CORROSION RESISTANCE ENHANCEMENT OF HYDROXYAPATITE AND MAGNESIUM OXIDE MULTILAYER COATING ON MAGNESIUM ALLOY AZ31 VIA ELECTROPHORETIC DEPOSITION

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ABSTRACT

Magnesium (Mg) and its alloys have recently gained attention among researchers due to their excellent biodegradable and mechanical properties. However, poor corrosion resistance of Mg and its alloys have limited their clinical application. Rapid corrosion of Mg and its alloys may cause an implant failed before the bone has fully restored. The aim of this research was to enhance corrosion resistance of biodegradable implant substrate that is suitable for biomedical applications. Multilayer coating of nano-powders HA/MgO were coated on AZ31 magnesium alloy substrate via electrophoretic deposition (EPD) technique. A coating was obtained by applying suitable EPD process parameters (applied voltage and deposition time) and coating approaches (single layer coating and multilayer coating). The results of coating behaviour were characterised by means of XRD, and SEM to examine the coating surface morphologies and their phases. Coating performances of HA, MgO, and HA/MgO coating were studied by immersion test, potentiodynamic test, and electrochemical impedance spectroscopy where all done in-vitro. Elemental analysis was carried out using EDS to verify that the composed elements are biodegradable and harmless to the human body. The results obtained in this research suggested that corrosion resistance of a coated sample was affected by its particles distribution structure. Particles distribution structure with higher compactness showed a homogeneous coating layer and smaller surface defects. In general, the multilayer coating approach has outperformed the single coating approach by demonstrating a higher compactness particle distribution structure. Corrosion results of each group were compared, and the optimum process parameters were determined. The optimum process parameters for single layer coating HA, MgO and HA/MgO were 2min/10V, 30V/1min, and 15V/1min, respectively. On the other hand, the optimum number of layers for multilayer coating HA, MgO and HA/MgO were 5 layers, 3 layers, and 2 layers, respectively. It was also found that composite coating of HA/MgO has successfully inherited the benefits and limitations of each coating powder. Furthermore, defects such as agglomeration and cracks were found significantly reduced to a lower degree in multilayer coating approach. Among all of the coated samples, Laco-HA/MgO 2 layers coated with 5V/10 min each layer showed the highest corrosion resistance. The significant improvement in inhibition efficiency achieved 99.76% against the uncoated AZ31. Based on these results, it was concluded that this sample has a great potential for biodegradable orthopaedic application. Lastly, it was recommended to conduct cell viability measurement, biological reaction, and cytotoxicity test on Laco-2 layers by biological field researchers in the future.

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ABSTRAK

Aloi berasaskan magnesium (Mg) akhir-akhir ini mendapat perhatian penyelidik-penyelidik dengan sifat biodegradasi dan mekanikalnya yang sangat baik. Walau bagaimanapun, ketahanan kakisan yang lemah telah menghadkan penggunaan klinikalnya. Kakisan Mg yang cepat boleh menyebabkan kegagalan implan sebelum tulang pulih sepenuhnya. Tujuan penyelidikan ini adalah untuk meningkatkan rintangan kakisan aloi biodegradasi implan substrak untuk kegunaan bioperubatan. Komposit pelbagai lapisan serbuk nano HA / MgO telah disalut pada substrak aloi magnesium AZ31 dengan teknik pemendapan elektroforetik (EPD) untuk meningkatkan ketahanan kakisannya. Lapisan tersebut dihasilkan dengan menggunakan proses parameter yang sesuai (voltan dikenakan dan masa pemendapan) dan pendekatan salutan (lapisan tunggal dan lapisan pelbagai lapisan) menggunakan teknik EPD. Hasil tingkah laku salutan dicirikan dengan XRD, dan SEM untuk memeriksa morfologi permukaan lapisan dan analisis fasa. Kadar kakisan lapisan HA, MgO dan HA / MgO dikaji dengan ujian rendaman, ujian potentiodynamic, dan spektroskopi impedans elektrokimia secara in-vitro. Analisis unsur dijalankan mengunakan EDS untuk mengesahkan produk sampingan kakisan terbiodegradasi dan tidak berbahaya terhadap kepada manusia. Hasil kajian menunjukkan bahawa ketahanan kakisan sampel bersalut dipengaruhi oleh struktur taburan partikel. Struktur taburan partikel dengan kepadatan ketumpatan yang lebih tinggi menunjukkan lapisan yang sekata dan kecacatan permukaan yang lebih kecil. Secara umumnya, pendekatan salutan pelbagai lapisan menghasilkan struktur taburan partikel dengan kepadatan ketumpatan yang lebih tinggi berbanding dengan lapisan salutan tunggal. Hasil kadar kakisan setiap kumpulan dibandingkan, dan parameter proses optimum ditentukan. Parameter proses optimum salutan tunggal HA, MgO dan HA / MgO masing-masing adalah 2min / 10V, 30V / 1min, dan 15V / 1min. Sebaliknya, bilangan lapisan optimum dalam pendekatan lapisan pelbagai lapisan untuk HA, MgO dan HA / MgO adalah 5, 3, dan 2 lapisan. Selain itu, didapati juga salutan komposit HA / MgO telah berjaya mewarisi faedah dan batasan setiap salutan. Selanjutnya, kecacatan seperti penggumpalan dan retakan telah berkurang pada tahap pembentukan yang lebih rendah dengan pendekatan pelbagai lapisan. Di antara semua sampel yang disaluti, 2 lapisan Laco-HA/MgO yang tersalut pada 5V/10min setiap lapisan menunjukkan ketahanan kakisan tertinggi. Sampel ini telah menunjukkan peningkatan yang ketara dengan 99.76% perencatan kecekapan berbanding dengan AZ31 yang tidak disaluti. Berdasarkan hasil kajian ini, disimpulkan bahawa, Laco-komposit 2 lapisan berpotensi besar untuk aplikasi ortopedik terbiodegradasi. Pada masa hadapan, pengukuran daya maju sel, tindak balas biologi dan pengukuran ujian sitotoksisiti disyorkan untuk dilakukan oleh penyelidik bidang biologi.

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CHAPTER 1

INTRODUCTION

1.1 **Background of Research**

A massive number of orthopaedic implant surgeries were performed in worldwide each year [1]. In general, implant devices have been categorised as permanent implant or as a temporary fracture fixation [2]. Nowadays, the most common bioimplant metallic alloys used are titanium alloy, 316L stainless steel or cobalt-chromium alloys. However, the limitations of these materials are (i) it required second time of surgery to remove the implant as the bone has healed, (ii) relatively high stiffness caused bone shielding effect [3] and (iii) toxic level as if high concentration were used [4, 5].

Magnesium and its alloy (Mg) have gained significant attention in biodegradable implant as a temporary fixation to solve the stated problems. Mg possesses with superior mechanical properties, bioactivity and lightweight (density of 1.77 g/cm³ which closer to natural bone 1.8-2.1 g/cm³). Mg²⁺ ion is the fourth most abundant cation in human body to helps metabolism activities in human body [6, 7] and to stimulate new bone growth [8, 9]. In the meantime, the decomposition element of Mg^{2+} is harmless to human body [10]. AZ31 is one of the most potential based material that performed a biodegradable, non-toxic, excellent strength materials. Details of AZ31 is stated at section 2.3.

However, Mg degrade rapidly in psychological environment (pH 7.4-7.6) has limited the use of Mg in orthopaedic applications. High degradation rate of the implant causes mechanical failure by losing its mechanical strength before bone has fully restored [7]. For decades, researchers has been trying to study the variable methods and solutions to fulfil the requirement for biomedical applications [11].

Clinical concern is ambivalent on a biodegradable implant degradation rate [12]. On one hand, a slow degradation rate is preferred in the initial stage to maintain the mechanical strength and biological favour [13, 14]. On the other hand, fast degradation is preferred after the bone has been consolidated. In principle, an implant shall not remain inside a human body for unnecessary long period of time [12]. Therefore, surface coating has become an interesting option where it could effectively to achieve ambivalent implant requirements [12].

In this research, an AZ31 Mg alloy is coated by using electrophoretic deposition (EPD). Other coating methods such as thermal spray, pulsed laser deposition, dip coating has the limitations of controlling the thickness, and high temperature coating process. Therefore, EPD comes out as relatively outstanding nano powder ceramic coating technique, easy fabricate, cost effective, and an excellent choice for ceramic nano powders colloid deposition as compared to other coating methods [15]. Besides, EPD is a coating process that allow the coating at room temperature, and with flexible coating thickness which suitable for low melting point Mg substrate. Many studies of EPD are relates to its kinetic movement of charged particles motion [16-19]. However, the studies on coating approach such as the effects of multilayer coating, the particles distribution of multilayer approach, the corrosion resistance by applying various layers have not been reported.

Hence, the aim of this research was to study the effects on suitable coating parameters of single layer approach and effectiveness of multilayer coating approach on HA and MgO coating powder. Nano powders size is used to enhance the fine dispersion of HA and MgO in the suspension. HA is one of the most common coating powders applied on implant to increase it biological favour while MgO natural protectiveness of Mg based alloy. A synthetic compact MgO coated on AZ31 with the purpose of increase it corrosion resistance. Through studies, MgO not commonly applied by using wet coating method such as EPD due to its hygroscopic character. In this research, the aim is not only applying MgO but also applying composite HA/MgO.

Composite coating of HA/MgO with co-deposition and layer by layer with codeposition method (named as Laco) was introduced to performance on HA/MgO. HA/MgO composite coating is coated to enhancing implant corrosion resistance and maintaining the biological stability at the same time. Multilayer coating of HA/MgO is to increase the coating compactness of the composite. Composite of HA/MgO with the advantages of enhancing biological favour and increase the corrosion resistance of the implant.

In this research, the focus on corrosion of biodegradable implant. Adhesion and substrate roughness of biodegradable coating are unlike permanent implant coating where the features of coating adhesion or roughness of substrate implant are not critical. A biodegradable coating is meant to be dissolved during the process and bonded with the new bone growth.

1.2 **Problem Statement**

A permanent implant is an implant that used to serve patient for the life span of the implant such as non-loading implant at joint. A biodegradable implant material it is expected to exist in human body, as a load bearing implant until the bone fully recover.

However, rapid corrosion behaviour of Mg in physiological environment has limited its clinical application. It reacts with high concentration of chloride (Cl⁻) in aqueous environment, and form soluble $MgCl₂$ and hydrogen gas, $H₂$ as the byproducts. Hydrogen gas evolved is harmless to human body but with a rapid corrosion, and considerable amount of H2 might resulting undesired inflammation and creates an empty space weaken the adhesion between the substrate with adjacent bone.

Rapid corrosion also results the risk of implant failure. Implant may lose its mechanical integrity before the bone has fully healed. Patient might face unnecessary pain and cost due to implant failure and extend the healing time.

Current coating methods are limited by low melting point substrate of magnesium in the range of 500-650°C. In general, the ceramic coating powder

requires a sintering process around 1300-1500°C. Coating such as HA on titanium are eligible to perform a high sintering process but it is not applicable on magnesiumbased material. Room temperature coating methods without sintering process caused coating powder in a loosen and less compactness condition.

Currently, a permanent implant material is applied with a biological favour coating material to increase it biological favour. However, a single coating as such insufficient for a biodegradable material to act as a biological favour and increase the corrosion resistance at the same time. A biodegradable implant is unlike the permanent implant which possesses with high corrosion resistance. Hence, it may lead to implant failure during the healing process.

Besides, the stress shielding effect remain the issue of permanent implant materials. Stress shielding effect is the effect that caused the low density of new bone growth as the implant is relatively too stiff to human bone. The healed new bone growth with low density may cause the bone too weak as compared to a healthy healing bone and without the strength that is supposed to support human activities.

1.3 **Purpose of Research**

This research addressed the problems described above namely, to increase the corrosion resistance of the magnesium alloy to synchronize between degradation rate of a magnesium alloy biodegradable implant and restoration period of a fracture bone. Firstly, to obtain a successful coated suspension parameters nano powders hydroxyapatite (HA), magnesium oxide (MgO) and nano composite HA/MgO respectively. Secondly, to investigate the effects of various process parameters of EPD coating on HA, MgO and HA/MgO for single coating layers approach for the corrosion protection of on bare material magnesium alloy (AZ31). Thirdly, to determine the multilayer coating approach of HA, MgO and HA/MgO to demonstrate enhancement of corrosion resistance of AZ31. Forth, to investigate the composite deposition of HA/MgO coating has influenced to the coating layer by complementary the limitation

embodied by each coating behaviour to increase the corrosion resistance by using multilayer coating approach.

1.4 **Objectives of the Research**

The principal objective of the research to increase corrosion resistance of magnesium alloy AZ31 by using electrophoretic deposition (EPD) with nano hydroxyapatite (HA), nano magnesium (MgO) and nano HA/MgO composite coatings with single layer and multilayer approaches for biomedical applications. Specific objectives include:

- (a) To study the effects EPD coating process parameters of applied voltage and deposition time of nano powders HA and MgO in terms of its surface morphologies relates to particles mobility rate, agglomeration, and particles distribution structures.
- (b) To investigate the effects of EPD multilayer coating approach of HA nano powders HA and MgO coating behaviour (surface morphologies examination, deposition yield measurement, and coating thickness measurement) and corrosion resistivity (evaluated by immersion test, potentiodynamic test, and electrochemical impedance spectroscopy).
- (c) To investigate the effects of co-deposition HA/MgO and Laco-deposition HA/MgO composite coating behaviour (surface morphologies examination, deposition yield measurement, and coating thickness measurement) and corrosion resistivity (evaluated by immersion test, potentiodynamic test, and electrochemical impedance spectroscopy).

1.5 **Significance of the Research**

The significant study of the research is to enhance the corrosion resistance of Mg base alloys with easy fabricate, and cost saving method. The experimental works

have undertaken, particularly in investigate the potential used of Mg biodegradable implant in extending knowledge and greater understanding level of electrophoretic deposition (EPD) and its possibility to increase the corrosion resistance.

This research also intended to extend the possible solution for biodegradable implant to apply practically. An extending of multilayer coating approach to produce a high compact density and corrosion resistance of Mg implant to replace permanent implant in temporary bone fixation applications. Hence, cost saving, and less suffering would have been borne by patients to conduct second time surgery all around the world.

The quantitative analysing data that have accumulated in depth with effects process parameters would provide valuable information for future research. An extensive study on Laco coating approach of HA/MgO extend the possibility to coat on Mg and provide the solution of Mg rapid corrosion.

1.6 **Scope of the Research**

The powder using are limited to nano hydroxyapatite (HA) at the length of 40 nm and nano magnesium oxide (MgO) is limited at the radius of 20 nm. Substrate of the research of Mg alloy AZ31 with the size of $30 \times 10 \times 3$ mm and coated with 3.0 cm² effective surface area. Electrophoretic deposition (EPD) as the only coating method in this research by the process parameters limit with voltage and deposition time.

The responses on the effect of the coating protection on AZ31 were limited to the coating characterizations (microstructure, deposition yield, phase, and coating thickness), and corrosion behaviour (immersion test, potentiodynamic test, and electrochemical impedance spectrometry test). Machines that used to investigate coating behaviours are including optical microscope, X-ray diffraction, scanning electron microscope (SEM), energy dispersive X-ray spectroscopy (EDS), and potentiostat. In addition, all corrosion performance of coated samples were investigated in *in vitro* condition only.

REFERENCES

- 1. Al-Mayahi, M., P. Vaudaux, L. Deabate, A. Lomessy, D. Suvà, and I. Uçkay, *5 - Diagnosis and treatment of implant-associated infections*, in *Biomaterials and Medical Device - Associated Infections*, L. Barnes and I.R. Cooper, Editors. 2015, Woodhead Publishing: Oxford. p. 83-99.
- 2. Jin, W. and P.K. Chu, *Orthopedic Implants*, in *Encyclopedia of Biomedical Engineering*, R. Narayan, Editor. 2019, Elsevier: Oxford. p. 425-439.
- 3. Weinans, H., D. R. Sumner, R. Igloria, and R.N. Natarajan, *Sensitivity of periprosthetic stress-shielding to load and the bone density–modulus relationship in subject-specific finite element models.* Journal of Biomechanics, 2000. **33**(7): p. 809-817.
- 4. Puleo, D.A. and W.W. Huh, *Acute toxicity of metal ions in cultures of osteogenic cells derived from bone marrow stromal cells.* Journal of Applied Biomaterials, 1995. **6**(2): p. 109-116.
- 5. Starosvetsky, D. and I. Gotman, *Corrosion behavior of titanium nitride coated Ni–Ti shape memory surgical alloy.* Biomaterials, 2001. **22**(13): p. 1853-1859.
- 6. Saris, N.-E.L., E. Mervaala, H. Karppanen, J.A. Khawaja, and A. Lewenstam, *Magnesium: An update on physiological, clinical and analytical aspects.* Clinica Chimica Acta, 2000. **294**(1–2): p. 1-26.
- 7. Staiger, M.P., A.M. Pietak, J. Huadmai, and G. Dias, *Magnesium and its alloys as orthopedic biomaterials: A review.* Biomaterials, 2006. **27**(9): p. 1728-1734.
- 8. Zreiqat, H., C.R. Howlett, A. Zannettino, P. Evans, G. Schulze-Tanzil, C. Knabe, and M. Shakibaei, *Mechanisms of magnesium-stimulated adhesion of osteoblastic cells to commonly used orthopaedic implants.* Journal of Biomedical Materials Research, 2002. **62**(2): p. 175-184.
- 9. Yamasaki, Y., Y. Yoshida, M. Okazaki, A. Shimazu, T. Uchida, T. Kubo, Y. Akagawa, Y. Hamada, J. Takahashi, and N. Matsuura, *Synthesis of functionally graded MgCO3 apatite accelerating osteoblast adhesion.* Journal of Biomedical Materials Research, 2002. **62**(1): p. 99-105.
- 10. Ye, X., M. Chen, M. Yang, J. Wei, and D. Liu, *In vitro corrosion resistance and cytocompatibility of nano-hydroxyapatite reinforced Mg–Zn–Zr composites.* Journal of Materials Science: Materials in Medicine, 2010. **21**(4): p. 1321-1328.
- 11. Jurgen Breme, C.J.K., and Roger Thull, *Metallic Biomaterial Interfaces*. 2008: Wiley-VCH.
- 12. Fischerauer, S.F., T. Kraus, X. Wu, S. Tangl, E. Sorantin, A.C. Hänzi, J.F. Löffler, P.J. Uggowitzer, and A.M. Weinberg, *In vivo degradation performance of micro-arc-oxidized magnesium implants: A micro-CT study in rats.* Acta Biomaterialia, 2013. **9**(2): p. 5411-5420.
- 13. Krause, A., N. von der Höh, D. Bormann, C. Krause, F.-W. Bach, H. Windhagen, and A. Meyer-Lindenberg, *Degradation behaviour and mechanical properties of magnesium implants in rabbit tibiae.* Journal of Materials Science, 2009. **45**(3): p. 624.
- 14. Kraus, T., S.F. Fischerauer, A.C. Hänzi, P.J. Uggowitzer, J.F. Löffler, and A.M. Weinberg, *Magnesium alloys for temporary implants in osteosynthesis: in vivo studies of their degradation and interaction with bone.* Acta biomaterialia, 2012. **8**(3): p. 1230-1238.
- 15. Besra, L. and M. Liu, *A review on fundamentals and applications of electrophoretic deposition (EPD).* Progress in Materials Science, 2007. **52**(1): p. 1-61.
- 16. Ferrari, B. and R. Moreno, *EPD kinetics: A review.* Journal of the European Ceramic Society, 2010. **30**(5): p. 1069-1078.
- 17. Moreno, R. and B. Ferrari, *Nanoparticles dispersion and the effect of related parameters in the EPD kinetics*, in *Electrophoretic Deposition of Nanomaterials*. 2012, Springer. p. 73-128.
- 18. Reynolds, G.J., Z.S. Barrett, H.N. McMurray, and G. Williams, *An investigation of the influence of physical vapour deposited aluminium layers on the kinetics of organic coating disbondment on iron.* Corrosion Science, (0).
- 19. Farnoush, H., J.A. Mohandesi, and D.H. Fatmehsari, *Effect of particle size on the electrophoretic deposition of hydroxyapatite coatings: a kinetic study based on a statistical analysis.* International Journal of Applied Ceramic Technology, 2013. **10**(1): p. 87-96.
- 20. Joon Park, R.S.L., *Biomaterials -An introduction* 3rd ed. 2007, Veriag New York: Springer.
- 21. Taljanovic, M.S., M.D. Jones, J.T. Ruth, J.B. Benjamin, J.E. Sheppard, and T.B. Hunter, *Fracture Fixation.* RadioGraphics, 2003. **23**(6): p. 1569-1590.
- 22. Williams, D.F., *The Williams Dictionary of Biomaterials*. 1999: Liverpool University Press.
- 23. StJohn, D.H., M.A. Easton, M. Qian, and J.A. Taylor, *Grain Refinement of Magnesium Alloys: A Review of Recent Research, Theoretical Developments, and Their Application.* Metallurgical and Materials Transactions A, 2013. **44**(7): p. 2935-2949.
- 24. Lee, C.C., *Environmental engineering dictionary*. 2005: Government Institutes.
- 25. Calderale, P.M. and F. Pipino, *Osteosynthesis means for the connection of bone fracture segments*. 1992, Google Patents.
- 26. Simon, J.-P. and G. Fabry, *An overview of implant materials.* Acta Orthop Belg, 1991. **57**(1): p. 1-5.
- 27. Witte, F., N. Hort, C. Vogt, S. Cohen, K.U. Kainer, R. Willumeit, and F. Feyerabend, *Degradable biomaterials based on magnesium corrosion.* Current Opinion in Solid State and Materials Science, 2008. **12**(5–6): p. 63-72.
- 28. Ozkan, K., I. Türkmen, A. Sahin, Y. Yildiz, S. Erturk, and M. Soylemez, *A biomechanical comparison of proximal femoral nails and locking proximal anatomic femoral plates in femoral fracture fixation A study on synthetic bones.* Indian Journal of Orthopaedics, 2015. **49**(3): p. 347-351.
- 29. Sealy, M.P. and Y.B. Guo, *Fabrication and Characterization of Surface Texture for Bone Ingrowth by Sequential Laser Peening Biodegradable Orthopedic Magnesium-Calcium Implants.* Journal of Medical Devices, 2011. **5**(1): p. 011003-011003.
- 30. Poinern, G.E.J., S.V. Brundavanam, and D. Fawcett, *Biomedical Magnesium Alloys: A Review of Material Properties, Surface Modifications and Potential as a Biodegradable Orthopaedic Implant.* American Journal of Biomedical Engineering, 2012. **2**(6): p. 218-240.
- 31. Uhthoff, H.K., P. Poitras, and D.S. Backman, *Internal plate fixation of fractures: short history and recent developments.* Journal of Orthopaedic Science, 2006. **11**(2): p. 118-126.
- 32. Song, Y.W., D.Y. Shan, and E.H. Han, *Electrodeposition of hydroxyapatite coating on AZ91D magnesium alloy for biomaterial application.* Materials Letters, 2008. **62**(17–18): p. 3276-3279.
- 33. Frost, H.M., *The Biology of Fracture Healing: An Overview for Clinicians. Part I.* Clinical Orthopaedics and Related Research, 1989. **248**: p. 283-293.
- 34. Sirbu, P.D., G. Berea, P. Botez, R. Asaftei, and T. Petreus, *Minimally Invasive Plate Osteosynthesis (MIPO) in Long Bone Fractures-Biomechanics-Design-Clinical Results*. 2011: INTECH Open Access Publisher.
- 35. Xin, Y., T. Hu, and P.K. Chu, *In vitro studies of biomedical magnesium alloys in a simulated physiological environment: A review.* Acta Biomaterialia, 2011. **7**(4): p. 1452-1459.
- 36. Hampp, C., N. Angrisani, J. Reifenrath, D. Bormann, J.-M. Seitz, and A. Meyer-Lindenberg, *Evaluation of the biocompatibility of two magnesium alloys as degradable implant materials in comparison to titanium as non*‐ *resorbable material in the rabbit.* Materials Science and Engineering: C, 2013. **33**(1): p. 317-326.
- 37. Xue, D., Y. Yun, Z. Tan, Z. Dong, and M.J. Schulz, *In Vivo and In Vitro Degradation Behavior of Magnesium Alloys as Biomaterials.* Journal of Materials Science & Technology, 2012. **28**(3): p. 261-267.
- 38. Witte, F., *The history of biodegradable magnesium implants: A review.* Acta Biomaterialia, 2010. **6**(5): p. 1680-1692.
- 39. Huse, E.C., *A new ligature ?*, in *The Chicago medical journal and examiner*, William H. Byford, J.N. Hyde, Ferd.Chotz, and E.F. Ingals, Editors. 1878, Chicago: Chicago. p. 172.
- 40. Payr, E., *Beiträge zur Technik der Blutgefäss- und Nervennaht nebst Mittheilungen über die Verwendung eines resorbirbaren Metalles in der Chirurgie.* Arch Klin Chir, 1900. **62**: p. 67-93.
- 41. Verbrugge, J., *L'utilisation du magnésium dans le traitement chirurgical des fractures.* Bull Mém Soc Nat Cir, 1937. **59**(59): p. 813-823.
- 42. Verbrugge, J., *Le matériel métallique résorbable en chirurgie osseuse.* Presse Med, 1934. **23**(460): p. 5.
- 43. Lambotte, A., *L'utilisation du magnesium comme materiel perdu dans l'osteosynthèse.* Bull Mem Soc Nat Chir, 1932. **28**: p. 1325-1334.
- 44. Lambotte, A., *Technique et indications de la prothèse perdue dans la traitement des fractures.* Presse Med Belge, 1909. **17**: p. 321-323.
- 45. Andrews, E., *Absorbable metal clips as substitutes for ligatures in wound closure.* JAMA, 1917. **28**: p. 669-680.
- 46. Seelig, M.G., *A study of magnesium wire as an absorbable suture and ligature material.* Archives of Surgery, 1924. **8**(2): p. 669-680.
- 47. Groves, E.W.H., *An experimental study of the operative treatment of fractures.* British Journal of Surgery, 1913. **1**(3): p. 438-501.
- 48. McBRIDE, E.D., *Magnesium screw and nail transfixion in fractures.* South Med Journal, 1938. **31**(5): p. 508-515.
- 49. Maier, O., *Über die Verwendbarkeit von Leichtmetallen in der Chirurgie (metallisches Magnesium als Reizmittel zur Knochenneubildung).* Deut Z Chir, 1940. **253**: p. 552-556.
- 50. McBRIDE, E.D., *Absorbable metal in bone surgery: a further report on the use of magnesium alloys.* Journal of the American Medical Association, 1938. **111**(27): p. 2464-2467.
- 51. Stone, P. and J. Lord Jr, *An experimental study of the thrombogenic properties of magnesium and magnesium-aluminum wire in the dog's aorta.* Surgery, 1951. **30**(6): p. 987.
- 52. Fontenier, G., R. Freschard, and M. Mourot, *Study of the corrosionin vitro andin vivo of magnesium anodes involved in an implantable bioelectric battery.* Medical and biological engineering, 1975. **13**(5): p. 683-689.
- 53. Wexler, B.C., *Pathophysiologic responses of spontaneously hypertensive rats to arterial magnesium—aluminum wire implants.* Atherosclerosis, 1980. **36**(4): p. 575-587.
- 54. Kammer, C., *Magnesium Taschenbuch, chapter Eigenschaften von reinem Magnesium, pages77-97*. 2000, Aluminium Verlag, Düsseldorf.
- 55. Wolf, F.I. and A. Cittadini, *Chemistry and biochemistry of magnesium.* Molecular aspects of medicine, 2003. **24**(1): p. 3-9.
- 56. Huan, Z.G., M.A. Leeflang, J. Zhou, L.E. Fratila-Apachitei, and J. Duszczyk, *In vitro degradation behavior and cytocompatibility of Mg–Zn–Zr alloys.* Journal of Materials Science. Materials in Medicine, 2010. **21**(9): p. 2623-2635.
- 57. Jahnen-Dechent, W. and M. Ketteler, *Magnesium basics.* Clinical Kidney Journal, 2012. **5**(Suppl_1): p. i3-i14.
- 58. Elin, R.J., *Magnesium metabolism in health and disease.* Disease-a-Month. **34**(4): p. 166-218.
- 59. Virtanen, S., *Biodegradable Mg Alloys: Corrosion, Surface Modification, and Biocompatibility*, in *Biomedical Applications*, S.S. Djokić, Editor. 2012, Springer US: Boston, MA. p. 101-125.
- 60. Wu, K. and I. Zhitomirsky, *Electrophoretic deposition of ceramic nanoparticles.* International Journal of Applied Ceramic Technology, 2011. **8**(4): p. 920-927.
- 61. Witte, F., V. Kaese, H. Haferkamp, E. Switzer, A. Meyer-Lindenberg, C.J. Wirth, and H. Windhagen, *In vivo corrosion of four magnesium alloys and the associated bone response.* Biomaterials, 2005. **26**(17): p. 3557-3563.
- 62. Lei, T., C. Ouyang, W. Tang, L.-F. Li, and L.-S. Zhou, *Enhanced corrosion protection of MgO coatings on magnesium alloy deposited by an anodic electrodeposition process.* Corrosion Science, 2010. **52**(10): p. 3504-3508.
- 63. Bakhsheshi-Rad, H.R., M.H. Idris, M.R. Abdul-Kadir, A. Ourdjini, M. Medraj, M. Daroonparvar, and E. Hamzah, *Mechanical and bio-corrosion properties of quaternary Mg-Ca-Mn-Zn alloys compared with binary Mg-Ca alloys.* Materials and Design, 2014. **53**: p. 283-292.
- 64. Wang, Z.-L., Y.-H. Yan, T. Wan, and H. Yang, *Poly (L-lactic acid)/hydroxyapatite/collagen composite coatings on AZ31 magnesium alloy for biomedical application.* Proceedings of the Institution of Mechanical Engineers, Part H: Journal of Engineering in Medicine, 2013. **227**(10): p. 1094- 1103.
- 65. Gao, J.-c., S. Wu, L.-y. Qiao, and Y. Wang, *Corrosion behavior of Mg and Mg-Zn alloys in simulated body fluid.* Transactions of Nonferrous Metals Society of China, 2008. **18**(3): p. 588-592.
- 66. Gray, J.E. and B. Luan, *Protective coatings on magnesium and its alloys a critical review.* Journal of Alloys and Compounds, 2002. **336**(1–2): p. 88-113.
- 67. Verbrugge, J., *La tolérance du tissu osseux vis-à-vis du magnésium métallique.* Presse méd, 1933. **55**: p. 1112-1114.
- 68. Kirkland, N.T., J. Lespagnol, N. Birbilis, and M.P. Staiger, *A survey of biocorrosion rates of magnesium alloys.* Corrosion Science, 2010. **52**(2): p. 287- 291.
- 69. Fekry, A. and M. Ameer, *Electrochemistry and impedance studies on titanium and magnesium alloys in Ringer's solution.* International journal of electrochemical science, 2011. **6**: p. 1342-1354.
- 70. Witte, F., I. Abeln, E. Switzer, V. Kaese, A. Meyer‐Lindenberg, and H. Windhagen, *Evaluation of the skin sensitizing potential of biodegradable magnesium alloys.* Journal of Biomedical Materials Research Part A, 2008. **86**(4): p. 1041-1047.
- 71. Tan, L., X. Yu, P. Wan, and K. Yang, *Biodegradable materials for bone repairs: a review.* Journal of Materials Science & Technology, 2013. **29**(6): p. 503-513.
- 72. Liu, C., Z. Ren, Y. Xu, S. Pang, X. Zhao, and Y. Zhao, *Biodegradable Magnesium Alloys Developed as Bone Repair Materials: A Review.* Scanning, 2018. **2018**: p. 15.
- 73. Sunil, B.R., A.A. Kumar, T.S. Sampath Kumar, and U. Chakkingal, *Role of biomineralization on the degradation of fine grained AZ31 magnesium alloy processed by groove pressing.* Materials Science and Engineering: C, (0).
- 74. Ren, Y., J. Huang, B. Zhang, and K. Yang, *Preliminary study of biodegradation of AZ31B magnesium alloy.* Frontiers of Materials Science in China, 2007. **1**(4): p. 401-404.
- 75. Dai, D., H. Wang, J.-z. Li, and X.-d. Wu, *Environmentally friendly anodization on AZ31 magnesium alloy.* Transactions of Nonferrous Metals Society of China, 2008. **18, Supplement 1**(0): p. s380-s384.
- 76. Zhao, H., Z. Huang, and J. Cui, *Electroless plating of copper on AZ31 magnesium alloy substrates.* Microelectronic Engineering, 2008. **85**(2): p. 253- 258.
- 77. Wan, T.-T., Z.-X. Liu, M.-Z. Bu, and P.-C. Wang, *Effect of surface pretreatment on corrosion resistance and bond strength of magnesium AZ31 alloy.* Corrosion Science, 2013. **66**(0): p. 33-42.
- 78. Witte, F., J. Fischer, J. Nellesen, H.-A. Crostack, V. Kaese, A. Pisch, F. Beckmann, and H. Windhagen, *In vitro and in vivo corrosion measurements of magnesium alloys.* Biomaterials, 2006. **27**(7): p. 1013-1018.
- 79. Song, G., A.L. Bowles, and D.H. StJohn, *Corrosion resistance of aged die cast magnesium alloy AZ91D.* Materials Science and Engineering: A, 2004. **366**(1): p. 74-86.
- 80. Zhiquan, H., H. Qingxue, M. Lifeng, L. Jinbao, and P. Zhining, *Effects of beta-Mg-17 Al-12 on Edge Crack of Roll-Casting AZ31B Magnesium Alloy Plate.* Rare Metal Materials and Engineering, 2014. **43**(5): p. 1199-1203.
- 81. Alvarez-Lopez, M., M.D. Pereda, J.A. del Valle, M. Fernandez-Lorenzo, M.C. Garcia-Alonso, O.A. Ruano, and M.L. Escudero, *Corrosion behaviour of AZ31 magnesium alloy with different grain sizes in simulated biological fluids.* Acta Biomaterialia, 2010. **6**(5): p. 1763-1771.
- 82. Li, N. and Y. Zheng, *Novel Magnesium Alloys Developed for Biomedical Application: A Review.* Journal of Materials Science & Technology, 2013. **29**(6): p. 489-502.
- 83. Nicholson, P.S., P. Sarkar, and X. Haung, *Electrophoretic deposition and its use to synthesize ZrO2/Al2O3 micro-laminate ceramic/ceramic composites.* Journal of Materials Science, 1993. **28**(23): p. 6274-6278.
- 84. Song, G.-L. and M. Liu, *The effect of Mg alloy substrate on "electroless" Ecoating performance.* Corrosion Science, 2011. **53**(11): p. 3500-3508.
- 85. Witte, F., F. Feyerabend, P. Maier, J. Fischer, M. Störmer, C. Blawert, W. Dietzel, and N. Hort, *Biodegradable magnesium–hydroxyapatite metal matrix composites.* Biomaterials, 2007. **28**(13): p. 2163-2174.
- 86. Lin, B., M. Zhong, C. Zheng, L. Cao, D. Wang, L. Wang, J. Liang, and B. Cao, *Preparation and characterization of dopamine-induced biomimetic hydroxyapatite coatings on the AZ31 magnesium alloy.* Surface and Coatings Technology, 2015. **281**: p. 82-88.
- 87. Li, Z., Z. Shang, X. Wei, and Q. Zhao, *Corrosion resistance and cytotoxicity of AZ31 magnesium alloy with N+ ion implantation.* Materials Technology, 2019. **34**(12): p. 730-736.
- 88. Adekanmbi, I., L. Ferguson, P. Tsimbouri, M. Riehle, H. Kubba, and K. Tanner, *Evaluation of Cell Viability of Biodegradable AZ31 Mg Alloy for Use in Paediatric Tracheal Stent after Long Term In Vitro Corrosion.* 2016.
- 89. Makar, G. and J. Kruger, *Corrosion of magnesium.* International materials reviews, 1993. **38**(3): p. 138-153.
- 90. Kirkland, N.T., N. Birbilis, and M.P. Staiger, *Assessing the corrosion of biodegradable magnesium implants: A critical review of current methodologies and their limitations.* Acta Biomaterialia, 2012. **8**(3): p. 925- 936.
- 91. Hornberger, H., S. Virtanen, and A.R. Boccaccini, *Biomedical coatings on magnesium alloys – A review.* Acta Biomaterialia, 2012. **8**(7): p. 2442-2455.
- 92. Yang, G.-l., F.-m. He, J.-a. Hu, X.-x. Wang, and S.-f. Zhao, *Biomechanical comparison of biomimetically and electrochemically deposited hydroxyapatite–coated porous titanium implants.* Journal of Oral and Maxillofacial Surgery, 2010. **68**(2): p. 420-427.
- 93. Paital, S.R., W. He, and N.B. Dahotre, *Laser pulse dependent micro textured calcium phosphate coatings for improved wettability and cell compatibility.* Journal of Materials Science: Materials in Medicine, 2010. **21**(7): p. 2187-2200.
- 94. Liu, F., J. Xu, F. Wang, L. Zhao, and T. Shimizu, *Biomimetic deposition of apatite coatings on micro-arc oxidation treated biomedical NiTi alloy.* Surface and Coatings Technology, 2010. **204**(20): p. 3294-3299.
- 95. Gross, K.A. and C. Berndt, *In vitro testing of plasma-sprayed hydroxyapatite coatings.* Journal of materials science: materials in medicine, 1994. **5**(4): p. 219-224.
- 96. Zhang, J., Y. Chan, and Q. Yu, *Plasma interface engineered coating systems for magnesium alloys.* Progress in Organic Coatings, 2008. **61**(1): p. 28-37.
- 97. Sreekanth, D. and N. Rameshbabu, *Development and characterization of MgO/hydroxyapatite composite coating on AZ31 magnesium alloy by plasma electrolytic oxidation coupled with electrophoretic deposition.* Materials Letters, 2012. **68**(0): p. 439-442.
- 98. Dey, A., A.K. Mukhopadhyay, S. Gangadharan, M.K. Sinha, and D. Basu, *Development of hydroxyapatite coating by microplasma spraying.* Materials and manufacturing processes, 2009. **24**(12): p. 1321-1330.
- 99. Iqbal, N., R. Nazir, A. Asif, A.A. Chaudhry, M. Akram, G.Y. Fan, A. Akram, R. Amin, S.H. Park, and R. Hussain, *Electrophoretic deposition of PVA coated hydroxyapatite on 316L stainless steel.* Current Applied Physics, 2012. **12**(3): p. 755-759.
- 100. Singh, I., C. Kaya, M.S.P. Shaffer, B.C. Thomas, and A.R. Boccaccini, *Bioactive ceramic coatings containing carbon nanotubes on metallic substrates by electrophoretic deposition.* Journal of Materials Science, 2006. **41**(24): p. 8144-8151.
- 101. Riccardis, M.F.D., *Ceramic Coatings Obtained by Electrophoretic Deposition: Fundamentals, Models, Post- Deposition Processes and Applications*, in *Ceramic Coatings - Applications in Engineering*, F. Shi, Editor. 2012, Intech.
- 102. Kwok, C., P. Wong, F. Cheng, and H. Man, *Characterization and corrosion behavior of hydroxyapatite coatings on Ti6Al4V fabricated by electrophoretic deposition.* Applied Surface Science, 2009. **255**(13): p. 6736-6744.
- 103. Santillán, M., F. Membrives, N. Quaranta, and A. Boccaccini, *Characterization of TiO2 nanoparticle suspensions for electrophoretic deposition.* Journal of Nanoparticle Research, 2008. **10**(5): p. 787-793.
- 104. Zhang, J. and C. Wu, *Corrosion protection behavior of AZ31 magnesium alloy with cathodic electrophoretic coating pretreated by silane.* Progress in Organic Coatings, 2009. **66**(4): p. 387-392.
- 105. Ma, J. and W. Cheng, *Deposition and packing study of sub-micron PZT ceramics using electrophoretic deposition.* Materials Letters, 2002. **56**(5): p. 721-727.
- 106. Mishra, M., S. Bhattacharjee, L. Besra, H.S. Sharma, T. Uchikoshi, and Y. Sakka, *Effect of pH localization on microstructure evolution of deposits during aqueous electrophoretic deposition (EPD).* Journal of the European Ceramic Society, 2010. **30**(12): p. 2467-2473.
- 107. Feuillard, G., D. Kuscer, L.P. Tran Huu Hue, E. Le Clezio, M. Kosec, and M. Lethiecq. *Electroacoustic performance of high frequency PZT based transducer fabricated by electrophoretic deposition: comparison with screen printing technique*. in *Ultrasonics Symposium (IUS), 2009 IEEE International*. 2009.
- 108. Cao, R.J., S.C. Zhao, G.R. Li, J.T. Zeng, and L.Y. Zheng, *Microstructure and Electrical Properties of the PNN-PZT Thick Films Prepared by Electrophoretic Deposition.* Ferroelectrics, 2010. **411**(1): p. 28-35.
- 109. Dorozhkin, S., *Calcium orthophosphates.* Journal of Materials Science, 2007. **42**(4): p. 1061-1095.
- 110. Wu, G., J.M. Ibrahim, and P.K. Chu, *Surface design of biodegradable magnesium alloys — A review.* Surface and Coatings Technology, 2013. **233**(0): p. 2-12.
- 111. Zhong, C., F. Liu, Y. Wu, J. Le, L. Liu, M. He, J. Zhu, and W. Hu, *Protective diffusion coatings on magnesium alloys: A review of recent developments.* Journal of Alloys and Compounds, 2012. **520**(0): p. 11-21.
- 112. Basu, R.N., C.A. Randall, and M.J. Mayo, *Fabrication of Dense Zirconia Electrolyte Films for Tubular Solid Oxide Fuel Cells by Electrophoretic Deposition.* Journal of the American Ceramic Society, 2001. **84**(1): p. 33-40.
- 113. Cao, X., R. Vassen, W. Fischer, F. Tietz, W. Jungen, and D. Stoever, *Lanthanum*–*Cerium Oxide as a Thermal Barrier*‐*Coating Material for High*‐*Temperature Applications.* Advanced Materials, 2003. **15**(17): p. 1438- 1442.
- 114. Jelinek, M., *23 Hybrid laser technology for biomaterials A2 Jelínková, Helena*, in *Lasers for Medical Applications*. 2013, Woodhead Publishing. p. 704-724.
- 115. Orava, J., T. Kohoutek, and T. Wagner, *9 Deposition techniques for chalcogenide thin films*, in *Chalcogenide Glasses*. 2014, Woodhead Publishing. p. 265-309.
- 116. Zhitomirsky, I. and L. Gal-Or, *Electrophoretic deposition of hydroxyapatite.* Journal of Materials Science: Materials in Medicine, 1997. **8**(4): p. 213-219.
- 117. Hadraba, H., D. Drdlik, Z. Chlup, K. Maca, I. Dlouhy, and J. Cihlar, *Laminated alumina/zirconia ceramic composites prepared by electrophoretic deposition.* Journal of the European Ceramic Society, 2012. **32**(9): p. 2053-2056.
- 118. Seuss, S., M. Lehmann, and A.R. Boccaccini, *Alternating current electrophoretic deposition of antibacterial bioactive glass-chitosan composite coatings.* International journal of molecular sciences, 2014. **15**(7): p. 12231- 12242.
- 119. Ata, M.S., Y. Sun, X. Li, and I. Zhitomirsky, *Electrophoretic deposition of graphene, carbon nanotubes and composites using aluminon as charging and film forming agent.* Colloids and Surfaces A: Physicochemical and Engineering Aspects, 2012. **398**(0): p. 9-16.
- 120. Heavens, S., *Electrophoretic deposition as a processing route for ceramics.* Noyes Publications, Advanced Ceramic Processing and Technology., 1990. **1**: p. 255-283.
- 121. Ahn, J.R., G.S. Kang, H.J. Lee, and C.J. Park, *Electrophoretic Deposition of Oxide Nanoparticles for Electron Emission Enhancement.* Journal of Materials Science & amp; Technology, 2010. **26**(11): p. 1032-1036.
- 122. Bartmanski, M., B. Cieslik, J. Glodowska, P. Kalka, L. Pawlowski, M. Pieper, and A. Zielinski, *Electrophoretic deposition (EPD) of nanohydroxyapatite nanosilver coatings on Ti13Zr13Nb alloy.* Ceramics International, 2017. **43**(15): p. 11820-11829.
- 123. Neirinck, B., T. Mattheys, A. Braem, J. Fransaer, O. Van der Biest, and J. Vleugels, *Porous titanium coatings obtained by electrophoretic deposition (EPD) of pickering emulsions and microwave sintering.* Advanced engineering materials, 2008. **10**(3): p. 246-249.
- 124. Assadian, M., H. Jafari, S.M. Ghaffari Shahri, M.H. Idris, and B. Gholampour, *Corrosion resistance of EPD nanohydroxyapatite coated 316L stainless steel.* Surface Engineering, 2014. **30**(11): p. 806-813.
- 125. Javidi, M., S. Javadpour, M.E. Bahrololoom, and J. Ma, *Electrophoretic deposition of natural hydroxyapatite on medical grade 316L stainless steel.* Materials Science and Engineering: C, 2008. **28**(8): p. 1509-1515.
- 126. Mehdipour, M., A. Afshar, and M. Mohebali, *Electrophoretic deposition of bioactive glass coating on 316L stainless steel and electrochemical behavior study.* Applied Surface Science, 2012. **258**(24): p. 9832-9839.
- 127. Hasegawa, K., S. Kunugi, M. Tatsumisago, and T. Minami, *Preparation of thick films by electrophoretic deposition using surface modified silica particles derived from sol-gel method.* Journal of sol-gel science and technology, 1999. **15**(3): p. 243-249.
- 128. Zhang, Z., Y. Huang, and Z. Jiang, *Electrophoretic deposition forming of SiC*‐*TZP composites in a nonaqueous sol media.* Journal of the American Ceramic Society, 1994. **77**(7): p. 1946-1949.
- 129. Sato, N., M. Kawachi, K. Noto, N. Yoshimoto, and M. Yoshizawa, *Effect of particle size reduction on crack formation in electrophoretically deposited YBCO films.* Physica C: Superconductivity, 2001. **357-360**: p. 1019-1022.
- 130. Santillán, M.J., N.E. Quaranta, and A.R. Boccaccini, *Titania and titania–silver nanocomposite coatings grown by electrophoretic deposition from aqueous suspensions.* Surface and Coatings Technology, 2010. **205**(7): p. 2562-2571.
- 131. Pang, X. and I. Zhitomirsky, *Electrophoretic deposition of composite hydroxyapatite-chitosan coatings.* Materials Characterization, 2007. **58**(4): p. 339-348.
- 132. Powers, R., *The Electrophoretic Forming of Beta Alumina Ceramic.* Journal of The Electrochemical Society, 1975. **122**(4): p. 490-500.
- 133. Negishi, H., H. Yanagishita, and H. Yokokawa, *Electrophoretic deposition of solid oxide fuel cell material powders.* Proceedings of the electrochemical society on electrophoretic deposition: fundamentals and applications, 2002. **21**: p. 214-221.
- 134. Dörner, L., P. Schmutz, R. Kägi, M.V. Kovalenko, and L.P.H. Jeurgens, *Electrophoretic Deposition of Nanoporous Oxide Coatings from Concentrated CuO Nanoparticle Dispersions.* Langmuir, 2020. **36**(28): p. 8075-8085.
- 135. Hadraba, H., K. Maca, and J. Cihlar, *Electrophoretic deposition of alumina and zirconia: II. Two-component systems.* Ceramics International, 2004. **30**(6): p. 853-863.
- 136. Wang, C., J. Ma, W. Cheng, and R. Zhang, *Thick hydroxyapatite coatings by electrophoretic deposition.* Materials Letters, 2002. **57**(1): p. 99-105.
- 137. Besra, L., P. Samantaray, S. Bhattacharjee, and B. Singh, *Electrophoretic deposition of alumina on stainless steel from non-aqueous suspension.* Journal of Materials Science, 2007. **42**(14): p. 5714-5721.
- 138. González‐Cuenca, M., P.M. Biesheuvel, and H. Verweij, *Modeling constant voltage electrophoretic deposition from a stirred suspension.* AIChE journal, 2000. **46**(3): p. 626-631.
- 139. Kim, S., S. Cho, J. Lee, S. Samal, and H. Kim, *Relationship between the process parameters and the saturation point in electrophoretic deposition.* Ceramics International, 2012. **38**(6): p. 4617-4622.
- 140. Nuswantoro, N.F., I. Budiman, A. Septiawarman, H.T. Djong, M. Manjas, and Gunawarman, *Effect of Applied Voltage and Coating Time on Nano Hydroxyapatite Coating on Titanium Alloy Ti6Al4V Using Electrophoretic Deposition for Orthopaedic Implant Application.* IOP Conference Series: Materials Science and Engineering, 2019. **547**: p. 1-11.
- 141. Meng, X., T. Kwon, Y. Yang, J.L. Ong, and K. Kim, *Effects of Applied Voltages on Hydroxyapatite Coating of Titanium by Electrophoretic Deposition.* Biomed. Mater. Res. Part B Appl. Biomater., 2006. **78B**: p. 373- 377.
- 142. Stojanovic, D., B. Jokic, D. Veljovic, R. Petrovic, P. Uskokovic, and D. Janackovic, *Bioactive glass–apatite composite coating for titanium implant synthesized by electrophoretic deposition.* Journal of the European Ceramic Society, 2007. **27**(2-3): p. 1595-1599.
- 143. Boccaccini, A.R., J. Cho, T. Subhani, C. Kaya, and F. Kaya, *Electrophoretic deposition of carbon nanotube–ceramic nanocomposites.* Journal of the European Ceramic Society, 2010. **30**(5): p. 1115-1129.
- 144. Nie, X., A. Leyland, and A. Matthews, *Deposition of layered bioceramic hydroxyapatite/TiO2 coatings on titanium alloys using a hybrid technique of micro-arc oxidation and electrophoresis.* Surface and Coatings Technology, 2000. **125**(1–3): p. 407-414.
- 145. Hosseinbabaei, F. and B. Raissidehkordi, *Electrophoretic deposition of MgO thick films from an acetone suspension.* Journal of the European Ceramic Society, 2000. **20**(12): p. 2165-2168.
- 146. Mehdipour, M. and A. Afshar, *A study of the electrophoretic deposition of bioactive glass–chitosan composite coating.* Ceramics International, 2012. **38**(1): p. 471-476.
- 147. Chen, Q., C. Wong, W. Lu, K. Cheung, J. Leong, and K. Luk, *Strengthening mechanisms of bone bonding to crystalline hydroxyapatite in vivo.* Biomaterials, 2004. **25**(18): p. 4243-4254.
- 148. Hench, L.L., *Chronology of bioactive glass development and clinical applications.* New Journal of Glass and Ceramics, 2013. **3**(02): p. 67.
- 149. Crowninshield, R., *An overview of prosthetic materials for fixation.* Clinical orthopaedics and related research, 1988(235): p. 166-172.
- 150. Geesink, R., K. de Groot, and C. Klein, *Bonding of bone to apatite-coated implants.* Bone & Joint Journal, 1988. **70**(1): p. 17-22.
- 151. Mondragon-Cortez, P. and G. Vargas-Gutierrez, *Electrophoretic deposition of hydroxyapatite submicron particles at high voltages.* Materials Letters, 2004. **58**(7-8): p. 1336-1339.
- 152. Bose, S. and S. Tarafder, *Calcium phosphate ceramic systems in growth factor and drug delivery for bone tissue engineering: A review.* Acta Biomaterialia, 2012. **8**(4): p. 1401-1421.
- 153. Hahn, B.-D., D.-S. Park, J.-J. Choi, J. Ryu, W.-H. Yoon, J.-H. Choi, H.-E. Kim, and S.-G. Kim, *Aerosol deposition of hydroxyapatite–chitosan composite coatings on biodegradable magnesium alloy.* Surface and Coatings Technology, 2011. **205**(8–9): p. 3112-3118.
- 154. Loghmani, S.K., M. Farrokhi-Rad, and T. Shahrabi, *Effect of polyethylene glycol on the electrophoretic deposition of hydroxyapatite nanoparticles in isopropanol.* Ceramics International, 2013. **39**(6): p. 7043-7051.
- 155. Farrokhi-Rad, M. and T. Shahrabi, *Effect of suspension medium on the electrophoretic deposition of hydroxyapatite nanoparticles and properties of obtained coatings.* Ceramics International, 2014. **40**(2): p. 3031-3039.
- 156. Grandfield, K. and I. Zhitomirsky, *Electrophoretic deposition of composite hydroxyapatite–silica–chitosan coatings.* Materials Characterization, 2008. **59**(1): p. 61-67.
- 157. Wei, M., A.J. Ruys, B.K. Milthorpe, C.C. Sorrell, and J.H. Evans, *Electrophoretic Deposition of Hydroxyapatite Coatings on Metal Substrates: A Nanoparticulate Dual-Coating Approach.* Journal of Sol-Gel Science and Technology, 2001. **21**(1-2): p. 39-48.
- 158. Farrokhi-Rad, M., S.K. Loghmani, T. Shahrabi, and S. Khanmohammadi, *Electrophoretic deposition of hydroxyapatite nanostructured coatings with controlled porosity.* Journal of the European Ceramic Society, 2014. **34**(1): p. 97-106.
- 159. Ma, J., C. Wang, and K.W. Peng, *Electrophoretic deposition of porous hydroxyapatite scaffold.* Biomaterials, 2003. **24**(20): p. 3505-3510.
- 160. Maleki-Ghaleh, H., V. Khalili, J. Khalil-Allafi, and M. Javidi, *Hydroxyapatite coating on NiTi shape memory alloy by electrophoretic deposition process.* Surface and Coatings Technology, 2012. **208**(0): p. 57-63.
- 161. Hu, J., Q. Li, X. Zhong, L. Zhang, and B. Chen, *Composite anticorrosion coatings for AZ91D magnesium alloy with molybdate conversion coating and silicon sol–gel coatings.* Progress in Organic Coatings, 2009. **66**(3): p. 199-205.
- 162. Collazo, A., A. Covelo, M. Izquierdo, X.R. Nóvoa, and C. Pérez, *Effect of the experimental setup in the behaviour of sol–gel coatings.* Progress in Organic Coatings, 2008. **63**(3): p. 291-298.
- 163. Wang, D. and G.P. Bierwagen, *Sol–gel coatings on metals for corrosion protection.* Progress in Organic Coatings, 2009. **64**(4): p. 327-338.
- 164. Singh, S., R.M. Kumar, K.K. Kuntal, P. Gupta, S. Das, R. Jayaganthan, P. Roy, and D. Lahiri, *Sol–Gel Derived Hydroxyapatite Coating on Mg-3Zn Alloy for Orthopedic Application.* JOM, 2015. **67**(4): p. 702-712.
- 165. Carter, C.B. and M.G. Norton, *Ceramics in Biology and Medicine*, in *Ceramic Materials: Science and Engineering*. 2013, Springer New York: New York, NY. p. 659-676.
- 166. Ducheyne, P., W. Van Raemdonck, J. Heughebaert, and M. Heughebaert, *Structural analysis of hydroxyapatite coatings on titanium.* Biomaterials, 1986. **7**(2): p. 97-103.
- 167. Kannan, M.B. and L. Orr, *In vitro mechanical integrity of hydroxyapatite coated magnesium alloy.* Biomedical Materials, 2011. **6**(4): p. 045003.
- 168. Sridhar, T.M., U. Kamachi Mudali, and M. Subbaiyan, *Preparation and characterisation of electrophoretically deposited hydroxyapatite coatings on type 316L stainless steel.* Corrosion Science, 2003. **45**(2): p. 237-252.
- 169. Jasim, M., M.K. Abbass, K. Salah, and A.N. Jasim, *Characterization of electrophoretic deposition parameters of nano hydroxyapatite coating on the Ti6Al4V alloy using DC current.* AIP Conference Proceedings, 2020. **2213**(1): p. 020203.
- 170. Nordlien, J., S. Ono, N. Masuko, and K. Nisancioglu, *A TEM investigation of naturally formed oxide films on pure magnesium.* Corrosion science, 1997. **39**(8): p. 1397-1414.
- 171. Ferrari, B., R. Moreno, P. Sarkar, and P.S. Nicholson, *Electrophoretic deposition of MgO from organic suspensions.* Journal of the European Ceramic Society, 2000. **20**(2): p. 99-106.
- 172. Castro, R.R., P.B. Marcos, E. Sakamoto, and D. Gouvêa, *Surface reactivity and electrophoretic deposition of ZrO2–MgO mechanical mixture.* Journal of Materials Science, 2007. **42**(16): p. 6946-6950.
- 173. Wang, Y.H., Q.Z. Chen, J. Cho, and A.R. Boccaccini, *Electrophoretic codeposition of diamond/borosilicate glass composite coatings.* Surface and Coatings Technology, 2007. **201**(18): p. 7645-7651.
- 174. Ferrari, B., S. González, R. Moreno, and C. Baudín, *Multilayer coatings with improved reliability produced by aqueous electrophoretic deposition.* Journal of the European Ceramic Society, 2006. **26**(1–2): p. 27-36.
- 175. Ahmed, Y. and M.A. Ur Rehman, *Improvement in the surface properties of stainless steel via zein/hydroxyapatite composite coatings for biomedical applications.* Surfaces and Interfaces, 2020. **20**: p. 100589.
- 176. Patil, P. and S. Nimbalkar-Patil, *Lost wax-bolus technique to process closed hollow obturator with uniform wall thickness using single flasking procedure.* Journal of the International Clinical Dental Research Organization, 2016. **8**(1): p. 84-88.
- 177. Cobo-Vázquez, C., D. Reininger, P. Molinero-Mourelle, J. González-Serrano, B. Guisado-Moya, and J. López-Quiles, *Effect of the lack of primary stability in the survival of dental implants.* Journal of clinical and experimental dentistry, 2018. **10**(1): p. e14-e19.
- 178. Mistry, G., O. Shetty, S. Shetty, and R. Singh, *Measuring implant stability: A review of different methods.* Journal of Dental Implants, 2014. **4**(2): p. 165-169.
- 179. Rasmusson, L., K.E. Kahnberg, and A. Tan, *Effects of implant design and surface on bone regeneration and implant stability: an experimental study in the dog mandible.* Clin Implant Dent Relat Res, 2001. **3**(1): p. 2-8.
- 180. Saini, M., Y. Singh, P. Arora, V. Arora, and K. Jain, *Implant biomaterials: A comprehensive review.* World journal of clinical cases, 2015. **3**(1): p. 52-57.
- 181. Khalili, V., J. Khalil-Allafi, J. Frenzel, and G. Eggeler, *Bioactivity and electrochemical behavior of hydroxyapatite-silicon-multi walled carbon nanotubes composite coatings synthesized by EPD on NiTi alloys in simulated body fluid.* Materials Science and Engineering: C, 2017. **71**: p. 473-482.
- 182. Chew, K.-K., S.H.S. Zein, A.L. Ahmad, D.S. McPhail, and M.F. Abdullah, *The electrochemical studies of the corrosion resistance behaviour of hydroxyapatite coatings on stainless steel fabricated by electrophoretic deposition.* Journal of Industrial and Engineering Chemistry, 2013. **19**(4): p. 1123-1129.
- 183. Janković, A., S. Eraković, M. Mitrić, I.Z. Matić, Z.D. Juranić, G.C.P. Tsui, C. y. Tang, V. Mišković-Stanković, K.Y. Rhee, and S.J. Park, *Bioactive hydroxyapatite/graphene composite coating and its corrosion stability in simulated body fluid.* Journal of Alloys and Compounds, 2015. **624**: p. 148- 157.
- 184. Ramay, H.R.R. and M. Zhang, *Biphasic calcium phosphate nanocomposite porous scaffolds for load-bearing bone tissue engineering.* Biomaterials, 2004. **25**(21): p. 5171-5180.
- 185. Meng, D., L. Francis, I. Roy, and A. Boccaccini, *Using electrophoretic deposition to identify protein charge in biological medium.* Journal of Applied Electrochemistry, 2011. **41**(8): p. 919-923.
- 186. Braem, A., T. Mattheys, B. Neirinck, J. Schrooten, O. Van der Biest, and J. Vleugels, *Porous titanium coatings through electrophoretic deposition of TiH2 suspensions.* Advanced Engineering Materials, 2011. **13**(6): p. 509-515.
- 187. Chitsaz-Khoyi, L., J. Khalil-Allafi, A. Motallebzadeh, and M. Etminanfar, *The effect of hydroxyapatite nanoparticles on electrochemical and mechanical performance of TiC/N coating fabricated by plasma electrolytic saturation method.* Surface and Coatings Technology, 2020. **394**: p. 125817.
- 188. Lu, X.J. and P. Xiao, *Constrained sintering of YSZ/Al2O3 composite coatings on metal substrates produced from eletrophoretic deposition.* Journal of the European Ceramic Society, 2007. **27**(7): p. 2613-2621.
- 189. Feil, F., W. Fürbeth, and M. Schütze, *Purely inorganic coatings based on nanoparticles for magnesium alloys.* Electrochimica Acta, 2009. **54**(9): p. 2478-2486.
- 190. Huehnerschulte, T.A., J. Reifenrath, B. von Rechenberg, D. Dziuba, J.M. Seitz, D. Bormann, H. Windhagen, and A. Meyer-Lindenberg, *In vivo assessment of the host reactions to the biodegradation of the two novel magnesium alloys ZEK100 and AX30 in an animal model.* Biomed Eng Online, 2012. **11**: p. 14.
- 191. Dorozhkin, S., *Calcium orthophosphate coatings, films and layers.* Progress in Biomaterials, 2012. **1**(1): p. 1-40.
- 192. Tseng, S.-T. and H.-T. Hu, *Finite element verification on constitutive law of AZ31 magnesium alloy at 400 °C.* Transactions of Nonferrous Metals Society of China, 2013. **23**(11): p. 3372-3382.
- 193. Liu, D.-M., *Densification of zirconia from submicron-sized to nano-sized powder particles.* Journal of Materials Science Letters, 1998. **17**(6): p. 467-469.
- 194. Feil, F., W. Fu¨rbeth, and M. Schu¨tze, *Nanoparticle based inorganic coatings for corrosion protection of magnesium alloys.* Surface Engineering, 2008. **24**(3): p. 6.
- 195. Kannan, M.B., *Electrochemical deposition of calcium phosphates on magnesium and its alloys for improved biodegradation performance: A review.* Surface and Coatings Technology, 2016. **301**: p. 36-41.
- 196. Petzow, G., *Metallographic etching: techniques for metallography, ceramography, plastography*. 1999: ASM international.
- 197. Duygulu, O., S. Ucuncuoglu, G. Oktay, D.S. Temur, O. Yucel, and A.A. Kaya, *Development of Rolling Technology for Twin Roll Casted 1500mm Wıde Magnesium AZ31 Alloy.* Submitted to Magnesium Technology 2009 TMS, San Fransisco, USA 2009, 2009.
- 198. ASTMInternational, *ASTM F2129-08: Test Method for Conducting Cyclic Potentiodynamic Polarization Measurements to Determine the Corrosion Susceptibility of Small Implant Devices*. 2008: West Conshohocken, PA.
- 199. ASTMInternational, *ASTM G31-72: Standard Practice for Laboratory Immersion Corrosion Testing of Metals*. 2004: West Conshohocken.
- 200. Murray, J.N., *Electrochemical test methods for evaluating organic coatings on metals: an update. Part III: Multiple test parameter measurements1Note: References 1, 2, 3, 4, 5, 6, 7, 8, 9, 10, 11, 12, 13, 14, 15, 16, 17, 18, 19, 20, 21, 22, 23, 24, 25, 26, 27, 28, 29, 30, 31, 32, 33, 34, 35, 36, 37, 38, 39, 40, 41, 42, 43, 44, 45, 46, 47, 48, 49, 50, 51, 52, 53, 54, 55, 56, 57, 58, 59, 60, 61, 62, 63, 64, 65, 66, 67, 68, 69, 70, 71, 72, 73, 74, 75, 76, 77, 78, 79, 80, 81, 82, 83, 84, 85, 86, 87, 88, 89, 90, 91, 92, 93, 94, 95, 96, 97, 98, 99, 100, 101, 102, 103, 104, 105, 106were initiated within Parts I and II of this three part article. References also cited within this Part III are repeated below as a convenience to the reader1.* Progress in Organic Coatings, 1997. **31**(4): p. 375-391.
- 201. Alsamuraee, A.M.A. and H.I. Jaafer, *Electrochemical impedance spectroscopic evaluation of corrosion protection properties of polyurethane/polyvinyl chloride blend coatings on steel Abdulkareem Mohammed Ali Alsamuraee1, Harith Ibraheem Jaafer2, Hani Aziz Ameen3 and Ahmed Qasim Abdullah4.* 2011.
- 202. Durán, A., Y. Castro, A. Conde, and J.J. de Damborenea, *Sol-gel protective coatings for metals.* Handbook of sol-gel science and technology. 3. Applications of sol-gel technology, 2004. **1**: p. 399.
- 203. Zomorodian, A., F. Brusciotti, A. Fernandes, M.J. Carmezim, T. Moura e Silva, J.C.S. Fernandes, and M.F. Montemor, *Anti-corrosion performance of a new silane coating for corrosion protection of AZ31 magnesium alloy in Hank's solution.* Surface and Coatings Technology, 2012. **206**(21): p. 4368-4375.
- 204. Yu, K., S.-t. Rui, X.-y. Wang, R.-c. Wang, and W.-x. Li, *Texture evolution of extruded AZ31 magnesium alloy sheets.* Transactions of Nonferrous Metals Society of China, 2009. **19**(3): p. 511-516.
- 205. Chen, Y., Q. Wang, J. Peng, C. Zhai, and W. Ding, *Effects of extrusion ratio on the microstructure and mechanical properties of AZ31 Mg alloy.* Journal of Materials Processing Technology, 2007. **182**(1–3): p. 281-285.
- 206. Zhang, B.-P., Y.-F. Tu, J.-Y. Chen, H.-L. Zhang, Y.-L. Kang, and H.G. Suzuki, *Preparation and characterization of as-rolled AZ31 magnesium alloy sheets.* Journal of Materials Processing Technology, 2007. **184**(1): p. 102-107.
- 207. Lu, L., T. Liu, Y. Chen, and Z. Wang, *Deformation and fracture behavior of hot extruded Mg alloys AZ31.* Materials Characterization, 2012. **67**(Supplement C): p. 93-100.
- 208. ASTMInternational, *ASTM B90/B90M-12: Standard Specification for Magnesium-Alloy Sheet and Plate*. 2012: West Conshohocken, PA.
- 209. Abdelaal, O.M. and S.H. Darwish, *Review of Rapid Prototyping Techniques for Tissue Engineering Scaffolds Fabrication*, in *Characterization and Development of Biosystems and Biomaterials*, A. Öchsner, L.F.M. Silva, and H. Altenbach, Editors. 2013, Springer Berlin Heidelberg. p. 33-54.
- 210. Seyedraoufi, Z.S. and S. Mirdamadi, *In vitro biodegradability and biocompatibility of porous Mg-Zn scaffolds coated with nano hydroxyapatite via pulse electrodeposition.* Transactions of Nonferrous Metals Society of China, 2015. **25**(12): p. 4018-4027.
- 211. Chávez-Valdez, A., M. Herrmann, and A.R. Boccaccini, *Alternating current electrophoretic deposition (EPD) of TiO2 nanoparticles in aqueous suspensions.* Journal of Colloid and Interface Science, 2012. **375**(1): p. 102- 105.
- 212. L, H.L. and W. June, *An Introduction To Bioceramics*. 1993: World Scientific Publishing Company.
- 213. Witte, F. and A. Eliezer, *Biodegradable metals*, in *Degradation of implant materials*. 2012, Springer. p. 93-109.
- 214. Shi, Z., M. Liu, and A. Atrens, *Measurement of the corrosion rate of magnesium alloys using Tafel extrapolation.* Corrosion Science, 2010. **52**(2): p. 579-588.
- 215. Poorqasemi, E., O. Abootalebi, M. Peikari, and F. Haqdar, *Investigating accuracy of the Tafel extrapolation method in HCl solutions.* Corrosion Science, 2009. **51**(5): p. 1043-1054.
- 216. Cheng, Y.-l., H.-I. Wu, Z.-h. Chen, H.-m. Wang, Z. Zhang, and Y.-w. Wu, *Corrosion properties of AZ31 magnesium alloy and protective effects of chemical conversion layers and anodized coatings.* Transactions of Nonferrous Metals Society of China, 2007. **17**(3): p. 502-508.
- 217. Bai, N., Tan, Cui , Li, Qing; *, Xi, Zhongxian, *Study on the corrosion resistance and anti-infection of modified magnesium alloy.* Bio-Medical Materials and Engineering 17 April 2017. **vol.28**(no.4): p. pp.339-345.
- 218. Dhal, J.P., M. Sethi, B.G. Mishra, and G. Hota, *MgO nanomaterials with different morphologies and their sorption capacity for removal of toxic dyes.* Materials Letters, 2015. **141**: p. 267-271.
- 219. Hiromoto, S., *4 Corrosion of metallic biomaterials*, in *Metals for Biomedical Devices*, M. Niinomi, Editor. 2010, Woodhead Publishing. p. 99-121.
- 220. Ferrari, B., A. Bartret, and C. Baudín, *Sandwich materials formed by thick alumina tapes and thin-layered alumina–aluminium titanate structures shaped by EPD.* Journal of the European Ceramic Society, 2009. **29**(6): p. 1083-1092.
- 221. Ferrari, B. and R. Moreno, *The conductivity of aqueous Al2O3 slips for electrophoretic deposition.* Materials letters, 1996. **28**(4-6): p. 353-355.
- 222. Vandeperre, L., O. Van der Biest, and W. Clegg. *Silicon carbide laminates with carbon interlayers by electrophoretic deposition*. in *Key Engineering Materials*. 1997. Trans Tech Publ.
- 223. Wu, C.-y. and J. Zhang, *State-of-art on corrosion and protection of magnesium alloys based on patent literatures.* Transactions of Nonferrous Metals Society of China, 2011. **21**(4): p. 892-902.
- 224. Hu, R.-G., S. Zhang, J.-F. Bu, C.-J. Lin, and G.-L. Song, *Recent progress in corrosion protection of magnesium alloys by organic coatings.* Progress in Organic Coatings, 2012. **73**(2–3): p. 129-141.
- 225. Abdal-hay, A., N.A.M. Barakat, and J.K. Lim, *Hydroxyapatite-doped poly(lactic acid) porous film coating for enhanced bioactivity and corrosion behavior of AZ31 Mg alloy for orthopedic applications.* Ceramics International, 2013. **39**(1): p. 183-195.
- 226. Lin, Z., Y. Xiao, L. Wang, Y. Yin, J. Zheng, H. Yang, and G. Chen, *Facile synthesis of enzyme-inorganic hybrid nanoflowers and their application as an immobilized trypsin reactor for highly efficient protein digestion.* RSC Advances, 2014. **4**(27): p. 13888-13891.
- 227. Vishnu, J., V. K Manivasagam, V. Gopal, C. Bartomeu Garcia, P. Hameed, G. Manivasagam, and T.J. Webster, *Hydrothermal treatment of etched titanium: A potential surface nano-modification technique for enhanced biocompatibility.* Nanomedicine: Nanotechnology, Biology and Medicine, 2019. **20**: p. 102016.
- 228. Wang, L., C.-H. Liu, Y. Nemoto, N. Fukata, K.C.W. Wu, and Y. Yamauchi, *Rapid synthesis of biocompatible gold nanoflowers with tailored surface textures with the assistance of amino acid molecules.* RSC Advances, 2012. **2**(11): p. 4608-4611.
- 229. Wen, C., S. Guan, L. Peng, C. Ren, X. Wang, and Z. Hu, *Characterization and degradation behavior of AZ31 alloy surface modified by bone-like hydroxyapatite for implant applications.* Applied Surface Science, 2009. **255**(13-14): p. 6433-6438.
- 230. Nayak, S., B. Satpati, R.K. Shukla, A. Dhawan, S. Bhattacharjee, and Y.S. Chaudhary, *Facile synthesis of nanostructured hydroxyapatite–titania bioimplant scaffolds with different morphologies: their bioactivity and corrosion behaviour.* Journal of Materials Chemistry, 2010. **20**(23): p. 4949-4954.
- 231. Dorozhkin, S.V., *Calcium orthophosphate coatings on magnesium and its biodegradable alloys.* Acta Biomaterialia, (0).
- 232. Zhou, H., Y.-Y. Jiang, S. Tan, L.-J. Liu, Q.-T. Yao, Y.-J. Xia, Y.-S. Fan, J.-P. Hu, Z.-F. Zhou, B.-Q. Lu, S.-S. He, and F. Chen, *Flower-like calcium phosphoserine complex as biomimetic mineral with high bioactivity.* Ceramics International, 2020. **46**(13): p. 20914-20922.
- 233. Wang, Z., S. Jiang, Y. Zhao, and M. Zeng, *Synthesis and characterization of hydroxyapatite nano-rods from oyster shell with exogenous surfactants.* Materials Science and Engineering: C, 2019. **105**: p. 110102.
- 234. Hu, Q., W. Jiang, Y. Li, X. Chen, J. Liu, T. Chen, and G. Miao, *The effects of morphology on physicochemical properties, bioactivity and biocompatibility of micro-/nano-bioactive glasses.* Advanced Powder Technology, 2018. **29**(8): p. 1812-1819.
- 235. Wang, Y., M. Wei, J. Gao, J. Hu, and Y. Zhang, *Corrosion process of pure magnesium in simulated body fluid.* Materials Letters, 2008. **62**(14): p. 2181- 2184.
- 236. Bakhsheshi-Rad, H., E. Hamzah, M. Daroonparvar, M. Yajid, M. Kasiri-Asgarani, M. Abdul-Kadir, and M. Medraj, *In-vitro degradation behavior of Mg alloy coated by fluorine doped hydroxyapatite and calcium deficient hydroxyapatite.* Transactions of Nonferrous Metals Society of China, 2014. **24**(8): p. 2516-2528.
- 237. Bakhsheshi-Rad, H.R., E. Hamzah, M. Daroonparvar, R. Ebrahimi-Kahrizsangi, and M. Medraj, *In-vitro corrosion inhibition mechanism of*

fluorine-doped hydroxyapatite and brushite coated Mg–Ca alloys for biomedical applications. Ceramics International, 2014. **40**(6): p. 7971-7982.

- 238. Chun-Yan, Z., Z. Rong-Chang, L. Cheng-Long, and G. Jia-Cheng, *Comparison of calcium phosphate coatings on Mg–Al and Mg–Ca alloys and their corrosion behavior in Hank's solution.* Surface and Coatings Technology, 2010. **204**(21): p. 3636-3640.
- 239. Cheng, Z., J. Lian, Y. Hui, and G. Li, *Biocompatible DCPD Coating Formed on AZ91D Magnesium Alloy by Chemical Deposition and Its Corrosion Behaviors in SBF.* Journal of Bionic Engineering, 2014. **11**(4): p. 610-619.
- 240. Rojaee, R., M. Fathi, and K. Raeissi, *Controlling the degradation rate of AZ91 magnesium alloy via sol–gel derived nanostructured hydroxyapatite coating.* Materials Science and Engineering: C, 2013. **33**(7): p. 3817-3825.
- 241. Wan, P., X. Lin, L. Tan, L. Li, W. Li, and K. Yang, *Influence of albumin and inorganic ions on electrochemical corrosion behavior of plasma electrolytic oxidation coated magnesium for surgical implants.* Applied Surface Science, 2013. **282**(0): p. 186-194.
- 242. Friedl, J., C.M. Bauer, A. Rinaldi, and U. Stimming, *Electron transfer kinetics of the VO2+/VO2+ – Reaction on multi-walled carbon nanotubes.* Carbon, 2013. **63**: p. 228-239.
- 243. Ishizaki, T., J. Hieda, N. Saito, N. Saito, and O. Takai, *Corrosion resistance and chemical stability of super-hydrophobic film deposited on magnesium alloy AZ31 by microwave plasma-enhanced chemical vapor deposition.* Electrochimica Acta, 2010. **55**(23): p. 7094-7101.
- 244. Woodrow, J., H. Chilton, and R.I. Hawes, *Forces Between Slurry Particles Due To Surface Tension.* Journal of Nuclear Energy. Part B. Reactor Technology, 1961. **1**(4): p. 229-IN3.
- 245. Xie, J., X. Mao, X. Li, B. Jiang, and L. Zhang, *Influence of moisture absorption on the synthesis and properties of Y2O3–MgO nanocomposites.* Ceramics International, 2017. **43**(1, Part A): p. 40-44.
- 246. Unluer, C. and A. Al-Tabbaa, *The role of brucite, ground granulated blastfurnace slag, and magnesium silicates in the carbonation and performance of MgO cements.* Construction and Building Materials, 2015. **94**: p. 629-643.
- 247. Kurapova, O.Y., V. Konakov, S. Golubev, V. Ushakov, and I.Y. Archakov, *Cryochemical methods for manufacturing nanosized ceramics and ceramic precursor powders with low agglomeration degree: a review.* Rev. Adv. Mater. Sci, 2012. **32**: p. 112-132.
- 248. Sarkar, P., D. De, T. Uchikochi, and L. Besra, *Electrophoretic Deposition (EPD): Fundamentals and Novel Applications in Fabrication of Advanced Ceramic Microstructures*, in *Electrophoretic Deposition of Nanomaterials*, J.H. Dickerson and A.R. Boccaccini, Editors. 2012, Springer New York: New York, NY. p. 181-215.
- 249. Negishi, H., K. Yamaji, T. Imura, D. Kitamoto, T. Ikegami, and H. Yanagishita, *Electrophoretic deposition mechanism of YSZ/n-propanol suspension.* Journal of The Electrochemical Society, 2004. **152**(2): p. J16.
- 250. MA, N., J. HUANG, J. SU, and L. YIN, *Effects of MgO Nanoparticles on Corrosion and Wear Behavior of Micro-arc Oxide Coatings Formed on AZ31B Magnesium Alloy.* Materials Review, 2018. **2018**(16): p. 12.
- 251. Li, C.-Y., X.-L. Fan, R.-C. Zeng, L.-Y. Cui, S.-Q. Li, F. Zhang, Q.-K. He, M.B. Kannan, H.-W. Jiang, D.-C. Chen, and S.-K. Guan, *Corrosion resistance of in-*

situ growth of nano-sized Mg(OH)2 on micro-arc oxidized magnesium alloy AZ31—Influence of EDTA. Journal of Materials Science & Technology, 2019. **35**(6): p. 1088-1098.

- 252. Wang, S., L. Fu, Z. Nai, J. Liang, and B. Cao, *Comparison of corrosion resistance and cytocompatibility of MgO and ZrO2 coatings on AZ31 magnesium alloy formed via plasma electrolytic oxidation.* Coatings, 2018. **8**(12): p. 441.
- 253. Xin, Y.C. and P.K. Chu, *11 Plasma immersion ion implantation (PIII) of light alloys*, in *Surface Engineering of Light Alloys*, H. Dong, Editor. 2010, Woodhead Publishing. p. 362-397.
- 254. Ryu, S., Y.J. Kwon, Y. Kim, and J.U. Lee, *Corrosion protection coating of three-dimensional metal structure by electrophoretic deposition of graphene oxide.* Materials Chemistry and Physics, 2020. **250**: p. 123039.
- 255. Shukla, P., *7 Thermodynamics of corrosion and potentiometric methods for measuring localized corrosion*, in *Techniques for Corrosion Monitoring*, L. Yang, Editor. 2008, Woodhead Publishing. p. 156-186.
- 256. Wang, L., T. Wu, D. Cao, J. Yin, G. Wang, and M. Zhang, *Self-growth of micro- and nano-structured Mg(OH)2 on electrochemically anodised Mg–Li alloy surface.* Journal of Experimental Nanoscience, 2015. **10**(1): p. 56-65.
- 257. Das, P.S., A. Dey, A.K. Mandal, N. Dey, and A.K. Mukhopadhyay, *Synthesis of Mg(OH)2 micro/nano flowers at room temperature.* Journal of Advanced Ceramics, 2013. **2**(2): p. 173-179.
- 258. Khalajabadi, S.Z., M.R. Abdul Kadir, S. Izman, and M. Marvibaigi, *The effect of MgO on the biodegradation, physical properties and biocompatibility of a Mg/HA/MgO nanocomposite manufactured by powder metallurgy method.* Journal of Alloys and Compounds, 2016. **655**: p. 266-280.
- 259. Lei, T., O. Chun, W. Tang, L.-F. Li, and L.-S. Zhou, *Preparation of MgO coatings on magnesium alloys for corrosion protection.* Surface and Coatings Technology, 2010. **204**: p. 3798-3803.
- 260. Feng, J., Y. Chen, X. Liu, T. Liu, L. Zou, Y. Wang, Y. Ren, Z. Fan, Y. Lv, and M. Zhang, *In-situ hydrothermal crystallization Mg(OH)2 films on magnesium alloy AZ91 and their corrosion resistance properties.* Materials Chemistry and Physics, 2013. **143**(1): p. 322-329.
- 261. Ishizaki, T., N. Kamiyama, K. Watanabe, and A. Serizawa, *Corrosion resistance of Mg(OH)2/Mg–Al layered double hydroxide composite film formed directly on combustion-resistant magnesium alloy AMCa602 by steam coating.* Corrosion Science, 2015. **92**: p. 76-84.
- 262. Roy, M., V.K. Balla, A. Bandyopadhyay, and S. Bose, *MgO-doped tantalum coating on Ti: microstructural study and biocompatibility evaluation.* ACS applied materials & interfaces, 2012. **4**(2): p. 577-580.
- 263. Khalajabadi, S.Z., A.B.H. Abu, N. Ahmad, M.R.A. Kadir, A.F. Ismail, R. Nasiri, W. Haider, and N.B.H. Redzuan, *Biodegradable Mg/HA/TiO2 nanocomposites coated with MgO and Si/MgO for orthopedic applications: A study on the corrosion, surface characterization, and biocompatability.* Coatings, 2017. **7**(10): p. 154.
- 264. Fayomi, O.S.I., O.O. Joseph, M.P. Mubiayi, B.M. Durodola, and O. Gabriel, *Microstructural evolution and characterization of super-induced MgO composite on zinc-rich coatings.* Egyptian Journal of Basic and Applied Sciences, 2016. **3**(1): p. 1-9.
- 265. Tan, L.-l., Q. Wang, F. Geng, X.-s. Xi, J.-h. Qiu, and K. Yang, *Preparation and characterization of Ca-P coating on AZ31 magnesium alloy.* Transactions of Nonferrous Metals Society of China, 2010. **20, Supplement 2**(0): p. s648 s654.
- 266. Bai, Y., M.P. Neupane, I.S. Park, M.H. Lee, T.S. Bae, F. Watari, and M. Uo, *Electrophoretic deposition of carbon nanotubes–hydroxyapatite nanocomposites on titanium substrate.* Materials Science and Engineering: C, 2010. **30**(7): p. 1043-1049.
- 267. Shih, T.-S., J.-B. Liu, and P.-S. Wei, *Oxide films on magnesium and magnesium alloys.* Materials Chemistry and Physics, 2007. **104**(2–3): p. 497- 504.
- 268. Kunjukunju, S., A. Roy, M. Ramanathan, B. Lee, J.E. Candiello, and P.N. Kumta, *A layer-by-layer approach to natural polymer-derived bioactive coatings on magnesium alloys.* Acta Biomaterialia, 2013. **9**(10): p. 8690-8703.
- 269. Wei, F., F. Ye, S. Li, L. Wang, J. Li, and G. Zhao, *Layer-by-layer coating of chitosan/pectin effectively improves the hydration capacity, water suspendability and tofu gel compatibility of okara powder.* Food Hydrocolloids, 2018. **77**: p. 465-473.

LIST OF PUBLICATIONS

- 1. C.Y. Chong, Tuty Asma Abu Bakar, Nor Akmal Fadil, Rafaqat Hussain, Electrophoretic Deposition of Hydroxyapatite Coatings on AZ31: The Effect of Nanoparticles Multiple Coating Approach, Advanced Materials Research, Vol. 1125, pp 484-488, 2015
- 2. N.H.Mohamad, T.Abubakar, N.H.Ahmad, C.Y.Chong, Effect of Sintering Temperature on Adhesion and Corrosion Performance of Eggshell Coated onto Mild Steel, 25th Scientific Conference of the Microscopy Society Malaysia, Dec, 2016, Conference Proceeding.
- 3. N. Mohamad Jafar, T.Abubakar, C.Y.Chong, N.H.Ahmad, Effect of Sintering Temperature on the Morphology and Adhesion Strength of Eggshell Coating on Mild Steel, Solid State Phenomena, Vol. 264, pp 190-193, 2017
- 4. Mohd Islahuddin Aziz, Hamidreza Ghandvar, Tuty Asma Abu Bakar, Chong Chin Yee, Effect of praseodymium addition on wear properties of Al-15%Mg2Si composite, SIE 2019 (published online in Materials Today: Proceeding), 2020
- 5. Chong Chin Yee, Tuty Asma Abu Bakar, Nur Akmal Fadil, Hamidreza Ghandvar, Effect of Applied Voltage on Electrophoretic Deposition (EPD) of Nano-Hydroxyapatite (HA) coating on Biodegradable AZ31 Magnesium Alloy and its Corrosion Properties, International Journal of Automotive and Mechanical Engineering (IJAME) on the process of submission