THE MORPHOLOGY AND STRUCTURAL PROPERTIES OF SILICON CARBIDE QUANTUM DOTS GROWN BY VERY HIGH FREQUENCY PLASMA ENHANCED CHEMICAL VAPOUR DEPOSITION

NUR FARAH NADIA BINTI ABD KARIM

A thesis submitted in fulfilment of the requirements for the award of the degree of Master of Philosophy (Physics)

> Faculty of Science Universiti Teknologi Malaysia

> > JUNE 2019

This thesis dedicated to My precious parents Abd Karim and Hasnah. My beloved husband Mohd Syafiq. My cutie pie Luthfiyaa Eryna. Thank you...

ACKNOWLEDGEMENT

Alhamdulillah, first and foremost, all praise to Allah Almighty, for giving me the courage, strength and patience to complete this research.

Over the years in UTM, I had the pleasure of meeting and working with many great people who contributed to this work in different ways. Surely, it is impossible to thank all of them properly for all their help and support in just a few lines. Nevertheless, I will make an attempt to express my gratitude:

I would like acknowledge my supervisor, Dr Abd Khamim bin Ismail for being resourceful, inspiring, supporting and understanding during my study. To all the staffs in Ibnu Sina Institute, University Industry Research Laboratory, Physics Department (UTM) thank you for the caring and encouragement given all along. I would also like to thank Dr Muhammad Firdaus Omar and Dr Nurhafizah Hasim for the support and motivation given during my study. Thanks also to my friends especially my group for discussions and sharing of ideas.

ABSTRACT

The growth and characterization of silicon carbide (SiC) quantum dots (QDs) are reported in this work. The SiC QDs were grown using plasma enhanced chemical vapour deposition (PECVD) at 150 MHz radio frequency (RF). A mixture of silane (SiH₄) and methane (CH₄) with a ratio of 1:4 and diluted in hydrogen (H₂) was used as precursor gaseous. By manipulating the growth parameters such as hydrogen flow rate, growth temperature, growth time and RF power, the morphological and structural properties of SiC ODs were studied. The surface morphology of samples was observed through atomic force microscopy (AFM) and field emission scanning electron microscopy (FESEM). The structural properties of the sample were determined using Fourier transform infra-red (FTIR), emission dispersive xray (EDX) and Raman spectroscopy. It was found that a combination of dots formed islands on the substrate and the size of each dot was below 50 nm. By observing the cross-section view of the substrate, this self-assembled quantum dots followed Stranski-Krastanow (S-K) mode in which a combination of both islands and layer mode was formed. FTIR result showed absorption peaks located at 500 to 1000 cm⁻¹ which proved the presence of SiC bonding. Further elemental mapping by EDX confirmed that the island and the layer underneath were formed from silicon, carbon and oxygen. This is in agreement with the observed SiC QDs cross-sectional FESEM image. The Raman spectra revealed three SiC polytypes which were 3C-SiC, 4H-SiC and 6H-SiC, and therefore in agreement with XRD results. All polytypes were considered as crystal due to small full width half maximum (FWHM) with crystallite size greater than 1 nm and sharp peaks were formed on Raman spectra. In conclusion, the SiC QDs growth parameters have shown a good impact to the morphological and structural properties of the grown quantum dots.

ABSTRAK

Pertumbuhan dan pencirian titik-titik kuantum (QDs) silikon karbida (SiC) dilaporkan dalam kerja ini. Titik-titik kuantum SiC telah ditumbuh menggunakan pemendapan wap kimia diperkuat plasma (PECVD) pada frekuensi radio (RF) 150 MHz. Campuran silana (SiH₄) dan metana (CH₄) dengan nisbah 1:4 dan dicairkan dalam hidrogen (H₂) telah digunakan sebagai gas pelopor. Dengan memanipulasi parameter pertumbuhan, seperti kadar aliran hidrogen, suhu pertumbuhan, masa pertumbuhan dan kuasa RF, ciri-ciri morfologi dan struktur titik-titik kuantum SiC telah dikaji. Morfologi permukaan sampel dicerap melalui mikroskopi daya atom (AFM) dan mikroskopi elektron pengimbasan pancaran medan (FESEM). Ciri-ciri struktur sampel ditentukan menggunakan transformasi Fourier infra-merah (FTIR), sebaran pancaran sinar-X (EDX) dan spektroskopi Raman. Didapati bahawa kombinasi titik-titik membentuk pulau-pulau pada substrat dan saiz setiap titik adalah di bawah 50 nm. Dengan mencerap pandangan keratan rentas substrat, titik kuantum yang terhimpun-sendiri ini mengikuti mod Stranski-Krastanow (S-K) di mana satu gabungan kedua-dua pulau dan mod lapisan dibentuk. Keputusan FTIR menunjukkan puncak penyerapan terletak pada 500 hingga 1000 cm⁻¹ yang membuktikan kehadiran ikatan SiC. Pemetaan unsur seterusnya menggunakan EDX mengesahkan bahawa pulau dan lapisan di bawahnya adalah dibentuk daripada silikon, karbon dan oksigen. Ini adalah sepadan dengan cerapan imej keratan rentas FESEM titiktitik kuantum SiC. Spektra Raman menunjukkan tiga polijenis SiC iaitu 3C-SiC, 4H-SiC dan 6H-SiC, dan oleh itu selaras dengan keputusan XRD. Kesemua polijenis ini dianggap sebagai hablur kerana lebar penuh setengah maksimum (FWHM) yang kecil dengan saiz hablur lebih besar daripada 1 nm dan puncak yang tajam terhasil pada spektra Raman. Kesimpulannya, parameter pertumbuhan titik-titik SiC telah menunjukkan impak baik kepada morfologi dan struktur titik-titik kuantum yang ditumbuhkan.

TABLE OF CONTENTS

TITLE	PAGE
DECLARATION	ii
DEDICATION	iii
ACKNOWLEDGEMENTS	iv
ABSTRACT	v
ABSTRAK	vi
TABLE OF CONTENTS	vii
LIST OF TABLES	Х
LIST OF FIGURES	xii
LIST OF SYMBOLS	xvii
LIST OF APPENDICES	XX

CHAPTER 1 INTRODUCTION

1.1	Overview	1
1.2	Background of Research	1
1.3	Problem Statement	6
1.4	Research Objectives	8
1.5	Scope of Research	8
1.6	Significant of Research	9
1.7	Thesis Outline	9

CHAPTER 2 LITERATURE REVIEW

2.1	Semiconductor Quantum Dots	11
2.2	Epitaxy	16
2.3	Growth of Self-Assembled	17

Semiconductor Nanomaterials

2.4	Morphological and Structural Properties	19
	of Self-assembled quantum Dots	
	2.4.1 Morphology	20
	2.4.2 Structural Properties of SiC	25
2.5	Si and SiC as semiconductor material	34
2.6	SiC Quantum Dots	40
2.7	Plasma Enhanced Chemical Vapour	41
	Deposition	

CHAPTER 3 RESEARCH METHODOLOGY

3.1	Introduction	47
3.2	Sample Preparation	49
	3.2.1 Substrate Cleaning	49
	3.2.2 Growth of Silicon Carbide	50
	Quantum Dots	
	3.2.3 Growth Parameters of Silicon	55
	Carbide Quantum Dots	
3.3	Characterization of Silicon Carbide	57
	Quantum Dots	
	3.3.1 Atomic Force Microscopy (AFM)	57
	3.3.2 Field Emission Scanning Electron	58
	Microscopy (FESEM) and Energy	
	Dispersive X-ray Spectroscopy (EDX)	
	3.3.3 Fourier Transform Infrared	59
	Spectroscopy (FTIR)	
	3.3.4 Raman Spectroscopy	60
	3.3.5 X-ray Diffraction Spectroscopy	61

CHAPTER 4 RESULTS AND DISCUSSION

4.1	Introduction	65
4.2	Morphology	66
	4.2.1 Effect of Hydrogen Flow Rate	67
	4.2.2 Effect of RF Power	73
	4.2.3 Effect of Growth Time	78
	4.2.4 Effect of Growth Temperature	83
4.3	Structural Properties of SiC	88
	4.3.1 Fourier Transform Infrared	88
	Spectroscopy	
	4.3.2 Micro Raman Spectroscopy and	93
	Energy Dispersive X-Ray	
	4.3.3 Energy Dispersive X-ray	97
	Spectroscopy	
	4.3.4 X-Ray Diffraction	100
CHAPTER 5	CONCLUSIONS AND	
	SUGGESTIONS	
5.1	Conclusions	105
5.2	Suggestion	107
REFERENCES	8	109

APPENDICES

139

119-

LIST OF TABLES

TABLE NO.	TITLE	PAGE
Table 2.1	Quantum Confined Structure	13
	Classification (Arivazhagan, 2013).	
Table 2.2	Table of Raman shift of SiC polytypes	29
	(Nakashima et al., 2003)	
Table 2.3	Properties of Silicon and SiC (Choi,	36
	2012)	
Table 2.4	Band gap and lattice parameters of	39
	common SiC polytypes (Harris, 1995;	
	Izhevskyi et al, 2000).	
Table 3.1	Growth Parameter for SiC Quantum Dots	56
Table 3.2	The parameter and setup condition of	63
	XRD	
Table 4.1	The surface roughness, mean size, mean	71
	diameter and density of samples grown at	
	different H ₂ flow rate.	
Table 4.2	The summary data of AFM surface	76
	analysis.	
Table 4.3	The roughness, mean size, mean diameter	83
	and density.	
Table 4.4	The roughness, mean size, mean diameter	88
	and island density.	

Table 4.5	Summary of observed peak position and	92
	FWHM in parentheses of Si-C related	
	phonon, stretching, deformation and	
	rocking mode.	
Table 4.6	Raman peak positions of SiC QDs	97
Table 4.7	Atomic percentages of Silicon and	100
	Carbon at three regions, map data 19, 20	
	and 21 of SiC quantum dots.	
Table 4.8	Summary of XRD data analysis for	103
	sample M,N and O	
Table 5.1	The Best Parameter of SiC Quantum Dots	107
	Deposition	

LIST OF FIGURES

FIGURE NO.	TITLE	PAGE
Figure 2.1	Schematic representation of system	15
	dimensionality on the density of states for	
	bulk semiconductor, quantum well,	
	quantum wire and quantum dots	
	(Zachbroeck & Bart, 1997; Ameruddin,	
	2010).	
Figure 2.2	Cross-section views of the three primary	19
	modes of heteroepitaxial growth (a)	
	Volmer-Weber (VW: island formation), (b)	
	Frank-van der Merwe (FM: layer-by-layer),	
	and (c) Stranski-Krastanov (SK: layer-plus-	
	island) (Barabasi, 1999; Ameruddin, 2010).	
Figure 2.3	(a) Laser, (b) cantilever with tip, (c)	21
	photodiode and (d) feedback circuitry to	
	maintain the applied force throughout	
	scanning (Morris, 2014).	
Figure 2.4	SiC nuclei growth by carbon vapour	22
	deposition (Andrivski, 2009).	
Figure 2.5	The signal emits when electron bombard	23
	sample (Hayat, 2000).	
Figure 2.6	FESEM image of SiC quantum dot grown at	24
	different H ₂ flow rate and different	
	deposition time (Huang et al., 2006).	
Figure 2.7	Principle working of ATR (Perkin Elmer,	26

2015)

Figure 2.8	General illustration of Raman scattering.	27
Figure 2.9	Raman quantum energy diagram.	27
Figure 2.10	Raman spectra of bulk 6H-SiC and 3C-SiC	30
	(Wasyluk <i>et al.</i> , 2010).	
Figure 2.11	EDX spectrum of SiC quantum dots (Huang	31
	et al., 2006).	
Figure 2.12	Bragg's law of periodic arrangement atoms	32
	(Prendergast, 2010)	
Figure 2.13	XRD spectra of (a) Si (111) substrate and	35
	(b) & (c) epitaxial 3C-SiC films on Si (111)	
	substrates grown at 1250°C (Gupta et al.,	
	2004).	
Figure 2.14	(a) Diamond lattice and (b) unit cell of	34
	silicon (Greve, 2012).	
Figure 2.15	The tetragonal bonding of a carbon atom	37
	with four nearest neighbors (Reddy, 2008).	
Figure 2.16	Polytype structures of 2H-SiC, 3C-SiC, 4H-	38
	SiC and 6H-SiC (Mukherjee, 2011)	
Figure 2.17	Chemical vapor deposition mechanism	46
	(Mukherjee, 2011).	
Figure 3.1	Research flow chart.	48
Figure 3.2	Penta Vacuum VHF-PECVD system.	51
Figure 3.3	Schematic diagram of 150MHz VHF-	52
	PECVD.	
Figure 3.4	Atomic force microscopy.	58
Figure 3.5	Field emission scanning electron	59
	microscopy in UTM	

Figure 3.6	FTIR/NIR spectrometer	60
Figure 3.7	Raman Spectrometer	61
Figure 3.8	XRD diffractometer	62
Figure 4.1	Cross section morphology of SiC quantum	66
	dots at 50K magnification.	
Figure 4.2	AFM image of SiC quantum dots of sample	68
	A at 140 sccm (a), B at 120 sccm (b), C at	
	100 sccm (c), D at 80 sccm (d) and E at 60	
	sccm (e).	
Figure 4.3	Hydrogen flow rate against surface	70
	roughness and dot density.	
Figure 4.4	30K magnification FESEM image of	73
	sample A at 140 sccm (a), B at 120 sccm	
	(b), C at 100 sccm (c), D at 80 sccm (d) and	
	E at 60 sccm (e)	
Figure 4.5	AFM image of SiC quantum dots with	75
	different RF power (a) 30W for smple F, (b)	
	25W for sample G, (c) 20W for sample H,	
	(d) 15W for sample I, and (e) 10W for	
	sample J.	
Figure 4.6	FESEM image of SiC quantum dots with	78
	different RF power (a) 30W for smple F, (b)	
	25W for sample G, (c) 20W for sample H,	
	(d) 15W for sample I, and (e) 10W for	
	sample J.	
Figure 4.7	AFM image for sample (a) K at 5 min, (b) L	80
	at 4 min, (c) M at 3min, (d) N at 2 min and	
	(e) O at 1 min.	

Figure 4.8	FESEM image with 30K magnification of	82
	SiC quantum dots with different time of	
	deposition a) 5 min, b) 4 min, c) 3 min, d) 2	
	min, and e) 1 min of sample K, L, M, N and	
	О.	
Figure 4.9	AFM image of sample (a) P at 400°C, (b) Q	84
	at 300 °C, (c) R at 200 °C , (d) S at 100 °C	
	and (e) T at 24 $^{\circ}$ C (room temperature)	
Figure 4.10	FESEM image of SiC quantum dots of	87
	sample (a) P at 400°C, (b) Q at 300 °C, (c) R	
	at 200 $^{\rm o}\rm{C}$, (d) S at 100 $^{\rm o}\rm{C}$ and (e) T at 24 $^{\rm o}\rm{C}$	
	(room temperature) at 30K magnification.	
Figure 4.11	Infrared spectrum of sample A.	90
Figure 4.12	Raman spectra of SiC quantum dots grown	93
	at different hydrogen flow rate of 120 sccm	
	(B), 100 sccm (C) and 80 sccm (D)	
Figure 4.13	Raman spectra of SiC quantum dots of	94
	sample G, H and J grown at 25W, 20W and	
	15W.	
Figure 4.14	Raman spectra of SiC quantum dots of	95
	sample N, M and O which grown at $T_s =$	
	120s, 180s and 60s.	
Figure 4.15	Raman spectra of SiC quantum dots growth	96
	at 300°C, 100°C and 200°C.	
Figure 4.16	(a) FESEM image of the SiC quantum dots	99
	island with three EDX spectrum point; (b)	
	EDX signals collected from map data 19.	
Figure 4.17	XRD spectrum of sample M, N and O from	101

 20° to 90.

Figure 4.18 XRD spectrum of sample M, N and O at 101 zoom mode.

LIST OF SYMBOLS

μs	- strain energy
0D	- zero-dimensional
1D	- one dimensional
2D	- two-dimensional
3D	- three-dimensional
a_e	- lattice constant of epitaxial layer
a_s	- lattice constant of substrate
d	- lattice distance
DOS	- density of state
E_i	- energy quantized state
f	- lattice mismatch
Ye	- surface energy
Yes	- interface energy
Ys	- surface energy of the substrate
m^*	- electron mass
п	- integer
q_B	- maximum wave vector of the Brillouin Zone of 3C SiC
t_c	- critical thickness
δ	- function at descrete energy
λ	- wavelength of incident x-ray
λ_{db}	- de Broglie's wavelength
ρ	- Momentum
χ	- reduce wave vector
AFM	- atomic force microscopy
Ar	- Argon gas

ATR	- attenuated total reflection
CBL	- Cluster-Bethe-lattice
CH_4	- methane
CH_4	- Methane gas
CVD	- Chemical vapor deposition
D	- crystallite size
DI-H ₂ O	- deionized water
EDX	- energy dispersive x-ray
FESEM	- field emission scanning electron microscopy
FLA	- folded longitudinal acoustic
F-M	- Frank-van der Merwe
FTIR	- fourier transform infrared
FWHM	- full width half maximum
H_2	- hydrogen gas
HF	- hydrofluoric acid
ICDD	- International Center for Diffraction Data
IPA	- isopropanol
LA	- longitudinal acoustic
LED	- Light emitting diode
LO	- longitudinal optical phonon
LO	- longitudinal optical phonon
MBE	- molecular beam epitaxy
MFC	- mass flow controller
N_2	- nitrogen gas
NEMS	- Nanoelectromechanical system
PDF	- powder diffraction file
PECVD	- plasma enhanced chemical vapor deposition
RAMAN	- Micro-raman spectrometry

RF	- radio frequency
R _{rms}	- Surface roughness
Si	- Silicon
SiC	- Silicon Carbide
SiH ₄	- Silane gas
SiO ₂	- silicon oxide
S-K	- Stranski-Krastanow
TEC	- Thermoelectric cooling
TEM	- Transmission electron microscopy
TMP	- turbo molecular pump
ТО	- transverse optical phonon
T_g	- growth temperature
T_s	- Growth time
VHF-PECVD	- Very high frequency plasma enhanced chemical vapor
	deposition
V-W	- Volmer-Weber
XRD	- X-ray diffraction

LIST OF APPENDIX

APPENDIX	TITLE	PAGE
Appendix A	FTIR Spectra of SiC quantum dots	119
Appendix B	EDX mapping image	138

CHAPTER 1

INTRODUCTION

1.1 Overview

This chapter gives an overview about background, problem statement, objectives, scope, significant and contribution of this research. Finally, the thesis outline is presented toward the end of this chapter.

1.2 Background of Research

In the past two decades, there has been a great interest of nanostructures materials fabrication. The materials comprising of submicron or nanoscale size with at least at one dimension and exhibit the size effect, had received much attention by researchers due to its interesting characteristics and properties compared to bulk materials (Tiwari *et al.*, 2012). Silicon (Si) based materials are the most popular candidate since the

techniques used for fabrication are largely correspond with available semiconductor production processes. However, its crucial limitation where it is not well-matched to handle large current densities and high voltage open up an opportunity to new materials of wide band gap such as silicon carbide (SiC) (Mukherjee, 2011).

Silicon carbide (SiC) is regarded as a promising substitute for silicon especially in high power, high temperature and high frequency devices. This emerging semiconductor material has received a great deal of attention due to its unique properties which fit for application in optoelectronics and microelectronics (Fan *et al.*, 2006; Cheng *et al.*, 2007) such as light emitting diode, electroluminescent devices, nanoelectromechanical system (NEMS) sensors fabrication and also thermoelectric cooling (TEC) devices for deployment in extreme environments (Saddaw, 2012). The most recent application of nanostructures SiC is quantum dots structures in optical device fabrication such as blue LED (Willander *et al.*, 2006).

Generally, thin film deposition method can be used to grow quantum dots structures. Si quantum dots structures are known to be grown with various methods. There are three type of quantum dots synthesis method which are solution-based chemical methods, chemical vapour methods and physical vapour deposition (Fan *et al.*, 2012). Now, quantum dots can be synthesized by chemical colloidal method, self-assembly method, lithography and etching, and split-gate approach.

Through chemical colloidal method, multilayered quantum dots can be obtained and this process is suitable for mass production. In self-assembly method, chemical vapour deposition process (CVD) or molecular beam epitaxy (MBE) is utilized. In this method, the principle of lattice mismatch is applied to ensure the growth of quantum dots can undergo selfpolymerization in a particular substrate. This technique is applicable for regularly arranged quantum dots mass production.

By electron beam lithography and etching, quantum dots can be fabricated by direct write of the electron beam to the photoresist on the substrate and followed by etching the substrate. However, it is time consuming and not suitable for mass production. The two-dimensional plane of the quantum well is generated from two-dimensional confinement when external voltage is applied is called split-gate approach. This approach is fit for academic research but not for mass production (Shi *et al.*, 2015).

Cheng *et al* (2007) had synthesized self-assemble SiC quantum dots grown on Si substrate by low-frequency inductively coupled plasma assisted RF magnetron sputtering. The effects of SiC target power and gas pressure on surface morphology and structural properties of SiC quantum dots were investigated. They discovered the growth dynamics of quantum dots obeys cubic root-law behavior. This law state that when an object undergoes a proportional increase in size, its new surface area is proportional to the square of multiplier and its new volume is proportional to the cube of multiplier. Thus, as the dimension increased, the volume will continue grow faster than the surface area. Moreover, the SiC quantum dots morphology is highly uniform and the average size increases as the pressure increase, below 1 Pa. Yet, the morphology became non-uniform and decreases in sized as pressure exceed 1 Pa. These behaviors attributed to scattering effect and surface mobility of the sputtered atoms (Cheng *et al.*, 2007).

In 2012, a group of researchers, Fan *et al* had fabricated SiC quantum dots from three different type of silicon carbide which are 3C, 6H and 4H SiC polytypes by electrochemical method and studied the dots based on its photoluminenscence. They noticed those polytypes show unexpected quite-similar photoluminescence, photoluminescence excitation, and transient photoluminescence properties which can be explained by polytypic transformations of the colloidal SiC quantum dots driven by ultrasonic waves. Although this method can produce SiC quantum dots but the size of the dots is varied from 1 nm to 8 nm. (Fan *et al.*, 2012). Both research done by Cheng *et al* in 2007 and Fan *et al* in 2012 did not discuss the type of quantum dots obtained whether it is amorphous or crystalline.

Mwania *et al* (2013) had produced quantum sized cluster of β -SiC by photo-assisted electrochemical corrosion of bulk powders. Transmission electron microscopy (TEM) result shows that β -SiC quantum dots are single crystalline. The process which is the amount of synthesized quantum dots and its size can be controlled via regulating the deposition time. However, indirect chemical etching through some of the in situ generated oxidizing species which cannot be excluded completely. Also, it requires external bias potential in order to control the size of the dots (Mwania *et al.*, 2013).

3C-SiC quantum dots fabricated via pulsed laser ablation method was carried out by Zhu *et al* in 2014. They successfully produced bulk quantity of crystalline 3C-SiC quantum dots with uniform diameter of about 2 nm. Even though they had fabricated crystalline quantum dots with uniform diameter, but the distribution of the quantum dots obtained is not being observed and discussed. Moreover, the 6H-SiC target used in their experiment to fabricate the dots is quite expensive (Zhu *et al.*, 2014).

This research aims to grow SiC quantum dots film over large areas (substrate) which has been predicted to have a crystalline quantum dots structure. Very high frequency plasma enhanced chemical vapour deposition, 150MHz (VHF-PECVD) technique will be utilized to deposit SiC quantum dots film in which variation in chamber pressure, substrate temperature, RF power and precursor gases flow rate will be investigated in order to get a crystalline quantum dots with uniform size and distribution. To the best of our knowledge, the report on SiC quantum dots structure growth by VHF-PECVD is not widely discovered.

Characterization of the growth quantum dots will be carried out to study the surface morphology, structural information and elemental composition using various techniques that serve the required characterization purposes. SiC quantum dots are expected to deposit with uniform size and distribution. Moreover, it is predicted to possess better structural as well as preserving the superior crystalline properties.

1.3 Problem Statement

Silicon carbide has many superior properties which make it a good candidate for semiconductor applications such as power transistor, thermoelectric, optoelectronics and as coating materials.

Many synthesis methods have been done by researchers in order to produce SiC quantum dots such as electrochemical technique, laser ablation, RF magnetron sputtering and electrochemical corrosion. Even though the size of quantum dots fabricated via RF magnetron sputtering has uniform and average size but it needs high power which is above 100 W and longer deposition time between 20 to 60 minute (Cheng *et al.*, 2007).

An ultrasmall quantum dots whose diameter below 1 nm had fruitfully obtained by electrochemical etching technique. The ultrasmall quantum dot was stem from bulk of SiC. However, only one type of SiC that can be produced from one bulk SiC. A different type of bulk SiC is needed in order to have various SiC polytypes (Fan *et al.*, 2012). Similar to the electrochemical technique, photo-assisted technique also produce one polytype at once depend on the SiC polytype powder used. Moreover, there is formation of oxidizing species which cannot be excluded which affected the properties of quantum dots produced. This method also needs an external bias potential to control the dots size (Mwania *et al.*, 2013).

6

Pulsed laser ablation technique had successfully produced 3C-SiC crystalline quantum dots with a uniform size of 2 nm. This technique needs high temperature and pressure during laser ablation which is analogous to thermal evaporation for SiC polytype from one (target) to another polytypes. For this technique, only one type of SiC polytype can be transformed at a time. In addition, the target used for this technique is expensive (Zhu *et al.*, 2014).

Based on previous research, the quantum dots formations are strongly dependent on the growth technique and the selected growth parameter. High temperature, pressure and power are needed to produce quantum dots. Some of the techniques can only produce one polytype at a time and depend on the polytypes of bulk or target used. In addition, the structural properties of SiC quantum dots obtained are not widely discussed and most researchers only focus on the fabrication technique and photoluminescence of the SiC quantum dots.

Therefore, in this research various polytypes of SiC quantum dots will be grown via VHF- PECVD at low temperature and limited growth time. The morphology and structural properties of the dots will be analyzed and discussed.

1.4 Research Objectives

- i. To grow silicon carbide quantum dots on silicon substrate using very high frequency chemical vapour deposition method.
- ii. To optimize the growth parameters (growth temperature, RF power, hydrogen flow rate and growth time) of silicon carbide quantum dots.
- iii. To characterize the morphology and structural properties of the deposited silicon carbide quantum dots.

1.5 Scope of Research

The fabrication of self-assembled SiC quantum dots was carried out using VHF-PECVD technique at 150 MHz in order to produce a crystalline quantum dots at low growth temperature and limited growth time. The Si substrate was used to grow the quantum dot. The structure was grown only limited to zero dimensions. The effect of varying growth temperature from 24 to 400°C, RF power from 10 to 25 W, hydrogen flow rate from 60 to 140 sccm and growth time from 1 to 5 minutes were observed and investigated. The morphology and structural properties of the SiC quantum dots grown were characterized by Atomic Force Microscopy, Field Emission Scanning Electron Microscopy, Fourier Transform Infra-Red spectroscopy, X-ray Diffraction Spectroscopy, Electron Dispersive X-ray, and Scanning/Transmission Electron Microscopy.

1.6 Significant of Research

Based on this research, the uniformity of dots size and distribution was controlled by 150MHz VHF-PECVD method. The different in polytypes and the size of the quantum dots are important for application in optoelectronics and microelectronics especially in extreme environment. This research would contribute to the understanding of the morphological and structural properties of SiC-quantum dots.

1.7 Thesis Outline

Chapter 1 presents some previous works that are related to this study. The problem statement, objectives, scope and significant of the research are also presented. In Chapter 2, literature reviews related with this research were discussed. This chapter covers the growth mechanism of quantum dots, Si and SiC as semiconductor materials, fundamental of PECVD method and the precursor gases used in SiC quantum dots growth. Chapter 3 described the methodology which included the fabrication of SiC quantum dots, operation procedure for VHF-PECVD system and followed by characterization techniques in order to observe the morphological and structural properties of SiC quantum dots. In Chapter 4, the results from each characterization were analysed. The influenced of each parameter upon quantum dots growth were critically studied. The conclusions and suggestions are stated in Chapter 5.

REFERENCES

- Abdesselem S., Aida M. S., Attaf N., Ouahab A., (2006). Growth Mechanism of Sputtered Amorphous Silicon Thin Films. *Physica B* 373: 33-41.
- Ameruddin A. S. (2010). Growth Of Islands And Quantum Dots Of Germainium And Indium Gallium Arsenide. MSc Thesis UTM.
- Andrieviski R. A., (2009). Synthesis, Structure And Properties Of Nanosized Silicon Carbide. *Review Advance Materials Science* 22: 1-20.
- Arivazhagan V., (2013). Investigation Of Quantum Confinement Effect In PbSe/ZnSe Multiple Quantum Well Structures Prepared By Thermal Evaporation Technique. PhD Thesis Karunya University.
- Arulsamy A. D., Rider A. E., Cheng Q. J., Xu S., Ostrikov K., (2009). Effect of Elemental Composition And Size On Electron Confinement in Self-Assembled SiC Quantum Dots: A Combinatorial Approach. *Journal of Appllied Physics* 105: 1-9.
- Aryanto D. (2009). Structural and Optical Properties of Self-Assembeled Indium Gallium Arsenide Quantum Dots. MSc Thesis UTM.
- Baraba'si A. L., (1999). Thermodynamic and kinetic mechanisms in selfassembled quantum dot formation. *Materials Science and Engineering* 67: 23–30.
- Bechstedt F., Fisselb A., Furthmu⁻llera J., Kaiserb U., Weisskera H. C., Weschb W., (2003). Quantum structures in SiC. Applied Surface Science 212–213: 820–825.
- Bin L., Sheng S. G., Fang L. X., Feng Z., Lin D., Liu Z., Guo Y. G., Bei L. S., Shun Z. W., Lei W., Ping Z. Y., Guang L. X., Guo W. Z, Fei Y.,

(2013). Fast Homoepitaxial Growth of 4H-SiC Films on 4° off-Axis Substrates in a SiH4-C2H4-H2 System. *Chinese Physics Letter* 30: 12.

- Cao G., (2004). Nanostructures and Nanomaterials Synthesis, Properties and Applications. Imperial College Press.
- Castelletto S., Johnson B. C., Parker A., (2013). Silicon Carbide: An Advanced Platform For Next Generation Quantum Devices. *Quantum Communications and Quantum Imaging* 8875: 88750.
- Chelikowsky J. R., (2004). *Electronic Structure of Clusters and Nanocrystals*. MSc Thesis University of Minnesota.
- Chen E., Du G., Zhang Y., Qin X., Lai H., Shi W., (2014). RF-PECVD Deposition And Optical Properties Of Hydrogenated Amorphous Silicon Carbide Thin Films. *Ceramic International* 2: 67.
- Cheng Q. J., Long J. D., Xu S., (2007). Growth Dynamics and Characterization of SiC Quantum Dots Synthesized by Low-Frequency Inductively Coupled Plasma Assisted Rf Magnetron Sputtering. *Journal of Applied Physics* 101: 1-7.
- Choi H., (2012). Overview of Silicon Carbide Power Devices. *Fairchild Semiconductor*.
- Chowdhury A., Mukhopadhyay S., Ray S., (2007). Structural and Transport Properties of Nanocrystalline Silicon Thin Films Prepared at 54.24 MHz Plasma Excitation Frequency. *Journal of Crystal Growth* 304 (2): 352–360.
- Drbohlavova J., Adam V., Kizek R., Hubalek J., (2009). Quantum Dots —
 Characterization, Preparation and Usage in Biological Systems.
 International *Journal of Molecular Sciences 10*: 656-673.
- Eaton P., West P., (2010). *Atomic Force Microscopy*. United State: Oxford University Press Inc. New York.

- Ewen S., Dent G. (2005). *Modern Raman Spectroscopy: A Practical Approach*. John Wiley & Son Ltd.
- Fan J. Y., Wu X. L., Chu P. K., (2006). Low-Dimensional SiC Nanostructures: Fabrication, Luminescence, and Electrical Properties. *Progress in Material Science* 5: 983–1031.
- Fan J., Li H., Wang J., Xiao M., (2012). Fabrication and Photoluminescence of SiC Quantum Dots Stemming from 3C, 6H, and 4H Polytypes of Bulk SiC. *Applied Physics.Letter* 101: 3-8.
- Forsberg U., (2001). CVD Growth of Silicon Carbide for High Frequency Applications. PhD Thesis Lingko pings University.
- Fukawa M., Suzuki S., Guo L., Kondo M., Matsuda A., (2001). High Rate Growth of Microcrystalline Silicon Using a High-Pressure Depletion Method with VHF Plasma. *Solar Energy Material and Solar Cells* 66: 217–223.
- Gupta A., Paramanik D., Varma S., Jacob C., (2004).CVD Growth And Characterization Of 3C-Sic Thin Films. *Bulletin of Material Science* 27: 445-451.
- Gurvinder S.B., Rakesh M.S. (2016). Raman Spectroscopy-Basic Principle, Instrumentation and Selected Applications for the Characterization of Drug of Abuse. *Egytian Jurnal of Forensic Science* 6: 2016-215.
- Gusev A. S., Ryndyab S. M., Zenkevicha A. V., Kargina N. I., Averyanova D. V., Grekhova M. M., (2015). Research Of Morphology And Structure Of 3C–Sic Thin Films On Silicon By Electron Microscopy And X-Ray Diffractometry. *Modern Electronic Materials* 1: 120-125.
- Habuka H., Ohmori H., Ando Y., (2017). Silicon Carbide Film Deposition At Low Temperatures Using Monomethylsilane Gas. Surface and Coating Technology 204: 1432–1437.

- Harris G.L., (1995). *Properties of Silicon Carbide*. London: The Institution of Electrical Engineers.
- Harrison P., (2005). Quantum Well, Wire and Dots: Theoretical and Computational Physics of Semiconductor Nanostructure. England : John Wiley & Sons.
- Hasim N., (2017). Optical and Catalyst Studies of Erbium/Neodymium Coodoped Lithium Niobate Tellurite Glass Embedded With Silver Nanoparticle. PhD Thesis UTM
- Hayat M. (2000). Principle and Technique of Electron Microscopy, Biological Applications 4th Edition. Cambridge: Cambridge University Press.
- Hu Z., Liao X., Diao H., Kong G., Zeng X., Xu Y., (2004). Amorphous Silicon Carbide Films Prepared by H₂ Diluted Silane-Methane Plasma. *Journal of Crystal Growth* 264: 7–12.
- Iwanowski I. J., Fronc K., Paszkowicz W., Heinonen M., (1999).XPS and XRD Study Of Crystalline 3C-Sic Grown By Sublimation Method. *Journal of Alloys and Compounds* 286: 143–147.
- Izhevskyi V. A., Genova L. A., Bressiani J. C., Bressiani A. H. A., (2000). Review article: Silicon Carbide Structure, Properties and Processing. *Cerâmica* 46: 297.
- Julian H. (2015). Energy Dispersive Spectroscopy 2nd Edition. Chichester: John Wiley & Son Ltd.
- Katharria Y. S., Kumar S., Prakash R., Choudhary R. J., Singh F., Phase D.
 M., Kanjilal D., (2007). Characterizations of Pulsed Laser Deposited SiC Thin Films. *Journal of Non-crystalline Solid* 353: 4660 – 4665.
- Kazmirak V. (2012). Scanning Electron Microscopy. Europe: In Tech Publisher.

- Kimoto T., (2016). Bulk And Epitaxial Growth Of Silicon Carbide. Progress in Crystal Growth and Characterization of Materials 62: 329–351.
- King S. W., French M., Bielefeld J., Lanford W. A., (2011). Fourier Transform Infrared Spectroscopy Investigation of Chemical Bonding in Low-k a-SiC:H Thin Films. *Journal of Non- Crystalline Solids* (375): 2970–2983.
- Latha H. K. E., Udayakumar A., Prasad V. S., (2010). Microstructure and electrical properties of nitrogen doped 3C–SiC thin films deposite dusing methyltrichlorosilane. *Materials Science in Semiconductor Processing* 29: 117-123.
- Li J., Shirai T., Fuji M., (2014). Silicon Carbide and Its Nanostructures. 先進 セラミックス研究センター年報 3:5-10.
- Manasreh O., (2011). Introduction to Nanomaterials and Devices. John Wiley & Son.
- Markov I. V., (2003). Crystal Growth for Beginners 2nd Edition: Fundamental of Nucleation, Crystal Growth and Epitaxy. Singapore : World Scientific Publishing.
- Marckx D.A., (2005). *Breakthrough in Power Electronics from SiC*. United States of America : NREL Technical Monitor.
- Matsuda A., Takai M., Nishimato M. K., (). Control of Plasma Chemistry for Preparing Highly Stablized Amorphousat High Growth Rate. Solid Energy Matter Solar Cells 78 (1): 3-26.
- Matsui T., Matsuda A., Kondo M., (2006). High-rate microcrystalline silicon deposition for p-i-n junction solar cells. *Solar Energy Material and*. *Solar Cells* 90: 3199–3204.

Melinon P., (2011). SiC Cage Like Based Materials. Croatia : Intech Europe.

- Monaghan E., (2014). VHF-PECVD and Analysis of Thin Nano-crystalline Silicon Films With a Multi-tile Plasma Source for Solar Energy Applications. PhD Thesis Dublin City University.
- Morris P. (2014). *Biomedica Imaging: Applications and Advance*. Wordhead Publishing.
- Mukherjee M., (2011). Silicon Carbide Materials, Processing and Applications in Electronic Devices. Intech Open (Europe).
- Mwania M., Janáky C., Rajeshwar K., Kroll P., (2013). Fabrication of β-SiC Quantum Dots by Photo-Assisted Electrochemical Corrosion of Bulk Powders. *Electrochemical Communication* 37: 1-4.
- Nakashima S., Nishimori A., Masuda Y., Sano H., Sorai M., (1991). Fe Mossbauer, ir and Raman spectra and heat capacity of the mixedvalance complex, 1',1'''-diethylbiferrocenium triiodide, showing fusion type averaging process of the mixed valence state. *Journal of Physics and Chemistry of Solids* 52: 1169-1180.
- Nakashima S., Higashihira M., Maeda K., (2003). Raman Scattering Characterization of Polytype in Silicon Carbide Ceramics: Comparison with X-ray Diffraction. *Journal of America Ceramic Society* 86 (5): 823–829.
- Omar M. F., (2015). Design and Development of VHF-PECVD for Nanostructured Silicon Carbide Thin Film Deposition. PhD Thesis UTM.
- Partha P. D., Khare A., (2016). Effect Of Substrate Temperature On Structural And Linear And Nonlinear Optical Properties Of Nanostructured PLD A-Sic Thin Films. *Materials Research Bulletin* 84: 105–117.

- Partha P. D., Khare A., (2016). Effect of Substrate Temperature on Structural And Linear And Nonlinear Optical Properties Of Nanostructured PLD a-SiC Thin Films. *Materials Research Bulletin* 84: 105-117.
- Peri B., Borah B., Dash R. K., (2014). Effect of RF Power and Gas Flow Ratio on The Growth and Morphology of The PECVD SiC Thin Films for MEMS Applications. Bulletin Materials of Science 38 (4): 1105-1112.
- Pluchery O., Costantini J., (2012). Infrared spectroscopy characterization of 3C–SiC epitaxial layers on silicon. *Journal Of Physics D: Applied Physics* (45):495101.
- Prendergast R., (2010). Structural Determination of Small and Large Molecule using Single Crystal X-ray Crystallography. MSc Thesis University of Manchester.
- Reddy J. D., (2008). *Mechanical properties of Silicon Carbide (SiC) Thin Films*. PhD Thesis University of South Florida.
- Saddaw S. E., (2012). Silicon Carbide Biotechnology: A Biocompatible Semiconductor for Advanced Biomedical Devices and Applications. Elsevier Inc.
- Saripalli S., (2008). Transport Properties in Nanocrystalline Silicon and Silicon Germanium. PhD Thesis Iowa State University.
- Shi D., Guo Z., Bedford N., (2015). Semiconductor Quantum Dots. Elsevier Inc.
- Tiwari J. N., Tiwari R. N., Kim K. S., (2012). Zero-Dimensional, One-Dimensional, Two-Dimensional and Three-Dimensional Nanostructured Materials for Advanced Electrochemical Energy Devices. *Progress in Material Science* 57(4): 724–803.

- Wang Y., Lin J., Huan C. H. (2002). Multiphase structure of hydrogenated amorphous silicon carbide thin films. *Materials Science and Engineering* 95: 43–50.
- Wasyluk J., Peroval T. S., Kukushkin S. A., Osipov A. V., Feoktistov N. A., Grudinkin S. A., (2010). Raman investigation of different polytypes in SiC thin films grown by solid-gas phase epitaxy on Si (111) and 6H-SiC substrates. *Materials Science Forum* 645-648: 359-362.
- Willander M., Friesel M., Wahab Q., Straumal B., (2006). Silicon Carbide and Diamond for High Temperature Device Applications. *Journal of Material Science: Materials in Electronics* 7: 1–25.
- Wu R., Zhou K., Yue C. Y., Wei J., Pan Y., (2015).Recent Progress In Synthesis, Properties And Potential Applications Of SiC Nanomaterials. *Progress in Materials Science* 72: 1–60.
- Yan G., Zhang F., Niu Y., Yang F., Liu X., Wang L., Zhou W., Sun G, Zeng Y., (2015). Effect of Hydrogen Flow rate on Growth of 3C-SiC heteroepitaxial Layers on Si (111) Substrates. *Applied Surface Science* 6: 172.
- Younus M. H., Khalajabadi S. Z., Abu A., Ameen O. F., Ahmad N., Redzuan N., Ibrahim R. K. R., (2018). Synthesis and characterization of nanocrystalline silicon carbide thin films on multimode fiber optic by means 150MHz VHF-PECVD. *International Journal of Biosensors & Bioelectronics* 4: 1.
- Zachbroeck V., & Bart J., (1997). Density of States Calculation. From http://ecee.colorado.edu/~bart/book/dos.htm
- Zhang H., Ding W., He K., Li M., (2010). Synthesis and Characterization of Crystalline Silicon Carbide Nanoribbons. *NanoScale Express Letter* (5): 1264–1271.

- Zhou W., Coleman J.J., (2016). Semiconductor quantum dots. *Current* Opinion in Solid State and Materials Science : 1359-0286
- Zhu J., Hu S., Xia W. W., Li T. H., Fan L., Chen H. T., (2014). Photoluminescence of ~2 nm 3C-SiC Quantum dots Fabricated from Polycrystalline 6H-SiC Target by Pulsed Laser Ablation. *Material Letter* 132: 210-213.