat femur for 12 weeks. Excellent bone integrity, good tissue integration without infection, and no sign of osteonecrosis were elucidated. The sample with 56% of porosity showed the highest bone remodeling proliferation. Another study by Lu et al. [190] produced an ace-centered cubic (FCC), and hexahedron (HCP) of Co29Cr9W3Cu with 65% of porosity to evaluate the bone ingrowth behavior. Both samples were implanted in vivo in goats for 12 weeks, and outstanding biocompatibility and osseointegration bone growth inside the pores were obtained. From the in vitro study of BMSCs, those samples excellently exhibited cell proliferation and non-cytotoxic under fluorescence staining. Similarly, by Caravaggi et al. [186] found that most of the CoCr strut scaffolds showed cell viability and high ALP; however, unstructured pore size had a reduced effect on the on cell vialibility and adhesion while smallest pore size structure demonstrated the least bone remodeling for the formation of anatomical bone.

Naujokat et al. [189] printed the nickel-titanium (NiTi) which belongs to the group of shape-memory alloys (SMAs). The clinical plantation demonstrated that there was no sign of local inflammation, or any wound healing disturbance *in vivo* of pigs. The implant which has direct contact with the pig showed well connective tissue between the struts and the bone overgrowth with nearly complete coverage of the implants.

4.4. Antibacterial

Implant infection caused by bacteria, also known as osteomyelitis (OM), is a serious issue in the field of biomedical implants that may result in bone loss known as osteolysis. These contaminations develop through the inflammatory response of the human body against microbials that spread through the bloodstream toward the implant. Then, the amount of antimicrobial on the surface of AM implants has been increasing due to the resistance of microbials to antibiotics [192]. The microbial can colonize the surface of AM implants by forming a thin film known as biofilm. Fig. 9 shows the biofilm formation process of bacterial on the SLM surface. It begins with the microorganisms attach to a surface via van der Waals interactions and then the bacteria colonized to the implant surface [193]. During colonization, collagen-binding adhesive proteins known as polysaccharides between the bacteria cause stronger hydrophobic/hydrophilic interactions [194]. Then, the bacteria develop, and the cells are accumulated through proliferation. A stable formation of bacteria that underwent mitosis is divided, followed by the maturation phase, which is distributed around the sample [195].





Fig. 8. Assessing the SAOS-2 cells' metabolic activity after seeded on the prefabricated Ti6Al4V surface [174].



Fig. 9. Stages of biofilm formation on the surface of implants [199].

During the active dispersal, the bacteria evolve in multiple layers, then forming a thick layer of biofilm [194]. Finally, the bacteria divide into multiple levels of biofilm formation, including initiation, maturation, and dispersal, colonizing on other location [196]. They are usually known as *Staphylococcus aureus* (*S. aureus*) and *Escherichia coli* (*E. coli*), both are gram-negative bacteria and highly resistant to most medicine and antibiotics [197]. It has been shown that microbial activity is responsible for more than 20% of overall implant failure and causes billions of dollars in damages each year [198]. Microbial infection is acknowledged to be one of the main causes of implant failure in biomedical industries. Table 7 summarized the methods to reduce the implant infections and toxicity from the previous study.

A study has been conducted to improve the antimicrobial mechanism

Table 7

Method to reduce implant infections and toxicity.

via copper (Cu)-bearing which an important ion in living organism act as an bactericidal element. Therefore, developing a new method in SLM for incorporating different elements into the alloy and maintaining its mechanical properties and biocompatibility is crucial. The new strategy in developing new alloys has been promoted back in 2004 by Cantor et al. [216] and Yeh et al. [217], whereby they proposed multicomponent of materials that exhibited the "core" properties of high entropy, lattice, and slug diffusion. These are known as entropy alloys or multicomponent ions. Using this method, the Cu is able to be designed with good antibacterial properties and excellent mechanical properties. A recent study by Zhou et al. [218] showed that the new type of antibacterial implant containing Cu exhibits good antimicrobial than the common 316L implant in the market. Similarly, Wang et al. [219] indicated the

Material	In Vitro/Antimicrobial Technique	Yield Strength (MPa)	Remark	Ref	Year
CoCrCu	E. Coli (ATCC25922) and S. aureus (ATCC 25923)	565.0	CoCrCu alloys have effective antibacterial and antibiofilm properties.	Ren et al. [200]	2016
Ti6Al4V	MRSA strain AMC201 and hMSCs	636.5	SLM PEO silver-releasing implants demonstrated significant antibacterial efficacy, with no evidence of cytotoxicity.	Hangel et al. [201]	2017
Ti6Al4V–Ag	E. Coli (ATCC25922)	860.0	Ag exhibit moderate antibacterial capability compared to Cu	Macpherson et al. [202]	2017
Ti6Al4V-xCu	S. aureus, (ATCC 6538) and E. coli (ATCC 2592)	-	Stronger antibacterial properties were demonstrated by SLMed Ti6Al4V alloys containing 4 and 6 wt% Cu.	Guo et al. [203]	2017
CoCrWCu	E. coli. (ATCC 25922)	540.0	The SLMed CoCrWCu alloy had outstanding antibacterial action of E. coli.	Lu et al. [204]	2018
Ti6Al4V	MG-63	_	Ti6Al4V alloys with 6 wt% Cu content exhibit good cytocompatibility.	Luo et al. [205]	2018
Ti6Al4V	MC3T3-E1	850.0	Samples containing zinc showed inhibitory effects on S.sanguinis.	Wu et al. [206]	2019
ZK30-Cu	S. aureus (ATCC25923)	_	As Cu content increased, the colonies of S. aureus decreased	Xu et al. [207]	2019
Ti6Al4V–TiO2/ Ag	E.Coli (MG 1655)	1110.0	Ceramics strengthened with titanium had a markedly impaired antibacterial action.	Rahmani et al. [208]	2019
316L	BMSCs, E. coli (ATCC 25922 and S. aureus (ATCC 25923)	470.0	stainless steel 3D printed with zeolite coatings containing silver have improved antibacterial activity	Qing et al. [209]	2020
Fe-xCu	E. coli (ATCC 25922)	_	The SLMed Fe-2.3Cu exhibits significant antibacterial activity.	Guo et al. [210]	2021
ZK30-Cu-Mn	MG63 and S. aureus, (ATCC 25923)	730.0	The cells adhered on the SLMed with spindle and round shape according to the cell morphologies. No detection of S.aureus after 72 h	Xie et al. [211]	2021
Ti-xCu	MG63 E. coli (ATCC 25922)	760.0	A continuous, very potent antimicrobial rate (antibacterial rate>99%) was demonstrated by the SLMed Ti-3Cu.	Ji et al. [212]	2021
Ti6Al4V	MC3T3-E1 S. aureus (ATCC 25923)	880.0	Increased alkaline phosphatase levels and pre-osteoblast metabolic activity	Fazel et al. [176]	2021
CoCrCuFeNi	E. coli (ATCC 25922) and S. aureus (ATCC 25923)	516.0	Excellent antibacterial properties of the CoCrFeCuNi HEA suitable for use in medical applications	Gao et al. [213]	2022
Ti6Al4V	MG63 and S.aureus (ATCC 25923)	950.0	Ca^{2+} ions with an atase TiO ² layer increased the bioactivity and adhesion of the scaffold	Rajendran et al. [214]	2022
Ti6Al4V–Ag	E. coli (ATCC 25922) and S. aureus (ATCC 25923)	890.0	The antibacterial capabilities increased after 12 days against <i>S. aureus and E. coli</i>	Sedaghat et al. [215]	2024

Abbreviation: MRSA-methicillin-resistant Staphylococcus aureus, and hMSCs PEO- Plasma electrolytic oxidation MG-63- human osteosarcoma cell, MC3T3-E1- murine calvarial cell line.

copper bearing of 316L results, whereby there was no cytotoxicity effect on rat mesenchymal stem cell and excellent antimicrobial against *E.Coli* and *S. aureus* after 24 h of incubation. This is attributed to the Cu that homogenously distributed on the microstructure of the sample. However, the *in vivo* test did not demonstrate anti-infection under mesenchymal stem cell.

Many attempts have been made to use antibacterial alloys and reduce the colonization of biofilm. For example, Chen et al. [220] evaluated the performance of blended powder, based on CoCrFeMnNi as a high entropy alloy (HEA) fabricated utilizing SLM technique. This study found that the HEA alloy can release more Cu2+, preventing biofilm formation. Also, Gao et al. [213] fabricated antibacterial CoCr-CuFeNi HEA alloy via SLM to determine the antibacterial capability and biofilm formation. After 24 h of immersion in the bacterial colony of *S. aureus* and *E. coli*, the sample possessed antibacterial rates against *S. aureus* and *E. coli*, almost 95% and 80%, respectively. A thin biofilm was formed with more dead bacteria on the surface, indicating an excellent antibiofilm formation capability. It is due to the release of Cu ions and the act of microphage that engulf the bacteria.

Many studies have explored CoCr material in SLM as base metal coping due to their favorable properties, including excellent mechanical strength, ductility, corrosion resistance, and biocompatibility. For instance, Ren et al. [200] fabricated novel metal copings of CoCrCu alloy by using SLM and proved that the addition of Cu increased the antibacterial and antibiofilm capabilities. Also, Lu et al. [69] used a highly dense CoCrWCu alloy to fabricate dental restoration following the ISO standard. A fine microstructure with a columnar grain was obtained, and the *E. coli* attached to the CoCrWCu were fewer after 24 h than the control group.

Currently, researchers have explored the optimum surface roughness to prevent microbial attachment on the surface of biomedical implants. Moore et al. [221] suggested that a fine surface roughness can resist the *S. aureus* attachment on the hip-stem SLM implant. Fig. 9 compares the gram-negative bacteria *S.aureus* attachment between a fine and rougher surface of the SLM implant. It was found that the Ra value of the hip stem, 78.440 \pm 4.85 µm showed higher biofilm colonization (Fig. 10(a)) in comparison to its smoother counterpart, of $17 \pm 0.42 \mu$ m (Fig. 10(b)). *S.aureus* is the most common pathogen of periprosthetic joint infections that prefers the creases, shard edges or ridges on the implant's surface. These fasten the delivery of nutrients from both sides of the biofilm,

encouraging bacterial growth and leading to infections.

Currently, Ti-based alloy has been modified by mixing some alloys like Cu or Ni to provide a strong antibacterial resistance in biomedical implant [203,205]. Guo et al. [203] doped Cu into Ti6Al4V powder while producing the implant by using SLM (Mlab-R, Conceptlaer). The effect on microstructure, phase constitutions, antibacterial properties and cytotoxicity of samples were observed. The Cu element is effectively fused from Ti6Al4V, resulting in stable properties of antibacterial activity against E. coli and S. aureus. An excellent cytocompatibility from an in vitro test by using BMSc was obtained; however, the density of the implant gradually decreased as the Cu content increased, attributed to the presence of micropores. Another study by Luo et al. [205] suggested that 6% of Cu added into SLM Ti6Al4V exhibited good cytocompatibility in vitro MG-63. The SEM and fluorescent staining demonstrated excellent cell morphology and adhesion on the sample. The most recent study by Bandyopadhyay et al. [222] manufactured a Ti3Al2V custom alloy with the addition of 10 wt% Ta and 3 wt% Cu to investigate the antibacterial resistance against P. aeruginosa and S. aureus at the initial stage of osseointegration. The bone implant contact was the highest and showed no gaps on the sample, and the presence of Cu increased the osteoid formation, which filled the pores of the Ti3Al2V implant. S. aureus and *P. aureus* decreased with the duration of the agar plate counting after 36 and 48 h compared to the conventional CpTi and Ti6Al4V implants, as shown in Fig. 11. This is attributed to the antibacterial effect of the TiO₂. Photocatalytic property that results in the lysis of the bacterial cell membrane.

Surface modification or coating on the surface of SLM implants is one as an effective method to enhance the bioactivity of Ti6Al4V. Rajendran et al. [214] treated the surface of SLM Ti6Al4V with silver nitrate (AgNO₃), an inorganic compound to enhance the bioactivity as well as antibacterial capability. The SLM Ti6Al4V was coated for 1 h at a temperature of 600 °C and demonstrated good bioactivity in the simulated body fluid (SBF), which corresponded to a higher number of Ca²⁺ and anatase TiO₂ layers. A good Ca:Ag ratio of 9.9:0.1 for the prevention of attachment of bacterial in *S.aureus* immersion was observed. Similarly, Fazel et al. [176] used silver nanoparticles (AgNP) to coat the SLM sample containing Ca⁻ and P⁻ ions by using the PEO method. The result showed a spindle-like hydroxyapatite(HA) formed which allow the anchorage of osteoblast and osteoclast cell for bone remodeling process. There was a higher level of alkaline phosphatase activity (ALP) in *in vitro*



Fig. 10. SEM images of bacteria adhered to surfaces with different roughness on femoral hip stem of (a) smooth counterpart and (b) rough counterpart [221].



Fig. 11. Bacterial viability and bacterial colonies on the CpTi, Ti6Al4V and Cu addition Ti3Al2V [222].

study, as well as a crystalline transition of the Ca-P compound formed on the surface. After 8 h of culture in S. aureus, the trace amount of AgNPs in the PEO layers significantly increased (***p < 0.001), improving their antibacterial activity. This is due to the embedded AgNPs that release the Ag ions, which prevents the bacteria from attaching and colonizing the implant surface. Recent study by Sedaghat et al. [223] doped AgNPs with Ti6Al4V and tested the antibacterial efficacy for a duration of 12 days in S. aureus and E. coli. There was an increase in the mineralization of MSCs attached to the AgNP-coated implant (p < 0.05). The S. aureus and E. coli bacterial properties decreased within 96 h, and after 288 h, S. aureus possessed a thicker peptidoglycan layer compared to E. coli, which acted as a protective barrier that prevented the penetration of antimicrobial agents. The On the other hand, Chen et al. [224] deposited Ag composite coating on the surface of the SLM implants with a varied time from 0.5 h to 4 h to evaluate the antibacterial rate. As the spraying time increased, the antibacterial rate of the E. coli on the implant increased to 60%. This corresponds to the higher release of Ag ions that generate reactive oxygen species (ROS) when in contact with human cells.

Alloying Ag with Ti is an alternative to increasing antibacterial capabilities and localizing antibiotics during implantation. A study by Xu et al. [225] demonstrated that the addition of Ag to Ti–Mo showed a minimal effect on cell viability compared to the control group. The cytotoxicity study indicates that the Ag exhibits no cytotoxicity to the Ti alloy, where the cells are clustered in a circular shape and connected by pseudopods. On the other hand, the antibacterial behavior of the Ti–Ag decreased up to 85%, attributed to the increase in Ag concentration. A similar study by Macpherson et al. [226] ound that Ag-containing Ti showed moderate antibacterial properties with a 300% increase in ductility. They concluded that it could be caused by the low alloying content of Ti6Al4V. Lowther et al. [226] assessed the Ti–Ag alloys that were fabricated by the powder blend method. The high Ti–Ag content, up to 35% intermetallic, showed an insignificant change in antimicrobial action compared to the control sample. However, Ag precipitate formed, attributed to the increased diffusion at the grain boundaries. In an *in vivo* study, Liu et al. [227] fabricated a Ti–Ag alloy with a nanotube coating and implanted the sample on the right femoral of a rat. There was an increase in alkaline phosphatase expression, and the highest cell attached to 4% of the Ag content. As for the low toxicity, the 2% content of Ag was optimal. The study was consistent with a previous study where it was demonstrated that 2% wt Ag–TiO2 exhibited the highest cell attachement [228,229].

Titania (TiO₂) is a ceramic material catalyst containing manganese and rhenium, which is used as a thin film or additive in composites to fasten the chemical reaction between the sample and the active coating site [230]. However, it has limitations to be used due to its high brittleness and low fracture toughness (2–3 MPa m^{1/2}) [231]. Rahmani et al. [208] fabricated lattice structure made of Ti6Al4V and coating with TiO₂-2.5% Ag by using plasma spark sintering. The result suggested that there was no significant effect on the antibacterial activity of *E. coli* after the addition of TiO2-2.5% Ag.

Among Mg alloys, ZK30 also known as Mg–3Zn-0.5Zr has attracted attention in biomedical field due to their good biodegradation resistance and mechanical properties. The antibacterial properties of ZK30 can be improved by alloying with any antimicrobial metal. Xu et al. [207] studied on the antibacterial and biodegradation resistance of ZK30 with the addition of antimicrobial Cu. The grain size of the ZK30 decreased, and the hardness increased with the increasing amount of Cu content. The antibacterial effect of *S. aureus* increased as the soaking time increased, attributed to the increase in surrounding pH that degraded Cu-ZK30. A recent study by Xie et al. [211] evaluated the combination of Cu-ZK30 with Mn, known as an essential element in body microenvironment to enhance the biodegradation and antibacterial of Cu-ZK30. After 7 days of MG63 cell culture, the number of stain cells on the surface increased, suggesting the SLM sample is cytocompatibility, and the colonies of *S. aureus* were none after 72 h of immersion in a pH 7.2 environment. The sample of the highest Mn content (8 wt%) demonstrated the strongest antibacterial ability and good cytocompatibility. However, The micro-galvanic corrosion between the Cu-containing intermetallic compounds and the alloy matrix causes Cu-containing ZK30 to erode in the 7.0 pH conditions [232–234].

Table 8 demonstrates the previous study of modifications done on the SLM metallic biomedical implants prior to their clinical application on their osseointegration, antibacterial, corrosion, and mechanical properties. Generally, this table provides a valuable study on the relationship between the modifications and their biocompatibility. It is important to recognize the limitations, of the SLM metallic implant like the high elastic modulus and corrosion resistance. However, due to their superior bone remodeling and antibacterial capabilities, Ti6Al4V and Cp-Ti dominated the SLM-fabricated implants used in the biomedical field. Thus, further study, such as further modification and alloying processes, needed to be done on other metallic implants to enhance their reliability prior to clinical application.

5. Post processing

In improving the antibacterial, biocompatibility, corrosion resistance, and clinical properties of SLM metallic to meet the appropriate applications and treatments, various surface modification technologies have emerged. The post-process modification techniques are divided into four categories: mechanical, physical, chemical, and biological modification, as shown in Table 9.

5.1. Mechanical and physical modification of SLM implants

The mechanical and physical modification of the implant is a method of exposing the surface to charge, laser, or plasma to improve their performance by altering the topography of the metallic substrate. These two techniques do not change the chemical composition and properties of the SLM metallic implant. They are commonly known as plasma spray [239], plasma immersion [240], and physical vapor deposition [241]. In order to improve crack resistance of implant, Li et al. [242] employed plasma spray ceramic coatings of Al₂O₃, ZrO₂, and Al₂O₃-ZrO₂ composite on SLM CoCrMo alloy. The coating adhesion reached 238 MPa and there were morphology enhancement with fewer microcrack and higher corrosion resistance. This is due to the melting temperature of Al₂O₃ (2050 °C) and ZrO₂ (2680 °C) below the melting temperature, which led to few porosity and micro-cracks upon solidification. In addition, the conventional mechanical techniques, such as machining, grinding, polishing [243,244], sandblasting [245] and friction stir has been widely explored to modify the surface of SLM fabricated sample. Likewise, Liang et al. [243] used the laser polished technique on SLM-built Ti6Al4V and found that the roughness of the sample reduced from 10.2 µm to 2.1 µm. The lowest roughness sample exhibited the highest cell attachment in in vitro of MC3T3E1. Besides, eliminating the partially melted powders, Zhao et al. [244] suggested a reduction of 75% of the roughness of SLM-fabricated SS 304L following electrochemical polishing of the sample. Similarly, Obeidi et al. [246] found that the surface roughness of SLM fabricated SS 316L was reduced from 10.4 μ m to 2.7 μ m after laser polishing, enhancing the apatite formation after 72 h of immersion in a pH 7.2 environment. Nevertheless, Bai et al. [247] demonstrated formation of oxide on the top surface of SLM SS 316L caused by the heated particles during polishing. Li et al. [248] polished the sample of Inconel 718 of temperature 1683 K and cooling of 2.46×106 K/s that showed reduction of porosity up to 65%, mainly caused by the equiaxed columnar grain formed on the surface.

Sandblasting involves the abrasive modification that is fed at high air flow directed to the substrate at high pressure. It is a common method to clean or to smooth and clean the surface of SLM implant. Yang et al. [249] employed sandblasting on SLM Ti and demonstrated a sample lowest water contact angle. It showed the osteoclast differentiation and bone osteoclastogenesis in in vitro study, reducing the bone resorption after implantation. Ly et al. [250] removed the spheroidized particles and corrosion cracks on SLM SS 316L sample surface. The pores formed after sandblasting inhibit the Ca^{2+} and P^- attachment under the SBF. Baciu et al. [251] polished the dental implant of SLM CoCr alloy by using three stages of sandblasting and found that a roughness of sample approximately 4.34 µm. The lamellar like cell of keratinocyte was observed on the surface. The physical and mechanical methods from previous study have shown simple technique to improve the surface of implant. However their ability in optimizing the surface of SLM metallic implants is limited yet inadequate for complex shapes of implants.

A chemical vapor deposition method called plasma vapor deposition is used to deposit a thin films originating from a gas vapor [252]. It has increased applications in engineering field for its ability to control temperature, deposition, and distance rate, allowing the substrate to be protected against wear, corrosion, and damage. Lassegue et al. [253] deposited CrZr on SLM Cu to enhance the optical absorption, however, the coating was oxidized causing it to leach. On the other hand, Wozniak et al. [254] coated SLM Cp-Ti with CrN and TiN, whereby the sample showed best pitting corrosion resistance with transpassivation potential of 3052 ± 143 mV and a maximum charge transfer resistance of 1132 k Ωcm^2 in the charge transfer resistance. This is attributed to the plasma vapor coating that formed double layer consisting of compact and a porous external layer. Nevertheless, there is still lack of study in plasma vapor deposition in biomedical field.

5.2. Chemical modification of SLM implants

Chemical modification method refers to the various processes that involve the alteration of the chemical composition or structure with the presence of a new substance to improve the performance of the metallic implant. These methods are mainly classified as chemical vapor deposition, electrochemical treatment, sol-gel, etc. The most common is chemical vapor deposition, whereby a solid material is deposited from a vaporization by chemical reactions. Konalov et al. [255] modified the surface of Ti-6.5Al–1Mo–1V–2Zr with Ni-based superalloy Ni-16Co-11Cr powder with coating thickness of 70 µm-130 µm and surface layer and the substrate is wave-shaped (Fig. 12). As shown in Fig. 12, microcracks started to appear throughout the coating due to mismatch between the substrate and the coatings coefficient of thermal expansion (CTE). Sitek et al. [256] modified the surface of nickel superalloy IN 718 with the participation of AlCl₃ vapor at a temperature of 1040 °C, and lasted 8 h at a reduced pressure of 150 kPa. Nevertheless, the IN 718 showed degradation in mechanical properties with visible microcracks after the aluminizing process. The reduction in mechanical properties was mainly caused by the brittle fracture on the sample.

The sol-gel method occurs in a solution based. Macera et al. [257] prepared silica sol-gel coating to enhance the corrosion resistance of SLM Al6061. The sample showed pH values ranging between pH 6.9 and pH 7, demonstrating a good corrosion resistance. Koo et al. [258] prepared sol-gel dense SiO2 insulation into SLM Fe-6.5 wt% alloy, whereby the coating is able to preserve the shell thickness of 2.2 mm. Xu et al. [237] improved the biological and bone regeneration of SLM Ti6Al4V through subsequent alkali treatment. Thus, previous study has stated that the development of surface modification technique of SLM metallic implant to satisfy the clinical needs. The chemical mechanism of these modifiers is relatively complex and expensive and often used in the preparation of complex samples.

Table 8

Summary of the SLM metallic implant's modification, osseointegration, antibacterial, corrosion, and mechanical properties.

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Material	In vitro/In Vivo Studies	Modification	Osseointegration	Antibacterial/ Toxicity	Corrosion	Tensile strength (MPa), Yield strength (MPa) and Young Modulus (GPa)	Ref
Ti6Al4V	rMSCs	Reinforced with growth factor BMP-2	The sample's cell growth was higher than that of the uncoated group				Lv et al. [235]
Zn	MG63	Al	There was an increase in macrophage labeling and fibrous encapsulation.		The least amount of corrosion is inhibited by lower Al concentration.	37, NR, 95	Guillory et al. [97]
Ti6Al4V	L-929 fibroblast cell	Simvastatin/ poloxamer 407 hydrogel	Improved neovascularization under <i>in vitro</i> study	Poloxamer 407 showed the lowest toxicity and weak immunogenicity.		960,NR, 90	Liu et al. [236]
CoCrCu	rMSCs	Cu	The addition of Cu increases cell adhesion and decreases biofilm formation on the sample.	On the surface of SLMCoCr alloys, 98% of <i>E. coli</i> and 92% of <i>S. aureus</i> survived.		991,636.5220	Ren et al. [200]
Ni	Dorsal sides of mice	Doped with zirconium (Zr)	After 4 days, the tissue was excised from the sample with no inflammation response.	A thin layer of biofilm formed on the sample.	When immersed in the PBS, the sample exhibits the least amount of corrosion.		Sonofuchi et al. [155]
Zn	HUVECs	Pure Zinc and Zinc Alloy	After four days of incubation, the Zn alloy had the highest cell viability.	Severy cytotoxicity of the pure Zn	The galvanic effect caused the release of higher Zn ions during corrosion.	225,130,85	Wang et al. [64]
SS 316L	rMSCs	Doped with Cu	The MTT assay demonstrated no cytotoxic effect <i>in vitro</i> .	The sample with Cu doping showed the highest antimicrobial activity against <i>E. coli</i> .		NR,45.2,20.9	Wang et al. [219]
Ti6Al4V	MC3T3-E1	Modified with HF and HNO_3 which of acid etching	After 8 weeks of <i>in vivo</i> . The etched sample exhibits a higher bone-to-implant contact, up to double that of the untreated sample.			900,850, 78	Xu et al. [237]
Zn	MC3T3-E1		Despite the release of passive corrosion products, no toxicity was evident after up to 20 months of implantation.		Zinc oxide, phosphate, and carbonate are the primary factor for the Zn implant's corrosion.	430, 325, 105	Drelish et al. [98]
Ti6Al4V	BMSCs	Cu	Higher Cu exhibits cytocompatibility when implanted <i>in vitro</i> .	A higher Cu content exhibits stable antibacterial capability against <i>E. coli</i> and <i>S. aureus</i> .	The corrosion of the Cu- doped sample was lower than that of the Cu-free sample.	80	Guo et al. [203]
Fe	Rabbit abdominal aorta	Nitride	The mass loss of the Nitride Fe sample was twice as high as that of the Fe sample, indicating that Fe is biocompatible for use as an implant.		The PBS study showed insoluble corrosion products were safe and were cleared by the macrophages.	92	Lin et al. [69]
Ti6Al4V		Cu and Ag	L	Cu's antibacterial capabilities were higher than Ag's.		930,860,113	Macperson et al. [202]
Mg–Zn	SBF	НА	Bone-like apatite formed a stable layer of MgCl ₂ after 1 week of immersion.		The grain refinement that formed on the SBF hindered hydrogen degeneration, resulting in decreased biodegradation.		Shuai et al. [91]
Ti6Al4V	hMSCs	AgNPs	In rat femurs with greater bone-to-implant contact, samples show no cytotoxicity.	The Ag sample exhibited antibacterial activity against pathogens and reduced inhibition of surface microbial growth.		900,790,110	Hangel et al. [201]
Ti6Al4V	Beagle		Osteoblast cells in the sample increased after 4 weeks of implantation.		TiO crevices formed on the sample due to the argon ion bombardment, increasing corrosion.		Zhao et al. [172]
SS 316L	SBF	Quenched	A globular and thick layer of apatite formed on the 316L sample compared to the quenched sample.		After 288 h in SBF, the passive current and corrosion current showed a higher corrosion potential on the 316L sample than the quenched sample.	614.3470.8,55	Kong et al. [170]
CoCr	E.coli	Cu and Mo	Cell attachments were higher on the Cu sample, specifically	Compared to Mo, Cu-bearing CoCr exhibits excellent		951,503, 215	Lu et al. [204]
						(continue	eu on next page)

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Table 8 (continued)

Material	In vitro/In Vivo Studies	Modification	Osseointegration	Antibacterial/ Toxicity	Corrosion	Tensile strength (MPa), Yield strength (MPa) and Young Modulus (GPa)	Ref
CoCr	MG63	Cu	on the cellular and columnar structures. The alkaline phosphatase activity was higher in the Cu- doped sample, promoting cell proliferation.	antibacterial capabilities against <i>E. coli</i> . 90% antibacterial activity against E. Coli and <i>S. aureus</i> was demonstrated by 4% Cu, while up to 99% antibacterial activity was demonstrated by 6% Cu.	Compared to 2% and 4%, 6% of Cu showed the least corrosion.		Lou et al. [205]
SS 316L	L929	НА	The HA-coated sample showed higher biocompatibility and exhibited no cell cytotoxicity.	·		730,NR, 214	Lou et al. [238]

Abbrevation: rMSCs-rat mesenchymal stem cells, BMP-2- Human Bone morphogenetic protein 2, Zn-zinc, MG63-human osteosarcoma cell, Al-aluminuium, L-929fibroblast cell, hMSCs-human mesenchymal stem cells, *E.Coli- Escherichia coli, S.aureus-Staphylococcus aureus*, Ni-nickel, Zr-zirconium, HUVECs- Human umbilical vein endothelial cells, MTT assay- (3-[4,5-dimethylthiazol-2-yl]-2,5 diphenyl tetrazolium bromide) assay, MC3T3-E1- murine calvarial cell line, HF- hydrofluoric acid, HNO₃- nitric acid, Fe-iron, PBS-phosphate buffered saline, SBF- simulated body fluid, HA-hydroxyapatite, AgNPs-silver nanoparticles, TiO-titanium(II) oxide, Momolybdenium, L929-mouse fibroblast cell.

Table 9

Surface modification categories to improve SLM based implants.

Mechanical Physical C	Chemical	Biological
 Grinding Blasting Polishing Machining Shot Ion Implantation Thermal Spray Microwave 	 Alkaline Acid Hydrogen peroxide Sol gel Chemical Vapor Deposition UV tradiction 	 Biomimetic Cross-linking Cell seeding Chemical conjunction





Fig. 12. The cross-section view of the material's microstructure [255].

5.3. Biological modification SLM implants

For the SLM-fabricated metallic implant to be used in human body, the substrate should be able to promote bone-to-implant contact with high biocompatibility, ensuring a long-term usage with efficacy. A common used biomodifications method is by coating the SLM metallic surface with a biological coatings, such as HA [238] and biomimetic coating [259,260]. Yan et al. [177] used biomimetic mineralization duplex treatment as a biomimetic coating on SLM Ti6Al4V sample. It demonstrated that a crystalline bone-like apatite and a thick titanate



Fig. 13. SEM imaging of SLM fabricated SS 316L coating with (a) 7.50 μ m and (b) surface morphology of the sample [238].

porous SLM Ti6Al4V similar to cancellous bone and used biomimetic calcium phosphate as a carrier to attach on bone protein. The biomimetic coating induced the osteogenesis and promoting the new bone formation in *in vivo* dog model evaluation. Sun et al. [260] employed the bioactive glass nanosphere coating (MBG) on SLM Ti6Al4V implant surface through the sol-gel process to obtain micro- and nano-topography. The hydrophilic water-to-implant contact angle of the sample was increased, and adhesion osteogenic differentiation was enhanced as compared to commercial CP-Ti in *in vitro* pre-osteoblast cells. The process-inherent appropriate surface roughness and cellular scaffold showed higher implant-bone contact for osseointegration. However, there is still lack of study in biological modification for SLM metallic implant which of Co–Cr, Mg and Zinc. Thus, more research are needed in SLM metallic to integrate the bone-to-implant for clinical application in the future.

5.4. Coating of pharmaceutical

Table 10

Drug delivery is a technology used to safely incorporate pharmaceutical or drug into the host for a specific therapeutic effect. These incorporation of drugs into human body acts as active materials to the site of actions and it is one of a challenges approaches. In reducing the restenosis rate of the implant, the implants are coated with drugs such as bioactive compounds [235,261–265]. The Ti implants are also deposited by using the and polymer coating to improve their biocompatibility [266,267]. These implants are modified and coated with different techniques, such as dip, spray or spin coating or even solvent casting as summarized in Table 10.

Surface modification of implant by using bioactive compound or also known as biomolecules enhance the surface of the implant and making them biocompatible which can act as a good drug delivery. Fina et al. [269] presented a study, whereby they employed SLM with a single step with an elevated resolution during the fabrication process. Then, paracetamols were loaded into the SLM sample and found that the paracetamols were released accordingly in the acidic environment of human stomach acid. Vaithilingam et al. [261] attached the paracetamols by using the immersion technique for a duration of 1 h. Prior to immersion, the sample was polished to improve the surface for drug attachment. The result demonstrated an improved wettability of up to 95° angle. This high wettability is due to the presence of oxides on implant surface. After four weeks of *in vitro* test, the Ti6Al4V demonstrated a stable layer of paracetamol, indicating that the drug can be efficiently used as a coating for implant to enhance the biocompatibility and reducing the

Table 10						
Biological modi	fication of SLM	metallic i	mplant on	drug d	eliverv te	echniques.

post-implant complications. Another study by Lv et al. [235], whereby they doped the surface of Ti6Al4V with growth doped fibrin glue to enhance the angiogenesis and osteogenesis in cases of bone defect. Their study focused on the controlled released of drug Bone-morphogenic protein-2 (BMP-2) and vascular endothelial growth factor (VEGF) which contributed to the main bone defect after implantation. The *in vitro* cell viability assay indicated there were no presence of toxicity to the cell, whereby there were an improvement of cell viability on Ti6Al4V scaffold surface.

Antibiotic is a well-known for the treatment of bacterial infections, whereby it helps in killing bacteria as well as preventing their growth in human body. Vancomycin is a well-known antibiotic and is recommended as a treatment for the skin infections as well as bone infections. Bezuidenhout et al. [263] and Zhang et al. [268] utilized the effectiveness of vancomycin in treating bone infections. Bezuidenhout et al. [263] incorporated vancomycin into Ti6Al4V implant and confirmed that there was an enhancement in antimicrobial activity against the S. aureus. The drug delivery of the vancomycin was also evaluated, whereby they found that the implant contact angle significantly improved the permeability of the implant surface. These allow the release of the vancomycin after implantation. On the other hand Zhang et al. [268] demonstrated that structure containing vancomycin had a good antibacterial effect on E. coli and S. aureus. They found that there was an increased osteogenic expression of MC3T3-E1 cell attach on the surface after in vitro cell culture.

Drug releasing implant requires an appropriate surface modification. Polymers are available in two types, such as synthetic and natural. The natural polymers occurs in nature and can be extracted naturally. The most common biocompatible and biodegradable polymers are chitosan, poly(ɛ-caprolactone) and poly(3-hydroxybutyrate), and can be deposited on implant surface by spin coating or adsorption. Micheletti et al. [266] rinsed the SLM sample with acetone and kept in air-dried, pen-strep were used for the loading of antibiotics. They found that the release of the antibiotics was about 17% for a single-layer polymer and was double percentage rate for the two-layer polymer. The in vivo study in rat model demonstrated that the polymer is able to modify the drug release pattern in rat model. Similarly, Chudinova et al. [267] used the method of adsorption on Ti6Al4V surface by using poly (allylamine hydrochloride) loaded with calcium carbonate. The in vitro test demonstrated that the cell growth pattern on the implant surface showed the cell grew normally on the surface in spindle shape. The drug release of the sample showed increase in percentage after 1 week of implantation. However, there are some limitations in using polymeric

Enhancement Technique	Drug Delivery	In Vitro/In Vivo	Remark	Ref	Year
Immersion deposition	Paracetamol	3T3-L1	Not an ideal drug for immobilization as there were contaminant present on surface of sample	Vaithilingam et al. [261]	2015
Injection	Reinforcement with growth factor doped fibrin	rMSCs	Biocompatible and provide slow release of the bioactive growth factors. The angiogenesis were enhance with the addition of growth factor.	Lv et al. [235]	2015
Custom-made gel chamber	Simvastatin/ poloxamer 407 thermosensitive hydrogel	White rabbits	Sample with simvastatin demonstrated higher bone mineral density, and composition.	Liu et al. [236]	2016
Integrated permeable structure	Vancomycin	E.Coli and S. aureus	Controlled release of the antibiotics the exhibit the periprosthetic joint infections	Bezuidenhout et al. [263]	2018
Electrochemical Anodization	MTANi	hMSCs	Enhance ALP and antibacterial formed firm film, preventing biofilm formation	Li et al. [265]	2020
micro-arc oxidized (MAO)	Vancomycin	E.Coli and S. aureus	Scaffold with vancomycin exhibit least of bone infection	Zhang et al. [268]	2020
Spin coating	Biodegradable polymers	-	Presence of polymers modified the pen-strep release under the <i>in vitro</i> test of rat model	Micheletti et al. [266]	2021
Adsorption	PSS and PAH	3T3-L1	The surface modification affected cell morphology <i>in vitro</i> . No cytotoxicity and improved cell adhesion	Chudinova et al. [267]	2021

Abbreviation: 3T3-L1-mouse embryonic fibroblast, rMSCs-rat mesenchymal stem cells, *E.Coli- Escherichia coli, S.aureus-Staphylococcus aureus*, MTANi-Methylthioadenosine nucleosidase inhibitor, hMSCs-human mesenchymal stem cells, PSS-polystyrene sulfonate, PAH-poly(allylamine)hydrochloride, 3T3-L1-mouse embryonic fibroblast.

carriers for biomedical applications, includes surface fractures and coating waviness, which can cause inflammatory and hypersensitivity reactions as well as unequal medication distribution.

6. Applications

Currently, SLM printed biometallics are mainly focused in producing implants for tissue repair, dentistry, orthopedic including cranial, dental, total hip replacement as well as prosthesis. This is due to their ability of tailoring to anatomical structure, lightweight and ease of handling during clinical applications. There are various applications of SLM printed materials in biomedical applications, including skull, joint and hip replacement, and dental.

6.1. Skull

Trauma, inflammatory diseases, or malignancies are the main causes of mandibular and craniofacial bone abnormalities. These defects cause patients to suffer from speech, mastication, and aesthetic issues as they alter the bone structure, resulting in the need for implants [270]. Mohammed et al. [271] fabricated a lower jaw to replace the patient's fractured mandible by using the Ti6Al4V SLM technique. The patient underwent a CT scan to obtain the shape of the lower jaw, and the results demonstrated that the implant was able to support the patient's contraction and relaxation during speech. In addition, Peel et al. [272] used Ti6Al4V to fabricate a porous orbital floor for treating a post-traumatic zygomatic osteotomy patient. The implant offered a high bending stiffness to weight ratio, which improved the patient's facial symmetry and enhanced the ability to fully close the eyelid.

Head trauma including cranial defects is one of the arduous task for most craniomaxillofacial surgeons. Previous study has demonstrated the application of SLM technique for the cranial prosthesis to restore the skulls' integrity, ensure adequate protections and functionality of the underneath brain. Likewise, Sharma et al. [273] created a biomimetic patient-specific cranial prosthesis with an interconnected strut to fit the patient's skull of head trauma by using SS 316L. It was fabricated in a vertical direction; however, this complex structure was overhanging between the interconnected network (Fig. 14(a)). The combination of flange (Fig. 14(b)) and angular fixation (Fig. 14(c)) was used to fix the problem. This design of lightweight (30 g) biomimetic pattern prostheses was fit to the anatomical biomodel (Fig. 14(d)).

6.2. Joint and hip replacement

Generally, the mass productions of implants fabricated by the conventional method can serve the most of the patients population. However, due to the complex structure of anatomical, the ready-made conventional implant does not fit the clinical applications scenario, such as hip and severe joint surgery. The problem related to total hip replacement, which entailed the acetabular cup dislocated or loosened, has been shown in earlier research. Thus, SLM patient-specific implants has shown as a valuable method for addressing the most common causes of hip replacements. Colen et al. [274] fabricated custom-made reinforcement ring by using SLM for the trabecular surface of the acetabular cup to increase long term stability and to encourage bone ingrowth. The cortex screws were also used to achieve a maximum balance, and after 10 months of radiological follow-up, there was good attachment of bone cells around the porous acetabular cup. In addition, Baauw et al. [275] fabricated acetabular components by using Ti6Al4V for 16 patients with Paprosky Type 3 defects to determine the accuracy and stability of the implant. Under CT data, the implant demonstrated good position of inclination, anteversion and center of rotation as compared to the pre-operative data. Wang et al. [276] fabricated the femoral neck of a Ti6Al4V implant for severe hip arthroplasties. The 24 months of clinical data showed that the implant postoperative weight bearing was better than that of the conventional, attributed to a higher femoral neck anteversion angle. The implant demonstrated good stability with the physiological and anatomical structures of the patients.

Treatment of complex acetabular fracture and defects can also get benefit from the SLM technology. The fabricated implants could also



Fig. 14. Biomimetic cranial prostheses and a skull biomodel tailored patient (a) wireframe pattern prosthesis highlighting issues with structural printing (red-arrows) (b) prosthesis with screw fixing points on a flange (c) prosthesis with points for angular screw fixing and (d) cranial biomodel [273].

customized to the shape of the pelvis and acetabular fracture [277]. Wei β mann et al. [278] manufactured different struts type of acetabular cup from Ti6Al4V. It was demonstrated that the mechanical strength was depended on the struts to fit the prosthesis and the implants were matched to the anatomical features of the patients with porous structure that enhance osseointegration. Femoral implants were also constructed by using CoCrMo SLM by Song et al. [279], and they demonstrated a surface roughness of 7.9 µm on the planate surface and 16.8 µm on the curved surface, allowing the attachment of bone cells. The implant exhibited good corrosion resistance and biocompatibility.

6.3. Dental

Oral health and dental hygiene play a crucial and significant role in overall human health. Tooth wear occurs naturally because teeth are exposed to loosening, fracture, harnessing, overall chemical stressors, including damage, misalignment, and discoloration [280]. Thus, taking care of it is important for preventing any cavities or plugs. AM has come a long way in fabricating medical implants by leveraging pioneering science and engineering technology, as well as in the field of dental implants. Biocompatibility of the dental implants for the implantations for the prolong time is the most critical factor, including other characteristics of resistance to corrosion, fracture, and material strength. In fabricating a dental implant, the corrosive nature of the materials may lead to harmful events due to the acidic pH of the saliva and may corrode the surface of the dental. The natural microenvironment of saliva, which consists of 99% water, proteins, and electrolytes, may as well cause ion interchange, leading to the corrosive behavior of the dental implant [281]. Thus, in order to replicate tooth wear and simulate real-world oral cavity interactions, researchers have created artificial saliva by chemical formulation.

Dental implant prosthetics are used in replacing missing teeth as well as treating partially or fully damaged teeth. The primary types of dental implant, including endosteal or endosseous, subperiosteal, and transosteal [282]. Endosteal dental implants are inserted within the mandible of the skeletal face, as shown in Fig. 15(a). Subperiosteal dental prosthetics are positioned at the jawbone's gum while are fastened through bone remodeling and osseointegration. It applications include the common dentures, which are fixed and removable. As for transosteal implants, they needed sufficient bone support from the mandible commonly used for patients with severe resorption and jaw damage [282]. Screws were used to pierce through the jawbone and bolted through the lower mandible to support the metal rods, which acts as a support beam as shown in Fig. 15(b). The common type of implants used are the Cp-Ti [283] for the base of the implants, however, Ti6Al4V [284] and Ti-Zi [285,286] are mainly used for these kind of implants due to their excellent corrosion resistance.

SLM dental implants mostly depend on the two factors of porosity and part density, which are mainly determined by the energy density input. Low energy input may be attributed to the imperfection on the surface of sample, which is the molten layer that eventually results in the large pores. Zhang et al. [287] analyzed the corrosion behavior of dental implant SLM Ti6Al4V under the environment of the artificial saliva of pH 2, pH 4 and pH 6. The corrosion test revealed that the SLM sample of the Ti6Al4V had lower corrosion than the wrought sample. On the other hand, Kim et al. [288] assessed the Co–Cr alloy mechanical characteristics and microstructure under *in vitro* study and they found that all the samples of casting, milling and milling/post-sintering showed excellent crystallinity under XRD examination (Fig. 16). Nevertheless, the CoCr alloy exhibit dominant γ -FCC and it showed that the SLM has the highest superior mechanical properties due to the full local melting and rapid solidification that minimizes the porosity.

7. Way forward and challenges

The current AM technologies require the recognising the obstacles



Fig. 15. Schematic Dental implant consisting of (a) Endosteal dental implants and (b) a transosteal implant [3].

and comprehending the reliable technologies, as well as their future impacts, at an early stage in the procedure for making decisions [289]. Therefore, this section addresses the current and crucial challenges in the adoption of SLM technologies in the market, which involve the standard design, selection of material, manufacturing, and trial for biomedical applications. The microstructure has been hindered by the lack of simulation-based printing processes, which has reduced the mechanical characteristics of the manufactured parts [290].

As stated earlier, the current SLM metallics are still limited in their ability to obtain the appropriate properties to the anatomical structure. This manufactured implant's strength remains greater than that of the



Fig. 16. The XRD analysis of the CoCr alloy with different manufacturing process [288].

bone tissues. For instance, the SLM metallic implant (density 1.63 g/ cm³) and porosity of 65% demonstrated more strength than anatomical bone [291]. Thus, it is worth developing topological optimization to control the load transfer to mimic the bone implant in the future. This geometrical modeling structure can be determined by using the finite element softwares, such as Abaqus and Ansys. The ideal shape can also be assessed before being used as a biomedical implant by validating the structural performance using von Mises failure criteria.

The metallic implant fabrication by the SLM needed more research due to the particular geometric constraint and the stringent requirements for producing the quality of the struts [292]. In this sense, the best build geometry with the required quality that ought to be generated can be optimal by using the SLM parameters. This means that the pore and the lack of fusion defects can be avoided during the fabrication. The internal porosity could be reduced by using the pore-free feedstock, and the lack of fusion defects could also be minimized by thoroughly controlling the SLM process [293,294]. In addition, The topology optimization can also be used and formulated to study bone internal architecture and provide optimal topological structure of the fabricated sample. This limitation necessitates manual involvement and a less-than-ideal design approach, adding needless complexity, danger, and expense to the manufacturing process of biomedical implants.

Apart from these challenges and limitations, the cracks and void formation of the metallic material fabricated by SLM significantly lowered its tensile performance [295]. The formation of voids leads to lack of interfacial bonding between the adjacent layers on the sample, results in the layer delamination upon printing. These will result in a discernible difference between the interior material and the edges of the layers, leading to the creation of anisotropic microstructures [296]. These problems are the major causes of the formation of cracks to occurs, reducing the mechanical strength of the metallic SLM sample. To overcome these problems, adjusting the input parameters throughout the SLM process. The hatching distance, laser power, scan speed, hatch distance, and layer thickness needed to be carefully addressed. Prior to the experimental study, optimization techniques such as the Taguchi method and the response surface method can be used to enhance the design. The tensile property of the SLM-fabricated metallic wear is lowered due to the rough strut surface, internal defects, and incomplete decomposed alpha' martensitic microstructure [297]. In general, the brittle fracture of the implants still remains a great concern for the wider applications and needs to be appropriately addressed prior to implantation in the human body. Post-processing operations may improve the as-fabricated sample, which includes powder removal, stress relief, and eliminating or lowering print-induced flaws including porosity and lack of fusion [297]. HIP is mostly used on the SLM to reduce the build sample's porosity, enhance mechanical qualities, and relieve stresses [298]. This process involves high temperatures and applies to the sample in an inert environment. According to the research, the microstructure changes caused the interior porosity and tensile strength to decrease after the HIP treatment.

Recent research has shown the significant micro- and nanoscale porous materials are for the adhesion, differentiation, migration, and proliferation of cells. The integration of SLM and dealloying with other metals, such as Cu, Ti, Na, and Tb has the potential to produce structures and surfaces that replicate the properties of anatomical structures at the macro-to nanoscale [299-301]. However, the effect of dealloying on an implant's mechanical properties has not yet been explored. Metallic implants are bio-inert, and the survival of cells is still a main issue. It has been shown that surface treatment and modification can enhance the biological activity of SLM implant, leading to easier fixation and improving long-term stability. The antibacterial properties of SLM metallic implant could be improved by the introduction of silver, antihistamines, and antibiotics, as discussed previously [302,303]. These coatings prevent the implant from with a wide range microorganisms that causes infections without affecting its characteristics. However, the surface coating may result in surface leaching, which would reduce implant integrity [304]. Thus, it is great to achieve access to a novel exploration in the future.

As for the load-bearing application, especially in the orthopaedic field, which includes multiple compressions, tensions, and bending applied during contraction and the relaxation of muscles tissue [305]. This main scenario is crucial for the SLM-fabricated metallic implant. Therefore, further study is needed to enhance the reliability of the SLM-fabricated parts and prevent the unfavorable effects caused by the

SLM on surface roughness and cell proliferation. In addition, the product cost of SLM is higher in comparison to conventional manufacturing of similar geometries and has been determined to be among the most overburdening elements. The large environmental impact mostly is caused by the high amount of waste powder. By using different grades according to the melt flow rate of the recycled powder, the material quality and the usage of fresh powder can be used efficiently. Thus, more optimal recycling and reusing the metallic implants from the SLM fabrication could reduce the cost to both patients and the clinical industry. In addition, the consumption of compressed air of nitrogen from the external reservoir might as well cause significant environmental impact. To avoid this, by sealing the process chamber appropriately so there is no longer needed of continuous filling of the chamber. The total production cost of the metal SLM should be reduced based on the optimal topological optimization of SLM in reducing material, labor, power, and equipment costs.

8. Conclusions

Metallic implants fabricated by SLM method have been manufactured and utilized in different medical procedures due to their excellent biocompatibility and non-toxicity. The surface modification on the treatment has broaden the usage of the SLM metallic in treating the fracture and bone diseases, particularly for load bearing applications. Therefore, this review article had summarized and demonstrated the previous works on the widely used powder-based SLM method of metallic implant fabrication. A large amount of research has been focused regarding how the parameters of the process relate to the characteristics of the final product, such as scanning speed, hear course, product microstructure, mechanical properties, corrosion, and defects. The SLM method used in clinical applications has also emerge, including surface modification, drug development, biomaterials and tissue engineering. Also, this review has demonstrated some attempts made in the evolution of SLM fabricated metallic implant to offer more understanding of the methods, parameters, properties of the fabricated parts including their structures. Thus, acquiring knowledge in producing a defect free product efficiently and sustainably. The innovations in the powder materials can as well be explored since the powder of unique properties have the potential to broaden the scope of applications.

Therefore, more research and development on process optimizations of SLM metallic implant is necessary in the future. Current SLM implants yet to focus on the material, printed quality, processing complexities and mechanical properties that has been discussed in this paper. Some of the future directions include the usage of composite and alloy powders can be uncovered due to the immense number of combinations. Then more focus on the post processing surface modification as well as the Zdirectional printing as it can reduce the martensite and anisotropy of the implant. Another method for reducing the chances of nozzle clogging and the cost of the feedstock, using multi-nozzles during SLM fabrications, would be interesting to explore. Finally, combining various printing lasers with a high deposition rate could increase productivity with good accuracy. This involves the re-melting process and postprocessing in reducing and repairing the defective parts on the sample.

Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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