

SUPERHYDROPHOBIC-SUPEROLEOPHILIC KAOLIN BASED
MICROFILTRATION MEMBRANE FOR OIL RECOVERY FROM OILFIELD
PRODUCED WATER

JAMILU USMAN

UNIVERSITI TEKNOLOGI MALAYSIA

SUPERHYDROPHOBIC-SUPEROLEOPHILIC KAOLIN BASED
MICROFILTRATION MEMBRANE FOR OIL RECOVERY FROM OILFIELD
PRODUCED WATER

JAMILU USMAN

A thesis submitted in partial fulfilment of the
requirements for the award of the degree of
Doctor of Philosophy

School of Chemical and Energy Engineering
Faculty of Engineering
Universiti Teknologi Malaysia

OCTOBER 2020

ACKNOWLEDGEMENT

Alhamdulillah, all praises are to Allah, the beneficent, the merciful, the God of all heavens and earth that gave me knowledge, health, patience, ability, guidance and protection through the period of my study. First and foremost, I will like to express my utmost gratitude to my supervisor, Assoc. Prof. Ts. Dr. Mohd Hafiz Dzarfan Othman for his unending guidance, encouragement, advice and support for this research. He is always supportive and willing to give all his best to help me to strive in my research through all forms of assistance. His selfless sharing of his wide knowledge, experience and personal guidance has given great value to this research.

I am very grateful to all the academic staff, administrative and technical staff in Advanced Membrane Technology Research Centre (AMTEC), Universiti Teknologi Malaysia (UTM) for their physical and emotional supports. Special appreciation goes to Tijjani Hassan El-badawy, Yusuf O. Raji, Dr Lukka Thuyavan, Dr Gbadamosi Afeez Olayinka Tai Zhong Sheng, Dr. Siti Khadija Hubadillah, Dr. Mohd Ridhwan Adam, Asif Hafeez, Stanley Chinedu Mamah for the motivation and being by side throughout the research period, thank you so very much.

I also owe a big thank to my beloved mother; Hajiya Rabi (Maman Jamilu), my beloved wife Amina Mohammad, my uncle Senator Garba Ilah Gada and Maryam Usman Ardo for their encouragement and motivation that help me to get over the twists and turns in this research. With the blessings of all the people who have been constantly helping me, I believe that I will be able to complete this project successfully. May Allah bless you for your endless love. I love you so very much.

My deepest appreciation is also dedicated to all my research colleagues who have been very helpful in guiding and assisting me in my research. Encouragement and help never failed to come from my friends, notably Umar Sani, Mubaraq Mohammad Nasir, Abubakar Yusuf, Abudullahi Abubakar, Ismail Rabo, Murtala Umar (Chairman), Shafiu Mohammad Dansanda, Mustapha Bature, Saleem Abubakar, Munir Maizuwa, Umar Garkuwa, Usman Mohammad Kyari, Abdulwahab Iya, Mal. Ibrahim Mohammad, Mal. Shamsudden Ahmad, Mal. Mustapha Gani, Mal. Nura Bashir, Mal. Anas Tukur Balarabe, Mal. Anas Physics, Umar Farouq Lawal and others.

I also wish to show my appreciation to the management of Sokoto State University, Sokoto, Nigeria for granting me study leave, and Petroleum Technology Development (PTDF), Nigeria for sponsoring me to undertake this programme. Finally, I thank and pray for all those whose efforts, courage, guidance, help, support, thought, love, care and concern has made me what I am today. May Allah continue to bless you all abundantly, Amen

ABSTRACT

The discharge of oilfield produced water (OPW) causes disruption of the ecosystem and environmental degradation. Herein, novel hybrid membrane coupled absorption-filtration technology is proposed for the recovery of oil from OPW. The present study aims to develop a superhydrophobic-superoleophilic kaolin-based hollow fibre ceramic membrane using phase inversion and sintering technique for the recovery of oil from synthetic OPW. To achieve the superhydrophobic-superoleophilic modification, organosilanes sol-gel coating was performed on kaolin-based hollow fibre ceramic membranes. Membrane morphology and surface roughness was analysed using field emission scanning electron microscopy (FESEM) and atomic force microscopy. The membrane surface functionality was studied using Fourier transform infrared, X-ray photoemission spectroscopy (XPS) and X-ray diffraction analysis. The membrane filtration performance was evaluated using cross flow module. In the first stage of the work, feasibility studies of Malaysian kaolin (MK) and Nigerian kaolin (NK) were studied on fabrication of kaolin-based hollow fibre membrane by varying the loading composition (34 to 37 wt.%) and sintering temperature (1200 to 1500°C). Experimental results show that increase of kaolin concentration and sintering temperatures decreases the flux rate. The physiochemical and performance analysis showed that 34 wt.% MK ceramic membrane exhibits better water flux (565.06 L/m²h) with desired pore size and stability than 34 wt.% NK membrane. It owes to the MK which hold higher degree of crystallinity and smaller particle size. In the second stage, for effective oil absorption-filtration, organosilane agents such as methyltriethoxysilane (MTES), fluoroalkylsilane, octadecyltrimethoxysilane, chlorotrimethylsilane chlorotrimethylsilane, and trichloro(octadecyl)silane were used for the modification of superhydrophobic-superoleophilic kaolin hollow fibre membrane. XPS and FESEM analysis clearly indicated that the organosilanes are bound firmly on the surface of kaolin membranes. The effect of coating cycle and oil concentration were also studied. Among the coated membranes, MTES coated kaolin membrane showed the maximum water contact angle of 161.3° and lowest oil contact angle of 0°. Resultantly, this depicts that the superhydrophobic-superoleophilic property were attained. In the third stage of the study, the oil recovery performance of the kaolin membranes with different organosilane agents were evaluated and compared. MTES-coated membranes showed maximum oil absorption capacity of 10 g/g, oil flux of 80 L/m²h, and oil separation efficiency 90%. The optimized MTES coated membranes were adopted to further optimization of process condition (oil concentration, feed flow and feed pH) in cross flow module for the effective oil flux and separation efficiency using response surface methodology (RSM). From the central composite design, maximum oil flux of 97.67 L/m²h and separation efficiency 98.41% were observed at oil concentration of 50 mg/L, feed flow of 300 mL/min, and feed pH of 4. The RSM model was good coherent with experimental data. Overall, this study portrays the development of economically viable superhydrophobic-superoleophilic kaolin hollow fibre membrane for the absorption combined filtration process for the separation of oil from produced water. This study would pave the way for researchers to eliminate the pollutants using hybrid absorption-filtration process.

ABSTRAK

Pembuangan air keluaran dari medan minyak (OPW) telah menyebabkan berlakunya gangguan terhadap ekosistem dan kemerosotan alam sekitar. Oleh itu membran hibrid baharu yang digabungkan dengan teknologi penyerapan-penapisan telah dicadangkan untuk perolehan minyak daripada OPW. Kajian ini bertujuan untuk menghasilkan membran seramik gentian berongga berasaskan kaolin superhidrofobik-superoleofilik menerusi penggunaan teknik penyongsangan fasa dan pensinteran bagi perolehan minyak daripada OPW sintetik. Bagi melaksanakan pengubahsuaian superhidrofobik-superoleofilik, penyalutan sol-gel organosilana telah dilakukan terhadap membran seramik gentian berongga berasaskan kaolin. Morfologi dan kekasaran permukaan membran telah dianalisis menggunakan mikroskopi elektron imbasan pancaran medan (FESEM) dan mikroskopi daya atom. Kefungsian permukaan membran dikaji menggunakan analisis daripada inframerah jelmaan Fourier, spektroskopi pelepasan cahaya sinar-X (XPS), dan belauan sinar-X. Prestasi penapisan membran telah dinilai menggunakan modul aliran silang. Pada peringkat pertama, kajiankebolehlaksanaan telah dilaksanakan terhadap kaolin Malaysia (MK) dan kaolin Nigeria (NK) bagi penghasilan membran gentian berongga berasaskan kaolin, dengan mempelbagaikan komposisi bahan (34 sehingga 37 wt.%) dan suhu pensinteran (1200 sehingga 1500°C). Hasil kajian telah menunjukkan bahawa peningkatan kepekatan kaolin dan suhu pensinteran mengurangkan kadar fluks. Hasil analisis terhadap fizikimia dan prestasi menunjukkan bahawa membran seramik MK dengan 34 wt.% menghasilkan fluks air yang lebih baik (565.06 L/m²h) pada saiz liang dan kestabilan sasaran berbanding membran NK dengan 34 wt.%. Keputusan ini berpunca daripada MK yang mempunyai darjah kehabluran yang lebih tinggi dan saiz zarah yang lebih kecil. Pada peringkat kedua, bagi mencapai keberkesanan penyerapan minyak-penapisan, agen organosilana seperti metiltritoksisilana (MTES), fluoroalkilsilana, oktadesiltrimetoksisilana, klorotrimetilsilana, dan trikloro(oktadesil)silana telah digunakan untuk pengubahsuaian membran gentian berongga kaolin superhidrofobik-superoleofilik. Hasil analisis daripada XPS dan FESEM jelas menunjukkan bahawa organosilana telah melekat dengan kuat pada permukaan membran kaolin. Kesan kitaran penyalutan dan kepekatan minyak turut dikaji. Dalam kalangan membran bersalut, membran kaolin bersalut MTES menunjukkan sudut sentuh air maksimum yang bernilai 161.3° dan sudut sentuh minyak paling rendah, iaitu 0°. Ia menunjukkan bahawa sifat superhidrofobik-superoleofilik berjaya diperolehi. Pada peringkat ketiga kajian, prestasi perolehan minyak bagi membran kaolin dengan agen organosilana yang berbeza turut dinilai dan dibanding. Membran bersalut MTES menunjukkan keupayaan serapan minyak yang maksimum iaitu 10 g/g, fluks minyak sebanyak 80 L/m²h dan kecekapan pemisahan minyak pada tahap 90%. Membran bersalut MTES teroptimum telah digunakan untuk pengoptimuman keadaan proses yang seterusnya (kepekatan minyak, aliran suapan dan pH suapan) dalam modul aliran silang untuk meningkatkan fluks minyak dan kecekapan pemisahan menggunakan kaedah sambutan permukaan (RSM). Berdasarkan reka bentuk komposit pusat, fluks maksimum minyak dan kecekapan maksimum pemisahan masing-masing bernilai 97.67 L/m²h dan 98.41% telah dicapai pada kepekatan minyak bernilai 50 mg/L, aliran suapan bernilai 300 mL/min, dan pH suapan bernilai 4. Model RSM yang diperolehi adalah setanding dengan data uji kaji. Secara keseluruhan, kajian ini menunjukkan pembangunan membrane gentian berlubang kaolin superhidrofobik-superoleofilik yang berdaya maju secara ekonomik untuk proses gabungan penyerapan dan penapisan bagi pemisahan minyak daripada air keluaran. Kajian ini mampu membantu para penyelidik dalam menyingkir bahan cemar menggunakan proses hibrid penyerapan-penapisan.

TABLE OF CONTENTS

	TITLE	PAGE
	DECLARATION	ii
	DEDICATION	iii
	ACKNOWLEDGEMENT	iv
	ABSTRACT	v
	ABSTRAK	vi
	TABLE OF CONTENTS	vii
	LIST OF TABLES	xiii
	LIST OF FIGURES	xv
	LIST OF ABBREVIATIONS	xix
	LIST OF SYMBOLS	xx
CHAPTER 1	INTRODUCTION	1
	1.1 Background of Research	1
	1.2 Problem Statement	5
	1.3 Research objectives	7
	1.4 Scope of Study	7
	1.5 Significant of the Study	9
	1.6 Thesis Organization	9
CHAPTER 2	LITERATURE REVIEW	13
	2.1 Oilfield Produced Water	13
	2.2 The Need of OPW Treatment	15
	2.3 Technologies for Removing Oil from Oilfield Produced Water	16
	2.3.1 Adsorption Technique	16
	2.3.2 Hydro-cyclones Technique	17
	2.3.3 Flotation	18
	2.3.4 Coagulation	19

2.3.5	Biological Technique	20
2.3.6	Membrane Filtration Technique	22
2.3.7	Hybrid Absorption-Filtration Technique	23
2.4	Surface Modification of Porous Substrate for Oil Removal	24
2.4.1	Copper as a Porous Substrate for Oil Recovery	26
2.4.2	Sponge as a Porous Substrate for Oil Recovery	28
2.4.3	Textile as a Porous Substrate for Oil Recovery	29
2.4.4	Stainless Steel as a Porous Substrate for Oil Recovery	31
2.5	Introduction to Membrane Separation Technology	33
2.5.1	Ceramic Membrane as Porous Substrate	37
2.5.2	Characteristics of Ceramic Membranes	39
2.6	Methods of Fabrication of Ceramic Membrane	41
2.6.1	Slip Casting Ceramic Membrane Fabrication Method	42
2.6.2	Tape Casting Ceramic Membrane Fabrication Method	44
2.6.3	Pressing Ceramic Membrane Fabrication Method	45
2.6.4	Extrusion Ceramic Membrane Fabrication Method	48
2.6.5	Phase Inversion/Sintering Technique	50
2.7	Kaolin	55
2.8	The Phase Transformation of Kaolin	56
2.9	Superhydrophobic-Superoleophilic Ceramic Membrane	57
2.10	Ceramic Membrane Surface Modification Techniques	63
2.10.1	Immersion Technique	64
2.10.2	Chemical Vapor Deposition Technique	67
2.9.3	Sol-Gel Technique	70
2.9.4	Lipid Solution Direct Grafting	73
2.11	Optimization Study using Design of Experiment	75
2.12	Response Surface Modelling (RSM)	76

2.13	Concluding Remarks of the Literature and Research Gap	78
CHAPTER 3	RESEARCH METHODOLOGY	81
3.1	Introduction	81
3.2	Material	83
3.2.1	Kaolin Clay	83
3.2.2	Solvents and Organosilane Agents Used	83
3.2.3	Dispersant	83
3.2.4	Polymer Binder	84
3.2.5	Synthetic Produced Water	84
3.3	Characterization of Raw Kaolin Clay Powder	85
3.3.1	X-ray Fluorescence (XRF)	85
3.3.2	Brunauer-Emmett-Teller Method (BET)	86
3.3.3	Field Emission Scanning Electron Microscopy (FESEM)	86
3.3.4	X-ray Diffraction (XRD)	86
3.3.5	Particle Size Analysis (PSA)	87
3.3.6	Thermogravimetric Analysis (TGA)	87
3.4	Fabrication of Kaolin-Based Hollow Fibre Ceramic Membrane	87
3.4.1	Preparation of Kaolin Dope Suspension	87
3.4.2	Extrusion of Kaolin Dope Suspension to Form Membrane Precursor via Phase Inversion	89
3.4.3	Sintering of Kaolin-Based Ceramic Hollow Fibre Membrane	90
3.5	Characterization of the Fabricated Kaolin-Based Hollow Fibre Ceramic Membrane	91
3.5.1	Viscosity Analysis	91
3.5.2	Membrane Morphological Study	92
3.5.3	Mechanical Strength Testing	92
3.5.4	Porosity and Pore Size Distribution Test	92
3.5.5	X-ray Diffraction (XRD)	93
3.5.6	Water Permeation Test	94

3.6	Surface Grafting of the Kaolin-Based Ceramic Hollow Fibre Membrane	94
3.6.1	Preparation of Silica Sol-Gel Solution	94
3.6.2	Coating of Superhydrophobic-Superoleophilic Layer onto the Pristine Kaolin-Based Hollow Fibre Ceramic Membrane	95
3.7	Characterization of the Modified Kaolin- Based Hollow Fibre Ceramic Membrane	98
3.7.1	Particle Size Analysis (PSA) and Zeta Potential Analysis	98
3.7.2	Morphological Study	98
3.7.3	Surface Roughness	98
3.7.3	Fourier Transform Infrared (FTIR)	99
3.7.4	X-ray Photoemission Spectroscopy (XPS)	99
3.7.5	Contact Angle Measurement	99
3.7.6	X-ray Diffraction (XRD) Analysis	100
3.7.7	Porosity and Pore Size Distribution Test	100
3.8	Performance and Optimization Study of Process Parameters on Surface Modified Kaolin-Based Hollow Fibre Membranes for Oil Absorption And Filtration Studies	101
3.8.1	Oil Absorption Capacity	101
3.8.2	Oil Flux filtration Studies	102
3.8.3	Optimization Process via Research Surface Methodology (RSM) Software	103
3.8.3.1	Analysis of Variance (ANOVA)	105
3.8.3.2	Confirmation Test	106
CHAPTER 4	FEASIBILITY STUDIES OF MALAYSIAN AND NIGERIAN KAOLIN FOR HOLLOW FIBRE MEMBRANE FABRICATION	107
4.1	Introduction	107
4.2	Results and Discussion	108
4.2.1	Physiochemical Analysis of Malaysian and Nigerian Kaolin Clay Powders	108
4.3	Effects of Loading Composition and Sintering Temperature on Malaysian and Nigerian Kaolin Based Hollow Fibre Ceramic Membrane	111

	4.3.1	Effects of Loading Composition	111
	4.3.2	Effects of Sintering Temperature	114
	4.4	Conclusion	122
CHAPTER 5		SURFACE COATING OF THE KAOLIN-BASED HOLLOW FIBRE CERAMIC MEMBRANE;	125
	5.1	Introduction	125
	5.2	Results and Discussions	126
	5.2.1	Silane Sol-Gel Characterization	126
	5.2.2	FESEM and EDS Mapping	127
	5.2.3	Surface Topology Analysis	130
	5.2.4	FTIR Analysis	132
	5.2.5	XPS Analysis	133
	5.2.6	Surface Wettability and Water Super- Repellency	137
	5.2.7	Crystallinity of the Membranes	141
	5.2.8	Porosity and Pore Size Distribution of the Coated Membrane	142
	5.3	Conclusions	144
CHAPTER 6		PERFORMANCE AND OPTIMIZATION STUDY OF PROCESS PARAMETERS ON SURFACE MODIFIED KAOLIN-BASED HOLLOW FIBRE MEMBRANES FOR OIL ABSORPTION AND FILTRATION STUDIES	145
	6.1	Introduction	145
	6.2	Results and Discussion	146
	6.2.1	Effect of Organosilane Agents on Oil Absorption Studies	146
	6.2.1.1	Effect of Different Hydrocarbon Absorption Capacity on the Pristine and Coated Membranes	146
	6.2.1.2	Effect of Different Crude Oil Concentration and Organosilane Coating Cycles on Oil Absorption Capacity of Membranes	147
	6.2.2	Crude Oil Filtration Studies	149

6.2.3	Optimization Study of the Separation Parameters on the Coated Membrane	150
6.2.3.1	Oil Flux Responses	152
6.2.3.2	Response Surface and Contour Plots of Separation Efficiency	156
6.2.4	Prediction Parameter Analysis for Oil Flux and Separation Efficiency	158
6.2.5	Optimization Desirability	159
6.2.6	Confirmatory Test	161
6.3	Conclusion	162
CHAPTER 7	GENERAL CONCLUSIONS AND RECOMMENDATIONS	165
7.1	General Conclusions	165
7.2	Recommendations	166
	REFERENCES	168
	LIST OF PUBLICATIONS	199

LIST OF TABLES

TABLE NO.	TITLE	PAGE
Table 2.1	Typical characteristics of OPW parameters worldwide	14
Table 2.2	common membrane separation processes	36
Table 2.3	Features of ceramic-based membrane filtration	40
Table 2.4	Some summary of ceramic membrane fabricated using the pressing method	46
Table 2.5	Summary of some ceramic membranes prepared by the extrusion method	49
Table 2.6	Comparison of the different ceramic membrane fabrication techniques	53
Table 2.7	Summary of hydrophobic surface modification of ceramic membrane fabrications processes	74
Table 3.1	Properties of the polyethersulfone (PESF)	84
Table 3.2	Characterization of the sample of the crude oil used	85
Table 3.3	Kaolin dope composition at different loadings	88
Table 3.4	Spinning condition for hollow fibre membrane precursor	89
Table 3.5	Factors and levels for response surface study	104
Table 3.6	The experimental layout of 2 ³ full factorial CCD	104
Table 4.1	Chemical oxide compositions of MK and NK used in this study	109
Table 5.1	Atomic composition of detected elemental species from XPS survey, average porosity and average pore size of the pristine and coated membranes	137
Table 6.1	Oil absorption capacity comparison data with the works of literature	150
Table 6.2	Experiment parameters and levels	151
Table 6.3	CCD Design 2 by 3 factorial and experimental responses	151
Table 6.4	ANOVA for oil flux	154
Table 6.5	ANOVA for Superhydrophobic-superoleophilic kaolin-based hollow fibre ceramic membrane oil separation efficiency	157

Table 6.6	ANOVA and Regression analysis of the Superhydrophobic-superoleophilic kaolin-based hollow fibre ceramic membrane oil flux and separation efficiency	158
Table 6.7	Suggested optimum responses for Superhydrophobic-superoleophilic kaolin-based hollow fibre ceramic membrane oil separation efficiency and oil flux	160
Table 6.8	Confirmatory test for optimum responses of Oil flux and Oil Separation Efficiency	162

LIST OF FIGURES

FIGURE NO.	TITLE	PAGE
Figure 1.1	Summary of overall thesis structure	10
Figure 2.1	Basic mechanism of coagulation process	19
Figure 2.2	Schematic illustration of the preparation process of superhydrophobic and super-oleophilic copper mesh, and the application in oil-water separation	27
Figure 2.3	Schematic illustration of the MF/PPy sponge fabrication process	29
Figure 2.4	Schematic process of coating textile for oil-water separation	31
Figure 2.5	Schematic construction process of superhydrophobic-superoleophilic stainless steel mesh	32
Figure 2.6	Basic membrane separation set-up displaying the feed, permselective membrane and the feed, permeate and retentate flows	33
Figure 2.7	Schematics of a dead-end (a) and cross-flow (b) filtration modes	34
Figure 2.8	Timeline of ceramic membrane applications	39
Figure 2.9	Ceramic membrane structural illustration	40
Figure 2.10	The schematic illustration of slip casting	43
Figure 2.11	Tape-casting method of ceramic fabrication process	45
Figure 2.12	Schematic illustration of the pressing method of ceramic fabrication	47
Figure 2.13	Extrusion method of ceramic fabrication	49
Figure 2.14	Kinds of spinneret (A) Double orifice, (B) Triple orifice and (C) Quadruple orifice	51
Figure 2.15	Schematic illustration of hollow fibre ceramic membranes fabrication process	51
Figure 2.16	Different type of membrane structural formation based on the different demixing process	53
Figure 2.17	The structural features of kaolinite	56

Figure 2.18	Water contact angles of hydrophilic, hydrophobic and superhydrophobic substrates.	59
Figure 2.19	Schematic illustration of liquid droplets using Young's (a), Wenzel (b) and Baxter-Cassie (c) models .	61
Figure 2.20	Bonding mechanism of silane coupling agent onto the ceramic membrane surface	65
Figure 2.21	Schematic illustration of immersion technique of surface modification	67
Figure 2.22	Schematic illustration of the CVD technique of surface modification	68
Figure 2.23	Sol-gel grafting process of the ceramic membrane	71
Figure 2.24	Some profiles of surface response generated from a quadratic model	78
Figure 3.1	Research flow chart	82
Figure 3.2	Schematic illustration of kaolin dope preparation, spinning process and sintering process	90
Figure 3.3	Sintering temperature profile for the Kaolin-based hollow fibre membrane	91
Figure 3.4	Schematic diagram showing the preparation of a silica sol-gel solution	95
Figure 3.5	Schematic illustration for reaction route of sol-gel preparation and coating procedure to fabricate superhydrophobic silica film on the ceramic membrane surface	97
Figure 3.6	The oil water separation set-up used in this study	102
Figure 4.1	FESEM images of (a) MK and (b) NK clay (powder	108
Figure 4.2	PSA (a), XRD (b), TG/DTA (c) and N ₂ adsorption-desorption isotherms and related BET surface areas (d) of the MK and NK clay samples	110
Figure 4.3	Viscosity of the MK and NK dope suspensions at shear rates between 1 and 100 s ⁻¹ containing 34, and 37 wt.%	111
Figure 4.4	SEM images of the kaolin membranes at different kaolin contents (MKM-37 & 34 wt. %; NKM-37 & 34 wt. %) and sintering temperatures of 1350 °C (F-stands for finger like voids while S-stands for sponge like voids).	113
Figure 4.5	Mechanical strength of the membranes (sintered at 1350 °C) with respect to loading composition (Average of 5	

	mechanical strength values was taken for each composition)	114
Figure 4.6	Overall cross-section SEM images of the kaolin membranes at different kaolin powder contents (MKM-37 & 34 wt. %; NKM-37 & 34 wt. %) and sintering temperatures (1350 °C, 1400 °C & 1500 °C)	115
Figure 4.7	Magnified cross section of the kaolin membranes at different sintering temperature	116
Figure 4.8	Effect of sintering temperature with respect to mechanical strength	117
Figure 4.9	Porosity of the MKM and NKM with respect to sintering temperature (a) and pore size distribution of the MKM and NKM at loading 34 wt. %	119
Figure 4.10	XRD patterns of the MKM and NKM sintered at 1350 – 1500 °C	121
Figure 4.11	Pure water flux evaluation of the MKM and NKM	122
Figure 5.1	Particle size distribution (a) Alkylated-halide silica sol solution and (b) Zeta potential of the alkylated-halide silica sol solution.	127
Figure 5.2	FESEM morphology of (a) - (f) inner surface and cross-sectional area (1) - (6) of pristine membrane, MTES-coated, FAS-coated, OTMS-coated, TCOS-coated and CTMS-coated membranes respectively.	129
Figure 5.3	Elemental mapping of the (a) pristine (b) MTES-coated (c) FAS-coated (d) OTMS-coated (e) TCOS-coated and (f) CTMS-coated membranes respectively. The respective colour codes below the mapping represent the individual element distribution.	130
Figure 5.4	AFM surface images of (a) pristine (b) MTES-coated (c) FAS-coated (d) OTMS-coated (e) TCOS-coated and (f) CTMS-coated membranes respectively	132
Figure 5.5	FTIR spectra of (a) pristine (b) MTES-coated (c) FAS-coated (d) OTMS-coated (e) TCOS-coated and (f) CTMS-coated membranes respectively.	133
Figure 5.6	XPS spectra of whole survey for (a) pristine, (b) MTES-coated, (c) FAS-coated, (d) OTMS-coated, (e)TCOS-coated and (f) CTMS-coated membranes.	135
Figure 5.7	XPS C1s spectrum for; (a) pristine, (b) MTES-coated, (c) FAS-coated, (d) OTMS-coated, (e) TCOS-coated and (f) CTMS coated membranes.	136

Figure 5.8	Wettability measurement of the pristine and coated membranes (Average of 5 WCA and OCA values was taken for each coated membranes).	139
Figure 5.9	Photographic images of water and oil droplet on the surface of the, a) pristine, b) MTES coated, c) FAS coated, (d) OTMS coated, (e) TCOS coated and (f) CTMS coated kaolin membrane.	140
Figure 5.10	Water adhesion test on various coated kaolin membranes a) MTES-Coated b) FAS-coated c) OTMS-coated d) TCOS-coated and e) CTMS-Coated membrane.	141
Figure 5.11	XRD patterns of the pristine and coated membranes	142
Figure 5.12	Pore size distribution of the pristine and the coated membranes.	143
Figure 6.1	Absorption capacities of: (a) kerosene, hexane and crude oils, (b) pristine and coated membranes at different crude oil concentrations, (c) coated kaolin membrane with respect to different coating cycles at 30 minutes cycle time.	148
Figure 6.2	Mechanism for absorption of oil on the coated porous kaolin membrane substrates.	148
Figure 6.3	Oil flux of the MTES coated kaolin membrane in the filtration of water-in-crude oil emulsions.	149
Figure 6.4	3D and Contour plot of Superhydrophobic-superoleophilic kaolin-based hollow fibre ceramic membrane oil flux	155
Figure 6.5	3D plot for Superhydrophobic-superoleophilic kaolin-based hollow fibre ceramic membrane oil separation efficiency (a) and Contour plot for the Superhydrophobic-superoleophilic kaolin-based hollow fibre ceramic membrane oil separation efficiency (b)	157
Figure 6.6	Normal plot residue of Superhydrophobic-superoleophilic kaolin-based hollow fibre ceramic membrane oil flux (a) and oil separation efficiency (b)	159
Figure 6.7	Overlay plot of optimization of the Superhydrophobic-superoleophilic kaolin-based hollow fibre ceramic membrane	161

LIST OF ABBREVIATIONS

AFM	-	Atomic force microscopy
ANOVA	-	Analysis of variance
CCD	-	Central composite design
CTMS	-	Chlorotrimethylsilane
DOE	-	Design of experiments
DTA	-	Differential thermal analysis
EDS	-	Electron dispersive spectroscopy
FAS	-	Flouroalkylsilane
FESEM	-	Field emission scanning electron microscopy
FTIR	-	Fourier transform infrared spectroscopy
MK	-	Malaysian kaolin
MKM	-	Malaysian kaolin membrane
MTES	-	Methyltriethoxysilane
NK	-	Nigerian kaolin
NKM	-	Nigerian kaolin membrane
OCA	-	Oil contact angle
OTMS	-	Octadecyltrimethoxysilane
RSM	-	Response surface methodology
TCOS	-	Trichloro(octadecyl)silane
TEOS	-	Tetraethylorthoxysilicate
TGA	-	Thermogravimetry analysis
WCA	-	Water contact angle
XPS	-	X-ray photoelectron spectroscopy
XRD	-	X-ray diffraction
XRF	-	X-ray fluorescence
OPW	-	Oilfield produced water

LIST OF SYMBOLS

Δt	-	Operation time
A	-	Area of membrane
f	-	Force
v	-	Velocity
p	-	Pressure
g	-	Gram
J_o	-	Oil flux
Q	-	Oil absorption capacity
m_f	-	Weight of membrane after coating
mL	-	Millilitre
V	-	Volume
$V_{d,final}$	-	Final volume of water
$V_{d,initial}$	-	Initial volume of water
σ_F	-	Bending strength
wt %	-	Weight percentage
mg/L	-	Milgram per liter
P_a	-	Pressure
t	-	Time
L	-	Permeate flux
m_i	-	Weight of membrane before coating
<i>Error (%)</i>	-	Percentage error
Y	-	Response
%	-	Percentage
μm	-	Micrometre
$^{\circ}C$	-	Degree Celsius

CHAPTER 1

INTRODUCTION

1.1 Background of Research

A large volume of oilfield produced water (OPW) is generated as refinery by-product during crude oil production, enhanced oil recovery and exploitation processes which are carried out in both offshore and onshore. The constituents are carcinogenic, toxic and persistent due to the presence of oils, hydrocarbons, dissolved formation minerals, suspended solids, dissolved gases, and injected chemical such as biocides, corrosion inhibitors, emulsion and reverse emulsion breakers [1–3]. Discharge of OPW to the environment can contaminate the quality of drinking water, groundwater, degrade soil, and deplete oxygen [4–6]. Furthermore, the discharge also unavoidably resulted to a great loss of valuable energy resources due to the presence of vast amount of unexplored crude oil in OPW.

Therefore, recovery of oil from OPW is a prerequisite to cost saving method to oil and gas industry as well as the protection of environment. Conventional methods of oil removal are gravity separation, coagulation and air flotation, electrocoagulation and electrostatic separation, microwave and heat treatment methods, oil absorbing, biodegradation, sonication, cyclones filters, sand filter, oxidation, photocatalytic treatment, ozone treatment, electrochemical process, Fenton process, flocculation, adsorption and ion exchange [7–14]. However, the aforementioned techniques have drawback such as less energy efficient, low product quality and creation of huge amount of sludge, high energy cost, requirement of large space for installation, complex separation equipment, use of toxic compounds, and difficult to clean up or recycle. To overcome this limitation, membrane technology has gained interests in treatment of water as well as recovery of oil in oil and gas industries.

Membrane is a selective barrier to separate the solute molecule based on size and charge based interaction. Globally, membrane technology has been implemented in various industrial wastewater treatment and public sewage as well as municipal waste management. The main advantage of membrane technology is low cost, high emulsion separation efficiency, no phase change, no addition of chemicals and simplicity of operation [15–17]. Polymer and ceramic are the common membrane material for industrial scale applications. Polymeric membranes are widely adopted in wide range of versatile applications such as desalination, wastewater treatment and bio-product purification [18, 19]. Ceramic membranes are not well explored but it has the unique characteristics of chemical and thermal stability, excellent resistance to fouling, pressure resistance, long lifetime, ease of cleaning and mechanical stability. Nonetheless, its shortcoming is in its limitation for large scale operation due to high cost. In this regards, recently low-cost raw materials such as kaolin [15–17], natural clay [20–22], waste material such as rice husk ash [23] and fly ash [24–26] corn cob ash [27] and palm oil fuel ash (POFA) [28, 29] are attempted. Among all the ceramic materials used for fabrication of ceramic membrane, kaolin is one of the most used and applied due to its distinct features like abundant availability, cost-effectiveness, low plasticity, ease of processing and high refractory properties to the membrane [30, 31].

Superhydrophobic-superoleophilic materials are preferred in the absorption and recovery of the low surface tension oil molecules. However, ceramic membranes are rich in hydrophilic groups thus tend to absorb water molecules. For the oil molecules absorption/recovery, hydrophobic-oleophilic modification is necessary to the ceramic membranes. Silica sol coating is a single step sol-gel technique which provides a cost effective and simple approach to improve the hydrophobic property of membrane through the decrease of surface free energy. There are several reports on various chemicals used in lowering the surface energy of a substrate which include; organophosphonic acids, steric acids, fatty acids, alkanethiols and organosilanes [32–37]. Organosilanes are the most widely and frequently used chemicals in both research and technology for alignment of proper surface topology and lowering of substrate surface energy through the hydrolytic condensation of tetraethylorthoxysilicate (TEOS) and organosilanes. It is preferred due to its simplicity, creation of superhydrophobic-superoleophilic substrate, cost-effective and less toxic [38].

Most of the organosilane agents used in superhydrophobic-superoleophilic modifications of substrates are methyltriethoxysilane (MTES), fluoroalkylsilane (FAS), octadecyltrimethoxysilane (OTMS), chlorotrimethylsilane (CTMS), and Trichloro(octadecyl)silane (TCOS). And also, these organosilanes are mostly used in surface coating of fabric, filter paper, porous glass, polymeric membranes, nanofibre mats, sponge substrates. Wen *et al.* [39] produced a superhydrophobic surface with water contact angle of 156° on a glass substrate via a sol-gel derived organic-inorganic hybrid emulsion. The hybrid sol-gel was prepared by co-hydrolysis and co-polycondensation reactions of TEOS, MTES and tri(isopropoxy)vinylsilane (TIPVS). Wang *et al.* [40] prepared a silica sol by co-hydrolysis and co-condensation of TEOS and FAS for superhydrophobic fabric, nanofibre mat, filter paper, glass slide and silicon wafer coatings. The coated substrates were reported to have a high degree of water contact angle above 170°. Another group of researchers [41] fabricated a robust superhydrophobic fabric bag for oil absorption and collection of oily water. The fabric bags were immersed into TEOS and OTMS, and ammonia solution, respectively. While SiO₂ nanoparticles functionalized with OTMS on the fabric surfaces. The substrate was reported to possess water contact angle above 150°. Meng and co-workers [42] also fabricated a hydrophobic nano-structure porous glass membrane by deposition of SiO₂ nanoparticles followed by grafting with CTMS on the membrane surface. Zhang and Seeger [43] used chemical vapour deposition of trichloro(octadecyl)silane (TCOS) to produce superhydrophobic-superoleophilic polyester textile material for selective oil absorption application. The resultant material showed a high-water contact angle above 150° demonstrating the superhydrophobic features.

For oil-water separation, ultrafiltration (UF) and microfiltration (MF) are widely deployed for the treatment of oil-water mixture. However, clogging of oil on the membrane pore and surface is a critical issue in filtration of oily wastewater feed, which leads to decline of the flux rate and membrane lifecycle. Hence, membrane modification gained more interest in tailoring of ceramic membranes for versatile applications. Adsorptive membranes are showing promising effect on the removal of pollutant such as metal ions, hydrocarbons and dyes from the contaminated sources.

In ceramic membrane, there are three types of modification process, which are chemical vapour deposition (CVD), immersion, and sol gel method. Immersion method is the simplest method but do not possessed any oil absorption capabilities. Meanwhile, CVD is dangerous method due to the thermal process involved. From literatures, it was reported that modification via sol gel method is always applied for the application of oily wastewater separation. As stated by Pierre (2013), there are many definitions of sol gel process exist. For instance, sol gel process takes into account multicomponent oxides that are homogeneous at the atomic level. In fact, the term “sol gel” is restricted to the gels synthesized from alkoxides in which from colloidal dispersion or from metal alkoxides. In other word, grafting process through sol gel method can be defined as a colloidal route used to synthesize ceramics with an intermediate stage including a sol and/or gel state. Literatures on the absorption of oil using ceramic membranes are less studied. Therefore, this study aimed to recover the oil from produced water using silane functionalized ceramic membranes.

The Jabatan Mineral dan Geosains Malaysia (JMG)) minerals reported that, Peninsular Malaysia produces up to 112 million tons of raw kaolin with three kaolin processing industries all located in Perak state [44]. However, most of the kaolin produced by these three industries are mainly used in production of paper as such less emphasis is given other research related application such as membrane fabrication. On the other hand, the raw material research and development council of Nigeria reported that, Nigeria have about 3 billion metric tons of kaolin deposit [44, 45]. The kaolin clay is mainly used in producing household, office utensils and for pharmaceutical used in the country. Also, they are used in fabricating ceramic utensils, paper, porcelain, white incandescent light bulbs, and paint, among others. The kaolin mineral from Nigeria are used for other purposes but its application for the fabrication of membrane is elusive in literature. This research both used Nigerian kaolin and Malaysian in fabrication of ceramic membrane for recovery of oil from produced water.

1.2 Problem Statement

Membrane fouling is a critical issue in filtration of oil-water emulsion and it causes the decline of the membrane performance and life span. To overcome this limitation, the advent of hybrid absorption combined membrane filtration technology has lured in specific removal of target pollutants. It is based on the mechanism of absorption of feed components with respect to surface functionality of membranes. Thereby, it prevents the blocking of solutes on membrane surface which endured for longer duration filtration. Surface modification are prevalently used to tailor the membranes for better oil-water separation. Literatures on various pollutants such as metal ions, ammonia, arsenic etc. are reported using ceramic membranes [46–48]. However, ceramic membranes are quite expensive and low-cost ceramic materials such as clay and biomass based green silica materials are paid more attention in development of membranes. Hence, this study aimed to develop low cost kaolin based hollow fibre ceramic membrane for the recovery of oil from produced water. Malaysian kaolin based ceramic membranes are used in versatile application for liquid [49] and gas separation [50, 51], owes to chemical and thermal stability. Similarly, Nigerian kaolin has a rich of silica groups required for ceramic membrane and used extensive as additives in cement industries and pharmaceutical applications [52–54]. Hence, the feasibility study of each kaolin was performed to aid comparative analysis for the development of low cost ceramic membranes. The main advantage of utilization of kaolin is low cost, abundant in nature and rich of hydroxyl silica groups. Yet, the conventional ceramic ultrafiltration membrane retains the oil molecules on surface and cause adsorption of oil. Thus, cost effective hybrid kaolin-based absorption membrane was proposed for the recovery of oil from produced water.

Surface hydrophobic modification are prevalently used to tailor the hybrid absorption-based membranes for better oil-water separation. Silane functionalization is the common method to enhance the super hydrophobic (anti-wetting properties) and superoleophilic (wetting properties) properties in ceramic membrane. Meng and co-workers [42] fabricated a hydrophobic nano-structure porous glass membrane by deposition of SiO_2 nanoparticles followed by grafting with chlorotrimethylsilane (CTMS) on the membrane surface for oil absorption. Similarly, Hubadillah et al [55]

used methyltriethoxysilane (MTES) to develop superhydrophobic-superoleophilic ceramic membrane for efficient oil separation. Another group of researchers used 1H,1H,2H,2H-perfluorooctyltriethoxysilane (FAS) to fabricate superhydrophobic ceramic membrane for carbon dioxide capture [50]. Based on the literature, different organosilane agents have significant influence in control of membrane hydrophobic properties. However, mechanistic study of different silane agents on superhydrophobic-superoleophilic ceramic membrane modification is limited for the oil-absorption. Hence, methyltriethoxysilane (MTES), fluoroalkylsilane (FAS), octadecyltrimethoxysilane (OTMS), chlorotrimethylsilane (CTMS) and trichloro(octadecyl)silane (TCOS) were chosen as organosilane agents for the modification of kaolin ceramic membrane to absorb the oil molecules such as hexane, kerosene and crude oil. Moreover, membrane filtration involves hydrodynamic conditions such as oil concentration, cross flow velocity and feed pH. Therefore, it is necessary to optimize the condition for enhanced filtration for longer durations. Response surface methodology is widely recognized tool to optimize the parameter and evaluation of interaction of parameters with minimal experiments. RSM was also used in this study to design the experiments and evaluation of parameters on oil absorption.

To the best of our knowledge, literatures on absorption and filtration of oil using ceramic membranes are limited, owes to ceramic membranes are expensive. Based on the literature, the study is organized into three section and it includes (i) Fabrication of kaolin hollow fibre ceramic membrane via phase inversion/sintering method using Malaysian and Nigerian Kaolin, (ii) Comparison of different organosilane agents (MTES, FAS, OTMS, CTMS and TCOS) on super hydrophobic-oleophilic modification of kaolin ceramic membrane, (iii) Performance evaluation and optimization of process variables on optimized super hydrophobic-oleophilic kaolin hollow fibre ceramic membrane in cross flow filtration set up.

1.3 Research Objectives

- i. To investigate the feasibility of Malaysian and Nigerian kaolin as the main ceramic materials for the fabrication of kaolin-based hollow fibre ceramic membrane at different ceramic loadings and sintering temperatures.
- ii. To examine the effects of different organosilane agents on the superhydrophobic-superoleophilic modification of kaolin-based hollow fibre membrane regarding surface morphology, phase transformation, wettability, surface topology, pore size and porosity distribution.
- iii. To evaluate the filtration performance of the superhydrophobic-superoleophilic kaolin-based hollow fibre membranes and the optimization process variables using response surface methodology (RSM) for the recovery of oil from synthetic OPW.

1.4 Scope of Study

The scope of this research is devised as follows;

- i. Screening and characterization of Nigerian and Malaysian kaolin precursor material for the fabrication of the kaolin-based hollow fibre ceramic membrane. Physiochemical properties of each kaolin were analyzed using particle size analyzer (PSA), thermogravimetry/differential thermal analysis (TG/DTA) and X-ray fluorescence (XRF), X-ray diffraction (XRD) and Brunauer-Emmett-Teller (BET) (to accomplish objective 1)
- ii. Fabrication of Nigerian and Malaysian kaolin based hollow fibre ceramic membrane based using kaolin loading composition of 34 to 37 wt.% and sintering temperature of 1200 to 1500 °C. Comparison of physio-chemical characterization and pure water flux performance of the Nigerian and Malaysian kaolin based hollow fibre ceramic membrane for water filtration studies. Physio-chemical characterization involves field

- emission scanning electron microscopy (FESEM), X-ray diffraction (XRD), atomic force microscopy (AFM), mercury intrusion porosimetry (MIP), contact angle measurements (CA (to accomplish objective 1)
- iii. Superhydrophobic-superoleophilic modification of optimum kaolin based hollow fibre ceramic membrane using different organosilane agents such as methyltriethoxysilane (MTES), fluoroalkylsilane (FAS), octadecyltrimethoxysilane (OTMS), chlorotrimethylsilane chlorotrimethylsilane (CTMS) and trichloro(octadecyl)silane (TCOS) through surface coating technique. Also, evaluation of 1 to 4 coating cycles of the organosilane agents on the kaolin-based hollow fibre ceramic membrane for the superhydrophobic-superoleophilic modification (to accomplish objective 2)
 - iv. Confirmation of individual silane agent modification of kaolin-based hollow fibre ceramic membrane using (FESEM), X-ray diffraction (XRD), energy dispersion spectroscopy (EDS), X-ray photoelectron spectroscopy (XPS), atomic force microscopy (AFM), Fourier transform infrared spectroscopy (FTIR), mercury intrusion porosimetry (MIP), contact angle measurements (CA) (to accomplish objective 2)
 - v. Optimum oil recovery with respect to different silane agent modified kaolin-based hollow fibre ceramic membrane for oil absorption capacity and filtration efficiency using synthetic produced water of oil concentration of 1000 mg/L, 1500 mg/L and 2000 mg/L (to accomplish objective 3)
 - vi. Studying the optimum parameters and conditions of the superhydrophobic-superoleophilic kaolin-based hollow fibre ceramic membrane towards optimization of the oil recovery from synthetic oilfield produced water by set of independent parameters (oil concentration of 50-10000 mg/L, feed pH 4-10 and feed flux of 150-300) with responses of oil flux and oil separation efficiency) (to accomplish objective 3)

1.5 Significant of the Study

Water crisis is a serious issue across the world in the present era. The produced water constitutes of various organic and inorganic substances, which are toxic to environment. The permissible limit of discharge of produced water is 10-15 ppm [56]. Membrane technology has acquired greater interest in production sustainable of quality potable water from industrial wastewater and to meet the strict environment regulation policy. This study contributes to development of cost effective functionalized superhydrophobic-superoleophilic kaolin hollow fibre ceramic membrane for the recovering oil from produced water. It also aids in novel scheme of hybrid absorption combined membrane filtration methodology for the recovery of oil from oily wastewater effluents. Apart, organosilanes functionalization effects also extend the frontier of knowledge on novel approach for modification of kaolin membrane for the effective filtration of low surface tension solutions. The developed prototype may also serve as a low-cost filtration module for the industrial wastewater treatment.

1.6 Thesis Organization

This section describes the organization of the different chapters of this thesis for the fabrication of superhydrophobic-superoleophilic kaolin-based hollow fibre ceramic membrane for the recovery of oil from oilfield produce water. The flow of the entire thesis is depicted in Figure 1.1 bellow.

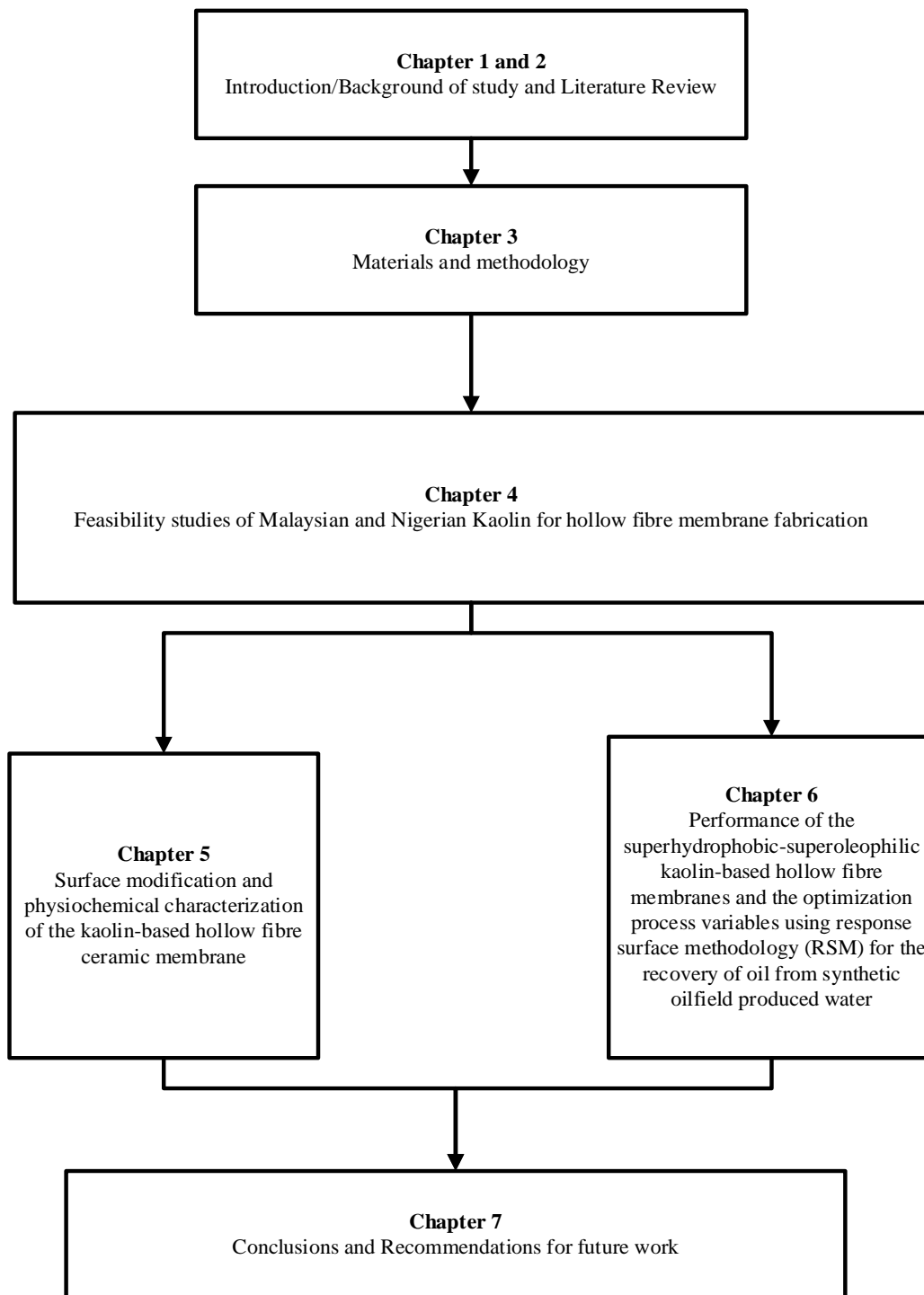


Figure 1.1 Summary of overall thesis structure

Chapter 1 briefly introduce the oilfield produce water contamination in the ecosystem and the current technologies used for the treatment which mainly devoted to this study. It followed by detail problem statement, objectives, scopes and the significance of the study.

Chapter 2 describes the scientific literature review on the produce water, the composition, available methods for the treatment were comprehensively discussed, with much focus on the membrane technology separation techniques for oily wastewater treatments. The adverse effects and the toxicity of oilfield produced water to the ecosystem were also discussed. The chapter also discussed the current available methods of surface modification on different substrate materials as well as their advantages and limitations.

Chapter 3 of the thesis describes the techniques, materials, working principles, modification process, characterization approaches used, membrane setup for oil recovery from produced water and complete operation framework.

Chapter 4 discusses the characteristics features of two different kaolin powders as an alternative material to fabricate kaolin hollow fibre ceramic membrane for the modification and subsequent oil recovery purpose. The characterizations include evaluation of chemical composition of the both kaolin, particle size distribution, crystallinity, morphology and surface area analysis. Afterwards the fabrication and characterization of defect-free high-performance kaolin-based hollow fibre ceramic membrane from the selected kaolin material using various loading composition and sintering temperature. The effect of loading composition, sintering temperature and also the thermal stability, phase transformation as well as pure water flux were evaluated. This chapter also describe the preliminary ceramic membrane fabrication and pure water flux evaluation of the fabricated membrane.

Chapter 5 discusses the processes of surface modification of the kaolin-based hollow fibre ceramic membrane. The effects of various surface coating agents on the membrane surface with respect to its concentration, coating cycle and coating time on the wettability, surface porosity and surface chemical composition.

Chapter 6 presents the potential of the oil recovery by the fabricated superhydrophobic-superoleophilic kaolin-based hollow fibre ceramic membrane was evaluated using the cross-flow filtration system. Also, the factors that influenced the separation efficiency were studied in detail. And it also discussed the optimization study of the superhydrophobic-superoleophilic kaolin-based hollow fibre ceramic membrane performance for 3 significant factors viz oil concentration, feed pH and feed flow rate via response surface research methodology approach. The desirability test was also performed to verify the adequacy of the developed model.

Chapter 7 finally presents the conclusions from present study, suggestions and recommendations for future researcher.

REFERENCES

1. Mohammadi T, Pak A. Effect of calcination temperature of kaolin as a support for zeolite membranes. *Sep Purif Technol* 2003; 30: 241–249.
2. Cui J, Zhang X, Liu H, et al. Preparation and application of zeolite/ceramic microfiltration membranes for treatment of oil contaminated water. *J Memb Sci* 2008; 325: 420–426.
3. Hua FL, Tsang YF, Wang YJ, et al. Performance study of ceramic microfiltration membrane for oily wastewater treatment. *Chem Eng J* 2007; 128: 169–175.
4. Pouloupoulos SG, Voutsas EC, Grigoropoulou HP, et al. Stripping as a pretreatment process of industrial oily wastewater. *J Hazard Mater* 2005; 117: 135–139.
5. Singh V, Purkait MK, Das C. Cross-Flow Microfiltration of Industrial Oily Wastewater: Experimental and Theoretical Consideration. *Sep Sci Technol* 2011; 46: 1213–1223.
6. Bai W, Xu J, Guan M, et al. Preparation of superhydrophobic polyimide microstructural layer on copper mesh for oil/water separation. *J Taiwan Inst Chem Eng* 2019; 95: 71–77.
7. Kumar S, Nandi BK, Guria C, et al. Oil Removal From Produced Water By Ultrafiltration Using Polysulfone Membrane. *Brazilian J Chem Eng* 2017; 34: 583–596.
8. Suresh K, Pugazhenth G. Cross flow microfiltration of oil-water emulsions using clay based ceramic membrane support and TiO₂ composite membrane. *Egypt J Pet* 2017; 26: 679–694.
9. Foght JM, Gutnick DL, Westlake DWS. Effect of Emulsan on Biodegradation of Crude Oil by Pure and Mixed Bacterial Cultures. *Appl Environ Microbiol* 1989; 55: 36–42.
10. Cañizares P, Martínez F, Jiménez C, et al. Coagulation and electrocoagulation of oil-in-water emulsions. *J Hazard Mater* 2008; 151: 44–51.

11. Canizares P, Sáez C, Martínez F, et al. The role of the characteristics of p-Si BDD anodes on the efficiency of wastewater electro-oxidation processes. *Electrochem Solid-State Lett* 2008; 11: 15–19.
12. Pintor, Ariana MA, Vitor JP Vilar, Cidália MS Botelho and RAB. Oil and grease removal from wastewaters: Sorption treatment as an alternative to state-of-the-art technologies. A critical review. *Chem Eng J* 2016; 297: 229–255.
13. Eckenfelder WW, Patoczka J, Watkin AT. Wastewater treatment. *Chem Eng (New York)* 1985; 92: 60–74.
14. Zhou H, Wang H, Niu H, et al. Robust, self-healing superamphiphobic fabrics prepared by two-step coating of fluoro-containing polymer, fluoroalkyl silane, and modified silica nanoparticles. *Adv Funct Mater* 2013; 23: 1664–1670.
15. Hubadillah SK, Othman MHD, Matsuura T, et al. Fabrications and applications of low cost ceramic membrane from kaolin: A comprehensive review. *Ceram Int* 2018; 44: 4538–4560.
16. Bouzera F, Harabi A, Achour S, et al. Porous ceramic supports for membranes prepared from kaolin and dolomite mixtures. *J Eur Ceram Soc* 2006; 26: 1663–1671.
17. Sheikhi M, Arzani M, Mahdavi HR, et al. Kaolinitic clay-based ceramic microfiltration membrane for oily wastewater treatment: Assessment of coagulant addition. *Ceram Int* 2019; 45: 17826–17836.
18. Chang YS, Leow HTL, Ooi BS. Membrane distillation for water recovery and its fouling phenomena. *Journal of Membrane Science and Research* 2020; 6: 107–124.
19. Otitoju TA, Ahmad AL, Ooi BS. Polyvinylidene fluoride (PVDF) membrane for oil rejection from oily wastewater: A performance review. *J Water Process Eng* 2016; 14: 41–59.
20. Jana S, Purkait MK, Mohanty K. Preparation and characterization of low-cost ceramic microfiltration membranes for the removal of chromate from aqueous solutions. *Appl Clay Sci* 2010; 47: 317–324.
21. Saffaj N, Persin M, Younsi SA, et al. Elaboration and characterization of microfiltration and ultrafiltration membranes deposited on raw support prepared from natural Moroccan clay: Application to filtration of solution containing dyes and salts. *Appl Clay Sci* 2006; 31: 110–119.

22. Mohamed Bazin M, Ahmad N, Nakamura Y. Preparation of porous ceramic membranes from Sayong ball clay. *J Asian Ceram Soc* 2019; 7: 417–425.
23. Hossain SKS, Mathur L, Roy PK. Rice husk/rice husk ash as an alternative source of silica in ceramics: A review. *Journal of Asian Ceramic Societies* 2018; 6: 299–313.
24. Hubadillah SK, Othman MHD, Harun Z, et al. A novel green ceramic hollow fiber membrane (CHFM) derived from rice husk ash as combined adsorbent-separator for efficient heavy metals removal. *Ceram Int* 2017; 43: 4716–4720.
25. Zou D, Qiu M, Chen X, et al. One step co-sintering process for low-cost fly ash based ceramic microfiltration membrane in oil-in-water emulsion treatment. *Sep Purif Technol* 2018; 210: 511–520.
26. Alfiyan B, Susanto H. Utilization of fly ash as ceramic support mixture for the synthesis of zeolite pervaporation membrane. *Adv Mater Res* 2014; 896: 74–77.
27. Kamarudin NH, Harun Z, Othman MHD, et al. Waste environmental sources of metakaolin and corn cob ash for preparation and characterisation of green ceramic hollow fibre membrane (h-MCa) for oil-water separation. *Ceram Int* 2020; 46: 1512–1525.
28. Tai ZS, Othman MHD, Hubadillah SK, et al. Low cost palm oil fuel ash based ceramic membranes for oily water separation. *Malaysian J Fundam Appl Sci* 2018; 14: 419–424.
29. Tai ZS, Hubadillah SK, Othman MHD, et al. Influence of pre-treatment temperature of palm oil fuel ash on the properties and performance of green ceramic hollow fiber membranes towards oil/water separation application. *Sep Purif Technol* 2019; 222: 264–277.
30. Hubadillah SK, Harun Z, Othman MHD, et al. Preparation and characterization of low cost porous ceramic membrane support from kaolin using phase inversion/sintering technique for gas separation: Effect of kaolin content and non-solvent coagulant bath. *Chem Eng Res Des* 2016; 112: 24–35.
31. Mittal P, Jana S, Mohanty K. Synthesis of low-cost hydrophilic ceramic-polymeric composite membrane for treatment of oily wastewater. *Desalination* 2011; 282: 54–62.
32. Qi Y, Yang Z, Chen T, et al. Fabrication of superhydrophobic surface with desirable anti-icing performance based on micro/nano-structures and organosilane groups. *Appl Surf Sci* 2020; 501: 0169–4332.

33. Van Alsten JG. Self-assembled monolayers on engineering metals: Structure, derivatization, and utility. *Langmuir* 1999; 15: 7605–7614.
34. Gao W, Dickinson L, Grozinger C, et al. Self-assembled monolayers of alkylphosphonic acids on metal oxides. *Langmuir* 1996; 12: 6429–6435.
35. Godin M, Williams PJ, Tabard-Cossa V, et al. Surface stress, kinetics, and structure of alkanethiol self-assembled monolayers. *Langmuir* 2004; 20: 7090–7096.
36. Jadhav SA. Self-assembled monolayers (SAMs) of carboxylic acids: An overview. *Cent Eur J Chem* 2011; 9: 369–378.
37. Qu M, Ma X, He J, et al. Facile Selective and Diverse Fabrication of Superhydrophobic, Superoleophobic-Superhydrophilic and Superamphiphobic Materials from Kaolin. *ACS Appl Mater Interfaces* 2017; 9: 1011–1020.
38. Li L, Li B, Dong J, et al. Roles of silanes and silicones in forming superhydrophobic and superoleophobic materials. *J Mater Chem A* 2016; 4: 13677–13725.
39. Wen XF, Wang K, Pi PH, et al. Organic-inorganic hybrid superhydrophobic surfaces using methyltriethoxysilane and tetraethoxysilane sol-gel derived materials in emulsion. *Appl Surf Sci* 2011; 258: 991–998.
40. Wang H, Fang J, Cheng T, et al. One-step coating of fluoro-containing silica nanoparticles for universal generation of surface superhydrophobicity. *Chem Commun* 2008; 877–879.
41. Li J, Yan L, Tang X, et al. Robust Superhydrophobic Fabric Bag Filled with Polyurethane Sponges Used for Vacuum-Assisted Continuous and Ultrafast Absorption and Collection of Oils from Water. *Adv Mater Interfaces* 2016; 3: 3–10.
42. Meng T, Xie R, Ju XJ, et al. Nano-structure construction of porous membranes by depositing nanoparticles for enhanced surface wettability. *J Memb Sci* 2013; 427: 63–72.
43. Zhang J, Seeger S. Polyester materials with superwetting silicone nanofilaments for oil/water separation and selective oil absorption. *Adv Funct Mater* 2011; 21: 4699–4704.
44. Yahya H, Othman MR, Ahmad ZA. Study on the potential of kaolinitic clay from Perak state, malaysia for aluminosilicate ceramic ball. *Mater Sci Forum* 2016; 840: 112–117.

45. RMRDC, <https://www.rmrdc.gov.ng/>.
46. Hubadillah SK, Othman MHD, Ismail AF, et al. A low cost hydrophobic kaolin hollow fiber membrane (h-KHFM) for arsenic removal from aqueous solution via direct contact membrane distillation. *Sep Purif Technol* 2019; 31–39.
47. Adam MR, Matsuura T, Othman MHD, et al. Feasibility study of the hybrid adsorptive hollow fibre ceramic membrane (HFCM) derived from natural zeolite for the removal of ammonia in wastewater. *Process Saf Environ Prot* 2019; 122: 378–385.
48. Yusof MSM, Othman MHD, Mustafa A, et al. Feasibility study of cadmium adsorption by palm oil fuel ash (POFA)-based low-cost hollow fibre zeolitic membrane. *Environ Sci Pollut Res* 2018; 25: 21644–21655.
49. Hubadillah SK, Othman MHD, Ismail AF, et al. The feasibility of kaolin as main material for low cost porous ceramic hollow fibre membrane prepared using combined phase inversion and sintering technique. *J Teknol* 2017; 79: 35–39.
50. Abdulhameed MA, Othman MHD, Ismail AF, et al. Carbon dioxide capture using a superhydrophobic ceramic hollow fibre membrane for gas-liquid contacting process. *J Clean Prod* 2017; 140: 1731–1738.
51. Abdulhameed MA, Othman MHD, Joda HNA Al, et al. Fabrication and characterization of affordable hydrophobic ceramic hollow fibre membrane for contacting processes. *J Adv Ceram* 2017; 6: 330–340.
52. Edomwonyi-Otu LC, Aderemi BO, Ahmed AS, et al. Influence of Thermal Treatment on Kankara Kaolin. *Opticon1826* 2013; 0: 1–5.
53. Christian ZC. Chemical investigation of kankara kaolin for its pharmaceutical applications. *Open Access J Transl Med Res* 2018; 2: 2–4.
54. Ayeni O. *Performance of a Nigerian Metakaolin-Based Geopolymer as a Sustainable Building Material (Doctoral dissertation)*. African University of Science and Technology Research, 2017.
55. Hubadillah SK, Kumar P, Dzarfan Othman MH, et al. A low cost, superhydrophobic and superoleophilic hybrid kaolin-based hollow fibre membrane (KHFM) for efficient adsorption-separation of oil removal from water. *RSC Adv* 2018; 8: 2986–2995.

56. Mendoza SMV, Moreno EA, Fajardo CAG, et al. Liquid-liquid continuous extraction and fractional distillation for the removal of organic compounds from the wastewater of the oil industry. *Water (Switzerland)* 2019; 11: 1452.
57. Jiménez S, Micó MM, Arnaldos M, et al. State of the art of produced water treatment. *Chemosphere* 2018; 192: 186–208.
58. Yang M. Measurement of oil in produced water. *Prod Water* 2011; 57–88.
59. Ebrahimi M, Kerker S, Schmitz O, et al. Evaluation of the fouling potential of ceramic membrane configurations designed for the treatment of oilfield produced water. *Sep Sci Technol* 2018; 53: 349–363.
60. Al-Ghouti MA, Al-Kaabi MA, Ashfaq MY, et al. Produced water characteristics, treatment and reuse: A review. *J Water Process Eng* 2019; 28: 222–239.
61. Nasiri M, Jafari I, Parniankhoy B. Oil and Gas Produced Water Management: A Review of Treatment Technologies, Challenges, and Opportunities. *Chem Eng Commun* 2017; 204: 990–1005.
62. Veil JA, Puder MG, Elcock D, et al. A White Paper Describing Produced Water from Production of Crude Oil, Natural Gas, and Coal Bed Methane. *US Dep Energy, Energy Technol Lab* 2004; 87.
63. Tibbetts PJC, Buchanan IT, Gawel LJ, et al. A Comprehensive Determination of Produced Water Composition. *Prod Water* 1992; 97–112.
64. Campos JC, Borges RMH, Oliveira Filho AM, et al. Oilfield wastewater treatment by combined microfiltration and biological processes. *Water Res* 2002; 36: 95–104.
65. Fakhru'l-Razi A, Pendashteh A, Abdullah LC, et al. Review of technologies for oil and gas produced water treatment. *J Hazard Mater* 2009; 170: 530–551.
66. Allen EW. Process water treatment in Canada's oil sands industry: II. A review of emerging technologies. *J Environ Eng Sci* 2008; 7: 499–524.
67. Fakhru'l-Razi A, Pendashteh A, Abdullah LC, et al. Review of technologies for oil and gas produced water treatment. *Journal of Hazardous Materials* 2009; 170: 530–551.
68. Szép A, Kohlheb R. Water treatment technology for produced water. *Water Sci Technol* 2010; 62: 2372–2380.

69. Diya'Uddeen BH, Daud WMAW, Abdul Aziz AR. Treatment technologies for petroleum refinery effluents: A review. *Process Saf Environ Prot* 2011; 89: 95–105.
70. Fouladi Tajar A, Kaghazchi T, Soleimani M. Adsorption of cadmium from aqueous solutions on sulfurized activated carbon prepared from nut shells. *J Hazard Mater* 2009; 165: 1159–1164.
71. Feng C, Khulbe KC, Matsuura T, et al. Recent progress in zeolite/zeotype membranes. *Journal of Membrane Science and Research* 2015; 1: 49–72.
72. Masooleh MS, Bazgir S, Tamizifar M, et al. Adsorption of petroleum hydrocarbons on organoclay Archive of SID. *J Appl Chem Res* 2010; 4: 19–23.
73. Duan J, Higuchi M, Horike S, et al. High CO₂/CH₄ and C₂ hydrocarbons/CH₄ selectivity in a chemically robust porous coordination polymer. *Adv Funct Mater* 2013; 23: 3525–3530.
74. Ngang HP, Ahmad AL, Low SC, et al. Adsorption-desorption study of oil emulsion towards thermo-responsive PVDF/SiO₂-PNIPAM composite membrane. *J Environ Chem Eng* 2017; 5: 4471–4482.
75. Apul OG, Karanfil T. Adsorption of synthetic organic contaminants by carbon nanotubes: A critical review. *Water Research* 2015; 68: 34–55.
76. Nasiri M, Jafari I. Produced water from oil-gas plants: A short review on challenges and opportunities. *Period Polytech Chem Eng* 2017; 61: 73–81.
77. Saleh TA, Gupta VK. Processing methods, characteristics and adsorption behavior of tire derived carbons: A review. *Advances in Colloid and Interface Science* 2014; 211: 93–101.
78. Stavropoulos GG, Samaras P, Sakellariopoulos GP. Effect of activated carbons modification on porosity, surface structure and phenol adsorption. *J Hazard Mater* 2008; 151: 414–421.
79. Hadi P, Xu M, Ning C, et al. A critical review on preparation, characterization and utilization of sludge-derived activated carbons for wastewater treatment. *Chemical Engineering Journal* 2015; 260: 895–906.
80. Rakić V, Rac V, Krmar M, et al. The adsorption of pharmaceutically active compounds from aqueous solutions onto activated carbons. *J Hazard Mater* 2015; 282: 141–149.

81. Souza JS, Paiva MKN, Farias FPM, et al. Hydrocyclone applications in produced water: a steady-state numerical analysis. *Brazilian J Pet Gas* 2012; 6: 133–143.
82. Sinker A. Produced Water Treatment Using Hydrocyclones : Theory and practical application. In: *14th Annual International Petroleum Environmental Conference*. 2007.
83. Lohne K. Separation of solids from produced water using hydrocyclone technology. *Chem Eng Res Des* 1994; 72: 169–175.
84. Bulk D. EPCON Dual Compact Flotation Unit.
85. Youyi Z, Qiang Z, Yabin N. Development of a New High Effective Flotation Device Used for Water Treatment. In: *Proceedings - SPE International Symposium on Oilfield Chemistry*. 2001.
86. Yu L, Han M, He F. A review of treating oily wastewater. *Arabian Journal of Chemistry* 2017; 10: S1913–S1922.
87. Zouboulis AI, Avranas A. Treatment of oil-in-water emulsions by coagulation and dissolved-air flotation. *Colloids Surfaces A Physicochem Eng Asp* 2000; 172: 153–161.
88. Wang LK, Shammas NK, Selke WA, et al. (eds). *Flotation Technology*. Humana Press.
89. Kundu P, Mishra IM. Treatment and reclamation of hydrocarbon-bearing oily wastewater as a hazardous pollutant by different processes and technologies: A state-of-the-art review. *Reviews in Chemical Engineering* 2019; 35: 73–108.
90. Moosai R, Dawe RA. Gas attachment of oil droplets for gas flotation for oily wastewater cleanup. *Sep Purif Technol* 2003; 33: 303–314.
91. Jameson GJ. Hydrophobicity and floc density in induced-air flotation for water treatment. *Colloids Surfaces A Physicochem Eng Asp* 1999; 151: 269–281.
92. Yan Y, Jameson GJ. Application of the Jameson Cell technology for algae and phosphorus removal from maturation ponds. *Int J Miner Process* 2004; 73: 23–28.
93. Zabel TF. Flotation in Water Treatment. In: *Innovations in Flotation Technology*. Springer Netherlands, pp. 431–454.
94. Li P, Tsuge H. Water Treatment by Induced Air Flotation Using Microbubbles. *J Chem Eng Japan* 2006; 39: 896–903.

95. Nazirah Wan Ikhsan S, Yusof N, Aziz F, et al. Malaysian Journal of Analytical Sciences a Review of Oilfield Wastewater Treatment Using Membrane Filtration Over Conventional Technology. *Malaysian J Anal Sci* 2017; 21: 643–658.
96. Zeng Y, Yang C, Zhang J, et al. Feasibility investigation of oily wastewater treatment by combination of zinc and {PAM} in coagulation/flocculation. *J Hazard Mater* 2007; 147: 991–996.
97. Cong LN, Liu YJ, Hao B. Synthesis and application of PAZSC in oily wastewater treatment. *Chem Eng* 2011; 1: 5–9.
98. Eweis JB, Ergas SJ, Chang DPY, et al. Bioremediation principles. *Bioremediation Princ.*
99. Kriipsalu M, Marques M, Nammari DR, et al. Bio-treatment of oily sludge: The contribution of amendment material to the content of target contaminants, and the biodegradation dynamics. *J Hazard Mater* 2007; 148: 616–622.
100. Yu L, Han M, He F. A review of treating oily wastewater. *Arab J Chem* 2017; 10: S1913--S1922.
101. Kulkarni SJ. Biological Treatment of Petroleum Wastewater: A Review on Research and Studies. *Int J Pet Petrochemical Eng*; 2.
102. Cerqueira VS, Hollenbach EB, Maboni F, et al. Biodegradation potential of oily sludge by pure and mixed bacterial cultures. *Bioresour Technol* 2011; 102: 11003–11010.
103. Song H, Zhou L, Zhang L, et al. Construction of a whole-cell catalyst displaying a fungal lipase for effective treatment of oily wastewaters. *J Mol Catal B Enzym* 2011; 71: 166–170.
104. Khondee N, Tathong S, Pinyakong O, et al. Airlift bioreactor containing chitosan-immobilized *Sphingobium* sp. P2 for treatment of lubricants in wastewater. *J Hazard Mater* 2012; 213–214: 466–473.
105. Xie W, Zhong L, Chen J. Treatment of slightly polluted wastewater in an oil refinery using a biological aerated filter process. *Wuhan Univ J Nat Sci* 2007; 12: 1094–1098.
106. Santo CE, Vilar VJP, Bhatnagar A, et al. Biological treatment by activated sludge of petroleum refinery wastewaters. *Desalin Water Treat* 2013; 51: 6641–6654.

107. Ahmad AL, Ooi BS, Mohammad AW, et al. Development of a highly hydrophilic nanofiltration membrane for desalination and water treatment. *Desalination* 2004; 168: 215–221.
108. Baker RW. *Membrane Technology and Applications*. 2012.
109. Mulder M. Basic Principles of Membrane Technology. *Zeitschrift für Physikalische Chemie* 1998; 72: 564.
110. Vatai GN, Krstic DM, Koris AK, et al. Ultrafiltration of oil-in-water emulsion: Comparison of ceramic and polymeric membranes. *Desalin Water Treat* 2009; 3: 162–168.
111. Khemakhem S, Larbot A, Ben R. New Ceramic Microfiltration Membranes from Tunisian Natural Materials: Application for the Cuttlefish Effluents Treatment. *Ceram Mater* 2010; 35: 55–61.
112. Chakrabarty B, Ghoshal AK, Purkait MK. Cross-flow ultrafiltration of stable oil-in-water emulsion using polysulfone membranes. *Chem Eng J* 2010; 165: 447–456.
113. Pendashteh AR, Abdullah LC, Fakhru’L-Razi A, et al. Evaluation of membrane bioreactor for hypersaline oily wastewater treatment. *Process Saf Environ Prot* 2012; 90: 45–55.
114. Kayvani Fard A, Bukenhoudt A, Jacobs M, et al. Novel hybrid ceramic/carbon membrane for oil removal. *J Memb Sci* 2018; 559: 42–53.
115. Arnot T., Field R., Koltuniewicz A. Cross-flow and dead-end microfiltration of oily-water emulsions: Part II. Mechanisms and modelling of flux decline. *J Memb Sci* 2000; 169: 1–15.
116. Dickhout JM, Moreno J, Biesheuvel PM, et al. Produced water treatment by membranes: A review from a colloidal perspective. *J Colloid Interface Sci* 2017; 487: 523–534.
117. Singh AK, Singh JK. Fabrication of durable super-repellent surfaces on cotton fabric with liquids of varying surface tension: Low surface energy and high roughness. *Appl Surf Sci* 2017; 416: 639–648.
118. Su B, Tian Y, Jiang L. Bioinspired Interfaces with Superwettability: From Materials to Chemistry. *Journal of the American Chemical Society* 2016; 138: 1727–1748.

119. Xiang Y, Pang Y, Jiang X, et al. One-step fabrication of novel superhydrophobic and superoleophilic sponge with outstanding absorbency and flame-retardancy for the selective removal of oily organic solvent from water. *Appl Surf Sci* 2018; 428: 338–347.
120. Gao N, Xu Z-K. Ceramic membranes with mussel-inspired and nanostructured coatings for water-in-oil emulsions separation. *Sep Purif Technol* 2019; 212: 737–746.
121. Cao H, Gu W, Fu J, et al. Preparation of superhydrophobic/oleophilic copper mesh for oil-water separation. *Appl Surf Sci* 2017; 412: 599–605.
122. Li J, Yan L, Hu W, et al. Facile fabrication of underwater superoleophobic TiO₂ coated mesh for highly efficient oil/water separation. *Colloids Surfaces A Physicochem Eng Asp* 2016; 489: 441–446.
123. Wang C, Yao T, Wu J, et al. Facile approach in fabricating superhydrophobic and superoleophilic surface for water and oil mixture separation. *ACS Appl Mater Interfaces* 2009; 1: 2613–2617.
124. Ge B, Zhang Z, Zhu X, et al. A superhydrophobic/superoleophilic sponge for the selective absorption oil pollutants from water. *Colloids Surfaces A Physicochem Eng Asp* 2014; 457: 397–401.
125. Yang C, Wang Y, Fu H, et al. A stable eco-friendly superhydrophobic/superoleophilic copper mesh fabricated by one-step immersion for efficient oil/water separation. *Surf Coatings Technol* 2019; 359: 108–116.
126. Hamzah N, Leo CP, Ooi BS. Superhydrophobic PVDF/TiO₂-SiO₂ Membrane with Hierarchical Roughness in Membrane Distillation for Water Recovery from Phenolic Rich Solution Containing Surfactant. *Chinese J Polym Sci (English Ed)* 2019; 37: 609–616.
127. Horiuchi Y, Fujiwara K, Kamegawa T, et al. An efficient method for the creation of a superhydrophobic surface: ethylene polymerization over self-assembled colloidal silica nanoparticles incorporating single-site Cr-oxide catalysts. *J Mater Chem* 2011; 21: 8543.
128. Tserepi AD, Vlachopoulou M-E, Gogolides E. Nanotexturing of poly(dimethylsiloxane) in plasmas for creating robust super-hydrophobic surfaces. *Nanotechnology* 2006; 17: 3977–3983.

129. Ghosh N, Singh AV, Vaidya AA. Water-Based Layer-by-Layer Surface Chemical Modification of Biomimetic Materials: Oil Repellency. *ACS Appl Mater Interfaces* 2013; 5: 8869–8874.
130. Cha SC, Her EK, Ko TJ, et al. Thermal stability of superhydrophobic, nanostructured surfaces. *J Colloid Interface Sci* 2013; 391: 152–157.
131. Song X, Zhai J, Wang Y, et al. Fabrication of Superhydrophobic Surfaces by Self-Assembly and Their Water-Adhesion Properties. *J Phys Chem B* 2005; 109: 4048–4052.
132. Xiang H, Zhang L, Wang Z, et al. Multifunctional polymethylsilsesquioxane (PMSQ) surfaces prepared by electrospinning at the sol–gel transition: Superhydrophobicity, excellent solvent resistance, thermal stability and enhanced sound absorption property. *J Colloid Interface Sci* 2011; 359: 296–303.
133. Wang S, Guo X, Xie Y, et al. Preparation of superhydrophobic silica film on Mg-Nd-Zn-Zr magnesium alloy with enhanced corrosion resistance by combining micro-arc oxidation and sol-gel method. *Surf Coatings Technol* 2012; 213: 192–201.
134. Feng L, Zhang Z, Mai Z, et al. A Super-Hydrophobic and Super-Oleophilic Coating Mesh Film for the Separation of Oil and Water. *Angew Chemie* 2004; 116: 2046–2048.
135. Pi P, Hou K, Zhou C, et al. Superhydrophobic Cu₂S@Cu₂O film on copper surface fabricated by a facile chemical bath deposition method and its application in oil-water separation. *Appl Surf Sci* 2017; 396: 566–573.
136. Zhu H, Guo Z. A Superhydrophobic Copper Mesh with Microrod Structure for Oil–Water Separation Inspired from Ramee Leaf. *Chem Lett* 2014; 43: 1645–1647.
137. Cao H, Fu J, Liu Y, et al. Facile design of superhydrophobic and superoleophilic copper mesh assisted by candle soot for oil water separation. *Colloids Surfaces A Physicochem Eng Asp* 2018; 537: 294–302.
138. Wang H, Wang E, Liu Z, et al. A novel carbon nanotubes reinforced superhydrophobic and superoleophilic polyurethane sponge for selective oil-water separation through a chemical fabrication. *J Mater Chem A* 2015; 3: 266–273.

139. Li B, Li L, Wu L, et al. Durable superhydrophobic/superoleophilic polyurethane sponges inspired by mussel and lotus leaf for the selective removal of organic pollutants from water. *Chempluschem* 2014; 79: 850–856.
140. Liu F, Sun F, Pan Q. Highly compressible and stretchable superhydrophobic coating inspired by bio-adhesion of marine mussels. *J Mater Chem A* 2014; 2: 11365–11371.
141. Huang S. Mussel-inspired one-step copolymerization to engineer hierarchically structured surface with superhydrophobic properties for removing oil from water. *ACS Appl Mater Interfaces* 2014; 6: 17144–17150.
142. Chen J, You H, Xu L, et al. Facile synthesis of a two-tier hierarchical structured superhydrophobic-superoleophilic melamine sponge for rapid and efficient oil/water separation. *J Colloid Interface Sci* 2017; 506: 659–668.
143. Zhang J, Seeger S. Polyester Materials with Superwetting Silicone Nanofilaments for Oil/Water Separation and Selective Oil Absorption. *Adv Funct Mater* 2011; 21: 4699–4704.
144. Xue CH, Ji PT, Zhang P, et al. Fabrication of superhydrophobic and superoleophilic textiles for oil-water separation. *Appl Surf Sci* 2013; 284: 464–471.
145. Zareei Pour F, Karimi H, Madadi Avargani V. Preparation of a superhydrophobic and superoleophilic polyester textile by chemical vapor deposition of dichlorodimethylsilane for Water–Oil separation. *Polyhedron* 2019; 159: 54–63.
146. Xiang M, Jiang M, Zhang Y, et al. Fabrication of a novel superhydrophobic and superoleophilic surface by one-step electrodeposition method for continuous oil/water separation. *Appl Surf Sci* 2018; 434: 1015–1020.
147. Wang J, Zou Z, Geng G. Construction of superhydrophobic copper film on stainless steel mesh by a simple liquid phase chemical reduction for efficient oil/water separation. *Appl Surf Sci* 2019; 486: 394–404.
148. Chen C, Weng D, Mahmood A, et al. Separation Mechanism and Construction of Surfaces with Special Wettability for Oil/Water Separation. *ACS Appl Mater Interfaces* 2019; 11: 11006–11027.
149. Feng L, Zhang Z, Mai Z, et al. A super-hydrophobic and super-oleophilic coating mesh film for the separation of oil and water. *Angew Chemie - Int Ed* 2004; 43: 2012–2014.

150. Oikawa Y, Minami T, Mayama H, et al. Preparation of self-organized porous anodic niobium oxide microcones and their surface wettability. *Acta Mater* 2009; 57: 3941–3946.
151. Steele A, Bayer I, Loth E. Inherently Superoleophobic Nanocomposite Coatings by Spray Atomization. *Nano Lett* 2009; 9: 501–505.
152. Pan Q, Wang M, Wang H. Separating small amount of water and hydrophobic solvents by novel superhydrophobic copper meshes. *Appl Surf Sci* 2008; 254: 6002–6006.
153. Wang J, Wen Y, Hu J, et al. Fine Control of the Wettability Transition Temperature of Colloidal-Crystal Films: From Superhydrophilic to Superhydrophobic. *Adv Funct Mater* 2007; 17: 219–225.
154. Li ZY, Xie S, Jiang G, et al. Bioremediation of Offshore Oily Drilling Fluids. *Energy Sources, Part A Recover Util Environ Eff* 2015; 37: 1680–1687.
155. Liu F, Ma M, Zang D, et al. Fabrication of superhydrophobic/superoleophilic cotton for application in the field of water/oil separation. *Carbohydr Polym* 2014; 103: 480–487.
156. Yang H, Pi P, Cai ZQ, et al. Facile preparation of super-hydrophobic and superoleophilic silica film on stainless steel mesh via sol-gel process. *Appl Surf Sci* 2010; 256: 4095–4102.
157. Zhang J, Huang W, Han Y. A Composite Polymer Film with both Superhydrophobicity and Superoleophilicity. *Macromol Rapid Commun* 2006; 27: 804–808.
158. Li K. *Ceramic Membranes for Separation and Reaction*. John Wiley & Sons, Ltd.
159. Pan Y, Nilges MJ. Electron paramagnetic resonance spectroscopy: Basic principles, experimental techniques and applications to earth and planetary sciences. *Rev Mineral Geochemistry* 2014; 78: 655–690.
160. Staude E. Marcel Mulder: Basic Principles of Membrane Technology, Kluwer Academic Publishers, Dordrecht, Boston, London, 1991, ISBN 0-7923-0978-2, 363 Seiten, Preis: DM 200,-. *Berichte der Bunsengesellschaft für Phys Chemie*.
161. Zhang W, Shi Z, Zhang F, et al. Superhydrophobic and superoleophilic PVDF membranes for effective separation of water-in-oil emulsions with high flux. *Adv Mater* 2013; 25: 2071–2076.

162. Brunetti A, Bernardo P, Drioli E, et al. Membrane Engineering: Progress and Potentialities in Gas Separations. In: *Membrane Gas Separation*. 2010.
163. Fukumoto LR, Delaquis P GB. Microfiltration and Ultrafiltration Ceramic Membranes for Apple Juice Clarification. *J Food Sci* 1998; 63: 845–850.
164. Abadi SRH, Sebzari MR, Hemati M, et al. Ceramic membrane performance in microfiltration of oily wastewater. *Desalination* 2011; 265: 222–228.
165. Ma C, Li Y, Nian P, et al. Fabrication of oriented metal-organic framework nanosheet membrane coated stainless steel meshes for highly efficient oil/water separation. *Sep Purif Technol* 2019; 229: 115835.
166. Li Z, Xu ZL, Huang BQ, et al. Three-channel stainless steel hollow fiber membrane with inner layer modified by nano-TiO₂ coating method for the separation of oil-in-water emulsions. *Sep Purif Technol* 2019; 222: 75–84.
167. Wang J-C, Lou H, Cui Z-H, et al. Fabrication of porous polyacrylamide/polystyrene fibrous membranes for efficient oil-water separation. *Sep Purif Technol* 2019; 222: 278–283.
168. Le NL, Nunes SP. Materials and membrane technologies for water and energy sustainability. *Sustainable Materials and Technologies* 2016; 7: 1–28.
169. Biron D da S, Santos V dos, Zeni M. *Ceramic membranes applied in separation processes*.
170. Noble R, S. Alexander Stern. Membrane Separations Technology Principles and Applications. *Elsevier* 1989; 4: 731.
171. Li K. *Ceramic Membranes for Separation and Reaction*. John Wiley & Sons, 2007.
172. Finnigan T, Shackleton R, Skudder P. Using ceramic microfiltration for the filtration of beer and recovery of extract. *Filtr Sep* 1989; 26: 198–200.
173. Hanley HJM. Thermal transpiration measurements on a porous ceramic. *Trans Faraday Soc* 1966; 62: 2395–2402.
174. Abadi SRH, Sebzari MR, Hemati M, et al. Ceramic membrane performance in microfiltration of oily wastewater. *Desalination* 2011; 265: 222–228.
175. Blumenschein S, Böcking A, Kätzel U, et al. Rejection modeling of ceramic membranes in organic solvent nanofiltration. *J Memb Sci*; 510.
176. Hosseinabadi SR, Wyns K, Buekenhoudt A, et al. Performance of Grignard functionalized ceramic nanofiltration membranes. *Sep Purif Technol* 2015; 147: 320–328.

177. Kingsbury BFK, Li K. A morphological study of ceramic hollow fibre membranes. *J Memb Sci* 2009; 328: 134–140.
178. Monash P, Pugazhenth G. Effect of TiO₂ addition on the fabrication of ceramic membrane supports: A study on the separation of oil droplets and bovine serum albumin (BSA) from its solution. *Desalination* 2011; 279: 104–114.
179. Bhave RR, Gillot J. The Developing Use of Inorganic Membranes: A Historical Perspective. In: Bhave RR (ed) *Inorganic Membranes Synthesis, Characteristics and Applications*. Dordrecht: Springer Netherlands, 1991, pp. 1–9.
180. Strathmann H. Inorganic membranes — synthesis, characteristics and applications. *Inorganica Chim Acta* 1992; 202: 119.
181. Yiqu F. Progress in research on surface properties of ceramic membranes, <https://www.semanticscholar.org/paper/Progress-in-research-on-surface-properties-of-Yiqu/6b032344d00208f5370a66c637c7d2fc6fbb3e> (2013, accessed 26 September 2019).
182. Zhang Y, Liu B, Fang D. Stress-induced phase transition and deformation behavior of BaTiO₃ nanowires. *J Appl Phys* 2011; 110: 054109.
183. De Lange RSA, Hekkink JHA, Keizer K, et al. Microstructural properties of non-supported microporous ceramic membrane top-layers obtained by the sol-gel process. *J Non Cryst Solids* 1996; 195: 203–217.
184. van der Horst HC, Hanemaaijer JH. Cross-flow microfiltration in the food industry. State of the art. *Desalination* 1990; 77: 235–258.
185. Bolduan P, Latz M. Ceramic membranes and their application in food and beverage processing. *Filtration and Separation* 2000; 37: 36–38.
186. Shackleton R. The application of ceramic membranes to the biological industries. *J Chem Technol Biotechnol* 2007; 37: 67–69.
187. De Vos RM, Maier WF, Verweij H. Hydrophobic silica membranes for gas separation. *J Memb Sci* 1999; 158: 277–288.
188. Aaron D, Tsouris C. Separation of CO₂ from flue gas: A review. *Sep Sci Technol* 2005; 40: 321–348.
189. Hyun SH, Kang BS. Synthesis of titania composite membranes by the pressurized sol-gel technique. *J Am Ceram Soc* 1996; 79: 279–282.

190. Gitis V. *Ceramic membranes: new opportunities and practical applications*. John Wiley & Sons, 2016.
191. Deng W, Long M, Zhou Q, et al. One-step preparation of superhydrophobic acrylonitrile-butadiene-styrene copolymer coating for ultrafast separation of water-in-oil emulsions. *J Colloid Interface Sci* 2018; 511: 21–26.
192. Horovitz I, Horovitz I, Gitis V, et al. Ceramic-based photocatalytic membrane reactors for water treatment - Where to next? *Rev Chem Eng* 2020; 36: 593–622.
193. Richerson DW. *Modern Ceramic Engineering*. CRC press, 2005.
194. Vijayan S, Wilson P, Prabhakaran K, et al. Preparation of ceramic foam spheres by injection molding of emulsions. *J Asian Ceram Soc* 2020; 8: 21–28.
195. Amin SK. An Overview of Production and Development of Ceramic Membranes. 2016; 11: 7708–7721.
196. Zyryanov V V, Karakchiev LG. Porous supports for conducting ceramic membranes. *Inorg Mater* 2011; 44: 429.
197. Souza LPDF, Mansur HS. Production and characterization of ceramic pieces obtained by slip casting using powder wastes. *J Mater Process Technol* 2004; 145: 14–20.
198. Onoda GY, Hench LL, of Florida U. *Ceramic processing before firing*. Wiley, https://books.google.com.my/books?id=8_AeAQAIAAJ (1978).
199. Meier LP, Urech L, Gauckler LJ. Tape casting of nanocrystalline ceria gadolinia powder. *J Eur Ceram Soc* 2004; 24: 3753–3758.
200. Yuping Z, Dongliang J, Greil P. Tape casting of aqueous Al₂O₃ slurries. *J Eur Ceram Soc* 2000; 20: 1691–1697.
201. Lindqvist K, Lidén E. Preparation of Alumina Membranes by Tape Casting and Dip Coating. *J Eur Ceram Soc* 1997; 17: 359–366.
202. Gongora-Rubio MR, Espinoza-Vallejos P, Sola-Laguna L, et al. Overview of low temperature co-fired ceramics tape technology for meso-system technology (MsST). *Sensors Actuators, A Phys* 2001; 89: 222–241.
203. Cooper AR, Eaton LE. Compaction Behavior of Several Ceramic Powders. *J Am Ceram Soc* 1962; 45: 97–101.
204. Wang YH, Cheng JG, Liu XQ, et al. Preparation and sintering of macroporous ceramic membrane support from titania sol-coated alumina powder. *J Am Ceram Soc* 2008; 91: 825–830.

205. Huang S-C, Huang C-T, Lu S-Y, et al. Ceramic/Polyaniline Composite Porous Membranes. *J Porous Mater* 1999; 6: 153–159.
206. Li G, Qi H, Fan Y, et al. Toughening macroporous alumina membrane supports with YSZ powders. *Ceram Int* 2009; 35: 1641–1646.
207. Dong Y, Zhou J er, Lin B, et al. Reaction-sintered porous mineral-based mullite ceramic membrane supports made from recycled materials. *J Hazard Mater* 2009; 172: 180–186.
208. Del Colle R, Fortulan CA, Fontes SR. Manufacture and characterization of ultra and microfiltration ceramic membranes by isostatic pressing. *Ceram Int* 2011; 37: 1161–1168.
209. Chen G, Qi H, Xing W, et al. Direct preparation of macroporous mullite supports for membranes by in situ reaction sintering. *J Memb Sci* 2008; 318: 38–44.
210. Luyten J, Coymans J, Smolders C, et al. Shaping of Multilayer Ceramic Membranes by Dip Coating. *J Eur Ceram Soc* 1997; 17: 273–279.
211. Dong Y, Lin B, Xie K, et al. Cost-effective macro-porous mullite-corundum ceramic membrane supports derived from the industrial grade powder. *J Alloys Compd* 2009; 477: 350–356.
212. Vercauteren S, Keizer K, Vansant EF, et al. Porous Ceramic Membranes: Preparation, Transport Properties and Applications. *J Porous Mater* 1998; 5: 241–258.
213. Wang YH, Liu XQ, Meng GY. Preparation and properties of supported 100% titania ceramic membranes. *Mater Res Bull* 2008; 43: 1480–1491.
214. Issaoui M, Limousy L. Low-cost ceramic membranes: Synthesis, classifications, and applications. *Comptes Rendus Chim* 2019; 22: 175–187.
215. Hubadillah SK, Tai ZS, Othman MHD, et al. Hydrophobic ceramic membrane for membrane distillation: A mini review on preparation, characterization, and applications. *Sep Purif Technol* 2019; 217: 71–84.
216. Malgaj T, Kocjan A, Jevnikar P. The effect of firing protocols on the resin-bond strength to alumina-coated zirconia ceramics. *Adv Appl Ceram* 2019; 1–9.
217. Handle F. *Extrusion in ceramic*. Springer, 2007.
218. Bengisu M. *Engineering Ceramics*. Springer Science & Business Media, 2013.

219. Substech,
https://www.substech.com/dokuwiki/doku.php?id=methods_of_shape_forming_ceramic_powders (accessed 17 May 2020).
220. Grida I, Evans JRG. Extrusion freeforming of ceramics through fine nozzles. *J Eur Ceram Soc* 2003; 23: 629–635.
221. Kastyl J, Chlup Z, Stastny P, et al. Machinability and properties of zirconia ceramics prepared by gelcasting method. *Adv Appl Ceram* 2019; 1–9.
222. Oun A, Tahri N, Mahouche-Chergui S, et al. Tubular ultrafiltration ceramic membrane based on titania nanoparticles immobilized on macroporous clay-alumina support: Elaboration, characterization and application to dye removal. *Sep Purif Technol* 2017; 188: 126–133.
223. Zhou J, Zhang X, Wang Y, et al. Elaboration and characterization of tubular macroporous ceramic support for membranes from kaolin and dolomite. *J Porous Mater* 2010; 17: 1–9.
224. Tang S, Fan Z, Zhao H, et al. Layered extrusion forming—a simple and green method for additive manufacturing ceramic core. *Int J Adv Manuf Technol* 2018; 96: 3809–3819.
225. Shokrkar H, Salahi A, Kasiri N, et al. Mullite ceramic membranes for industrial oily wastewater treatment: Experimental and neural network modeling. *Water Sci Technol* 2011; 64: 670–676.
226. Yang L, Tang S, Li G, et al. Performance characteristics of collapsible CaO-SiO₂ based ceramic core material via layered extrusion forming. *Ceram Int* 2019; 45: 7681–7689.
227. Abbasi M, Salahi A, Mirfendereski M, et al. Oily wastewater treatment using mullite ceramic membrane. *Desalin Water Treat* 2012; 37: 21–30.
228. Trunec M. Fabrication of zirconia- and ceria-based thin-wall tubes by thermoplastic extrusion. *J Eur Ceram Soc* 2004; 24: 645–651.
229. Chaves AC, Neves GA, Lira HL, et al. Use of the processed waste from kaolin and granite sawing in the manufacture of tubular ceramic membranes. *Mater Sci Forum* 2015; 805: 337–342.
230. Vinoth Kumar R, Kumar Ghoshal A, Pugazhenti G. Elaboration of novel tubular ceramic membrane from inexpensive raw materials by extrusion method and its performance in microfiltration of synthetic oily wastewater treatment. *J Memb Sci* 2015; 490: 92–102.

231. da Silva VJ, da Silva MF, Gonçalves WP, et al. Porous mullite blocks with compositions containing kaolin and alumina waste. *Ceram Int* 2016; 42: 15471–15478.
232. Loeb S, Sourirajan S. Sea Water Demineralization by Means of an Osmotic Membrane. 1963, pp. 117–132.
233. Nidal Hilal, Ahmad Fauzi Ismail CW. Membrane Fabrication - Google Books. *CRC Press*.
234. Luyten J, Buekenhoudt A, Adriansens W, et al. Preparation of LaSrCoFeO_{3-x} membranes. *Solid State Ionics* 2000; 135: 637–642.
235. Kingsbury BFK, Wu Z, Li K. A morphological study of ceramic hollow fibre membranes: A perspective on multifunctional catalytic membrane reactors. *Catal Today* 2010; 156: 306–315.
236. Paiman SH, Rahman MA, Othman MHD, et al. Morphological study of yttria-stabilized zirconia hollow fibre membrane prepared using phase inversion/sintering technique. *Ceram Int* 2015; 41: 12543–12553.
237. Li T, Wu Z, Li K. A dual-structured anode/Ni-mesh current collector hollow fibre for micro-tubular solid oxide fuel cells (SOFCs). *J Power Sources* 2014; 251: 145–151.
238. Kanawka K, Othman MHD, Wu Z, et al. A dual layer Ni/Ni-YSZ hollow fibre for micro-tubular SOFC anode support with a current collector. *Electrochem commun* 2011; 13: 93–95.
239. Othman MHD, Droushiotis N, Wu Z, et al. Dual-layer hollow fibres with different anode structures for micro-tubular solid oxide fuel cells. *J Power Sources* 2012; 205: 272–280.
240. He T, Mulder MHV, Wessling M. Preparation of porous hollow fiber membranes with a triple-orifice spinneret. *J Appl Polym Sci* 2003; 87: 2151–2157.
241. Li T, Wu Z, Li K. Single-step fabrication and characterisations of triple-layer ceramic hollow fibres for micro-tubular solid oxide fuel cells (SOFCs). *J Memb Sci* 2014; 449: 1–8.
242. Lee M, Wang B, Li K. New designs of ceramic hollow fibres toward broadened applications. *J Memb Sci* 2016; 503: 48–58.

243. Makhtar SNNM, Rahman MA, Ismail AF, et al. Preparation and characterization of glass hollow fiber membrane for water purification applications. *Environ Sci Pollut Res* 2017; 24: 15918–15928.
244. Liu S, Li K, Hughes R. Preparation of porous aluminium oxide (Al₂O₃) hollow fibre membranes by a combined phase-inversion and sintering method. *Ceram Int* 2003; 29: 875–881.
245. Li L, Chen M, Dong Y, et al. A low-cost alumina-mullite composite hollow fiber ceramic membrane fabricated via phase-inversion and sintering method. *J Eur Ceram Soc* 2015; 36: 2057–2066.
246. Hubadillah SK, Othman MHD, Harun Z, et al. Superhydrophilic, low cost kaolin-based hollow fibre membranes for efficient oily-wastewater separation. *Mater Lett* 2017; 191: 119–122.
247. Guillen GR, Pan Y, Li M, et al. Preparation and characterization of membranes formed by nonsolvent induced phase separation: A review. *Ind Eng Chem Res* 2011; 50: 3798–3817.
248. Sawaiyan A. *Development of Low Clay Whiteware Bodies (PhD Thesis)*. National Institute of Technology Rourkela, Odisha, 2015.
249. Lagaly G, Ogawa M, Dékány I. Clay mineral-organic interactions. In: *Developments in Clay Science*. 2013, pp. 435–505.
250. Chakraborty AK, Chakraborty AK. Final Conclusion on the Thermal Effects of Kaolinite. In: *Phase Transformation of Kaolinite Clay*. 2014.
251. Babu Valapa R, Loganathan S, Pugazhenth G, et al. Chapter 2 - An Overview of Polymer–Clay Nanocomposites. In: Jlassi K, Chehimi MM, Thomas SBT-C-PN (eds). Elsevier, pp. 29–81.
252. Rollmann LD, Valyocsik EW, Shannon RD. Zeolite Molecular Sieves. In: *Inorganic Syntheses*. 2006, pp. 61–68.
253. Granizo ML, Blanco-Varela MT, Palomo A. Influence of the starting kaolin on alkali-activated materials based on metakaolin. Study of the reaction parameters by isothermal conduction calorimetry. *J Mater Sci* 2000; 35: 6309–6315.
254. Brindey GW, Nakihira M. Kinetics of Dehydroxylation of Kaolinite and Hallosite. *J Am Ceram Soc* 1957; 40: 346–350.
255. Oberli L, Caruso D, Hall C, et al. Condensation and freezing of droplets on superhydrophobic surfaces. *Adv Colloid Interface Sci* 2014; 210: 47–57.

256. Thomas YI. An essay on the cohesion of fluids. *Philos Trans R Soc London* 1805; 95: 65–87.
257. Crick CR, Parkin IP. Preparation and characterisation of super-hydrophobic surfaces. *Chem - A Eur J* 2010; 16: 3568–3588.
258. Song J, Rojas OJ. Approaching super-hydrophobicity from cellulosic materials: A review. *Nord Pulp Pap Res J* 2013; 28: 216–238.
259. Wenzel RN. Resistance of solid surfaces to wetting by water. *Ind Eng Chem* 1936; 28: 988–994.
260. Cassie ABD, Baxter S. Wettability of porous surfaces. *Trans Faraday Soc* 1944; 40: 546–551.
261. Cassie ABD. Contact angles. *Discuss Faraday Soc* 1948; 3: 11–16.
262. Kim SH. Fabrication of superhydrophobic surfaces. *J Adhes Sci Technol* 2008; 22: 235–250.
263. Prusi A, Arsov L, Haran B, et al. Anodic Behavior of Ti in KOH Solutions. *J Electrochem Soc* 2002; 149: B491.
264. Search H, Journals C, Contact A, et al. Roughness-induced non-wetting. *Europhys Lett* 2000; 52: 165–170.
265. Wu J, Xia J, Lei W, et al. Fabrication of superhydrophobic surfaces with double-scale roughness. *Mater Lett* 2010; 64: 1251–1253.
266. Feng L, Li S, Li Y, et al. Super-hydrophobic surfaces: From natural to artificial. *Adv Mater* 2002; 14: 1857–1860.
267. Cao L, Hu HA, Gao D. Design and fabrication of micro-textures for inducing a superhydrophobic behavior on hydrophilic materials. *Langmuir* 2007; 23: 4310–4314.
268. Li Z, Rana D, Wang Z, et al. Synergic effects of hydrophilic and hydrophobic nanoparticles on performance of nanocomposite distillation membranes: An experimental and numerical study. *Sep Purif Technol* 2018; 202: 45–58.
269. Rácz G, Kerker S, Kovács Z, et al. Theoretical and experimental approaches of liquid entry pressure determination in membrane distillation processes. *Period Polytech Chem Eng* 2014; 58: 81–91.
270. Alkhudhiri A, Darwish N, Hilal N. Membrane distillation: A comprehensive review. *Desalination* 2012; 287: 2–18.

271. Ahmad NA, Leo CP, Ahmad AL, et al. Membranes with great hydrophobicity: A review on preparation and characterization. *Sep Purif Rev* 2015; 44: 109–134.
272. García-Payo MC, Izquierdo-Gil MA, Fernández-Pineda C. Wetting study of hydrophobic membranes via liquid entry pressure measurements with aqueous alcohol solutions. *J Colloid Interface Sci* 2000; 230: 420–431.
273. Krajewski SR, Kujawski W, Bukowska M, et al. Application of fluoroalkylsilanes (FAS) grafted ceramic membranes in membrane distillation process of NaCl solutions. *J Memb Sci* 2006; 281: 253–259.
274. Razmjou A, Arifin E, Dong G, et al. Superhydrophobic modification of TiO₂ nanocomposite PVDF membranes for applications in membrane distillation. *J Memb Sci* 2012; 415–416: 850–863.
275. Padaki M, Surya Murali R, Abdullah MS, et al. Membrane technology enhancement in oil-water separation. A review. *Desalination* 2015; 357: 197–207.
276. Witucki GL. Silane primer: chemistry and applications of alkoxy silanes. *J Coatings Technol* 1993; 65: 57–60.
277. Mittal KL. *Silanes and other coupling agents*. 2004.
278. Agents SC. *Silane Coupling Agents: Combination of organic and Inorganic Materials Catalog*. 2017.
279. Corning D. A guide to silane solutions from Dow Corning. USA Dow Corning Retrieved July, <http://scholar.google.com/scholar?hl=en&btnG=Search&q=intitle:Guide+to+Silane+Solutions+from+Dow+Corning#0> (2005).
280. Corning D. A guide to silane solutions from Dow Corning. USA: Dow Corning., <http://scholar.google.com/scholar?hl=en&btnG=Search&q=intitle:Guide+to+Silane+Solutions+from+Dow+Corning#0> (accessed 13 January 2020).
281. Dumée LF, Allieux FM, Reis R, et al. Qualitative spectroscopic characterization of the matrix-silane coupling agent interface across metal fibre reinforced ion exchange resin composite membranes. *Vib Spectrosc* 2014; 75: 203–212.
282. Kong L, Chen X, Yang G, et al. Preparation and characterization of slice-like Cu₂(OH)₃NO₃ superhydrophobic structure on copper foil. *Appl Surf Sci* 2008; 254: 7255–7258.

283. Kujawski W, Kujawa J, Wierzbowska E, et al. Influence of hydrophobization conditions and ceramic membranes pore size on their properties in vacuum membrane distillation of water-organic solvent mixtures. *J Memb Sci* 2016; 499: 442–451.
284. Liu H, Szunerits S, Xu W, et al. Preparation of superhydrophobic coatings on zinc as effective corrosion barriers. *ACS Appl Mater Interfaces* 2009; 1: 1150–1153.
285. Khemakhem M, Khemakhem S, Ben Amar R. Emulsion separation using hydrophobic grafted ceramic membranes by. *Colloids Surfaces A Physicochem Eng Asp* 2013; 436: 402–407.
286. Tan J, Hao J, An Z, et al. Superhydrophobic surfaces on brass substrates fabricated: Via micro-etching and a growth process. *RSC Adv* 2017; 7: 26145–26152.
287. Choy KL. Chemical vapour deposition of coatings. *Prog Mater Sci* 2003; 48: 57–170.
288. Mahmood T, Afzal U. Security analytics: Big data analytics for cybersecurity: A review of trends, techniques and tools. In: *Conference Proceedings - 2013 2nd National Conference on Information Assurance, NCIA 2013*. 2013, pp. 129–134.
289. Subhash Lathe S, Basavraj Gurav A, Shridhar Maruti C, et al. Recent Progress in Preparation of Superhydrophobic Surfaces: A Review. *J Surf Eng Mater Adv Technol* 2012; 02: 76–94.
290. Vlassioux I, Fulvio P, Meyer H, et al. Large scale atmospheric pressure chemical vapor deposition of graphene. *Carbon N Y* 2013; 54: 58–67.
291. Yan TX and. *Chemical Vapour Deposition*. London: Springer, 2010.
292. Zhang Q, Sando D, Nagarajan V. Chemical route derived bismuth ferrite thin films and nanomaterials. *J Mater Chem C* 2016; 4: 4092–4124.
293. Hozumi A, Cheng DF, Yagihashi M. Hydrophobic/superhydrophobic oxidized metal surfaces showing negligible contact angle hysteresis. *J Colloid Interface Sci* 2011; 353: 582–587.
294. Li YY, Nomura T, Sakoda A, et al. Fabrication of carbon coated ceramic membranes by pyrolysis of methane using a modified chemical vapor deposition apparatus. *J Memb Sci* 2002; 197: 23–35.

295. Rezaei S, Manoucheri I, Moradian R, et al. One-step chemical vapor deposition and modification of silica nanoparticles at the lowest possible temperature and superhydrophobic surface fabrication. *Chem Eng J* 2014; 252: 11–16.
296. Sugimura H. Self-Assembled Monolayer on Silicon. *Nanocrystalline Mater* 2006; 57–91.
297. Yim JH, Rodriguez-Santiago V, Williams AA, et al. Atmospheric pressure plasma enhanced chemical vapor deposition of hydrophobic coatings using fluorine-based liquid precursors. *Surf Coatings Technol* 2013; 234: 21–32.
298. Hsieh C Te, Chen WY, Wu FL. Fabrication and superhydrophobicity of fluorinated carbon fabrics with micro/nanoscaled two-tier roughness. *Carbon N Y* 2008; 46: 1218–1224.
299. Zheng Z, Gu Z, Huo R, et al. Superhydrophobicity of polyvinylidene fluoride membrane fabricated by chemical vapor deposition from solution. *Appl Surf Sci* 2009; 255: 7263–7267.
300. Himma NF, Prasetya N, Anisah S, et al. Superhydrophobic membrane: Progress in preparation and its separation properties. *Rev Chem Eng* 2019; 35: 211–238.
301. Ahmad NA, Leo CP, Ahmad AL, et al. Membranes with great hydrophobicity: A review on preparation and characterization. *Sep Purif Rev* 2015; 44: 109–134.
302. Xue C-H, Jia S-T, Zhang J, et al. Large-area fabrication of superhydrophobic surfaces for practical applications: an overview. *Sci Technol Adv Mater* 2010; 11: 033002.
303. Hubadillah SK, Kumar P, Dzarfan Othman MH, et al. A low cost, superhydrophobic and superoleophilic hybrid kaolin-based hollow fibre membrane (KHFM) for efficient adsorption-separation of oil removal from water. *RSC Adv* 2018; 8: 2986–2995.
304. Yang X, Ding L, Wolf M, et al. Pervaporation of ammonia solution with γ -alumina supported organosilica membranes. *Sep Purif Technol* 2016; 168: 141–151.
305. Su C, Xu Y, Zhang W, et al. Porous ceramic membrane with superhydrophobic and superoleophilic surface for reclaiming oil from oily water. *Appl Surf Sci* 2012; 258: 2319–2323.
306. Purcar V, Donescu D, Spataru IC, et al. Surface modification of sol-gel hybrid films using fluorinated silica nanoparticles. *Rev Roum Chim* 2013; 58: 37–42.

307. Ahmad NA, Leo CP, Ahmad AL. Synthesis of superhydrophobic alumina membrane: Effects of sol-gel coating, steam impingement and water treatment. *Appl Surf Sci* 2013; 284: 556–564.
308. Gurav AB, Xu Q, Latthe SS, et al. Superhydrophobic coatings prepared from methyl-modified silica particles using simple dip-coating method. *Ceram Int* 2015; 41: 3017–3023.
309. Van De Mark MR, Sandefur K. Vegetable oils in paint and coatings. *Inf - Int News Fats, Oils Relat Mater* 2005; 16: 478.
310. Alam M, Akram D, Sharmin E, et al. Vegetable oil based eco-friendly coating materials: A review article. *Arab J Chem* 2014; 7: 469–479.
311. Sharmin E, Zafar F, Akram D, et al. Recent advances in vegetable oils based environment friendly coatings: A review. *Ind Crops Prod* 2015; 76: 215–229.
312. Romero J, Draga H, Belleville MP, et al. New hydrophobic membranes for contactor processes - Applications to isothermal concentration of solutions. *Desalination* 2006; 193: 280–285.
313. Hablot E, Zheng D, Bouquey M, et al. Polyurethanes based on castor oil: Kinetics, chemical, mechanical and thermal properties. *Macromol Mater Eng* 2008; 293: 922–929.
314. Meshram PD, Puri RG, Patil AL, et al. Synthesis and characterization of modified cottonseed oil based polyesteramide for coating applications. *Prog Org Coatings* 2013; 76: 1144–1150.
315. Yang H, Pi P, Cai ZQ, et al. Facile preparation of super-hydrophobic and super-oleophilic silica film on stainless steel mesh via sol-gel process. *Appl Surf Sci* 2010; 256: 4095–4102.
316. Khatib SJ, Oyama ST. Silica membranes for hydrogen separation prepared by chemical vapor deposition (CVD). *Sep Purif Technol* 2013; 111: 20–42.
317. Ahmad AL, Idrus NF, Othman MR. Preparation of perovskite alumina ceramic membrane using sol-gel method. *J Memb Sci* 2005; 262: 129–137.
318. Karimifard S, Alavi Moghaddam MR. Application of response surface methodology in physicochemical removal of dyes from wastewater: A critical review. *Sci Total Environ* 2018; 640–641: 772–797.
319. Ferreira SLC, Bruns RE, Ferreira HS, et al. Box-Behnken design: An alternative for the optimization of analytical methods. *Anal Chim Acta* 2007; 597: 179–186.

320. Khedmati M, Khodaii A, Haghshenas HF. A study on moisture susceptibility of stone matrix warm mix asphalt. *Constr Build Mater* 2017; 144: 42–49.
321. Sakkas VA, Islam MA, Stalikas C, et al. Photocatalytic degradation using design of experiments: A review and example of the Congo red degradation. *J Hazard Mater* 2010; 175: 33–44.
322. Suresh K, Pugazhenti G, Uppaluri R. Fly ash based ceramic microfiltration membranes for oil-water emulsion treatment: Parametric optimization using response surface methodology. *J Water Process Eng* 2016; 13: 27–43.
323. Li Y, Zhu X, Zhou X, et al. A facile way to fabricate a superamphiphobic surface. *Appl Phys A Mater Sci Process* 2014; 115: 765–770.
324. Baig N, Alghunaimi FI, Dossary HS, et al. Superhydrophobic and superoleophilic carbon nanofiber grafted polyurethane for oil-water separation. *Process Saf Environ Prot* 2019; 123: 327–334.
325. Salah A, Noshadi I, Badrnezhad R, et al. Nano-porous membrane process for oily wastewater treatment: Optimization using response surface methodology. *J Environ Chem Eng* 2013; 1: 218–225.
326. Han LF, Xu ZL, Cao Y, et al. Preparation, characterization and permeation property of Al₂O₃, Al₂O₃-SiO₂ and Al₂O₃-kaolin hollow fiber membranes. *J Memb Sci* 2011; 372: 154–164.
327. Hubadillah SK, Othman MHD, Ismail AF, et al. The feasibility of kaolin as main material for low cost porous ceramic hollow fibre membrane prepared using combined phase inversion and sintering technique. *J Teknol* 2017; 79: 35–39.
328. Hubadillah SK, Harun Z, Othman MHD, et al. Effect of kaolin particle size and loading on the characteristics of kaolin ceramic support prepared via phase inversion technique. *J Asian Ceram Soc* 2016; 4: 164–177.
329. Hubadillah SK, Harun Z, Othman MHD, et al. Effect of kaolin particle size and loading on the characteristics of kaolin ceramic support prepared via phase inversion technique. *J Asian Ceram Soc* 2016; 4: 164–177.
330. Nandi BK, Uppaluri R, Purkait MK. Preparation and characterization of low cost ceramic membranes for micro-filtration applications. *Appl Clay Sci* 2008; 42: 102–110.

331. Tang W, Yuan Y, Lin D, et al. Kaolin-reinforced 3D MBG scaffolds with hierarchical architecture and robust mechanical strength for bone tissue engineering. *J Mater Chem B* 2014; 2: 3782–3790.
332. Silva MC Da, Lira HDL, Lima RDCDO, et al. Effect of sintering temperature on membranes manufactured with clays for textile effluent treatment. *Adv Mater Sci Eng* 2015; 2015: 1–7.
333. Chen CY, Lan GS, Tuan WH. Microstructural evolution of mullite during the sintering of kaolin powder compacts. *Ceram Int* 2000; 26: 715–720.
334. Woo SH, Lee JS, Lee HH, et al. Preparation Method of Crack-free PVDF Microfiltration Membrane with Enhanced Antifouling Characteristics. *ACS Appl Mater Interfaces* 2015; 7: 16466–16477.
335. Murray HH. Overview - clay mineral applications. *Appl Clay Sci* 1991; 5: 379–395.
336. Fane AG, Fell CJD, Waters AG. The relationship between membrane surface pore characteristics and flux for ultrafiltration membranes. *J Memb Sci* 1981; 9: 245–262.
337. Hu K, Dickson JM. Membrane Processes for Dairy Ingredient Separation. In: Hu K, Dickson JM (eds) *Membrane Processes for Dairy Ingredient Separation*. John Wiley & Sons, Ltd, pp. 1–275.
338. Xue CH, Ji PT, Zhang P, et al. Fabrication of superhydrophobic and superoleophilic textiles for oil-water separation. *Appl Surf Sci* 2013; 284: 464–471.
339. Lin YM, Rutledge GC. Separation of oil-in-water emulsions stabilized by different types of surfactants using electrospun fiber membranes. *J Memb Sci* 2018; 563: 247–258.
340. Sun D, Kang S, Liu C, et al. Effect of zeta potential and particle size on the stability of SiO₂ nanospheres as carrier for ultrasound imaging contrast agents. *Int J Electrochem Sci* 2016; 11: 8520–8529.
341. Cheng Y, Huang T, Shi X, et al. Removal of ammonium ion from water by Na-rich birnessite: Performance and mechanisms. *J Environ Sci (China)* 2017; 57: 402–410.
342. Zhao X, Hao H, Duan Y, et al. A robust superhydrophobic and highly oleophobic coating based on F-SiO₂-copolymer composites. *Prog Org Coatings* 2019; 135: 417–423.

343. Muzakir S, Salim N, Abu Bakar NH, et al. Fabrication of hydrophobic compressed oil palm trunk surface by sol-gel process. In: *IOP Conference Series: Materials Science and Engineering*. 2018.
344. Hsieh C Te, Chen WY, Wu FL, et al. Superhydrophobicity of a three-tier roughened texture of microscale carbon fabrics decorated with silica spheres and carbon nanotubes. *Diam Relat Mater* 2010; 19: 26–30.
345. Mostofi Sarkari N, Darvish F, Mohseni M, et al. Surface characterization of an organosilane-grafted moisture-crosslinked polyethylene compound treated by air atmospheric pressure non-equilibrium gliding arc plasma. *Appl Surf Sci* 2019; 490: 436–450.
346. Zhang Q, Yan Z, Ouyang J, et al. Chemically modified kaolinite nanolayers for the removal of organic pollutants. *Appl Clay Sci* 2018; 157: 283–290.
347. Wang S, Liu C, Liu G, et al. Fabrication of superhydrophobic wood surface by a sol-gel process. *Appl Surf Sci* 2011; 258: 806–810.
348. Cui M, Shen Y, Tian H, et al. Influence of water adhesion of superhydrophobic surfaces on their anti-corrosive behavior. *Surf Coatings Technol* 2018; 347: 38–45.
349. Monash P, Pugazhenth G, Saravanan P. Various fabrication methods of porous ceramic supports for membrane applications. *Rev Chem Eng* 2013; 29: 357–383.
350. Wahid H, Ahmad S, Nor MAM, et al. *Gelest Silane functionalization*. Gelest Inc, 2017.
351. Zhou YN, Li JJ, Luo ZH. PhotoATRP-Based Fluorinated Thermosensitive Block Copolymer for Controllable Water/Oil Separation. *Ind Eng Chem Res* 2015; 54: 10714–10722.
352. Li J, Cui M, Tian H, et al. Facile fabrication of anti-corrosive superhydrophobic diatomite coatings for removal oil from harsh environments. *Sep Purif Technol* 2017; 189: 335–340.
353. Liu M, Li J, Guo Z. Electrochemical route to prepare polyaniline-coated meshes with controllable pore size for switchable emulsion separation. *Chem Eng J* 2016; 304: 115–120.
354. Gao N. Evaluation of the oleophilicity of different alkoxy silane modified ceramic membranes through wetting dynamic measurements. *Appl Sci* 2013; 66: 863–870.

355. Jamalludin MR, Hubadillah SK, Harun Z, et al. Novel superhydrophobic and superoleophilic sugarcane green ceramic hollow fibre membrane as hybrid oil sorbent-separator of real oil and water mixture. *Mater Lett* 2019; 240: 136–139.
356. Parsaie A, Mohammadi-Khanaposhtani M, Riazi M, et al. Magnesium stearate-coated superhydrophobic sponge for oil/water separation: Synthesis, properties, application. *Sep Purif Technol* 2020; 251: 117105.
357. Gao Y, Zhou YS, Xiong W, et al. Highly efficient and recyclable carbon soot sponge for oil cleanup. *ACS Appl Mater Interfaces* 2014; 6: 5924–5929.
358. Sun H, Zhu Z, Liang W, et al. Reduced graphene oxide-coated cottons for selective absorption of organic solvents and oils from water. *RSC Adv* 2014; 4: 30587–30591.
359. Nine MJ, Kabiri S, Sumona AK, et al. Superhydrophobic/superoleophilic natural fibres for continuous oil-water separation and interfacial dye-adsorption. *Sep Purif Technol* 2020; 233: 116062.
360. Yu Y, Chen H, Liu Y, et al. Superhydrophobic and Superoleophilic Porous Boron Nitride Nanosheet/Polyvinylidene Fluoride Composite Material for Oil-Polluted Water Cleanup. *Adv Mater Interfaces* 2015; 2: 1–10.
361. Hua FL, Tsang YF, Wang YJ, et al. Performance study of ceramic microfiltration membrane for oily wastewater treatment. *Chem Eng J* 2007; 128: 169–175.
362. Chakrabarty B, Ghoshal AK, Purkait MK. Ultrafiltration of stable oil-in-water emulsion by polysulfone membrane. *J Memb Sci* 2008; 325: 427–437.
363. Chandio ZA, Ramasamy M, Mukhtar HB. Temperature effects on solubility of asphaltenes in crude oils. *Chem Eng Res Des* 2015; 94: 573–583.
364. Banerjee A, Ray SK. {PVA} modified filled copolymer membranes for pervaporative dehydration of acetic acid-systematic optimization of synthesis and process parameters with response surface methodology. *J Memb Sci* 2018; 549: 84–100.
365. Zolfaghari R, Fakhru'l-Razi A, Abdullah LC, et al. Demulsification techniques of water-in-oil and oil-in-water emulsions in petroleum industry. *Sep Purif Technol* 2016; 170: 377–407.
366. Yuan X, Liu J, Zeng G, et al. Optimization of conversion of waste rapeseed oil with high FFA to biodiesel using response surface methodology. *Renew Energy* 2008; 33: 1678–1684.

367. Sakinah AMM, Ismail AF, Hassan O, et al. Influence of starch pretreatment on yield of cyclodextrins and performance of ultrafiltration membranes. *Desalination* 2009; 239: 317–333.
368. Suárez-Escobar A, Pataquiva-Mateus A, López-Vasquez A. Electrocoagulation - Photocatalytic process for the treatment of lithographic wastewater. Optimization using response surface methodology (RSM) and kinetic study. *Catal Today* 2016; 266: 120–125.

LIST OF PUBLICATIONS

Journal with Impact Factor

1. **Jamilu Usman**; Mohd Hafiz Dzarfan Othman*; Ahmad Fauzi Ismail; Muklish A Rahman; Juhana Jaafar; Yusuf Olabode Raji; Tijjani Hassan El-badawy; Gbadamosi Afeez, Impact of Organosilanes Modified Superhydrophobic-Superoleophilic Kaolin Ceramic Membrane for Efficient Oil Recovery from Produced Water *Journal of Chemical Technology & Biotechnology*, 2020, Wiley (**Q1, IF: 2.75**)

Indexed Journal

1. **Jamilu Usman**; Mohd Hafiz Dzarfan Othman*; Ahmad Fauzi Ismail; Muklish A Rahman; Juhana Jaafar; Tijjani Abudllahi, Comparative study of Malaysian and Nigerian Kaoli-based ceramic hollow fibre membranes for filtration application, *Malaysian Journal of Fundamental and Applied Sciences*, 16:2(2020) 182-185 (**Indexed by ISI**)
2. **Jamilu Usman**; Mohd Hafiz Dzarfan Othman*; Ahmad Fauzi Ismail; Yusuf Olabode Raji; Tijjani Hassan El-badawy; Tijjani Abdullahi; Jamila Baba Ali, Facile Approach in Development of Superhydrophobic-Superoleophilic Kaolin-Based Hollow Fibre Ceramic Membrane, *International Journal of Advance Science and Technology*, 29:108(2020), 3451-3459 (**Indexed by Scopus**)

Non-Indexed Conference Proceedings

1. **Jamilu Usman**; Mohd Hafiz Dzarfan Othman*; Ahmad Fauzi Ismail; Muklish A Rahman; Juhana Jaafar; Tijjani Abudllahi, Influence of kaolin on fabrication of ceramic hollow fibre membrane for oily wastewater treatment, National Congress on Membrane Technology, 2017, Pulau Springs Resort, Skudai, Johor, Malaysia

2. **Jamilu Usman**; Mohd Hafiz Dzarfan Othman*; Ahmad Fauz Ismail; Mukhlis A Rahman; Juhana Jaafar; Khairul Anwar Mohamad Said; Yusuf Olabode Raji, Superhydrophobic-Superoleophilic Modification of Kaolin-Based Ceramic Membrane Via Silanization Technique, International Conference on Sustainable Environment and Technology, 2019, Double Tree Hilton Hotel, Johor, Malaysia

3. **Jamilu Usman**; Mohd Hafiz Dzarfan Othman*; Ahmad Fauzi Ismail; Mukhlis A. Rahman; Juhana Jafar; Yusuf Olabode Raji; Tijjani Hassan El Badawy, Superhydrophobic-Superoleophilic Modification of Ceramic Membrane Via Nano-Silica Based Silane Sol, Malaysian-Japan International Conference on Nanoscience, Nanotechnology and nanoengineering 2020, Universiti Teknologi Mara (UiTM), Shah Alam, Malaysia