ETHYLENE YIELD FROM A LARGE SCALE NAPHTHA PYROLYSIS CRACKING UTILIZING RESPONSE SURFACE METHODOLOGY

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DEDICATION

This thesis is dedicated to my late Abah (Zakria Mohamed), beloved Ummi (Rabiah Awang), Aboh (Jaafar Ali), Mama (Zakiah Ismail), supportive wife (Amirah Jaafar), and beloved children (Farhah Amni, Hanan Sufiyyah, Iffah Mardhiyyah, and Wafa Nadheera).

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ABSTRACT

It is desirable in industry to optimize the production yield from olefin plant, for instance, to achieve the highest profit from the yield of ethylene. However, there are gaps associated with the restricted usage of the proprietary simulation software and the lack of a specific model to correlate the relationship between variables that has a significant impact on the processing parameters. In this study, response surface methodology (RSM) was used to evaluate the impact of critical operating parameters from large scale naphtha pyrolysis cracking. Those parameters include hearth burner flow, integral burner flow, naphtha feed flow, dilution steam flow, and coil outlet temperature (COT), with the addition of propylene yield towards the ethylene yield. The data was collected at the steam cracker furnace using process information management system (PIMS) software, PI process book version 2015. The analysis was conducted for naphtha feedstock with paraffins content at 57.60 - 70.73 vol % to evaluate the impact of operating at different naphtha feedstock compositions on the ethylene yield. Propylene yield, hearth burner flow, and naphtha feed flow consistently showed a significant relationship with ethylene yield from surface response analysis with the interaction factor ranges at -10.07 to 192.3, -0.001698 to 0.01938, and -2.383 to 820, respectively. The final equation models were successfully concluded in the form of quadratic with 2-way interaction at the high paraffins content and linear relation at the lower paraffins content after the models' validation using probability plot, scatterplot, and Mann-Whitney hypothesis test. The maximum ethylene yield generated from response optimizer was observed dissimilar at 31.46 - 34.97 % for different paraffins content in naphtha feedstock, with the highest reading observed for the naphtha feedstock having the highest paraffins content. The best ethylene yield with consideration to the production cost for the naphtha with the highest paraffins content of 70.73 vol % was identified at the range 34.41 - 34.97 %, using the recommended process ranges at 12.22 - 12.25 % of propylene yield, 11033.90 -11816.40 kg/hr of hearth burner flow, 66.67 - 67.05 t/hr of naphtha feed flow and 816.38 °C of COT. It is recommended for other large scale plants to adopt the same methodology that was proven successful in this study, for process monitoring and optimization.

ABSTRAK

Proses mengoptimumkan hasil pengeluaran daripada loji olefin adalah suatu keperluan di industri, antaranya untuk mendapatkan keuntungan tertinggi daripada hasil etilena. Walau bagaimanapun, terdapat kekangan disebabkan oleh keterbatasan pemilikan perisian simulasi, serta kekurangan model untuk menghubungkaitkan pembolehubah-pembolehubah yang mempunyai kesan ketara terhadap parameter proses. Kaedah tindak balas permukaan (RSM) telah digunakan dalam kajian ini untuk menilai kesan pembolehubah operasi kritikal daripada proses pemecahan pirolis nafta berskala besar. Parameter yang dimaksudkan adalah aliran penunu perapian, aliran penunu integral, aliran nafta, aliran wap pencairan, suhu salur keluar gegelung (COT) dengan penambahanan hasil propilena ke arah keberhasilan etilena. Data dikumpul di relau stim menggunakan perisian sistem pengurusan maklumat proses (PIMS), buku proses PI versi 2015. Analisis dijalankan kepada bahan mentah nafta pada kandungan parafin 57.60 – 70.73 vol % untuk menilai kesan operasi komposisi nafta yang berbeza terhadap hasil etilena. Hasil propilena, aliran penunu perapian, dan aliran nafta secara konsisten menunjukkan hubungan signifikan terhadap hasil etilena semasa analisis tindak balas permukaan dijalankan, dengan julat faktor interaksi masing-masing dalam -10.07 hingga 192.3, -0.001698 hingga 0.01938, dan -2.383 hingga 820. Model akhir berjaya diwujudkan dalam bentuk model persamaan kuadratik dengan interaksi 2 hala pada kandungan parafin tinggi, dan hubungan linear pada kandungan parafin rendah selepas pengesahan model dijalankan melalui plot kebarangkalian, plot serakan, dan ujian hipotesis Mann-Whitney. Hasil etilena maksimum yang dijana daripada tindak balas pengoptimum diperhatikan tidak sama pada kadar 31.46 - 34.97 % untuk kandungan parafin dalam nafta yang berbeza, dengan bacaan tertinggi dicatatkan oleh nafta yang mempunyai kandungan parafin tertinggi. Hasil terbaik etilena pada kandungan parafin tertinggi iaitu 70.73 vol % dengan mengambil kira kos pengeluaran dikenal pasti pada kadar 34.41 – 34.97 %, menggunakan julat proses yang disyorkan pada 12.22 - 12.25 % hasil propilena, 11033.90 - 11816.40 kg/j aliran penunu perapian, 66.67 – 67.05 t/j aliran nafta dan 816.38 °C COT. Adalah disyorkan kepada loji berskala besar lain untuk menggunakan metodologi yang telah terbukti berjaya dalam kajian ini, untuk pemantauan dan pengoptimuman proses.

TABLE OF CONTENTS

TITLE

DE	CLARATION	iii
DE	DICATION	iv
AC	KNOWLEDGEMENT	v
AB	STRACT	vi
AB	STRAK	vii
TA	BLE OF CONTENTS	viii
LIS	T OF TABLES	XV
LIS	T OF FIGURES	xvii
LIS	T OF SYMBOLS	xxii
LIS	T OF ABBREVIATIONS	xxiv
LIS	T OF APPENDICES	xxviii
CHAPTER 1	INTRODUCTION	1
1.1	Preface	1
1.2	Problem Statement	4
1.3	Objective of Study	5
1.4	Scope of Study	5
1.5	Significance of Study	7
1.6	Thesis Outline	9
CHAPTER 2	LITERATURE REVIEW	11
2.1	Introduction	11
2.2	Olefins in Oil and Gas Industry	11
	2.2.1 Ethylene and Propylene Generation	12
	2.2.2 Olefins Production Technology	13
	2.2.3 Feedstock Selection to Produce Olefins	15
2.3	Steam Cracker Furnace	16

	2.3.1	Steam C	racking	18
	2.3.2	Reaction	Mechanism	19
	2.3.3	Coke Fo	rmation in Steam Cracker Furnace	23
	2.3.4	Steam C	racker Furnace Configuration	25
		2.3.4.1	Radiation Section	27
		2.3.4.2	Convection Section	30
		2.3.4.3	Draft Section	31
	2.3.5	Impact c	of Operating Variables	31
		2.3.5.1	Effect of Temperature	32
		2.3.5.2	Effect of Steam Amount	35
		2.3.5.3	Effect of Feed Compositions	37
2.4	Statist	tical Analy	/sis	39
	2.4.1	Applicat cox Trar	ion for Non-normal Data using Box- sformation Approach	40
	2.4.2	Surface	Response Analysis	41
2.5	Olefin	ns Yield Ir	nprovement	42
	2.5.1	Simulati	on-Based Software	43
	2.5.2	Statistica	al-Based Software	44
	2.5.3	Improve	ment Study at Steam Cracker Furnace	46
2.6	Summ	nary		56
CHAPTER 3	RESI	EARCH N	METHODOLOGY	57
3.1	Introd	luction		57
3.2	Frame	ework of S	btudy	57
	3.2.1	The Stud	lied Plant Operation	59
	3.2.2	The Stud	lied SRT VII Operation	62
		3.2.2.1	Determination of Controlled Operating Parameters	66
		3.2.2.2	Naphtha Feedstock Composition During Study	67
	3.2.3	Data Ext cox	raction and Transformation using Box-	69
	3.2.4	Outline	of Data Analysis	69

3.3	Analysis of	Significance and Interaction Factors	71
	3.3.1 Stab	bility Test	71
	3.3.2 Nor	mality Test	72
	3.3.3 Data	a Clearance	72
	3.3.4 ANG Fact	OVA for Significance and Interaction	72
3.4	Establishme	ent of Equation Model and Relationship	73
	3.4.1 Vari Ana	iable Elimination in Surface Response lysis	73
	3.4.2 Fina	l Equation Model	74
	3.4.3 Mod	del Validation via Existing and New Data	75
3.5	Investigatic Ethylene Yi	on of Different Paraffins Content towards ield	77
3.6	Evaluation Production	of Ethylene Yield with Consideration to Cost	78
3.7	Summary		78
CHAPTER 4	INTERAC MODEL	CTION FACTOR AND EQUATION	81
4.1	Introduction	n	81
4.2	Naphtha Fe	edstock Case 1	81
	4.2.1 Ana Fact	lysis of Significance and Interaction for Naphtha Feedstock Case 1	82
	4.2.	1.1 Stability Test using Boxplot for Data Validation Case 1	82
	4.2.	1.2 Stability Test using I-MR Chart for Data Validation Case 1	84
	4.2.	1.3 Stability Test using Run Chart for Data Validation Case 1	87
	4.2.	1.4 Normality Test using Probability Plot for Data Validation Case 1	91
	4.2.	1.5 Normality Test using Graphical Summary for Data Validation Case 1	93
	4.2.	1.6 Data Clearance for Case 1	97
	4.2.	1.7 Significant Variable with Factorial for Case 1	98

4.2.2	Establish Validatio	ment of Equation Model with n for Naphtha Feedstock Case 1	99
	4.2.2.1	Initial Surface Response Analysis for Case 1	100
	4.2.2.2	Variable Elimination from First Cycle of Surface Response Analysis for Case 1	100
	4.2.2.3	Variable Elimination from Next Cycle of Surface Response Analysis for Case 1	106
	4.2.2.4	Final Equation Model for Case 1	110
	4.2.2.5	Model Validation using Existing Data for Case 1	113
	4.2.2.6	Model Validation using New Data Generation for Case 1	117
Napht	ha Feedsto	ock Case 2	118
4.3.1	Analysis Factors fo	of Significance and Interaction or Naphtha Feedstock Case 2	119
	4.3.1.1	Stability Test using Boxplot for Data Validation Case 2	119
	4.3.1.2	Stability Test using I-MR Chart for Data Validation Case 2	121
	4.3.1.3	Stability Test using Run Chart for Data Validation Case 2	124
	4.3.1.4	Normality Test using Probability Plot for Data Validation Case 2	128
	4.3.1.5	Normality Test using Graphical Summary for Data Validation	130
	4316	Data Clearance for Case 2	133
	4.3.1.7	Significant Variable with Factorial for Case 2	133
4.3.2	Establish Validatio	ment of Equation Model with n for Naphtha Feedstock Case 2	136
	4.3.2.1	Initial Surface Response Analysis for Case 2	136
	4.3.2.2	Variable Elimination from First Cycle of Surface Response Analysis for Case 2	137

4.3

		4.3.2.3	Variable Elimination from Next Cycle of Surface Response Analysis for Case 2	142
		4.3.2.4	Final Equation Model for Case 2	147
		4.3.2.5	Model Validation using Existing Data for Case 2	150
		4.3.2.6	Model Validation using New Data Generation for Case 2	153
4.4	Napht	ha Feedsto	ock Case 3	154
	4.4.1	Analysis Factors fo	of Significance and Interaction or Naphtha Feedstock Case 3	155
		4.4.1.1	Stability Test using Boxplot for Data Validation Case 3	155
		4.4.1.2	Stability Test using I-MR Chart for Data Validation Case 3	158
		4.4.1.3	Stability Test using Run Chart for Data Validation Case 3	161
		4.4.1.4	Normality Test using Probability Plot for Data Validation Case 3	165
		4.4.1.5	Normality Test using Graphical Summary for Data Validation Case 3	166
		4.4.1.6	Data Clearance for Case 3	169
		4.4.1.7	Significant Variable with Factorial for Case 3	170
	4.4.2	Establish Validatio	ment of Equation Model with on for Naphtha Feedstock Case 3	172
		4.4.2.1	Initial Surface Response Analysis for Case 3	172
		4.4.2.2	Variable Elimination from First Cycle of Surface Response Analysis for Case 3	173
		4.4.2.3	Variable Elimination from Next Cycle of Surface Response Analysis for Case 3	178
		4.4.2.4	Final Equation Model for Case 3	183
		4.4.2.5	Model Validation using Existing Data for Case 3	186

		4.4.2.6	Model Validation using New Data Generation for Case 3	189
4.5	Summ Cases	nary of M	Iodel Establishment for All Studied	190
CHAPTER 5	ETH MOI	YLENE Y DEL	IELD FROM FINAL EQUATION	191
5.1	Introd	uction		191
5.2	Invest Paraff	igation c ins Conter	of Comparison between Different nt towards Ethylene Yield	191
	5.2.1	Relations for Naph	ship of Identified Significant Variables tha Feedstock Case 1	192
		5.2.1.1	Interaction Plot for Case 1	192
		5.2.1.2	Response Optimizer for Case 1	193
		5.2.1.3	Surface Plot for Case 1	195
	5.2.2	Relations for Naph	ship of Identified Significant Variables tha Feedstock Case 2	198
		5.2.2.1	Interaction Plot for Case 2	198
		5.2.2.2	Response Optimizer for Case 2	199
		5.2.2.3	Surface Plot for Case 2	201
	5.2.3	Relations for Naph	ship of Identified Significant Variables tha Feedstock Case 3	203
		5.2.3.1	Interaction Plot for Case 3	204
		5.2.3.2	Response Optimizer for Case 3	205
		5.2.3.3	Surface Plot for Case 3	206
	5.2.4	Ethylene	Yield Maximization	208
	5.2.5	Impact of	f Ethylene Yield on Propylene Yield.	212
	5.2.6	Summary Feedstoc	y of Impact at Different Naphtha k Compositions on Ethylene Yield	214
5.3	Evalu Produ	ation of E ction Cost	Ethylene Yield with Consideration to for Significant Variables	214
	5.3.1	Ethylene Productio	Yield with Consideration to on Cost for Naphtha Feedstock Case 1	215
	5.3.2	Ethylene Productio	Yield with Consideration to on Cost for Naphtha Feedstock Case 2	219

	5.3.3 Ethylene Yield with Consideration to Production Cost for Naphtha Feedstock Case 3	222
	5.3.4 Effective Process Condition to Enhance Profits	226
5.4	Summary of Ethylene Yield at Different Naphtha Feedstock Compositions	227
CHAPTER 6	CONCLUSIONS AND RECOMMENDATIONS	229
6.1	Conclusions	229
6.2	Recommendations	231
REFERENCES		235
LIST OF PUBLI	CATIONS	347

LIST OF TABLES

TABLE NO.	TITLE	PAGE
Table 2.1	The product yield from different naphtha feedstock compositions	38
Table 2.2	Available simulation software for steam cracker furnace simulation	43
Table 2.3	Available statistical software potentially be used for olefin yield study using surface response analysis at steam cracker furnace	45
Table 2.4	Previous improvement and optimization studies using simulation and statistical software at the steam cracker furnace	48
Table 3.1	SRT VII specification at the studied plant	65
Table 3.2	The studied variables with process ranges for the analysis	66
Table 3.3	Naphtha feedstock specification during the study	68
Table 3.4	Associated costs utilized for the studied variables	78
Table 4.1	ANOVA for the final equation model for naphtha feedstock Case 1	98
Table 4.2	Model summary for naphtha feedstock Case 1	111
Table 4.3	ANOVA for the final equation model for naphtha feedstock Case 2	135
Table 4.4	Model summary for naphtha feedstock Case 2	148
Table 4.5	ANOVA for the final equation model for naphtha feedstock Case 3	171
Table 4.6	Model summary for naphtha feedstock Case 3	184
Table 5.1	Multiple response prediction of X1 (ethylene yield) for Case 1	194
Table 5.2	Multiple response prediction of X1 (ethylene yield) for Case 2	200
Table 5.3	Multiple response prediction of X1 (ethylene yield) for Case 3	205
Table 5.4	Summary of maximum X1 (ethylene yield) with the relationship of identified significant variables established	

	from final equation models at different naphtha feedstock compositions	209
Table 5.5	The options of process condition to achieve maximum X1 (ethylene yield) using composite desirability calculation from Response Optimizer for Case 1	215
Table 5.6	X1 (ethylene yield) with consideration to the production cost for naphtha feedstock Case 1	218
Table 5.7	The options of process condition to achieve maximum X1 (ethylene yield) using composite desirability calculation from Response Optimizer for Case 2	219
Table 5.8	X1 (ethylene yield) with consideration to the production cost for naphtha feedstock Case 2	221
Table 5.9	The options of process condition to achieve maximum X1 (ethylene yield) using composite desirability calculation from Response Optimizer for Case 3	223
Table 5.10	X1 (ethylene yield) with consideration to the production cost for naphtha feedstock Case 3	225
Table 5.11	Summary of process conditions to achieve maximized profit for Case 1, Case 2, and Case 3	226

LIST OF FIGURES

FIGURE NO	TITLE	PAGE
Figure 1.1	Configuration of SRT VII in the studied plant	3
Figure 2.1	Various technologies and production routes for the olefins using different feedstocks	14
Figure 2.2	Process Flow Diagram for the typical naphtha cracking process in the olefin plant	17
Figure 2.3	The main reactions that occur during the steam cracking of higher alkanes (paraffins)	22
Figure 2.4	Mechanism of coke and aromatic formation as a side product in steam cracking	24
Figure 2.5	The steam cracker furnace configuration in a typical olefin	26
Figure 2.6	A transfer line exchanger with its configuration	29
Figure 2.7	Light olefin yield plotted with the cracking temperature	33
Figure 2.8	The effect of COT on the product yields in the steam cracking process	34
Figure 2.9	The light olefin yield from the steam to oil ratio utilized for the study	36
Figure 2.10	Graphical representation between factors and surface response analysis	41
Figure 3.1	Research framework	58
Figure 3.2	Process Flow Diagram (PFD) at the studied plant	60
Figure 3.3	The SRT furnaces in the studied plant	62
Figure 3.4	Location of the selected controlled variables for the study	63
Figure 3.5	Outline procedures of data analysis at the studied SRT VII	70
Figure 4.1	Boxplot for the studied variables; (a) A1 (propylene yield), (b) B (hearth burner flow), (c) C (integral burner flow), (d) D (dilution steam flow), (e) E (naphtha feed flow), and (f) F (COT) for naphtha feedstock Case 1	83
Figure 4.2	I-MR Chart for the studied variables; (a) A1 (propylene yield), (b) B (hearth burner flow), (c) C (integral burner flow), (d) D (dilution steam flow), (e) E (naphtha feed flow), and (f) F (COT) for naphtha feedstock Case 1	86

Figure 4.3	Run Chart for the studied variables; (a) A1 (propylene yield), (b) B (hearth burner flow), (c) C (integral burner flow), (d) D (dilution steam flow), (e) E (naphtha feed flow), and (f) F (COT) for naphtha feedstock Case 1	90
Figure 4.4	Probability Plot for the for the studied variables; (a) A1 (propylene yield), (b) B (hearth burner flow), (c) C (integral burner flow), (d) D (dilution steam flow), (e) E (naphtha feed flow), and (f) F (COT) for naphtha feedstock Case 1	92
Figure 4.5	Summary Report for the studied variables; (a) A1 (propylene yield), (b) B (hearth burner flow), (c) C (integral burner flow), (d) D (dilution steam flow), (e) E (naphtha feed flow), and (f) F (COT) for naphtha feedstock Case 1	96
Figure 4.6	Improved Run Chart with the removed abnormal data for D (dilution steam flow) for naphtha feedstock Case 1	97
Figure 4.7	Pareto Chart of final equation model for naphtha feedstock Case 1	112
Figure 4.8	Residuals Plots of X1 (ethylene yield) from final equation model for naphtha feedstock Case 1	113
Figure 4.9	Probability Plot of residuals from final equation model for naphtha feedstock Case 1	114
Figure 4.10	Mann-Whitney test using existing data for Case 1	115
Figure 4.11	Scatterplot for X1 (ethylene yield) for the actual data versus the calculated value in BFIT1 for naphtha feedstock Case 1 with the R-squared at 94.3%	116
Figure 4.12	Final equation model validation using new 1.5 days of actual plant data for naphtha feedstock Case 1	117
Figure 4.13	Boxplot for the studied variables; (a) A1 (propylene yield), (b) B (hearth burner flow), (c) C (integral burner flow), (d) D (dilution steam flow), (e) E (naphtha feed flow), and (f) F (COT) for naphtha feedstock Case 2	120
Figure 4.14	I-MR Chart for for the studied variables; (a) A1 (propylene yield), (b) B (hearth burner flow), (c) C (integral burner flow), (d) D (dilution steam flow), (e) E (naphtha feed flow), and (f) F (COT) for naphtha feedstock Case 2	123
Figure 4.15	Run Chart for for for the studied variables; (a) A1 (propylene yield), (b) B (hearth burner flow), (c) C (integral burner flow), (d) D (dilution steam flow, (e) E (naphtha feed flow), and (f) F (COT) for naphtha feedstock Case 2	127
Figure 4.16	Probability Plot for the studied variables; (a) A1 (propylene yield), (b) B (hearth burner flow), (c) C (integral burner	

	flow), (d) D (dilution steam flow), (e) E (naphtha feed flow), and (f) F (COT) for naphtha feedstock Case 2	129
Figure 4.17	Summary Report for the studied variables; (a) A1 (propylene yield), (b) B (hearth burner flow), (c) C (integral burner flow), (d) D (dilution steam flow), (e) E (naphtha feed flow), and (f) F (COT) for naphtha feedstock Case 2	132
Figure 4.18	Revised Run Chart with the removed abnormal data for F (COT) for naphtha feedstock Case 2	134
Figure 4.19	Pareto Chart of final equation model for naphtha feedstock Case 2	149
Figure 4.20	Residuals Plots of X1 (ethylene yield) from final equation model for naphtha feedstock Case 2	149
Figure 4.21	Probability Plot of residuals from final equation model for naphtha feedstock Case 2	150
Figure 4.22	Mann-Whitney test using existing data for Case 2	151
Figure 4.23	Scatterplot for X1 (ethylene yield) for the actual data versus the calculated value in BFIT1 for naphtha feedstock Case 2 with the R-squared at 75.8%	152
Figure 4.24	Final equation model validation using new 1.5 days of actual plant data for naphtha feedstock Case 2	153
Figure 4.25	Boxplot for the studied variables; (a) A1 (propylene yield), (b) B (hearth burner flow), (c) C (integral burner flow), (d) D (dilution steam flow), (e) E (naphtha feed flow), and (f) F (COT) for naphtha feedstock Case 3	156
Figure 4.26	I-MR Chart for the studied variables; (a) A1 (propylene yield), (b) B (hearth burner flow), (c) C (integral burner flow), (d) D (dilution steam flow), (e) E (naphtha feed flow), and (f) F (COT) for naphtha feedstock Case 3	160
Figure 4.27	Run Chart for the studied variables; (a) A1 (propylene yield), (b) B (hearth burner flow), (c) C (integral burner flow), (d) D (dilution steam flow), (e) E (naphtha feed flow), and (f) F (COT) for naphtha feedstock Case 3	163
Figure 4.28	Probability Plot for the studied variables; (a) A1 (propylene yield), (b) B (hearth burner flow), (c) C (integral burner flow), (d) D (dilution steam flow, (e) E (naphtha feed flow), and (f) F (COT) for naphtha feedstock Case 3	166
Figure 4.29	Summary Report for the studied variables; (a) A1 (propylene yield), (b) B (hearth burner flow), (c) C (integral burner flow), (d) D (dilution steam flow, (e) E (naphtha feed flow), and (f) F (COT) for naphtha feedstock Case 3	169

Figure 4.30	Revised Run Chart with the removed abnormal data for F (COT) for naphtha feedstock Case 3	170
Figure 4.31	Pareto Chart of final equation model for naphtha feedstock Case 3	185
Figure 4.32	Residuals Plots of X1 (ethylene yield) from final equation model for naphtha feedstock Case 3	185
Figure 4.33	Probability Plot of residuals from final equation model for naphtha feedstock Case 3	187
Figure 4.34	Mann-Whitney test using existing data for Case 3	187
Figure 4.35	Scatterplot for X1 (ethylene yield) for the actual data versus the calculated value in BFIT1 for naphtha feedstock Case 3 with the R-Squared at 87.6%	188
Figure 4.36	Final equation model validation using new 1.5 days of actual plant data for naphtha feedstock Case 3	189
Figure 5.1	Interaction Plot of X1 (ethylene yield) established from final surface response analysis for Case 1	193
Figure 5.2	Response Optimizer with the process condition to achieve maximized X1 (ethylene yield) established from final surface response analysis for naphtha feedstock Case 1	194
Figure 5.3	Surface Plot established from the final surface response analysis and the reference line to achieve the maximized X1 (ethylene yield) at 34.9676 % identified from Response Optimizer for (a) A1 (propylene yield) versus B (integral burner flow), (b) A1 (propylene yield) versus E (naphtha feed flow), (c) A1 (propylene yield) versus F (COT), (d) B (hearth burner flow) versus E (naphtha feed flow), (e) B (hearth burner flow) versus F (COT), and (f) E (naphtha feed flow) versus F (COT), for naphtha feedstock Case 1	196
Figure 5.4	Interaction Plot of X1 (ethylene yield) established from final surface response analysis for Case 2	199
Figure 5.5	Response Optimizer with the process condition to achieve maximized X1 (ethylene yield) established from final surface response analysis for naphtha feedstock Case 2	200
Figure 5.6	Surface Plot established from the final surface response analysis and the reference line to achieve the maximized X1 (ethylene yield) at 33.1706 % identified from Response Optimizer for (a) A1 (propylene yield) versus B (hearth burner flow), (b) A1 (propylene yield) versus C (integral burner flow), (c) A1 (propylene yield) versus E (naphtha feed flow), (d) B (hearth burner flow) versus C (integral burner flow), (e) B (hearth burner flow) versus E (naphtha	

	feed flow), and (f) C (integral burner flow) versus E (naphtha feed flow), for naphtha feedstock Case 2.	202
Figure 5.7	Interaction Plot of X1 (ethylene yield) established from final surface response analysis for Case 3	204
Figure 5.8	Response Optimizer with the process condition to achieve maximized X1 (ethylene yield) established from final surface response analysis for naphtha feedstock Case 3	205
Figure 5.9	Surface Plot established from the final surface response analysis and the reference line to achieve the maximized X1 (ethylene yield) at 31.4605 % identified from Response Optimizer for (a) A1 (propylene yield) versus B (hearth burner flow), (b) A1 (propylene yield) versus C (integral burner flow), (c) A1 (propylene yield) versus E (naphtha feed flow), (d) B (hearth burner flow) versus C (integral burner flow), (e) B (hearth burner flow) versus E (naphtha feed flow), (a) C (integral burner flow) versus E (naphtha	
	(naphtha feed flow), for naphtha feedstock Case 3	207

LIST OF SYMBOLS

6σ	-	Six-sigma
λ	-	Lambda (for Box-cox transformation)
A1	-	Propylene yield
В	-	Hearth burner flow
BCRE1	-	Storage in Minitab (transformed value of response)
BFIT1	-	Storage in Minitab (fitted value of original response)
С	-	Integral burner flow
C1	-	Carbon group 1
C2	-	Carbon group 2
C3	-	Carbon group 3
C4	-	Carbon group 4
C5	-	Carbon group 5
C_nH_{2n}	-	Standard formulae for alkenes (olefins)
$C_nH_{2n+2} \\$	-	Standard formulae for alkanes (paraffins)
C-C	-	Carbon to carbon bond
C=C	-	Carbon double bond
С-Н	-	Carbon to hydrogen bond
COEF1	-	Storage in Minitab (coefficient)
D	-	Dilution steam flow
DFIT1	-	Storage in Minitab (for DFIT)
E	-	Naphtha feed flow
F	-	Coil outlet temperature
FITS	-	Storage in Minitab (fitted values)
g	-	Geometric means (for Box-cox transformation)
H_0	-	Null hypothesis
H_1	-	Alternative hypothesis
Ν	-	Number of tests
RESI1	-	Storage in Minitab (residuals)
X1	-	Ethylene yield
vol %	-	Volume percent

wt % - Weight percent

LIST OF ABBREVIATIONS

2D	-	2-Dimensional
3D	-	3-Dimensional
AGO	-	Atmospheric Gas Oil
AHP	-	Analytic Hierarchy Process
AL	-	Arabian Light
ANN	-	Artificial Neural Network
ANOVA	-	Analysis of Variance
API	-	American Petroleum Institute
ASL	-	Arab Super Light
AXL	-	Arab Extra Light
BATH	-	Bio-acid Acetone to Hydrocarbon
BDU	-	Butadiene Unit
BFW	-	Boiler Feed Water
BFWP	-	Boiler Feed Water Preheat
CAPEX	-	Capital Expenditure
CC	-	Catalytic Cracking
CCD	-	Central Composite Design
CFD	-	Computational Fluid Dynamics
CGC	-	Charge Gas Compressor
CI	-	Confidence Interval
СОР	-	Coil Outlet Pressure
СОТ	-	Coil Outlet Temperature
DCC	-	Deep Catalytic Cracking
DCS	-	Distributed Control System
DEACM	-	Data Envelopment Analysis Cross Model
DF	-	Degree of Freedom
DH	-	De-hydration Process
DoE	-	Design of Experiment
DOX	-	Disperse Oil Extraction
DS	-	Dilution Steam

DSSH	-	Dilution Steam Superheat
DV	-	Decoke Valve
EBST	-	Ethane Buffer Storage Tank
EOEG	-	Ethylene Oxide Ethylene Glycol
EOR	-	End of Run
EPC	-	Engineering, Procurement, Construction
EPCC	-	Engineering, Procurement, Construction, and Commissioning
FCC	-	Fluidized Catalytic Cracking
FP	-	Flash Pyrolysis
FPH	-	Feed Preheat
FT	-	Fischer-Tropsch Synthesis
GA	-	Genetic Algorithm
GF	-	Gasoline Fractionator
GS	-	Gas Stream (Reactor Technology)
GSO	-	Group Search Optimization
HC	-	Hydrocarbon
HCR	-	Hydrocracker Residue
HDoE	-	Historical Design of Experiment
HEx	-	Heat Exchanger
HG	-	Hydrogenation
HHP	-	High Pressure (Super High Pressure)
HP	-	Hydro-pyrolysis (Specific to Figure 2.1)
HP	-	High Pressure
HPS	-	High Pressure Steam
HTUL	-	Hydro-thermal Upgrading Liquefaction
HVGO	-	Heavy Vacuum Gas Oil
IDF	-	Induced Draft Fan
I-MR	-	Individual – Moving Range
KTA	-	Kilo Tonne Per Annum
KTI	-	Kinetics Technology International
LCL	-	Lower Control Limit
LDV	-	Large Decoke Valve
LFPH	-	Lower Feed Preheat

LMP	-	Lower Mix Preheat
LPG	-	Liquefied Petroleum Gas
LSSH	-	Lower Steam Superheat
LSSVM	-	Least Square Support Vector Machine
LTHT	-	Lummus Technology Heat Transfer
MAPD	-	Methyl Acetylene Propadiene
MC	-	Management Committee
MINLP	-	Mixed-Integer Nonlinear Programming
MOC	-	Management of Change
MOO	-	Multi-Objective Optimization
MS	-	Mean of Square
MSSH	-	Medium Steam Superheat
MTO	-	Methanol to Olefins
MW	-	Molecular Weight
NBST	-	Naphtha Buffer Storage Tank
NLP	-	Nonlinear Programming
OA	-	Outer Approximation
OC	-	Oxidative Coupling
OM	-	Olefin Metathesis
OPEX	-	Operating Expenditure
OU	-	Olefins Upgrading
PD	-	Propane Dehydrogenation
PFD	-	Process Flow Diagram
PFO	-	Pyrolysis Fuel Oil
PGO	-	Pyrolysis Gas Oil
PI	-	Predictive Interval
PIMS	-	Process Information Management System
PIONA	-	n-Paraffins, Isoparaffins, Olefins, Naphthenes, Aromatics
PONA	-	Paraffins, Olefins, Naphthenes, Aromatics
PSA	-	Pressure Swing Adsorption
PSO	-	Particle Swarm Optimization
PWS	-	Process Water Stripper
QT	-	Quench Tower

RFCC	-	Residual Fluidized Catalytic Cracking
ROG	-	Refinery Off-Gas
RSM	-	Response Surface Methodology
RVP	-	Reid Vapor Pressure
SAS	-	Statistical Analysis Software
SC	-	Steam Cracking
SDV	-	Small Decoke Valve
SE	-	Standard Error
SH	-	Super High
SHPS	-	Super High Pressure Steam
SQP	-	Sequential Quadratic Programming
SR	-	Steam Reforming
SRN	-	Straight Run Naphtha
SRT	-	Short Residence Time
SS	-	Sum of Square
SSH	-	Steam Superheat
SSPS	-	Statistical Package for Social Sciences
TBP	-	True Boiling Point
TLE	-	Transfer Line Exchanger
TLV	-	Transfer Line Valve
TMT	-	Tube Metal Temperature
UCL	-	Upper Control Limit
UMP	-	Upper Mix Preheat
USA	-	United State of America
USD	-	United State Dollar
USSH	-	Upper Steam Superheat
UTM	-	Universiti Teknologi Malaysia
UTP	-	Universiti Teknologi PETRONAS
VFD	-	Variable Frequency Drive
VIF	-	Variance Inflation Factor
WRGO	-	Wide Range Gas Oil

LIST OF APPENDICES

APPENDIX	TITLE	PAGE
Appendix A	Coded Coefficients for Surface Response Analysis Case 1	257
Appendix B	Coded Coefficients for Surface Response Analysis Case 2	277
Appendix C	Coded Coefficients for Surface Response Analysis Case 3	297
Appendix D	Data from Final Surface Response Analysis for Naphtha Feedstock Case 1	319
Appendix E	Data from Final Surface Response Analysis for Naphtha Feedstock Case 2	326
Appendix F	Data from Final Surface Response Analysis for Naphtha Feedstock Case 3	333
Appendix G	Validation of Final Equation Model using New Data for Naphtha Feedstock Case 1	339
Appendix H	Validation of Final Equation Model using New Data for Naphtha Feedstock Case 2	341
Appendix I	Validation of Final Equation Model using New Data for Naphtha Feedstock Case 3	343
Appendix J	Summary of Overall Findings for the Study	345

CHAPTER 1

INTRODUCTION

1.1 Preface

Olefins are unsaturated hydrocarbons comprised of a single double bond with a chemical formula of C_nH_{2n} . They are one of the most important chemicals in the petrochemical industry that form the basis for most of the essential applications such as pharmaceuticals, insulation, plastic, cosmetics product, and synthetic fiber (Bender, 2014; Fakhroleslam & Sadrameli, 2020; Rahimi & Karimzadeh, 2011; Sadrameli, 2015, 2016; Zhu et al., 2017).

Olefin plants utilizing steam cracking in the steam cracker furnace are often defined as the core of the petrochemical industry (Fan et al., 2015; Nikolaidis et al., 2018). Its plant performance is important in indicating the development of the chemical industry in the country and region (Gong et al., 2017) due to its significant contribution to the industry.

Ethylene and propylene are examples of the most important olefins produced widely in the olefin plants to meet the demands in the petrochemical industry. Their yield monitoring from the steam cracker furnace is therefore essential as their values translate to the profit generation and sustainability of the olefin plant, especially to the steam cracker furnace performance (Díaz & Bandoni, 1996; Gholami et al., 2021; Khor et al., 2014; Lashkajani et al., 2016; Leo et al., 2018; Luo et al., 2015; Petracci et al., 1993; Ruckaert et al., 1978).

Previous studies had been established to utilize various process simulations to improve the olefins' yield (Geng et al., 2016; Leo et al., 2018; Song & Tang, 2018; Yu et al., 2018). These simulation studies were successfully developed and significantly improve the olefins' yield by controlling various controlled variables in the olefin

process. The effort to enhance the olefin process continues, targeting various operating parameters for steam cracking utilizing gaseous and liquid feedstocks. Most of the studies were carried out using olefin simulation software, with some being verified at the lab or small pilot plant.

Besides the well-utilized simulation software for olefin process evaluation and improvement, there is also a bright prospect to utilize statistical software that was proven robust such as Minitab (Arminian & Ozgur, 2020; Martin & Roberts, 1996; Ozgur, 2019), Stata (Fuad et al., 2015; Shim et al., 2016), SSPS (Cuesta-Lozano et al., 2020; Ozgur, 2019; Sahud Alotaibi, 2020), SAS (Gunst, 2012; Ozgur, 2019; Sullivan & Greenland, 2012), and GraphPad Prism (Mavrevski et al., 2018; Mitteer et al., 2019) as they seem more practical to be practiced by Operation personnel compared to the complex and expensive simulation software equipped by Olefin Licensors that come with a high price and restricted access to safeguard the proprietary design knowledge by Olefin Licensors.

ABB Lummus, KTI-Technip, Linde AG (Pyrocrack), Stone and Webster, M.W. Kellogg (Brayden et al., 2008; Sadrameli, 2015), and Sinopec (Pu & Shi, 2013; Wang et al., 2021) are examples of the established Olefin Licensors which having mature, reliable, and proven olefin technologies. There are also various available simulation software from Olefins Licensors and simulation companies that is currently available such as SPYRO (Dente et al., 1979; Van Goethem et al., 2001), SHAHAB (Toufighi et al., 2004), CRACKER (Joo et al., 2000; Joo & Park, 2001), CHEMKIN (Kee et al., 2006; Reyniers et al., 2017; Van Cauwenberge et al., 2017) and CRACKSIM (Hillewaert et al., 1988; Van Geem et al., 2008; Willems & Froment, 1988).

Surface response analysis is an example of established mathematical and statistical approaches used for modeling and analyzing a process (Haladu et al., 2022; Montgomery, 2017) that may utilize the available statistical software in the market. In surface response analysis, the response of interest is affected by the number of variables (Braimah et al., 2016) in multi-relations. The recent studies conducted for various normal type furnaces utilizing surface response analysis (Ganesh et al., 2018; Sun, Yang, et al., 2016; Sun, Zhang, et al., 2016) had also successfully improved the

related process. However, the surface response implementation towards the ethylene yield from special type furnaces, such as steam cracker furnace in the actual fluctuating large scale olefin plant with non-normal data, is not found in the literature.

This study was conducted with a focus on ethylene yield at the newly commissioned olefin plant with the design capacity to produce 1,100 KTA of polymer grade ethylene from the steam cracker furnace, Short Residence Time (SRT) VII type. The feedstock used in the studied plant is a straight run naphtha (SRN) from the upstream plant and is cracked in the SRT VII with the operating tube metal temperature (TMT) of 1,050 °C - 1,180 °C. Conducting the study in actual olefin plant conditions is challenging, resulting from various process fluctuations (Feli et al., 2017; Lin et al., 2021; Zakria, 2018; Zakria et al., 2016) and variations in the upstream process, downstream readiness, utility shortage, and frequent changes in the feedstock compositions. Figure 1.1 shows the configuration of the SRT VII furnace with its support auxiliaries at the studied plant.



Figure 1.1 Configuration of SRT VII in the studied plant

This study is essential as a baseline and provides the guidelines to the Operations personnel in the actual olefin plant to systematically monitor and improve the ethylene yield. It adopts practical statistical software from both academic and industrial perspectives, providing a reliable alternative to the simulation software offered by Olefin Licensors.

1.2 Problem Statement

Surface response analysis is the established statistical multi-response analysis proven successful in improving the process (Ganesh et al., 2018; Sun, Yang, et al., 2016; Sun, Zhang, et al., 2016). However, it is not widely discussed at the large scale olefin process as the olefin plant often relies on the Olefin Licensors to run their proprietary simulation software should there is any process upset. This causes process troubleshooting and evaluation to become longer and more challenging, especially for the fluctuating process with non-normal data that is frequently observed at the large scale olefin plant (Feli et al., 2017; Lin et al., 2021; Zakria, 2018; Zakria et al., 2016). No surface response analysis study was found from the literature on the steam cracker furnace in the large scale olefin plants to find various variables' significant and interaction factors towards the ethylene yield focusing on non-normal data from fluctuating plant conditions. Besides, the output provided from Olefins Licensor's simulation software is in the form of a few operating cases which require further interpretation, without a specific linear or quadratic equation model with coefficient factors that can directly correlate the relationship of controlling variables for faster decision by Operations personnel. In addition, due to the dynamics of naphtha feedstock compositions in the olefin plant, a fast but reliable analysis tool to handle the evaluation of different naphtha feedstock compositions, especially on the paraffins content towards ethylene yield, is also required. Furthermore, previous studies on improving the olefins process mostly focused on increasing the olefin yield (Berreni & Wang, 2011; Khor et al., 2014; Riverol & Pilipovik, 2007; Song & Tang, 2018) via increasing the Coil Outlet Temperature (COT) with less focus to the operating cost at the large scale application. However, operating a large scale steam cracker furnace at the highest COT is impractical where it may cause a higher fuel gas consumption and,

therefore largely increase the production cost. The practical guideline to be established for Operations personnel at the large scale olefin plant to monitor and improve the ethylene yield effectively and economically.

1.3 Objective of Study

This study focuses on evaluating the ethylene yield from steam cracking using naphtha as a feedstock at the actual plant condition with various process fluctuations. The study is conducted at the newly commissioned steam cracker furnace, SRT VII type, with the objectives as follows:

- To analyze the significance and interaction factors for the controlled variables in the SRT VII furnace.
- To establish the linear or quadratic equation model with two-way interactions between significant variables for the ethylene yield from the actual fluctuating plant condition using surface response analysis.
- iii) To investigate the impact of operating the SRT VII furnace at different naphtha feedstock compositions towards ethylene yield.
- iv) To evaluate the process condition in achieving the best ethylene yield with consideration to the production cost through the proposed operating parameters in Response Optimizer.

1.4 Scope of Study

The scope of study mainly focuses on developing the ethylene yield model using surface response analysis at one dedicated SRT VII furnace in the studied plant. The study also will cover the following scopes:

- i) Analysis of the significance and interaction factors for the selected operating variables towards the ethylene yield. The process range for data analysis was limited to the actual plant operating condition for the studied variables during the study duration with hearth burner flow (10.08 12.19 t/hr), integral burner flow (0.61 0.74 t/hr), dilution steam flow (40.59 40.79 t/hr), naphtha feed flow (61.33 72.89 t/hr), COT $(809.27 816.38 \,^{\circ}\text{C})$ and propylene yield $(10.63 12.25 \,^{\circ}\text{o})$. The analysis proceeded regardless of the non-normal data was extracted from PI Process Book to represent the actual large scale operating plant condition using the Box-cox data transformation approach throughout the data analysis.
- ii) Establishment of final equation model based on the linear or quadratic equation model for the ethylene yield. The data validation for the final model utilized the existing plant data used for the analysis and also the new data. For existing data, the validation covered a full range of ethylene yield using Probability Plot of residuals and Scatterplot with the support of Mann-Whitney hypothesis test for actual versus predicted ethylene yield. For model validation using the new data, the analysis is limited to only 1.5 days (36 hours) of actual plant data, established after the surface response analysis was completed. This was due to the studied SRT VII limitation resulting from coke continuous built-up in the coils which may interfere on the validation of the established model.
- iii) Investigation of ethylene yield from different naphtha feedstock compositions at 3 cases based on the main component, which was paraffins, ranging from 57.6 70.7 vol %. Only paraffins content was studied for comparison as this study did not intend to evaluate the full aspect of naphtha feedstock composition towards ethylene yield. The analysis of the relationship between identified significant variables towards the ethylene yield was presented using the available graphical tools in Minitab, which are Interaction Plot, Surface Plot, and Response

Optimizer. In this step, the maximized ethylene yield comparison between these cases was established using Response Optimizer without considering the production cost.

 iv) Evaluation of ethylene yield considering the production cost, limited to the proposed operating conditions in Response Optimizer. The actual costs for the significant variables were inserted into the evaluation table and calculated based on the previously validated final equation model to see the impact of the production cost from the identified significant variables on the ethylene yield. The range of most profitable process conditions at the studied SRT VII was determined from the evaluation table, limited to the identified significant variables in the validated final equation model.

1.5 Significance of Study

Analyzing the non-normal process data is practical to be applied at the large scale olefin plant due to frequent process fluctuation (Feli et al., 2017; Lin et al., 2021; Zakria, 2018; Zakria et al., 2016) observed at the steam cracker furnace, such as SRT VII. It will also significantly reduce the time from excessive data clearance, which is not practical to be conducted due to time-consuming. Besides, deleting the vast non-normal data to make the data normal may also unnecessarily remove the essential actual process variability at the actual plant condition. Minitab software version 21 can run the analysis to find the significance and interaction factors of the studied variables in the SRT VII directly by transforming the non-normal data using the Box-cox transformation approach (Minitab, 2022).

Besides, the established equation model using surface response analysis will show the impact of each significant variable in the form of the coefficient factors on the linear or quadratic equation model. The coefficient factors on this equation model may guide Operations personnel to focus on which controlled variables to be given closer attention to in generating a better ethylene yield. Besides, the capability of surface response analysis in Minitab to establish the multiple response relationships in the form of the equation model is practical and may help Operations personnel to act fast and avoid unnecessary opportunities lost for the studied plant compared to waiting for the response from Olefins' Licensor on their proprietary simulation result.

Surface response analysis may also investigate the ethylene yield under various conditions, including the different paraffins content in naphtha feedstock using a large amount of data. This flexibility ensures various conditions at the studied steam cracker furnace can be analyzed using the surface response analysis without the need for integration with other methods. Besides, the graphical tools in the Minitab, such as Interaction Plot, Surface Plot, and Response Optimizer, are effective in evaluating the impact of operating the steam cracker furnace at different naphtha feedstock compositions towards the ethylene yield. These graphical representations are also practical for Operation personnel to directly assess the effect of each studied naphtha composition towards the ethylene yield through the graphical representation without the complex evaluation through the calculations and additional detailed analysis.

The evaluation of ethylene yield with consideration to the production cost is important in the actual plant condition. Achieving the maximum olefins yield is usually the goal for any operating olefin plant, as the higher olefins yield typically translates to higher profit generation (Díaz & Bandoni, 1996; Gholami et al., 2021; Khor et al., 2014; Lashkajani et al., 2016; Leo et al., 2018; Luo et al., 2015; Petracci et al., 1993; Ruckaert et al., 1978). The actual large scale olefin plant depends on various operating parameters to remain profitable. Therefore, the operational process evaluation based on profit generation is essential. This study also will evaluate the ethylene yield with consideration to the production cost from the established significant variables using the validated final equation model. This approach is necessary to achieve the best olefins' yield with consideration to the production cost to keep a healthy financial flow to the studied plant.

1.6 Thesis Outline

The thesis is organized into six chapters. Chapter 1 starts with a brief introduction about the olefins, information on the steam cracker furnace, and the background of the surface response analysis. Besides, this chapter also presents the problem statement, objectives, scope of work, and significance of the study. Chapter 2 discusses the detailed literature taken from previous, current, and future works in improving ethylene yield. This chapter presents the information on the olefins production, reaction mechanism, steam cracker furnace, statistical analysis focusing on surface response, and previous olefin yield improvement studies. Chapter 3 describes the detailed research methodology, starting from selecting controlled variables, data transformation from non-normal data, and methods used to illustrate the impact of the process on the production cost. This chapter also discusses the basic procedure for data analysis using surface response analysis with the validation steps for the final equation model. Chapter 4 discusses the detailed analysis results of the study's significant factors and equation model establishment. This chapter critically explains one-by-one variable elimination using surface response analysis for all studied cases: Case 1, Case 2, and Case 3. Chapter 5 focuses on the ethylene yield relationship from the final equation model validated in Chapter 4. It explains the investigation findings on the relationship between the significance variables assisted by graphical tools in Minitab for the maximized and optimized condition (further evaluation considering production cost). Finally, Chapter 6 summarizes the research conclusion and responds to all the objectives planned for the study. It also provides recommendations for future studies on ethylene yield valuation at the end of the chapter.

REFERENCES

- Abbasali, S. M., Farsi, M., & Rahimpour, M. R. (2017). Simulation and Dynamic Optimization of an Industrial Naphtha Thermal Cracking Furnace Based on Time Variant Feeding Policy. *Chemical Product and Process Modeling*, 13. https://doi.org/10.1515/cppm-2017-0032
- Abghari, S., Towfighi, J., Karimzadeh, R., & Omidkhah, M. (2008). Determination of yield distribution in olefin production by thermal cracking of atmospheric gasoil. *Korean Journal of Chemical Engineering*, 25(4), 681-692. https://doi.org/10.1007/s11814-008-0112-4
- Akah, A., & Al-Ghrami, M. (2015). Maximizing propylene production via FCC technology. *Applied Petrochemical Research*, 5(4), 377-392. https://doi.org/10.1007/s13203-015-0104-3
- Akporiaye, D., Jensen, S., Olsbye, U., Rohr, F., Rytter, E., Rønnekleiv, M., & Spjelkavik, A. I. (2001). A Novel, Highly Efficient Catalyst for Propane Dehydrogenation. *Industrial & Engineering Chemistry Research*, 40(22), 4741–4748. https://doi.org/10.1021/ie010299+
- Al-Absi, A. A., Aitani, A. M., & Al-Khattaf, S. S. (2020). Thermal and catalytic cracking of whole crude oils at high severity. *Journal of Analytical and Applied Pyrolysis*, 145, 104705. https://doi.org/10.1016/j.jaap.2019.104705
- Al-Absi, A. A., & Al-Khattaf, S. S. (2018). Conversion of Arabian Light Crude Oil to Light Olefins via Catalytic and Thermal Cracking. *Energy & Fuels*, 32(8), 8705-8714. https://doi.org/10.1021/acs.energyfuels.8b01932
- Albright, L. F., Crynes, B. L., & Corcoran, W. H. (1983). Pyrolysis, theory and industrial practice. Academic Press. https://doi.org/ 10.1002/ aic.690300635
- Albright, L. F., & Marek, J. C. (1988). Mechanistic model for formation of coke in pyrolysis units producing ethylene. *Industrial & Engineering Chemistry Research*, 27(5), 755-759. https://doi.org/10.1021/ie00077a006
- Alotaibi, F. M., González-Cortés, S., Alotibi, M. F., Xiao, T., Al-Megren, H., Yang, G., & Edwards, P. P. (2018). Enhancing the production of light olefins from heavy crude oils: Turning challenges into opportunities. *Catalysis Today*, *317*, 86-98. https://doi.org/10.1016/j.cattod.2018.02.018

Anderson, M. J., & Whitcomb, P. J. (2007). DOE Simplified (2 ed.). Productivity Press.

- Anderson, M. J., & Whitcomb, P. J. (2016). *RSM simplified : optimizing processes using response surface methods for design of experiments.*
- Arminian, A., & Ozgur, C. (2020). Advanced statistical approaches for data analysis by MINITAB: A step-by-step education. *Decision Sciences Journal of Innovative Education*, 18.
- Astruc, D. (2005). The Metathesis Reactions: From a Historical Perspective to Recent Developments. *New Journal of Chemistry*, 29(1), 42-56. https://doi.org/ 10.1039/b412198h
- Atkinson, A. (2020). The box-cox transformation: review and extensions. LSE Research Online Documents on Economics 103537. Retrieved 02/07, from http://eprints.lse.ac.uk/id/eprint/103537
- Atkinson, A., Riani, M., & Corbellini, A. (2021). The Box–Cox Transformation: Review and Extensions. *Statistical Science*, 36(2), 239 - 255. https://doi.org/ 10.1214/20-STS778
- Barazandeh, K., Dehghani, O., Hamidi, M., Aryafard, E., & Rahimpour, M. R. (2015). Investigation of coil outlet temperature effect on the performance of naphtha cracking furnace. *Chemical Engineering Research and Design*, 94, 307-316. https://doi.org/10.1016/j.cherd.2014.08.010
- Barendregt, S., Valkenburg, P. J. M., Wagner, E. S., Dente, M., & Ranzi, E. (2002). History and Recent Developments in SPYRO®, a Review. AIChE Ethylene Producers Conference Proceedings, 11, 30-70.
- Belohlav, Z., Zamostny, P., & Herink, T. (2003). The kinetic model of thermal cracking for olefins production. *Chemical Engineering and Processing: Process Intensification*, 42(6), 461-473. https://doi.org/10.1016/S0255-2701(02)00062-4
- Bender, M. (2014). An Overview of Industrial Processes for the Production of Olefins
 C4 Hydrocarbons. *ChemBioEng Reviews*, 1(4), 136-147. https://doi.org/ 10.1002/cben.201400016
- Berreni, M., & Wang, M. (2011). Modelling and dynamic optimization of thermal cracking of propane for ethylene manufacturing. *Computers & Chemical Engineering*, 35(12), 2876-2885. https://doi.org/10.1016/j.compchemeng. 2011.05.010

- Braimah, M., Anozie, A., & Odejobi, O. (2016). Utilization of Response Surface Methodology (RSM in the Optimization of Crude Oil Refinery Process, New Port Harcourt Refinery, Nigeria. *Journal of Multidisciplinary Engineering, Science and Technology*, 3(3), 4361-4369.
- Brayden, M., Hines, D., Graham, J., & Pickett, T. (2008). Top 5 Contaminants in Ethylene Production Unit Feedstock AIChE 2012 - 2012 AIChE Annual Meeting, New Orleans, Lousiana.
- Broemeling, L. (2014). *Box-Cox Transformation Introduction*. John Wiley & Sons, Inc. https://doi.org/10.1002/9781118445112.stat00404
- Caballero, D. Y., Biegler, L. T., & Guirardello, R. (2015). Simulation and Optimization of the Ethane Cracking Process to Produce Ethylene. *Computer Aided Chemical Engineering*, 37, 917-922. https://doi.org/10.1016/B978-0-444-63578-5.50148-1
- Cai, H., Krzywicki, A., & Oballa, M. C. (2002). Coke formation in steam crackers for ethylene production. *Chemical Engineering and Processing: Process Intensification*, 41(3), 199-214. https://doi.org/10.1016/S02552701(01) 00135-0
- Cai, Y., Yang, S., Fu, S., Zhang, D., & Zhang, Q. (2017). Investigation of Portevin– Le Chatelier Band Strain and Elastic Shrinkage in Al-Based Alloys Associated with Mg Contents. *Journal of Materials Science & Technology*, 33(6), 580-586. https://doi.org/10.1016/j.jmst.2016.05.012
- Castillo, F. J. L., & Dhole, V. R. (1995). Pressure analysis of the ethylene cold-end process. *Computers & Chemical Engineering*, 19, 89-94. https://doi.org/ 10.1016/0098-1354(95)87020-2
- Chang, H. (2001). Exergy Analysis and Exergoeconomic Analysis of An Ethylene Process. Journal of Applied Science and Engineering, 4(2). https://doi.org/ 10.6180/JASE.2001.4.2.03
- Chang, H., & Li, J.-W. (2005). A new exergy method for process analysis and optimization. *Chemical Engineering Science*, 60(10), 2771-2784. https://doi.org/10.1016/j.ces.2004.12.029
- Chauvel, A., & Lefebvre, G. (1989). *Petrochemical Processes* (2nd ed.). Editions Technip Paris.
- Che, Y., Hao, J., Zhang, J., Qiao, Y., Li, D., & Tian, Y. (2018). Vacuum Residue Thermal Cracking: Product Yield Determination and Characterization Using

Thermogravimetry–Fourier Transform Infrared Spectrometry and a Fluidized Bed Reactor. *Energy & Fuels*, *32*(2), 1348-1357. https://doi.org/10.1021/acs.energyfuels.7b03364

- Choudhary, S., Sharma, A., Gupta, S., Purohit, D. H., & Sachan, S. (2020). Use of RSM Technology for the Optimization of Received Signal Strength for LTE Signals Under the Influence of Varying Atmospheric Conditions. *Evergreen*, 7(4), 500-509. https://doi.org/10.5109/4150469
- Cuesta-Lozano, D., Simón López, L. C., Mirón-González, R., Garcia, M., Bonito-Samino, D., & Asenjo-Esteve, Á. (2020). Prevalence Rates of Loneliness and Its Impact on Lifestyle in the Healthy Population of Madrid, Spain. *International Journal of Environmental Research and Public Health*, 17(14), 5121. https://doi.org/10.3390/ijerph17145121
- Dalena, F., Basile, A., & Rossi, C. (2017). Bioenergy systems for the future: prospects for biofuels and biohydrogen (1 ed.). Woodhead Publishing.
- Darvishi, A., Davand, R., Khorasheh, F., & Fattahi, M. (2016). Modeling-based optimization of a fixed-bed industrial reactor for oxidative dehydrogenation of propane. *Chinese Journal of Chemical Engineering*, 24(5), 612-622. https://doi.org/10.1016/j.cjche.2015.12.018
- Dente, M., Ranzi, E., & Goossens, A. G. (1979). Detailed prediction of olefin yields from hydrocarbon pyrolysis through a fundamental simulation model (SPYRO). Computers & Chemical Engineering, 3(1), 61-75. https://doi.org/ 10.1016/0098-1354(79)80013-7
- Depeyre, D., & Flicoteaux, C. (1991). Modeling of thermal steam cracking of nhexadecane. *Industrial & Engineering Chemistry Research*, 30(6), 1116-1130. https://doi.org/10.1021/ie00054a008
- Depeyre, D., Flicoteaux, C., Arbabzadeh, F., & Zabaniotou, A. (1989). Modeling of thermal steam cracking of an atomspheric gas oil. *Industrial & Engineering Chemistry Research*, 28(7), 967-976. https://doi.org/10.1021/ie00091a013
- Díaz, M. S., & Bandoni, J. A. (1996). A mixed integer optimization strategy for a large scale chemical plant in operation. *Computers & Chemical Engineering*, 20(5), 531-545. https://doi.org/10.1016/0098-1354(95)00209-X
- Dominov, P., Gilyazetdinova, R., Zhirnov, B., Tarasov, I., & Khlestkin, R. (2009).
 Overview world technologies of pyrolysis and perspective of development. *1*, 23–32.

- Edwards, L., & Hamilton, S. (1995). Errors-in-variables and the Box-Cox transformation. *Computational Statistics & Data Analysis*, 20, 131-140. https://doi.org/10.1016/0167-9473(94)00039-L
- Eliceche, A., Petracci, N., Hoch, P., & Brignole, E. (1995). Optimal operation of an ethylene plant at variable feed conditions. *Computers & Chemical Engineering*, 19(1), 223–228. https://doi.org/10.1016/0098-1354(95)87040-7
- Epstein, L. G. (1978). The Le Chatelier Principle in optimal control problems. *Journal* of Economic Theory, 19(1), 103-122. https://doi.org/10.1016/0022-0531(78)90058-3
- Ewadh, H. (2020). Statistically analysis using Response Surface Methodology (RSM) for analysis and modelling of wastewater treatment (Review Paper) International Scientific Conference of the University of Babylon, University of Babylon.
- Fakhroleslam, M., & Sadrameli, S. M. (2019). Thermal/catalytic cracking of hydrocarbons for the production of olefins; a state-of-the-art review III: Process modeling and simulation. *Fuel*, 252, 553-566. https://doi.org/10.1016/j.fuel.2019.04.127
- Fakhroleslam, M., & Sadrameli, S. M. (2020). Thermal cracking of hydrocarbons for the production of light olefins; A review on optimal process design, operation, and control. *Industrial & Engineering Chemistry Research*, 59(27), 12288-12303. https://doi.org/10.1021/acs.iecr.0c00923
- Fan, T.-J., Luo, R., Xia, H., & Li, X. (2015). Using LMDI method to analyze the influencing factors of carbon emissions in China's petrochemical industries. *Natural Hazards*, 75(2), 319-332. https://doi.org/10.1007/s11069-014-1226-0
- Feli, Z., Darvishi, A., Bakhtyari, A., Rahimpour, M. R., & Raeissi, S. (2017). Investigation of propane addition to the feed stream of a commercial ethane thermal cracker as supplementary feedstock. *Journal of the Taiwan Institute of Chemical Engineers*, 81, 1-13. https://doi.org/10.1016/j.jtice.2017.10.025
- Froment, G. F. (1990). Coke formation in the thermal cracking of hydrocarbons. *Reviews in Chemical Engineering*, *6*, 293-328.
- Fuad, M., Lye, M. s., Ibrahim, N., Phang, C.-K., Ismail, S., & Nasir, B. (2015). t-test using STATA software. *Education in Medicine Journal*, 7(2), 64. https://doi.org/10.5959/eimj.v7i2.330

- Gaitonde, V. N., Manjaiah, M., Maradi, S., Karnik, S. R., Petkar, P. M., & Paulo Davim, J. (2017). Multiresponse optimization in wire electric discharge machining (WEDM) of HCHCr steel by integrating response surface methodology (RSM) with differential evolution (DE). In J. Paulo Davim (Ed.), *Computational Methods and Production Engineering* (pp. 199-221). Woodhead Publishing. https://doi.org/10.1016/B978-0-85709-481-0.00007-0
- Gál, T., & Lakatos, B. (2008). Thermal cracking of recycled hydrocarbon gas-mixtures for re-pyrolysis: Operational analysis of some industrial furnaces. *Applied Thermal Engineering*, 28((2-3)), 218–225. https://doi.org/10.1016/j.applthermaleng.2007.03.020
- Ganesh, H., Ezekoye, O., Edgar, T., & Baldea, M. (2018). Improving energy efficiency of an austenitization furnace by heat integration and real-time optimization IEEE International Conference on Automation, Quality and Testing, Robotics (AQTR), Cluj-Napoca, Romania. https://ieeexplore.ieee.org/document/8402763/
- Gao, X., Chen, B., & He, X. (2006). An agent-oriented architecture for modeling and optimization of naphtha pyrolysis process. *Computer Aided Chemical Engineering*, 21, 475-481. https://doi.org/10.1016/S1570-7946(06)80091-X
- Gao, X., Chen, B., He, X., Qiu, T., Li, J., Wang, C., & Zhang, L. (2008). Multiobjective optimization for the periodic operation of the naphtha pyrolysis process using a new parallel hybrid algorithm combining NSGA-II with SQP. *Computers & Chemical Engineering*, 32(11), 2801-2811. https://doi.org/ 10.1016/j.compchemeng.2008.01.005
- Geerts, M., Ristic, N., Djokic, M., Ukkandath Aravindakshan, S., Marin, G. B., & Van Geem, K. M. (2020). Crude to Olefins: Effect of Feedstock Composition on Coke Formation in a Bench-Scale Steam Cracking Furnace. *Industrial & Engineering Chemistry Research*, 59(7), 2849-2859. https://doi.org/ 10.1021/acs.iecr.9b06702
- Geng, Z., Wang, Z., Zhu, Q., & Han, Y. (2016). Multi-objective operation optimization of ethylene cracking furnace based on AMOPSO algorithm. *Chemical Engineering Science*, 153. https://doi.org/10.1016/j.ces.2016.07.009
- Gerzeliev, I. M., Fairuzov, D. K., Gerzelieva, Z. I., & Maksimov, A. L. (2019). Production of Ethylene from Ethane Fraction by a Method Alternative to Steam

Cracking. *Russian Journal of Applied Chemistry*, 92(11), 1549-1557. https://doi.org/10.1134/S1070427219110120

- Ghafari, S., Aziz, H. A., Isa, M. H., & Zinatizadeh, A. A. (2009). Application of response surface methodology (RSM) to optimize coagulation–flocculation treatment of leachate using poly-aluminum chloride (PAC) and alum. *Journal* of Hazardous Materials, 163(2), 650-656. https://doi.org/10.1016/ j.jhazmat.2008.07.090
- Ghashghaee, M., & Karimzadeh, R. (2011). Multivariable optimization of thermal cracking severity. *Chemical Engineering Research and Design*, 89(7), 1067-1077. https://doi.org/10.1016/j.cherd.2010.12.002
- Gholami, Z., Gholami, F., Tišler, Z., & Vakili, M. (2021). A Review on the Production of Light Olefins Using Steam Cracking of Hydrocarbons. *Energies*, 14(23), 8190. https://doi.org/10.3390/en14238190
- Gómez, A., Fueyo, N., & Díez, L. I. (2008). Modelling and simulation of fluid flow and heat transfer in the convective zone of a power-generation boiler. *Applied Thermal Engineering*, 28(5), 532-546. https://doi.org/10.1016/j.applthermaleng.2007.04.019
- Gong, S., Shao, C., & Zhu, L. (2017). Energy Efficiency Evaluation based on DEA integrated Factor Analysis with respect to Operation Classification in Ethylene Production. *Chinese Journal of Chemical Engineering*, 25(6), 793-799. https://doi.org/10.1016/j.cjche.2016.10.023
- Gorman, J. M., Sparrow, E. M., & Ahn, J. (2019). In-line tube-bank heat exchangers: Arrays with various numbers of thermally participating tubes. *International Journal of Heat and Mass Transfer*, 132, 837-847. https://doi.org/ 10.1016/j.ijheatmasstransfer.2018.11.167
- Gunst, R. (2012). Regression and ANOVA: An Integrated Approach Using SAS Software. *Technometrics*, 45, 170-171. https://doi.org/10.1198/tech.2003.s159
- Haaland, P. D. (1989). Experimental design in biotechnology. Marcel Dekker.
- Hair, J. F., Anderson, R. E., Tatham, R. L., & Black, W. C. (1995). *Multivariate Data Analysis with Readings* (4 ed.). Prentice-Hall.
- Haladu, S. A., Dalhat Mu'azu, N., Ali, S. A., Elsharif, A. M., Odewunmi, N. A., & Abd El-Lateef, H. M. (2022). Inhibition of mild steel corrosion in 1 M H2SO4 by a gemini surfactant 1,6-hexyldiyl-bis-(dimethyldodecylammonium bromide): ANN, RSM predictive modeling, quantum chemical and MD

simulation studies. *Journal of Molecular Liquids*, 350, 118533. https://doi.org/10.1016/j.molliq.2022.118533

- Han, L., Ding, C., & Lui, H. (2013). Studies on Olefin Production by Steam Cracking of Waste Oil Blended with Naphtha. *Applied Mechanics and Materials*, 291-294, 738-743. https://doi.org/10.4028/www.scientific.net/AMM.291-294.738
- Han, Y., Geng, Z., Wang, Z., & Mu, P. (2016). Performance analysis and optimal temperature selection of ethylene cracking furnaces: A data envelopment analysis cross-model integrated analytic hierarchy process. *Journal of Analytical and Applied Pyrolysis*, 122, 35-44. https://doi.org/10.1016/ j.jaap.2016.10.025
- Hernandez, A. (2012). A model for the prediction of olefin production and coke deposition during thermal cracking of light hydrocarbons Universidad Nacional de Colombia.
- Heynderickx, G. J., Schools, E. M., & Marin, G. B. (2006). Optimization of the Decoking Procedure of an Ethane Cracker with a Steam/Air Mixture. *Industrial & Engineering Chemistry Research*, 45(22), 7520-7529. https://doi.org/10.1021/ie060381a
- Hillewaert, L. P., Dierickx, J. L., & Froment, G. F. (1988). Computer generation of reaction schemes and rate equations for thermal cracking. *AIChE JOURNAL*, 34(1), 17-24. https://doi.org/10.1002/aic.690340104
- Hsu, C. S., & Robinson, P. R. (2019). *Petroleum Science and Technology*. Springer International Publishing.
- Huang, S.-N., & Shao, H.-H. (1994). Application of pattern recognition to ethylene production optimization. *Engineering Applications of Artificial Intelligence*, 7(3), 329-333. https://doi.org/10.1016/0952-1976(94)90060-4
- Jarullah, A. T., Hadi, A. J., & Hameed, S. A. (2015). Optimal design of industrial reactor for naphtha thermal cracking process. *Diyala Journal of Engineering Science*, 8(3), 139-161. https://doi.org/10.24237/djes.2015.08312
- Jin, Y., Li, J., Du, W., & Qian, F. (2015). Multi-Objective Optimization of Pseudo-Dynamic Operation of Naphtha Pyrolysis by a Surrogate Model. *Chemical Engineering & Technology*, 38(5), 900-906. https://doi.org/10.1002/ ceat.201400162
- Joo, E. (2000). *Modelling of Industrial Naphtha Thermal Cracking Furnaces* Korea Advanced Institute of Science and Technology]. Tecnico Lisboa Library.

- Joo, E., Lee, K., Lee, M., & Park, S. (2000). CRACKER a PC based simulator for industrial cracking furnaces. *Computers & Chemical Engineering*, 24(2), 1523-1528. https://doi.org/10.1016/S0098-1354(00)00558-5
- Joo, E., & Park, S. (2001). Pyrolysis Reaction Mechanism for Industrial Naphtha Cracking Furnaces. Industrial & Engineering Chemistry Research, 40(11), 2409-2415. https://doi.org/10.1021/ie0007740
- Junfeng, Z., Zhiping, P., Delong, C., Qirui, L., Jieguang, H., & Jinbo, Q. (2019). A Method for Measuring Tube Metal Temperature of Ethylene Cracking Furnace Tubes Based on Machine Learning and Neural Network. *IEEE Access*, 7, 158643-158654. https://doi.org/10.1109/ACCESS.2019.2950419
- Karaba, A., Dvořáková, V., Patera, J., & Zámostný, P. (2020). Improving the steamcracking efficiency of naphtha feedstocks by mixed/separate processing. *Journal of Analytical and Applied Pyrolysis*, 146, 104768. https://doi.org/10.1016/j.jaap.2019.104768
- Karimi, H., Cowperthwaite, E., Olayiwola, B., Farag, H., & McAuley, K. (2017). Modelling of heat transfer and pyrolysis reactions in an industrial ethylene cracking furnace. *Canadian Journal of Chemical Engineering*, 96(1), 33-48. https://doi.org/10.1002/cjce.22844
- Karimzadeh, R., Godini, H. R., & Ghashghaee, M. (2009). Flowsheeting of steam cracking furnaces. *Chemical Engineering Research and Design*, 87(1), 36-46. https://doi.org/10.1016/j.cherd.2008.07.009
- Katta, V., Jones, E., & Roquemore, W. (1993). Development of global/chemistry model for jet-fuel thermal stability based on observations from static and flowing experiments. 81st AGARD Symposium on Fuels and Combustion Technology for Advanced Aircraft Engines,
- Kee, R. J., Rupley, F. M., Miller, J. A., Coltrin, M. E., Grcar, J. F., Meeks, E., Moffat, H. K., Lutz, A. E., Dixon-Lewis, G., & Smooke, M. D. (2006). *CHEMKIN Release 4.1*. Reaction Design.
- Keyvanloo, K., Sedighi, M., & Towfighi, J. (2012). Genetic algorithm model development for prediction of main products in thermal cracking of naphtha: Comparison with kinetic modeling. *Chemical Engineering Journal*, 209, 255-262. https://doi.org/10.1016/j.cej.2012.07.130
- Keyvanloo, K., Towfighi, J., Sadrameli, S. M., & Mohamadalizadeh, A. (2010). Investigating the effect of key factors, their interactions and optimization of

naphtha steam cracking by statistical design of experiments. *Journal of Analytical and Applied Pyrolysis*, 87(2), 224-230. https://doi.org/10.1016/j.jaap.2009.12.007

- Khan, W. A., Culham, J. R., & Yovanovich, M. M. (2006). Convection heat transfer from tube banks in crossflow: Analytical approach. *International Journal of Heat and Mass Transfer*, 49(25), 4831-4838. https://doi.org/10.1016/ j.ijheatmasstransfer.2006.05.042
- Khor, C. S., Lee, T. F., Nhlapo, D., & Lau, K. K. (2014). Optimal synthesis of ethylene production process. *Chemical Engineering Transactions*, 39, 1585-1590. https://doi.org/10.3303/CET1439265
- Kopinke, F. D., Bach, G., & Zimmermann, G. (1993). New results about the mechanism of TLE fouling in steam crackers. *Journal of Analytical and Applied Pyrolysis*, 27(1), 45-55. https://doi.org/10.1016/0165-2370(93) 80021-Q
- Kopinke, F. D., Zimmermann, G., Reyniers, G. C., & Froment, G. F. (1993). Relative rates of coke formation from hydrocarbons in steam cracking of naphtha. 2.
 Paraffins, naphthenes, mono-, di-, and cycloolefins, and acetylenes. *Industrial & Engineering Chemistry Research*, 32(1), 56-61. https://doi.org/10.1021/ie00013a009
- Krungsri. (2021). Petrochemicals. Industry Indicators, 1-2. https://www.krungsri.com/en/research/industry/industry-outlook/ Petrochemicals/Petrochemicals/IO
- Kucora, I., Paunjoric, P., Tolmac, J., Vulovic, M., Speight, J., & Radovanovic, L. (2017). Coke formation in pyrolysis furnaces in the petrochemical industry. *Petroleum Science and Technology*, 35(3), 213-221. https://doi.org/ 10.1080/10916466.2016.1198810
- Landsberg, W. O., Vanyai, T., McIntyre, T. J., & Veeraragavan, A. (2020). Dual/scram-mode combustion limits of ethylene and surrogate endothermically-cracked hydrocarbon fuels at Mach 8 equivalent highenthalpy conditions. Proceedings ofthe *Combustion* Institute. https://doi.org/10.1016/j.proci.2020.07.003
- Lashkajani, K. H., Ghorbani, B., Amidpour, M., & Hamedi, M.-H. (2016). Superstructure optimization of the olefin separation system by harmony search

and genetic algorithms. *Energy*, 99, 288-303. https://doi.org/10.1016/ j.energy.2016.01.045

- Leo, M. B., Dutta, A., & Farooq, S. (2018). Process Synthesis and Optimization of Heat Pump Assisted Distillation for Ethylene-Ethane Separation. *Industrial & Engineering Chemistry Research*, 57(34), 11747-11756. https://doi.org/ 10.1021/acs.iecr.8b02496
- Li, C., Zhu, Q., & Geng, Z. (2007). Multi-objective Particle Swarm Optimization Hybrid Algorithm: An Application on Industrial Cracking Furnace. *Industrial* & Engineering Chemistry Research, 46(11), 3602-3609. https://doi.org/ 10.1021/ie051084t
- Li, L., Gao, J., & Meng, X. (2005). The Influencing Factors of the Catalytic Pyrolysis Processes and Their Product Distribution. *Petroleum Science and Technology*, 23(3-4), 243-255. https://doi.org/10.1081/LFT-200028279
- Li, Z.-Z., Shen, Y.-D., Xu, H.-L., Lee, J.-W., Heo, K.-S., & Seol, S.-Y. (2008). Optimal design of high temperature vacuum furnace using response surface method. *Journal of Mechanical Science and Technology*, 22(11), 2213-2217. https://doi.org/10.1007/s12206-008-0617-0
- Lin, X., Zhao, L., Du, W., He, W., & Qian, F. (2021). Data-Driven Modeling and Cyclic Scheduling for Ethylene Cracking Furnace System with Inventory Constraints. *Industrial & Engineering Chemistry Research*, 60(9), 3687–3698. https://doi.org/10.1021/acs.iecr.0c06085
- Luo, Y., Kong, L., & Yuan, X. (2015). A systematic approach for synthesizing a lowtemperature distillation system. *Chinese Journal of Chemical Engineering*, 23(5), 789-795. https://doi.org/10.1016/j.cjche.2014.06.041
- Manafzadeh, H., Sadrameli, S. M., & Towfighi, J. (2003). Coke deposition by physical condensation of poly-cyclic hydrocarbons in the transfer line exchanger (TLX) of olefin plant. *Applied Thermal Engineering*, 23(11), 1347-1358. https://doi.org/10.1016/S1359-4311(03)00088-7
- Mangrulkar, C. K., Dhoble, A. S., Chakrabarty, S. G., & Wankhede, U. S. (2017). Experimental and CFD prediction of heat transfer and friction factor characteristics in cross flow tube bank with integral splitter plate. *International Journal of Heat and Mass Transfer*, 104, 964-978. https://doi.org/10.1016/ j.ijheatmasstransfer.2016.09.013

- Martin, P., & Roberts, L. (1996). Efficiency in Minitab. *Teaching Statistics*, *18*, 26-27. https://doi.org/10.1111/j.1467-9639.1996.tb00892.x
- Masoumi, M., Sadrameli, S. M., Towfighi, J., & Niaei, A. (2006). Simulation, optimization and control of a thermal cracking furnace. *Energy*, 31(4), 516-527. https://doi.org/10.1016/j.energy.2005.04.005
- Mavrevski, R., Traykov, M., Trenchev, I., & Trencheva, M. (2018). Approaches to modeling of biological experimental data with Graphpad prism software. WSEAS Transactions on Systems and Control, 13, 242-247.
- McMurry, J. (2013). *Fundamentals of Organic Chemistry* (7 ed.). Cengage Learning, Inc.
- Meng, X., Xu, C., & Gao, J. (2007). Effect of steam on heavy oil catalytic pyrolysis. *Petroleum Chemistry*, 47(2), 83-86. https://doi.org/10.1134/ S096554410702003X
- Meng, X., Xu, C., Gao, J., & Li, L. (2005). Studies on catalytic pyrolysis of heavy oils: Reaction behaviors and mechanistic pathways. *Applied Catalysis A: General*, 294(2), 168-176. https://doi.org/10.1016/j.apcata.2005.07.033
- Meng, X., Xu, C., Gao, J., & Zhang, Q. (2004). Effect of catalyst to oil weight ratio on gaseous product distribution during heavy oil catalytic pyrolysis. *Chemical Engineering and Processing: Process Intensification*, 43(8), 965-970. https://doi.org/10.1016/j.cep.2003.09.003
- Minitab. (2022). *Minitab Express Support*. Minitab, LLC. https://support.minitab.com /en-us/minitab-express/1/
- Mitteer, D., Greer, B., Randall, K., & Briggs, A. (2019). Further Evaluation of Teaching Behavior Technicians to Input Data and Graph Using GraphPad Prism. 20(2), 81-93. https://doi.org/10.1037/bar0000172
- Montgomery, D. C. (2017). *Design and analysis of experiments* (8th ed.). John Wiley & Sons.
- Nabavi, S., Rangaiah, G., Niaei, A., & Salari, D. (2009). Multiobjective Optimization of an Industrial LPG Thermal Cracker using a First Principles Model. *Industrial & Engineering Chemistry Research*, 48(21), 9523–9533. https://doi.org/10.1021/ie801409m
- Nabavi, S., Rangaiah, G., Niaei, A., & Salari, D. (2011). Design Optimization of an LPG Thermal Cracker for Multiple Objectives. *International Journal of*

Chemical Reactor Engineering, 9(1), 1-34. https://doi.org/10.1515/1542-6580.2507

- Nian, X., Wang, Z., & Qian, F. (2013). A Hybrid Algorithm Based on Differential Evolution and Group Search Optimization and Its Application on Ethylene Cracking Furnace. *Chinese Journal of Chemical Engineering*, 21(5), 537-543. https://doi.org/10.1016/S1004-9541(13)60531-5
- Nian, X., Wang, Z., & Qian, F. (2015). Strategy of changing cracking furnace feedstock based on improved group search optimization. *Chinese Journal of Chemical Engineering*, 23(1), 181-191. https://doi.org/10.1016/j.cjche. 2014.09.027
- Nighswander, J. A., Huntrods, R. S., Mehrotra, A. K., & Behie, L. A. (1989). Quench time modeling in propane ultrapyrolysis. *The Canadian Journal of Chemical Engineering*, 67(4), 608-614. https://doi.org/10.1002/cjce.5450670413
- Nikolaidis, I. K., Franco, L. F. M., Vechot, L. N., & Economou, I. G. (2018). Modeling of physical properties and vapor – liquid equilibrium of ethylene and ethylene mixtures with equations of state. *Fluid Phase Equilibria*, 470, 149-163. https://doi.org/10.1016/j.fluid.2018.01.021
- Osborne, J. (2010). Improving your data transformations: Applying the Box-Cox transformation. *Practical Assessment, Research and Evaluation*, 15(12), 1-9. https://doi.org/10.7275/QBPC-GK17
- Ozgur, C. (2019). R, Python, Excel, SPSS, SAS, and MINITAB in Research. *Journal* of Data Science, 17, 45-61.
- Pandey, S., & Rangaiah, G. P. (2013). Multiobjective Optimization of Cold-End Separation Process in an Ethylene Plant. *Industrial & Engineering Chemistry Research*, 52(48), 17229-17240. https://doi.org/10.1021/ie4027764
- Pashchenko, D. (2020). How to choose endothermic process for thermochemical waste-heat recuperation? *International Journal of Hydrogen Energy*, 45(38), 18772-18781. https://doi.org/10.1016/j.ijhydene.2020.04.279
- Peng, Z., Zhao, J., Yin, Z., Gu, Y., Qiu, J., & Cui, D. (2019). ABC-ANFIS-CTF: A Method for Diagnosis and Prediction of Coking Degree of Ethylene Cracking Furnace Tube. *Processes*, 7(12), 909. https://doi.org/10.3390/pr7120909
- Petracci, N., Eliceche, A. M., Bandoni, A., & Brignole, E. A. (1993). Optimal operation of an ethylene plant utility system. *Computers & Chemical Engineering*, 17, S147-S152. https://doi.org/10.1016/0098-1354(93)80221-8

- Petracci, N., Hoch, P. M., & Eliceche, A. M. (1996). Flexibility analysis of an ethylene plant. Computers & Chemical Engineering, 20, S443-S448. https://doi.org/10.1016/0098-1354(96)00084-1
- Pinzi, S., Lopez-Gimenez, F. J., Ruiz, J., & Dorado, M. P. (2010). Response surface modeling to predict biodiesel yield in a multi-feedstock biodiesel production plant. *Bioresource technology*, 101(24), 9587-9593. https://doi.org/ 10.1016/j.biortech.2010.07.076
- Pu, X., & Shi, L. (2013). Commercial test of the catalyst for removal of trace olefins from aromatics and its mechanism. *Catalysis Today*, 212, 115-119. https://doi.org/10.1016/j.cattod.2012.09.010
- Pyl, S. P., Schietekat, C. M., Reyniers, M.-F., Abhari, R., Marin, G. B., & Van Geem,
 K. M. (2011). Biomass to olefins: Cracking of renewable naphtha. *Chemical Engineering Journal*, 176-177, 178-187. https://doi.org/10.1016/ j.cej.2011.04.062
- Rahimi, N., & Karimzadeh, R. (2011). Catalytic cracking of hydrocarbons over modified ZSM-5 zeolites to produce light olefins: A review. *Applied Catalysis* A: General, 398(1), 1-17. https://doi.org/10.1016/j.apcata.2011.03.009
- Razani, S., Farsi, M., Rahimpour, M. R., & Bolhassani, A. (2021). Improving Cracking Severity in an Ethane Thermal Cracker Based on Dynamic Optimization Considering Process Limitations. *Iranian Journal of Chemistry and Chemical Engineering (IJCCE)*, 40(4), 1277-1288. https://doi.org/10.30492/ ijcce.2020.119569.3909
- Reddy, P. V., Reddy, B., & Ramulu, P. (2020). Mathematical modelling for prediction of tube hydroforming process using RSM and ANN. *International Journal of Industrial and Systems Engineering*, 35(1), 13. https://doi.org/10.1504/ IJISE.2020.106848
- Ren, T., Patel, M., & Blok, K. (2006). Olefins from conventional and heavy feedstocks: Energy use in steam cracking and alternative processes. *Energy*, 31(4), 425-451. https://doi.org/10.1016/j.energy.2005.04.001
- Ren, T., Patel, M. K., & Blok, K. (2008). Steam cracking and methane to olefins: Energy use, CO2 emissions and production costs. *Energy*, 33(5), 817-833. https://doi.org/10.1016/j.energy.2008.01.002
- Ren, Y., Guo, G., Liao, Z., Yang, Y., Sun, J., Jiang, B., Wang, J., & Yang, Y. (2020). Kinetic modeling with automatic reaction network generator, an application to

naphtha steam cracking. *Energy*, 207, 118204. https://doi.org/10.1016/ j.energy.2020.118204

- Reyniers, P. A., Schietekat, C. M., Kong, B., Passalacqua, A., Van Geem, K. M., & Marin, G. B. (2017). CFD simulations of Industrial Steam Cracking Reactors: Turbulence–Chemistry Interaction and Dynamic Zoning. *Industrial & Engineering Chemistry Research*, 56(51), 14959-14971. https://doi.org/10.1021/acs.iecr.7b02492
- Rice, F., & Dooley, M. (2002). The Thermal Decomposition of Organic Compounds from the Standpoint of Free Radicals. XII. The Decomposition of Methane. *Journal of the American Chemical Society*, 56. https://doi.org/ 10.1021/ja01327a069
- Rice, F. O. (1931). The Thermal Decomposition of Organic Compounds from The Standpoint of Free Radicals. I. Saturated Hydrocarbons. *Journal of the American Chemical Society*, 53(5), 1959-1972. https://doi.org/ 10.1021/ja01356a053
- Ringle, C., Wende, S., & Becker, J.-M. (2015). SmartPLS 3. http://www.smartpls.com
- Riverol, C., & Pilipovik, M. V. (2007). Optimization of the pyrolysis of ethane using fuzzy programming. *Chemical Engineering Journal*, 133(1), 133-137. https://doi.org/10.1016/j.cej.2007.02.009
- Robertson, R. W. J., & Hanesian, D. (1975). An Optimization Study of the Pyrolysis of Ethane in a Tubular Reactor. *Industrial & Engineering Chemistry Process Design and Development*, 14(3), 216-221. https://doi.org/10.1021/ i260055a004
- Ruckaert, M. J., Martens, X. M., & Desarnauts, J. (1978). Ethylene plant optimization by geometric programming. *Computers & Chemical Engineering*, 2(2), 93-97. https://doi.org/10.1016/0098-1354(78)80013-1
- Sadrameli, S. M. (2015). Thermal/catalytic cracking of hydrocarbons for the production of olefins: A state-of-the-art review I: Thermal cracking review. *Fuel*, 140, 102-115. https://doi.org/10.1016/j.fuel.2014.09.034
- Sadrameli, S. M. (2016). Thermal/catalytic cracking of liquid hydrocarbons for the production of olefins: A state-of-the-art review II: Catalytic cracking review. *Fuel*, 173, 285-297. https://doi.org/10.1016/j.fuel.2016.01.047

- Safwat Wilson, A., & Khalil Bassiouny, M. (2000). Modeling of heat transfer for flow across tube banks. *Chemical Engineering and Processing: Process Intensification*, 39(1), 1-14. https://doi.org/10.1016/S0255-2701(99)00069-0
- Sahud Alotaibi, S. (2020). Identifying the Relationship between Academic Selfmotivation and the Mathematical Thinking: A Case Study of Secondary School Students. Universal Journal of Educational Research, 8(12A), 7239-7245. https://doi.org/10.13189/ujer.2020.082506
- Salari, D., Niaei, A., & Nabavi, R. (2008). Multi-Objective Genetic Optimization of Ethane Thermal Cracking Reactor. *Iranian Journal of Chemical Engineering*, 5(3). https://www.sid.ir/en/Journal/ViewPaper.aspx?ID=136217
- Sedighi, M., Keyvanloo, K., & Towfighi, D. J. (2010a). Olefin production from heavy liquid hydrocarbon thermal cracking: kinetics and product distribution. *Iranian Journal of Chemistry & Chemical Engineering*, 29(4), 135-147.
- Sedighi, M., Keyvanloo, K., & Towfighi, J. (2010b). Experimental study and optimization of heavy liquid hydrocarbon thermal cracking to light olefins by response surface methodology. *Korean Journal of Chemical Engineering*, 27, 1170-1176. https://doi.org/10.1007/s11814-010-0217-4
- Seifzadeh Haghighi, S., Rahimpour, M. R., Raeissi, S., & Dehghani, O. (2013). Investigation of ethylene production in naphtha thermal cracking plant in presence of steam and carbon dioxide. *Chemical Engineering Journal*, 228, 1158-1167. https://doi.org/10.1016/j.cej.2013.05.048
- Shahrokhi, M., & Nejati, A. (2002). Optimal Temperature Control of a Propane Thermal Cracking Reactor. *Industrial & Engineering Chemistry Research*, 41(25), 6572-6578. https://doi.org/10.1021/ie0106783
- Shen, L., Gong, J., & Liu, H. (2015). Effect of Coking Size on the Thermal Diffusion and Stress Distribution of Cr25Ni35Nb and Cr35Ni45Nb Austenitic Steels. *Applied Mechanics and Materials*, 750, 192-197. https://doi.org/ 10.4028/www.scientific.net/AMM.750.192
- Shi, H., Su, C., Cao, J., Li, P., Liang, J., & Zhong, G. (2015). Nonlinear Adaptive Predictive Functional Control Based on the Takagi–Sugeno Model for Average Cracking Outlet Temperature of the Ethylene Cracking Furnace. *Industrial & Engineering Chemistry Research*, 54(6), 1849-1860. https://doi.org/10.1021/ ie503531z

- Shim, S., Shin, I.-S., & Bae, J.-M. (2016). Intervention Meta-Analysis Using STATA Software. Journal of Health Informatics and Statistics, 41, 123-134. https://doi.org/10.21032/jhis.2016.41.1.123
- Song, G., & Tang, L. (2018). Optimization Model for the Transfer Line Exchanger System. Computer Aided Chemical Engineering, 44, 1015-1020. https://doi.org/10.1016/B978-0-444-64241-7.50164-6
- Song, H., Su, C.-l., Shi, H., Li, P., & Cao, J.-t. (2019). Improved predictive functional control for ethylene cracking furnace. *Measurement and Control*, 52(5-6), 526-539. https://doi.org/10.1177/0020294019842602
- Speybroeck, V., Hemelsoet, K., Minner, B., Marin, G., & Waroquier, M. (2007). Modeling elementary reactions in coke formation from first principles. *Molecular Simulation*, 33(9), 879-887. https://doi.org/10.1080/ 08927020701308315
- Sullivan, S., & Greenland, S. (2012). Bayesian regression in SAS software. International journal of epidemiology, 42(1), 308-317. https://doi.org/ 10.1093/ije/dys213
- Sun, X., & Shen, L. (2017). Research progress of coking mechanism and prevention measures for ethylene cracking furnace tubes. *Corrosion Science and Protection Technology*, 29(5), 575-580. https://doi.org/10.11903/1002.6495. 2017.034
- Sun, Y., Yang, G., Li, K., Zhang, L., & Zhang, L. (2016). CO2 mineralization using basic oxygen furnace slag: process optimization by response surface methodology. *Environmental Earth Sciences*, 75(19), 1-10. https://doi.org/ 10.1007/s12665-016-6147-7
- Sun, Y., Zhang, J., & Zhang, L. (2016). NH4Cl Selective Leaching of Basic Oxygen Furnace Slag: Optimization Study Using Response Surface Methodology. *Environmental Progress & Sustainable Energy*, 35(5), 1387-1394. https://doi.org/10.1002/ep.12365
- Sundaram, K. M., & Froment, G. F. (1977a). Modeling of thermal cracking kinetics—
 I: Thermal cracking of ethane, propane and their mixtures. *Chemical Engineering Science*, 32(6), 601-608. https://doi.org/10.1016/0009-2509(77)80225-X
- Sundaram, K. M., & Froment, G. F. (1977b). Modeling of thermal cracking kinetics— II: Cracking of iso-butane, of n-butane and of mixtures ethane—propane—n-

butane. Chemical Engineering Science, 32(6), 609-617. https://doi.org/ 10.1016/0009-2509(77)80226-1

- Sundaram, K. M., & Froment, G. F. (1979). Kinetics of coke deposition in the thermal cracking of propane. *Chemical Engineering Science*, 34(5), 635-644. https://doi.org/10.1016/0009-2509(79)85108-8
- Sundaram, K. M., Van Damme, P. S., & Froment, G. F. (1981). Coke deposition in the thermal cracking of ethane. AIChE JOURNAL, 27(6), 946-951. https://doi.org/ 10.1002/aic.690270610
- Symoens, S. H., Olahova, N., Muñoz Gandarillas, A. E., Karimi, H., Djokic, M. R., Reyniers, M.-F., Marin, G. B., & Van Geem, K. M. (2018). State-of-the-art of Coke Formation during Steam Cracking: Anti-Coking Surface Technologies. *Industrial & Engineering Chemistry Research*, 57(48), 16117-16136. https://doi.org/10.1021/acs.iecr.8b03221
- Toufighi, J., Karimzadeh, R., Saedi, G., Hosseini, S., Morafahi, M., Mokhtarani, B., Niaee, A., & Sadr, A. M. (2004). SHAHAB-A PC-Based Software for Simulation of Steam Cracking Furnaces (Ethane and Naphtha). *Iranian Journal of Chemical Engineering*, 1(2), 55-70.
- Towfighi, J., & Karimzadeh, R. (1993). Development of a mechanistic model for pyrolysis of naphtha. 6th Conference of the Asia Pacific Confederation of Chemical Engineering,
- Towfighi, J., Sadrameli, S. M., & Niaei, A. (2002). Coke Formation Mechanisms and Coke Inhibiting Methods in Pyrolysis Furnaces. *Journal of Chemical Engineering of Japan*, 35, 923-937. https://doi.org/10.1252/jcej.35.923
- Trinh, T. K., & Kang, L. S. (2011). Response surface methodological approach to optimize the coagulation–flocculation process in drinking water treatment. *Chemical Engineering Research and Design*, 89(7), 1126-1135. https://doi.org/10.1016/j.cherd.2010.12.004
- Van Cauwenberge, D. J., Vandewalle, L. A., Reyniers, P. A., Van Geem, K. M., Marin, G. B., & Floré, J. (2017). Periodic reactive flow simulation: Proof of concept for steam cracking coils. *AIChE JOURNAL*, 63(5), 1715-1726. https://doi.org/10.1002/aic.15530
- Van de Vijver, R., Vandewiele, N., Bhoorasingh, P., Slakman, B., Seyedzadeh Khanshan, F., Carstensen, H. H., Reyniers, M. F., Marin, B., West, R., & Van Geem, K. (2015). Automatic Mechanism and Kinetic Model Generation for

Gas- and Solution-Phase Processes: A Perspective on Best Practices, Recent Advances, and Future Challenges. *International Journal of Chemical Kinetics*, *47*(4), 199-231. https://doi.org/10.1002/kin.20902

- Van Geem, K., Dhuyvetter, I., Prokopiev, S., Reyniers, M.-F. o., Viennet, D., & Marin, B. (2009). Coke Formation in the Transfer Line Exchanger during Steam Cracking of Hydrocarbons. *Industrial & Engineering Chemistry Research*, 48(23). https://doi.org/10.1021/ie900124z
- Van Geem, K., Reyniers, M., & Marin, G. (2008). Challenges of Modeling Steam Cracking of Heavy Feedstocks. Oil & Gas Science and Technology, 63(1), 79-94. https://doi.org/10.2516/ogst:2007084
- van Goethem, M. W. M., Barendregt, S., Grievink, J., Moulijn, J. A., & Verheijen, P. J. T. (2010). Model-based, thermo-physical optimisation for high olefin yield in steam cracking reactors. *Chemical Engineering Research and Design*, 88(10), 1305-1319. https://doi.org/10.1016/j.cherd.2010.02.003
- Van Goethem, M. W. M., Barendregt, S., Grievink, J., Verheijen, P. J. T., Dente, M., & Ranzi, E. (2013). A kinetic modelling study of ethane cracking for optimal ethylene yield. *Chemical Engineering Research and Design*, 91(6), 1106-1110. https://doi.org/10.1016/j.cherd.2013.01.006
- Van Goethem, M. W. M., Kleinendorst, F. I., van Leeuwen, C., & van Velzen, N. (2001). Equation-based SPYRO® model and solver for the simulation of the steam cracking process. *Computers & Chemical Engineering*, 25(4), 905-911. https://doi.org/10.1016/S0098-1354(01)00655-X
- Van Speybroeck, V., Van Neck, D., Waroquier, M., Wauters, S., Saeys, M., & Marin, G. B. (2003). Ab initio study on elementary radical reactions in coke formation. *91*(3), 384-388. https://doi.org/10.1002/qua.10458
- Vangaever, S., Reyniers, P. A., Symoens, S. H., Ristic, N. D., Djokic, M. R., Marin, G. B., & Van Geem, K. M. (2020). Pyrometer-based control of a steam cracking furnace. *Chemical Engineering Research and Design*, 153, 380-390. https://doi.org/10.1016/j.cherd.2019.10.023
- Velez, J., Correa, J., & Marmolejo-Ramos, F. (2015). A new approach to the Box–Cox transformation. *Frontiers in Applied Mathematics and Statistics*, 1. https://doi.org/10.3389/fams.2015.00012
- Wan, J., Wei, Y., Liu, Z., Li, B., Qi, Y., Li, M., Xie, P., Meng, S., He, Y., & Chang,F. (2008). A ZSM-5-based Catalyst for Efficient Production of Light Olefins

and Aromatics from Fluidized-bed Naphtha Catalytic Cracking. *Catalysis Letters*, 124(1), 150-156. https://doi.org/10.1007/s10562-008-9445-1

- Wan Omar, W. N. N., Nordin, N., Mohamed, M., & Saidina Amin, N. A. (2009). A Two-Step Biodiesel Production from Waste Cooking Oil: Optimization of Pre-Treatment Step. *Journal of Applied Sciences*, 9(17), 3098-3103. https://doi.org/10.3923/jas.2009.3098.3103
- Wang, D., Fan, X., & Feng, X. (2015). Optimization framework for energy-induced separation network: Application to the chilling train system in ethylene plants. *Chemical Engineering Transactions*, 45, 91-96. https://doi.org/10.3303/ CET1545016
- Wang, H., Yang, Z., Liu, J., Li, G., & Zhang, X. (2020). Activating ABO3-type coating by additive for coke inhibition in supercritical thermal cracking of endothermic hydrocarbon fuel. *Fuel Processing Technology*, 198, 106229. https://doi.org/10.1016/j.fuproc.2019.106229
- Wang, X., & Tang, L. (2013). Multiobjective Operation Optimization of Naphtha Pyrolysis Process Using Parallel Differential Evolution. *Industrial & Engineering Chemistry Research*, 52(40), 14415-14428. https://doi.org/ 10.1021/ie401954d
- Wang, Z., Han, Y., Li, C., Geng, Z., & Fan, J. (2021). Input-output networks considering graphlet-based analysis for production optimization: Application in ethylene plants. *Journal of Cleaner Production*, 278, 123955. https://doi.org/10.1016/j.jclepro.2020.123955
- Wang, Z., Li, Z., Feng, Y., & Rong, G. (2016). Integrated short-term scheduling and production planning in an ethylene plant based on Lagrangian decomposition. *The Canadian Journal of Chemical Engineering*, 94(9), 1723-1739. https://doi.org/10.1002/cjce.22544
- Wauters, S., & Marin, G. B. (2001). Computer generation of a network of elementary steps for coke formation during the thermal cracking of hydrocarbons. *Chemical Engineering Journal*, 82(1), 267-279. https://doi.org/10.1016/ S1385-8947(00)00354-5
- Wauters, S., & Marin, G. B. (2002). Kinetic Modeling of Coke Formation during Steam Cracking. *Industrial & Engineering Chemistry Research*, 41(10), 2379-2391. https://doi.org/10.1021/ie010822k

- Willems, P. A., & Froment, G. F. (1988). Kinetic modeling of the thermal cracking of hydrocarbons. 1. Calculation of frequency factors. *Industrial & Engineering Chemistry Research*, 27(11), 1959-1966. https://doi.org/10.1021/ie00083a001
- Wojtowicz, J. A. (2005). *Ozone*. John Wiley & Sons, Inc. https://doi.org/10.1002/ 0471238961.1526151423151020.a01.pub2
- Xiang, D., Qian, Y., Man, Y., & Siyu, Y. (2014). Techno-economic analysis of the coal-to-olefins process in comparison with the oil-to-olefins process. *Applied Energy*, 113, 639-647. https://doi.org/10.1016/j.apenergy.2013.08.013
- Yan, M. (2000). Simulation and optimization of an ethylene plant Texas Tech University. TTU University Libraries.
- Yang, R., Yi, N., & Xu, S. (2006). Box–Cox transformation for QTL mapping. Genetica, 128, 133-143. https://doi.org/10.1007/s10709-005-5577-z
- Yu, K., Wang, X., & Wang, Z. (2015). Self-adaptive multi-objective teachinglearning-based optimization and its application in ethylene cracking furnace operation optimization. *Chemometrics and Intelligent Laboratory Systems*, 146, 198-210. https://doi.org/10.1016/j.chemolab.2015.05.015
- Yu, K., Wang, X., & Wang, Z. (2016). Multiple learning particle swarm optimization with space transformation perturbation and its application in ethylene cracking furnace optimization. 96, 156–170. https://doi.org/10.1016/j.knosys.2015. 12.020
- Yu, K., While, L., Reynolds, M., Wang, X., Liang, J. J., Zhao, L., & Wang, Z. (2018).
 Multiobjective optimization of ethylene cracking furnace system using selfadaptive multiobjective teaching-learning-based optimization. *Energy*, 148, 469-481. https://doi.org/10.1016/j.energy.2018.01.159
- Zainal Abideen, M. (2016). Optimization of Coagulation Process in Water Treatment Plant Using Statistical Approach Universiti Teknologi Malaysia. UTM Institutional Repository.
- Zakria, M. H. (2018). Evaluation of Mercury Removal by Sulphur Impregnated Activated Carbon in a Large Scale Ethylene Plant Universiti Teknologi PETRONAS, Malaysia. UTPedia.
- Zakria, M. H., Omar, A. A., & Bustam, M. A. (2016). Mercury Removal of Fluctuating Ethane Feedstock in a Large Scale Production by Sulphur Impregnated Activated Carbon. *Procedia Engineering*, 148, 561-567. https://doi.org/ 10.1016/j.proeng.2016.06.511

- Zarinabadi, S., & Samimi, A. (2010). Modeling and Simulation for Olefin Production in Amir Kabir Petrochemical. World Congress on Engineering and Computer Science,
- Zhang, T., & Yang, B. (2016a). Box-Cox Transformation in Big Data. *Technometrics*, 59. https://doi.org/10.1080/00401706.2016.1156025
- Zhang, T., & Yang, B. (2016b). Box-Cox Transformation in Big Data. *Technometrics*, 59(2), 189-201. https://doi.org/10.1080/00401706.2016.1156025
- Zhang, Y., Reyniers, P. A., Du, W., Qian, F., Van Geem, K. M., & Marin, G. B. (2017). Incident Radiative Heat Flux Based Method for the Coupled Run Length Simulation of Steam Cracking Furnaces. *Industrial & Engineering Chemistry Research*, 56(14), 4156-4172. https://doi.org/10.1021/acs.iecr.6b05013
- Zhu, G., Xie, C., Li, Z., & Wang, X. (2017). Catalytic Processes for Light Olefin Production. In C. S. Hsu & P. R. Robinson (Eds.), Springer Handbook of Petroleum Technology (pp. 1063-1079). Springer International Publishing. https://doi.org/10.1007/978-3-319-49347-3_36
- Zimmermann, H., & Walzl, R. (2009). *Ethylene*. Ullmann's Encyclopedia of Industrial Chemistry, Wiley.

LIST OF PUBLICATIONS

- Zakria, M. H., Mohd Ghazali, M. N., & Abdul Rahman, M. R. (2021). Ethylene Yield from a Large Scale Naphtha Pyrolysis Cracking Utilizing Response Surface Methodology. *Pertanika Journal of Science & Technology*, 29(2), 791-808. https://doi.org/10.47836/pjst.29.2.06 (Indexed by WOS ESCI & SCOPUS)
- Zakria, M. H., Mohd Ghazali, M. N., Abdul Rahman, M. R., & Saudi, M. A. (2021). Ethylene yield in a large scale olefin plant utilizing regression analysis. *Polyolefins Journal*, 8(2), 105-113. https://doi.org/10.22063/poj.2021. 2795.1169 (Indexed by SCOPUS)
- Zakria, M. H., Mohd Nawawi, M. G., & Abdul Rahman, M. R. (2021a). Ethylene Yield from Pyrolysis Cracking in Olefin Plant Utilizing Regression Analysis. *E3S Web of Conference*, 287, 1-6. https://doi.org/ 10.1051/e3sconf/202128703004 (Indexed by WOS CPCI & SCOPUS)
- Zakria, M. H., Mohd Nawawi, M. G., & Abdul Rahman, M. R. (2021b). Propylene Yield from Olefin Plant Utilizing Box-Cox Transformation in Regression Analysis. *E3S Web of Conference*, 287, 1-6. https://doi.org/10.1051/e3sconf/202128703013 (Indexed by WOS CPCI & SCOPUS)
- Zakria, M. H., Mohd Ghazali, M. N., Abdul Rahman, M. R. (2022). Propylene yield from naphtha pyrolysis cracking using surface response analysis. *Polyolefins Journal*, 9(1), 15-24. https://doi.org/10.22063/POJ.2021. 2902.1183 (Indexed by SCOPUS)
- Zakria, M. H., Mohd Ghazali, M. N., Abdul Rahman, M. R. (2022). Propylene Yield Assessment Utilizing Response Surface Methodology for Naphtha

Cracking in Olefin Process. *Evergreen Journal, Accepted.* (Indexed by SCOPUS)

 Zakria, M. H., Mohd Ghazali, M. N., Abdul Rahman, M. R. (2022). Ethylene Yield Valuation Utilizing Response Surface Methodology for Steam Cracker Furnace in Olefin Process. *Materials Science and Engineering Technology*, *Accepted*. (Indexed by WOS SCIE IF: 0.854 & SCOPUS)