KAOLIN STABILISATION BY UTILISING GROUND GRANULATED BLAST SLAG FOR CARBON DIOXIDE REDUCTION

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DEDICATION

To my late Mother Somaya

Who taught me to believe in myself, who had been the kindest soul and the gentle voice calling me everyday to infuse me with patience, courage and determination to complete this research. You are not here with me today to witness the results of my work but I know you will be proud.

To my Father Mohammed

Who taught me hard work always pays off and to strive for the best. Who has been the source of strength and resilience for me and my siblings in the most difficult of times. Your love and faithful prayers carried me to complete this thesis to the end.

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ABSTRACT

The aim of this study is to identify the effect of ground granulated blast furnaces slag (GGBS) in stabilising kaolin using the accelerated carbonation process while sequestering CO₂ to reduce its level in the atmosphere. The effects of incorporating GGBS and carbonated GGBS on mechanical properties of soil and the underlying mechanisms of stabilisation are investigated by carrying out physical characterisations, strength test and microstructure analysis (i.e., Field Emission Scanning Electron Microscope, X-Ray Diffraction). In addition, thermogravimetric analysis (TGA) and Acid Digestion method were used to quantify the CO2 that has been sequester in the soil. The results show that GGBS was found to be a good candidate for kaolin stabilisation (under ambient condition) in which the strength keeps increasing with time, and as the content of GGBS increases. 20% and 25% GGBS-kaolin mixtures strength have exceeded the value requires for the subgrade layer at 7 curing days. The highest strength recorded was for 25% GGBS-kaolin mixture at 28 days curing as the strength reached 8.4 MPa. For the mineral carbonation process, the outcome indicates that the strength rises as the carbonation period rises, and a substantial strength gain was noted during the first hour of the carbonation phase. Likewise, unconfined compressive strength (UCS) rises as the CO₂ pressure rises from 100 kPa to 200 kPa. Acid digestion method and TGA shows that the GGBS-kaolin mixtures sequester between 4.5 to 5% CO₂ from its original dry weight. XRD analysis shows no peaks were observed for the magnesium-carbonated products. On the other hand, a single peak of calcium-carbonated products was seen in the diffractogram of the carbonated GGBS-kaolin mixture. Field emission scanning electron microscopy (FESEM) analysis for carbonated samples shows a needle-like structures seen on the surface of the kaolin particles which is Aragonite (CaCO₃). It could be concluded that augmentation of the strength is due to carbonated calcium and magnesium products which stuff the soil voids.

ABSTRAK

Tujuan kajian ini adalah untuk mengenal pasti kesan terak tungku letupan berbutir tanah (GGBS) dalam proses menstabilkan kaolin dengan menggunakan kaedah proses karbonasi dipercepatkan disamping mengasingkan CO₂ supaya dapat mengurangkan tahap intensitinya di atmosfera. Kesan kombinasi GGBS tidak berkarbonat dan GGBS berkarbonat pada sifat mekanikal tanah dan mekanisme penstabilannya telah dikenalpasti dengan menjalankan ujian ciri fizikal tanah, ujian kekuatan dan analisis struktur (mikro Imbasan mikroskop Electron (FESEM), Serakan Tenaga X-Ray). Sebagai tambahan, analisis Gravimetrik Terma (TGA) dan kaedah Pencernaan Asid digunakan untuk mendapatkan kuantiti CO₂ yang telah diasingkan dalam pengstabilan tanah. Hasil kajian menunjukkan bahawa GGBS terbukti menjadi bahan yang baik dalam proses pengstabilan kaolin (dalam keadaan tidak berkarbonat) dimana kekuatannya terus meningkat seiring tempoh masa yang semakin lama, dan pada kandungan GGBS yang kian meningkat. Kekuatan campuran 20% dan 25% GGBS-kaolin telah melebihi nilai yang diperlukan untuk lapisan paling bawah permukaan tanah pada 7 hari tempoh pengawetan. Kekuatan GGBS-Kaolin tertinggi yang dicatatkan adalah pada campuran 25% GGBS-kaolin dalam 28 hari pengawetan iaitu mencapai kekuatan sehingga 8.4 MPa. Manakala bagi proses karbonasi mineral, keputusan kajian menunjukkan bahawa kekuatan meningkat sepanjang tempoh karbonasi meningkat dan peningkatan kekuatan yang banyak dicatat pada jam pertama fasa karbonasi (3jam). Begitu juga keputusan kekuatan mampatan tidak terkurung kekuatan mampatan yang tidak terkawal (UCS) semakin meningkat ketika tekanan CO2 meningkat daripada 100 kPa hingga 200 kPa. Kaedah pencernaan asid dan TGA pula menunjukkan bahawa GGBS-kaolin mengasingkan antara 4.5% hingga 5% CO₂ daripada berat asal kering. Analisis XRD menunjukkan tidak ada puncak yang diperhatikan untuk produk berkarbonat magnesium. Sebaliknya, satu puncak produk kalsium berkarbonat dilihat dalam diffractogram campuran GGBS-kaolin berkarbonat. Analisis mikro Imbasan mikroskop electron (FESEM) untuk sampel berkarbonat menunjukkan struktur seperti jarum yang dilihat pada permukaan zarah kaolin yang merupakan Aragonite (CaCO₃). Kesimpulannya, peningkatan kekuatan disebabkan oleh kandungan kalsium dan magnesium berkarbonat yang mengisi kekosongan partikel dalam tanah

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LIST OF ABBREVIATIONS

Al - Aluminum

ASTM - American society of testing material

BS - British standard
CaO - Calcium oxide

CaCO₃ - Calcite

CSH - Calcium silicate hydrate

 $\begin{array}{cccc} CO_2 & - & Carbon \ dioxide \\ E_{50} & - & Secant \ Modulus \\ G_s & - & Specific \ Gravity \\ MgO & - & Magnesium \ Oxide \\ \end{array}$

 $MgCO_3$ - Magnesite

PI - Plasticity Index
PL - Plastic limit

SEM - Scanning electron microscopy

SiO₂ - Silica

TGA - Thermogravimetric analysis

UCS - Unconfined compressive strength test

XRD - X-ray diffraction

CHAPTER 1

INTRODUCTION

1.1 Background

Malaysia is known as a developing country that will develop more in the next few years. This development is due to population growth, which required more building, roads, and other facilities. Tropical countries such as Malaysia experience humid and wet weather. Ground tropical soils are generally affected by moisture content fluctuation, which causes a volume change. Hence, it leads to a low strength capacity that causes structural failure.

Chemical stabilisation using pozzolanic material that has involves a significant amount of calcium, silica, and aluminum in the presence of water was proven to be very useful in improving soil strength (Walker and Pavia, 2011). The most common pozzolanic materials used for this purpose are cement and lime. Cement is preferable as it is cheaper and more effective than lime (Raftari et al., 2014). Nevertheless, in Malaysia, lime in the form of hydrated lime is widely used; it can be found easily and abundantly, making it an economical choice for stabilising tropical soil. However, cement and lime production comes with disadvantages, such as the high amount of carbon dioxide (CO₂) emits during the process and the fact that they are not sustainable materials. (Phummiphan et al., 2018). In the late 1940s, the need to find alternative sustainable materials becomes essential.

The necessity to find alternative materials to replace ordinary Portland cement (OPC) partially or fully is essential, especially in Malaysia, where green and recycled material has been suggested. Among them, fly ash and bottom ash from the power plant (Abubakar A.U. & Baharudin K.S., 2012), ground granulated blast furnace slag (GGBS), which is a by-product from iron production, and other by-product materials (Pathak et al., 2014). Fly ash and ground granulated blast furnaces dust are pozzolanic

materials according to ASTM: C-618 and ASTM: C-989, respectively (Hashim et al., 2015). Bagasse ash, a by-product of natural agricultural waste generated during bagasse combustion in the sugar industry (Jamsawang et al., 2017), and waste marble powder is a by-product of the marble industry (Choksi, Mishra and Patel 2018).

Malaysia generates about 22,000 tons of waste material per day, as reported in 2012; this quantity will rise annually to reach more than 30,000 tons per day by 2020. (Zainuddin et al., 2016) The essential sources for those waste materials are construction sector solid waste, representing about 41% of Malaysia's total waste, followed by industrial, household, commercial, and institutions, representing about 25% and 33%. It has been recorded that 76% of solid waste is successfully collected in Malaysia. 5% of the collected waste has been recycled while the remaining have been disposed to 112 landfills (Papargyropoulou et al., 2011).

It is well known that waste or by-product materials are considered to be most likely a toxic, ignitable, corrosive, or reactive pose that can cause severe damage to human health and the environment. Alternatively, the utilisation of these materials is an option in soil stabilisation instead of been disposed of (Rasool and Kapoor, 2016).

Currently, amidst the world climate change consideration, mineral carbonation (MC) is the most promising methodology for permanently storing CO₂. It encompasses minerals comprising calcium (Ca) and magnesium (Mg) with CO₂ to produce Ca/Mg carbonated products. It has been proven to enhance soil strength (He et al., 2013; Mun and Cho, 2013; Hu et al., 2017; Jimoh et al., 2018).

GGBS is considered as a waste material generates from cooling molten slag in water or steam it followed by drying it (Mujtaba et al., 2018). It is obtainable in Malaysia, and it is mainly composed of calcium, magnesium, silicates, and aluminum (Islam et al., 2014). GGBS is the primary stabiliser in this study.

1.2 Problem Statement

Kaolin clay is a tropical clay associated with volume change due to the changes in moisture content. It is also known for its low shear strength, high plasticity, and poor workability; such a soil cannot afford the load from structures and vehicles to a certain extent. For this matter, the need for soil improvement/stabilisation is essential to overcome the problems.

OPC is used globally to treat different soil types. However, OPC's global production is responsible for 5% to 8% of the total CO₂ emission worldwide. Producing 1 ton from OPC leads to 0.85 to 0.95 t CO₂ emission. CO₂ is considered the major greenhouse gas, including methane (CH₄) and nitrous oxide (N₂O). The CO₂ causes 50% of the global warming impacts on the climate change temperature and sea level; those impacts is considered one of the main causes of extinction. The other concern about OPC is the energy consumed, about 12 to 15% of the industrial area's energy. Compared to OPC, 1 ton of GGBS leads to 0.05 to 0.07 t CO₂ emission, much lower than OPC.

Globally annual emission of CO₂ increased from 17.78 billion tons in 1980 to reach 32.1 billion tons. This increment is about 100% in 35 years and is expected to reach double this value in 2035. In Malaysia, CO₂ emission increased by 69% from 2000 to 2020 to reach about 285.7 million tons (Jorat et al., 2018).

In the recent years, a new chemical stabilization technique that is more environmentally friendly has been introduced. The technique involves the use of calcium (Ca⁺) and magnesium (Mg⁺) rich material (cement, serpentine, olivine, wollastonite, steel slag, and reactive magnesia cement). These materials have the potential to absorb and sequester CO₂ permanently. Thus, cause the reduction of CO₂ in the atmosphere while beneficial for increasing the soil strength quickly. The process also is favorable as it does not rely much on pozzolanic reaction as conventional soil stabilisers.

1.3 Aim and Objectives of Study

This study aims to identify the effect of ground granulated blast furnaces slag in stabilising kaolin using the accelerated carbonation process while sequestering CO₂ to reduce its level in the atmosphere. In order to achieve this aim, the following objectives are conducted:

- 1. To evaluate the effect of GGBS contents on the strength of the kaolin clay under different curing period.
- 2. To evaluate the effect of carbon dioxide pressures on the treated kaolin clay with the effective content of GGBS under different carbonation period.
- 3. To quantify the content of carbon dioxide that has been sequestered by the stabilise soil.
- 4. To identify the changes occurs in the microstructure and mineralogical characteristics for the treated soil under ambient and carbonation condition.

1.4 Scope of Study

The scope of this study can be divided into the material to be used in this research and the laboratory tests:

1.4.1 Material

The soil used in this study is kaolin clay. It is considered as an unsuitable material for construction purposes as it is considered a weak soil due to its low strength. It is available in Malaysia and can be obtained in two colors; white without a sufficient amount of iron and changes to brown when iron is acceptable. GGBS is used

as a stabiliser to improve the engineering characteristics of kaolin. GGBS has both calcium (Ca⁺) and magnesium (Mg⁺) in its chemical composition, known for their ability to react and sequester CO₂ as carbonated products.

1.4.2 Testing

Physical tests were carried out on the untreated kaolin clay to determine its properties and classification. A compaction test will be carried out on the treated kaolin clay with different content (5%, 10%, 15%, 20%, and 25%) of GGBS to determine the optimum moisture content and the maximum dry density, which were used later to prepare the samples for unconfined compressive strength (UCS). No further addition of GGBS was conducted (beyond 25%) as the strength already achieved the requirement limit at earlier curing. Also, the fact that the load cell capacity reaches its highest limit but the sample did not fail at 25% GGBS cured kaolin for 14 days. After preparing the samples, those samples were cured for 0, 7, 14, and 28 days under ambient conditions before the testing. The effective content of GGBS stabilised kaolin was then identified.

Chemical compositions of lime and cement from various studies show that lime has calcium oxide (CaO) between 80% to 90% from its total chemical composition, while cement with its different types has a CaO content between 60% to 70%. The chemical composition of GGBS from previous literature shows that the content of CaO in this waste material is between 40% to 45%, which half of the amount of CaO in lime or cement, respectively. Thus, it can be concluded that the initial content of GGBS for soil stabilisation to occur was 5%, which is twice the one used in lime and 1% greater than cement.

Next, different samples were prepared with the effective content of GGBS (20% and 25%) and cured for 0 and 7 days under ambient conditions. Those samples were then placed in a carbonation cell; the carbonation was performed by applying different CO₂ pressure (100 and 200 kPa) for other carbonation period (1, 3, 6, and 24 hours).

The quantification of CO₂ sequestered within GGBS-kaolin mixtures was determined using two methods: The acid digestion method using 10% nitrite acid solution (chemical method) and thermogravimetric analysis (TGA) (thermal method). Microstructure analyses such as X-ray Diffraction (XRD), Field Emission Scanning Electron Microscope (FESEM), and X-ray fluorescence (XRF) are conducted on selected samples to determine the changes in soil structure and its chemical composition.

1.5 Significance of Study

This study is contributed to the chemical stabilisation of kaolin clay with the non-traditional stabiliser represent in this study by GGBS. GGBS has a significant amount of Calcium (Ca⁺), so it can be used alone as a stabiliser due to its cementitious and pozzolanic properties. However, since GGBS has a low hydration rate, it leads to a low strength gained at the early stages. GGBS is considered as waste material with low CO₂ emission and low energy during manufacturing. Thus, it leads to a dual advantage; it will be cost-effective and environmentally friendly.

Previous studies have shown that magnesium and calcium can react with CO₂ and sequester it in the form of carbonates (He et al., 2013; Hu et al., 2017; Jimoh et al., 2018). This process is known as carbonation or mineral carbonation. It has been used effectively for the permanent storage of CO₂. Calcium carbonation that involves the reaction between calcium hydroxide (Ca(OH)₂) and carbon dioxide (CO₂) to produce calcium carbonate known as calcite with formula CaCO₃ have shown a more significant effect in enhancing soil strength at early stages. However, the reaction between calcium silicate hydrate C-S-H with CO₂ reduces soil strength. Materials rich with calcium can be used to carbonate and improve soil strength, such as cement (Nakarai and Yoshida, 2015; Cao et al., 2017; Ho et al., 2017). Magnesium carbonation using magnesium-rich material (reactive magnesia and natural olivine) have been proven to be very useful in soil stabilisation (Yi et al., 2013a; Yi et al., 2013b; Y.L et al., 2013; Cai et al., 2015; Guang-hua et al., 2015; Yi et al., 2016; Cai and Liu, 2017; Liu et al., 2017; Song-yu et al., 2017).

GGBS, with its high content of CaO about 47.5% and reasonable MgO of about 4.9%, can be used as an alternative for cement, reactive magnesia, and olivine as it has another advantage (by-product material) over them. Calcium (Ca) or magnesium (Mg) rich materials as cement, reactive magnesia cement or olivine have the ability to sequester the CO₂ by the reaction between CO₂ and calcium or magnesium. This reaction is a chemical expansion reaction that fills the soil's voids, which increases soil strength and durability. The ability of such an effect is being tested for ground granulated furnaces slag by conducting different laboratory tests.

Nonetheless, limited studies was conducted on soil stabilisation using carbonation where only cement (industrial material) (Nakarai and Yoshida, 2015; Ho et al., 2017) reactive magnesia (industrial material) (Yi et al., 2013a; Yi et al., 2013b; Y.L et al., 2013; Cai et al., 2015; Guang-hua et al., 2015; Yi et al., 2016; Cai and Liu, 2017; Liu et al., 2017; Song-yu et al., 2017) and olivine (natural material) (Fasihnikoutalab et al., 2016a; Fasihnikoutalab et al., 2016b; Fasihnikoutalab et al., 2017) have been used previously. Up to date, there are no studies emphasized on the utilisation of GGBS as stabiliser material with consideration of its effect on carbonation. This study contributes to calcium and magnesium carbonation to enhance the soil strength while reducing CO₂ levels in the atmosphere by using sustainable material (GGBS). GGBS has been used widely in soil stabilisation due to its significant amount of calcium alone and other materials. GGBS can be used as an alternative for cement, reactive magnesia, and olivine as it has another advantage (by-product material) over them. Therefore, in this study, the abovementioned is covered.

1.6 Thesis Outline

This thesis consists of five chapters. Chapter one consists of an introduction on waste generation, the probability of using pozzolanic waste material as a stabiliser in soil stabilisation, and the new method for soil stabilisation using minerals carbonation, enhancing soil strength and sequestering CO₂. This chapter also includes the problem statement, objectives of the study, scope of the study, and significant of the study.

Chapter two provides an overview of kaolin formation, GGBS, waste management in Malaysia, and ground improvement role in CO₂ reduction. The role of calcium and magnesium carbonation in enhancing soil strength and the factors affecting it from previous findings is discussed in detail. Moreover, the methods which are used in the quantification of sequester CO₂ are introduced.

Chapter three includes the laboratory tests used in this study to achieve its objectives and the standards used for each test. The methodology used to set up the carbonation cell used to carbonate different mixtures of GGBS-kaolin. The criteria used to accelerate the carbonation process is also discussed.

Chapter four analyses and discuss the result obtained from the laboratory testing designed in chapter three. Macrostructure and microstructure behaviour, together with an integration of the basic engineering properties of treated kaolin and the strength behaviour under ambient conditions and carbonation process, are also discussed and compared to previous findings are included in this chapter.

The last chapter summarizes the whole result as an integral part of the study. It also provides recommendations for future works related to this topic.

REFERENCES

- A.Sanna, M.Uibu, G.Caramanna, R.Kuusik, and Maroto-Valer, M. M. (2014). A review of Mineral Carbonation Technologies to sequester CO₂. *Chemical Society Reviews*, 43, 1–49. https://doi.org/10.1039/C4CS00035H
- Abas, M. A., and Wee, S. T. (2014). The Issues of Policy Implementation on Solid Waste Management in Malaysia. *International Journal of Conceptions on Management and Social Sciences*, 2(3), 12–17.
- Abubakar A.U. & Baharudin K.S. (2012). Potential Use of Malaysian Thermal Power Plants Coal Bottom Ash in Construction. *International Journal of Sustainable Construction Engineering & Technology*, 3(2), 25–37.
- Aimaro, S., and Mercedes, M. (2017). CO₂ Sequestration By Ex-situ Mineral Carbonation (1st ed.). World Scientific Publisher
- Andreani, M., Luquot, L., Gouze, P., Godard, M., Hoise, E., and Gibert, B. (2009). Experimental Study of Carbon Sequestration Reactions Controlled by the Percolation of CO₂ -Rich Brine through Peridotites. *Environmental Science & Technology*, 43(4), 1226–1231.
- Attahiru, Y. B., Aziz, M. M. A., Kassim, K. A., Shahid, S., Bakar, W. A. W. A., NSashruddin, T. F., ... Ahamed, M. I. (2019). A review on green economy and development of green roads and highways using carbon neutral materials. *Renewable and Sustainable Energy Reviews*, 101, 600–613.
- Azhar, A. T. S., Fazlina, M. I. S., Nizam, Z. M., Fairus, Y. M., Hakimi, M. N. A., Riduan, Y., and Faizal, P. (2017). Shear Strength of Stabilized Kaolin Soil Using Liquid Polymer. *IOP Conference Series: Materials Science and Engineering*, 226(1). https://doi.org/10.1088/1757-899X/226/1/012063
- Bachu, S., Bonijoly, D., Bradshaw, J., Burruss, R., Holloway, S., Christensen, N. P., and Mathiassen, O. M. (2007). CO₂ storage capacity estimation: Methodology and gaps. *International Journal of Greenhouse Gas Control*, 1, 430–443.

- https://doi.org/10.1016/S1750-5836(07)00086-2
- Baciocchi, R., Costa, G., Lategano, E., Marini, C., Polettini, A., Pomi, R., ... Rocca, S. (2010). Accelerated carbonation of different size fractions of bottom ash from RDF incineration. *Waste Management*, 30(7), 1310–1317. https://doi.org/10.1016/j.wasman.2009.11.027
- Baciocchi, Renato, Costa, G., Polettini, A., and Pomi, R. (2009). Influence of particle size on the carbonation of stainless steel slag for CO₂ storage. *Energy Procedia*, *1*(1), 4859–4866. https://doi.org/10.1016/j.egypro.2009.02.314
- Bagonza, S., Peete, J. M., Newill, D., and Freer-Hewish, R. (1987). Carbonation of Stabilised Soil-Cement and Soil-Lime Mixtures. *Transport Research Laboratory*, 29–48.
- Baioumy, H. M., and Ibrahim, A. R. (2012). Mineralogical Variations among the Kaolin Deposits in Malaysia. In *Annual International Conference on Geological & Earth Sciences* (pp. 133–140). https://doi.org/10.5176/2251-3361
- Baldovino, J. A., Moreira, E. B., Teixeira, W., Izzo, R. L. S., and Rose, J. L. (2018). Effects of lime addition on geotechnical properties of sedimentary soil in Curitiba, Brazil. *Journal of Rock Mechanics and Geotechnical Engineering*, 10(1), 188–194. https://doi.org/10.1016/j.jrmge.2017.10.001
- Baldyga, J., Henczka, M., and Sokolnicka, K. (2011). Mineral carbonation accelerated by dicarboxylic acids as a disposal process of carbon dioxide. *Chemical Engineering Research and Design*, 89(9), 1841–1854. https://doi.org/10.1016/j.cherd.2011.02.034
- Bessaim, M. M., Bessaim, A., Missoum, H., and Bendani, K. (2018). Effect of quick lime on physicochemical properties of clay soil. *MATEC Web of Conferences*, 149, 1–5. https://doi.org/10.1051/matecconf/201714902065
- Boot-Handford, M. E., Abanades, J. C., Anthony, E. J., Blunt, M. J., Brandani, S., Dowell, N. Mac, ... Fennell, P. S. (2014). Carbon capture and storage update. *Energy & Environmental Science*, 130–189. https://doi.org/10.1039/c3ee42350f

- BS5930:2015. (2015). Code of practice for ground investigations (BS 5930:2015. BSI Standards Limited 2015.
- Cai, G. H., Du, Y. J., Liu, S. Y., and Singh, D. N. (2015). Physical properties, electrical resistivity, and strength characteristics of carbonated silty soil admixed with reactive magnesia. *Canadian Geotechnical Journal*, 52, 1699–1713.
- Cai, G., and Liu, S. (2017). Compaction and Mechanical Characteristics and Stabilization Mechanism of Carbonated Reactive MgO-Stabilized Silt. *KSCE Journal of Civil Engineering*, 21, 2641–2654. https://doi.org/10.1007/s12205-017-1145-1
- Cao, Z., Zhang, T., and Zhang, D. (2017). Effect of carbonation on the engineering properties and microstructure of cement-stabilized lead-contaminated soils. *Geotechnical Frontiers*, 76(21), 526–533. https://doi.org/10.1007/s12665-017-7071-1
- Chakraborty, S., and Jo, B. W. (2018). *Aqueous-based carbon dioxide sequestration*. *Carbon Dioxide Sequestration in Cementitious Construction Materials*. Elsevier Ltd. https://doi.org/10.1016/b978-0-08-102444-7.00003-4
- Choksi, R., Mishra, C. B., and Patel, N. (2018). Pursuance of Waste Marble Powder to Improve Soil Stabilization. *International Research Journal of Engineering and Technology*, *5*(5), 1695–1698.
- Cuellar-Franca, R. M., and Azapagic, A. (2015). Carbon capture, storage and utilisation technologies: A critical analysis and comparison of their life cycle environmental impacts. *Journal of CO₂ Utilization*, 9, 82–102. https://doi.org/10.1016/j.jcou.2014.12.001
- Das, B. M. (2008). *Advanced Soil Mechanics*. *Journal of Medicinal Plants* (Third edit, Vol. 11). Taylor & Francis.
- Donnelly, F. C., Purcell-Milton, F., Framont, V., Cleary, O., Dunne, P. W., and Gun'ko, Y. K. (2017). Synthesis of CaCO₃ nano- and micro-particles by dry ice carbonation. *Chemical Communications*, 53(49), 6657–6660. https://doi.org/10.1039/c7cc01420a

- Du, Y. J., Wei, M. L., Reddy, K. R., and Wu, H. liang. (2016). Effect of carbonation on leachability, strength and microstructural characteristics of KMP binder stabilized Zn and Pb contaminated soils. *Chemosphere*, 144, 1033–1042. https://doi.org/10.1016/j.chemosphere.2015.09.082
- Eisazadeh, A., Kassim, K. A., and Nur, H. (2012). Stabilization of tropical kaolin soil with phosphoric acid and lime. *Natural Hazards*, *61*(3), 931–942. https://doi.org/10.1007/s11069-011-9941-2
- Elias, R. S., Wahab, M. I. M., and Fang, L. (2018). Retrofitting carbon capture and storage to natural gas- fired power plants: A real-options approach. *Journal of Cleaner Production*, 192, 722–734. https://doi.org/10.1016/j.jclepro.2018.05.019
- Eloneva, S., Teir, S., Salminen, J., Fogelholm, C. J., and Zevenhoven, R. (2008). Fixation of CO₂ by carbonating calcium derived from blast furnace slag. *Energy*, 33(9), 1461–1467. https://doi.org/10.1016/j.energy.2008.05.003
- Fagerlund, J., and Zevenhoven, R. (2011). An experimental study of Mg(OH)₂ carbonation. *International Journal of Greenhouse Gas Control*, 5, 1406–1412. https://doi.org/10.1016/j.ijggc.2011.05.039
- Fasihnikoutalab, M. H., Asadi, A., Huat, B. B. K., Ball, R. J., Pourakbar, S., and Singh, P. (2016). Utilisation of carbonating olivine for sustainable soil stabilisation.

 Journal of Environmental Geotechnics, (February). https://doi.org/10.1680/jenge.15.00018
- Fasihnikoutalab, M. H., Asadi, A., Huat, B. K., Westgate, P., Ball, R. J., and Pourakbar, S. (2016). Laboratory-scale model of carbon dioxide deposition for soil stabilisation. *Journal of Rock Mechanics and Geotechnical Engineering*, 8(2), 178–186. https://doi.org/10.1016/j.jrmge.2015.11.001
- Fasihnikoutalab, M. H., Asadi, A., Unluer, C., Huat, B. K., Ball, R. J., and Pourakbar, S. (2017). Utilization of Alkali-Activated Olivine in Soil Stabilization and the Effect of Carbonation on Unconfined Compressive Strength and Microstructure.

 Journal of Materials in Civil Engineering, 1–11. https://doi.org/10.1061/(ASCE)MT.1943-5533.0001833.

- Fasihnikoutalab, M. H., Westgate, P., Huat, B. B. K., Asadi, A., Ball, R. J., Nahazanan, H., and Singh, P. (2015). New Insights into Potential Capacity of Olivine in Ground Improvement. *Electronic Journal of Geotechnical Engineering*, 20, 2137–2148.
- Fernández Bertos, M., Simons, S. J. R., Hills, C. D., and Carey, P. J. (2004). A review of accelerated carbonation technology in the treatment of cement-based materials and sequestration of CO₂. *Journal of Hazardous Materials*, 112(3), 193–205. https://doi.org/10.1016/j.jhazmat.2004.04.019
- Gerdemann, S. J., Dahlin, D. C., O'Connor, W. K., Penner, L. R., and Rush, G. E. (2004). Ex-Situ and In-Situ Mineral Carbonation as a Means to Sequester Carbon Dioxide. In *Pittsburgh Coal Conference (PCC)*, 1249 Benedum Hall, University of Pittsburgh, Pittsburgh, PA, 15261, USA (pp. 1–17).
- Goh, K. M. (2004). Carbon sequestration and stabilization in soils: Implications for soil productivity and climate change. *Soil Science and Plant Nutrition*, *50*(4), 467–476. https://doi.org/10.1080/00380768.2004.10408502
- Goodarzi, A. R., Akbari, H. R., and Salimi, M. (2016). Enhanced stabilization of highly expansive clays by mixing cement and silica fume. *Applied Clay Science*, *132–133*, 675–684. https://doi.org/10.1016/j.clay.2016.08.023
- Guang-hua, C., Song-yu, L., Yan-jun, D., Ding-wen, Z., and Xu, Z. (2015). Strength and deformation characteristics of carbonated reactive magnesia treated silt soil. *Journal of Central South University*, 22, 1859–1868. https://doi.org/10.1007/s11771-015-2705-5
- Gupta, D., Priyadarshee, A., Sharma, S. K., and Singh, H. (2018). Behaviour of cement treated kaolin clay mixed with fiber and rice husk ash. *International Research Journal of Engineering and Technology*, 5(2), 250–255.
- H.Abdullah, H., A.Shahin, M., and Sarker, P. (2018). Use of Fly-Ash Geopolymer Incorporating Ground Granulated Slag for Stabilisation of Kaolin Clay Cured at Ambient Temperature. *Geotechnical and Geological Engineering*, 2. https://doi.org/10.1007/s10706-018-0644-2

- Hammond, G., and Jones, C. (2008). *Inventory of Carbon & Energy (ICE)*. Version 1.6a. *University of Bath*.
- Hänchen, M., Prigiobbe, V., Baciocchi, R., and Mazzotti, M. (2008). Precipitation in the Mg-carbonate system-effects of temperature and CO₂ pressure. *Chemical Engineering Science*, 63(4), 1012–1028. https://doi.org/10.1016/j.ces.2007.09.052
- Hariharan, S., Leopold, C., Werner, M. R., and Mazzotti, M. (2017). A Two-step CO₂ Mineralization Process. *Energy Procedia*, 114, 5404–5408. https://doi.org/10.1016/j.egypro.2017.03.1684
- Hasan, A., and Fatehi, P. (2019). Flocculation of kaolin particles with cationic lignin polymers. *Scientific Reports*, *9*(1), 1–12. https://doi.org/10.1038/s41598-019-39135-z
- Hasan, M., Alrubaye, A. J., Seng, L. K., Ideris, M. S., and Marto, A. (2016). Undrained Shear Strength of Soft Clay Mixed With Different Percentages of Lime and Silica Fume. *Jurnal Teknologi*, 78(8) 23–30. DOI: 10.11113/jt.v78.9606.
- Hashim, A. N., Razak, K. A., Abdullah, M. M. A. B., Nan, N. M. M., and Noor, N. A. M. (2015). Characteristics of Mortars with Various Composition of Ground Granulated Blast Furnace Slag. *Applied Mechanics and Materials*, 754–755, 305–309. https://doi.org/10.4028/www.scientific.net/AMM.754-755.305
- He, J., Wang, X., Su, Y., Li, Z., and Shi, X. (2019). Shear Strength of Stabilized Clay Treated with Soda Residue and Ground Granulated Blast Furnace Slag. *Journal of Material in Civil Engineering*, 31(3), 1–8. https://doi.org/10.1061/(ASCE)MT.1943-5533.0002629.
- He, L., Yu, D., Lv, W., Wu, J., and Xu, M. (2013). A novel method for CO₂ sequestration via indirect carbonation of coal fly ash. *Industrial and Engineering Chemistry Research*, 52(43), 15138–15145. https://doi.org/10.1021/ie4023644
- Ho, L. S., Nakarai, K., Ogawa, Y., Sasaki, T., and Morioka, M. (2017). Strength development of cement-treated soils: Effects of water content, carbonation, and pozzolanic reaction under drying curing condition. *Construction and Building*

- Materials, 134, 703–712. https://doi.org/10.1016/j.conbuildmat.2016.12.065
- Holloway, S. (2005). Underground sequestration of carbon dioxide a viable greenhouse gas mitigation option. *Energy*, *30*, 2318–2333. https://doi.org/10.1016/j.energy.2003.10.023
- Holt, D. G. A., Jefferson, I., Braithwaite, P. ., and Chapman, D. . (2010). Sustainable Geotechnical Design. In *GeoFlorida 2010: Advances in Analysis, Modeling & Design* (pp. 2925–2932).
- Hu, J., Liu, W., Wang, L., Liu, Q., Chen, F., Yue, H., ... Li, C. (2017). Indirect mineral carbonation of blast furnace slag with (NH₄)₂SO₄ as a recyclable extractant. *Journal of Energy Chemistry*, 26(5), 927–935. https://doi.org/10.1016/j.jechem.2017.06.009
- Huijgen, W. J. J. (2007). *Carbon dioxide sequestration by mineral carbonation*. Energy research Centre of the Netherlands, The Netherlands. ISBN: 90-8504-573-8.
- Islam, A., Alengaram, U. J., Jumaat, M. Z., and Bashar, I. I. (2014). The development of compressive strength of ground granulated blast furnace slag-palm oil fuel ashfly ash based geopolymer mortar. *Materials and Design*, *56*, 833–841. https://doi.org/10.1016/j.matdes.2013.11.080
- Jafer, H. M., W.Atherton, F.Ruddock, and E.Loffil. (2015). Assessing the Potential of a Waste Material for Cement Replacement and the Effect of Its Fineness in Soft Soil Stabilisation. *International Journal of Geological and Environmental Engineering*, 9(8), 915–921.
- Jafer, H. M., W.Atherton, and Ruddock, F. M. (2015). Soft Soil Stabilisation Using High Calcium Waste Material Fly Ash. In *12th International Post-Graduate Research Conference* (pp. 847–857).
- Jamsawang, P., Poorahong, H., Yoobanpot, N., Songpiriyakij, S., and Jongpradist, P. (2017). Improvement of soft clay with cement and bagasse ash waste.

 *Construction** and *Building** Materials*, 154, 61–71.

 https://doi.org/10.1016/j.conbuildmat.2017.07.188

- Jha, A. K., and Sivapullaiah, P. V. (2018). Potential of fly ash to suppress the susceptible behavior of lime-treated gypseous soil. *Soils and Foundations*, 58(3), 654–665. https://doi.org/10.1016/j.sandf.2018.02.024
- Jiang, N., Du, Y., and Liu, K. (2018). Durability of lightweight alkali-activated ground granulated blast furnace slag (GGBS) stabilized clayey soils subjected to sulfate attack. *Applied Clay Science*, *161*(April), 70–75. https://doi.org/10.1016/j.clay.2018.04.014
- Jimoh, O. A., Ariffin, K. S., Hussin, H. Bin, and Temitope, A. E. (2018). Synthesis of precipitated calcium carbonate: a review. *Carbonates and Evaporites*, *33*(2), 331–346. https://doi.org/10.1007/s13146-017-0341-x
- Jorat, M.Ehsan, A.Aziz, M., Marto, A., Zaini, N., Jusoh, S. N., and Manning, D. A. C. (2018). Sequestration Atmosperic CO₂ Inorganically: A Solution for Malaysia's CO₂ Emission. *Geosciences Journal*, 8, 483. https://doi.org/10.3390/geosciences8120483
- Jorat, Mohammad Ehsan, Marto, A., Namazi, E., and Amin, M. F. M. (2011). Engineering characteristics of kaolin mixed with various percentages of bottom ash. *Electronic Journal of Geotechnical Engineering*, *16 H*(January), 841–850.
- K.Phillips, E., M.Shillaber, C., K.Mitchel, J., P.E.Dist, E.Dove, J., and Filz, G. M. (2016). Sustainability Comparison of a Geosynthetic-Reinforced Soil Abutment and a Traditionally-Founded Abutment: A Case History. In *Geotechnical and Structural Engineering Congress* (pp. 699–711).
- Kamaruddin, M. A., Yusoff, M. S., Rui, L. M., Isa, A. M., Zawawi, M. H., and Alrozi, R. (2017). An overview of municipal solid waste management and landfill leachate treatment: Malaysia and Asian perspectives. *Environmental Science and Pollution Research*, 24(35), 26988–27020. https://doi.org/10.1007/s11356-017-0303-9
- Karim, N. A. A., Rashid, A. S. A., Noor, N. M., and Yaacob, H. (2014). Effect of Cement Stabilized Kaolin Subgrade on Strength Properties. *Journal of Applied Sciences*, 842–845.

- Katz, L. E., Rauch, A. F., Liljestrand, H. M., Harmon, J. S., Shaw, K. S., and Albers,
 H. (2001). Mechanisms of soil stabilization with liquid ionic stabilizer.
 Transportation Research Record, (1757), 50–57. https://doi.org/10.3141/1757-06
- Kim, M. J., Pak, S. Y., Kim, D., and Jung, S. (2017). Optimum conditions for extracting Ca from CKD to store CO₂ through indirect mineral carbonation. KSCE Journal of Civil Engineering, 21(3), 629–635. https://doi.org/10.1007/s12205-016-0913-7
- Kwon, S., Fan, M., DaCosta, H. F. M., and Russell, A. G. (2011). Factors affecting the direct mineralization of CO2 with olivine. *Journal of Environmental Sciences*, 23(8), 1233–1239. https://doi.org/10.1016/S1001-0742(10)60555-4
- Lackner, K. S., Wendt, C. H., Butt, D. P., Joyce, E. L., and Sharp, D. H. (1995). Carbon dioxide disposal in carbonate minerals. *Energy*, 20(11), 1153–1170. https://doi.org/10.1016/0360-5442(95)00071-N
- Lanas, J., and Alvarez, J. I. (2004). Dolomitic lime: thermal decomposition of nesquehonite. *Thermochimica Acta*, 421, 123–132. https://doi.org/10.1016/j.tca.2004.04.007
- Latifi, N. (2014). Geotechnical and Micro-structural Behaviour of Chemically Stabilized Tropical Residual Soil. PhD Thesis, Universiti Teknologi Malaysia, Skudai.
- Latifi, N., Horpibulsuk, S., Meehan, C. L., Majlid, M. Z. A., Tahir, M. M., and Mohamad, E. T. (2017). Improvement of Problematic Soils with Biopolymer An Environmentally Friendly Soil Stabilizer. *Journal of Materials in Civil Engineering*Engineering, 29(2), 1–11. https://doi.org/10.1061/(ASCE)MT.1943-5533.0001706.
- Latifi, N., and Meehan, C. L. (2017). Strengthening of Montmorillonitic and Kaolinitic Clays with Calcium Carbide Residue: A Sustainable Additive for Soil Stabilization. Geotechnical Frontiers 2017, 154–163. https://doi.org/10.1061/9780784480441.017

- Latifi, N., Vahedifard, F., Ghazanfari, E., and A. Rashid, A. S. (2018). Sustainable Usage of Calcium Carbide Residue for Stabilization of Clays. *Journal of Materials in Civil Engineering*, 30(6), 04018099. https://doi.org/10.1061/(ASCE)MT.1943-5533.0002313
- Lee, S. M., Lee, S. H., Jeong, S. K., Youn, M. H., Nguyen, D. D., Chang, S. W., and Kim, S. S. (2017). Calcium extraction from steelmaking slag and production of precipitated calcium carbonate from calcium oxide for carbon dioxide fixation. *Journal of Industrial and Engineering Chemistry*, 53, 233–240. https://doi.org/10.1016/j.jiec.2017.04.030
- Leung, D. Y. C., Caramanna, G., and Maroto-valer, M. M. (2014). An overview of current status of carbon dioxide capture and storage technologies. *Renewable and Sustainable Energy Reviews*, *39*, 426–443. https://doi.org/10.1016/j.rser.2014.07.093
- Liu, S., Cai, G., Cao, J., and Wang, F. (2017). Influence of soil type on strength and microstructure of carbonated reactive magnesia- treated soil. *European Journal of Environmental and Civil Engineering*, 8189(September), 1–19. https://doi.org/10.1080/19648189.2017.1378925.
- M.Andrew, R. (2018). Global CO₂ emissions from cement production. *Earth System Science Data*, 10, 195–217.
- Mandal, S., and Singh, J. P. (2016). Stabilization of Soil using Ground Granulated Blast Furnace Slag and Fly Ash. *International Journal of Innovative Research in Science, Engineering and Technology*, 5(12), 21121–21126. https://doi.org/10.15680/IJIRSET.2016.0512038
- Maneli, A., Kupolati, W. K., Abiola, O. S., and Ndambuki, J. M. (2015). Influence of fly ash, ground-granulated blast furnace slag and lime on unconfined compressive strength of black cotton soil. *Road Materials and Pavement Design*, (September). https://doi.org/10.1080/14680629.2015.1066703
- Manjunath, K. V, Shekhar, H., Kumar, M., Kumar, P., and Kumar, R. (2012). Stabilization of Black Cotton Soil Using Ground Granulated Blast Furnace Slag. In *Proceedings of International Conference on Advances in Architecture and*

- Civil Engineering (Vol. 1, pp. 387–390).
- Marto, A., Boss, S., Makhtar, A. M., and Latifi, N. (2015). Strength and Ductility of High Strength Concrete Shear Walls. *Jurnal Teknologi*, 2(December), 89–94. https://doi.org/DOI: 10.11113/jt.v76.5438
- Mattila, H. (2014). *Utilization of steelmaking waste materials for production of calcium carbonate (CaCO3)*. Doctor of Technology Thesis, Åbo Akademi University, Turku, Finland.
- Mo, L., and Panesar, D. K. (2012). Effects of accelerated carbonation on the microstructure of Portland cement pastes containing reactive MgO. *Cement and Concrete Research*, 42(6), 769–777. https://doi.org/10.1016/j.cemconres.2012.02.017
- Mohammadia, M., H.M.Mallikarjuna, and Hussain, A. (2017). Stabilization of Clay Subgrade Soils for Pavements Using Ground Granulated Blast Furnace Slag. *International Journal of Engineering Development and Research*, 5(4), 764–773.
- Mohd Yunus, N. Z., Ayub, A., Wahid, M. A., Satar, M. K. I. M., Abudllah, R. A., Yaacob, H., ... Hezmi, M. A. (2019). Strength behaviour of kaolin treated by demolished concrete materials. In *IOP Conf. Series: Earth and Environmental Science* 220. https://doi.org/10.1088/1755-1315/220/1/012001
- Mousavi, S., and Wong, L. S. (2015). Mechanical Behavior of Compacted and Stabilized Clay with Kaolin and Cement. *Jordan Journal of Civil Engineering*, 9(4), 477–486.
- Muhmed, A., and Wanatowski, D. (2013). Effect of Lime Stabilisation on the Strength and Microstructure of Clay. *IOSR Journal of Mechanical and Civil Engineering*, 6(3), 2320–2334. https://doi.org/10.6088/ijes.2013030600005
- Mujtaba, H., Aziz, T., Farooq, K., Sivakugan, N., and Das, B. M. (2018). Improvement in Engineering Properties of Expansive Soils using Ground Granulated Blast Furnace Slag. *Journal of the Geological Society of India*, 92(3), 357–362. https://doi.org/10.1007/s12594-018-1019-2

- Mun, M., and Cho, H. (2013). Mineral carbonation for carbon sequestration with industrial waste. *Energy Procedia*, *37*, 6999–7005. https://doi.org/10.1016/j.egypro.2013.06.633
- Mustapa, S. I., and Bekhet, H. A. (2016). Analysis of CO₂ emissions reduction in the Malaysian transportation sector: An optimisation approach. *Energy Policy*, 89, 171–183. https://doi.org/10.1016/j.enpol.2015.11.016
- Nakarai, K., and Yoshida, T. (2015). Effect of carbonation on strength development of cement-treated Toyoura silica sand. *Soils and Foundations*, *55*(4), 857–865. https://doi.org/10.1016/j.sandf.2015.06.016
- Narendra, K., and Prasad, A. D. (2017). Effect of Ground Granulated Blast Furnace Slag on Expansive Soils Under Static Loading. *VFSTR Journal of Science, Technology, Engineering & Management*, Vol. 03, No. 01, 2455-2062.
- Oelkers, E. H., Gislason, S. R., and Matter, J. (2008). Mineral Carbonation of CO₂. *Elements*, 4, 333–338. https://doi.org/10.2113/gselements.4.5.333
- Olajire, A. A. (2013). A review of mineral carbonation technology in sequestration of CO₂. *Journal of Petroleum Science and Engineering*, 109, 364–392. https://doi.org/10.1016/j.petrol.2013.03.013
- Ozdemir, M. A. (2016). Improvement in Bearing Capacity of a Soft Soil by Addition of Fly Ash. In *Procedia Engineering (Advances in Transportation Geotechnics 3 . The 3rd International Conference on Transportation Geotechnics)* (Vol. 143, pp. 498–505). Elsevier B.V. https://doi.org/10.1016/j.proeng.2016.06.063
- Pakir, F. B. (2017). *Physicochemical, Microstructural and Engineering Behaviour of Non-Traditional Stabilizer Treated Marine Clay*. PhD Thesis, Universiti Teknologi Malaysia, Skudai.
- Papargyropoulou, E., Preece, C., Padifield, R., and Abdullah, A. A. B. (2011). Sustainable Construction Waste Management in Malaysia: A Contractor's Perspective. *Management and Innovation for a Sustainable Built Environment*.
- Patel, M. A., and Patel, H. S. (2012). A Review on Effects of Stabilizing Agents for

- Stabilization of Weak Soil. *Civil and Environmental Research*, 2(6), 1–7.
- Pathak, A. K., Pandey, V., Murari, K., Singh, J. P., and Prof, A. (2014). Soil Stabilisation Using Ground Granulated Blast Furnace Slag. *Journal of Engineering Research and Applications Www.Ijera.Com*, 4(5), 164–171. https://doi.org/10.1161/CIRCULATIONAHA.114.009107
- Phummiphan, I., Horpibulsuk, S., Rachan, R., Arulrajah, A., Shen, S. L., and Chindaprasirt, P. (2018). High calcium fly ash geopolymer stabilized lateritic soil and granulated blast furnace slag blends as a pavement base material. *Journal of Hazardous Materials*, 341, 257–267. https://doi.org/10.1016/j.jhazmat.2017.07.067
- Prigiobbe, V., Hänchen, M., Werner, M., Baciocchi, R., and Mazzotti, M. (2009). Mineral carbonation process for CO₂ sequestration. *Energy Procedia*, 1, 4885–4890. https://doi.org/10.1016/j.egypro.2009.02.318
- Priyanga.G, Divya Krishnan K, and P.T.Ravichandran. (2018). Characteristics of Rubberized Soil with Ground Granulated Blast furnace Slag as Binder Material. In *Materials Today: Proceedings* (pp. 8655–8661). https://doi.org/10.1016/j.matpr.2017.12.290
- R.Walker, and S.Pavia. (2011). Physical properties and reactivity of pozzolans, and their influence on the properties of lime pozzolan pastes. *Materials and Structures*, 44, 1139–1150. https://doi.org/10.1617/s11527-010-9689-2
- Raftari, M., A.Rashid, A. S., Kassim, K. A., and Moayedi, H. (2014). Evaluation of kaolin slurry properties treated with cement. *Measurement: Journal of the International Measurement Confederation*, 50(1), 222–228. https://doi.org/10.1016/j.measurement.2013.12.042
- Rahman, F. A., Aziz, M. M. A., Saidur, R., AbuBakar, W. A. W., Hainin, M. R., Putrajaya, R., and Hassan, N. A. (2017). Pollution to solution: Capture and sequestration of carbon dioxide (CO₂) and its utilization as a renewable energy source for a sustainable future. *Renewable and Sustainable Energy Reviews*, 71(January), 112–126. https://doi.org/10.1016/j.rser.2017.01.011

- Rahman, F. A., Aziz, M. M. A., Saidur, R., Bakar, W. A. W. A., Hainin, M. R., Putrajaya, R., and Hassan, N. A. (2017). Pollution to solution: Capture and sequestration of carbon dioxide (CO₂) and its utilization as a renewable energy source for a sustainable future. *Renewable and Sustainable Energy Reviews*, 71(January), 112–126. https://doi.org/10.1016/j.rser.2017.01.011
- Ranjan, G., and Rao, A. S. R. (2000). Basic and Applied Soil Mechanics.
- Rasool, R., and Kapoor, K. (2017). Comparative Study On Stabilization of Soil With Ground Granulated Blast Furnace Slag (GGBS). *International Journal of Latest Research in Science and Technology*, 6(3).
- Regnault, O., Lagneau, V., and Schneider, H. (2009). Experimental measurement of portlandite carbonation kinetics with supercritical CO₂. *Chemical Geology*, 265(1–2), 113–121. https://doi.org/10.1016/j.chemgeo.2009.03.019
- Saeed, K. A. H., Kassim, K. A., Nur, H., and Yunus, N. Z. M. (2013). Characterization of Hydrated Lime-Stabilized Brown Kaolin Clay. *International Journal of Engineering Research & Technology*, 2(11), 3722–3727.
- Saeed, K. A., Kassim, K. A., and Nur, H. (2014). Physicochemical characterization of cement treated kaolin clay. *Gradjevinar*, 66(6), 513–521. https://doi.org/10.14256/JCE.976.2013
- Saeeda, K. A. H., Kassima, K. A., Yunusa, N. Z. M., and Nur, H. (2015). Physico-Chemical Characterization Of Lime Stabilized Tropical Kaolin. *Jurnal Teknologi*, 72(3), 83–90.
- Safaai, M., Sharliza, N., Noor, Z., Hashim, H., and Ujang, Z. (2011). Projection of CO₂ Emissions in Malaysia. *Environmental Progress & Sustainable Energy*, 30(December), 658–665. https://doi.org/10.1002/ep
- Salih, A. G. (2012). Review on granitic residual soils' geotechnical properties. Electronic Journal of Geotechnical Engineering, 17, 2645–2658.
- Samtani, M., Dollimore, D., and Alexander, K. S. (2002). Comparison of dolomite decomposition kinetics with related carbonates and the effect of procedural

- variables on its kinetic parameters. *Thermochimica Acta*, 393, 135–145.
- Saravanan, R., Udhayakumar, T., Dinesh, S., Venkatasubramanian, C., and Muthu, D. (2017). Effect of addition of GGBS and lime in soil stabilisation for stabilising local village roads in Thanjavur region. In *IOP Conf. Series: Earth and Environmental Science*.
- Sauvé, J. L., Goddard, T. W., and Cannon, K. R. (2000). A Preliminary Assessment of Carbon Dioxide Emissions from Agricultural Soils. In *Proceedings of 37th Annual Alberta Soil Science Workshop* (pp. 1–6).
- Saygili, A., and Dayan, M. (2019). Freeze-thaw behavior of lime stabilized clay reinforced with silica fume and synthetic fibers. *Cold Regions Science and Technology*, *161*, 107–114. https://doi.org/10.1016/j.coldregions.2019.03.010
- Sharma, A. K., and Sivapullaiah, P. V. (2016). Ground granulated blast furnace slag amended fly ash as an expansive soil stabilizer. *Soils and Foundations*, 56(2), 205–212. https://doi.org/10.1016/j.sandf.2016.02.004
- Sharma, N. K., Swain, S. K., and Sahoo, U. C. (2012). Stabilization of a Clayey Soil with Fly Ash and Lime: A Micro Level Investigation, *30*, 1197–1205. https://doi.org/10.1007/s10706-012-9532-3
- Sharma, R. K., and Verma, G. (2019). Effect of Ground Granulated Blast Furnace Slag and Metakaolin on Geotechnical Properties of Clayey Soil. In *Proceedings of the 1st International Conference on Sustainable Waste Management through Design* (pp. 386–392). Springer International Publishing. https://doi.org/10.1007/978-3-030-02707-0
- Shillaber, C. M., Mitchell, J. K., and Dove, J. E. (2015a). Energy and Carbon Assessment of Ground Improvement Works . I: Definitions and Background. *Journal of Geotechnical and Geoenvironmental Engineering*, 1–9. https://doi.org/10.1061/(ASCE)GT.1943-5606.0001410.
- Shillaber, C. M., Mitchell, J. K., and Dove, J. E. (2015b). Energy and Carbon Assessment of Ground Improvement Works . II: Working Model and Example. *Journal of Geotechnical and Geoenvironmental Engineering*, 1–11.

- https://doi.org/10.1061/(ASCE)GT.1943-5606.0001411.
- Shillaber, C. M., Mitchell, J. K., Dove, J. E., and Ostrum, Z. A. (2016). Framework to Account for Uncertainty in Energy and Carbon Assessment of Ground Improvement Works. *Journal of Geotechnical and Geoenvironmental Engineering*, 1–12. https://doi.org/10.1061/(ASCE)GT.1943-5606.0001649.
- Shillaber, C. M., Pearce, A. R., Mitchell, J. K., and Dove, J. E. (2016). Uncertainty in the Estimates of Embodied Energy and CO₂ Emissions for Ground Improvement: The Influence of Material Haul Distance. In *Geo-Chicago 2016 GSP 269* (pp. 722–731).
- Silva, P. De, Bucea, L., and Sirivivatnanon, V. (2009). Chemical, microstructural and strength development of calcium and magnesium carbonate binders. *Cement and Concrete Research*, 39(5), 460–465. https://doi.org/10.1016/j.cemconres.2009.02.003
- Singh, R., Mehrotra, A., and Khan, M. A. (2018). A Review Paper on Comparative Study of Soil Stabilization with Widely used Admixtures Like Lime, Cement, Flyash and Bitumen Emulsion. *International Journal of Engineering Trends and Technology*, 58(2), 96–99.
- Singh, U. (2013). Carbon capture and storage: an effective way to mitigate global warming. *Current Science*, 105, 914–922.
- Song-yu, L., Guang-hua, C., Yan-jun, D., Heng, Z., and Ping, W. (2017). Engineering properties of carbonated reactive magnesia-stabilized silt under different activity index. *Procedia Engineering*, 189(May), 158–165. https://doi.org/10.1016/j.proeng.2017.05.026
- Song, K., Song, J., Lee, B. Y., and Yang, K. (2014). Carbonation Characteristics of Alkali-Activated Blast-Furnace Slag Mortar. *Advances in Materials Science and Engineering*, 2014, 1–11.
- Strózyk, J., and Tankiewicz, M. (2016). The elastic undrained modulus Eu50 for stiff consolidated clays related to the concept of stress history and normalized soil properties. *Studia Geotechnica et Mechanica*, 38(3), 67–72.

- https://doi.org/10.1515/sgem-2016-0025
- Suresh, D., and Nagaraju, K. (2015). "Ground Granulated Blast Slag (GGBS) In Concrete A Review." *IOSR Journal of Mechanical and Civil Engineering*, 12(4), 76–82. https://doi.org/10.9790/1684-12467682
- Tamilselvi Dananjayan, R. R., Kandasamy, P., and Andimuthu, R. (2016). Direct mineral carbonation of coal fly ash for CO₂ sequestration. *Journal of Cleaner Production*, 112, 4173–4182. https://doi.org/10.1016/j.jclepro.2015.05.145
- Tasong, W. A., Wild, S., and Tilley, R. J. D. (1999). Mechanisms by which ground granulated blastfurnace slag prevents sulphate attack of lime-stabilised kaolinite. *Cement and Concrete Research*, 29(0008), 975–982.
- ToolBox, E. (2003). Air Composition and Molecular Weight. Retrieved from https://www.engineeringtoolbox.com/air-composition-d_212.html
- Uliasz-Bochenczyk, A., Mokrzycki, E., Mazurkiewicz, M., and Piotrowski, Z. (2006).

 Utilization of carbon dioxide in fly ash and water mixtures. *Chemical Engineering Research and Design*, 84, 843–846. https://doi.org/10.1205/cherd.05146
- Unluer, C., and Al-tabbaa, A. (2013). Impact of hydrated magnesium carbonate additives on the carbonation of reactive MgO cements. *Cement and Concrete Research*, *54*, 87–97. https://doi.org/10.1016/j.cemconres.2013.08.009
- Unluer, C., and Al-Tabbaa, A. (2014). Characterization of light and heavy hydrated magnesium carbonates using thermal analysis. *Journal of Thermal Analysis and Calorimetry*, 115, 595–607. https://doi.org/10.1007/s10973-013-3300-3
- Unluer, C., and Al-Tabbaa, A. (2015a). The role of brucite, ground granulated blastfurnace slag, and magnesium silicates in the carbonation and performance of MgO cements. *Construction and Building Materials*, *94*, 629–643. https://doi.org/10.1016/j.conbuildmat.2015.07.105
- Unluer, C., and Al-Tabbaa, A. (2015b). The role of brucite, ground granulated blastfurnace slag, and magnesium silicates in the carbonation and performance

- of MgO cements. *Construction and Building Materials*, *94*, 629–643. https://doi.org/10.1016/j.conbuildmat.2015.07.105
- Wang, Y., Zhao, L., Otto, A., Robinius, M., and Stolten, D. (2017). A Review of Post-combustion CO₂ Capture Technologies from Coal-fired Power Plants. *Energy Procedia*, 114(November 2016), 650–665. https://doi.org/10.1016/j.egypro.2017.03.1209
- Wu, J. C.-S., Sheen, J.-D., Chen, S.-Y., and Fan, Y.-C. (2001). Feasibility of CO₂ Fixation via Artificial Rock Weathering. *Industrial & Engineering Chemistry Research*, 40(18), 3902–3905. https://doi.org/10.1021/ie0102221
- Y.L, Y., Liska, M., Unluer, C., and A, A.-T. (2013). Initial investigation into the carbonation of MgO for soil stabilisation. In *Proceedings of the 18th International Conference on Soil Mechanics and Geotechnical Engineering* (Vol. 5, pp. 2641–2644).
- Yadu, L., and Tripathi, R. K. (2013). Stabilization of Soft Soil With Granulated Blast Furnace Slag and Fly Ash. *International Journal of Research in Engineering and Technology*, 2(1997), 115–119.
- Yan, H., Zhang, J., Zhao, Y., Liu, R., and Zheng, C. (2015). CO₂ Sequestration by direct aqueous mineral carbonation under low-medium pressure conditions. *Journal of Chemical Engineering of Japan*, 48(11), 937–946. https://doi.org/10.1252/jcej.14we381
- Yan, H., Zhang, J., Zhao, Y., and Zheng, C. (2013). CO₂ Sequestration from flue gas by direct aqueous mineral carbonation of wollastonite. *Science China Technological Sciences*, 56(9), 2219–2227. https://doi.org/10.1007/s11431-013-5318-y
- Yi, Y., Gu, L., and Liu, S. (2015). Microstructural and mechanical properties of marine soft clay stabilized by lime-activated ground granulated blastfurnace slag. *Applied Clay Science*, 103, 71–76. https://doi.org/10.1016/j.clay.2014.11.005
- Yi, Y., Gu, L., Liu, S., and Jin, F. (2016). Magnesia reactivity on activating efficacy for ground granulated blast furnace slag for soft clay stabilisation. *Applied Clay*

- Science, 126, 57–62. https://doi.org/10.1016/j.clay.2016.02.033
- Yi, Y., Liska, M., Akinyugha, A., Unluer, C., and Al-Tabbaa, A. (2013). Preliminary Laboratory-Scale Model Auger Installation and Testing of Carbonated Soil-MgO Columns. *Geotechnical Testing Journal*, 36(March). https://doi.org/10.1520/GTJ20120052
- Yi, Y., Liska, M., and Al-tabbaa, A. (2014). Properties of Two Model Soils Stabilized with Different Blends and Contents of GGBS, MgO, Lime, and PC. *Journal of Material in Civil Engineering*, 26(FEBRUARY), 267–274. https://doi.org/10.1061/(ASCE)MT.1943-5533.0000806.
- Yi, Y., Liska, M., Unluer, C., and Al-tabbaa, A. (2013). Carbonating magnesia for soil stabilization. *Canadian Geotechnical Journal*, *50*(June), 899–905.
- Yi, Y., Lu, K., Liu, S., and Al-tabbaa, A. (2016). Property changes of reactive magnesia stabilized soil subjected to forced carbonation. *Canadian Geotechnical Journal*, *53*, 314–325.
- Yin, C., Zhang, W., Jiang, X., and Huang, Z. (2018). Effects of initial water content on microstructure and mechanical properties of lean clay soil stabilized by compound calcium-based stabilizer. *Materials*, 11(10). https://doi.org/10.3390/ma11101933
- Zainuddin, N. E. B., Yunus, N. Z. M., Marto, A., Al-bared, M. A. M., Mashros, N., and Abdullah, R. A. (2016). A review: Reutilization of waste material to stabilize marine clay. *Malaysian Journal of Civil Engineering*.
- Zainuddin, N. E. B., Yunus, N. Z. M., Marto, A., Al-Bared, M. A. M., Mashros, N., and Abdullah, R. A. (2016). A review: Reutilization of waste material to stabilize marine clay. *Malaysian Journal of Civil Engineering*, (December).
- Zevenhoven, R., Jens Kohlmann, and Mukherjee, A. B. (2002). Direct Dry Mineral Carbonation for CO₂ Emissions Reduction in Finland. In *Proc. of the 27th International Technical Conference on Coal Utilization & Fuel Systems*. Retrieved from http://users.abo.fi/rzevenho/Clw27_31.PDF

- Zhang, J., Zhang, R., Geerlings, H., and Bi, J. (2010). A novel indirect wollastonite carbonation route for CO₂ sequestration. *Chemical Engineering and Technology*, 33(7), 1177–1183. https://doi.org/10.1002/ceat.201000024
- Zhou, Y., Pan, L., Tang, Q., Zhang, Y., Yang, N., and Lu, C. (2018). Evaluation of Carbonation Effects on Cement-Solidified Contaminated Soil Used in Road Subgrade. *Advances in Materials Science and Engineering*, 2018, 1–15. https://doi.org/10.1155/2018/5271324

LIST OF PUBLICATIONS

Journal with Impact Factor

- Ahmed Mohammed Awad Mohammed, Nor Zurairahetty Mohd Yunus, Muhammad Azril, Ahmad Safuan A Hezmi (2021). Sequestration of Carbon Dioxide Using Ground Granulated Blast Furnaces Slag and kaolin mixtures. "Global NEST Journal". DOI: https://doi.org/10.30955/gnj.003487. (Q4, IF: 0.983).
- Ahmed Mohammed Awad Mohammed, Nor Zurairahetty Mohd Yunus, Muhammad Azril, Dayang Zulaika Abang Hasbollah, Ahmad Safuan A Hezmi (2021). Ground improvement and its role in carbon dioxide reduction: a review. "Environmental Science and Pollution Research". DOI:10.1007/s11356-021-12768-2 (Q2, IF: 3.056).
- Ahmed Mohammed Awad Mohammed, Nor Zurairahetty Mohd Yunus, Muhammad Azril Hezmi, Ahmad Safuan A. Rashid & Suksun Horpibulsuk (2021). Carbonated ground granulated blast furnace slag stabilising brown kaolin. "Environmental Science and Pollution Research".
 DOI: https://doi.org/10.1007/s11356-021-14718-4 (Q2, IF: 3.056).