

EFFECTS OF TYPE OF SALT ON NANOPARTICLES-STABILIZED CARBON
DIOXIDE FOAM IN ENHANCED OIL RECOVERY

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To my beloved mother and father

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ABSTRACT

Foam is employed to improve the efficiency by which the displacing fluid sweeps the reservoir and contacts and recovers oil. However, it is known that foam are both thermodynamically and kinetically unstable systems that readily destruct as a result of film drainage, drying, film breakup and gas diffusion. Study had found that solid nanoparticle may promote foam stabilization. The foams stability depends on the particle size, shape, concentration and hydrophobicity, as well as on the type of surfactant used. The objectives of this study are to determine the effectiveness of different type of salt on CO₂ foam stability by using various concentrations of nanoparticle at fixed surfactant concentration (AOS) and dispersion salinity, and to determine oil recovery performance of each type of salt and compare the results. Three types of salt (Sodium Chloride, Calcium Chloride and Magnesium Chloride) were used to create six different compositions of brine with total salinity of 3 wt%. A silicon dioxide nanoparticle was used in this study at four different concentrations; 0 wt%, 0.1 wt%, 0.5 wt% and 1.0 wt%. Alpha olefin surfactant (AOS) was used as the foaming agent at fixed concentration of 0.5 wt%. Red dye paraffin oil is used to represent the oil in formation. In the foam stability test, it was found that different type of salt requires different concentration of SiO₂ to give optimum stabilization on foam. From this study, it was found that different type of salt would affect the stability of nanoparticle-stabilized foam. The 3 wt% CaCl₂ at 0.5 wt% SiO₂ gave most stable foam and highest oil recovery because of the greater stability of micelles to cause enhanced foam stability. Thus, one can conclude that the foam stabilities can only be enhanced by packing micelles of high stability in the foam lamellae.

ABSTRAK

Busa digunakan untuk meningkatkan kecekapan bendalir anjakan menyapu reservoir dan menyentuh dan mengeluarkan minyak. Walau bagaimanapun, sistem busa secara termodinamik dan kinetiknya adalah tidak stabil, dan mudah musnah akibat daripada saliran filem, pengeringan, pepecahan filem dan peresapan gas. Kajian telah mendapati bahawa nanopartikel pepejal boleh menggalakkan penstabilan busa. Kestabilan busa bergantung kepada saiz partikel, bentuk, kepekatan dan hidrofobisiti, dan juga jenis surfaktan yang digunakan. Objektif kajian ini adalah untuk menentukan keberkesanan berlainan jenis garam ke atas kestabilan busa CO₂ pada perlbagai kepekatan nanopartikel dengan kepekatan tetap bagi surfaktan (AOS) dan kemasinan larutan; dan untuk menentukan prestasi perolehan minyak bagi setiap jenis garam dan membandingkan keputusannya. Tiga jenis garam (Natrium Klorida, Kalsium Klorida dan Magnesium Klorida) telah digunakan untuk menghasilkan enam komposisi yang berbeza dengan jumlah kemasinan 3% berat. Nanopartikel silika dioksida digunakan dalam kajian ini pada kepekatan yang berbeza; 0% berat, 0.1% berat, 0.5% berat dan 1.0% berat. Surfaktan Alfa Olefina (AOS) digunakan sebagai ejen pembusa pada kepekatan tetap 0.5% berat. Minyak parafin diwarnai merah untuk mewakili minyak dalam formasi. Dalam ujian kestabilan, didapati bahawa komposisi air garam yang berbeza akan memberi kesan kepada kestabilan busa yang distabilkan oleh nanosilika. 3% berat CaCl₂ pada 0.5% berat SiO₂ menghasilkan busa yang paling stabil dan perolehan minyak yang paling tinggi kerana misela yang lebih stabil mempertingkatkan kestabilan busa. Dengan itu, boleh disimpulkan bahawa kestabilan busa hanya boleh dipertingkatkan dengan kepadatan misela yang berkestabilan tinggi dalam lamela busa.

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LIST OF ABBREVIATIONS

AOS	-	Alpha Olefin Sulfonate
CaCl ₂	-	Calcium Chloride
CMC	-	Critical Micelle Concentration
CO ₂	-	Carbon Dioxide
EOR	-	Enhanced Oil Recovery
HCPV	-	Hydrocarbon Pore Volume
IFT	-	Interfacial Tension
MgCl ₂	-	Magnesium Chloride
NaCl	-	Sodium Chloride
NBF	-	Newton Black Film
OOIP	-	Oil Originally In Place
SC	-	Stable Supercritical
WAG	-	Water Alternating Gas

CHAPTER 1

INTRODUCTION

1.1 Background

World production capabilities of oil by conventional means will no longer meet energy demands. As a result, oil price will continue to soar. The soaring of oil price is speculated could make enhanced oil recovery very economically attractive, and could be the beginning of an era when unconventional petroleum sources become economic. These potential oil resources include heavy oils, tar sands, oil shale, and enhanced recovery from know reservoirs. Despite the convenience and adaptable of petroleum as energy resource, it is non-renewable. However, through enhanced oil recovery technique, it is possible to increase or maintain the current levels of production further for many years to come.

Enhanced oil recovery technique can improve the percentage of discovered oil that can be used. It can be classified into thermal processes, chemical processes, and miscible displacement. One of the techniques of enhanced oil recovery is miscible gas injection. Usually, conventional gas or water-drive will leave 25-50% of the original oil in the reservoir. A portion of this oil can be recovered if the oil is contacted by a fluid with which it is miscible. Carbon dioxide (CO₂) and flue gas has become recognised as a useful substitute for light hydrocarbon gases in enhance oil recovery of crude oil where CO₂, know to be highly soluble in crude oils, and in water, causing 20 to 90% reduction in crude oil viscosities, and also an appreciable swelling (up to 50%) of crude oil when it is saturate with CO₂ above about 700 psi. However, most gas flooding field projects often faced with problem of early CO₂

breakthrough, poor sweep efficiency, and inefficient oil recovery due to viscous fingering resulted from low gas phase viscosity and an unfavourable mobility control.

One potential solution for reducing gas mobility is the use of foam. By the use of foam, the gas mobility is reduced by liquid films trapping gas in the porous media and reducing the gas fraction available for flow. Foam are also much efficient than water in controlling gas fingering. It can also decrease aqueous permeability by increasing gas saturation. Furthermore, the surfactant concentration required to formed foams is low, thus resulted in a potential cost effective process. Foam system is also reversible and is best suited where total blockage is not desired or feasible. However, it is known that foam are both thermodynamically and kinetically unstable systems that readily destruct as a result of film drainage, drying, film breakup and gas diffusion.

Solid particles may promote foam stabilization. Studies found that foams stability depends on the particle size, shape, concentration and hydrophobicity, as well as on the type of surfactant used. Much study had been conducted to study the stability of CO₂ foam. This study focused on the effects of different type of salt with various concentrations of nanoparticle on stabilized CO₂ foam for mobility control in immiscible flooding with fixed concentration of anionic surfactant (AOS) and brine salinity.

1.2 Problem Statement

Gases used in gas-flooding (such as CO₂, Hydrocarbons, N₂, etc.) are normally less viscous (more than one order of magnitude less) and less dense than both water and crude oil, which results in gas channelling through the high permeability zones and gravity overriding. Thus, gas flooding normally has poor volumetric sweep efficiency, especially in an immiscible displacement, with the displacing phase being a lower viscosity. The main advantage of gas is its better microscopic sweep leading to lower residual oil saturation in the pores compared to waterflood however, the major challenge associated with gas injection is its poor

volumetric sweep efficiency, as a result of which gas does not contact a large fraction of oil and, thus, the overall recovery remains low.

A need for mobility control in gas flooding has led to the use of foam for sweep improvement and profile modification. Foam is employed to improve the efficiency by which the displacing fluid sweeps the reservoir and contacts and recovers oil.

Foam stability and mobility reduction characteristics depend on the properties of rock and fluids and process-design parameters such as formation permeability, injection foam quality and the size of the chemical slug. The effects of these parameters on the performance of the foam flooding process needs to be ascertained in order to determine its optimal potential for EOR. Most foam owes its existence to the presence of surfactants, that is, materials which are surface active.

They are concentrated at the interface of a fluid and act to reduce the surface tension between interfaces. More importantly for preventing foam termination, they stabilize the thin fluid film against rupture. In aqueous foams, surfactant molecules are amphiphilic; their two parts are hydrophobic and hydrophilic so that they can stay on the water surface. In porous media, foam exists as gas bubbles whose shapes conform to the solid matrix. Hirasaki (1989) defined foam in porous media as “a dispersion of a gas in a liquid such that the liquid phase is continuous and at least some part of the gas phase is made discontinuous by thin liquid films called lamellae.”

Solid-stabilized emulsions, known as “pickering emulsions”, have drawn active research interest for use under harsh conditions, because solid particles are generally more stable than surfactant under high temperature, and/or high salinity conditions. Their application in the upstream oil has been limited because the solids available to use were colloidal size with wide size distribution; and consequently it is not feasible for them to propagate long distance in oil reservoir formations with pore throat having similar sizes to those particles. It was also difficult to make with those colloids very stable and dense emulsions that are useful for upstream applications.

With the recent rapid nanotechnology advancement, nanoparticles with uniform size and shape and desired surface properties can be produced in large

quantities at economic cost. The surface properties (“wettability”) of the nanoparticle can be adjusted to generate desired types of emulsion droplet with desired uniform sized. Being solid particles, the nanoparticles can not only endure the harsh reservoir conditions, but also carry some additional functionality, such as paramagnetism or as a catalyst. Very stable emulsions and foams with desired internal structure and functionality can now be accordingly produced for potential oil recovery applications.

1.3 Objective of study

The objectives of this study are:

- a) To determine the effectiveness of different type of salt on CO₂ foam stability by using various concentration of nanoparticles at fixed surfactant concentration (AOS) and dispersion salinity.
- b) To determine oil recovery performance of each type of salt and compare the results.

1.4 Scope of study

In this thesis, the interested parameters to enhance the performance of foam are type of salt (Sodium chloride (NaCl), Calcium chloride (CaCl₂) and Magnesium Chloride (MgCl₂)) and nanoparticles concentration (0 wt%, 0.1 wt%, and 0.5 wt%, 1.0 wt %) with fixed concentration of 0.5 wt% anionic surfactant (AOS) and 3 wt% water salinity.

For the purpose of this study, glass bead pack models with 30.5 cm of length and 5.4 cm of diameter made from Acrylic (Perspex) were used and packed with granule sized of glass bead in range between 90 to 150 μm. Porosity and permeability of glass bead packs were measured.

The stability test was prepared and modified based on ASTM-D 6082-62, D892-06 and D1881-87 (Borole and Caneba, 2013). In this study, the stability of the

aqueous foam was evaluated by the Ross-Miles method, using half-life measurements. The technique used in to generate the foam is air expansion this technique was used to evaluate the stability of foam in presence of nanoparticles, surfactant and brine.

Displacement tests were performed by using seven unconsolidated glass bead pack models. The glass bead packs were located in the horizontal position to neglect the effect of gravity forces. Initially, the glass bead pack was saturated with 3 wt% NaCl brine for measurement of porosity and absolute permeability. Then, the glass bead pack models were saturated with red dye paraffin oil to create the initial oil saturation. Next, water flooding was done by injection of 2 PV of 3 wt% brine at flow rate of 3 cc/min. The oil recovery after water flooding was calculated. Finally, the glass bead pack model were subjected to foam flooding by injecting aqueous nanoparticles dispersion with flow rate 3 cc/min followed by co-injection of CO₂ gases with 9 cc/min for further oil recovery and better volumetric sweep efficiency.

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