

CHARACTERISATION AND OPTIMISATION OF FLAME-RETARDING
PALM OIL-BASED POLYURETHANE/MONTMORILLONITE-
AMMONIUM POLYPHOSPHATE FOAM

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DEDICATION

To those who chose to rise up and fight further

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ABSTRACT

The organic chemical composition in polyurethane (PU) foam has adversely affected the flammability of the foam that causes high susceptibility to fire. Many researchers and industrialists have invested vast efforts to counter this issue, including incorporation of synthetically-produced flame retardants and fillers, and chemical modification on the constituting materials of PU foam. All these methods are not environmentally-friendly. Montmorillonite (MMT) has been known to have flame-retarding properties in numerous polymer composite systems, however, its effects in PU foam application has not been investigated. The aim of this project was to develop and optimise palm oil-based PU foam reinforced with ammonium polyphosphate (APP) and MMT in the form of blended (PU/MMT/APP) and hybrid (PU/APP-MMT) systems. Palm oil-based PU foam was chosen as a matrix due to its renewable resources and sustainability. Two types of multi filler systems were prepared: PU/MMT/APP blend and PU/APP-MMT hybrid. The APP-MMT hybrid filler was prepared through an ion-exchanged surface-treatment method. PU foams were then fabricated at different MMT/APP and APP-MMT loadings, and were characterized for their compressive properties, fire retardancies, thermal stabilities, and morphologies. Hybrid PU/APP-MMT system had improved the fire retardancy where the limiting oxygen index (LOI) had reached 24.19 at 10 wt. % filler loading as opposed to 23.85 showed by blended PU/MMT/APP filler system. This probably indicates the synergistic effect between APP and MMT in fire-retarding mechanism where a more stable alumino-phosphate species was formed during the combustion, which enhanced the thermal-insulating and fire protection properties. However, the compressive modulus of hybrid APP-MMT showed the highest value (2.857 MPa) at 6 wt. %, where further filler inclusion beyond this point had deteriorated the strengths of the foams. Hindered intermolecular hydrogen-bonding was thought as the main contributor for the reduction. Response Surface Methodology was then used to find the optimised filler formulation for both multi-filler systems with compressive modulus and LOI as the responses. The optimisation yielded filler combination of 0.9 wt. % Na-MMT and 4.9 wt. % APP was the best for blended system, whereby the optimised hybrid APP-MMT value was computed as 9.2 wt. %. The characterization of both optimised filler formulations portrayed some improvements, when compared to its pristine counterpart. The LOI values were improved up to 44 %, with compromising only 5 % of compressive modulus. From the results obtained, bio-based PU foam has a great potential to be used in structural panel-related applications that requires moderate load-bearing and flame-retardancy capabilities.

ABSTRAK

Komposisi kimia organik dalam busa poliuretena (PU) telah memudaratkan kemudahbakaran busa tersebut dengan menyebabkannya mudah terbakar. Ramai penyelidik dan ahli perindustrian telah menggembung tenaga untuk mengatasi isu ini, termasuk menggunakan bahan cegah kebakaran dan pengisi yang dihasilkan secara sintetik, dan pengubahsuai kimia bahan-bahan penghasilan busa PU. Kesemua kaedah ini adalah tidak mesra alam. Montmorillonit (MMT) telah diketahui mempunyai sifat cegah kebakaran dalam berbagai sistem komposit polimer, namun kesannya terhadap sistem busa PU belum pernah dikaji. Tujuan kajian ini adalah untuk membangun dan mengoptimumkan busa PU berasaskan kelapa sawit bertetulang ammonium polifosfat (APP) dan tanah liat nano MMT, di dalam bentuk campuran (PU/MMT/APP) dan hibrid (PU/APP-MMT). Busa PU berasaskan kelapa sawit telah dipilih sebagai matriks kerana sifat sumber boleh diperbaharui dan kelestariannya. Dua jenis sistem pelbagai pengisi telah disediakan: PU/MMT/APP campuran dan juga PU/APP-MMT hibrid. Pengisi APP-MMT hibrid telah disediakan melalui proses rawatan permukaan pertukaran ion. Busa PU kemudiannya dihasilkan dengan muatan MMT/APP dan APP-MMT yang berbeza, dan dicirikan mengikut sifat mampatan, kecegahbakaran, kestabilan haba, dan morfologi. Sistem hibrid APP-MMT telah meningkatkan kecegahbakaran yang mana nilai Indeks Oksigen Pengehad (LOI) mencapai 24.19 pada 10 wt. % muatan pengisi, berbanding 23.85 menggunakan sistem campuran Na-MMT/APP. Ini mungkin menunjukkan kesan sinergi antara MMT dan APP dalam mekanisme mencegah kebakaran yang mana spesis alumino-fosfat lebih stabil telah terbentuk ketika pembakaran yang meningkatkan sifat penebat haba dan kalis api. Walau bagaimanapun, modulus mampatan PU/APP-MMT mencapai nilai tertinggi (2.857 MPa) pada 6 wt. % yang mana penambahan pengisi melebihi nilai ini menyebabkan kemerosotan pada kekuatan busa. Penyumbang utama kepada kemerosotan ini adalah dikaitkan dengan ikatan hidrogen antara-molekul yang terhalang. Kaedah Gerak Balas Permukaan kemudiannya digunakan untuk mencari formulasi pengisi optimum bagi kedua-dua sistem busa PU pelbagai pengisi dengan nilai LOI dan modulus mampatan sebagai tindak balas. Pengoptimuman telah menghasilkan nilai gabungan pengisi 0.9 wt. % Na-MMT dan 4.9 wt. % APP sebagai terbaik untuk sistem campuran MMT/APP, manakala nilai tero optimum hibrid APP-MMT adalah 9.2 wt. %. Pencirian kedua-dua formulasi tero optimum telah menunjukkan beberapa peningkatan berbanding busa tanpa pengisi. Nilai LOI meningkat sehingga 44 % dengan hanya pengurangan nilai modulus mampatan sebanyak 5 %. Melalui hasil dapatan kajian, busa PU berasaskan bio mempunyai potensi yang besar untuk digunakan dalam aplikasi berkaitan panel struktur yang memerlukan beban galas dan keupayaan pencegahan kemudahbakaran yang sederhana.

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LIST OF ABBREVIATIONS

ANOVA	-	Analysis of variance
AP	-	Adequate precision
APP	-	Ammonium polyphosphate
ASTM	-	American Society of Testing and Materials
ATH	-	Aluminium trihydroxide
ATR	-	Attenuated Total Reflectance
CCD	-	Central composite design
CDA	-	Chlorhexidine diacetate
CDWs	-	Construction and Demolition Wastes
CEC	-	Cation exchange capacity
CNTs	-	Carbon nanotubes
CTAB	-	Cetyl trimethylammonium bromide
CV	-	Coefficient of variance
DAMP	-	Diamino methylpentane
DAO	-	Diamino octane
DAP	-	Diamino propane
DOE	-	Design of experiment
DOPO	-	Dihydro oxaphosphaphenanthrene oxide
DTG	-	Derivative thermogravimetry
EGs	-	Expandable graphites
FESEM	-	Field Emission Scanning Electron Microscopy
FRs	-	Fire-retardants
FTIR	-	Fourier-Transform Infrared
HCl	-	Hydrochloric acid
HNTs	-	Halloysite nanotubes
HPEPO	-	Epoxidized palm oil hydrogel
HRR	-	Heat release rate
IFRs	-	Intumescent fire retardants
LOI	-	Limiting oxygen index
MDH	-	Magnesium hydroxide

MFAPP	-	Melamine-formaldehyde ammonium polyphosphate
MMT	-	Montmorillonite nanoclay
MWCNT	-	Multi-walled carbon nanotubes
Na-MMT	-	Sodium montmorillonite nanoclay
NIPU	-	Non-isocyanate polyurethane
OMMT	-	Organically-modified montmorillonite nanoclay
PA6	-	Polyamide 6
PEG	-	Poly(ethylene glycol)
PET	-	Polyethylene terephthalate
PF	-	Phenol formaldehyde
p-HRR	-	Peak heat release rate
PHU	-	Polyhydroxyurethane
PIR	-	polyisocyanurate
PMC	-	Polymer matrix composite
POP	-	Palm oil polyol
PP	-	Polypropylene
PPG	-	Poly(propylene glycol)
pphp	-	Parts per hundred parts of polyol
PS	-	Polystyrene
PU, PUR	-	Polyurethane
PVC	-	Poly(vinyl chloride)
RBD	-	Refined, bleached, and deodorised
RSM	-	Response Surface Methodology
SEM	-	Scanning Electron Microscopy
TGA	-	Thermogravimetry analysis
THE	-	Total heat evolved
THR	-	Total heat release
TML	-	Total mass loss
TPP	-	Triphenyl phosphate
VMFAPP	-	Vinyl-melamine-formaldehyde ammonium polyphosphate
XRD	-	X-ray diffraction

LIST OF SYMBOLS

A	-	Area in microscopy image
Al(OH)_3	-	Aluminium hydroxide
Al_2O_3	-	Aluminium oxide
A_o	-	Initial cross section area of specimen
APP-1	-	Pristine ammonium polyphosphate
APP-2	-	Silane-treated ammonium polyphosphate
Ca^{2+}	-	Calcium ion
CaO	-	Calcium oxide
C_F	-	Final value of measured oxygen concentration
CO	-	Carbon monoxide gas
CO_2	-	Carbon dioxide gas
C_V	-	Closed cell content of foam
d, d -spacing	-	Spacing between diffractioal lattice planes
d_{001}	-	Spacing between lattice planes in [001] orientation
d_{LOI}	-	Interval between oxygen concentrations
E	-	Compressive modulus of foam
h	-	Initial height of foam
H_2	-	Helium gas
k	-	Combustion pattern-dependent factor in LOI determination
K^+	-	Potassium ion
M	-	Microscopy image magnification
m	-	Mass of specimen
Mg(OH)_3	-	Magnesium hydroxide
MgO	-	Magnesium oxide
MnO_2	-	Manganese dioxide
N	-	Cellular density
n	-	Number of cells in microscopy image
N_2	-	Nitrogen gas
Na^+	-	Sodium ion
NCO	-	Isocyanate group

NH_2	- Amine group
NH_3	- Ammonia group
NH_4^+	- Ammonium ions
O_2	- Oxygen gas
OH	- Hydroxyl group
O_V	- Open cell content of foam
$P_{10\%}$	- Load at 10 % deformation
<i>p</i> -MDI	- Diphenylmethane diisocyanate
R^2	- Coefficient of multiple determination
$R_{H\text{-}bond}$	- Carbonyl hydrogen index
<i>sg</i>	- Specific gravity of material in solid form
SiCl_4	- Silicon tetrachloride
SiO_2	- Silicon dioxide
SiO_4	- Siloxane group
SiOH	- Silanol group
T_5	- Onset degradation temperature at 5 % mass loss
T_{50}	- Onset degradation temperature at 50 % mass loss
T_{bonded}	- Intensity of transmittance at bonded urethane peak
T_{free}	- Intensity of transmittance at free urethane peak
TiO_2	- Titanium dioxide
V	- Volume of foam determined by measurement
V-0, V-1, V-2	- Ratings in UL-94 Vertical Burning Test
V_{SPEC}	- Volume of foam determined by pycnometer
δ	- Deformation of foam
θ	- diffraction angle
λ	- X-ray radiation wavelength
ρ_{foam}	- Density of foam
ρ_{solid}	- Density of solid polymer
σ	- Compressive strength of foam

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CHAPTER 1

INTRODUCTION

1.1 Background of Study

It is not an overstatement to claim that today, polyurethane (PU) has found its niche among the most widely utilised polymer products in the world. Ever since its discovery by Professor Otto Bayer and colleagues in 1938 [1], PU can be found across numerous applications, ranging from consumer level products such as shoe insoles, cushions, mattresses, piping insulations, and food packaging; to high-end applications such as in maritime and aeronautics [2–6]. This is perhaps owed to PU's versatility and ease of tailoring to suit the end product, as PU can be synthesised across numerous forms such as elastomers, flexible foams, rigid foams, shape memory foams, coatings, films, and sealants [4,7–9]. Because of its versatility, rigid PU foams emerge as the plastic with the second highest global consumption, accounting for 25 % within the PU family [10]. Although commonly found as thermal insulators and barriers, due to its improved thermal insulating properties, it is not uncommon to discover rigid PU foams in structural applications – especially as a core material in sandwich composites – due to its superior strength-to-weight ratio, high chemical and moisture resistance, high weatherability, and high durability [6,11,12].

Like other polymeric materials, PU foams depend on petrochemical feedstock. Issues on sustainability, environmental impact, and the global crisis of petroleum supply depletion however, has caused PU industrialists to find alternatives for polymeric foam production from bio-renewable resources [13]. Vegetable oils such as rapeseed oil, soybean oil, canola oil, castor oil, and palm oil have been proven, through various research projects, to be suitable replacements for petroleum-based polyols. The vegetable oils however, need to undergo modifications like transesterification and glycerolysis to introduce hydroxyl groups, which is essential to react with isocyanates to form urethane linkages [11,13–15]. Though, this process is not necessary for castor

oil as it already possess built-in hydroxyl groups [11,13]. Numerous studies have been conducted to replace petroleum-based polyol, such as polyethylene glycol (PEG), with bio-renewable polyols in the production of PU foams [12,13,16–18]. Apart from acting as an alternative for petrochemical-based polyols, bio-renewable polyols also improve the properties of the foam. It has been found that PU foams produced from vegetable oils possess a higher thermal stability than those prepared with polypropylene oxide polyols [19]. Among these bio-based polyols, palm oil polyols have garnered more attention due to their low cost in production and their abundance in countries across regions like Malaysia, Indonesia, and Thailand.

Despite, however, possessing superior mechanical properties, one major drawback of polymeric materials is its high flammability due to it being organic. And since they are widely applied in industrial settings, efforts are being undertaken to ensure that the polymers are produced with the least flammability possible. Flame retardants are therefore commonly introduced to the polymers in order to reduce their susceptibility to fires. Common strategies to incorporate these flame retardants include the addition of non-flammable fillers, inclusion of substances that promote char formation, use of endothermic-decomposing materials, and use of materials that release gas or vapours in fire scenarios [20,21]. For this last approach mentioned, halogenated compounds are usually utilised, from which fire will liberate hydrogen halides. While this method offers a low-cost solution without compromising the polymers' mechanical properties, halogen combustions release hazardous toxic gases. This toxicity is a worldwide concern and has led to the restrict the use of halogen-based flame retardants, and to find safer, more environmental-friendly counterparts [22].

One of the solutions achieved in dealing with the toxic combustion gas issue is by employing non-halogen flame retarding additives such as ammonium polyphosphate (APP), triphenyl phosphate (TPP), and melamine polyphosphate. It has been affirmed in past research literature of the flame retarding properties of these materials across different polymeric systems [23–26]. These additives respond to combustion by forming carbonaceous char which acts as a thermal barrier and hinders the transfer of heat between the polymer and burning surface; resulting in insufficient

heat to sustain the combustion and hence breaking the ‘combustion cycle’. These phosphate-based additives also combat flame growth and spread by imparting ‘*fuel cooling*’ effect [27,28], which absorbs part of the heat and again interferes with the ‘combustion cycle.’ However, even though these flame retardants have been successful in countering the fire vulnerability of polymeric materials, they are synthetically produced. One of the greener alternatives has been to incorporate nanoclays, such as montmorillonite (MMT) into the polymeric system. It has been extensively shown in past publications that these nanoclays are able to decrease fire susceptibility of polymers via the rapid formation of char layering, while at the same time enhancing the thermal stability of the polymeric system [29–31].

Apart from thermal and fire-response improvement, incorporation of MMT into polymer systems also enables improved mechanical properties for load-bearing capabilities, as reported across previous research [17,32–35]. The polymer system, however, suffers a distinguishable drop in mechanical properties with an over-inclusion of MMT. This is understood to be due to a segregation of the nanoclays and forcing the nanoclays to be intercalated within the polymer gallery rather than exfoliated [17,32,36]. Surface treatment of the MMT, via an ion exchange method, can modify MMT to be hydrophobic and facilitates its assimilation within the polymer system; allowing the clays to be exfoliated within the polymer gallery. Previous studies have shown enhanced mechanical properties and thermal stability with the inclusion of organically-modified MMT nanoclays up to a certain limit [32,37,38]. Apart from this limitation, MMT modification can be conducted with a wide array of different suitable surfactants which opens up possibilities of ‘attaching’ various material-enhancing properties onto the clay, and subsequently onto the polymeric foam being embedded by the clay – such as anti-bacterial and fire-proofing properties [39–41].

In line with global efforts towards ‘*Green Chemistry*’ [42], it is hoped that this research will not only be able synthesise PU foam from renewable resources that possess superior mechanical and flame-retarding properties, but the products will also be on par with its synthetically-manufactured counterparts.

1.2 Problem Statement

PU foams are one of the largest polymeric materials used in many cellular applications which need on-going studies to improve their properties to meet the demands. As faced by other non-renewable petroleum-based polymers and non-biodegradability characteristics, PU foams also contribute to the environmental impact and sustainability. Renewable palm oil-based polyol is a good alternative to overcome the issue, owing to its abundance and eco-friendly. However, the process of converting palm oil into polyol (one of the raw materials to make PU) has shorten the molecular chain of palm oil which negatively influences the mechanical and thermal properties of the fabricated PU foam. Furthermore, the presence of various organic chemical compounds in palm oil such as fatty acids contribute to the flammability of the foam. Issue on poor mechanical strength and thermal stability can be countered back by incorporating filler such as montmorillonite (MMT) nanoclay into the foam. A drawback, a hydrophilic nature of MMT nanoclay tends to segregate upon the incorporation into the hydrophobic PU matrix. As such, surface modification via ion-exchange method can be carried out to increase the dispersion of MMT into the matrix. On the other hand, issue on the high flammability of PU foam can be diminished by adding fire retardants (FRs) such as ammonium polyphosphate (APP) into the formulation. However, PUs with a single-FR system struggled to achieve the desired level of commercially flame-retardant systems; and as such, multiple FRs system is often applied. Apart from this, another common issue is inefficient dispersion of FRs in the PU matrix, especially at higher loadings. This problem is more prominent in multiple-filler PU system. Studies show that surface modification of MMT with an ionic-bearing compound such as amine via ionic exchange process is able to increase the interfacial adhesion between the MMT and the matrix. Hence, this study had adopted the technique by manipulating the cationic-bearing ammonium groups, $-\text{NH}_4^+$ in APP to undergo the cation exchange process with the MMT to produce APP-MMT hybrid filler. APP-MMT is thought able to impart the synergistic effect in flame retarding and smoke suppression, and also can be regarded as a ‘green filler’ since it is non-halogenated and releases no toxic gases. Furthermore, the ammonium tail of APP-MMT could form the hydrogen bonding with the hydroxyl groups (OH) of PU chain, thus improving the strength of the foam. The later caters the issue on the shorten

molecular chain of palm oil polyol. To overcome the issue of imbalanced enhancement and deterioration of properties, the amounts of APP-MMT loadings were optimised, using response surface methodology (RSM) as the optimization tool. Optimisation using RSM is an important process to evaluate and determine the governing effects in a multiple-fillers system. To obtain an optimised PU foam formulation, the parameters involved are the Limiting Oxygen Index (LOI) and the compressive modulus, E_c .

1.3 Research Objectives

The main aim of this study is to develop a nature-based rigid polyurethane (PU) foam which possesses high modulus and flame-retarding capabilities for use in structural applications, such as wall panels and doors. Specific objectives include:

1. To investigate the effect of polyol substitution from petrochemical-based to palm oil-based on the flame retardancy, thermal stability, material, and mechanical properties of PU foam.
2. To investigate the influence of different types and loadings of fillers on the flame retardancy, thermal stability, material, and mechanical properties of palm oil-based PU foam.
3. To determine the effect of adding MMT modified with APP on the flame retardancy, thermal stability, material, and mechanical properties of palm oil-based PU foam.
4. To obtain optimised filler loading of palm oil-based PU foam with respect to its Limiting Oxygen Index (LOI) and compressive modulus.

1.4 Scope of Study

The scope for this study includes the following:

1. Fabrication of PU foam utilizing either poly(ethylene glycol) (PEG) or palm oil as its polyol source, prior to conducting characterisation tests on both types of foam.
2. Production of several single-filler palm oil-based PU foam formulation systems; incorporating either montmorillonite (MMT) nanoclay, pristine and silane-treated ammonium polyphosphates (APPs), fumed nanosilica particles, or basalt-based rockwool fibres at varied amounts. This is followed by characterisation process on the filled-foam systems.
3. Development of multi-filler palm oil-based PU foam systems, comprising of MMT nanoclay and silane-treated APP. One system utilises both fillers separately; while the other system incorporates APP-treated MMT organoclay as filler. The filler in latter PU system is obtained by first carrying out ion exchange surface modification process on the MMT nanoclay with silane-treated APP as its surfactant.
4. Foam formulation optimisation via surface response methodology (RSM); with compressive modulus and Limiting Oxygen Index (LOI) as its output responses. Filler loading is set as the input, with two different input groups are used for optimisation process, namely combined fillers (comprising of MMT and APP) and also single filler (comprising of APP-treated MMT organoclay).

The characterisation tests involved are as follows:

- (a) Optical Microscopy
- (b) Field Emission Scanning Electron Microscopy (FESEM)
- (c) X-ray Diffraction (XRD)
- (d) Fourier Transform Infrared (FTIR) spectroscopy

- (e) Thermogravimetric Analysis (TGA)
- (f) Helium gas pycnometry – ASTM D6226
- (g) Apparent Density – ASTM D1622
- (h) Limiting Oxygen Index (LOI) – ASTM D2863
- (i) Vertical Burning Test UL-94 – ASTM D3801
- (j) Compressive properties – ASTM D1621

1.5 Significance of Study

Production of conventional polyurethane (PU) foam relies on petrochemical feedstock. With global consensus towards environment sustainability, it is becoming vital for the utilisation of bio-based resources in the production of polymers such as PU. This study seeks to develop a nature-based PU foam composite that possesses both high modulus and flame-retarding properties, on par with its conventional counterparts. This polymeric foam is synthesised using palm oil-based polyol and distilled water as the blowing agent, and uses nature-based resources in its production. The fillers also comprise of bio-based resources such as montmorillonite (MMT) nanoclay and basalt-based rockwool, which possess latent potential to enhance flame-retardancy, thermal stability, mechanical and material properties. To date, little emphasis has been given to the study of flame-retardancy and other property enhancements using nature-based materials such as MMT in polymeric foams – let alone investigating the flame, thermal, material, and mechanical response of ammonium polyphosphate (APP) -modified nanoclays in palm oil-based PU foam – a novel aspect of this study.

1.6 Summary of Thesis

CHAPTER 1 is the introduction chapter. This chapter introduces the study by explaining its background, the problem statement, the research objectives and scopes, and also the significance of the study.

CHAPTER 2 is the study's literature review which is divided into three main parts. The first part elaborates polyurethane (PU) foam attributes. Topics covered include the background of polyurethane and different types of bio-based polyurethane foam is reviewed in detail in this sub-chapter. The second part considers the flame retardancy component of the research, which covers topics such as mechanics and modes of flame retardancy, as well as its methods of employment in polymers. The third and last part of this chapter discusses this study's optimisation, specifically on the response surface methodology (RSM). Topics that are investigated in this sub-chapter include the theory behind RSM and the application of RSM as an optimisation tool.

CHAPTER 3 explains the research methodology used in the research. The research methodology includes details on the materials used, the fabrication process of foams, clay modification process, and also the characterisation testing methods.

CHAPTER 4 explains the outcome of this research. Detailed explanations and discussions are provided based on the data obtained from the characterisation tests. Results from this study are also compared to other studies, where applicable.

CHAPTER 5 concludes the study. A brief and concise concluding remark is provided in this chapter, which highlights the overall outcome of this research. The end of this chapter includes a number of recommendations for future research enquiries.

REFERENCES

1. Akindoyo, J. O., Beg, M. D. H., Ghazali, S., Islam, M. R., Jeyaratnam, N. and Yuvaraj, A. R. Polyurethane types, synthesis and applications – a review. *RSC Advances*, 2016. 6(115):114453–114482.
2. Kurańska, M., Prociak, A., Cabulis, U., Kirpluks, M., Ryszkowska, J. and Auguścik, M. Innovative porous polyurethane-polyisocyanurate foams based on rapeseed oil and modified with expandable graphite. *Industrial Crops and Products*, 2017. 95:316–323.
3. Lee, Y. J., Park, C. K. and Kim, S. H. Fabrication of Castor-Oil/Polycaprolactone Based Bio-Polyurethane Foam Reinforced with Nanocellulose. *Polymer Composites*, 2018. 39(6):2004–2011.
4. Somarathna, H. M. C. C., Raman, S. N., Mohotti, D., Mutalib, A. A. and Badri, K.H. The use of polyurethane for structural and infrastructural engineering applications: A state-of-the-art review. *Construction and Building Materials*, 2018. 190:995–1014.
5. Yang, L. -T., Zhao, C. -S., Dai, C. -L., Fu, Y. -L. and Lin, S. -Q. Thermal and Mechanical Properties of Polyurethane Rigid Foam Based on Epoxidized Soybean Oil. *Journal of Polymers and the Environment*, 2012. 20:230–236.
6. Kausar, A. Polyurethane Composite Foams in High-Performance Applications: A Review. *Polymer-Plastics Technology and Engineering*, 2018. 57(4):1–24.
7. Kattiyaboot, T. and Thongpin, C. Effect of Natural Oil Based Polyols on the Properties of Flexible Polyurethane Foams Blown by Distilled Water. *Energy Procedia*, 2016. 89:177–185.
8. Mahmud, S. N. S., Jusoh, M. A. and Jasim, S. E. A Review: Method of Preparing Palm Oil Based Polyurethane. *Proceedings of The National Conference for Postgraduate Research 2016*. Pahang: Universiti Malaysia Pahang. 2016. 655–662.

9. Singh, I., Samal, S. K., Mohanty, S. and Nayak, S. K. Recent Advancement in Plant Oil Derived Polyol-Based Polyurethane Foam for Future Perspective – A Review. *European Journal of Lipid Science and Technology*, 2020. 122(3):1900225.
10. Gama, N. V., Ferreira, A. and Barros-Timmons, A. Polyurethane foams: Past, present, and future. *Materials*, 2018. 11(10):1841.
11. Marcovich, N. E., Kurańska, M., Prociak, A., Malewska, E. and Bujok, S. The effect of different palm oil based bio-polyols on foaming process and selected properties of porous polyurethanes. *Polymer International*, 2017. 66(11):1522–1529.
12. Li, Q. F., Feng, Y. L., Wang, J. W., Yin, N., Zhao, Y. H., Kang, M. Q. and Wang, X. W. Preparation and properties of rigid polyurethane foam based on modified castor oil. *Plastics, Rubber and Composites*, 2016. 45(1):16–21.
13. Prociak, A., Rojek, P. and Pawlik, H. Flexible polyurethane foams modified with natural oil based polyols. *Journal of Cellular Plastics*, 2012. 48(6):489–499.
14. Marcovich, N. E., Kuranska, M., Prociak, A., Malewska, E. and Kulpa, K. Open cell semi-rigid polyurethane foams synthesized using palm oil-based bio-polyol. *Industrial Crops and Products*, 2017. 102:88–96.
15. Ng, W. S., Lee, C. S., Chuah, C. H. and Cheng, S. -F. Preparation and modification of water-blown porous biodegradable polyurethane foams with palm oil-based polyester polyol. *Industrial Crops and Products*, 2017. 97:65–78.
16. Kurańska, M., Prociak, A., Kirpluks, M. and Cabulis, U. Porous polyurethane composites based on bio-components. *Composite Science and Technology*, 2013. 75:70–76.
17. Dzulkifli, M. H., Yahya, M. Y., Akhir, F. S. M., and Majid, R. A. Development of Rigid Biocomposite Polyurethane Foam for Load Bearing Application. *Jurnal Teknologi*, 2014. 68(3):53-56.

18. Zlatanić, A., Javni, I., Ionescu, M., Bilić, N. and Petrović, Z. S. Polyurethane molded foams with high content of hyperbranched polyols from soybean oil. *Journal of Cellular Plastics*, 2015. 51(3):289–306.
19. Stirna, U., Lazdiņa, B., Vilsone, D., Lopez, M. J., Vargas-Garcia, M. d. C., Suárez-Estrella, F. and Moreno, J. Structure and properties of the polyurethane and polyurethane foam synthesized from castor oil polyols. *Journal of Cellular Plastics*, 2012. 48(6):476-488.
20. Mouritz, A. P. and Gibson, A. G. *Fire Properties of Polymer Composite Materials*. Dordrecht: Springer. 2006.
21. Price, D., Anthony, G. and Carty, P. Introduction: polymer combustion, condensed phase pyrolysis and smoke formation. In: Horrocks, A. R. and Price, D. eds. *Fire Retardant Materials*. Cambridge: Woodhead Publishing Limited. 1–30; 2001.
22. Kiliaris, P. and Papaspyrides, C. D. Polymers on Fire. In: Papaspyrides, C. D. and Kiliaris, P. eds. *Polymer Green Flame Retardant*. New York: Elsevier. 1–40; 2014.
23. Guo, C., Zhou, L. and Lv, J., Effects of expandable graphite and modified ammonium polyphosphate on the flame-retardant and mechanical properties of wood flour-polypropylene composites. *Polymer and Polymer Composites*, 2013. 21(7):449–456.
24. Shen, M. Y., Chen, W. J., Kuan, C. F., Kuan, H. C., Yang, J. -M. and Chiang, C.-M. Preparation, characterization of microencapsulated ammonium polyphosphate and its flame retardancy in polyurethane composites. *Materials Chemistry and Physic*, 2016. 173:205–212.
25. Lu, C., Yang, D., Cao, Q. -Q., Huang, X. -H., Liu, J. -C. and Zhang, Y. -Q. Influence of morphology on the flame retardancy of polystyrene/nylon-6/ammonium polyphosphate/clay blends. *High Performance Polymer*, 2014. 26(5):507–516.

26. Feng, C., Liang, M., Jiang, J., Huang, J. and Liu, H. Synergistic effect of a novel triazine charring agent and ammonium polyphosphate on the flame-retardant properties of halogen-free flame-retardant polypropylene composites. *Thermochimica Acta*, 2015. 627–629:83–90.
27. Morgan, A. B. and Worku, A. Z. Flame Retardants: Overview. In: Ley, C. ed. *Kirk-Othmer Encyclopedia of Chemical Technology*. 4th Ed. Vol. 10. John Wiley & Sons. 450–485; 2015.
28. Dasari, A., Yu., Z. -Z., Cai, G. -P. and Mai, Y. -W. Recent developments in the fire retardancy of polymeric materials. *Progress in Polymer Science*, 2013. 38(9): 1357–1387.
29. Niu, M. and Wang, G. J. Study on the nanocomposite foam of cardanol phenolic resin and organo-modified montmorillonite. *Advanced Materials Research*, 2013. 712–715:147–155.
30. He, X., Zhang, W., Yi, D. and Yang, R. Flame retardancy of ammonium polyphosphate-montmorillonite nanocompounds on epoxy resin. *Journal of Fire Science*, 2016. 34(3):212–225.
31. Hanna, A. A., Nour, M. A., Souaya, E. R., Sherief, M. A. and Abdelmoaty, A. S. Studies on the flammability of polypropylene/ammonium polyphosphate and montmorillonite by using the cone calorimeter test. *Open Chemistry*, 2018. 16(1):108–115.
32. Nik Pauzi, N. N. P., Majid, R. A., Dzulkifli, M. H. and Yahya, M. Y. Development of Rigid Bio-Based Polyurethane Foam Reinforced with Nanoclay. *Composites Part B: Engineering*, 2014. 67:521–526.
33. Ali, E. S. and Ahmad, S. Bionanocomposite hybrid polyurethane foam reinforced with empty fruit bunch and nanoclay. *Composites Part B: Engineering*, 2012. 43:2813–2816.
34. Harikrishnan, G., Patro, T. U. and Khakhar, D. V. Polyurethane Foam - Clay Nanocomposites: Nanoclays as Cell Openers. *Industrial and Engineering Chemistry Research*, 2006. 45(21):7126–7134.

35. El-Fattah, M. A., El Saeed, A. M., Dardir, M. M. and El-Sockary, M. A. Studying the effect of organo-modified nanoclay loading on the thermal stability, flame retardant, anti-corrosive and mechanical properties of polyurethane nanocomposite for surface coating. *Progress in Organic Coatings*, 2015. 89:212–219.
36. Chuayjuljit, S., Maungchareon, A. and Saravari, O. Preparation and Properties of Palm Oil-Based Rigid Polyurethane Nanocomposite Foams. *Journal of Reinforced Plastics and Composites*, 2010. 29(2):218–225.
37. Fan, H., Tekeei, A., Suppes, G. J. and Hsieh, F. H. Properties of biobased rigid polyurethane foams reinforced with fillers: Microspheres and nanoclay. *International Journal of Polymer Science*, 2012. 2012:474803.
38. Liang, K. and Shi, S. Q. Nanoclay Filled Soy-Based Polyurethane Foam. *Journal of Applied Polymer Science*, 2011. 119(3):1857–1863.
39. Meng, N., Zhou, N., Zhang, S. and Shen, J. Controlled release and antibacterial activity chlorhexidine acetate (CA) intercalated in montmorillonite. *International Journal of Pharmaceutics*, 2009. 382(1–2):45–49.
40. Chang, J. -H. and An, Y. U. Nanocomposites of polyurethane with various organoclays: Thermomechanical properties, morphology, and gas permeability. *Journal of Polymer Science Part B: Polymer Physics*, 2002. 40(7):670–677.
41. Beyer, G. Flame retardancy of thermoplastic polyurethane and polyvinyl chloride by organoclays. *Journal of Fire Sciences*, 2007. 25(1):65–78.
42. Das, S., Pandey, P., Mohanty, S. and Nayak, S. K. An Insight on Castor Oil Based Polyurethane and Nanocomposites: Recent Trends and Development. *Polymer-Plastics Technology and Engineering*, 2017. 56(14):1556–1585.
43. Eaves, D. Foam Fundamentals. In: Eaves, D. ed. *Handbook of Polymer Foams*. United Kingdom: Rapra Technology Ltd. 1–8; 2004.

44. Gibson, L. J. and Ashby, M. F. *Cellular solids – Structure and Properties*. 2nd Ed. Cambridge University Press: Cambridge. 1998.
45. Cunningham, A. and Hilyard, N. C. Physical behaviour of polymeric foams – an overview. In: Cunningham, A. and Hilyard, N. C. eds. *Low density cellular plastics – Physical basis of behavior*. Netherland: Springer Netherlands. 1–21; 1994.
46. Singh, S. N. *Blowing Agents for Polyurethane Foams*. United Kingdom: Rapra Technology Limited. Report 142, Vol. 12, No. 10. 2002.
47. Eaves, D. Rigid Polyurethane Foams. In: Eaves, D. ed. *Handbook of Polymer Foams*. United Kingdom: Rapra Technology Limited. 55–84; 2004.
48. Housel, T. Flexible Polyurethane Foams In: Eaves, D. ed. *Handbook of Polymer Foams*. United Kingdom: Rapra Technology Limited. 85–122; 2004.
49. Thomas, N. L. Rigid PVC Foams. In: Eaves, D. ed. *Handbook of Polymer Foams*. United Kingdom: Rapra Technology Limited. 123–154; 2004.
50. Howick, C. J. Flexible PVC Foams. In: Eaves, D. ed. *Handbook of Polymer Foams*. United Kingdom: Rapra Technology Limited. 155–172; 2004.
51. Njuguna, J., Michałowski, S., Pielichowski, K., Kayvantash, K. and Walton, A. C. Fabrication, Characterization and Low-Velocity Impact Testing of Hybrid Sandwich Composites with Polyurethane/Layered Silicate Foam Cores. *Polymer Composites*, 2011. 32(1):6-13.
52. Tan, S., Abraham, T., Ference, D. and Macosko, C. W. Rigid polyurethane foams from a soybean oil-based polyol. *Polymer*, 2011. 52(13):2840-2846.
53. Tenorio-Alfonso, A., Sánchez, M. C. and Franco, J. M. A Review of the Sustainable Approaches in the Production of Bio-based Polyurethanes and Their Applications in the Adhesive Field. *Journal of Polymers and the Environment*, 2020. 28:749–774.

54. Kurańska, M. and Prociak, A. The influence of rapeseed oil-based polyols on the foaming process of rigid polyurethane foams. *Industrial Crops and Products*, 2016. 89:182–187.
55. Mills, N. J. *Polymer foams handbook: Engineering and Biomechanics Applications and Design Guide*. Burlington, Massachusetts: Elsevier Limited. 2007.
56. Ionescu, M. Polyol. In: Ionescu, M. ed. *Chemistry and Technology of Polyols for Polyurethanes*. United Kingdom: Rapra Technology Limited. 1–12; 2008.
57. Levchik, S. V. and Weil, E. D. Thermal decomposition, combustion and fire-retardancy of polyurethanes - A review of the recent literature. *Polymer International*, 2004. 53(11):1585–1610.
58. Ashida, K. *Polyurethane and Related Foam*. Florida: CRC Press. 2006.
59. Mat Rejab, M. R. *Polymeric Foam as Core of Side-Door Impact Beam*. Master Thesis. Universiti Teknologi Malaysia; 2004.
60. Karak, N. *Biobased Smart Polyurethane Nanocomposites - From Synthesis to Applications*. United Kingdom: Royal Society of Chemistry. 2017.
61. Touchet, T. J. and Cosgriff-Hernandez, E. Hierarchical structure–property relationships of segmented polyurethanes. In: Cooper S. L. and Guan J. eds. *Advances in Polyurethane Biomaterials*. Massachusetts: Woodhead Publishing Limited. 3–22; 2016.
62. Brereton, G., Emanuel Jr., R. M., Lomax, R., Pennington, K., Ryan, T., Tebbe, H., Timm, M., Ware, P., Winkler, K., Yuan, T., Zhu, Z., Adam, N., Avar, G., Blankenheim, H., Friederichs, W., Giersig, M., Weigand, E., Halfmann, M., Wittbecker, F. -W., Larimer, D. -R., Maier, U., Meyer-Ahrens, S., Noble, K. - L. and Wussow, H. -G. Polyurethanes. in: *Ullmann's Encyclopedia of Industrial Chemistry*. 1–76; 2019.
63. Lim, H., Kim, S. H., and Kim, B. K. Effects of silicon surfactant in rigid polyurethane foams. *eXPRESS Polymer Letters*, 2008. 2(3):194-200.

64. Park, S. B., Choi, S. W., Kim, J. H., Bang, C. S. and Lee, J. M. Effect of the blowing agent on the low-temperature mechanical properties of CO₂- and HFC-245fa-blown glass-fiber-reinforced polyurethane foams. *Composites Part B: Engineering*, 2016. 93:317–327.
65. Kraitape, N. and Thongpin, C. Influence of Recycled Polyurethane Polyol on the Properties of Flexible Polyurethane Foams. *Energy Procedia*, 2016. 89:186–197.
66. Zhang, X. D., Macosko, C. W., Davis, H. T., Nikolov, A. D. and Wasan, D. T. Role of Silicone Surfactant in Flexible Polyurethane Foam. *Journal of Colloid and Interface Science*, 1999. 215(2):270–279.
67. Han, M. S., Choi, S. J., Kim, J. M., Kim, Y. H., Kim, W. N., Lee, H. S. and Sung, J. Y. Effects of Silicone Surfactant on the Cell Size and Thermal Conductivity of Rigid Polyurethane Foams by Environmentally Friendly Blowing Agents. *Macromolecular Research*, 2009. 17(1):44–50.
68. Semenzato, S., Lorenzetti, A., Modesti, M., Ugel, E., Hrelja, D., Besco, S., Michelin, R. A., Sassi, A., Facchin, G., Zorzi, F. and Bertani, R. A novel phosphorus polyurethane FOAM/montmorillonite nanocomposite: Preparation, characterization, and thermal behaviour. *Applied Clay Science*, 2009. 44(1–2):35–42.
69. Alis, A., Majid, R. A. and Mohamad, Z. Morphologies and thermal properties of palm-oil based rigid polyurethane/halloysite nanocomposite foams. *Chemical Engineering Transactions*, 2019. 72:415–420.
70. Anil, D. *Halloysite Containing Polyurethane Foams as Insulation Materials with Enhanced Flame Retardance*. Master Thesis. Sabanci University; 2019.
71. Nikje, M. M. A. and Yaghoubi, A. Preparation and properties of polyurethane/functionalized multi-walled carbon nanotubes rigid foam nanocomposites. *Polimery-Warsaw*, 2014. 59(11–12):776–782.
72. Caglayan, C., Gurkan, I., Gungor, S. and Cebeci, H. The effect of CNT-reinforced polyurethane foam cores to flexural properties of sandwich

- composites. *Composites Part A: Applied Science and Manufacturing*, 2018. 115:187–195.
73. Adnan, S. A., Zainuddin, F., Hidayah, N. and Zaidi, A. The effect of magnesium content on the properties of palm oil based polyurethane foam. *AIP Conference Proceedings*, 2018. 2030(1):020050.
74. Wahab, H. A., Rus, A. Z. M., Abdullah, M. F. L. and Abdullah, N. M. Design of experiment for sound absorption materials of microporous polymer. *Proceedings of 2019 International Conference on Information Science and Communication Technology (ICISCT)*. March 9–10, 2019. Karachi: IEEE. 2019. 1–7.
75. Seo, W. J., Park, J. H., Sung, Y. T., Hwang, D. H., Kim, W. N. and Lee, H. S. Properties of water-blown rigid polyurethane foams with reactivity of raw materials. *Journal of Applied Polymer Science*, 2004. 93(5):2334–2342.
76. Rokicki, G., Parzuchowski, P. G. and Mazurek, M. Non-isocyanate polyurethanes: Synthesis, properties, and applications. *Polymers for Advanced Technologies*, 2015. 26(7):707–761.
77. Ghasemlou, M., Daver, F., Ivanova, E. P. and Adhikari, B. Bio-based routes to synthesize cyclic carbonates and polyamines precursors of non-isocyanate polyurethanes: A review. *European Polymer Journal*, 2019. 118:668–684.
78. Xi, X., Pizzi, A., Gerardin, C. and Du, G. Glucose-Biobased Non-Isocyanate Polyurethane Rigid Foams. *Journal of Renewable Material*, 2019. 7(3):301–312.
79. Xi, X., Pizzi, A., Gerardin, C., Lei, H., Chen, X. and Amirou, S. Preparation and Evaluation of Glucose Based Non-Isocyanate Polyurethane Self-Blowing Rigid Foams. *Polymers*, 2019. 11(11):1802.
80. Kostopoulos, V. and Kollia, E. Development and Mechanical Characterization of a Non-Isocyanate Rigid Polyurethane Foam. *Research and Development in Material Science*, 2019. 10(3):1149–1156.

81. Clark, J. H., Farmer, T. J., Ingram, I. D. V., Lie, Y. and North, M. Renewable Self-Blowing Non-Isocyanate Polyurethane Foams from Lysine and Sorbitol. *European Journal of Organic Chemistry*, 2018. 2018(31):4265–4271.
82. Booysen, J., Marx, S., Muller, L. C., Vermeulen, U. and Grobler, A. Synthesis of Novel Non-Isocyanate Polyhydroxyurethane from L-Lysine and its Application. *Proceedings of 7th International Conference on Latest Trends in Engineering & Technology (ICLTET'2015)*. November 26–27, 2015. Pretoria, South Africa: IIE. 2015. 184–190.
83. Birukov, O., Potashnikova, R., Leykin, A., Figovsky, O. and Shapovalov, L. Advantages in chemistry and technology of non-isocyanate polyurethane. *Scientific Israel Technological Advantages*, 2014. 16(3):92–102.
84. Lauth, M., Mulhaupt, R. and Blattmann, H. *Non-Isocyanate Polyurethane Foams*. US10450416B2. 2019.
85. Petrović, Z. S. Polyurethanes from vegetable oils. *Polymer Reviews*, 2008. 48(1):109–155.
86. Gu, R., Konar, S. and Sain, M. Preparation and characterization of sustainable polyurethane foams from soybean oils. *Journal of the American Oil Chemists' Society*, 2012. 89:2103–2111.
87. Ji, D., Fang, Z., He, W., Luo, Z., Jiang, X., Wang, T. and Guo, K. Polyurethane rigid foams formed from different soy-based polyols by the ring opening of epoxidised soybean oil with methanol, phenol, and cyclohexanol. *Industrial Crops and Products*, 2015. 74:76–82.
88. Bresolin, D., Valério, A., de Oliveira, D., Lenzi, M. K., Sayer, C. and de Araújo, P. H. H. Polyurethane Foams Based on Biopolyols from Castor Oil and Glycerol. *Journal of Polymers and the Environment*, 2018. 26:2467–2475.
89. Zhang, M., Pan, H., Zhang, L., Hu, L. and Zhou, Y. Study of the mechanical, thermal properties and flame retardancy of rigid polyurethane foams prepared from modified castor-oil-based polyols. *Industrial Crops and Products*, 2014. 59:135–143.

90. Hilmi, H., Zainuddin, F., Cheng, T. S. and Lan, D. N. U. Mechanical properties of palm oil based bio-polyurethane foam of free rise and various densities. *AIP Conference Proceedings*, 2017. 1901(1):070002.
91. Czlonka, S., Bertino, M. F., Kośny, J., Strąkowska, A., Masłowski, M. and Strzelec, K. Linseed oil as a natural modifier of rigid polyurethane foams. *Industrial Crops and Products*, 2018. 115:40–51.
92. Pillai, P. K. S., Li, S., Bouzidi, L. and Narine, S. S. Polyurethane foams from chlorinated and non-chlorinated metathesis modified canola oil polyols. *Journal of Applied Polymer Science*, 2018. 135(33):46616.
93. Victor, E. I., Okechukwu, J. O., Femi, M. A. and Ogbonneya, O. M. Polyurethane Foam Production using Sunflower oil and Soybean oil as Polyol and Surfactant. *International Journal of Modern Science and Technology*, 2019. 4(2):48–53.
94. Zhu, S., Chen, K., Xu, J., Li, J. and Mo, L. Bio-based polyurethane foam preparation employing lignin from corn stalk enzymatic hydrolysis residues. *RSC Advances*, 2018. 8(28):15754–15761.
95. Wahid, M. B., Abdullah, S. N. A. and Henson, I. E. Oil palm - Achievements and potential. *Plant Production Science*, 2005. 8(3):288–297.
96. Sawpan, M. A. Polyurethanes from vegetable oils and applications: a review. *Journal of Polymer Research*, 2018. 25:184.
97. Awalludin, M. F., Sulaiman, O., Hashim, R. and Nadhari, W. N. A. W. An overview of the oil palm industry in Malaysia and its waste utilization through thermochemical conversion, specifically via liquefaction. *Renewable and Sustainable Energy Reviews*, 2015. 50:1469–1484.
98. Khatun, R., Reza, M. I. H., Moniruzzaman, M. and Yaakob, Z. Sustainable oil palm industry: The possibilities. *Renewable and Sustainable Energy Reviews*, 2017. 76:608–619.

99. Nik Pauzi, N. N. P. *Synthesis and Characterization of Rigid Polyurethane-Palm Oil Based Polyol/Diaminopropane-Montmorillonite Nanocomposite Foam*. Master thesis. Universiti Teknologi Malaysia; 2013.
100. Tajau, R., Rohani, R., Wan Isahak, W. N. R., Salleh, M. Z. and Ghazali, Z. Development of new bio-based polyol ester from palm oil for potential polymeric drug carrier. *Advances in Polymer Technologies*, 2018. 37(8):3552–3560.
101. Badri, K. H., Ahmad, S. H. and Zakaria, S. Development of zero ODP rigid polyurethane foam from RBD palm kernel oil. *Journal of Material Science Letters*, 2000. 19:1355–1356.
102. Zubir, S. A., Saad, N. M., Harun, F. W., Ali, E. S. and Ahmad, S. Incorporation of palm oil polyol in shape memory polyurethane: Implication for development of cardiovascular stent. *Polymer for Advanced Technologies*, 2018. 29(12):2926–2935.
103. Zaimahwati, Agusnar, H., Rihayat, T., Reflianto, D. and Gea, S. The manufacture of palm oil-based polyurethane nanocomposite with organic montmorillonite nanoparticle as paint coatings. *International Journal of ChemTech Research*, 2015. 7(5):2537–2544.
104. Maznee, T. I. T. N., Norin, Z. K. S., Ooi, T. L., Salmiah A. and Gan, A. Effects of Additives on Palm-Based Polyurethane Foams. *Journal of Oil Palm Research*, 2001. 13(2):7–15.
105. Ain, N. H., Maznee, T. I. T. N., Noor, M. A. M., Srihanum, A., Devi, K. P. P., Mohd, N. S., Mohdnoor, N., Kian, Y S., Hassan, H. A., Campara, I., Schiffman, C. M., Pietrzyk, K., Sendijarevic, V. and Sendijarevic, I. Structure-property performance of natural palm olein polyol in the viscoelastic polyurethane foam. *Journal of Cellular Plastics*, 2017. 53(1):65–81.
106. Arniza, M. Z., Hoong, S.S., Idris, Z., Yeong, S.K., Hassan, H. A., Din, A. K. and Choo, Y. M. Synthesis of Transesterified Palm Olein-Based Polyol and Rigid Polyurethanes from this Polyol. *Journal of the American Oil Chemists' Society*, 2015. 92:243–255.

107. Chuayjuljit, S., Sangpakdee, T. and Saravari, O. Processing and Properties of Palm Oil-Based Rigid Polyurethane Foam. *Journal of Metals, Materials, and Minerals*, 2007. 17(1):17–23.
108. Pillai, P. K. S., Li, S., Bouzidi, L. and Narine, S. S. Metathesized palm oil & novel polyol derivatives: Structure, chemical composition and physical properties. *Industrial Crops and Products*, 2016. 84:205–223.
109. Pillai, P. K. S., Li, S., Bouzidi, L. and Narine, S. S. Metathesized palm oil polyol for the preparation of improved bio-based rigid and flexible polyurethane foams. *Industrial Crops and Products*, 2016. 83:568–576.
110. Lee, C. S., Ooi, T. L., Chuah, C. H. and Ahmad, S. Rigid polyurethane foam production from palm oil-based epoxidized diethanolamides. *Journal of the American Oil Chemists' Society*, 2007. 84(12):1161–1167.
111. Chian, K. S. and Gan, L. H. Development of a Rigid Polyurethane Foam from Palm Oil. *Journal of Applied Polymer Science*, 1998. 68(3):509–515.
112. Badri, K. H., Ahmad, S. H. and Zakaria, S. Production of a high-functionality RBD palm kernel oil-based polyester polyol. *Journal of Applied Polymer Science*, 2001. 81(2):384–389.
113. Sari, K. A. M., Sani, A. R. M., Mat, S. and Badri, K. H. Optimization of polyurethane lightweight aggregate with the addition of palm-based polyol. *Applied Mechanics and Materials*, 2015. 773–774:1017–1021.
114. Tanaka, R., Hirose, S., and Hatakeyama, H. Preparation and characterization of polyurethane foams using a palm oil-based polyol. *Bioresource Technology*, 2008. 99(9):3810–3816.
115. Pawlik, H. and Prociak, A. Influence of Palm Oil-Based Polyol on the Properties of Flexible Polyurethane Foams. *Journal of Polymers and the Environment*, 2012. 20:438–445.
116. Pitts, W. M. Role of two stage pyrolysis in fire growth on flexible polyurethane foam slabs. *Fire and Materials*, 2014. 38(3):323–338.

117. Vahabi, H., Rastin, H., Movahedifar, E., Antoun, K., Brosse, N. and Saeb, M. R. Flame Retardancy of Bio-Based Polyurethanes: Opportunities and Challenges. *Polymers*, 2020. 12(6):1234.
118. Pagacz, J. and Pielichowski, K. Preparation and Characterization of PVC/Montmorillonite Nanocomposites — A Review. *Journal of Vinyl and Additive Technology*, 2009. 15(2):61–76.
119. Bee, S. -L., Abdullah, M. A. A., Bee, S. -T., Sin, T. L. and Rahmat, A. R. Polymer nanocomposites based on silylated-montmorillonite: A review. *Progress in Polymer Science*, 2018. 85:57–82.
120. Liang, Z., Yin, J. and Xu, H. Polyimide/montmorillonite nanocomposites based on thermally stable, rigid-rod aromatic amine modifiers. *Polymer*, 2003. 44(5):1391–1399.
121. Gul, S., Kausar, A., Muhammad, B. and Jabeen, S. Technical Relevance of Epoxy/Clay Nanocomposite with Organically Modified Montmorillonite: A Review. *Polymer-Plastics Technology and Engineering*, 2016. 55(13):1393–1415.
122. Zeng, Q. H., Yu, A. B., Lu, G. Q. and Paul, D. R. Clay-Based Polymer Nanocomposites: Research and Commercial Development. *Journal of Nanoscience and Nanotechnology*, 2005. 5(10):1574–1592.
123. Dzulkifli, M. H. *Rigid Bio-Nanocomposite Polyurethane Foam Reinforced with Organically-Modified Montmorillonite Nanoclay*. Master thesis. Universiti Teknologi Malaysia; 2016.
124. Usmani, M. A., Khan, I., Ahmad, N., Bhat, A. H., Sharma, D. K., Rather, J. A. and Hassan, S. I. Modification of Nanoclay Systems: An Approach to Explore Various Applications. In: Jawaid, M., Qaiss, A. el -K. and Bouhfid, R. eds. *Nanoclay Reinforced Polymer Composites - Nanocomposites and Bionanocomposites*. Singapore: Springer. 57–83; 2016.
125. Raji, M., Mekhzoum, M. E. M., Qaiss, A. el -K. and Bouhfi, R. Nanoclay Modification and Functionalization for Nanocomposites Development: Effect

- on the Structural, Morphological, Mechanical and Rheological Properties. In: Jawaid, M., Qaiss, A. el -K. and Bouhfid, R. eds. *Nanoclay Reinforced Polymer Composites - Nanocomposites and Bionanocomposites*. Singapore: Springer. 1–34; 2016.
126. Bhattacharya, S. N., Gupta, R. K. and Kamal, M. R. *Polymeric Nanocomposites – Theory and Practice*. Munich: Carl Hanser Publishers. 2008.
127. Ohrdorf, K. -H. and Flachberger, H. Processing of calcium montmorillonites for use in epolymers. In: Laske S. ed. *Polymer Nanoclay Composites*. Oxford, UK: Elsevier. 1–25; 2015.
128. Pradhan, K. C. and Nayak, P. L. Synthesis and Characterization of Polyurethane Nanocomposite from Castor Oil-Hexamethylene Diisocyanate (HMDI). *Advances in Applied Science Research*, 2012. 3(5):3045-3052.
129. Monti, M., Tsampas, S. A., Fernberg, S. P., Blomqvist, P., Cuttica, F., Fina, A. and Camino, G. Fire reaction of nanoclay-doped PA6 composites reinforced with continuous glass fibers and produced by commingling technique. *Polymer Degradation and Stability*, 2015. 121:1–10.
130. Ribeiro, S. P. S., Estevão, L. R. M. and Nascimento, R. S. V. Effect of clays on the fire-retardant properties of a polyethylenic copolymer containing intumescence formulation. *Science and Technology of Advanced Materials*, 2008. 9(2):024408.
131. Kanny, K. and Mohan, T. P. Resin infusion analysis of nanoclay filled glass fibre laminates. *Composites Part B: Engineering*, 2014. 58:328-334.
132. Zahedi, M., Khanjanzadeh, H., Pirayesh, H. and Saadatnia, M. A. Utilization of natural montmorillonite modified with dimethyl, dehydrogenated tallow quaternary ammonium salt as reinforcement in almond shell flour-polypropylene bio-nanocomposites. *Composites Part B: Engineering*. 2015. 71:143–151.

133. Mohan, T. P., and Kanny, K. Influence of nanoclay on rheological and mechanical properties of short glass fiber-reinforced polypropylene composites. *Journal of Reinforced Plastics and Composites*, 2011. 30(2):152–160.
134. Zhang, X., Xu, R., Wu, Z. and Zhou, C. The synthesis and characterization of polyurethane/clay nanocomposites. *Polymer International*, 2003. 52(5):790–794.
135. Gopakumar, T. G., Lee, J. A., Kontopoulou, M. and Parent, J. S. Influence of clay exfoliation on the physical properties of montmorillonite/polyethylene composites. *Polymer*, 2002. 43(20):5483–5491.
136. Hassan, A., Lee H. Y. and Hasan, M. Effect of Ammonium Polyphosphate on Flame Retardancy, Thermal Stability, and Mechanical Properties of Unsaturated Polyester/Phenolic/Montmorillonite Nanocomposites. *Advances in Polymer Technology*, 2017. 36(3):278–283.
137. Woldensenbet, E. and Sankella, N. Flexural properties of nanoclay syntactic foam sandwich structures. *Journal of Sandwich Structure Materials*, 2009. 11(5):425–444.
138. Yang, Z., Wu, M., Chen, G., Li, S., Peng P. and Zhang, Q. The effect of montmorillonite modification on crystallization behaviour of polypropylene/Montmorillonite composites. *Polymers and Polymer Composites*, 2016. 24(5):331–340.
139. Widjonarko, D. M., Mayasari, O. D., Wahyuningsih, S. and Nugrahaningtyas, K. D. Modification of Montmorillonite with Cetyl Trimethylammonium Bromide and Tetra Ethyl Ortho Silicate. *IOP Conference Series: Materials Science and Engineering*, 2018. 333:012408.
140. Zaini, M., Majid, R. A. and Nikbakht, H. Modification of Montmorillonite with Diamine Surfactants. *Applied Mechanics and Materials*, 2014. 695:224-227.
141. Boaro, L. C. C., Campos, L. M., Varca, G. H. C., Santos, T. M. R. d., Marques, P. A., Sugii, M. M., Saldanha, N. R., Cogo-Müller, K., Brandt, W. C., Braga,

- R. R. and Parra, D. F. Antibacterial resin-based composite containing chlorhexidine for dental applications. *Dental Materials*, 2019. 35(6):909–918.
142. Singla, P., Mehta, R. and Upadhyay, S. N. Clay Modification by the Use of Organic Cations. *Green and Sustainable Chemistry*, 2012. 2(1):21–25.
143. Banik, I. and Sain, M. M. Nanoclay Modified Water-Blown Polyurethane Foams Derived from Bifunctional Soybean Oil-Based Polyol. *Polymer-Plastics Technology and Engineering*, 2010. 49(7):701–706.
144. Duquesne, S. and Futterer, T. Intumescence Systems. In: Morgan A. B. and Wilkie C. A. eds. *The Non-Halogenated Flame Retardant Handbook*. Massachusetts: Scrivener Publishing. 293–346; 2014.
145. Watanabe, M., Sakurai, M. and Mioko, M. Preparation of Ammonium Polyphosphate and Its Application to Flame Retardant. *Phosphorus Research Bulletin*, 2009. 23:35–44.
146. Hahn, H., Heumann, H., Liebing, H., Schweppe, M. and Hilt, W. *Process for the Manufacture of Ammonium Polyphosphate*. 4104362. 1978.
147. Yao, W., Wang, H., Guan, D., Fu, T., Zhang, T. and Dou, Y. The Effect of Soluble Ammonium Polyphosphate on the Properties of Water Blown Semirigid Polyurethane Foams. *Advances in Materials Science and Engineering*, 2017. 2017:5282869.
148. Modesti., M., Lorenzetti, A., Simioni, F. and Checchin, M. Influence of different flame retardants on fire behaviour of modified PIR/PUR polymers. *Polymer Degradation and Stability*, 2001. 74(3):475–479.
149. Luo, F., Wu, K., Lu, M., Nie, S., Li, X. and Guan, X. Thermal degradation and flame retardancy of microencapsulated ammonium polyphosphate in rigid polyurethane foam. *Journal of Thermal Analysis and Calorimetry*, 2015. 120(2):1327–1335.

150. Kawaguchi, M. Dispersion Stabilities and Rheological Properties of Fumed Silica Suspensions. *Journal of Dispersion Science and Technology*, 2017. 38(5):626–660.
151. Gun'ko, V. M., Zarko, V. I., Turov, V. V., Voronin, E. F., Mironyuk, I. F. and Chuiko, A. A. Structural and Adsorptive Characteristics of Fumed Silicas in Different Media. In: Bergna, H. E. and Roberts, W. O. eds. *Colloidal Silica: Fundamentals and Applications*. Florida: CRC Press. 499–530; 2006.
152. Barthel, H., Rosch, L. and Weis, J. Fumed Silica - Production, Properties, and Applications. In: Auner N. and Weis J. eds. *Organosilicon Chemistry*. New York: Wiley-VCH. 761–778; 1996.
153. Liu, J., Kutsovsky, Y. E. and Fotou, G. P. *Process for the Production of Fumed Silica*. US 2007/0253884 A1. 2007.
154. Liu, T., Mao, L., Liu, F., Jiang, W., He, Z. and Fang, P. Preparation, structure, and properties of flexible polyurethane foams filled with fumed silica. *Wuhan University Journal of Natural Sciences*, 2011. 16(1):29–32.
155. Huang, W., Xu, H., Fan, Z., Ao, Y. and Liu, J. Compressive response of composite ceramic particle-reinforced polyurethane foam. *Polymer Testing*, 2020. 87:106514.
156. Santiago-Calvo, M., Tirado-Mediavilla, J., Ruiz-Herrero, J. L., Rodríguez-Pérez, M.A. and Villafaña, F. The effects of functional nano fillers on the reaction kinetics, microstructure, thermal and mechanical properties of water blown rigid polyurethane foams. *Polymer*, 2018. 150:138–149.
157. Nikje, M. M. A. and Tehrani, Z. M. Thermal and Mechanical Properties of Polyurethane Rigid Foam/Modified Nanosilica Composite. *Polymer Engineering and Science*, 2010. 50(3):468–473.
158. Viramgama, N., Garg, A., Thomas, K. and Krishnan, P. Thermomechanical Characterization of Vacuum Resin Infusion-Molded Ceramic Rock-Derived Natural Wool-Reinforced Epoxy and Cashew Nut Shell Liquid-Based Composites. In: Khan, A., Rangappa, S. M., Jawaid, M., Siengchin, S. and

- Asiri, A. M. eds. *Hybrid Fiber Composites: Materials, Manufacturing, Process Engineering*. Weinheim: Wiley-VCH. 265–306; 2020.
159. Zihlif, A. M. and Ragosta, G. A. Study on the Physical Properties of Rock Wool Fiber–Polystyrene Composite. *Journal of Thermoplastic Composite Materials*, 2003. 16(3):273–283.
 160. Öztürk, B. Hybrid Effect in the Mechanical Properties of Jute/Rockwool Hybrid Fibres Reinforced Phenol Formaldehyde Composites. *Fibers and Polymers*, 2010. 11(3):464–473.
 161. da Rocha, E. B. D., de Sousa, A. M. F. and Furtado, C. R. G. Properties Investigation of novel nitrile rubber composites with rockwool fibers. *Polymer Testing*, 2020. 82:106291.
 162. Elsacker, E., Vandelook, S., Brancart, J., Peeters, E. and Laet, L. D. Mechanical, physical and chemical characterisation of mycelium-based composites with different types of lignocellulosic substrates. *PLoS One*, 2019 14(7):0213954.
 163. Yoshida, S., Tamura, M., Okada, A., Sakamoto, E. and Yamasawa, A. *High Strength Rock Wool and Process for Producing Same*. EP0754652B1. 2001.
 164. Širok, B., Blagojević, B. and Bullen, P. *Mineral Wool: Production and Properties*. England: Cambridge International Science Publishing. 2008.
 165. Medeiros, M. G., Nadaleti, W. C., Rocha, J. C., Cheriaf, M., Gleise, P.J. P. and de Castilhos Jr., A. B. A cleaner material production by the incorporation of the rockwool waste into portland cement matrices. *Journal of Cleaner Production*, 2021. 293:126059.
 166. Kinnunen, P., Yliniemi, J., Talling, B. and Illikainen, M. Rockwool waste in fly ash geopolymers composites. *Journal of Material Cycles and Waste Management*, 2017. 19(3):1220–1227.

167. Mróz, K., Hager, I. and Korniejenko, K. Material Solutions for Passive Fire Protection of Buildings and Structures and Their Performances Testing. *Procedia Engineering*, 2016. 151:284–291.
168. Llantoy, N., Chàfer, M. and Cabeza, L. F. A comparative life cycle assessment (LCA) of different insulation materials for buildings in the continental Mediterranean climate. *Energy and Buildings*, 2020. 225:110323.
169. Väntsi, O. and Kärki, T. Mineral wool waste in Europe: a review of mineral wool waste quantity, quality, and current recycling methods. *Journal of Materials Cycles and Waste Management*, 2014. 16:62–72.
170. Savvas, D. and Lenz, F., Effects of NaCl or nutrient-induced salinity on growth, yield, and composition of eggplants grown in rockwool. *Scientia Horticulturae*, 2000. 84:37–47.
171. Awang, Y. B. and Atherton, J. G. Growth and fruiting responses of strawberry plants grown on rockwool to shading and salinity. *Scientia Horticulturae*, 1995, 62:25–31.
172. Acuña, R. A., Bonachela, S., Magán, J. J., Marfà, O., Hernández, J. H. and Cáceres, R. Reuse of rockwool slabs and perlite grow-bags in a low-cost greenhouse: Substrates' physical properties and crop production. *Scientia Horticulturae*, 2013. 160:139–147.
173. Bonachela, S., Quesada, J., Acuña, R. A., Magán, J. J. and Marfà, O. Oxyfertigation of a greenhouse tomato crop grown on rockwool slabs and irrigated with treated wastewater: Oxygen content dynamics and crop response. *Agricultural Water Management*, 2010. 97(3):433–438.
174. Cheng, A., Lin, W. T. and Huang, R. Application of rock wool waste in cement-based composites. *Materials and Design*, 2011. 32(2):636–642.
175. Kubiliute, R., Kaminskas, R. and Kazlauskaite, A. Mineral wool production waste as an additive for Portland cement. *Cement and Concrete Composites*, 2018. 88:130–138.

176. Ramírez, C. P., Merino, M. d. R., Arrebola, C. V, Barriguete, A. V. and Kosior-Kazberuk, M. Analysis of the mechanical behaviour of the cement mortars with additives of mineral wool fibres from recycling of CDW. *Construction and Building Materials*, 2019. 210:56–62.
177. Lin, W. T., Cheng, A., Huang, R. and Zou, S. Y. Improved microstructure of cement-based composites through the addition of rock wool particles. *Materials Characterization*, 2013. 84:1–9.
178. Väntsi, O. and Kärki, T. Different coupling agents in wood-polypropylene composites containing recycled mineral wool: A comparison of the effects. *Journal of Reinforced Plastics and Composites*, 2015. 34(11):879–895.
179. Keskisaari, A., Butylina, S. and Kärki, T. Use of construction and demolition wastes as mineral fillers in hybrid wood-polymer composites. *Journal of Applied Polymer Science*, 2016. 133(19):43412.
180. Seraji, M. M., Kianersi, S., Hosseini, S.H., Davarpanah, J. and Elahi, S. Performance evaluation of glass and rock wool fibers to improve thermal stability and mechanical strength of monolithic phenol-formaldehyde based carbon aerogels. *Journal of Non-Crystalline Solids*, 2018. 491:89–97.
181. Laoutid, F., Bonnaud, L., Alexandre, M., Lopez-Cuesta, J.-M. and Dubois, Ph. New prospects in flame retardant polymer materials: From fundamentals to nanocomposites. *Material Science and Engineering R*, 2009. 63(3):100–125.
182. Levchik, S. V. Introduction to Flame Retardancy and Polymer Flammability. In: Morgan A. B. and Wilkie C. A. eds. *Flame Retardant Polymer Nanocomposites*. New Jersey: John Wiley & Son. Ltd. 1–29; 2007.
183. Hu, Y. and Wang, X. Introduction. In: Hu Y. and Wang X. eds. *Flame Retardant Polymer Materials – A Handbook*. Florida: CRC Press. 3–12; 2019.
184. Irvine, D. J., McCluskey, J. A. and Robinson, I. M. Fire Hazards and some common polymers. *Polymer Degradation and Stability*, 2000. 67(3):383–396.

185. Morgan, A. B. and Gilman, J. W. An overview of flame retardancy of polymeric materials: application, technology, and future directions. *Fire and Materials*, 2013. 37(4):259–279.
186. Lewin, M. Physical and Chemical Mechanisms of Flame Retarding of Polymers. In: Le Bras, M., Camino, G., Bourbigot, S. and Delobel, R. eds. *Fire Retardancy of Polymers: The Use of Intumescence*. Cambridge: Athenaeum Press Ltd. 3–34; 1998.
187. Lewin, M. and Weil, E. D. Mechanisms and modes of action in flame retardancy of polymers. In: Horrocks, A. R. and Price, D. eds. *Fire Retardant Materials*. Cambridge: Woodhead Publishing Limited. 31–68; 2001.
188. Le Bras, M. and Bourbigot, S. Fire Retarded Intumescent Thermoplastics Formulations, Synergy and Synergistic Agents – A Review. In: Le Bras, M., Camino, G., Bourbigot, S. and Delobel, R. eds. *Fire Retardancy of Polymers: The Use of Intumescence*. Cambridge: Athenaeum Press Ltd. 64–75; 1998.
189. Patel, R. H. and Patel, K. S. Synthesis and characterization of flame retardant hyperbranched polyurethanes for nano-composite and nano-coating applications. *Progress in Organic Coatings*, 2015. 88:283–292.
190. Yew, M. C., Sulong, N. H. R., Yew, M. K., Amalina, M. A. and Johan, M. R. Influences of flame-retardant fillers on fire protection and mechanical properties of intumescent coatings. *Progress in Organic Coatings*, 2015. 78:59–66.
191. Hamdani-Devarennesa, S., El Hage, R., Dumazert, L., Sonnier, R., Ferry, L., Lopez-Cuesta, J.–M. and Bert, C. Water-based flame retardant coating using nano-boehmite for expanded polystyrene (EPS) foam. *Progress in Organic Coatings*, 2016. 99:32–46.
192. Kandola, B. K., Bhatti, W. and Kandare, E. A comparative study on the efficacy of varied surface coatings in fireproofing glass/epoxy composites. *Polymer Degradation and Stability*, 2016. 97(11):2418–2427.

193. Liu, Z., Dai, M., Zhang, Y., Gao, X. and Zhang, Q. Preparation and performances of novel waterborne intumescence fire-retardant coatings. *Progress in Organic Coatings*, 2016. 95:100–106.
194. Kurańska, M., Cabulis, U., Auguścik, M., Prociak, A., Ryszkowska, J. and Kirpluks, M. Bio-based polyurethane-polyisocyanurate composites with an intumescence flame retardant. *Polymer Degradation and Stability*, 2016. 127:11–19.
195. Ciecielska, E., Jurczyk-Kowalska, M., Bazarnik, P., Gloc, M., Kulesza, M., Kowalski, M., Krauze, S. and Lewandowska, M. Flammability, mechanical properties and structure of rigid polyurethane foams with different types of carbon reinforcing materials. *Composite Structures*, 2016. 140:67–76.
196. Wang, W., Pan., Y., Pan, H., Yang, W., Liew, K. M., Song, L. and Hu, Y. Synthesis and characterization of MnO₂ nanosheets based multilayer coating and applications as a flame retardant for flexible polyurethane foam. *Composite Science and Technology*, 2016. 123:212–221.
197. Zhang, L., Zhang, M., Zhou, Y. and Hu, L. The study of mechanical behavior and flame retardancy of castor oil phosphate-based rigid polyurethane foam composites containing expanded graphite and triethyl phosphate. *Polymer Degradation and Stability*, 2013. 98(12):2784–2794.
198. Naveen, A. N. and Manoj, N. Rheological and Thermal Analysis of Polystyrene–Kaolin Nanocomposite Prepared by Solution Intercalation Technique. *Procedia Technology*, 2016. 24:749–753.
199. Kandola, B. K. Nanocomposites. In: Horrocks, A. R. and Price, D. eds. *Fire retardant materials*. Cambridge: Woodhead Publishing Limited. 31–68; 2001.
200. Lan, T. and Beyer, G. Introduction to flame retardancy of polymer-clay nanocomposites. In: Mittal, V. ed. *Thermally Stable and Flame Retardant Polymer Nanocomposites*. New York: Cambridge University Press. 161–185; 2011.

201. Zheng, X., Wang, G. and Xu, W. Roles of organically-modified montmorillonite and phosphorus flame retardant during the combustion of rigid polyurethane foam. *Polymer Degradation and Stability*, 2014. 101:32–39.
202. Lee, S. K., Bai, B. C., Im, J. S., In, S. J. and Lee, Y.-S. Flame retardant epoxy complex produced by addition of montmorillonite and carbon nanotube. *Journal of Industrial and Engineering Chemistry*, 2010. 16(6):891–895.
203. Kashiwagi, T., Harris Jr., R. H., Zhang, X., Briber, R. M., Cipriano, B. H., Raghavan, S. R., Awad, W. H. and Shields, J. R. Flame retardant mechanism of polyamide 6-clay nanocomposites. *Polymer*, 2004. 45(3):881–891.
204. Gaan, S., Liang S., Mispreuve, H., Perler, H., Naescher, R. and Neisius, M. Flame retardant flexible polyurethane foams from novel DOPO-phosphonamidate additives. *Polymer Degradation and Stability*, 2015. 113:180–188.
205. Zhang, L., Zhang, M., Hu, L. and Zhou, Y. Synthesis of rigid polyurethane foams with castor oil-based flame retardant polyols. *Industrial Crops and Products*, 2014. 52:380–388.
206. Hejna, A., Kirpluks, M., Kosmela, P., Cabulis, U., Haponiuk, J. and Piszczyk, Ł. The influence of crude glycerol and castor oil-based polyol on the structure and performance of rigid polyurethane-polyisocyanurate foams. *Industrial Crops and Products*, 2017. 95:113–125.
207. Heinen, M., Gerbase, A. E. and Petzhold, C. L. Vegetable oil-based rigid polyurethanes and phosphorylated flame retardants derived from epoxidized soybean oil. *Polymer Degradation and Stability*, 2014. 108: 76–86.
208. Datta, S., Roy, S. and Davim, J. P. Optimization Techniques: An Overview. In: Datta, S. and Davim, J. P. eds. *Optimization in Industry: Present Practices and Future Scope*. Switzerland: Springer Nature. 1–11; 2019.
209. Krause, F. -L., Kind, C. and Biantoro, C. The Application of a Statistical Design of Experiment for Quantitative Analysis and Optimisation of Development Process. In: Krause, F.-L. *The Future of Product Development –*

Proceedings of the 17th CIRP Design Conference. Berlin: Springer. 493–502; 2007.

210. Sarafrazi, M., Hamadanian, M., and Ghasemi, A. R. Optimize epoxy matrix with RSM/CCD method and influence of multi-wall carbon nanotube on mechanical properties of epoxy/polyurethane. *Mechanics of Materials*, 2019. 138:103154.
211. Bahraeian, S. *Polypyrrole-Palladium Coated Poly(Styrene-Butyl Acrylate) Nanocomposite Emulsion*. Ph.D. thesis. Universiti Teknologi Malaysia; 2016.
212. Myers, R. H., Montgomery, D. C. and Anderson-Cook, C. M. *Response Surface Methodology - Process and Product Optimization Using Designed Experiments*. 4th ed. New Jersey: John Wiley & Son. Ltd. 2016.
213. Tajulruddin, W. N. W. *Synthesis and Characterization of Epoxidized Palm Oil Hydrogel*. Master thesis. Universiti Teknologi Malaysia; 2016.
214. Soundhar, A., Rajesh, M., Jayakrishna, K., Sultan, M. T. H. and Shah, A. U. M. Investigation on mechanical properties of polyurethane hybrid nanocomposite foams reinforced with roselle fibers and silica nanoparticles. *Nanocomposites*, 2019. 5(1):1–12.
215. Allan, D., Daly, J. H. and Liggat, J. J. Thermal volatilisation analysis of a TDI-based flexible polyurethane foam containing ammonium polyphosphate. *Polymer Degradation and Stability*, 2014. 102:170–179.
216. Li, Q., Wang, J., Chen, L., Shi, H. and Hao, J. Ammonium polyphosphate modified with β -cyclodextrin crosslinking rigid polyurethane foam: Enhancing thermal stability and suppressing flame spread. *Polymer Degradation and Stability*, 2019. 161:166–174.
217. Gao, L., Zheng, G., Zhou, Y., Hu, L. and Feng, G. Thermal performances and fire behaviors of rosin-based rigid polyurethane foam nanocomposites. *Journal of Thermal Analysis and Calorimetry*, 2015. 119:411–424.

218. Gao, L., Zheng, G., Zhou, Y., Hu, L. and Feng, G. Improved mechanical property, thermal performance, flame retardancy and fire behavior of lignin-based rigid polyurethane foam nanocomposite. *Journal of Thermal Analysis and Calorimetry*, 2015. 120:1311–1325.
219. Zhao, X. -L., Chen, C. -K. and Chen, X. -L. Effects of carbon fibers on the flammability and smoke emission characteristics of halogen-free thermoplastic polyurethane/ammonium polyphosphate. *Journal of Materials Science*, 2016. 51:3762–3771.
220. Yi, D. and Yang, R. Ammonium Polyphosphate/Montmorillonite Nanocompounds in Polypropylene. *Journal of Applied Polymer Science*, 2010. 118(2):834–840.
221. Piszczek, Ł., Danowska, M., Miatlarek-Kropidłowska, A., Szyszka, M. and Strąkowski, M. Synthesis and thermal studies of flexible polyurethane nanocomposite foams obtained using nanoclay modified with flame retardant compound. *Journal of Thermal Analysis and Calorimetry*, 2014. 118:901–909.
222. Gelest Incorporated. *Sodium Montmorillonite Clay Safety Data Sheet*. SIS6985.0. Pennsylvania, 2015.
223. Novista Group Limited. *Crystalline II Ammonium Polyphosphate ProFlame APP 801 Technical Data Sheet*. Shandong, 2017.
224. Novista Group Limited. *Silane Treated Ammonium Polyphosphate II ProFlame APP 802 Technical Data Sheet*. Shandong, 2017.
225. Sigma-Aldrich. *Fumed Silica Product Information*. Missouri, 2008.
226. American Society for Testing and Materials. *Standard Test Method for Measuring the Minimum Oxygen Concentration to Support Candle-Like Combustion of Plastics (Oxygen Index)*. Pennsylvania, USA. D2863. 2000.
227. American Society for Testing and Materials. *Standard Test Method for Measuring the Comparative Burning Characteristics of Solid Plastics in a Vertical Position*. Pennsylvania, USA. D3801. 2000.

228. Song, S. A., Lee, Y., Kim, Y. S. and Kim, S. S. Mechanical and thermal properties of carbon foam derived from phenolic foam reinforced with composite particles. *Composite Structures*, 2017. 173:1–8.
229. American Society for Testing and Materials. *Standard Test Method for Open Cell Content of Rigid Cellular Plastic*. Pennsylvania, USA. D6226. 2005.
230. American Society for Testing and Materials. *Standard Test Method for Apparent Density of Rigid Cellular Plastics*. Pennsylvania, USA. D1622. 2008.
231. American Society for Testing and Materials. *Standard Test Method for Compressive Properties of Rigid Cellular Plastics*. Pennsylvania, USA. D1621. 2010.
232. Mistry, B. D. *A Handbook of Spectroscopy Data Chemistry*. Jaipur: Oxford Book Company. 2009.
233. Yadav, L. D. S. *Organic Spectroscopy*. Dordrecht: Springer-Science. 2005.
234. Khairuddin, Pramono, E., Utomo, S. B., Wulandari V., Zahrotul A. W. and Clegg, F. FTIR studies on the effect of concentration of polyethylene glycol on polymerization of Shellac. *Journal of Physics: Conference Series*, 2016. 776(1): 012053.
235. Paciorek-Sadowska, J., Borowicz, M., Czupryński, B., Tomaszewska, E. and Liszkowska, J. *Oenothera biennis* seed oil as an alternative raw material for production of bio-polyol for rigid polyurethane-polyisocyanurate foams. *Industrial Crops and Products*, 2018. 126:208–217.
236. Toldy, A., Harakály, Gy., Szolnoki, B., Zimonyi, E. and Marosi, Gy. Flame retardancy of thermoplastics polyurethanes. *Polymer Degradation and Stability*, 2012. 97(12):2524–2530.
237. Papa, A. J. and Proops, W. R. A. New Mechanism of Flame Retardancy for Polyurethanes. *Journal of Applied Polymer Science*, 1973. 17(8):2463–2483.

238. Chattopadhyay, D. K. and Webster, D. C. Thermal stability and flame retardancy of polyurethanes. *Progress in Polymer Science*, 2009. 34(10):1068–1133.
239. Günther, M., Levchik, S. V. and Schartel, B. Bubbles and collapses: Fire phenomena of flame-retarded flexible polyurethane foams. *Polymers for Advanced Technologies*, 2020. 31(10):2185–2198.
240. Zhang, M. Luo, Z., Zhang, J., Chen, S. and Zhou, Y. Effects of a novel phosphorus-nitrogen flame retardant on rosin-based rigid polyurethane foams. *Polymer Degradation and Stability*, 2015. 120:427–434.
241. Modesti, M. Lorenzetti, A. Besco, S., Hrelja, D., Semenzato, S., Bertani, R. and Michelin, R. A. Synergism between flame retardant and modified layered silicate on thermal stability and fire behaviour of polyurethane nanocomposite foams. *Polymer Degradation and Stability*, 2008. 93(12):2166–2171.
242. Wang, C., Zheng, Y., Xie, Y., Qiao, K., Sun, Y. and Yue, L. Synthesis of bio-castor oil polyurethane flexible foams and the influence of biotic component on their performance. *Journal of Polymer Research*, 2015. 22:145.
243. Carriço, C. S., Fraga, T., Carvalho, V. E. and Pasa, V. M. D. Polyurethane Foams for Thermal Insulation Uses Produced from Castor Oil and Crude Glycerol Biopolyols. *Molecules*, 2017. 22(7):1091.
244. Trovati, G., Natali, M. V. S., Sanches, E. A., Campelo, P. H., Neto, R. B., Neto, S. C. and Trovati, L. R. Production and characterization of polyurethane castor oil (*Ricinus communis*) foam for nautical fender. *Polymer Testing*, 2019. 73:87–93.
245. Ibrahim, S., Ahmad, A. and Mohamed, N. S. Synthesis and characterization of castor oil-based polyurethane for potential application as host in polymer electrolytes. *Bulletin of Materials Science*, 2015. 38(5):1155–1161.
246. Amado, J. C. Q. Thermal Resistance Properties of Polyurethanes and Its Composites. In: Evingür, G. A., Pekcan, Ö. and Achilias, D. eds. *Thermosoftening Plastics*. IntechOpen. 2020.

247. Guo, A., Javni, I. and Petrović, Z. Rigid polyurethane foams based on soybean oil. *Journal of Applied Polymer Science*, 2000. 77(2):467–473.
248. Prociak, A., Malewska, E., Kurańska, M., Bąk, S. and Budny, P. Flexible polyurethane foams synthesized with palm oil-based bio-polyols obtained with the use of different oxirane ring opener. *Industrial Crops and Products*, 2018. 115:69–77.
249. Artavia, L. D. and Macosko, C. W. Polyurethane flexible foam formation. In: Hilyard, N. C. and Cunningham, A. eds. *Low Density Cellular Plastics*. Dordrecht: Springer-Science. 22–55; 1994.
250. Priester Jr., R. D. and Turner, R. B. The morphology of flexible polyurethane matrix polymers. In: Hilyard, N. C. and Cunningham, A. eds. *Low Density Cellular Plastics*. Dordrecht: Springer-Science. 78–103; 1994.
251. Gharehbagh, A. and Ahmadi, Z. Polyurethane Flexible Foam Fire Behavior. In: Zafar, F. and Sharmin, E. eds. *Polyurethane*. Rijeka, Croatia: IntechOpen. 101–120; 2012.
252. Rashmi, B. J., Rusu, D., Prashanta, K., Lacrampe, M. F. and Krawczak, P. Development of water-blown bio-based thermoplastic polyurethane foams using bio-derived chain extender. *Journal of Applied Polymer Science*, 2013. 128(1):292–303.
253. Palanisamy, A. Water-blown polyurethane–clay nanocomposite foams from biopolyol—effect of nanoclay on the properties. *Polymer Composites*, 2013. 34(8):1306–1312.
254. Zhang, S., Xiang, A., Tian, H. and Rajulu, A. V. Water-Blown Castor Oil-Based Polyurethane Foams with Soy Protein as a Reactive Reinforcing Filler. *Journal of Polymers and the Environment*, 2018. 26:15–22.
255. Tuan Ismail, T. N. M., Ibrahim, N. A., Noor, M. A. M., Soi, S. H., Palam, K. D. P., Kian, Y. K., Idris, Z., Schiffman, C. M., Sendijarevic, I., Malek, E. A., Zainuddin, N. and Sendijarevic, V. Oligomeric composition of palm olein-

- based polyols: The effect of nucleophiles. *European Journal of Lipid Science and Technology*, 2017. 120(4):1700354.
256. Pfister, D. P., Xia, Y. and Larock, R. C. Recent advances in vegetable oil-based polyurethanes. *ChemSusChem*, 2011. 4(6):703–717.
257. Oprea, S. and Potolinca, V. O. Synthesis and characterization of novel linear and cross-linked polyurethane urea elastomers with 2,3-diaminopyridine in the main chain. *High Performance Polymers*, 2013. 25(2):147–155.
258. Xu, Z., Tang, X., Gu, A. and Fang, Z. Novel Preparation and Mechanical Properties of Rigid Polyurethane Foam/Organoclay Nanocomposites. *Journal of Applied Polymer Science*, 2007. 106(1):439–447.
259. Nguyen, Q. T. and Baird, D. G. Preparation of Polymer–Clay Nanocomposites and Their Properties. *Advances in Polymer Technology*, 2006. 25(4):270–285.
260. Sarier, N. and Onder, E. Organic modification of montmorillonite with low molecular weight polyethylene glycols and its use in polyurethane nanocomposite foams. *Thermochimica Acta*, 2010. 510(1–2):113–121.
261. Xu, Z. -B., Kong, W. -W., Zhou, M. -X. and Peng, M. Effect of Surface Modification of Montmorillonite on The Properties of Rigid Polyurethane Foam Composites. *Chinese Journal of Polymer Science*, 2010. 28(7):615–624.
262. Zou, J., Lei, Y., Liang, M. and Zou, H. Effect of nano-montmorillonite as cell opener on cell morphology and resilient performance of slow-resilience flexible polyurethane foams. *Journal of Polymer Research*, 2015. 22:201.
263. Ray, S. S. and Kuruma, M. *Halogen-Free Flame-Retardant Polymers*. Cham, Switzerland: Springer-Nature. 2020.
264. Beyer, G. Flame Retardant Properties of Organoclays and Carbon Nanotubes and Their Combinations with Alumina. In: Morgan, A. B. and Wilkie, C. A. eds. *Flame Retardant Polymer Nanocomposites*. New Jersey: John Wiley & Sons. Ltd. 163–190; 2007.

265. Kiliaris, P. and Papaspyrides, C. D. Polymer/layered silicate (clay) nanocomposites: An overview of flame retardancy. *Progress in Polymer Science*, 2010. 35(7):902–958.
266. Wen, P., Wang, D., Liu, J. J., Zhan, J., Hu, Y. and Yuen, R. K. K. Organically modified montmorillonite as a synergist for intumescence flame retardant against the flammable polypropylene. *Polymers for Advanced Technologies*, 2017. 28(6):679–685.
267. Prociak, A., Kurańska, M., Cabulis, U., Ryszkowska, J., Leszczyńska, M., Uram, K. and Kirpluks, M. Effect of bio-polyols with different chemical structures on foaming of polyurethane systems and foam properties. *Industrial Crops and Products*, 2018. 120:262–270.
268. Wilkie, A. A. and Morgan, A. B. Nanocomposites I: Current developments in nanocomposites as novel flame retardants. In: Horrocks, A. R. and Price, D. eds. *Advances in Fire Retardant Materials*. Cambridge: Woodhead Publishing Limited. 95–123; 2008.
269. Lecouvet, B., Sclavons, M., Bailly, C. and Bourbigot, S. A comprehensive study of the synergistic flame retardant mechanisms of halloysite in intumescent polypropylene. *Polymer Degradation and Stability*, 2013. 98(11):2268–2281.
270. Bhattacharya, S. S. and Aadhar, M. Studies on Preparation and analysis of Organoclay Nano Particles. *Research Journal of Engineering Sciences*, 2014. 3(3):10–16.
271. Wang, J., Sun, K., Hao, W., Du, Y. and Pan, C. Structure and properties research on montmorillonite modified by flame-retardant dendrimer. *Applied Clay Science*, 2014. 90:109–121.
272. Nguyen, D. L., Le, S. H., Malin, M., Weisser, J., Walter, T., Schnabelrauch, M. and Seppälä, J. Synthesis and characterization of castor oil-segmented thermoplastic polyurethane with controlled mechanical properties. *European Polymer Journal*, 2016. 81:129–137.

273. Rihayat, T., Saari, M., Suraya, A. R., Mahmood, M. H., Dahlan, K. Z. H. M., Yunus, W. M. Z. W. and Sapuan, S. M. Synthesis and thermal characterization of polyurethane/clay nanocomposites based on palm oil polyol. *Polymer - Plastics Technology and Engineering*, 2006. 45(12):1323–1326.
274. Lorusso, C., Vergaro, V., Conciauro, C., Ciccarella, G. and Congedo, P. M. Thermal and mechanical performance of rigid polyurethane foam added with commercial nanoparticles. *Nanomaterials and Nanotechnology*, 2017. 7:1–9.
275. Anadão, P. The Use of Montmorillonite Clay in Polymer Nanocomposite Foams. In: Mittal, V. ed. *Polymer Nanocomposite Foams*. Florida: CRC Press. 149–168; 2014.
276. Li, Y., Chen, Z. and Zeng, C. Poly(Methyl Methacrylate) (PMMA) Nanocomposite Foams. In: Mittal, V. ed. *Polymer Nanocomposite Foams*. Florida: CRC Press. 1–33; 2014.
277. Lee, L. J., Zeng, C., Cao, X., Han, X., Shen, J. and Xu, G. Polymer nanocomposite foams. *Composites Science and Technology*, 2005. 65(15–16):2344–2363.
278. Estravís, S., Tirado-Mediavilla, J., Santiago-Calvo, M., Ruiz-Herrero, J. L., Villafaña, F. and Rodríguez-Pérez, M. Á. Rigid polyurethane foams with infused nanoclays: Relationship between cellular structure and thermal conductivity. *European Polymer Journal*, 2016. 80:1–15.
279. Panda, S. S., Kamal, S. K., Mohanty, S. and Nayak, S. K. Preparation, characterization, and properties of castor oil-based flexible polyurethane/Cloisite 30B nanocomposites foam. *Journal of Composite Materials*, 2017. 52(4):531–542.
280. Zhang, C., Tong, X., Deng, C., Wen, H., Huang, D., Guo, Q. and Liu, X. The foaming dynamic characteristics of polyurethane foam. *Journal of Cellular Plastics*, 2020. 56(3):279–285.

281. Zou, J., Chen, Y., Liang, M. and Zou, H. Effect of hard segments on the thermal and mechanical properties of water blown semi-rigid polyurethane foams. *Journal of Polymer Research*, 2015. 22:120.
282. Liu, G., Liu, X. and Yu, J. Ammonium Polyphosphate with Crystalline Form V by Ammonium Dihydrogen Phosphate Process. *Industrial and Engineering Chemistry Research*, 2010. 49(12):5523–5529.
283. Levchik, S. Phosphorus-based FRs. In: Morgan, A. B. and Wilkie, C. A. eds. *The Non-Halogenated Flame Retardant Handbook*. New Jersey: Wiley. 17–74; 2014.
284. Kishore, K. and Mohandas, K. Effect of ammonium halides on the combustion of polystyrene. *Journal of Fire Sciences*, 1983. 1(2): 81–95.
285. Fink, J. *Reactive Polymers Fundamentals and Applications*. Oxford, United Kingdom: Elsevier. 2013.
286. Chapple, S. and Anadjiwala, R. Flammability of Natural Fiber-reinforced Composites and Strategies for Fire Retardancy: A Review. *Journal of Thermoplastic Composite Materials*, 2010. 23(6):871–893.
287. Hörold, S. Phosphorus-based and Intumescent Flame Retardants. In: Kiliaris, P. and Papaspyrides, C. D. eds. *Polymer Green Flame Retardant*. Oxford: Elsevier. 221–254; 2014.
288. Reshetnikov, I. S., Yablokova, M. Y. and Khalturinskij, N. A. Intumescent Chars. In: Le Bras, M., Camino, G., Bourbigot, S. and Delobel, R. eds. *Fire Retardancy of Polymers - The Use of Intumescence*. Cambridge: Athenaeum Press Ltd. 88–103; 1998.
289. Liu, J. -C., Xu, M. -J., Lai, T. and Li, B. Effect of Surface-Modified Ammonium Polyphosphate with KH550 and Silicon Resin on the Flame Retardancy, Water Resistance, Mechanical and Thermal Properties of Intumescent Flame Retardant Polypropylene. *Industrial and Engineering Chemistry Research*, 2015. 54(40):9733–9741.

290. Kilinc, M. Silicon Based Flame Retardants. In: Morgan, A. B. and Wilkie, C. A. eds. *The Non-Halogenated Flame Retardant Handbook*. New Jersey: Wiley. 169–199; 2014.
291. Lei, Z., Cao, Y., Xie, F. and Ren, H. Study on Surface Modification and Flame Retardants Properties of Ammonium Polyphosphate for Polypropylene. *Journal of Applied Polymer Science*, 2011. 124(1):781–788.
292. Xing, W. and Wang, J. Recent Advances in Silicon-Containing Flame Retardants. In: Hu, Y. and Wang, X. eds. *Flame Retardant Polymeric Materials – A Handbook*. Florida: CRC Press. 77–95; 2019.
293. Shi, Y., Xing, W., Wang, B., Hong, N., Zhu, Y., Wang, C., Gui, Z., Yuen, R. K. K. and Hu, Y. Synergistic effect of graphitic carbon nitride and ammonium polyphosphate for enhanced thermal and flame retardant properties of polystyrene. *Materials Chemistry and Physics*, 2016. 177:283–292.
294. Zhu, F., Xin, Q., Feng, Q., Liu, R. and Li, K. Influence of nano-silica on flame resistance behavior of intumescent flame retardant cellulosic textiles: Remarkable synergistic effect? *Surface and Coatings Technology*, 2016. 294:90–94.
295. Jiang, M., Zhang, Y., Yu, Y., Zhang, Q., Huang, B., Chen, Z., Chen, T. and Jiang, J. Flame retardancy of unsaturated polyester composites with modified ammonium polyphosphate, montmorillonite, and zinc borate. *Journal of Applied Polymer Science*, 2019. 136(11):47180.
296. Chen, M., Tang, M., Qi, F., Chen, X. and He, W. Microencapsulated ammonium polyphosphate and its application in the flame retardant polypropylene composites. *Journal of Fire Sciences*, 2015. 33(5):374–389.
297. Decsov, K., Bocz, K., Szolnoki, B., Bourbigot, S., Fontaine, G., Vadas, D. and Marosi, G. Development of Bioepoxy Resin Microencapsulated Ammonium-Polyphosphate for Flame Retardancy of Polylactic Acid. *Molecules*, 2019. 24(22):4123.

298. Saba, N., Paridah, M. T., Abdan, K. and Ibrahim, N. A. Physical, structural and thermomechanical properties of oil palm nano filler/kenaf/epoxy hybrid nanocomposites. *Materials Chemistry and Physics*, 2016. 184:64–71.
299. Sachse, S., Poruri, M., Silva, F., Michalowski, S., Pielichowski, K. and Njuguna, J. Effect of nanofillers on low energy impact performance of sandwich structures with nanoreinforced polyurethane foam cores. *Journal of Sandwich Structures and Materials*, 2014. 16(2):173–194.
300. Thirumal, M., Khastgir, D. Singha, N. K., Manjunath, B. S. and Naik, Y. P. Effect of Foam Density on the Properties of Water Blown Rigid Polyurethane Foam. *Journal of Applied Polymer Science*, 2008. 108(3):1810–1817.
301. Li, K. -M., Jiang, J. -G., Tian, S. -C., Chen, X. -J. and Yan, F. Influence of Silica Types on Synthesis and Performance of Amine – Silica Hybrid Materials Used for CO₂ Capture. *The Journal of Physical Chemistry C*, 2014. 118(5):2454–2462.
302. Monteiro, M. S. d. S. d. B., Neto, R. P. C., Santos, I. C. S., da Silva, E. O. and Tavares, M. I. B. Inorganic-Organic Hybrids Based on Poly (ε -Caprolactone) and Silica Oxide and Characterization by Relaxometry Applying Low-field NMR. *Materials Research*, 2012. 15(6):825–832.
303. Salahshoori, I., Nasirian, D., Rashidi, N., Hossain, M. K., Hatami, A. and Hassanzadeganroudsari, M. The effect of silica nanoparticles on polysulfone–polyethylene glycol (PSF/PEG) composite membrane on gas separation and rheological properties. *Polymer Bulletin*, 2020.
304. Fambri, L., Dabrowska, I., Ceccato, R. and Pegoretti, A. Effects of Fumed Silica and Draw Ratio on Nanocomposite Polypropylene Fibers. *Polymers*, 2017. 9(2):41.
305. Chen, C. and Morgan, A. B. Mild processing and characterization of silica epoxy hybrid nanocomposite. *Polymer*, 2009. 50(26):6265–6273.

306. Chen, L., Wang, X., Jia, Z., Luo, Y. and Jia, D. Use of precipitated silica with silanol groups as an inorganic chain extender in polyurethane. *Materials and Design*, 2015. 87:324–330.
307. Alexandru, M., Cazacu, M., Vlad, S. and Iacomi, F. Polydimethylsiloxane – silica Composites. Influence of the Silica on the Morphology and the Surface, Thermal, Mechanical Properties. *High Performance Polymers*, 2009. 21(4):379–392.
308. Hopmann, C. and Latz, S. Foaming technology using gas counter pressure to improve the flexibility of foams by using high amounts of CO₂ as a blowing agent. *Polymer*, 2015. 56:29–36.
309. Francés, A. B. and Bañón, M. V. N. Effect of silica nanoparticles on polyurethane foaming process and foam properties. *IOP Conference Series: Materials Science and Engineering*, 2014. 64(1):012020.
310. de Vries, D. V. W. M. *Characterization of polymeric foams*. Bachelor's thesis. Eindhoven University of Technology; 2009.
311. Lobos, J. and Velankar, S. How much do nanoparticle fillers improve the modulus and strength of polymer foams? *Journal of Cellular Plastics*, 2016. 52(1):57–88.
312. Fan, J. and Chen, A. Studying a Flexible Polyurethane Elastomer with Improved Impact-Resistant Performance. *Polymers*, 2019. 11(3):467.
313. Trovati, G., Sanches, E. A., Neto, S. C., Mascarenhas, Y. P. and Chierice, G. O. Characterization of Polyurethane Resins by FTIR, TGA, and XRD. *Journal of Applied Polymer Science*, 2009. 115(1):263–268.
314. Hu, Y. and Zhang, Y. Mechanisms and Modes of Action in Flame Retardancy of Polymers. In: Hu, Y. and Wang, X. eds. *Flame Retardant Polymeric Materials - A Handbook*. Florida: CRC Press. 13–34; 2019.

315. Kandola, B. K. and Horrocks, A. C. Composites. In: Horrocks, A. R. and Price, D. eds. *Fire retardant materials*. Cambridge: Woodhead Publishing Limited. 182–203; 2001.
316. Kurańska, M., Barczewski, M., Uram, K., Lewandowski, K., Prociak, A. and Michałowski, S. Basalt waste management in the production of highly effective porous polyurethane composites for thermal insulating applications. *Polymer Testing*, 2019. 76:90–100.
317. Haupert, F. and Wetzel, B. Reinforcement of Thermosetting Polymers by the Incorporation of Micro- and Nanoparticles. In: Friedrich, K., Fakirov, S. and Zhang, Z. eds. *Polymer Composites: From Nano- to Macro-scale*. Boston, Massachusetts: Springer. 45–62; 2005.
318. Fellahi, S., Boukobbal, S. and Boudjenana, F. Study of the Effect of Fumed Silica on Rigid PVC Properties. *Journal of Vinyl Technology*, 1993. 15(1):17–21.
319. Hamedi, N., Hassanajili, S. and Sajedi, M. T. An Investigation of Mechanical Properties of Polyurethane Nanocomposites with Various Silicas: Experimental Study and Modeling of Finite Deformation Response. *Silicon*, 2018. 10(4):1243–1255.
320. Nikje, M. M. A., Garmarudi, A. B., Haghshenas, M. and Mazaheri, Z. Improving the Performance of Heat Insulation Polyurethane Foams by Silica Nanoparticles. In: Bittnar, Z., Bartos, P. J. M., Němeček, J., Šmilauer, V. and Zeman, J. eds. *Nanotechnology in Construction 3*. Berlin: Springer. 149–154; 2009.
321. Parvinzadeh, M., Moradian, S., Rashidi, A. and Yasdanshenas, M. -E. Surface characterization of polyethylene terephthalate/silica nanocomposites. *Applied Surface Science*, 2010. 256(9):2792-2802.
322. Xu, W., Wang, G. and Zheng, X. Research on highly flame-retardant rigid PU foams by combination of nanostructured additives and phosphorus flame retardants. *Polymer Degradation and Stability*, 2014. 111:142–150.

323. Ibeh, C. C. and Bubacz, M. Current trends in nanocomposite foams. *Journal of Cellular Plastics*, 2008. 44(6):493–515.
324. Cao, X., Lee, L. J., Widya, T. and Macosko, C. Polyurethane/clay nanocomposites foams: processing, structure and properties. *Polymer*, 2005. 46(3):775–783.
325. Madaleno, L., Pyrz, R., Crosky, A., Jensen, L. R., Rauhe, J. C. M., Dolomanova, V., Timmons, A. M. M. V. d. B., Pinto, J. J. C. and Norman, J. Processing and characterization of polyurethane nanocomposite foam reinforced with montmorillonite-carbon nanotube hybrids. *Composites Part A: Applied Science and Manufacturing*, 2013. 44:1–7.
326. Olad, A. and Rashidzadeh, A. Preparation and anticorrosive properties of PANI/Na-MMT and PANI/O-MMT nanocomposites. *Progress in Organic Coatings*, 2008. 62(3):293–298.
327. Sun, Y., Yuan, B., Shang, S., Zhang, H., Shi, Y., Yu, B., Qi, C., Dong, H., Chen, X. and Yang, X. Surface modification of ammonium polyphosphate by supramolecular assembly for enhancing fire safety properties of polypropylene. *Composites Part B: Engineering*, 2020. 181:107588.
328. Yan, H. -w., Wei, J. -l., Yin, B. and Yang, M. -b. Effect of the surface modification of ammonium polyphosphate on the structure and property of melamine-formaldehyde resin microencapsulated ammonium polyphosphate and polypropylene flame retardant composites. *Polymer Bulletin*, 2015. 72(11):2725–2737.
329. Mahesh, K. R. V., Murthy, H. N. R., Kumaraswamy, B. E., Raghavendra, N., Sridhar, R., Krishna, M., Pattar, N., Pal, R. and Sherigara, B. S. Synthesis and characterization of organomodified Na-MMT using cation and anion surfactants. *Frontiers of Chemistry in China*, 2011. 6:153–158.
330. Malicka, A. and Domka, L. Structural study of polyethylene/montmorillonite systems. *Acta Physica Polonica Series A*, 2008. 114(2):423–431.

331. Qin, S., Yao, Y., He, W., Yu, J., He, M., Xu, C., Xu, G. and Qin, Z. Effect of intercalation method and intercalating agent type on the structure of silane-grafted montmorillonite. *Clays and Clay Minerals*, 2013. 61:580–589.
332. Wang, B., Sheng, H., Shi, Y., Hu, W., Hong, N., Zeng, W., Ge, H., Yu, X., Song, L. and Hu Y. Recent advances for microencapsulation of flame retardant. *Polymer Degradation and Stability*, 2015. 113:96–109.
333. Fu, M., Zhang, Z., Wu, L., Zhuang, G., Zhang, S., Yuan, J. and Liao, L. Investigation on the co-modification process of montmorillonite by anionic and cationic surfactants. *Applied Clay Science*, 2016. 132–133:694–701.
334. Chen, D., Zhu, J. X., Yuan, P., Yang, S. J., Chen, T. -H. and He, H. P. Preparation and Characterization of Anion – Cation Surfactants Modified Montmorillonite. *Journal of Thermal Analysis and Calorimetry*, 2008. 94(3):841–848.
335. Hojiyev, R., Ersever, G., Karaağaoğlu, İ. E., Karakaş, F. and Boylu, F. Changes on montmorillonite characteristics through modification. *Applied Clay Science*, 2016. 127–128:105–110.
336. Piszczyk, Ł., Strankowski, M., Danowska, M., Hejna, A. and Haponiuk, J. T. Rigid polyurethane foams from a polyglycerol-based polyol. *European Polymer Journal*, 2014. 57:143–150.
337. Fridrihsone, A., Stirna, U., Lazdiņa, B., Misāne, M. and Vilsone, D. Characterization of polyurethane networks structure and properties based on rapeseed oil derived polyol. *European Polymer Journal*, 2013. 49(6):1204–1214.
338. Izadi-Vasafi, H., Sadeghi, G. M. M., Babaei, A. and Ghayoumi, F. A novel biodegradable polyurethane based on hydroxylated polylactic acid and tung oil mixtures. I. Synthesis, physicochemical and biodegradability characterization. *Fibers and Polymers*, 2016. 17(3):311–323.

339. Salahuddin, N., Abo-El-Enein, S. A., Selim, A. and Salah El-Dien, O. Synthesis and characterization of polyurethane/organo-montmorillonite nanocomposites. *Applied Clay Science*, 2010. 47(3–4):242–248.
340. Das, B., Konwar, U., Mandal, M. and Karak, N. Sunflower oil based biodegradable hyperbranched polyurethane as a thin film material. *Industrial Crops and Products*, 2013. 44:396–404.
341. Sipaut, C. S., Murni, S., Saalah, S., Hoon, T. C., Ibrahim, M. N. M., Rahman, I. A. and Abdullah, A. A. Synthesis and characterization of polyols from refined cooking oil for polyurethane foam formation. *Cellular Polymers*, 2012. 31(1):19–38.
342. Mizera, K. and Ryszkowska, J. Polyurethane elastomers from polyols based on soybean oil with a different molar ratio. *Polymer Degradation and Stability*, 2016. 132:21–31.
343. Alaa, M. A., Yusoh, K. and Hasany, S. F. Synthesis and characterization of polyurethane–organoclay nanocomposites based on renewable castor oil polyols. *Polymer Bulletin*, 2014. 72(1):1–17.
344. Rihayat, T., Saari, M., Mahmood, M., Wan Yunus, W. M. Z., Suraya, A. R., Dahlan, K. Z. H. M. and Sapuan, S. M. Mechanical Characterisation of Polyurethane/Clay Nanocomposites. *Polymer and Polymer Composites*, 2007. 15(8):647–652.
345. John, B. Structural and Physical Properties of Polyurethane Nanocomposites and Foams. In: Pandey, J. K., Reddy, K. R., Mohanty, A. K. and Misra, M. eds. *Handbook of Polymernanocomposites: Processing, Performance and Application. Volume A – Layered Silicates*. Berlin:Springer-Verlag. 341–359; 2014.
346. Marosi, Gy., Márton, A., Szép, A., Csontos, I., Keszei, S., Zimonyi, E., Toth, A., Almeras, X. and Le Bras, M. Fire retardancy effect of migration in polypropylene nanocomposites induced by modified interlayer. *Polymer Degradation and Stability*, 2003. 82(2):379–385.

347. Marney, D. C. O., Russell, L. J., Wu, D. Y., Nguyen, T., Cramm, D., Rigopoulos, N., Wright, N. and Greaves, M. The suitability of halloysite nanotubes as a fire retardant for nylon 6. *Polymer Degradation and Stability*, 2008. 93(10):1971–1978.
348. Ionescu, M., Radojčić, D., Wan, X., Shrestha, M. L., Petrović, Z. S. and Upshaw, T. A. Highly functional polyols from castor oil for rigid polyurethanes. *European Polymer Journal*, 2016. 84:736–749.
349. Wang, G. and Xu., W. Influence of caged bicyclic phosphate and CaCO₃ nanoparticles on char-forming property of PU rigid foams. *Polymer Degradation and Stability*, 2013. 98(11):2323–2330.
350. Zhang, P., Fan, H., Hu, K., Gu, Y., Chen, Y., Yan, J., Tian, S. and He, Y. Solvent-free two-component polyurethane conjugated with crosslinkable hydroxyl-functionalized ammonium polyphosphate: Curing behaviors, flammability and mechanical properties. *Progress in Organic Coatings*, 2018. 120:88–99.
351. Qian, X., Song, L., Wang, B., Hu, Y. and Yuen, R. K. K. Synthesis of organophosphorus modified nanoparticles and their reinforcements on the fire safety and mechanical properties of polyurea. *Materials Chemistry and Physics*, 2013. 139(2–3):443–449.
352. Seo, W. J., Sung, Y. T., Kim, S. B., Lee, Y. B., Choe, K. H., Choe, S. H., Sung, J. Y. and Kim, W. N. Effects of Ultrasound on the Synthesis and Properties of Polyurethane Foam/Clay Nanocomposites. *Journal of Applied Polymer Science*, 2006. 102(4):3764–3773.
353. Widya, T. and Macosko, C. W. Nanoclay-modified rigid polyurethane foam. *Journal of Macromolecular Science Part B: Physics*, 2005. 44(6):897–908.
354. Mondal, P. and Khakhar, D. V. Rigid Polyurethane – Clay Nanocomposite Foams: Preparation and Properties. *Journal of Applied Polymer Science*, 2007. 103(5):2802–2809.

355. Montgomery, D. C. *Design and Analysis of Experiments*. 9th ed. New Jersey: Wiley and Sons. 2017.
356. Dorraji, M. S. S., Rasoulifard, M. H., Shajeri, M., Ashjari, H. R. and Azizi, M. The role of prepared ZnO nanoparticles on improvement of mechanical and antibacterial properties of flexible polyurethane foams: experimental modeling. *Polymer Bulletin*, 2018. 75(4):1519–1533.
357. Valipour, F., Dehghan, S. F. and Hajizadeh, R. The effect of nano - and microfillers on thermal properties of Polyurethane foam. *International Journal of Environmental Science and Technology*, 2021.
358. Li, H., Xu, C., Yuan, Z. and Wei, Q. Synthesis of bio-based polyurethane foams with liquefied wheat straw: Process optimization. *Biomass and Bioenergy*, 2018. 111:134–140.
359. Kirpluks, M., Vanags, E., Abolins, A., Michalowski, S., Fridrihsone, A. and Cabulis, U. High Functionality Bio-Polyols from Tall Oil and Rigid Polyurethane Foams Formulated Solely Using Bio-Polyols. *Materials*, 2020. 13(8):1985.

LIST OF PUBLICATIONS

Indexed Journal

1. **Dzulkifli, M. H.**, Yahya, M. Y. and Majid, R. A. Rigid palm oil-based polyurethane foam reinforced with diamine-modified montmorillonite nanoclay. *IOP Conference Series: Materials Science and Engineering*, 2017. 204: 012024. <https://doi.org/10.1088/1757-899X/204/1012024>. (**Indexed by SCOPUS**)

Non-Indexed Conference Proceedings

1. **Dzulkifli, M. H.**, Yahya, M. Y. and Majid, R. A. Bio-based rigid polyurethane foam reinforced with diamine-modified montmorillonite. In *2017 International Conference on Bio and Nanotechnology (ICBN 2017)*.