IMPORVED CHARACTERISTICS OF OPTIMIZED YTTRIA-STABILIZED ZIRCONIA THIN FILM GROWN USING SOL-GEL DIP-COATING TECHNIQUE

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DEDICATION

To my beloved husband and daughter, for your unconditional love, motivation, support and especially time. We finally go through this together. This is for you both.

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ABSTRACT

Clean and sustainable energy production has become the priority of the 21^{st} century to save Earth's environment from ever-increasing pollution and climate change. Solid Oxide Fuel Cell (SOFC) is one of the emergent technologies that which can produce a vast amount of clean energy using water; thus, considered to be renewable, sustainable, and pollution-free. All previous studies have used high-temperature SOFC electrolyte preparation methods which were expensive and cumbersome. Based on these factors, in this study, good quality yttriastabilized zirconia (YSZ) electrolyte thin film was synthesizing using soda lime glass and sapphire as a substrate by sol-gel dip-coating method which use low temperature preparation method for YSZ thin film. Several solvents were tested to obtain homogeneous suspension of YSZ such as water, acetone and alcohol and it was found out that alcohol is the suitable one. The samples then were characterized by diverse analytical techniques to determine their potency for the SOFC applications. As-deposited thin films were characterized using X-ray diffractometer (XRD) analyses, atomic force microscopy (AFM), Raman spectroscopy, field emission scanning electron microscopy (FESEM) and Four-point probe techniques. The impact of various substrates, sintering temperatures and layer-count on the structural, morphological and electrical properties of these electrolyte thin films was evaluated. Furthermore, the optimum thin film was chosen to demonstrate its viability as an effective SOFC electrolyte. XRD analysis showed that the YSZ thin films deposited on soda-lime glass and sapphire were highly amorphous and crystalline in nature, respectively. Hence, the sapphire substrate was selected for further deposition of thin films. To improve the overall characteristics of the YSZ films, two types of heat treatments like sintering and normalizing, were performed. The film which had been sintered at 1300 °C revealed the formation of crystallites with lower density than the normalized film. Conversely, the film which had been sintered above 1300 °C showed much better morphology compared to those obtained using the normalized heat treatment. The single-layer YSZ film which had been sintered at 1300 °C exhibited excellent crystallinity, dense morphology with fewer cracks, very low porosity and high electrical conductivity (2.905 S/m). The double-layered YSZ film sintered at 1400 °C displayed the highest conductivity (3.552 S/m) and best crystalline density (5.976 g/cm³). An increase in the number of layers from three to ten was found to degrade the crystallinity, density and morphology of the grown films. The YSZ electrolyte films deposited with up to three layers were not suitable for practical use in the SOFC. Therefore, the double-layered film which had been sintered at 1400 °C was chosen as the optimum film. The growth of the proposed films was explained using various mechanisms to provide new insight into the preparation of the YSZ electrolyte thin films with tailored properties via the low-cost and simple method at low temperatures. This study focused only on two types of substrates, a few sintering temperatures and one preparation method. However, another synthesis method, substrate and characterization technique such as spin-coating, porous anodic alumina and UV-Visible absorption or emission spectroscopy, respectively, may be useful for better sample optimization. The use of lower temperature for the deposition of the thin films can yield higher electrical conductivity of the electrolyte useful for room temperature operation of SOFC. Based on the results it can be concluded that the interplay of the type of substrate, sintering temperature and the number of coating layers plays a vital role in obtaining high-quality electrolyte thin film, showing that synthesis of the YSZ thin film is essential thus the best sample is established.

ABSTRAK

Pengeluaran tenaga yang bersih dan mampan menjadi keutamaan pada abad ke-21 ini untuk menyelamatkan alam sekitar daripada pencemaran dan perubahan iklim yang semakin meningkat. Sel Fuel Oksida Pepejal (SOFC) jalah salah satu teknologi baru yang boleh menghasilkan sejumlah besar tenaga yang bersih dengan menggunakan air, dengan itu ia dianggap boleh diperbaharui, mampan dan juga bebas pencemaran. Kesemua kajian yang terdahulu menggunakan suhu tinggi dalam kaedah penyediaan elektrolit SOFC yang sangat mahal dan rumit. Berdasarkan faktor-faktor ini, dalam kajian ini, filem nipis elektrolit yttria zirkonia terstabil (YSZ) berkualiti tinggi telah disintesis menggunakan substrat kaca sodakapur dan batu nilam sebagai substrat dengan kaedah salutan celup dalam sol-gel yang menggunakan suhu rendah untuk filem nipis YSZ. Beberapa pelarut telah diuji untuk mendapatkan ampaian homogen YSZ seperti air, aseton dan alkohol dan telah dikenalpasti bahawa alkohol adalah yang paling sesuai. Sampel kemudiannya dicirikan oleh pelbagai teknik analisis bagi menentukan potensi mereka untuk penggunaan aplikasi SOFC. Filem nipis telah dicirikan menggunakan analisis pembelauan sinar-X (XRD), mikroskopi daya atom (AFM), spektroskopi Raman, mikroskop imbasan pancaran medan elektron (FESEM) dan penduga empat titik. Kesan kepelbagaian substrat, suhu pensinteran dan kiraan lapisan kepada sifat struktur, morfologi dan elektrik filem nipis ini telah dinilai. Tambahan lagi, filem nipis optimum dipilih untuk menunjukkan kelangsungannya sebagai elektrolit SOFC yang berkesan. Analisis XRD menunjukkan bahawa filem YSZ yang dimendapkan pada substrat kaca dan nilam masing-masing adalah sangat amorfus dan berhablur. Oleh itu, substrat nilam telah dipilih untuk pemendapan filem selanjutnya. Bagi menambah baik ciri-ciri keseluruhan filem YSZ, dua jenis rawatan iaitu pensinteran dan penormalan telah dilakukan. Filem yang disinter pada suhu 1300 °C mendedahkan pembentukan hablur dengan ketumpatan yang lebih rendah daripada filem yang dinormalkan. Sebaliknya, filem yang disinter di atas 1300 °C menunjukkan morfologi yang lebih baik berbanding dengan yang diperoleh menggunakan rawatan haba ternormal. Filem YSZ lapisan tunggal yang disinter pada suhu 1300 °C mempamerkan kehabluran yang sangat baik, morfologi yang tumpat dengan retakan yang sedikit, keliangan yang sangat rendah dan kekonduksian elektrik yang tinggi (2.9047 S/m). Dua lapisan filem YSZ yang disinter pada suhu 1400 °C memaparkan kekonduksian tertinggi (3.5516 S/m) dan ketumpatan hablur terbaik (5.976 g/cm³). Peningkatan bilangan lapisan daripada tiga ke sepuluh lapisan didapati merendahkan kehabluran, ketumpatan dan morfologi filem. Filem elektrolit YSZ yang dimendapkan sebanyak tiga lapisan ke atas tidak sesuai untuk kegunaan praktikal dalam SOFC. Oleh itu, dua lapisan filem yang disinter pada suhu 1400 °C telah dipilih sebagai sampel yang optimum. Pertumbuhan filem telah dijelaskan dengan kepelbagaian mekanisme untuk memberikan pandangan yang baru tentang penyediaan filem nipis elektrolit YSZ dengan sifat yang sesuai melalui kaedah yang berkos rendah dan mudah pada suhu rendah. Walaupun kajian ini hanya tertumpu kepada dua jenis substrat, beberapa suhu pensinteran dan satu kaedah penyediaan, bagaimanapun, kaedah sintesis, substrat dan teknik pencirian yang lain seperti salutan putaran, alumina anodik berliang dan spektroskopi penyerapan atau pelepasan ultralembayung-cahaya nampak, masing-masing mungkin berguna untuk pengoptimuman sampel yang lebih baik. Penggunaan suhu yang lebih rendah untuk filem nipis boleh menghasilkan elektrolit dengan kekonduksian eletrik yang lebih tinggi untuk digunakan dalam SOFC beroperasi suhu bilik. Berdasarkan keputusan ini, ia dapat disimpulkan bahawa interaksi diantara jenis substrat, suhu pensinteran dan bilangan lapisan salutan memainkan peranan penting dalam mendapatkan filem nipis elektrolit berkualiti tinggi lantas menunjukkan bahawa sintesis filem nipis YSZ adalah penting justeru itu sampel yang terbaik telah dibuktikan.

TABLE OF CONTENTS

TITLE

	DEC	LARATION	iiii
	DED	DICATION	iv
	ACK	v	
	ABS'	TRACT	vi
	ABS'	vii	
	TAB	LE OF CONTENTS	viii
	LIST	Γ OF TABLES	xii
	LIST	FOF FIGURES	xiv
	LIST	COF ABBREVIATIONS	xvxvii
	LIST	COF SYMBOLS	xvxviii
	LIST	COF APPENDICES	xix
СНАРТЕ	R 1	INTRODUCTION	1
	1.1	Background	1
	1.2	Problem Statement	3
	1.3	Research Objectives	5
	1.4	Scope of Study	5
	1.5	Significance of Study	7
	1.6	Outline of Thesis	7

CHAPTER 2 LITERATURE REVIEW 9 9 2.1 Introduction 2.2 9 Basic Of The Fuel Cell 2.3 Solid Oxide Fuel Cell 10 2.3.1 Planar SOFC 12 2.3.2 Tubular SOFC 13 Component of SOFC 2.4 14

2.4.1 Electrolyte 15

	2.4.2	Anode		18
	2.4.3	Cathode		19
2.5		Oxide Fue nability	l Cell contribution for environmental	20
2.6	Zircon	ium, Zirco	nia and Yttria Stabilized Zirconia	21
2.7	Growt Film	h Mechani	sm of Yttria Stabilized Zirconia Thin	22
	2.7.1	Nucleatio	n	23
		2.7.1.1	Homogeneous Nucleation	24
		2.7.1.2	Heterogeneous Nucleation	24
	2.7.2	Growth S	tage	27
2.8	Sol-ge	l Techniqu	ie	28
	2.8.1	Sol-gel T	hin Film	30
	2.8.2	Dip-coati	ng Technique	30
	2.8.3	Spin-coat	ing Technique	32
	2.8.4	Comparat	tive Literature Summary	34
2.9	Substr	ates		38
	2.9.1	Sapphire	Wafer	38
	2.9.2	Soda Lim	e Glass	40
2.10	Therm	al Treatme	ent of Thin Film	41
	2.10.1	Sintering	of YSZ Thin Film	42
	2.10.2	Normaliz	ing of YSZ Thin Film	43
2.11	Backg	round Stuc	lies on Characterization	43
	2.11.1	Structural	Properties	43
		2.11.1.1	X-ray Diffraction	43
		2.11.1.2	Raman Spectroscopy	47
	2.11.2	Morpholo	ogical Properties	50
		2.11.2.1	Atomic Force Microscopy	50
		2.11.2.2	Field Emission Scanning Electron Microscopy	55
	2.11.3	Electrical	Properties	57
		2.11.3.1	Four-Point Probe	57

CHAPTER 3	RESE	EARCH M	IETHODOLOGY	61
3.1	Introd	uction		61
3.2	Prepa	ration of Y	SZ electrolyte Thin Film	63
	3.2.1	Chemica	ls reagents for sample preparation	63
	3.2.2	Substrate	e Cleaning	63
3.3	Sol-ge	el Suspens	ion Preparation	63
3.4	Dip co	oating Pro	cess	64
3.5	Grow	th Parame	ters	65
	3.5.1	Sintering	g Temperature	65
	3.5.2	Layer of	YSZ Thin Film Coating	66
CHAPTER 4	RESU	JLTS AN	D DISCUSSION	69
4.1	Introd	uction		69
4.2	Influe	nce of Sul	ostrates on YSZ Thin Films Growth	69
	4.2.1		f Soda Lime Glass Substrate on YSZ m Structures	69
	4.2.2	Thin F	f Sapphire Wafer Substrate on YSZ ilm Structural, Morphological and al Characteristic	72
		4.2.2.1	Heat Treatment Influence on Various Properties	72
4.3			er of Layers and Sintering Temperature ies of YSZ Thin Films	77
	4.3.1	Impact of	of One Layer on YSZ Thin Film	77
		4.3.1.1	XRD Analysis	77
		4.3.1.2	Raman Spectra of Grown Film	80
		4.3.1.3	Atomic Force Micrographs of Films	83
		4.3.1.4	Field Emission Scanning Electron Images	85
		4.3.1.5	Electrical Properties of Thin Films	88
	4.3.2	Impact of	of Two Layer on YSZ Thin Film	91
		4.3.2.1	XRD Analysis	91
		4.3.2.2	Raman Spectra of Grown Film	93
		4.3.2.3	Atomic Force Topographs of Films	94

		4.3.2.4	Field Emission Scanning Electron Images	96
		4.3.2.5	Electrical Properties of Thin Films	98
	4.3.3	Impact of	of Three Layer on YSZ Thin Film	102
		4.3.3.1	XRD Analysis	102
		4.3.3.2	Atomic Force Micrographs of Films	103
		4.3.3.3	Field Emission Scanning Electron Images	104
	4.3.4	Impact of	of Five and 10 Layer on YSZ Thin Film	107
		4.3.4.1	XRD Analysis	107
		4.3.4.2	Atomic Force Micrographs of Films	108
		4.3.4.3	Field Emission Scanning Electron Images	109
	4.3.5	Results Films	for Optimum YSZ Electrolyte Thin	111
CHAPTER 5	CON	CLUSIO	N AND RECOMMENDATIONS	121
5.1	Introd	uction		121
5.2	Concl	usions		121
5.3	Future	e Works		123
REFERENCES				125
LIST OF PUBLI	(CATIO	ONS		143

LIST OF TABLES

TABLE NO.	TITLE	PAGE
Table 2.1	Previous literature for dip coating method	35
Table 2.2	The properties of Al ₂ O ₃	39
Table 2.3	Properties of the soda lime glass	40
Table 2.4	Raman bands assignment and corresponding modes	50
Table 3.1	Sintering Temperature and Layers of Coating used as parameters	67
Table 4.1	FWHM calculated for both sintering and normalizing process	74
Table 4.2	Estimate average value of crystallite size, lattice parameter, microstrain and density of the proposed samples woth codes (Chapter 3) thin films	80
Table 4.3	Raman band assignments	82
Table 4.4	Estimated value of RMS roughness, average roughness, grain size and also crystallite size from XRD analysis of sintered YSZ thin films	85
Table 4.5	Comparison of estimated grain size and crystallite size of YSZ thin films electrolyte between FESEM, AFM and XRD analysis	86
Table 4.6	Summary of structrual, morphological and electrical properties analysis of YSZ thin films with one-layer coating toward different sintering temperature	90
Table 4.7	Estimate average value of crystallite size, lattice parameter, microstrain and density of YSZ thin films with two-layers coating sintered at 1000 °C, 1300 °C, and 1400 °C	93
Table 4.8	Estimate value of RMS roughness, Average roughness and Grain Size of YSZ thin films with two-layers coating sintered at 1000 °C, 1300 °C, and 1400 °C	95
Table 4.9	Summary of structural and electrical properties analysis of YSZ thin films with one and two-layers coating toward different sintering temperature	100
Table 4.10	Comparative result for three-layers coating at different sintering temperature with one-and two-layers coating	106

Table 4.11	Summary results of different coating laters from one- to 10 layers sintered at 1300 °C	111
Table 4.12	Summary of the observed structural, morphological and electrical properties of YSZ electrolyte thin films (N: None, H: High and L: Low)	113
Table 4.13	Comparative summary of the sample properties and their usefulness (L: Low usage, H: High usage)	116
Table 4.14	Comparison of the present work with other reports (NR: Not Report).	120

LIST OF FIGURES

FIGURE NO	D. TITLE	PAGE
Figure 2.1	A simple schematic diagram of a fuel cell	10
Figure 2.2	Schematic for a planar SOFC design	13
Figure 2.3	Schematic for a tubular SOFC design	14
Figure 2.4	Schematic diagram of ZrO2 crystal structure: (a) monoclinic, (b) tetragonal, (c) cubic	16
Figure 2.5	YSZ cubic fluorite structure	17
Figure 2.6	Schematic of foam morphology difference between homogeneous and heterogeneous nucleation	24
Figure 2.7	Schematic illustrating of the heterogeneous nucleation process with all related surface in equilibrium	25
Figure 2.8	A schematic diagrams of three fundamental modes of early nucelation in the film growth. (a) Island or Volmer-Weber growth, (b) Layer or Frank-van der Merwe growth, (c) Island-layer or Stranski-Krastonov growth	27
Figure 2.9	Fundamental stages of sol-gel dip coating	31
Figure 2.10	Schematic of the four stage of spin coating	33
Figure 2.11	Al ₂ O ₃ hexagonal crystal structure	38
Figure 2.12	The adjacent SiO ₄ tetrahedral-type unit cell	40
Figure 2.13	Schematic diagram of the reflection of Bragg from lattice planes	44
Figure 2.14	XRD instrument schematic diagram	46
Figure 2.15	Raman scattering due to excitation of molecule by laser beam	48
Figure 2.16	Scattering of Rayleigh, Stokes and Anti-Stokes	49
Figure 2.17	Schematic diagram of Atomic Force Microscopy	51
Figure 2.18	Different modes of Atomic Force Microscopy	52
Figure 2.19	Elastic and inelastic interaction of electron beam with an atom	54
Figure 2.20	FESEM schematic diagram	55

Figure 2.21	SEM images of the surface of YSZ film sintered for 4 h at: (a)1300 °C, (b)1350 °C, (c)1400 °C, (d)1450 °C, (e)1500 °C, (f)1600 °C	57
Figure 2.22	A schematic diagram of four-point probe showing current flow and voltage measurement	58
Figure 3.1	Flow chart of the study including sample characterization of the samples	62
Figure 3.2	Schematic diagram of YSZ sol-gel suspension preparation	64
Figure 3.3	Graphical represent the heat treatment for both soda lime glass and sapphire	66
Figure 4.1	Sintering temperature dependant XRD patterns of YSZ film for one layer	70
Figure 4.2	Sintering temperature dependant XRD patterns of YSZ film for seven layer	70
Figure 4.3	XRD patterns of YSZ thin films containing one, five and seven layers sintered at 500 $^{\circ}$ C	71
Figure 4.4	XRD patterns of YSZ thin films containing one, five and seven layers sintered at 600 $^{\circ}$ C	71
Figure 4.5	XRD patterns between (a) normalizing and (b) sintering at heat treatment 1300 $^{\circ}\mathrm{C}$	74
Figure 4.6	Images of sintering process (a) 1300 °C, (c) 1400 °C, (e)1500 °C and normalizing process (b) 1300 °C, (d) 1400 °C, (f) 1500 °C	76
Figure 4.7	(a) The XRD patterns of YSZ thin films with one-layer coating sintered at different temperatures and (b) magnified view of one-layer coating sintered at 1300 and 1400 °C	79
Figure 4.8	Raman spectra of YSZ thin films sintered at 1300 °C, 1400 °C and 1500 °C	82
Figure 4.9	3D AFM images of YSZ (a) YSZ1_10, (b) YSZ1_11, (c) YSZ1_12, (d) YSZ1_13, (e) YSZ1_14 and (f) YSZ1_15	84
Figure 4.10	FESEM morphology images of YSZ thin films with particle size distribution for sintering temperature (a) 1300 °C, (b) 1400 °C and (c) 1500 °C	87
Figure 4.11	The grain size controlled by different sintering temperature	88
Figure 4.12	Conductivity value of one-layer film coating sintered at 1000 °C, 1300 °C and 1400 °C	90

Figure 4.13	(a) XRD patterns for YSZ thin films with two-layers coating at sintering temperature 1000 °C, 1300 °C and 1400 °C and (b) magnified view of highest intensity peak for cubic YSZ thin films occurred at θ : 30.2° refer to the cubic phase of the film and highest intensity for monoclinic phase appear at θ : 28.1° and 31.4°	92
Figure 4.14	Raman Spectra of YSZ thin films with two-layers coating sintered at 1300 °C and 1400 °C	94
Figure 4.15	3D images of AFM for YSZ thin films of two-layers coating sintered at (a) YSZ2_10, (b) YSZ2_13 and (c) YSZ2_14	96
Figure 4.16	Top view (left) of FESEM images and grain size distribution of YSZ thin films with two-layers coating sintered at (a) 1000 °C, (b) 1300 °C and (c) 1400 °C	97
Figure 4.17	Grain Size results of two-layers coating YSZ thin films sintered at 1000 °C, 1300 °C and 1400 °C	98
Figure 4.18	Conductivity value of two-layers films sintered at 1000 °C, 1300 °C and 1400 °C	99
Figure 4.19	XRD patterns of YSZ thin films with three-layers sintered 1300 $^{\circ}\mathrm{C}$ and 1400 $^{\circ}\mathrm{C}$	103
Figure 4.20	3D-images of AFM for YSZ thin films with three-layers sintered at 1300 $^{\circ}$ C and 1400 $^{\circ}$ C	104
Figure 4.21	Top view (above) of FESEM images and grain size distribution (below) of YSZ thin films with three-layers sintered at 1300 $^{\circ}$ C	105
Figure 4.22	XRD patterns for YSZ thin films with five- and 10-layers coating sintered at 1300 $^{\circ}$ C	108
Figure 4.23	3D Images of AFM for YSZ thin films with (a) five-layers and (b) 10-layers coating sintered at 1300 $^{\circ}$ C	109
Figure 4.24	Surface view (above) and grain size distribution (below) of (a) YSZ5_13 and (b) YSZ10_13 sintered at 1300 °C	111

LIST OF ABBREVIATIONS

SOFC	-	Solid Oxide Fuel Cell
YSZ	-	Yttria Stabilized Zirconia
XRD	-	X-ray Diffractometer
AFM	-	Atomic Force Microscopy
FESEM	-	Field Emission Scanning Electron Microscopy
UV-	-	Ultra Violet Visible
Visible		
UTM	-	Universiti Teknologi Malaysia
SDGs	-	Sustainable Development Goals
Y_2O_3	-	Yttria Oxide
RAMAN	-	Raman Spectroscopy
NO _x	-	Nitrogen Oxide
CO_2	-	Carbon Dioxide
PEMFC	-	Proton Exchange Membrane Fuel Cell
		\mathbf{D}^{\prime} (M (1 1 $\mathbf{\Gamma}$ 1 \mathbf{C} 11
DMFC	-	Direct Methanol Fuel Cell
DMFC MCFC	-	Molten Carbonate Fuel Cell
	- -	
MCFC	- - -	Molten Carbonate Fuel Cell
MCFC EPD		Molten Carbonate Fuel Cell Electrophoretic Deposition
MCFC EPD NiO		Molten Carbonate Fuel Cell Electrophoretic Deposition Nickel Oxide
MCFC EPD NiO LSC		Molten Carbonate Fuel Cell Electrophoretic Deposition Nickel Oxide Lanthanum Strontium Cobaltite
MCFC EPD NiO LSC LSM		Molten Carbonate Fuel Cell Electrophoretic Deposition Nickel Oxide Lanthanum Strontium Cobaltite Lanthanum Strontium Manganite
MCFC EPD NiO LSC LSM GDC		Molten Carbonate Fuel Cell Electrophoretic Deposition Nickel Oxide Lanthanum Strontium Cobaltite Lanthanum Strontium Manganite Gadolinium deposited Ceria
MCFC EPD NiO LSC LSM GDC Ni-YSZ		Molten Carbonate Fuel Cell Electrophoretic Deposition Nickel Oxide Lanthanum Strontium Cobaltite Lanthanum Strontium Manganite Gadolinium deposited Ceria Nickel-Yttria Stabilized Zirconia
MCFC EPD NiO LSC LSM GDC Ni-YSZ TEC		Molten Carbonate Fuel Cell Electrophoretic Deposition Nickel Oxide Lanthanum Strontium Cobaltite Lanthanum Strontium Manganite Gadolinium deposited Ceria Nickel-Yttria Stabilized Zirconia Thermal Expansion Coefficient

LIST OF SYMBOLS

MW	-	Megawatt
kW	-	Kilowatt
°C	-	Celsius
W	-	Watt
E, e	-	Electron
f	-	Lattice mismatch
as	-	the unstrained lattice constant of the substrate
a _f	-	unstrained lattice constant of the film
М	-	Metal
R	-	An alkyl group
Х	-	Valence state of the metal
XOH	-	Hydroxylated compounds
V	-	Volt
g	-	Gram
cm	-	Centimetre
nm	-	Nanometre
m	-	Meter
μm	-	Micrometre
Mpa	-	Megapascal
Psi	-	Pounds per square inch
Κ	-	Kelvin
ŋ	-	Refractive Index
ρ	-	Density
k	-	Thermal Conductivity
$C_{ m P}$	-	Specific Heat
LCA	-	Life Cycle Assessment

LIST OF APPENDICES

APPENDIX	TITLE	PAGE
	The figure of YSZ suspension preparation using dip- coating method	144
Appendix 3.B I	nstruments for Characterization	145
Appendix 4.A C	Calculation of crystallite size from XRD analysis	149
Appendix 4.B C	Calculation of lattice parameter from XRD analysis	156
Appendix 4.C C	Calculation of microstrain from XRD analysis	159
Appendix 4.D (Calculation of conductivity	166
Appendix 4.E C	Calculation of Film Density	168

CHAPTER 1

INTRODUCTION

1.1 Background

A thin film is a low-dimensional substance made by condensing atomic, molecular, or ionic species of matter one at a time. A thin film's thickness is usually less than a few microns. For more than half a century, thin films have been used in electrical appliances, hard coatings industry, optical coatings, and aesthetic components. Thin films are still being made on a regular basis these days, as they are an important component in the creation of novel materials in the twenty-first century, such as nanometre materials. Since the quantity utilized is restricted to only the surface or as a thin film layer, the materials and equipment necessary for making thin films with the least amount of hazardous ingredients are accessible. Thin film processing can also minimise energy usage during production, making it an ecologically friendly material technology for the twenty-first century (Wasa *et al.*, 2004).

Thermal evaporation, chemical breakdown, or irradiation of energetic particles or photons are all methods for depositing thin films on a substrate. Film structure, crystal phase and orientation, thickness, and microstructure are all important variables to consider. All of these properties can be controlled by deposition conditions. Other than these, thin film acquires several unique properties compared to bulk materials, such as the unique properties of materials from the atomic growth process and the size of the effect can be characterized by the thickness, orientation of crystallization and multilayer characteristics (Wasa *et al.*, 2004).

Thin film technology can be used in energy applications like fuel cells and solar cells. It is needed in improving the usefulness of the cells, reducing working temperature and expanding their life span. Fuel cells produce electrical energy from chemical energy. They generate energy through the oxidation of natural gas, hydrogen

or other fuels. They can operate continuously as long as they can access to the supply such as hydrogen and oxygen. Fuel cell applications may be divided into three categories: portable power production, fixed power generation, and transportation power. There are many different types of fuel cells, but they always include an anode, a cathode, and an electrolyte that permits positively charged hydrogen ions to travel between them. Solid Oxide Fuel Cells (SOFCs) is a major area of interest within the field of fuel cells. SOFCs use a solid ceramic oxide as an electrolyte. SOFCs operate at a high operating temperature of around 800 °C to 1000 °C. Current research tends to focus on lowering the operating temperature so that SOFCs can also be applied to suitable portable devices such as for transportation or mobile devices. Usually, power generation for residential use is around 1-10kW, while for commercial use, it is around 2-250MW (Choudhury et al., 2013). However, by decreasing the operating temperature, the electrolyte ohmic resistance losses will increase due to the thermally activated behaviour of ionic transport (Litzelman et al., 2008). Introducing thin electrolyte layers into SOFCs can decrease the internal cell resistance and thus drastically enhance the power density (Menzler et al., 2013).

This SOFC research has been using thin film for a long time since this technology is the most suitable for SOFC application. William Groove introduced the fuel cell idea for the first time in 1893. Grove was studying the electrolysis of water when he noticed that when the current was turned off, a small current continued to flow through the circuit in the opposite direction as a result of a reaction between the electrolysis products, hydrogen and oxygen, which was catalysed by the platinum electrodes. Grove also made the critically important insight that there must be a "notable surface of action" between the gas, the electrode, and the electrolyte phases in a cell. Grove recognised the idea of connecting several of these in series to create a gaseous voltaic battery. Maximizing the "triple-phase boundary," or the area of contact between the electrode, electrolyte, and gaseous reagent, continues to be at the forefront of fuel cell research and development (Ormerod, 2003)

The choice of YSZ as the material of electrolyte based on its high ionic conductivity material, stable in both oxidizing and reducing environments and good high temperature mechanical properties (Jacobson, 2010).

Ultimately, we are targeting to contribute by giving ideas in relation to solid oxide fuel cells, to achieve high-efficiency cells to make it applicable in people's daily lives. By achieving this, energy would be saved at a time when the climate is changing and global energy demand is continuously increasing. Thus, we have undertaken the task by increasing the output performance of SOFC electrolyte thin films.

On 25 September 2015, United Nations Member States met at the United Nations in New York and adopted the 2030 Agenda for Sustainable Development. These 17 Sustainable Development Goals (SDGs) set out an agenda for sustainable development, peace and prosperity for all people and the planet, for now and for a future that embraces growth in the economy and social inclusion while protecting the environment. Some issues supported in the SDGs were water accessibility, energy security, urbanization, transportation and climate crisis. Out of 17 SDGs, this research had targeted 4 goals and would be able to directly contribute to 2 goals. They were Goal 7: Affordable and Clean Energy and Goal 13: Climate Action. Another 2 goals which we have indirectly focused on were Goal 3: Good Health and Well-being and Goal 11: Sustainable Cities and Communities (Kroll *et al.*, 2019).

1.2 Problem Statement

During the last two decades, with the energy crisis and renewable energy development, SOFC materials have emerged as a starting material on the horizon of material science to overcome the problem of environmental pollution resulting from fossil fuel burning and excessive use of non-renewable energy such as from fossil fuel, crude oil and natural gas. Extensive research has been performed to find the optimum YSZ material for making the SOFC electrolyte (da Silva and de Souza, 2017; Abdalla *et al.*, 2018; Zakaria *et al.*, 2020). Various researchers have produced SOFC thin films mostly with high-temperature processing (more than 100 °C), but only a few reports exist on low-temperature processing (room temperature) (Ramesh *et al.*, 2016; Son *et al.*, 2018). However, the quality of the grown film depends on the optimum thickness, structural and morphological properties. To determine the effect of optimum growth

parameters on these properties' thorough and careful characterization and preparation of the samples are required.

Several studies have been performed to determine the effect of sintering temperature and sol-gel processing parameters using different methods. However, it has been found that growth temperature is the most crucial and significantly affects the layer properties and the layer thickness. Based on these facts, it has been found that most of the techniques had used high temperature and had required a lot of other processing parameters which makes it more expensive and complicated compared to the low-temperature processing method, where samples prepared by sol-gel method can be used to achieve better properties (Lee et al., 2016; Ma et al., 2016; Venkataramana et al., 2017). One of the advantages of the sol-gel method compared to other methods is that densification can easily be achieved at low temperatures. As an inexpensive process, chemical composition of the product can be controlled. A small amount of base material can be applied to the solution and can take part in affecting the final result (Lee et al., 2016). However, the regularity of the thin film structure, size of the particle and morphology of the particle distribution has not yet been clarified. In this view, the present study attempted to overcome the limitation of past studies to try to improve the properties by making a high-quality film for SOFC electrolyte application.

The requirement for low temperature during thin films deposition is needed to obtain high electrical conductivity of the electrolyte so that SOFC can be operated at room temperature since there has been increasing concern that SOFCs are being disadvantaged when operating at high temperature due to its start-up time, materials demand and application being limited to only large stationary sources (Timurkutluk *et al.*, 2016).

It is important to establish the correlation among the low-temperature processing, environmental friendliness and the cost-effectiveness of this sol-gel grown thin film. Therefore, it was important for the SOFC application that any material must meet the environmental sustainability criteria, especially since the SOFC material was for the renewable energy sector. As per this rationale, the present research work attempted to establish this yet-to-be established correlation.

1.3 Research Objectives

This research had three objectives, which were as follows:

- (a) To synthesis YSZ thin films in soda lime glass and sapphire substrates using sol-gel dip-coating technique;
- (b) To determine the effects of growth parameters such as substrates, coating layer numbers and sintering temperatures on the modified structural, morphological and electrical characteristics of YSZ thin films;
- (c) To optimize the growth parameters of YSZ thin films via structural, morphological and electrical characterizations useful for SOFC applications.

1.4 Scope of Study

The study used the method of sol-gel by dip-coating of YSZ electrolyte thin films. Growth parameters for the experiment included layers of coating (1 to 10 layers) and sintering temperature (300 °C to 1300 °C). The time of coating (60 seconds) and speed of coating (150 mm/min) remained constant.

The solid electrolyte material had been chosen due to the general criteria, including strong ionic conductivity, low electronic conductivity, oxidizing and reducing environment durability, and excellent mechanical properties (Jacobson, 2010; Biswas and Sadanala, 2013). Zirconia-based oxide ion fulfilled the criteria mentioned above. Addition of rare earth, alkaline earth and lanthanide oxides could

stabilize monoclinic zirconia to cubic fluorite phase at room temperature. The ionic conductivity also increases the oxygen partial pressure range during the SOFC operation. One of the oxides which needed to be used as a dopant for zirconia was yttria (Y_2O_3) (Figueiredo and Marques, 2013). Yttria-Stabilized Zirconia (YSZ) has a wide temperature range starting from room temperature until 2300 °C to achieve stability. In this study, special attention was given to the characteristic of the electrolyte thin film. Dense and low-porosity thin film is very important for ionic transportation when electrons transfer from anode to cathode. The stabilization of the cubic phase was needed to achieve very high ionic conductivity. The solid solution of cubic form induces the presence of oxygen vacancies according to the substitution of Zr^{4+} and Y^{3+} in the cationic network (Hidalgo *et al.*, 2011; Figueiredo and Marques, 2013).

Sapphire substrate was chosen in this research due to its nearest lattice mismatched between the substrate and thin film. YSZ lattice constant is 5.12 while sapphire is a: 4.785, c: 12.991. Additionally, to produce high crystallinity YSZ film, a high sintering temperature is needed. Sapphire substrate melting point can be up to 2040 °C. Glass substrate was used to test and compare the results. However, glass substrate can only withstand up to 700 °C which is not sufficient for YSZ thin film to achieve crystallinity. The dip-coating method is the simplest method to produce a thin film. Apart from that, it is easier to control the thickness of the film and other growth parameters. Time (60 seconds) and speed (150 mm/min) of coating remained unchanged. For growth parameters, the layer of coating (1 to 10 layers) and sintering temperature (1000 °C to 1500 °C) (Zhou *et al.*, 2014; Jang *et al.*, 2016) were discussed. The sintering temperature and coating layers is a very important parameters in determining a suitable temperature to produce high-quality YSZ thin film.

The properties of each component were to be characterized using X-ray Diffractometer (XRD) and field emission scanning electron microscope (FESEM) for the phase and structure of samples. Field Emission Scanning Electron Microscope (FESEM), Raman Spectroscopy (RAMAN) and atomic force microscope (AFM) were to be used for morphological characterization. The 4-point probe was to be used for conductivity and resistivity of electrolyte thin film as an application in SOFC.

Structural and electrical properties are important in producing YSZ thin film. From structural properties we can identify crystallization of thin film, phase structure and crystallite size. Morphologies of thin film would tell whether the thin film is dense as well as whether it had no-crack properties and pores. All these properties would help to get high conductivity and less resistivity in terms of electrical properties.

1.5 Significance of Study

The findings of this study may help other researchers when investigating the effects of Yttria-Stabilized Zirconia thin film using the sol-gel method in term of electrical, structural and morphological characteristics when deposited on the sapphire wafer and soda lime glass for the solid oxide fuel cell application. This study could also contribute to the improvement and development of low pollution and clean technology. The technique used in this study which is sol-gel dip-coating method is one of the simplest and low-cost method to fabricate thin film thus can contribute in cost-effectiveness of the sample. The use of thin film in this study also supports the increasing interest in nanotechnology. In the present, the thin film is useful for small and portable devices. So, this research could help researchers to improve the performance of SOFC electrolyte thin films. This study will also assist to gather new understanding and important to discover variations in resistivity, conductivity, crystal structure, morphologies, thickness, and porosity of a YSZ electrolyte useful for SOFC applications.

1.6 Outline of Thesis

This thesis explains the preparation and synthesis of growth YSZ thin film on soda lime glass and sapphire substrate, characterization and analysis of various properties of the thin film. The simple and less-cost wet deposition, which was the solgel dip-coating method, was used to prepare the thin film. The sintering heat treatment process was applied to produce growth thin film. The optimization of the YSZ thin film finally were determined based on the analysis given. Chapter 1 presents an introduction and the background of the current study. The problem statement, objectives, scope, limitations of the study and significance of the study have been discussed in this section.

Chapter 2 explains relevant literature regarding SOFC applications including type of SOFC, components of SOFC, thin film technologies with growth mechanism explanation and technique used and lastly the mechanism of characterization used to analyse thin film.

Chapter 3 describes the experimental procedure to prepare and to synthesize the YSZ thin film. In addition, information regarding the characterization of the thin film based on its structural and electrical properties have been emphasized.

Chapter 4 highlights all the results obtained from the comparison test of the different substrates and choosing a suitable one. The growth parameter in terms of sintering temperature and different coating layers were chosen. The optimum result from the parameter was determined by characterization analysis using XRD, AFM, and Raman for structural properties, FESEM was used to study morphology and a four-point probe was used to investigate the electrical properties.

Chapter 5 concludes the entire thesis that has been carried out to fulfil the objectives. The most optimized YSZ thin film electrolyte that could be used in SOFC application was prepared. By using the less-cost method, the production cost of the SOFC could be controlled, which is very useful as SOFC needs to be widely used and well-known in the future for better sustainable energy in line with protecting our environment with increased use of green energy.

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