CORROSION PREVENTION THROUGH PERFORMANCE OPTIMIZATION OF BIOLOGICAL CATHODIC PROTECTION SYSTEM

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DEDICATION

Specially dedicated to my beloved parent

(Azman Bin Ali and Che Ramlah Binti Yusof)

who gave me inspiration, encouragement, motivation and valuable support towards
the success of this study

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ABSTRACT

Unprotected pipelines, whether exposed to the air, submerged in seawater or buried underground are susceptible to corrosion and this phenomenon has been identified as one of the major mechanism that affects failures in oil and gas pipelines. At present, impressed current cathodic protection (ICCP) system is one of the methods used to control the corrosion of the steel pipeline. In this study, a new system was developed based on ICCP namely biological cathodic protection (CP) system. This system used the concept of microbial fuel cells (MFCs) to generate a protective current to the cathode (pipe). Little to know on the effect of both hydraulic retention time (HRT) and electrode surface area to anode-compartment volume ratio (ESAVR) simultaneously on the voltage output particularly for biological CP system. Therefore, this study analysed the optimum value of HRT and ESAVR of the biological CP system in generating adequate protective current for the carbon steel pipe with respect to copper/copper sulfate (Cu/CuSO₄) reference electrode which is -850 mV. Optimization of these two parameters on the biological CP system was done using response surface methodology based on central composite design (CCD) approach. CCD predicted values and experimental results showed a strong agreement, with an R² value of 0.7953. The estimated optimal conditions for HRT and ESAVR of anode were at 9.21 h and 0.239 cm²/cm³ respectively. Consequently, the polarization study was conduted to determine the main limiting factors that occurred in the system. It was found that the main losses that occurred was due to the concentration losses whilst the maximum power point recorded was 1.1908 mW/cm² (Run 11). The methodology supporting the purpose of this study by providing protective current to carbon steel pipe and thus verified the ability of this system to protect the pipe from corrosion.

ABSTRAK

Saluran paip yang tidak dilindungi, sama ada terdedah kepada udara, tenggelam di dalam laut atau tertanam di dalam tanah adalah terdedah terhadap kakisan dan fenomena ini telah dikenal pasti sebagai salah satu mekanisme utama yang menyebabkan kegagalan dalam saluran paip minyak dan gas. Pada masa ini, sistem perlindungan katod arus bekasan (ICCP) adalah salah satu kaedah yang digunakan untuk mengawal kakisan pada saluran paip keluli. Dalam kajian ini, satu sistem baharu dicipta berdasarkan ICCP iaitu sistem perlindungan katod secara biologi (CP). Sistem ini menggunakan konsep sel bahan api mikrob untuk menghasilkan arus pelindung ke katod (paip). Secara amnya, pengetahuan mengenai kesan kedua-dua nilai masa pengekalan hidraulik (HRT) dan luas permukaan elektrod kepada nisbah isipadu anod (ESAVR) secara serentak pada pengeluaran tenaga terutamanya untuk sistem CP biologi adalah terhad. Oleh itu, kajian ini menganalisis nilai optimum HRT dan ESAVR dalam menghasilkan arus pelindung yang cukup bagi paip keluli karbon berdasarkan elektrod rujukan tembaga / tembaga sulfat (Cu / CuSO₄) iaitu -850 mV. Pengoptimuman kedua-dua parameter sistem ini dilakukan dengan menggunakan kaedah sambutan permukaan berdasarkan pendekatan reka bentuk komposit pusat (CCD). Nilai ramalan CCD dan hasil eksperimen menunjukkan kolerasi yang kuat, dengan nilai R² bersamaan 0.7953. Anggaran optimum bagi HRT dan ESAVR anod masing-masing pada 9.21 jam dan 0.239 cm²/cm³. Seterusnya, kajian polarasi dilakukan untuk menentukan factor pembatas utama yang berlaku dalam sistem. Kehilangan utama yang berlaku adalah dari kehilangan kepekatan manakala titik maksima kuasa yang direkodkan adalah 1.1908 mW/cm² (Ujian 11). Metodologi menyokong tujuan kajian ini untuk menyediakan arus pelindung kepada paip keluli karbon dan dengan demikian mengesahkan kemampuan sistem ini untuk melindungi paip daripada kakisan.

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LIST OF ABBREVIATION S

CCD - Central Composite Design

CCI - Central Composite Inscribed

CE - Coulombic Efficiency

COD - Chemical Oxygen Demand

CP - Cathodic Protection

CR - Corrosion Rate

ESAVR - Electrode Surface Area / Anode-Compartment Volume Ratio

HRT - Hydraulic Retention Time

ICCP - Impressed Current Cathodic Protection

MLSS - Mixed Liquor Suspended Solid

MPP Maximum Power Point

OLR - Organic Loading Rates

RSM - Response Surface Methodology

LIST OF SYMBOLS

C₂H₃NaO₂ - Sodium acetate

CuSO₄ - Copper sulphate

h - hours

HCl - Hydrochloric acid

HNO₃ - Nitric acid

KCl - Potasium chloride

NaCl - Sodium chloride

R_i - Internal resistance

Sb₂O₃ - Antimony trioxide

SnCl₂ - Stannous chloride

 Ω - Ohms

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CHAPTER 1

INTRODUCTION

1.1 Research Background

According to National Association of Corrosion Engineers (NACE) International, corrosion is defined as the deterioration of a substance or its properties because of a reaction with its environment (Norsworthy, 2014). Popoola *et al.* (2013) stated that pipelines corrosion has been identified as one of the major mechanism that gives rise to failures in oil and gas pipelines. The consequences of pipelines corrosion leakages will result in the environmental impact, contamination, and shutdowns of the plant (Ngobiri *et al.*, 2015). The problem of corrosion is worth to be investigated because every year, this problem contributes to a high percentage of total costs for oil and gas producing companies worldwide (Finšgar and Jackson, 2014). In placing more emphasis, Ossai (2012) stated that, from 1980 until 2006, 50% of the European major hazards of loss arising from technical plants failures are due to ageing of plants mechanism caused by corrosion, erosion and fatigue. As the consequences, pipelines corrosion leakages can result in the environmental impact, contamination, and shutdowns of the plant. Besides that, contamination of products with residual rust leads to low quality production and finally loss of business reputation.

According to Paul *et al.* (2015), carbon steels are the most common materials used in the transportation of oil and gas from one region to another due to its cost-effectiveness and applications. However, the corrosion of carbon steel pipe still occurred due to contact with the environment. All metallic structures will tend to corrode when in contact with water, concrete, soil and moist air. Coatings in pipelines usually are not perfect and may have defects such as scratch of the coatings, pinholes, and flaws. Chemical species, groundwater, and corrosive gases may pass through the defects and reach the pipe surface. This phenomenon will cause coating disbonding and eventually pipeline corrosion at the defected area.

Therefore, to overcome this problem, all buried gas pipelines applied the concept of cathodic protection (CP) system (Ashworth, 2010; Brenna et al., 2017; Kuang and Cheng, 2015). This system reduces the corrosion of the metal surface by constructing an electrochemical cell and making the surface of metal as cathode. There are two types of CP systems which are sacrificial anode CP system and impressed current cathodic protection (ICCP) system. The concept of sacrificial anode is the driving voltage is provided by the natural potential difference that occurs between the structure and a second metal in the surrounding without the need of power source. In contrast to the concept of sacrificial anode, in ICCP method, the driving voltage for protective current is supplied from a power source. The power supply is usually consist of a rectifier that converts alternating current (AC) to direct current (DC). In order to ensure adequate protection for any structure, the polarized potential of the protected structure should be measured and compared to a certain criterion. For example, the criteria for carbon steel pipe is -850mV with respect to copper/copper sulfate (Cu/CuSO₄) reference electrode and the cathodic protection criterion differs according to the type of metals and the soil resistivity. Despite the fact that ICCP system can be applied to a wide range of structures, it can affect other structures if it is not properly monitored (A.S. Hamdy Makhlouf and M. Aliofkhazrael, 2002). Other than that, installation of ICCP system required an external power source and have a high maintenance cost, including the mechanical maintenance of power supply and equipment (Byrne et al., 2016).

Therefore, this study introduced a new approach by using the concept of microbial fuel cells (MFCs) to represent a new protection method namely biological CP system. MFCs are a sustainable energy technology which microorganisms present in the wastewater work as a catalyst to convert chemical energy into electrical (Oliveira *et al.*, 2013; Ortiz-Martínez *et al.*, 2015; Zhou *et al.*, 2011). Electricity is generated by MFCs as the microbes in the anode compartment consume substrates and thus generating electron and protons. The transfer of electrons from anode (negative terminal) to the cathode (positive terminal) through the external circuit will produce electricity. Thus, the capability of the microorganism to generate electricity can be used as a new approach for biological CP to protect the pipelines.

However, the electricity production of the system undeniably depends on the growth of microorganism. According to Al-Qadiri *et al.* (2008), the growth of microorganisms can be divided into four phases which are lag phase, exponential phase, stationary phase and death phase. During the lag phase, the microbes will adapt themselves to the growth environments. The next phase is followed by exponential growth where the microbes will grow rapidly and divide at a maximal rate possible. The next phase is the stationary phase where the growth rate of the microbes is limited due to the limiting factors such as the depletion of nutrients. Throughout this phase, the growth rate of the microbes also equals the death rate. Lastly, the population of microbes will decline in population during the death phase. Therefore, to maintain the microbial population in the wastewater, this study was constructed in continuous-mode as the microorganisms were continuously fed and removed from the system.

In the last few years, MFCs technology has been improved significantly. There are many factors that can influence the performance of this system. These measures may include different types of electrode (Zhou *et al.*, 2011), different substrate used in the system (Pandey *et al.*, 2016; Pant *et al.*, 2010), various types of separator materials used (Kim *et al.*, 2007; Li *et al.*, 2011), as well as different MFCs design and configurations (Das and Mangwani, 2010; Oliveira *et al.*, 2013; Walter *et al.*, 2016). Different design and configurations of MFCs are subject to some important considerations such as hydraulic retention time (HRT) and electrode surface area to anode-compartment volume ratio (ESAVR) of the system.

According to Akman *et al.* (2013), hydraulic retention time (HRT) is an important parameter as it can affect the design, investment cost and energy requirement. Generally, high capital cost is required for a system with higher HRT and varying HRT also has a major effect on the power generation of MFC. Regarding the ESAVR, it has been reported that large-volume units of MFCs tend to have lower power densities and hence decreased efficiency (Walter *et al.*, 2016). A study by Ieropoulos *et al.* (2008) also proved that a smaller MFCs have a greater power density compared to large vessels because of their higher surface area to volume (SA:V) ratios.

Therefore, by utilizing the capabilities of microorganisms to generate electricity as well as the energy recovery benefits from MFCs system, biological CP offers an attractive solution in protecting the corrosion of gas pipelines. Thus, to obtain the right configurations of the system, this research investigated the optimum HRT and ESAVR by using RSM to achieve higher power density output of biological CP system. Besides, the corrosion rate of the protected pipe and reference pipe were analysed to measure the abilities of the protective current provided by the system to protect the system.

1.2 Problem Statement

One of the primary problems that resulting in failures of oil and gas pipelines is the phenomenon of corrosion at the underground steel pipeline. This problem resulted from an electrochemical reaction that takes place around the pipeline (Kuang and Cheng, 2015). CP is the most important method of all approaches to control the corrosion. CP can be achieved in two methods which are sacrificial anode CP system and ICCP system (Ashworth, 2010). However, installation of sacrificial anode has its own drawbacks such as unable to control the current and shorter lifespan. Besides that, ICCP method also has its own disadvantages. ICCP method required the installation of external power and monitoring systems and these systems are vulnerable to damage and atmospheric corrosion. This system also is prone to failure of switches, fuses and therefore it has a high maintenance cost (Byrne *et al.*, 2016).

Recently, great attention have been paid to MFCs due to their ability to produce electricity directly from the degradation of organic matters (Rahimnejad *et al.*, 2015). However, there has been no previous work describing the utilization of MFCs to generate the protective current in CP system. In this study, a new approach was developed to protect the gas pipeline namely biological CP system by utilizing the concept of MFC to generate the electricity. Still, the power generation of the MFC system fluctuated and varied depending on the configurations of the system including the HRT and ESAVR.

It was reported that high HRT of the system may not provide enough substrate for the microorganism to sustain longer while at low HRT, the biofilm may not build completely thus resulting in low power generation (Haavisto *et al.*, 2017). Besides that, recent studies have demonstrated that increasing ESAVR improved the power generation of the system (Penteado *et al.*, 2018; Walter *et al.*, 2016). Nevertheless, previous study that considers both HRT and ESAVR simultaneously is less reported particularly for biological CP system. Therefore, the right HRT and ESAVR of the system need to be studied to achieve optimal voltage output for cathodic protection method. Therefore, to study these two parameters, the experimental run was designed by using the central composite design by using Design Expert software to determine the optimal condition for high voltage output. Besides, the main limiting factors of the system also was determined by polarization study.

1.3 Research Objectives

The main objective of this study is to investigate the optimum HRT and ESAVR of the biological CP system to protect carbon steel pipe from corrosion. The sub-objectives of this study are as follows:

- To optimize the hydraulic retention time (HRT) and electrode surface area to anode-compartment volume ratio (ESAVR) of the biological CP system by using RSM via central composite design (CCD)
- 2. To analyse the maximum power point and limiting factors of the system by polarization study

1.4 Research Scopes

In order to achieve the objectives of this research, the following scopes have been identified:

- Preparing and characterizing of wastewater collected from Indah Water Gravity Thickener Pond at Taman Bukit Senang.
- 2. The experimental rig was fabricated and constructed with local materials based on two compartments which were anode and cathode compartment. The anode compartment was constructed with a dimension of 32.0 cm × 20.5 cm × 10.0 cm (length × width × height) and a graphite rod was used as the anode electrode. Carbon steel pipe (schedule 40 pipe dimension) was selected as the cathode electrode and salt bridge was used to connect the anode and cathode compartment. Pumps were used to feed the wastewater and substrate continuously into the experimental rig (anode compartment) through plastic tubing.
- 3. Determination of optimum power density output of the system via Central Composite Design (CCD) in response surface methodology (RSM) based on two independent variables (inlet flowrate of wastewater and length of graphite anode). The inlet flowrate of wastewater was varying between 2.0 10.0 mL/min (HRT 9.11 h -27.33 h) while the length of graphite electrode was between 6.0 cm to 15.0 cm (ESAVR of 1005.31 1570.80 cm²/cm³).

REFERENCES

- A.S. Hamdy Makhlouf, and M. Aliofkhazrael. (2002). *Cathodic protection principles* (Article)o. Document Number)
- Ahmad, Z. (2006). CHAPTER 5 CATHODIC PROTECTION. In Z. Ahmad (Ed.), Principles of Corrosion Engineering and Corrosion Control (pp. 271-351). Oxford: Butterworth-Heinemann.
- Akman, D., Cirik, K., Ozdemir, S., Ozkaya, B., and Cinar, O. (2013). Bioelectricity generation in continuously-fed microbial fuel cell: Effects of anode electrode material and hydraulic retention time. *Bioresource Technology*, *149*, 459-464.
- Al-Qadiri, H. M., Al-Alami, N. I., Lin, M., Al-Holy, M., Cavinato, A. G., and Rasco, B. A. (2008). Studying of the Bacterial Growth Phases using Fourier Transform Infrared Spectroscopy and Multivariate Analysis. *Journal of Rapid Methods & Automation in Microbiology*, 16(1), 73-89.
- Almatouq, A., and Babatunde, A. O. (2018). Identifying optimized conditions for concurrent electricity production and phosphorus recovery in a mediator-less dual chamber microbial fuel cell. *Applied Energy*, 230, 122-134.
- Anderson, M. J., and Whitecomb, P. J. (2005). RSM Simplified: Optimizing Processes

 Using Response Surface Methods for Design of Experiments. New York:

 Producivity Press.
- Ashworth, V. (2010). Principle of Cathodic Protection. 2, 2747-2762.
- Barbato, R., and Gronwald, F. (2018). A Study of Soil Based Microbial Fuel Cells. *1*, 60-157.
- Behera, M., Jana, P. S., More, T. T., and Ghangrekar, M. M. (2010). Rice mill wastewater treatment in microbial fuel cells fabricated using proton exchange membrane and earthen pot at different pH. *Bioelectrochemistry*, 79(2), 228-233.
- Bezerra, M. A., Santelli, R. E., Oliveira, E. P., Villar, L. S., and Escaleira, L. A. (2008). Response surface methodology (RSM) as a tool for optimization in analytical chemistry. *Talanta*, 76(5), 965-977.

- Bose, D., Dhawan, H., Kandpal, V., Vijay, P., and Gopinath, M. (2018). Sustainable power generation from sewage and energy recovery from wastewater with variable resistance using microbial fuel cell. *Enzyme and Microbial Technology*, 118, 92-101.
- Brenna, A., Beretta, S., Uglietti, R., Lazzari, L., Pedeferri, M., and Ormellese, M. (2017). Cathodic protection monitoring of buried carbon steel pipeline: measurement and interpretation of instant-off potential. *Corrosion Engineering, Science and Technology*, 52(4), 253-260.
- Bushman, J. B., and Engineer, P. P. C. (2010). Corrosion and cathodic protection theory. *Bushman & Associates Inc.*, *Medina*.
- Byrne, A., Holmes, N., and Norton, B. (2016). State-of-the-art review of cathodic protection for reinforced concrete structures. *Magazine of Concrete Research*, 68(13), 664-677.
- Caines, S., Khan, F., and Shirokoff, J. (2013). Analysis of pitting corrosion on steel under insulation in marine environments. *Journal of Loss Prevention in the Process Industries*, 26(6), 1466-1483.
- Capodaglio, A., Molognoni, D., Dallago, E., Liberale, A., Cella, R., Longoni, P., et al. (2013). Microbial Fuel Cells for Direct Electrical Energy Recovery from Urban Wastewaters. *The Scientific World Journal*, 2013, 634738.
- Chae, K.-J., Choi, M.-J., Lee, J.-W., Kim, K.-Y., and Kim, I. S. (2009). Effect of different substrates on the performance, bacterial diversity, and bacterial viability in microbial fuel cells. *Bioresource Technology*, 100(14), 3518-3525.
- Chang, I. S., Jang, J. K., Gil, G. C., Kim, M., Kim, H. J., Cho, B. W., et al. (2004). Continuous determination of biochemical oxygen demand using microbial fuel cell type biosensor. *Biosensors and Bioelectronics*, 19(6), 607-613.
- Chen, X., and Zhao, Y. (2017). Research on Corrosion Protection of Buried Steel Pipeline. *Engineering*, 09, 504-509.
- Cheng, S., Kiely, P., and Logan, B. E. (2011). Pre-acclimation of a wastewater inoculum to cellulose in an aqueous–cathode MEC improves power generation in air–cathode MFCs. *Bioresource Technology*, *102*(1), 367-371.
- Cheng, S., and Logan, B. E. (2011). Increasing power generation for scaling up single-chamber air cathode microbial fuel cells. *Bioresource Technology*, 102(6), 4468-4473.

- Das, S., and Mangwani, N. (2010). Recent developments in microbial fuel cells: A review (Vol. 69).
- Daud, S. M., Kim, B. H., Ghasemi, M., and Daud, W. R. W. (2015). Separators used in microbial electrochemical technologies: Current status and future prospects. *Bioresource Technology*, 195, 170-179.
- David, E. J., and James, D. R. (2018). *Corrosion Science and Technology*: Boca Raton:CRC Press.
- Dewan, A., Beyenal, H., and Lewandowski, Z. (2008). *Scaling up Microbial Fuel Cells* (Vol. 42).
- Drani, A., Alam, M., and Muyibi, S. (2011). Screening of Potential Strains and Cosubstrate for Production of Cellulase Enzyme using Sewage Treatment Plant Sludge as Major Substrate. *Biotechnology(Faisalabad)*, 10, 197-202.
- Evitts, R. W. (2012). 12 Cathodic Protection. In M. Kutz (Ed.), *Handbook of Environmental Degradation of Materials (Second Edition)* (pp. 359-380). Oxford: William Andrew Publishing.
- Fajardo, V., Canto, C., Brown, B., and Nesic, S. (2007). *Effect Of Organic Acids In CO2 Corrosion*. Paper presented at the CORROSION 2007.
- Fazli, N., Mutamim, N. S. A., Jafri, N. M. A., and Ramli, N. A. M. (2018). Microbial Fuel Cell (MFC) in treating spent caustic wastewater: Varies in Hydraulic Retention Time (HRT) and Mixed Liquor Suspended Solid (MLSS). *Journal of Environmental Chemical Engineering*, 6(4), 4339-4346.
- Feng, Y., Cai, Z., Li, H., Du, Z., and Liu, X. (2013). Response surface optimization of fluidized roasting reduction of low-grade pyrolusite coupling with pretreatment of stone coal (Vol. 49).
- Finšgar, M., and Jackson, J. (2014). Application of corrosion inhibitors for steels in acidic media for the oil and gas industry: A review. *Corrosion Science*, 86, 17-41.
- Foad marashi, S. k., and Kariminia, H.-R. (2015). Performance of a single chamber microbial fuel cell at different organic loads and pH values using purified terephthalic acid wastewater. *Journal of Environmental Health Science and Engineering*, 13.
- Ge, Z., Ping, Q., Xiao, L., and He, Z. (2013). Reducing effluent discharge and recovering bioenergy in an osmotic microbial fuel cell treating domestic wastewater. *Desalination*, 312, 52-59.

- Ghangrekar, M. M., and Shinde, V. B. (2007). Performance of membrane-less microbial fuel cell treating wastewater and effect of electrode distance and area on electricity production. *Bioresource Technology*, *98*(15), 2879-2885.
- Ghangrekar, M. M., and Shinde, V. B. (2008). Simultaneous sewage treatment and electricity generation in membrane-less microbial fuel cell. *Water Science and Technology*, 58(1), 37-43.
- Guerrini, E., Cristiani, P., Pierpaolo, S., and Trasatti, M. (2013). *Relation of anodic* and cathodic performance to pH variations in membraneless microbial fuel cells (Vol. 38).
- Haavisto, J. M., Kokko, M. E., Lay, C.-H., and Puhakka, J. A. (2017). Effect of hydraulic retention time on continuous electricity production from xylose in up-flow microbial fuel cell. *International Journal of Hydrogen Energy*, 42(45), 27494-27501.
- Ieropoulos, I., Greenman, J., and Melhuish, C. (2008). Microbial fuel cells based on carbon veil electrodes: Stack configuration and scalability. *International Journal of Energy Research*, 32(13), 1228-1240.
- Ishii, S. i., Suzuki, S., Norden-Krichmar, T. M., Wu, A., Yamanaka, Y., Nealson, K. H., et al. (2013). Identifying the microbial communities and operational conditions for optimized wastewater treatment in microbial fuel cells. *Water Research*, 47(19), 7120-7130.
- Jiang, H.-m., Luo, S.-j., Shi, X.-s., Dai, M., and Guo, R.-b. (2013). A system combining microbial fuel cell with photobioreactor for continuous domestic wastewater treatment and bioelectricity generation. *Journal of Central South University*, 20(2), 488-494.
- Khan, M., Kakar, F., Khan, S., and Athar, S. (2018). Efficiency and Cost Analysis of Power Sources in Impressed Current Cathodic Protection System for Corrosion Prevention in Buried Pipelines of Balochistan, Pakistan. *IOP Conference Series: Materials Science and Engineering*, 414, 012034.
- Khattak, M. A., Zareen, N., Mukhtar, A., Kazi, S., Jalil, A., Ahmed, Z., et al. (2016). Root cause analysis (RCA) of fractured ASTM A53 carbon steel pipe at oil & Engineering Failure Analysis, 7, 1-8.

- Kiely, P. D., Rader, G., Regan, J. M., and Logan, B. E. (2011). Long-term cathode performance and the microbial communities that develop in microbial fuel cells fed different fermentation endproducts. *Bioresource Technology*, 102(1), 361-366.
- Kim, B. H., Chang, I. S., and Gadd, G. M. (2007). Challenges in microbial fuel cell development and operation. *Applied Microbiology and Biotechnology*, 76(3), 485.
- Kuang, D., and Cheng, Y. F. (2015). Effect of alternating current interference on coating disbondment and cathodic protection shielding on pipelines. *Corrosion Engineering, Science and Technology*, 50(3), 211-217.
- Kumaran, P., Hephzibah, D., Sivasankari, R., Saifuddin, N., and Shamsuddin, A. H. (2016). A review on industrial scale anaerobic digestion systems deployment in Malaysia: Opportunities and challenges. *Renewable and Sustainable Energy Reviews*, 56, 929-940.
- Lee, H.-S., Parameswaran, P., Kato-Marcus, A., Torres, C. I., and Rittmann, B. E. (2008). Evaluation of energy-conversion efficiencies in microbial fuel cells (MFCs) utilizing fermentable and non-fermentable substrates. *Water Research*, 42(6), 1501-1510.
- Li, W.-W., Sheng, G.-P., Liu, X.-W., and Yu, H.-Q. (2011). Recent advances in the separators for microbial fuel cells. *Bioresource Technology*, 102(1), 244-252.
- Li, Y., Wu, Y., Puranik, S., Lei, Y., Vadas, T., and Li, B. (2014). Metals as electron acceptors in single-chamber microbial fuel cells. *Journal of Power Sources*, 269(Supplement C), 430-439.
- Liu, H., Cheng, S., and Logan, B. E. (2005). Production of Electricity from Acetate or Butyrate Using a Single-Chamber Microbial Fuel Cell. *Environmental Science* & Technology, 39(2), 658-662.
- Liu, H., Ramnarayanan, R., and Logan, B. E. (2004). Production of Electricity during Wastewater Treatment Using a Single Chamber Microbial Fuel Cell. Environmental Science & Technology, 38(7), 2281-2285.
- Liu, J., Wang, J., Leung, C., and Gao, F. (2018). A Multi-Parameter Optimization Model for the Evaluation of Shale Gas Recovery Enhancement. *Energies*, 11(3), 654.

- Logan, B. E., Hamelers, B., Rozendal, R., Schröder, U., Keller, J., Freguia, S., et al. (2006). Microbial fuel cells: methodology and technology. *Environmental science & technology*, 40(17), 5181-5192.
- Ma, D., Jiang, Z.-H., Lay, C.-H., and Zhou, D. (2016). Electricity generation from swine wastewater in microbial fuel cell: Hydraulic reaction time effect. *International Journal of Hydrogen Energy*, 41(46), 21820-21826.
- Mäkelä, M. (2017). Experimental design and response surface methodology in energy applications: A tutorial review. *Energy Conversion and Management*, 151, 630-640.
- Makhlouf, A. S. H., Herrera, V., and Muñoz, E. (2018). Chapter 6 Corrosion and protection of the metallic structures in the petroleum industry due to corrosion and the techniques for protection. In A. S. H. Makhlouf and M. Aliofkhazraei (Eds.), *Handbook of Materials Failure Analysis* (pp. 107-122): Butterworth-Heinemann.
- Min, B., Cheng, S., and Logan, B. E. (2005). Electricity generation using membrane and salt bridge microbial fuel cells. *Water Res*, 39(9), 1675-1686.
- Nalli, K. (2010). Corrosion and its mitigation in the oil & gas industry-an overview.
- Ngobiri, N. C., Oguzie, E. E., Oforka, N. C., and Akaranta, O. (2015). Comparative study on the inhibitive effect of Sulfadoxine–Pyrimethamine and an industrial inhibitor on the corrosion of pipeline steel in petroleum pipeline water. *Arabian Journal of Chemistry*.
- Nor, M. H. M., Mubarak, M. F. M., Elmi, H. S. A., Ibrahim, N., Wahab, M. F. A., and Ibrahim, Z. (2015). Bioelectricity generation in microbial fuel cell using natural microflora and isolated pure culture bacteria from anaerobic palm oil mill effluent sludge. *Bioresource Technology*, 190, 458-465.
- Norsworthy, R. (2014). 1 Understanding corrosion in underground pipelines: basic principles A2 Orazem, Mark E. In *Underground Pipeline Corrosion* (pp. 3-34): Woodhead Publishing.
- Oliveira, V. B., Simões, M., Melo, L. F., and Pinto, A. M. F. R. (2013). Overview on the developments of microbial fuel cells. *Biochemical Engineering Journal*, 73, 53-64.
- Ortiz-Martínez, V. M., Salar-García, M. J., de los Ríos, A. P., Hernández-Fernández, F. J., Egea, J. A., and Lozano, L. J. (2015). Developments in microbial fuel cell modeling. *Chemical Engineering Journal*, *271*(Supplement C), 50-60.

- Ossai, C. I. (2012). Advances in Asset Management Techniques: An Overview of Corrosion Mechanisms and Mitigation Strategies for Oil and Gas Pipelines. *ISRN Corrosion*, 2012, 10.
- Pandey, P., Shinde, V. N., Deopurkar, R. L., Kale, S. P., Patil, S. A., and Pant, D. (2016). Recent advances in the use of different substrates in microbial fuel cells toward wastewater treatment and simultaneous energy recovery. *Applied Energy*, 168, 706-723.
- Pant, D., Van Bogaert, G., Diels, L., and Vanbroekhoven, K. (2010). A review of the substrates used in microbial fuel cells (MFCs) for sustainable energy production. *Bioresource Technology*, 101(6), 1533-1543.
- Paul, O., Shakoor, A., and Mohamed, A. M. A. (2015). Environmental Factors Affecting Corrosion of Pipeline Steel: A Review. *International Journal of Mechanical and Production Engineering Research and Development*, 5(5), 14.
- Penteado, E. D., Fernandez-Marchante, C. M., Zaiat, M., Gonzalez, E. R., and Rodrigo, M. A. (2018). Optimization of the performance of a microbial fuel cell using the ratio electrode-surface area / anode-compartment volume. Brazilian Journal of Chemical Engineering, 35, 141-146.
- Picioreanu, C., Head, I. M., Katuri, K. P., van Loosdrecht, M. C. M., and Scott, K. (2007). A computational model for biofilm-based microbial fuel cells. *Water Research*, *41*(13), 2921-2940.
- Pierre, R. (2012). *Handbook of Corrosion Engineering 2/E*: McGraw-Hill Professional.
- Popoola, L. T., Grema, A. S., Latinwo, G. K., Gutti, B., and Balogun, A. S. (2013). Corrosion problems during oil and gas production and its mitigation. *International Journal of Industrial Chemistry*, 4(1), 35.
- Puthalath, R., Surendranathan, A. O., and Murthy, C. S. N. (2012). Corrosion mitigation of the oil well steels using organic inhibitors-A review. *Journal of Materials and Environmental Science*, *3*, 856-869.
- Qiao, G., Guo, B., Ou, J., Xu, F., and Li, Z. (2016). Numerical optimization of an impressed current cathodic protection system for reinforced concrete structures. *Construction and Building Materials*, 119(Supplement C), 260-267.
- Rahimnejad, M., Adhami, A., Darvari, S., Zirepour, A., and Oh, S.-E. (2015). Microbial fuel cell as new technology for bioelectricity generation: A review. *Alexandria Engineering Journal*, *54*(3), 745-756.

- Rahimnejad, M., Ghoreyshi, A. A., Najafpour, G., and Jafary, T. (2011). Power generation from organic substrate in batch and continuous flow microbial fuel cell operations. *Applied Energy*, 88(11), 3999-4004.
- Singh, R. (2014). Chapter One Need for the Study of Corrosion. In *Corrosion Control* for *Offshore Structures* (pp. 3-6). Boston: Gulf Professional Publishing.
- Speight, J. (2014). Chapter 6 Corrosion Monitoring and Control. In *Oil and Gas Corrosion Prevention* (pp. 109-149). Boston: Gulf Professional Publishing.
- Stefanova, A., Angelov, A., Bratkova, S., Genova, P., and Nikolova, K. (2018).

 INFLUENCE OF ELECTRICAL CONDUCTIVITY AND TEMPERATURE IN

 A MICROBIAL FUEL CELL FOR TREATMENT OF MINING WASTE WATER.
- Tatode, A., Patil, A., and Umekar, M. (2018). Application of response surface methodology in optimization of paclitaxel liposomes prepared by thin film hydration technique. *International Journal of Applied Pharmaceutics*, 10, 62.
- Tsynaeva, A. (2015). Cathode Protection Systems of Cross-country Pipelines. *Procedia Engineering*, 111, 777-782.
- Vanaei, H. R., Eslami, A., and Egbewande, A. (2017). A review on pipeline corrosion, in-line inspection (ILI), and corrosion growth rate models. *International Journal of Pressure Vessels and Piping*, *149*(Supplement C), 43-54.
- Wadikar, D., Nanjappa, C., Premavalli, K., and Bawa, A. (2010). Development of ready-to-eat appetisers based on pepper and their quality evaluation. *Journal of food science and technology*, 47, 638-643.
- Walter, X. A., Forbes, S., Greenman, J., and Ieropoulos, I. A. (2016). From single MFC to cascade configuration: The relationship between size, hydraulic retention time and power density. *Sustainable Energy Technologies and Assessments*, 14, 74-79.
- Wang, M., Yan, Z., Huang, B., zhao, j., and Liu, R. (2013). Electricity Generation by Microbial Fuel Cells Fuelled with Enteromorpha Prolifera Hydrolysis. *International Journal of Electrochemical Science*, 8, 2104-2111.
- Wen, Q., Wu, Y., Zhao, L., and Sun, Q. (2010). Production of electricity from the treatment of continuous brewery wastewater using a microbial fuel cell. *Fuel*, 89(7), 1381-1385.

- Witek-Krowiak, A., Chojnacka, K., Podstawczyk, D., Dawiec, A., and Pokomeda, K. (2014). Application of response surface methodology and artificial neural network methods in modelling and optimization of biosorption process. *Bioresource Technology*, *160*, 150-160.
- Yang, Q., Wang, X., Feng, Y., Lee, H., Liu, J., Shi, X., et al. (2012). Electricity generation using eight amino acids by air—cathode microbial fuel cells. *Fuel*, 102, 478-482.
- Yuan, Y., Zhao, B., Zhou, S., Zhong, S., and Zhuang, L. (2011). Electrocatalytic activity of anodic biofilm responses to pH changes in microbial fuel cells. *Bioresource Technology*, 102(13), 6887-6891.
- Zhao, F., Slade, R., and Varcoe, J. (2009). Techniques for the study and development of microbial fuel cells: An electrochemical perspective. *Chemical Society reviews*, 38, 1926-1939.
- Zhou, M., Chi, M., Luo, J., He, H., and Jin, T. (2011). An overview of electrode materials in microbial fuel cells. *Journal of Power Sources*, 196(10), 4427-4435.
- Zhu, G.-F., Wu, P., Wei, Q., Jianyi, L., Gao, Y.-L., and Liu, H.-N. (2010). Biohydrogen production from purified terephthalic acid (PTA) processing wastewater by anaerobic fermentation using mixed microbial communities. *International Journal of Hydrogen Energy*, 35, 8350-8356.
- Zvirko, O. I., Savula, S. F., Tsependa, V. M., Gabetta, G., and Nykyforchyn, H. M. (2016). Stress corrosion cracking of gas pipeline steels of different strength. *Procedia Structural Integrity*, 2, 509-516.