

COMPARISON STUDY OF SILICA AND IRON OXIDE NANOPARTICLES AS
ENHANCED OIL RECOVERY AGENTS

HAMZA JEES ADAM

UNIVERSITI TEKNOLOGI MALAYSIA

COMPARISON STUDY OF SILICA AND IRON OXIDE NANOPARTICLES AS
ENHANCED OIL RECOVERY AGENTS

HAMZA JEES ADAM

A project report submitted in partial fulfilment of the
requirements for the award of the degree of
Master of Petroleum Engineering

School of Chemical and Energy Engineering
Faculty of Engineering
Universiti Teknologi Malaysia

February 2021

Acknowledgment

Firstly, in the name of Allah the most gracious, and most merciful. Alhamdulillah, thanks to Allah SWT, giving me the chance to complete this Master project, peace, and blessing of Allah be upon his last prophet Mohammed (sallallahu-alayhi wasalam). I would like to want to express my gratitude to the one and only supervisor Assoc. Prof. Dr. Mohd Zaidi Jaafar for his support, kindness, and ideas that he had given me throughout these last two semesters. The final semesters have been of a great experience for me, thanks to my supervisor.

I want to thanks all lecturers and staff from the School of chemical and energy Engineering. Also, I am very grateful to my family who's always on my side Without the financial support from my family I couldn't achieve my goals. Lastly, I would also like to be grateful to my friends and my classmates at University Technology Malaysia for their daily support which made me stay and study in Johor a more enjoyable life.

ABSTRACT

Fossil fuel has been the primary energy supply compared to other alternative energy sources. Due to economic constraints, locating new reserves becomes harder. Approximately 35 % of the original oil in place can be retrieved from the reservoir by primary and secondary methods, while the rest remained trapped, in the last decade nanoparticles have been considered as an alternative recovery method to boost the oil recovery process and to extract trapped oil. The purpose of this study is to compare the effect of silica (SiO_2) and iron oxide (Fe_3O_4) nanoparticles for their recovery mechanism, the parameters involved in this study include, interfacial tension measurement using tensiometer, wettability alteration using contact angle method, and flooding using glass bead pack model. Silica (SiO_2) and iron oxide Fe_3O_4 with the concentration of (0.02-0.1) wt.% was prepared using deionized water, Sodium chloride (NaCl) as brine with the concentration of (0.7wt% - 2.2wt) has been used. The interfacial tension result showed for silica SiO_2 and iron oxide (Fe_3O_4) 43% and 33% reduction of IFT with an optimum concentration of 0.05 wt.% and 0.1 wt.% respectively. For the contact angle measurement, both silica SiO_2 and iron oxide (Fe_3O_4) nanoparticles alter the oil-wet system to the water-wet system by reducing contact angle 19% and 18% respectively. Finally, the additional oil recovery for SiO_2 and (Fe_3O_4) was 10.7% and 9.8% respectively. This study concludes silica SiO_2 nanofluids are more effective on that brine concentration for an enhanced oil recovery application.

ABSTRAK

Fosil bahan api telah menjadi bekalan tenaga utama membandingkan dengan alternatif lain untuk tenaga sumber. Dengan kekurangan ekonomi, lokasi rizab baru menjadi lebih sukar. Aproximately 35 % daripada asal minyak di tempat boleh diambil dari takungan dengan kaedah rendah dan menengah, manakala selebihnya kekal terperangkap, dalam nanopartikel dekad lalu telah dianggap sebagai kaedah pemulihan alternatif untuk meningkatkan proses pemulihan minyak dan ekstrak minyak terperangkap. Maksudnya ini kajian adalah untuk membandingkan kesan nanopartikel silika (SiO_2) dan oksida besi (Fe_3O_4) untuk mekanisme pemulihannya, parameter yang terlibat dalam kajian ini termasuk, pengukuran ketegangan antara muka menggunakan tensiometer, perubahan kebasahan menggunakan kaedah sudut kontak, dan banjir menggunakan model pekmanik kaca. Silika (SiO_2) dan besi oksida Fe_3O_4 dengan kepekatan (0,02-0,1)% berat dibuat dengan menggunakan air deionisasi, Natrium klorida (NaCl) sebagai air garam dengan kepekatan (0,7wt% - 2,2wt) mempunyai telah digunakan. Hasil ketegangan antara muka menunjukkan untuk silika SiO_2 dan besi oksida (Fe_3O_4) Pengurangan IFT sebanyak 43% dan 33% dengan kepekatan optimum masing-masing 0,05% berat dan 0,1% berat. Untuk pengukuran sudut kontak, kedua-dua nanopartikel silika SiO_2 dan besi oksida (Fe_3O_4), mengubah sistem basah-minyak ke sistem basah-air dengan mengurangkan sudut kontak masing-masing 19% dan 18%. Akhirnya, pemulihan minyak tambahan untuk SiO_2 dan (Fe_3O_4) masing-masing adalah 10.7% dan 9.8%. Kajian ini menyimpulkan bahawa nanofluid silika SiO_2 lebih berkesan pada kepekatan air garam untuk aplikasi pemulihan minyak yang dipertingkatkan.

TABLE OF CONTENT

TITLE	PAGES
DECLARATION	iv
DEDICATION	vi
Acknowledgment	vii
ABSTRACT	vi
ABSTRAK	vii
LIST OF TABLES	xi
LIST OF FIGURES	xii
LIST OF ABBREVIATIONS	xiii
Chapter 1	1
Introduction	1
1.1 Background of Study	1
1.2 Problem statement	2
1.3 Objectives	2
1.4 Scope of the Study	3
CHAPTER 2	4
LITERATURE REVIEW	4
2.1 introduction	4
2.2 Nanoparticles	6
2.3 Nanoparticles for EOR	7
2.3.1 Nanofluids	8
2.3.2 disjoining pressure	8
2.3.3 wettability alteration mechanism	10
2.3.4. interfacial tension mechanism	11

2.4 factors that influence EOR recovery:	12
2.5 Nanofluid stability:	13
2.5.1 Factors affecting nanofluid stability	14
2.5.2 DVLO Theory	14
2.5.3 Zeta potential	16
2.5.4 Isoelectric Point (IEP)	17
2.6 Silica NPs	18
2.6.1 Application of SNPs in EOR	19
2.7 Iron Oxide NP	20
2.7.1 Application of iron nanoparticles (IONPs) for EOR	21
CHAPTER 3	25
RESEARCH METHODOLOGY	25
3.1 overview:	25
3.2 Materials	26
3.2.1 Silica nanoparticle	26
3.2.2 Iron oxide nanoparticle	26
3.2.3 brine	26
3.2.4: Oil properties	26
3.2.5 Deionized water	27
3.3 Preparation of nanoparticle solution	28
3.4: IFT Measurement	28
3.5 Wettability alteration experiment :	29
3.5.1 Rock cleaning and aging:	30
3.5.2 Nanofluid treatment :	30
3.5.3 Contact angle measurement :	31
3.6 Porous media preparation	31
3.6.1 Porosity measurement.	31

3.6.2 Permeability measurement.	32
3.6.3 Irreducible Water Saturation (Swir) and Initial Oil Saturation (Soi) Measurement	33
3.6.4 wettability alteration of the glass bead pack	35
3.6.5 Oil Displacement test	35
CHAPTER 4	36
RESULT AND DISCUSSION:	36
4.1 introduction	36
4. 2. IFT Measurement	36
4.2.1 Effect of brine on Interfacial tension reduction	36
4.2.2 Effect of silica and iron oxide nanoparticle Interfacial tension reduction:	37
4.3 wettability measurement	40
4.3.1 Effect of silica and iron oxide nanoparticle Contact angle reduction.	40
4.4 OIL recovery performance for silica and iron oxide nanofluids:	43
CHAPTER 5	46
CONCLUSIONS AND RECOMMENDATION	46
5.1 Conclusion.	46
5.2 Recommendation	46
References:	47

LIST OF TABLES

TABLE NO.	TITLE	PAGE
Table 2.1	summary of applications of NPs in EOR.....	22
Table 2..2:	Recent applications of NPs in EOR.....	24
Table 3.1	Crude oil properties	27
Table 4.1	shows summary of properties of Glass bead pack model.....	43

LIST OF FIGURES

FIGURE NO.	TITLE	PAGE
Figure 2.1	classification scheme of nano structured materials , their composition dimension	.5
Figure 2.2	Comparison of length scale Adapted from (Liu, Zhang et al. 2020).....	6
Figure 2.3	Schematic of EOR mechanism of nano-fluids	8
Figure 2.4	. a)the stimulated shape of the meniscus profile in the wedge region from (Chengara, Nikolov et al. 2004). , b) nanoparticles structuring in wedge-film resulting in structural disjoining pressure gradient at wedge vertex from (Wasan, Nikolov et al. 2011)	9
Figure 2..5	schematic of wettability alteration mechanism of nanofluids adapted from.(Eltoum, Yang et al. 2020).....	10
Figure 2.6	Steric and electrostatic stabilization (Panneerselvam & Choi, 2014)	15
Figure 2.7	illustration of the surface and zeta potential (K. Pate and P. Safier, 201	16
Figure 2.8	isoelectric points.....	17
Figure 3.1	: Flow chart for methodology	25
Figure 3.2	Anton paar stabinger viscometer	27
Figure 3.3	Kruss K6 tensiometer	29
Figure 3.4	Kruss advance drop shape analyzer.....	30
figure 3.5	dimention of the glass bead	31
Figure 3.6:	Injection of crude oil into the glass bead pack model	34
Figure 3.7:	Experimental setup of oil displacement test by Nanofluid flooding	35
Figure 4.1	Effect of brine on Interfacial tension reduction.....	37
Figure 4.2.	Effect of bare silica concentration on interfacial tension reduction.	38
Figure 4.3	effect of brine concentration and SiO ₂ on interfacial tension reduction.	38
Figure 4.4	effect of different concentration IONPs on interfacial tension reduction.	39
Figure 4..5	Effect of of brine concentration and IONPS on interfacial tension reduction. 39	39
Figure 4..6	effect of brine concentration and silica on contact angle reduction.	41
Figure 4.7	effect of brinet concentration of and silica on contact angle reduction	Error!
Bookmark not defined.		
Figure 4..8	effect of bare IONPs on contact angle reduction.....	41
Figure 4..9	effect of brine concentration of iron oxide interfacial tension reduction.	42

Figure 4.11 show the log jamming process adapted from (Sun, Zhang et al. 2017)45

LIST OF ABBREVIATIONS

EOR:	Enhanced Oil Recovery
HLB:	Hydrophilic- Lipophilic Balance
IFT:	Interfacial Tension
IONPs:	iron oxide nanoparticles

Chapter 1

Introduction

1.1 Background of Study

Over the past three centuries, fossil fuel was the primary energy supply compared to other alternative energy supplies. On the other hand, due to economic constraints locating new reserves becomes harder. Approximately around 35% of total oil can be recovered from the reservoir by means of the primary and secondary method while the remaining oil will remain trapped in the reservoir (Sukesh and Deka 2017). The remaining oil is essential after water floods and cannot be overlooked at this time of high energy demand. As a result, major oil companies are preferred in utilizing enhanced oil recovery (EOR) methods to achieve energy demands (Manrique and Alvarado 2010).

Several Enhanced oil recovery (EOR) methods have been developed to extract the trapped oil in the reservoir. These EOR are classified into thermal and non-thermal EOR. Thermal EOR is limited to shallow depth reservoir application, hence non-thermal EOR, which includes gas injection, microbial and chemical flooding, receives tremendous attention for oil recovery. Chemical EOR has been considered one of the most promising in EOR applications due to its high efficiency for extracting residual oil compared to other non-thermal EOR methods. Chemical processes are usually constrained by chemicals' exorbitant price, losses of chemicals, and possible formation damages (Chang et al., 2006). As such, more efficient, less expensive, and environmentally safe EOR methods are in great need.

1.2 Problem statement

Nanoparticles provide new routes to resolve unsolved challenges. NPS exhibits several superior properties as EOR agents compared to conventional methods such as chemical method, where pore plugging trapped of injection chemicals in the porous media are among the most critical challenges to chemical processes, which causes permeability(k) reduction of the formation and increasing the cost of chemical processes (Luka, Ahmadi et al. 2015)The commonly used NPs are SiO₂, TiO₂ and Al₂O₃ are in the range of 100nm to 1 nm. which is smaller than the pore through and pore size. therefore, nanoparticles smoothly flowed and transported across the porous media, eliminating any permeability (k) reduction and loss of the chemicals, Nps can increase the sweep efficiency resulting in higher macroscopic efficiency, which leads to higher recovery. Silica is considered one of the most used promising nanoparticle recovery for improving the EOR mechanism.

The purpose of this study is to compare the effect of silica (SiO₂) and iron oxide (Fe₃O₄) nanoparticles for their recovery mechanism, the parameters involved in this study include, interfacial tension measurement using tensiometer, wettability alteration using contact angle method, and flooding using glass bead pack model.

1.3 Objectives

The objectives of this research are as follow:

- i. To investigate Silica and iron oxide nanoparticles' effects to reduce interfacial tension at different concentrations and salinity.
- ii. To investigate the Silica and iron oxide nanoparticle to alter wettability at different concentration salinity.
- iii. To compare the recovery performance of both silica oxide nanoparticles and iron oxide nanoparticles through flooding tests.

1.4 Scope of the Study

the scope of this study has been categorized as flowing:

- i. Measurement of IFT reduction in DI water, at 24⁰C, with concentration (0.020wt%, 0.05 wt%, 0.1wt%, for both silica and iron oxide NP to identify optimum concentration.
- ii. Measurement of IFT reduction of both Silica and iron oxide nanoparticles with variable NaCl concentration of 0.7wt%,1.2wt%,1.7 wt% 2.2wt% to determine the influence of salinity on IFT reduction.
- iii. Measurement of the contact angle in DI water, at 24⁰C, with concentration 0.02wt%, 0.05 wt%, 0.1wt%, for both silica and Iron oxide NPs.
- iv. Measurement of the contact angle in a 3-phase system using different concentrations of Silica and iron oxide with variable NaCl concentration of 0.7wt%,1.2wt%,1.7wt% 2.2wt% to determine the effect of salinity on contact angle.
- v. Conducting flooding test using artificial heterogeneous glass beads pack flooding model at optimum silica oxide and iron oxide nanoparticles concentration at 24⁰C.

References:

Agi, A., et al. (2018). "Mechanism governing nanoparticle flow behavior in porous media: insight for enhanced oil recovery applications." International Nano Letters **8**(2): 49-77.

Al-Anssari, S., et al. (2016). "Wettability alteration of oil-wet carbonate by silica nanofluid." Journal of colloid and interface science **461**: 435-442.

Aminian, A. and B. ZareNezhad (2019). "Wettability alteration in carbonate and sandstone rocks due to low salinity surfactant flooding." Journal of Molecular Liquids **275**: 265-280.

Binks, B. P. and A. T. Tyowua (2016). "Oil-in-oil emulsions stabilised solely by solid particles." Soft Matter **12**(3): 876-887.

Campos, E. A., et al. (2015). "Synthesis, characterization and applications of iron oxide nanoparticles-a short review." Journal of Aerospace Technology and Management **7**(3): 267-276.

Chengara, A., et al. (2004). "Spreading of nanofluids driven by the structural disjoining pressure gradient." Journal of colloid and interface science **280**(1): 192-201.

Cheraghian, G. and L. Hendraningrat (2016). "A review on applications of nanotechnology in the enhanced oil recovery part A: effects of nanoparticles on interfacial tension." International Nano Letters **6**(2): 129-138.

Eltoum, H., et al. (2020). "The effect of nanoparticles on reservoir wettability alteration: a critical review." Petroleum Science: 1-18.

Giraldo, J., et al. (2013). "Wettability alteration of sandstone cores by alumina-based nanofluids." Energy & Fuels **27**(7): 3659-3665.

Hendraningrat, L., et al. (2013). "A coreflood investigation of nanofluid enhanced oil recovery." Journal of Petroleum Science and Engineering **111**: 128-138.

Hendraningrat, L. and O. Torsæter (2014). Understanding fluid-fluid and fluid-rock interactions in the presence of hydrophilic nanoparticles at various conditions. SPE Asia Pacific Oil & Gas Conference and Exhibition, Society of Petroleum Engineers.

Kazemzadeh, Y., et al. (2015). "Impact of Fe₃O₄ nanoparticles on asphaltene precipitation during CO₂ injection." Journal of Natural Gas Science and Engineering **22**: 227-234.

Kothari, S., et al. (2010). "Implications for GAAP from an analysis of positive research in accounting." Journal of Accounting and Economics **50**(2-3): 246-286.

Liu, D., et al. (2020). "Review on nanoparticle-surfactant nanofluids: formula fabrication and applications in enhanced oil recovery." Journal of Dispersion Science and Technology: 1-15.

Luka, G., et al. (2015). "Microfluidics integrated biosensors: A leading technology towards lab-on-a-chip and sensing applications." Sensors **15**(12): 30011-30031.

Manrique, E. and V. Alvarado (2010). "Enhanced oil recovery: An update review." Energies **3**: 1529-1575.

Olayiwola, S. O. and M. Dejam (2020). Effect of Silica Nanoparticles on the Oil Recovery During Alternating Injection with Low Salinity Water and Surfactant into Carbonate Reservoirs. SPE Annual Technical Conference and Exhibition, Society of Petroleum Engineers.

Panneerselvam, S. and S. Choi (2014). "Nanoinformatics: emerging databases and available tools." International journal of molecular sciences **15**(5): 7158-7182.

Rezvani, H., et al. (2019). "A new insight into Fe₃O₄-based nanocomposites for adsorption of asphaltene at the oil/water interface: an experimental interfacial study." Journal of Petroleum Science and Engineering **177**: 786-797.

Shojaati, F., et al. (2017). "Investigating the effect of salinity on the behavior of asphaltene precipitation in the presence of emulsified water." Industrial & Engineering Chemistry Research **56**(48): 14362-14368.

Skauge, T., et al. (2010). Nano-sized particles for EOR. SPE improved oil recovery symposium, OnePetro.

Sukesh, C. M. and B. Deka (2017). Nano Particle Based Polymer Flooding for Enhanced Oil Recovery: A Review. International Conference on Nano for Energy and Water, Springer.

Sun, X., et al. (2017). "Application of nanoparticles in enhanced oil recovery: a critical review of recent progress." Energies **10**(3): 345.

Wasan, D., et al. (2011). "The wetting and spreading of nanofluids on solids: Role of the structural disjoining pressure." Current Opinion in Colloid & Interface Science **16**(4): 344-349.

Huh C, Nizamidin N, Pope GA, Milner TE, Wang B. Hydrophobic paramagnetic nanoparticles as intelligent crude oil tracers. U.S. Patent application 14/765,426[P]. 2015-12-31.

Rezaei Gomari KA, Denoyel R, Hamouda AA. Wettability of calcite and mica modified by different long-chain fatty acids (C18 acids). J Colloid Interface Sci. 2006;297:470–9.

Kothari N, Raina B, Chandak KB, Iyer V, Mahajan, HP. Application of ferrofluids for enhanced surfactant flooding in IOR. In: SPE EUROPEC/EAGE annual conference and

exhibition. Society of Petroleum Engineers, Barcelona, Spain; 2010. <https://doi.org/10.2118/131272-MS>.

Zeyghami M, Kharrat R, Ghazanfari M. Investigation of the applicability of nano silica particles as a thickening additive for polymer solutions applied in EOR processes. *Energy Sources Part A*. 2014;36:1315–24. <https://doi.org/10.1080/15567036.2010.551272>.

Ogolo N, Olafuyi O, Onyekonwu M. Enhanced oil recovery using nanoparticles. In: SPE Saudi Arabia section technical symposium and exhibition, 8–11 Apr, Al-Khobar, Saudi Arabia; 2012. <https://doi.org/10.2118/160847-MS>.

Joonaki E, Ghanaatian S. The application of nanofluids for enhanced oil recovery: effects on interfacial tension and coreflooding process. *Pet Sci Technol*. 2014;32:2599–607. <https://doi.org/10.1080/10916466.2013.855228>.

Kamal MS, Adewunmi AA, Sultan A, Al-Hamad MF, Mehmood U. Recent advances in nanoparticles enhanced oil recovery: rheology, interfacial tension, oil recovery, and wettability alteration. *J Nanomater*. 2017. <https://doi.org/10.1155/2017/2473175>.

Hendraningrat L, Li S, Torsaeter O. A coreflood investigation of nanofluid enhanced oil recovery. *J Pet Sci Eng*. 2013;111:128–38. <https://doi.org/10.1016/J.Petrol.2013.07.003>.

Agista MN, Guo K, Yu Z. A state-of-the-art review of nanoparticles application in petroleum with a focus on enhanced oil recovery. *Appl Sci*. 2018;8(6):871–99. <https://doi.org/10.3390/app8060871>.

Chuan Lim, E. W. and Feng, R. (2012) „Agglomeration of magnetic nanoparticles“, *Journal of Chemical Physics*, 136(12). doi: 10.1063/1.3697865.

Derjaguin, B. and Landau, L. (1993) „Theory of the stability of strongly charged lyophobic sols and of the adhesion of strongly charged particles in solutions of electrolytes“, *Progress in Surface Science*, 43(1–4), pp. 30–59. doi: 10.1016/0079-6816(93)90013-L.

Dishon, M., Zohar, O. and Sivan, U. (2009) „From repulsion to attraction and back to repulsion: The effect of NaCl, KCl, and CsCl on the force between silica surfaces in aqueous solution“, *Langmuir*, 25(5), pp. 2831–2836. doi: 10.1021/la803022b.

Dong, S., Sun, Y., Gao, B., Shi, X., Xu, H., Wu, J. and Wu, J. (2017) „Retention and transport of graphene oxide in water-saturated limestone media“, *Chemosphere*. Elsevier Ltd, 180, pp. 506–512. doi: 10.1016/j.chemosphere.2017.04.052.

Duster, T. A., Na, C., Bolster, D. and Fein, J. B. (2017) „Transport of Single-Layered Graphene Oxide Nanosheets through Quartz and Iron Oxide – Coated Sand Columns“, 143(2), pp. 1–9. doi: 10.1061/(ASCE)EE.1943-7870.0001156.

Dutta, J. and Hoffmann, H. (2005) *Nanomaterials*.

El-diasty, A. I., Aly, A. M., Technical, T. P. S. and Services, P. (2015) „Understanding the Mechanism of Nanoparticles Applications in Enhanced Applications of Nanoparticles in EOR“, Paper SPE 175806 - North Africa Technical Conference (Cairo / Egipto), 0, pp. 1–19. doi: 10.2118/175806- MS.

El-sayed, G. M., Kamel, M. ., Morsy, N. S. and Taher, F. A. (2013) „Encapsulation of Nano Disperse Red 60 via Modified Miniemulsion Polymerization. I. Preparation and Characterization“, *Polymers and Polymer Composites*, 21(7), pp. 449–456. doi: 10.1002/app.

Esfandyari Bayat, A., Junin, R., Samsuri, A., Piroozian, A. and Hokmabadi, M. (2014) „Impact of Metal Oxide Nanoparticles on Enhanced Oil Recovery from Limestone Media at Several Temperatures“, *Energy & Fuels*, 28(10), pp. 6255–6266. doi: 10.1021/ef5013616.

Esfandyari Bayat, A., Junin, R., Shamshirband, S. and Tong Chong, W. (2015) „Transport and retention of engineered Al₂O₃, TiO₂, and SiO₂ nanoparticles through various sedimentary rocks“, *Scientific Reports*, 5(1), p. 14264. doi: 10.1038/srep14264.

Fakoya, M. F. and Shah, S. N. (2017) „Emergence of nanotechnology in the oil and gas industry: Emphasis on the application of silica nanoparticles“, *Petroleum*. Elsevier Ltd, pp. 1–15. doi: 10.1016/j.petlm.2017.03.001.

Fan, H., Resasco, D. E. and Striolo, A. (2011) „Amphiphilic silica nanoparticles at the decane-water interface: Insights from atomistic simulations“, *Langmuir*, 27(9), pp. 5264–5274. doi: 10.1021/la200428r.

Fan, W., Jiang, X. H., Yang, W., Geng, Z., Huo, M. X., Liu, Z. M. and Zhou, H. (2015) „Transport of graphene oxide in saturated porous media: Effect of cation composition in mixed Na-Ca electrolyte systems“, *Science of the Total Environment*. Elsevier B.V., 511, pp. 509–515. doi: 10.1016/j.scitotenv.2014.12.099.

Operti, L., Rabezzana, R., & Vaglio, G. A. (2006). Negative gas-phase ion chemistry of silane: a quadrupole ion trap study. *Rapid Communications in Mass Spectrometry*, 20(18), 2696–2700. doi:10.1002/rcm.2662

Ortega, D. J. S., Kim, H. B., James, L. A., Johansen, T. E., & Zhang, Y. (2016, November 7–10). The effectiveness of silicon dioxide SiO₂ nanoparticle as an enhanced oil recovery agent in Ben Nevis formation, Hebron field, offshore Eastern Canada. Paper presented at the Abu Dhabi International Petroleum Exhibition & Conference, UAE. doi:10.2118/183546-MS

Osei-Bonsu, K., Grassia, P., & Shokri, N. (2017). Relationship between bulk foam stability, surfactant formulation and oil displacement efficiency in porous media. *Fuel*, 203, 403–410. doi:10.1016/j.fuel.2017.04.114

Samin, A. M., Manan, M. A., Idris, A. K., Said, M., & Alghol, A. (2017). Protein foam application for enhanced oil recovery. *Journal of Dispersion Science and Technology* 38(4), 604-609. doi:10.1080/01932691.2016.1185014

Sanchez, J. M., & Hazlett, R. D. (1992). Foam flow through oil-wet porous medium: A laboratory study. *SPE Reservoir Engineering*, 7(01), 91–97. doi:10.2118/19687-PA

Sastry, N. V., Séquaris, J.-M., & Schwuger, M. J. (1995). Adsorption of polyacrylic acid and sodium dodecylbenzenesulfonate on kaolinite. *Journal of Colloid and Interface Science*, 171(1), 224–233. doi:10.1006/jcis.1995.1171