

POLYMER INCLUSION MEMBRANE SAMPLING PROBE FOR ELECTRIC  
FIELD DRIVEN EXTRACTION OF DOXORUBICIN

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## **DEDICATION**

Specially dedicated to my beloved family members for all support and encouragement in completing this study.

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## ABSTRACT

Sample preparation for the analysis of biological fluid samples posed significant challenges to the clinical laboratory. The current sample preparation practice involves multi-steps procedure that are time consuming and may lead to analytes lost. In this study, a low-cost, single-use sampling probe was introduced for electric field driven extraction. The sampling probe consists of a non-conductive glass capillary substrate, and a polymer inclusion membrane (PIM) dip-coated on the glass capillary substrate. The probe features a user-friendly design that can electrically extract targeted analytes from biological fluid sample. The glass capillary, closed at one end, was dipped in a homogeneous membrane solution that consists of an optimum composition of cellulose triacetate (CTA), 2-nitrophenyl octyl ether (2-NPOE) and 1-ethyl-3-methylimidazolium bis(trifluoromethylsulfonyl)imide ([EMIM][NTf<sub>2</sub>]), yielded a PIM with thickness of 30  $\mu\text{m}$  on the sampling probe. The effects of dipping cycles (thickness), voltage applied, and sample pH on the extraction efficiency of the sampling probe were thoroughly investigated. The concept of electrokinetic extraction using the PIM sampling probe was first demonstrated on the cationic model analyte, Rhodamine 6G at 500 V for 60 mins. The proven concept was then successfully applied to the extraction of the anticancer drug doxorubicin from samples in liquid or dried form, including dried blood spot (DBS), human plasma and human serum samples. The practicability and reliability of the electrokinetic extraction process were evaluated using liquid chromatography with tandem mass spectrometry (LC-MS/MS) to quantify the desorption of extracted doxorubicin from the PIM sampling probe. Under optimised conditions, the new method provided good linearity over a concentration range of 0.2 to 20 ng/mL; additionally, quantification limits of 0.2 to 2 ng/mL were achieved for the three biological samples and the relative recoveries ranged from 82.7 to 113.8% for DBS, human plasma and human serum samples; good method reproducibility was also achieved, with relative standard deviations (RSDs) ranging from 0.9 and 4.6%. The PIM sampling probe was further integrated into a portable battery-device for safe, lower-voltage (36 V) electrokinetic extraction. The correlation coefficients,  $r$ , for the two pairs of data were determined using the laboratory setup and the battery-powered device; values in the range of 0.9926 to 0.9996 were found, indicating an acceptable agreement. This new electrokinetic extraction approach represents a new opportunity for processing samples during sampling and transportation, saving time and reducing manual handling to produce more reliable test results efficiently.

## ABSTRAK

Penyediaan sampel untuk analisis sampel biologi menimbulkan cabaran besar kepada makmal klinikal. Amalan penyediaan sampel semasa melibatkan prosedur berbilang langkah yang memakan masa dan boleh menyebabkan kehilangan analit. Dalam kajian ini, *probe* persampelan sekali guna, kos rendah telah diperkenalkan untuk pengekstrakan dipacu medan elektrik. *Probe* persampelan terdiri daripada substrat kapilari kaca bukan konduktif, dan membran rangkum polimer (PIM) bersalut celup pada substrat kapilari kaca. *Probe* ini mempunyai reka bentuk mesra pengguna yang boleh mengekstrak analit disasarkan daripada sampel biologi di bawah aplikasi medan elektrik. Kapilari kaca yang ditutup pada satu hujung telah dicelup dalam larutan membran homogen mengandungi komposisi optimum yang terdiri daripada selulosa triasetat (CTA), 2-nitrofenil oktil eter (2-NPOE) dan 1-etil-3-metilimidazolium bis(trifluorometilsulfonil)imida ([EMIM][NTf<sub>2</sub>]) untuk menghasilkan PIM dengan ketebalan 30 µm pada *probe* persampelan. Kesan kitaran pencelupan (ketebalan), voltan yang dikenakan dan pH sampel ke atas kecekapan pengekstrakan *probe* persampelan telah dikaji dengan teliti. Konsep pengekstrakan elektrokinetik menggunakan *probe* persampelan PIM adalah kali pertama ditunjukkan pada analit model kationik, Rhodamine 6G pada 500 V selama 60 minit. Konsep kemudiannya terbukti berjaya digunakan untuk pengekstrakan drug antikanser doksorubisin daripada sampel dalam bentuk cecair atau kering termasuk tompok darah kering (DBS), sampel plasma manusia dan sampel serum manusia. Kebolehamalian dan kebolehpercayaan proses pengekstrakan elektrokinetik telah dinilai menggunakan kromatografi cecair dengan spektrometri jisim tandem (LC-MS/MS) untuk mengukur penyaherapan doksorubisin yang diekstrak daripada *probe* pensampelan PIM. Di bawah keadaan yang dioptimumkan, kaedah baharu memberikan kelinearan yang baik pada julat kepekatan 0.2 hingga 20 ng/mL; di samping itu, had kuantifikasi 0.2 hingga 2 ng/mL telah dicapai untuk tiga sampel biologi dan pemulihan relatif adalah antara 82.7 hingga 113.8% untuk sampel DBS, plasma manusia dan serum manusia; kebolehulangan kaedah yang baik juga dicapai, dengan sisihan piawai relatif (RSD) di antara 0.9 dan 4.6%. *Probe* persampelan PIM disepadukan lagi ke dalam peranti bateri mudah alih untuk pengekstrakan elektrokinetik voltan rendah (36 V) yang selamat. Pekali korelasi, *r*, untuk dua pasangan data telah ditentukan dengan menggunakan persediaan makmal dan peranti berkuasa bateri; nilai dalam julat 0.9926 hingga 0.9996 didapati, menunjukkan persetujuan yang boleh diterima. Pendekatan pengekstrakan elektrokinetik ini mewakili peluang baharu untuk memproses sampel semasa pensampelan dan pengangkutan, menjimatkan masa dan mengurangkan pengendalian manual untuk menghasilkan keputusan ujian yang lebih dipercayai dengan cekap.

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## LIST OF ABBREVIATIONS

[EMIM][NTf <sub>2</sub> ]	-	1-ethyl-3-methylimidazolium bis(trifluoromethylsulfonyl)imide
2-NPOE	-	2-nitrophenyl octyl ether
2-NPPE	-	2-nitrophenyl pentyl ether
6-mMP	-	6-methylmercaptapurine
6-TG	-	6-thioguanine
ACN	-	Acetonitrile
AMPA	-	Aminomethylphosphonic acid
AMTCI	-	4-amino-5-(methylthio)carbonyl imidazole
AMX	-	Amoxicillin
ANT	-	Anthranilic acid
ATR	-	Attenuated total reflectance
ATR-FTIR	-	Attenuated total reflectance-Fourier transform infrared
B2EHP	-	bis-(2-ethylhexyl) phosphate
BET	-	Brunauer-Emmett-Teller
Bio-SPME-IVLP	-	Bio-solid-phase microextraction-in vivo lung perfusion
BJH	-	Barrett-Joyner-Halenda
BLM	-	Bulk liquid membrane
C <sup>4</sup> D	-	Capacitively coupled contactless conductivity
CAGR	-	Compound annual growth rate
CAP	-	Cellulose acetate propionate
CBZ	-	Carbamazepine
CE	-	Capillary electrophoresis
CE-C <sup>4</sup> D	-	Capillary electrophoresis-capacitively coupled contactless conductivity detection
CE-ICDOF-LED- FLD	-	Capillary electrophoresis with in-column double optical- fiber light emitting diode-induced fluorescence detection
CE-UV	-	Capillary electrophoresis-ultraviolet detection
CTA	-	Cellulose triacetate
CTB	-	Cellulose tributyrate



CV	-	Cyclic voltammetry
CZE	-	Capillary zone electrophoresis
D2EHPA	-	di-(2-ethylhexyl)phosphoric acid
DART	-	Direct analysis in real time
DBS	-	Dried blood spot
DBSA	-	Dried blood spot autosampler
DC	-	Direct current
DCM	-	Dichloromethane
DDSME	-	Drop-to-drop microextraction
DESI	-	Desorption electrospray ionisation
DI	-	Deionised
DI-SDME	-	Direct immersion single drop microextraction
DI-SPME	-	Direct immersion solid-phase microextraction
DLLME	-	Dispersive liquid-liquid microextraction
DMF	-	Digital microfluidic device
DMPK	-	Drug metabolism and pharmacokinetic
DNA	-	Deoxyribonucleic acid
DNNS	-	Dinonylnaphthalene sulfonic acid
DOP	-	Dioctyl phthalate
DSC	-	Differential scanning calorimetry
DTX	-	Docetaxel
ED	-	Electrodialysis
EE	-	Electroextraction
EE-SPME	-	Electroenhanced solid-phase microextraction
ELISA	-	Enzyme-linked immunosorbent assay
ELM	-	Emulsion liquid membrane
EM-SPME	-	Electromembrane-surrounded solid-phase microextraction
EME	-	Electromembrane extraction
EME-PIM	-	Electromembrane extraction-polymer inclusion membrane
ESI	-	Electrospray ionisation
FA	-	Formic acid
FDA	-	Food drug administration
FE-SEM	-	Field emission-scanning electron microscopy

FIA-UV-Vis	-	Flow injection analysis-ultraviolet visible
FTA	-	Fast transient analysis
FTIR	-	Fourier transform infrared
GAC	-	Green analytical chemistry
GC	-	Gas chromatography
GC-MS	-	Gas chromatography-mass spectrometry
GLYP	-	Glyphosate
HCl	-	Hydrochloric acid
HF-LPME	-	Hollow fiber-liquid-phase microextraction
HIP	-	Hippuric acid
HPLC	-	High-performance liquid chromatography
HPLC-DAD	-	High-performance liquid chromatography-diode array detector
HPLC-FLD	-	High-performance liquid chromatography-fluorescence detector
HS-SPME	-	Headspace-solid-phase microextraction
HSA	-	Human serum albumin
IBU	-	Ibuprofen
IC	-	Ion chromatography
IM	-	Imatinib
InC.	-	Incorporated
IV	-	Intravenous
KTP	-	Ketoprofen
LC	-	Liquid chromatography
LC-MS/MS	-	Liquid chromatography tandem mass spectrometry
LED	-	Light emitting diode
LIF	-	Laser-induced fluorescence
LLE	-	Liquid-liquid extraction
LOD	-	Limit of detection
LOQ	-	Limit of quantification
LPME	-	Liquid-phase microextraction
LTG	-	Lamotrigine
m/z	-	Mass-to-charge ratio

MB	-	Methylene blue
MDMA	-	3,4-methylenedioxy-N-methylamphetamine
MeOH	-	Methanol
MG	-	Malachite green
MMS	-	Miniature mass spectrometer
MRM	-	Multiple reaction monitoring
MS	-	Mass spectrometry
mSPE	-	Magnetic solid-phase extraction
NaOH	-	Sodium hydroxide
NAX	-	Naproxen
NIC	-	Nicotinic acid
NSAID	-	Non-steroidal anti-inflammatory drugs
OFAT	-	one-factor-at-a-time
PADs	-	Paper-based analytical devices
PAN	-	Polyaniline
PCL	-	poly $\epsilon$ -caprolactone
PCR	-	Polymerase chain reaction
PDMS	-	Polydimethylsiloxane
PEDOT	-	Poly(3,4-ethylenedioxythiophene)
pH	-	Power of hydrogen
PIM	-	Polymer inclusion membrane
POC	-	Point-of-care
POU	-	Point-of-use
ppb	-	parts per billion
PPy	-	Polypyrrole
Pt	-	Platinum
PTFE	-	Polytetrafluoroethylene
PVC	-	Polyvinyl chloride
PVDF	-	Polyvinylidene fluoride
PVDF-HFP	-	Poly(vinylidene fluoride-hexafluoropropylene)
RNA	-	Ribonucleic acid
RSD	-	Relative standard deviation
RT	-	Room temperature

RTK	-	Rapid test kit
SAL	-	Salicylic acid
SALLE	-	Salting-out liquid-liquid extraction
SCAP	-	Sample card and preparation
SEM	-	Scanning electron microscopy
SERS	-	Surface-enhanced Raman spectroscopy
SIA-HPLC-MS	-	Sequential injection analysis-high performance liquid chromatography-mass spectrometry
SLM	-	Supported liquid membrane
SPE	-	Solid-phase extraction
SPE-LC-MS/MS	-	Solid-phase extraction-liquid chromatography tandem mass spectrometry
SPME	-	Solid-phase microextraction
TBEP	-	Tris(2-butoxyethyl)phosphate
TDM	-	Therapeutic drug monitoring
TEHP	-	Tris(2-ethylhexyl)phosphate
TGA	-	Thermogravimetric analyser
THF	-	Tetrahydrofuran
TIC	-	Total ion chromatogram
TOMATS	-	Trioctylmethylammonium thiosalicylate
TPU	-	Polyurethane
UPLC	-	Ultra-performance liquid chromatography
UV-Vis	-	Ultraviolet-visible
VAMS	-	Volumetric absorptive microsampling
VPA	-	Valproic acid

## LIST OF SYMBOLS

°	-	degree
%	-	percent
%T	-	percent transmittance
%v/v	-	% volume per volume
%w/v	-	% weight per volume
°C	-	degree celcius
°C/min	-	degree celcius per minute
cm <sup>-1</sup>	-	wavenumber
g/L	-	gram per liter
g/mol	-	gram per mole
J/g	-	Joule per gram
K	-	Kelvin
kV	-	kilovolt
L/Hr	-	liter per hour
m <sup>2</sup> /g	-	square meter per gram
mAH	-	milliamp hour
mg/L	-	milligram per liter
mg/mL	-	milligram per milliliter
mL/g	-	mililiter per gram
mL/min	-	mililiter per minute
mM	-	milimolar
MΩ-cm	-	megaohm-centimeter
ng/mL	-	nanogram per mililiter
ppm	-	parts per million
rpm	-	revolutions per minute
r <sup>2</sup>	-	regression coefficient
x g	-	times gravity
wt% (m/m)	-	percent by weight (mass per mass)
V	-	volt
V/cm	-	volt per centimeter

$\mu\text{A}$	-	microampere
$\mu\text{g/L}$	-	microgram per liter
$\mu\text{g/mL}$	-	microgram per milliliter
$\mu\text{M}$	-	micromolar

# CHAPTER 1

## INTRODUCTION

### 1.1 Background of the Study

Biological fluid samples—such as whole blood, plasma, urine and saliva—generally contain complex matrices which are known to be major obstacles to obtaining an accurate analytical outcome. These matrices, ranging from various inorganic ions and endogenous and exogenous organic molecules to biomacromolecules, are the coexisting impurities that can be adsorbed by the pores of the stationary phase in the analytical columns and adhere to the inner surface of the pipelines. This may lead to a reduction in the lifetime and efficiency of the column as well as to the blockage of the pipeline system (1). Many efforts have been made—in sample collection, sample preparation and instrumental analysis—to improve the existing analytical procedure, remove the coexisting interfering substances, accelerate the procedure, and improve the accuracy of the final analytical outcome. Despite tremendous advances having been made in the development of state-of-the-art analytical instruments for the qualitative and quantitative analysis of analytes in biological fluid samples, sample preparation remains a challenging task for researchers.

Sample preparation is a multi-step procedure that is employed with the purpose of pre-concentration of samples; the process involves the removal of the complex matrices and impurities in order to obtain a higher purity of the target analyte and thus more accurate results. Analytes may be lost by applying inappropriate sample preparation methods. Hence, an appropriate sample preparation method that is able to purify and pre-concentrate target analytes is a fundamental element of reliable sample analysis, particularly for samples with an extremely low concentration of target analytes. Traditional sample preparation methods such as liquid-liquid extraction (LLE) and solid-phase extraction (SPE) are widely employed prior to the analysis of biological fluid samples (2). However, the procedures of LLE and SPE are tedious,

time-consuming and harmful to the environment due to the high volume of solvent consumption. Miniaturized versions of the above sample preparation methods—namely, solid-phase microextraction (SPME), liquid-phase microextraction (LPME) and hollow fiber-liquid-phase microextraction (HF-LPME)—were introduced, offering such advantages as reduced analysis time, low solvent consumption, high extraction efficiency and environmental friendliness (3, 4).

The majority of the miniaturized sample preparation techniques—including SPME, LPME and liquid membrane extraction—are based on passive diffusion, which results in poor efficiency of extraction and unsatisfactory selectivity, particularly for the extraction of polar compounds and charged molecules. Traditionally, derivatisation, alkalization or acidification have been employed to decrease the polarity of the polar compounds in order to improve the partition coefficients of the analytes, hence, the efficiency of extraction (5). However, the derivatisation steps involve expensive and toxic reagents, while some analytes are unstable in extremely acidic or alkali conditions. Therefore, auxiliary energy—including acoustic wave, microwave, thermal and electrical energy—is introduced to the sample preparation procedure to promote energy exchange and the mass transfer of molecules, thus accelerating the entire extraction process, reducing the operation time and improving the efficiency (6). Among these field-assisted techniques, the appeal of electric field driven sample preparation has increased significantly in recent years due to the considerable improvement in the efficiency of extraction that it brings (7).

Electromembrane extraction (EME) is a novel LPME technique that was developed in 2006 by Pedersen-Bjergaard and Rasmussen (8, 9). EME has demonstrated that analytes can be electrokinetically isolated across a supported liquid membrane (SLM). The SLM is a porous hydrophobic supported membrane with a microliter volume of water-immiscible organic solvent immobilised in the pores. Notably, target analytes are ionised and extracted from the sample (donor solution) through an SLM, to an acceptor solution, in the presence of an electric field. In EME, the direct current (DC) electric potential across the SLM is the driving force for electrokinetic migration (10). Moreover, the electric potential controls the selectivity of the extraction as ions move based on the direction of the electric field. For the



extraction of cations, the cathode is placed in the acceptor solution, and the cations move towards the cathode trapped in the acceptor solution; for the extraction of anions, meanwhile, the anode is placed in the acceptor solution and the anions move towards the anode (direction of acceptor solution) (2). The EME method has considerable technical and analytical advantages over the conventional sample preparation methods. A comparative study has been made between HF-LPME and EME in the extraction of polar drugs from biological samples (11). The results proved that EME demonstrated a higher pre-concentration factor and a shorter extraction time; thus it was concluded to be a more efficient method than HF-LPME for extracting and pre-concentrating polar drugs from complex matrices (11). The success of EME in sample preparation has attracted considerable attention regarding the application of electric fields in sample preparation techniques.

For instance, electromembrane-surrounded solid-phase microextraction (EM-SPME) is an electrically enhanced solid-phase extraction technique that combines the merits of EME and SPME. In EM-SPME, the conductive SPME fiber acts as one of the electrodes and is located in the lumen of the hollow fiber, together with the SLM and the acceptor solution (12, 13). Under the application of an electric field, the analytes in the sample solution migrate through the SLM into the acceptor solution and are further adsorbed by the SPME fibers in the hollow fiber lumen (14). EM-SPME demonstrated better sample clean-up than the conventional SPME approach since the SLM around the SPME fibers acts as a “first line of defence” filter that prohibits the complex matrices from entering the acceptor solution, hence avoiding the risk of fiber saturation in the analysis of samples with complicated matrices (15, 16).

The polymer inclusion membrane (PIM) was introduced as an alternative to the SLM in the EME system. The PIMs demonstrated higher mechanical stability and improved mechanical robustness in the presence of an electric field (17). The PIM is a self-supporting liquid membrane that consists of a base polymer, a plasticizer and a carrier. The base polymer is the backbone of the PIM and provides mechanical support, while the plasticizer acts as a solvent in the membrane, providing it with flexibility and softness. The carrier in the PIM plays a significant role as it is the extractant that determines the selectivity of the PIM. A carrier, generally an ionic liquid, initiates ion

exchange reactions and forms ion complexes with target ions. Generally, the PIM is prepared by dissolving the base polymer, plasticizer and carrier in a volatile organic solvent, before casting the membrane solution as a thin film following the evaporation of the solvent. To date, a number of studies on the applications of PIMs in the aforementioned EME system have been published, in which the extraction of target analytes from feed solutions to the acceptor solution through the PIM under the application of an electric field has been demonstrated (18-20). These studies have proven the enhanced stability of the PIM over the SLM; hence, PIMs are more preferred in separation technologies nowadays, particularly in the EME system. Due to the presence of carriers and plasticizers, PIMs have good ionic conductivity and are thus effective ion transporters. By employing the concept of electrophoresis, the electrokinetic extraction of small drug analytes from a dried blood spot (DBS) matrix was conducted on a dried PIM in the absence of liquid reagents (21). The analytes were migrated and extracted through the thin film PIM under the application of an electric field, emphasising the potential and suitability of PIMs for electroextraction. The portability of the developed electrokinetic extraction method allowed for in-transit sample preparation, thus reducing the turn-around time for the analysis (21).

In the sample preparation approaches published thus far, PIMs have generally taken the form of thin films (19, 20, 22, 23) or hollow fibers (24, 25). In this study, the PIM was cast on a non-conductive glass capillary substrate using the dip-coating method and was applied in the form of a miniaturized sampling probe for electrokinetic extraction. No acceptor phase was involved and the voltage was directly applied onto the conductive PIM surface. The PIM-based sampling probe simultaneously acted as an electrode and as the extraction phase in the electrokinetic extraction. The extraction efficiency of this developed PIM sampling probe was studied using a charged fluorescent dye molecule, Rhodamine 6G, under the application of an electric field. The PIM sampling probe is potentially capable of extracting, preconcentrating and storing target analytes prior to the sample analysis.

With the rising awareness of the importance of immediate decentralised sample processing to prevent degradation and adulteration of samples, a large amount of research has been conducted on the development of miniaturized and portable

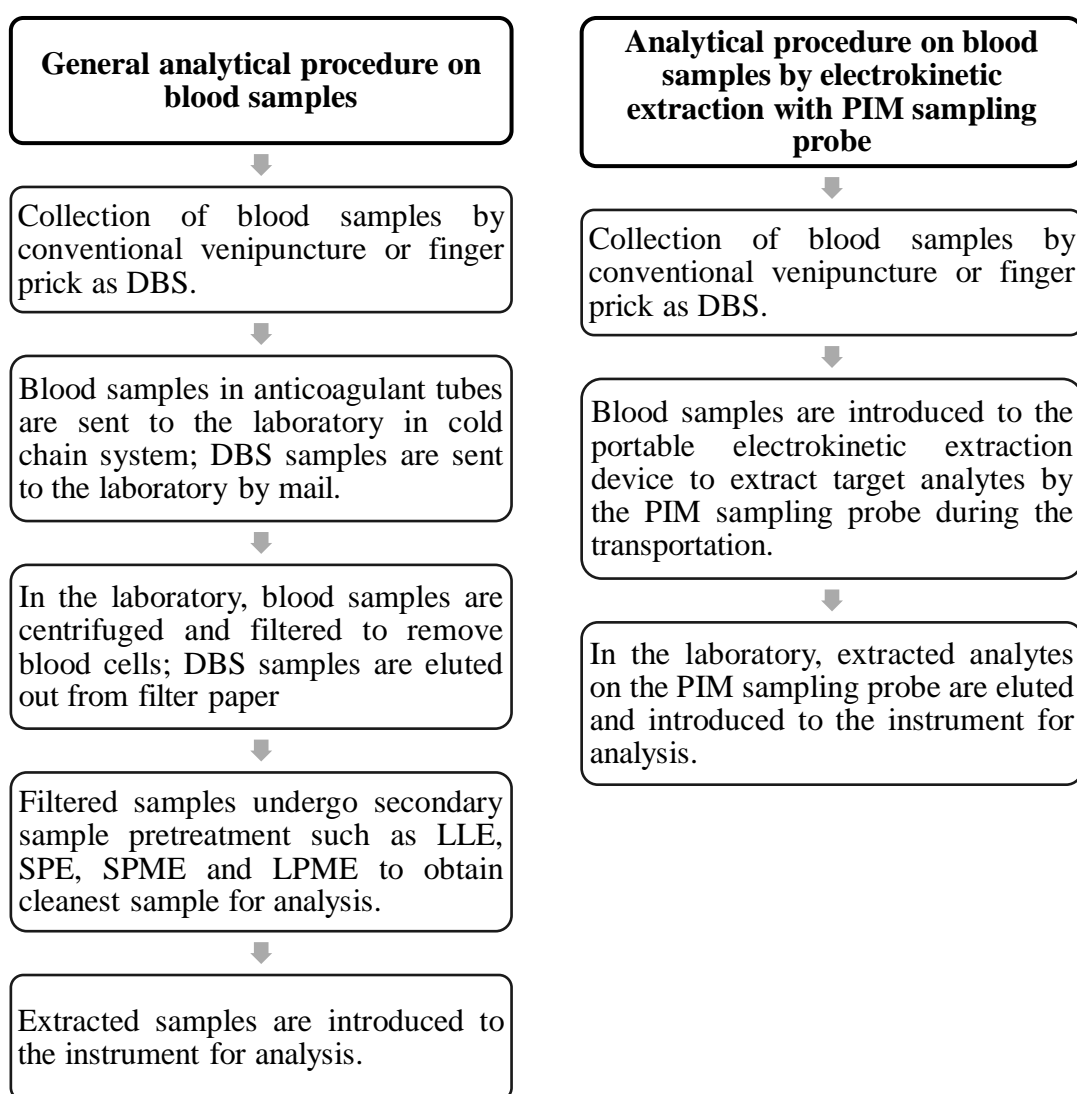
analytical methods. For example, a number of portable and easy-to-use point-of-care (POC) diagnostic tests have been developed for the immediate analysis of patients' samples at the sampling point for disease screening (26).

Patient-centric microsampling techniques such as DBS and volumetric absorptive microsampling (VAMS) (27) are less invasive blood sampling methods that extract only a microliter volume of capillary blood from a finger prick. These approaches allow patients to conduct self-sampling in a familiar and convenient environment (such as the home) thus reducing their frequency of visits to health facilities (28). They have particular benefits for newborn screening, therapeutic drug monitoring (TDM) and disease diagnostics (29). In spite of the advantages and the wide range of applications of microsampling techniques, the majority of the DBS samples are sent by mail and processed in the laboratory. The general sample for preparation of DBS samples comprises sonication, centrifugation, extraction and reconstitution; these processes are required to be conducted in the designated laboratory. To date, the development of a portable sample preparation technique for processing DBS samples during the transportation is hitherto unreported.

In this study, the optimised PIM sampling probe was employed to electrokinetically extract the anticancer drug doxorubicin from DBS, human plasma and human serum samples. Doxorubicin has demonstrated a narrow therapeutic window and serious side effects of cardiotoxicity (30). It is essential that the drug intake is monitored, particularly for childhood cancer survivors treated with doxorubicin, in order to prevent severe cardiac adverse events. In contrast to the existing analytical workflow, the sampling probe allows sample preparation to be conducted during transportation, with an electric field as its driving force. Consequently, it accelerates the entire analytical workflow for TDM and facilitates immediate analytical outcomes for patients. Figure 1.1 shows the comparison between the general analytical workflow and the analytical procedure by using electrokinetic extraction with a PIM sampling probe. PIM sampling probe simplifies the analytical workflow by electrokinetic extraction of analytes during the transportation of samples to the laboratory. To enable the portability of the electrokinetic extraction set up for sample preparation during transportation, the PIM sampling probe was further

integrated into a portable battery-powered electrokinetic extraction device. This study focuses on the design and demonstration of the prototype of the device. A lower voltage was applied to the PIM sampling probe for the electrokinetic extraction. The extracted analytes on the PIM sampling probe were further desorbed and quantified using liquid chromatography with tandem mass spectrometry (LC-MS/MS) in the laboratory in order to obtain more accurate and higher-sensitivity analytical results.

Figure 1.1 Comparison between the general analytical workflow and the analytical procedure by using electrokinetic extraction with PIM sampling probe.



## 1.2 Problem Statement

Nowadays, the preanalytical stage of sample collection and preparation is a time-consuming procedure, and samples are generally processed and analysed following their arrival in the laboratory. Many test results are inconclusive, sometimes incorrect, due to the fact that the sample was not stabilised for analysis by the central laboratory test. Moreover, there are also significant challenges in terms of the transportation of stabilised samples as the transportation network between the sampling site and many central laboratories are inadequate. The inconvenience of transporting samples from rural, remote areas to the centralised laboratory services may lead to low-quality of analytical and diagnostic services. In addition, traditional methods for the collection of liquid samples in bulk quantities involve high-cost cold chain systems during transportation to preserve the liquid samples for subsequent pretreatment and analysis in the laboratory. Sample preparation steps that employ huge amounts of toxic solvents for extraction are not environmentally friendly and are commonly performed in the laboratory after samples have been received. Patients are required to frequently visit healthcare premises for their routine medical check-ups, which is inconvenient, particularly for the elderly, children and patients who require TDM. Even before the outbreak of the COVID-19 pandemic, the decentralisation and miniaturization of sample preparation procedures have been a topic of great interest to analytical scientists.

The world has had to change and evolve in a variety of ways in order to adapt to the ongoing COVID-19 pandemic. Since the outbreak of the pandemic, there has been a growing trend for the decentralisation and digitalisation of healthcare services since most people have deemed visiting healthcare facilities for routine visits to be a high-risk course of action (31). Therefore, the only way for sample collection and preparation methods to keep up with this trend will be to introduce a simple and user-friendly portable device for patients to be able to closely monitor their conditions during pandemics.

The DBS method is a less-invasive and patient-friendly microsampling technique that can be easily performed by patients at home with adequate training.

Blood samples are collected from a finger prick and dried on a paper substrate for at least three hours before being sent to the laboratory for sample preparation and instrumental analysis. DBS samples are relatively stable; hence they are sent by mail to the laboratory, avoiding the need for a cold-chain system. Nevertheless, the surrounding humidity and moisture content might destroy the DBS samples. DBS require a significant sample preparation procedure to meet satisfactory levels of reliability on analytical results. In the laboratory, DBS samples are desorbed with suitable solvents and are further processed using traditional sample preparation techniques, including protein precipitation and LLE (32, 33). To date, no research has been conducted on the processing of DBS samples during transportation using an electric field, hence the motivation for this study.

A portable battery-powered electrokinetic extraction device using electric field driven sample preparation was developed in this study. The device is designed for the electrokinetic extraction of drug analytes from DBS, human plasma and human serum samples during sample transportation. The core of the device is a miniaturized PIM-based sampling probe, the PIM being fabricated using a simple dip-coating method. The PIM is robust, with a strong mechanical structure, and exhibits excellent performance in adsorbing targeted analytes in sample matrices under an electric field. In addition, the new sampling probe is quick and easy to manufacture and, most importantly, cost-effective, thus benefiting end users in the commercial market.

### **1.3 Objectives of the Study**

This project focuses on the development of a new self-assembly PIM sampling probe for electric field driven extraction. The aims of this proposal are:

- (a) To evaluate the capability and performance of a new self-assembly PIM sampling probe—comprising a PIM dip-coated on a non-conductive glass capillary substrate—in an electric field driven extraction.
- (b) To determine the optimal PIM composition and operating conditions for the electric field driven extraction of target charged analytes on the PIM sampling probe.

- (c) To validate and apply the developed approach as a new method of electrokinetic extraction of doxorubicin from human plasma, human serum and DBS samples.
- (d) To investigate the efficiency and feasibility of the PIM sampling probe integrated with a portable battery-powered device for electrokinetic extraction of doxorubicin.

#### **1.4 Scopes of the Research**

In this study, a PIM-based sampling probe was developed to electrokinetically extract doxorubicin from DBS, human plasma and human serum samples. The potential of the PIM sampling probe for application in electrokinetic extraction was investigated and monitored using the cationic fluorescent dye, Rhodamine 6G. In this work, the PIM sampling probes were designed, prepared and characterised for electrokinetic extraction. The PIM sampling probes were characterised using field emission-scanning electron microscopy (FE-SEM), nitrogen adsorption/ desorption analysis and attenuated total reflectance-Fourier transform infrared (ATR-FTIR) spectroscopy analysis. The thermal stability and phase transitions of the PIM were studied using thermogravimetric analysis (TGA) and differential scanning calorimetry (DSC) (Objective a).

A series of optimisation studies were performed to determine the composition of the PIM, including the amount of carrier (1-ethyl-3-methylimidazolium bis(trifluoromethylsulfonyl)imide [EMIM][NTf<sub>2</sub>]) in the range of 0 to 75 mg, the amount of plasticizer (2-nitrophenyloctyl ether [2-NPOE]) in the range of 0 to 150 mg, the amount of base polymer (cellulose triacetate [CTA]) in the range of 50 to 150 mg, and the thickness of the PIM on the glass capillary (controlled by the number of dipping cycles of the glass capillary in the membrane solution). The electrophoretic parameters, such as voltage applied (0 – 500 V) and pH of the solution (pH 3 – pH 8), were optimised to determine the effectiveness of the extraction. Analytical performances of the developed PIM sampling probe on the extraction of Rhodamine 6G dye were evaluated and visualised using a handheld fluorescence microscope. The

fluorescence intensity of the extracted Rhodamine 6G was quantified using ImageJ software (Objective b).

The performance of the PIM sampling probe was further evaluated by extracting the fluorescent anticancer drug doxorubicin from real biological fluid samples under the application of an electric field. The doxorubicin in DBS, human plasma and human serum samples was electrokinetically extracted using the PIM sampling probe and then desorbed in the suitable desorption solvent for LC-MS/MS analysis. The LC-MS/MS procedure assessed the practicability and reliability of the PIM sampling probe by quantifying the desorption of extracted doxorubicin from the sampling probe. The probe was applied to the electrokinetic extraction of doxorubicin from human plasma, human serum and DBS samples in order to evaluate and validate the analytical performance of the developed method. The method validation parameters include linearity and range, the limit of quantification (LOQ), the limit of detection (LOD), recovery and repeatability (Objective c). Ultimately, the PIM sampling probe was integrated into a portable battery-powered electrokinetic extraction device prototype to improve the portability of the setup, and the effectiveness of electrokinetic extraction of the developed device was evaluated (Objective d).

## **1.5 Significance of the Study**

In this study, the electrokinetic extraction capabilities of a portable miniaturized PIM-based sampling probe were investigated. The developed PIM sampling probe is envisioned to provide significant improvements in sampling efficiency and sample preparation procedures as well as in the transportation of samples in the presence of an electric field. In addition, the presence of this electric field in the extraction process means that liquid samples can be extracted with solvents in microliter volumes, resulting in a more environmentally friendly, cost-effective procedure.



Furthermore, the process of shipping conventional samples is both costly and time-consuming, particularly in some large countries; thus, our developed portable battery-powered electrokinetic extraction device brings significant benefits to the nation. The ability to conduct sample preparation during transportation using this portable device will reduce the incidence of mortality resulting from delayed diagnoses. It will shorten the time required to reach a clinical outcome following sample collection. Since the analytes are extracted and stored in the PIM sampling probe during sample transportation, it will also lead to reduced shipping costs. Transformation of remote clinical outposts into high-quality analytical and diagnostic units is also achievable with the introduction of this device. In addition, with this novel portable sampling probe, Malaysia can position itself as a pioneer in the development of portable sample preparation technology.

From an economic standpoint, the global sample preparation market size is estimated to increase at a compound annual growth rate (CAGR) of 6.6% from \$7.57 billion in 2021 to \$8.06 billion in 2022. The market is projected to rise to \$9.84 billion in 2026 at a CAGR of 5.1% (34). Therefore, the development of this portable electrokinetic extraction device into a commercially viable product is expected to bring significant economic gains to the country as one of the advanced sample preparation techniques.

## **1.6 Outline of the Thesis**

This thesis comprises five chapters.

Chapter 1 gives a detailed description of the research background, problem statement, objectives, scope and significance of the project.

Chapter 2 provides a review of the literature on existing miniaturized sample preparation techniques, electric field enhanced sample preparation techniques, PIMs, miniaturized blood sampling techniques, miniaturized analytical tools, TDM and doxorubicin.

Chapter 3 describes the method of fabrication of the PIM sampling probe, the set parameters for each characterisation technique, the sample and standard solution preparation procedure, the electrokinetic extraction platform, the electrokinetic extraction procedure in real sample matrices, and the quantification assays using HPLC-FLD and LC-MS/MS.

Chapter 4 discusses the configuration and physical characteristics of the PIM sampling probe, followed by a discussion on the optimisations of PIM composition, the number of dipping cycles in the PIM sampling probe fabrication process and operating conditions for electrokinetic extraction. The application of the developed electrokinetic extraction setup was firstly demonstrated through the visualisation of the fluorescent dye Rhodamine 6G extracted on the PIM sampling probe. This chapter also reports the application of the PIM sampling probe in the electrokinetic extraction of the anticancer drug doxorubicin from biological fluid samples, including human plasma, human serum and DBS samples. Spiked doxorubicin was electrokinetically extracted using the PIM sampling probe; this was followed by desorption of the extracted analytes in a suitable solvent for LC-MS/MS analysis. The PIM sampling probe was further integrated into a portable battery-powered device for in-transit sample preparation. The performance of the PIM sampling probe in the electrokinetic extraction of doxorubicin in real sample matrices was validated.

Chapter 5 provides a conclusion to the thesis. The main elements of the study—including the design, the optimal conditions and the applications of the electrokinetic extraction technique—are summarised. Suggestions for future studies are also presented.

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## LIST OF PUBLICATIONS

### Journal with Impact Factor

1. **Tey, H. Y.** & See, H. H. (2021). A review of recent advances in microsampling techniques of biological fluids for therapeutic drug monitoring. *J Chromatogr A*, 1635, pp. 461731. doi:10.1016/j.chroma.2020.461731 (Q1, IF: 4.759)
2. **Tey, H. Y.**, Breadmore, M. C. & See, H. H. (2022). Electrokinetic Extraction of Drug Analytes from Biological Fluids by Polymer Inclusion Membrane (PIM). *Analytical Chemistry* (Submitted) (Q1, IF: 8.008)

### Intellectual Property (IP)

#### Patents

1. See, H. H., **Tey, H. Y.**, Breadmore, M. C. & Nur H. (2021). *Polymer Inclusion Membrane Sampling Probe for Solvent-less Electric Field Driven Extraction, Methods and Assembly Thereof* (Patent No: PI2021006275)

### Conference Presentations

1. **Tey, H. Y.** & See, H. H., “Solid Sampling Probe based on a Polymer Inclusion Membrane.” Oral presentation at 36<sup>th</sup> International Symposium on Microscale Separations and Bioanalysis (e-MSB 2020). September 27 – 30, 2020. Virtual symposium organised by The Society for Microscale Separations and Bioanalysis (SMSB).
2. **Tey, H. Y.** & See, H. H., “Polymer Inclusion Membrane (PIM) Sampling Probe for Electric Field Driven Extraction of Drug Analytes from Biological Fluids.” Oral presentation at 38<sup>th</sup> International Symposium on Microscale Separations and Bioanalysis (MSB 2022). Liège, Belgium, July 3 – 6, 2022. Symposium organised by The Society for Microscale Separations and Bioanalysis (SMSB).

3. **Tey, H. Y.** & See, H. H., “Polymer Inclusion Membrane Sampling Probe for Electric Field Driven Extraction of Drug Analytes from Biological Fluids.” Oral presentation at 2022 AOAC SEA 1<sup>st</sup> Annual Conference, Singapore, October 12 – 13, 2022. Annual Meeting organised by AOAC SEA Section.